University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: CJP/ Course name: Academic English PFAJAKA/07

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: combined, present

Number of credits: 2

Recommended semester/trimester of the course:

Course level: I., II., N

Prerequisities:

Conditions for course completion:

kontrolný písomný test, aktivita na hodine

záverečný písomný test povolené max. 2 absencie

stupnica hodnotenia: A 93-100, B 86-92, C 79-85, D 72-78, E 65-71, FX 64 a menej

aktivita na hodine

predmet končí hodnotením, t.j. povolený je 1 opravný test

Learning outcomes:

Osvojenie si a rozvíjanie užitočných techník akademického písomného ako aj ústneho prejavu so zameraním na rozvoj jazykových kompetencií študenta, na upevňovanie a rozvíjanie všetkých jazykových zručností na stredne pokročilej až pokročilej úrovni ovládania jazyka (B2/C1 podľa Spoločného európskeho referenčného rámca pre jazyky). Predmet kladie dôraz na používanie akademickej angličtiny v akademickom prostredí.

Brief outline of the course:

Akademická angličtina a jej charakteristiky

Čítanie odborných článkov, analýza, parafrázovanie

Spájacie slová v akademickom písaní

Formálna a neformálna angličtina a ich črty

Vyjadrovanie príčiny, následku v akademickom jazyku

Čítanie odbornej publikácie, analýza, parafrázovanie

Slovotvorba v anglickom jazyku- predpony a prípony

Ako prezentovať v angličtine

Parafrázovanie a definovanie

Ako písať abstrakt

Slovosled v akademickom diškurze

Recommended literature:

Seal B.: Academic Encounters, CUP, 2002

T. Armer: Cambridge English for Scientists, CUP 2011

M. McCarthy M., O'Dell F. - Academic Vocabulary in Use, CUP 2008

Zemach, D.E, Rumisek, L.A: Academic Writing, Macmillan 2005

Olsen, A.: Active Vocabulary, Pearson, 2013

www.bbclearningenglish.com

Cambridge Academic Content Dictionary, CUP, 2009

Course language:

Notes:

Course assessment

Total number of assessed students: 292

A	В	С	D	Е	FX
29.11	22.26	16.1	11.3	8.22	13.01

Provides: PaedDr. Gabriela Bednáriková

Date of last modification: 06.02.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/

Course name: Analytical Chemistry

ANCHU/03

Course type, scope and the method:

Course type: Lecture / Practice

Recommended course-load (hours): Per week: 3 / 1 Per study period: 42 / 14

Course method: present

Number of credits: 6

Recommended semester/trimester of the course: 3.

Course level: I.

Prerequisities: ÚCHV/VCHU/14 or ÚCHV/VCHU/15 or ÚCHV/VCHU/10 or ÚCHV/VACH/10

Conditions for course completion:

3x test of analytical calculations.

Examination

Learning outcomes:

Survey of basic principles and tasks of analytical chemistry and applications of analytical methods in research and practice.

Brief outline of the course:

Subject and role of analytical chemistry. General principles and procedures - sampling, sample pretreatment. Preparation of solutions. Evaluation of the results.

Classification of analytical reactions. Qualitative analysis of cations and anions. Basic principles of organic analysis.

Methods of quantitative analysis. General principles of gravimetry. Volumetric analysis.

Instrumental methods of analytical chemistry (basic principles, instrumentaion and applications) - electroanalytical, optical and separation methods.

Recommended literature:

Skoog D.A.: Principles of Instrumental Analysis. Saunders Col. Publishing, New York 1985.

D.Harvey: Modern Analytical Chemistry. McGraw Hill, Boston, 2000.

Course language:

Notes:

Course assessment

Total number of assessed students: 507

A	В	С	D	Е	FX
17.16	18.34	25.25	26.04	8.68	4.54

Provides: doc. RNDr. Taťána Gondová, CSc.

Date of last modification: 03.02.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: KFaDF/ | Course name: Antique Philosophy and Present Times

AFS/05

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of credits: 2

Recommended semester/trimester of the course: 6.

Course level: I., II.

Prerequisities:

Conditions for course completion:

Learning outcomes:

Brief outline of the course:

Recommended literature:

Course language:

Notes:

Course assessment

Total number of assessed students: 30

A	В	С	D	Е	FX
83.33	6.67	6.67	0.0	3.33	0.0

Provides: doc. PhDr. Pavol Tholt, PhD., mim.prof., Doc. PhDr. Peter Nezník, CSc.

Date of last modification: 26.01.2014

University: P. J. Šafárik University in Košice						
Faculty: Faculty of Science						
Course ID: ÚFV/ ZP2a/04	Course name: Bachelor T	hesis				
Course type, scope a Course type: Recommended cou Per week: Per stud Course method: pre	rse-load (hours): ly period: esent					
Number of credits: 2						
	ster/trimester of the cours	e: 5.				
Course level: I.						
Prerequisities:						
Conditions for cours	se completion:					
Learning outcomes:						
Brief outline of the c	ourse:					
Recommended litera	nture:					
Course language:						
Notes:	,					
Course assessment Total number of asse	ssed students: 59					
	abs	n				
100.0 0.0						
Provides: doc. RNDr	Provides: doc. RNDr. Rastislav Varga, DrSc.					
Date of last modifica	Date of last modification: 18.02.2014					
Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.						

University: P. J. Šafárik University in Košice						
Faculty: Faculty of Science						
Course ID: ÚFV/ ZP2b/04	Course name: Bachelor T	nesis				
Course type, scope and the method: Course type: Recommended course-load (hours): Per week: Per study period: Course method: present						
Number of credits: 6						
Recommended seme	ster/trimester of the cours	e: 6.				
Course level: I.						
Prerequisities:						
Conditions for cours	e completion:					
Learning outcomes:						
Brief outline of the c	ourse:					
Recommended litera	iture:					
Course language:						
Notes:						
Course assessment Total number of assessed students: 57						
	abs	n				
100.0 0.0						
Provides: doc. RNDr. Rastislav Varga, DrSc.						
Date of last modification: 18.02.2014						
Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.						

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ Course name: Bachelor Work BP1a/04 Course type, scope and the method: **Course type: Recommended course-load (hours):** Per week: Per study period: Course method: present Number of credits: 2 **Recommended semester/trimester of the course:** 5. Course level: I. **Prerequisities: Conditions for course completion: Learning outcomes:** Individual scientific work of students. Publishing of obtained results in a written form and as a public presentation. **Brief outline of the course: Recommended literature: Course language: Notes:** Course assessment Total number of assessed students: 384 abs n 100.0 0.0 **Provides:** Date of last modification: 03.02.2014 Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ Course name: Bachelor Work BP1b/04 Course type, scope and the method: **Course type: Recommended course-load (hours):** Per week: Per study period: Course method: present **Number of credits:** 6 Recommended semester/trimester of the course: 6. Course level: I. **Prerequisities: Conditions for course completion: Learning outcomes:** Individual scientific work of students. Publishing of obtained results in a written form and as a public presentation. **Brief outline of the course: Recommended literature: Course language: Notes:** Course assessment Total number of assessed students: 387 abs n 99.48 0.52 **Provides:** Date of last modification: 03.02.2014 **Approved:** prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/

Course name: Basis of Mineralogy

MINU/03

Course type, scope and the method:

Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14

Course method: present

Number of credits: 5

Recommended semester/trimester of the course: 4.

Course level: I.

Prerequisities: ÚCHV/VCHU/10 or ÚCHV/ZAC2/10 or ÚCHV/VACH/10 or ÚCHV/CHG/09 or

ÚCHV/ZCF/03

Conditions for course completion:

Verification of theoretical knowledge and recognizing minerals.

Semester project, practical test from recognizing of minerals, optional oral examination.

Learning outcomes:

To recognize the beauty of nature and to obtain basic knowledge from mineralogy. To familiarize students with properties of usual minerals and to recognize these minerals.

Brief outline of the course:

Basic terms and definitions, origin of minerals in nature. Basis of morphological and structural crystallography: characteristic properties of crystals, crystallographic laws, crystal structure, unit cells and their parameters, crystallographic systems with examples of minerals. Crystallochemistry: types of bonds and structures and their effect on the properties of minerals. Physical properties of minerals and their utilize in minerals classification. Basis of genetic and systematic mineralogy. Structure of silicates.

Recommended literature:

M. Košuth: Mineralógia. Elfa, s.r.o. Košice, 2001 V. Radzo: Mineralógia, Alfa Bratislava, 1987.

Course language:

Slovak

Notes:

Course assessment

Total number of assessed students: 40

A	В	С	D	Е	FX
67.5	25.0	7.5	0.0	0.0	0.0

Provides: doc. RNDr. Ivan Potočňák, PhD.

Date of last modification: 03.02.2014

COURSE INFORMATION LETTER
University: P. J. Šafárik University in Košice
Faculty: Faculty of Science
Course ID: ÚCHV/ BCHU/03 Course name: Biochemistry
Course type, scope and the method: Course type: Lecture Recommended course-load (hours): Per week: 3 Per study period: 42 Course method: present
Number of credits: 5
Recommended semester/trimester of the course: 5.
Course level: I.
Prerequisities: ÚCHV/VCHU/10 or ÚCHV/VCHU/15 or ÚCHV/VACH/10
Conditions for course completion: test + oral examination
Learning outcomes: The aim of biochemistry teaching is to acquire knowledge in the field of living organisms on the basis of their molecular structure and metabolism.
Brief outline of the course: 1. Protein Structure and Function, Exploring proteins 2. DNA and RNA and the Flow of Genetic Information, Exploring genes 3. Enzymes: Basic Concepts and Kinetics, Catalytic Strategies and Regulatory Strategies 4. Carbohydrates (Monosaccharides, Disaccharides, Polysaccharides – Functions and Properties) 5. Lipids and Cells Membranes, Membrane Channels and Pumps 6. Metabolis: Basic Concepts and Design, Signal-Transduction Pathways 7. Glycolysis and Gluconeogenesis, Glycogen Metabolism 8. The Citric Acid Cycle and Glyoxylate Cycle 9. Oxidative Phosphorylation, The Light Reactions of Photosyntesis 10. The Calvine Cycle and the Pentose Phosphate Pathway 11. Fatty Acids Metabolism, Urea Cycle 12. DNA Replication, Transcription (RNA Synthesis) 13. Protein Synthesis & Degradation, the Integration of Metabolism
Recommended literature: Škárka: Biochémia. Alfa, 1992 Voet a Voetová: Biochemie. Victoria Publishing, Praha, 1994 Stryer, L.: Biochemistry, W.H. Freeman and Company, New York, 1988 Course language:

Notes:

Course assessment						
Total number of assessed students: 876						
Α	В	С	D	Е	FX	
16.44	18.49	21.12	21.92	19.06	2.97	

Provides: doc. RNDr. Erik Sedlák, PhD., RNDr. Nataša Tomášková, PhD.

Date of last modification: 03.02.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/

Course name: Biochemistry Practical

PBCHU/03

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 4 Per study period: 56

Course method: present

Number of credits: 5

Recommended semester/trimester of the course: 6.

Course level: L

Prerequisities: ÚCHV/BCHU/03

Conditions for course completion:

Protocols + 75 % continuous evaluation.

Learning outcomes:

Brief outline of the course:

The most important biochemical laboratory methods. The qualitative tests for amino acids and proteins. Time-dependent course of enzyme-catalyzed reaction: determination of enzymatic activity, determination of the first order rate constant, calculations of math models (examples), effect of a substrate concentration on initial rate of reaction, determination of Km and Vmax for urease. Isolation and detection of nucleic acids.

Recommended literature:

http://kosice.upjs.sk/~kbch/

Course language:

Notes:

Course assessment

Total number of assessed students: 272

A	В	С	D	Е	FX
58.09	25.0	11.76	2.21	2.21	0.74

Provides: doc. RNDr. Mária Kožurková, CSc., RNDr. Nataša Tomášková, PhD., RNDr. Rastislav Varhač, PhD., RNDr. Danica Sabolová, PhD., Mgr. Eva Žilecká

Date of last modification: 03.02.2014

Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.

Page: 13

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ Co

Course name: Bioinorganic Chemistry I

BAC1/04

Course type, scope and the method: Course type: Lecture / Practice

Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14

Course method: present

Number of credits: 5

Recommended semester/trimester of the course: 5.

Course level: I., II.

Prerequisities:

Conditions for course completion:

Test or seminar works

examination

Learning outcomes:

The basic knowledges about biometal interactions with biomolecules, biomaterials, biominerals, biocatalysis, metals in biology and medicine, metal-based drugs, toxic metals for biosystems and metals in the environment.

Brief outline of the course:

Metalic and non-metalic elements and their roles in biological systems (biometals, bulk biological elements, essential trace elements). Biocoordination compounds, bioligands. Biocatalyzers. Oxygen carriers and oxygen transport proteins. Photochemical process. Catalysis and regulation processes. Calcium biominerals and biomineralization. Toxic metals. Application of knowledge of bioinorganic chemistry in pharmacy, chemotherapy (e.g. platinum complexes in cancer therapy) radiodiagnostics, mineral biotechnology, ecology and in other branches of life.

Recommended literature:

- 1. Shriver D. F., Atkins P. W., Overton T. L., Rourke J.P., Weller M.T., Amstrong F.A.: Shiver & Atkins. Inorganic Chemistry. Oxford University Press, Oxford 2006.
- 2. Kaim W., Schwederski B.: Bioinorganic Chemistry: Inorganic Elements in the Chemistry of Life. Wiley, Chichester 1998.
- 3. Wilkins P. C., Wilkins R. G.: Inorganic Chemistry in Biology. OCP, Oxford 1997.

Course language:

Notes:

Course assessment

Total number of assessed students: 145

A	В	С	D	Е	FX
44.14	31.03	15.86	2.07	6.9	0.0

Provides: doc. RNDr. Zuzana Vargová, Ph.D.

 $\textbf{Date of last modification:}\ 03.02.2014$

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚBEV/

Course name: Biology of Children and Adolescents

BDD/05

Course type, scope and the method:

Course type: Lecture / Practice

Recommended course-load (hours): Per week: 2 / 0 Per study period: 28 / 0

Course method: present

Number of credits: 2

Recommended semester/trimester of the course: 4., 6.

Course level: I.

Prerequisities:

Conditions for course completion:

Written test

Learning outcomes:

The aim of the subject is to gain the particular level of knowledge about human body and its development. It is neccessary for the understanding of specific biological characteristics of children and adolescents linked to development.

Brief outline of the course:

Human ontogenesis. Postnatal development. Age specific features of skeletal and muscalar, circulatory, respiratory, gastrointestinal and urinary systems. Reproductive system. Endocrine system. Nervous system. Age specifics of selected diseases and drug dependence arise. Human population and environment.

Recommended literature:

Drobný I., Drobná M.: Biológia dieťaťa pre špeciálnych pedagógov I. a II. Bratislava, PdF UK, 2000

Lipková V.: Somatický a fyziologický vývoj dieťaťa. Osveta Bratislava, 1980

Malá H., Klementa J.: Biológia detí a dorastu. Bratislava, SPN, 1989

Course language:

Notes:

Course assessment

Total number of assessed students: 1069

A	В	С	D	Е	FX
36.3	23.48	16.0	15.9	7.86	0.47

Provides: doc. RNDr. Monika Kassayová, CSc.

Date of last modification: 13.02.2014

Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.

Page: 16

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: KGER/ **Course name:** Communication Competence in the German Language NJKK/07 Course type, scope and the method: Course type: Practice **Recommended course-load (hours):** Per week: 2 Per study period: 28 Course method: present Number of credits: 2 Recommended semester/trimester of the course: Course level: I., II. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 42 \mathbf{C} Α В D Е FX 57.14 14.29 7.14 4.76 14.29 2.38 Provides: Mgr. Eva Černáková, PhD. Date of last modification: 05.02.2014 Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.

Page: 17

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: CJP/ Course name: Communicative Competence in English

PFAJKKA/07

(A/07

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: combined, present

Number of credits: 2

Recommended semester/trimester of the course:

Course level: I., II., N

Prerequisities:

Conditions for course completion:

ontrolný písomný test, aktivita na hodine

záverečný písomný test

stupnica hodnotenia A 93-100, B 86 - 92, C 79-85, D 72-78, E 65-71, FX menej ako 64

Povolené max. 2 absencie počas semestra

predmet končí hodnotením, možnosť jedného opravného testu

Learning outcomes:

Uplatnenie a aktívne používanie svojich teoretických vedomostí v praktických komunikačných situáciách. Zdokonalenie jazykových vedomostí a zručností študenta, rečovej, pragmatickej a vecnej kompetencie, predovšetkým zlepšujú komunikáciu, schopnosť prijímať a formulovať výpovede, efektívne vyjadrovať svoje myšlienky ako aj orientovať sa v obsahovom pláne výpovede. Precvičovanie rečových intencií kontaktných (napr. pozdravy, oslovenia, pozvanie, oslovenie), informatívnych (napr. získavanie a podávanie informácií, vyjadrenie priestorových a časových vzťahov), regulačných (napr. prosba, poďakovanie, zákaz, pochvala, súhlas, nesúhlas) a hodnotiacich (napr. vyjadrenie vlastného názoru, stanoviska, želania, emócií). Výsledkom budovania praktickej jazykovej kompetencie majú byť vedomosti a zručnosti zodpovedajúce požiadavkám a kritériám dokumentu Spoločný európsky referenčný rámec pre vyučovanie jazykov - úroveň B2.

Brief outline of the course:

Rodina, jej formy a problémy

Vyjadrovanie pocitov a dojmov

Dom, bývanie a budúcnosť

Formy a dialekty v anglickom jazyku

Život v meste a na vidieku

Kolokácie a idiomy, zaužívané slovné spojenia

Prázdniny a sviatky vo svete

Životné prostredie a ekológia

Výnimky zo slovosledu

Frázové slovesá a ich použitie

Charakteristiky neformálneho diškurzu

Recommended literature:

McCarthy M., O'Dell F.: English Vocabulary in Use, 1994

Misztal M.: Thematic Vocabulary, 1998

Fictumova J., Ceccarelli J., Long T.: Angličtina, konverzace pro pokročilé, Barrister and

Principal, 2008

Peters S., Gráf T.: Time to practise, Polyglot, 2007

www.bbclearningenglish.com

Jones L.: Communicative Grammar Practice, CUP, 1985 Alexander L.G.: Longman English Grammar, Longman, 1988

Course language:

Notes:

Course assessment

Total number of assessed students: 174

A	В	С	D	Е	FX
36.78	22.41	18.39	9.77	8.05	4.6

Provides: PaedDr. Gabriela Bednáriková, Mgr. Silvia Marcinová, PhD.

Date of last modification: 06.02.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: CJP/ Course name: Communicative Grammar in English

PFAJGA/07

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: combined, present

Number of credits: 2

Recommended semester/trimester of the course:

Course level: I., II., N

Prerequisities:

Conditions for course completion:

kontrolná písomná práca, záverečná písomná práca

stupnica hodnotenia: A 93-100, B 86-92, C 79-85, D 65-71, 64 a menej - FX

aktivita na hodinách, povolené 2 absencie

predmet je ukončený hodnotením, možnosť jedného opravného testu

Learning outcomes:

Identifikovanie a odstránenie najfrekventovanejších gramatických chýb v ústnom prejave, ako aj v písomnom styku. Rozvoj jazykových kompetencií študenta so zameraním na funkcie gramatiky anglického jazyka v každodennej interakcii, v komunikačnom akte na stredne pokročilej úrovni ovládania jazyka (B2 podľa Spoločného európskeho referenčného rámca pre jazyky).

Brief outline of the course:

Zvieratá a rastliny na zemi

Zločin a trest

Cestovanie po mori a vzduchom

Jedlá a reštaurácie, národná kuchyňa

Vzdelanie na vysokých školách

História a viera

Vybrané problémy anglickej výslovnosti, gramatiky (nepriama reč, slovotvorba, predložkové väzby, anglická syntax, kondicionály v angličtine a slovnej zásoby príslušného zamerania Vybrané funkcie praktického odborného jazyka potrebné na prácu s odborným textom

Recommended literature:

Misztal M.: Thematic Vocabulary, 1994

McCarthy, O'Dell: English Vocabulary in Use, 1994

Alexander L.G.: Longman English Grammar, Longman, 1988 Jones I. - Communicative Grammar Practice, CUP, 1992

Vince M.: Macmillan Grammar in Context, Macmillan, 2008

www.bbclearningenglish.com

Gráf T., Peters S.: Time to practise, Polyglot, 2007

Course language:									
Notes:									
Course assessr Total number of	nent of assessed studen	ts: 378							
A	В	B C D E FX							
39.42	42 18.25 17.2 8.73 5.82 10.58								
Provides: PaedDr. Gabriela Bednáriková									
Date of last modification: 06.02.2014									
Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.									

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚFV/ Course name: Computer-Based Physical Measurement

PPFM/08

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of credits: 3

Recommended semester/trimester of the course: 4.

Course level: L

Prerequisities:

Conditions for course completion:

active participation at all labworks

written laboratory records with data analysis

Learning outcomes:

Students is able to measure physical quantities and gains skills important for measuring and data processing with the help of computer. The result is deeper conceptual understanding of physical phenomena involved in the labworks that is connected mainly with the content of courses General Physics I,II,III.

Brief outline of the course:

The content of the course involves labworks in physics aimed at selected problems of General Physics I,II,II. Student learns about different methods of measurement of physical quantities, he gains skills concerning measurement and data processing with the help of computer. The set of labworks involves analysis of different phenomena followed by the data processing and written report.

Recommended literature:

- 1. Halliday, Hajko, V., Daniel-Szabó, J.: Základy fyziky, Veda Bratislava 1983
- 2. Veis, Š., Maďar, J., Martišovitš, V.: Všeobecná fyzika 1, Alfa, Bratislava, 1987
- 3. Hlavička, A. a kol.: Fyzika pre pedagogické fakulty, SPN Praha, 1971
- 4. Halliday, D., Resnick, R., Walker, J.: Fyzika, part1-4, VUT Brno, 2000

Course language:

Slovak

Notes:

Course assessment

Total number of assessed students: 37

A	В	С	D	Е	FX
40.54	40.54	18.92	0.0	0.0	0.0

Provides: doc. RNDr. Zuzana Ješková, PhD., doc. RNDr. Marián Kireš, PhD.

 $\textbf{Date of last modification:}\ 18.02.2014$

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/

Course name: Coordination Chemistry

KCHU/03

Course type, scope and the method: Course type: Lecture / Practice

Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14

Course method: present

Number of credits: 4

Recommended semester/trimester of the course: 5.

Course level: I.

Prerequisities: ÚCHV/ACHU/03

Conditions for course completion:

Final written exam

Learning outcomes:

The student acquires basic knowledge on the coordination compounds, preparation, isomerism and properties of coordination compounds as well as about the chemical bonding in coordination compounds.

Brief outline of the course:

Definition and nomenclature of coordination compounds. Central atom and ligands, coordination numbers. Isomerism, preparation and stability of coordination compounds, chemical bonding in coordination compounds.

Recommended literature:

- J. Ribas: Coordination Chemistry, Wiley-VCH, Weinheim, 2008.
- J. C. Huheey, E. A. Keiter, R. L. Keiter: Inorganic Chemistry, Haper Collins, New York, 1993.
- G. A. Lawrance: Introduction to Coordination Chemistry, Wiley, 2010.

Course language:

Notes:

Course assessment

Total number of assessed students: 60

A	В	C	D	Е	FX
56.67	23.33	13.33	3.33	3.33	0.0

Provides: prof. RNDr. Juraj Černák, CSc., RNDr. Juraj Kuchár, PhD.

Date of last modification: 03.02.2014

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ **Course name:** Defence of bachelor thesis OBPC/03 Course type, scope and the method: **Course type: Recommended course-load (hours):** Per week: Per study period: Course method: present Number of credits: 0 **Recommended semester/trimester of the course:** Course level: I. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course:** Presentation of the thesis before the state exam committee. **Recommended literature:** Course language: **Notes: Course assessment** Total number of assessed students: 366 В \mathbf{C} Ε FX Α D 83.33 10.66 4.37 0.82 0.55 0.27 **Provides:** Date of last modification: 03.02.2014

Page: 25

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚFV/ Course name: Defence of Bachelor Thesis OZP/05 Course type, scope and the method: **Course type: Recommended course-load (hours):** Per week: Per study period: Course method: present Number of credits: 0 **Recommended semester/trimester of the course:** Course level: I. **Prerequisities: Conditions for course completion:** Learning outcomes: **Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 110 C A В D Е FX 88.18 9.09 0.0 0.0 2.73 0.0 **Provides:** Date of last modification: 08.08.2014 Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: KPE/ Course name: Education-related Legislation SL1/05 Course type, scope and the method: Course type: Practice **Recommended course-load (hours):** Per week: 2 Per study period: 28 Course method: present Number of credits: 2 Recommended semester/trimester of the course: 6. Course level: I., II. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 337 C Α В D Е FX 39.17 31.16 16.91 4.15 1.78 6.82

Provides: PaedDr. Renáta Orosová, PhD., Mgr. Zuzana Nováková, PhD.

Date of last modification: 04.02.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚFV/ | **Course name:** Electonics Practical

ELP1/01

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 3 Per study period: 42

Course method: present

Number of credits: 3

Recommended semester/trimester of the course: 6.

Course level: I.

Prerequisities: ÚFV/ELE1/07

Conditions for course completion:

Debate with students during practice, trial preparation and processing of theoretical and experimental results of their defense.

Summary evaluation of student activities while working on set topics of study practices.

Learning outcomes:

Practical work of students in the design, construction and properties of the measurements of electronic circuits and interpretation of the results obtained to verify and consolidate the theoretical knowledge acquired in lectures on the subject Electronics.

Brief outline of the course:

- 1. Combinatorial logical circuits. 2.Logical memory circuits. 3. Logical sequence circuits. 4. Rectifiers, filters, stabilizers. 5. Amplifier with bipolar transistor. 6. Stabilized DC power supplies.
- 7. Generators of harmonic signals. 8. Operational amplifiers and operational network interfaces. 9. Digital-to-analog converters. 10. Analog-to-digital converters. 11 Reserve.

Recommended literature:

- 1. Delaney C.F.G.: Electronics for the Physicist with Aplications. John Willey & Sons, New York, 1980.
- 2. Zbar P.B., Malvino A.P., Miller M.A.: Basic Electronics: a Text-Lab Manual. Macmillan/McGraw Hill. New York. 1994.

Course language:

slovak or english

Notes:

Course assessment

Total number of assessed students: 26

A	В	С	D	Е	FX
100.0	0.0	0.0	0.0	0.0	0.0

Provides: doc. RNDr. Rastislav Varga, DrSc., RNDr. Erik Čižmár, PhD., Mgr. Vladimír Komanický, PhD.

 $\textbf{Date of last modification:}\ 18.02.2014$

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚFV/ | Course name: Electronics

ELE1/07

Course type, scope and the method:

Course type: Lecture

Recommended course-load (hours): Per week: 3 Per study period: 42

Course method: present

Number of credits: 5

Recommended semester/trimester of the course: 5.

Course level: I.

Prerequisities: ÚFV/VF1b/03

Conditions for course completion:

Exam

Learning outcomes:

To explain physical principles of classical electronic components and systems and technologies of their realization. To perform analysis of properties and functions of basic electronic elements, electronic circuits and information transmission and processing systems. To introduce student into basic elements and devices in area of nanoelectonics and to explain methods of their fabrication and principles of their functioning.

Brief outline of the course:

Structure, properties and physical principles of the activity of selected electronic elements. Analysis of functions and properties of basic analog and digital electronic circuits. Nanoelectronics and selected building components of nanoelectronics: graphene, carbon nanotubes, selected types of nanodevices their properties, fabrication and integration to functional systems.

Recommended literature:

- 1. Brown P.B., Frantz G.N., Moraff H.: Electronics for the Modern Scientist. Elsevier, 1982.
- 2. Delaney C.F.G.: Electronics for the Physicist with Aplications. John Willey & Sons, 1980.
- 3. Wolt E. L.: Quantum Nanoelectronics, An introduction to electronic nanotechnology and quantum computing, Wiley-VCh, 2009

Course language:

Slovak

Notes:

Course assessment

Total number of assessed students: 257

A	В	С	D	Е	FX
30.35	27.24	27.63	7.78	2.72	4.28

Provides: Mgr. Vladimír Komanický, PhD.

Date of last modification: 18.02.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: CJP/

Course name: English Language of Natural Science

PFAJ4/07

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of credits: 2

Recommended semester/trimester of the course: 4.

Course level: I.

Prerequisities:

Conditions for course completion:

test na slovnú zásobu, ústna prezentácia, záverečný písomný test, účasť na seminároch (max. 2 absencie)

stupnica hodnotenia: A 93-100, B 86-92, C 79-85, D 72-78, E 65-71, FX 64 a menej

Learning outcomes:

Rozvoj jazykových kompetencií študentov príslušného študijného odboru, upevňovanie a rozvíjanie všetkých jazykových zručností (hovorenie, písanie, čítanie, počúvanie) predovšetkým v odbornej/profesnej angličtine, na stredne pokročilej úrovni ovládania jazyka (B2). Dôraz sa kladie na aktívne správne používanie odbornej/profesnej angličtiny.

Brief outline of the course:

ANGLICKÝ JAZYK PRE GEOGRAFOV:

Veda a výskum. Odbor geografia.

Planéta Zem. Naša slnečná sústava. Litosféra, hydrosféra, atmosféra, biosféra.

Zem - dynamická planéta. Tektonické platne. Sopečná činnosť.

Zemetrasenia.

Svetové oceány. Morské prúdy. Tsunami.

Veľký koralový útes.

Atmosféra - zloženie atmosféry.

Kontinenty. Európa - krajiny, národnosti.

ANGLICKÝ JAZYK PRE EKOLÓGOV:

Veda a výskum. Odbor ekológia.

Životné prostredie. Znečistenie a dôsledky.

Sopečná činnosť, zemetrasenia.

Great Pacific Garbage Patch.

Globálne otepľovanie a dôsledky. Ľadovce.

Počasie a klíma. Búrky, hurikány, tsunami.

Život na Zemi. Ohrozené rastlinné a živočíšne druhy.

ANGLICKÝ JAZYK PRE BIOLÓGOV:

veda a výskum, odbor biológia

morfológia rastlín, koreň

stonka, list

rozmnožovanie rastlín, kvet

biológia človeka - telesné sústavy

slovná zásoba z oblasti botanickej a zoologickej nomenklatúry

ANGLICKÝ JAZYK PRE MATEMATIKOV:

Veda a výskum, odbor matematika

čísla a tvary v matematike

Elementárna algebra

Elementárna geometria

Výpočty v matematike

Pytagoras, Pytagorova veta

Grafy a diagramy

Štatistika

ANGLICKÝ JAZYK PRE FYZIKOV

Veda a výskum, odbor fyzika

Atómy a molekuly

Hmota a jej premeny

Elektrina, jej využitie

Zvuka, jeho prenos

Svetlo

Solárny systém

Matematické operácie

ANGLICKÝ JAZYK PRE CHEMIKOV:

Veda a výskum, odbor chémia:

História, alchímia

Nomenklatúra

Laboratórium a jeho vybavenie

Periodická tabuľka

Hmota a jej premeny

Organická chémia

Anorganická chémia

ANGLICKÝ JAZYK PRE INFORMATIKOV:

Veda a výskum, informatika

Život s počítačom

Typický PC

Zdravie a bezpečnosť, ergonomika

Programovanie

Emailovanie

Cybercrime

Trendy budúcnosti

Recommended literature:

študijné materiálny dodané vyučujúcim

Velebná, V. English for Chemists.

Redman, S.: English Vocabulary in Use, Pre-intermetdiate, Intermediate. Cambridge University Press. 2003.

Powel, M.: Dynamic Presentations. CUP, 2010

Armer, T.: Cambridge English for Scientists. CUP, 2011

Wharton J.: Academic Encounters. The Natural World, CUP: 2009.

Murphy, R.: English Grammar in Use. Cambridge University Press. 1994.

Redman, s.: English Vocabulary in Use, Pre-intermetdiate, Intermediate. Cambridge University Press. 2003.

P. Fitzgerald: English for ICT studies, Garnet Publishing, 2011

Course language:

Notes:

Course assessment

Total number of assessed students: 1860

A	В	С	D	Е	FX
31.72	25.54	18.28	11.94	9.52	3.01

Provides: PhDr. Helena Petruňová, CSc., PaedDr. Gabriela Bednáriková, Mgr. Marianna Škultétyová, Mgr. Silvia Marcinová, PhD.

Date of last modification: 06.02.2014

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: KPE/ Course name: Essentials of School Pedagogy ZSKP/05 Course type, scope and the method: Course type: Lecture / Practice **Recommended course-load (hours):** Per week: 1 / 1 Per study period: 14 / 14 Course method: present Number of credits: 2 Recommended semester/trimester of the course: 6. Course level: I. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 268 C A В D Е FX 11.19 16.42 26.49 19.03 21.27 5.6 Provides: PaedDr. Renáta Orosová, PhD. Date of last modification: 04.02.2014 Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/

Course name: Food chemistry

PCH1/00

Course type, scope and the method: Course type: Lecture / Practice

Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14

Course method: present

Number of credits: 4

Recommended semester/trimester of the course: 5.

Course level: I., II.

Prerequisities:

Conditions for course completion:

Learning outcomes:

Based o excursions to food plants and analytical laboratories specialized on food analysis together with own prepared projects during seminars the students should gain general overview about food chemistry, basic legal documents, additives.

Brief outline of the course:

The main categories of substances in the most important group of food. Contamination of food. Physical and chemical properties of food and chemical reactions relative to obtaining, storing and preparing of food. Analytical methods for determination of quality of the food.

Recommended literature:

Course language:

Notes:

Course assessment

Total number of assessed students: 238

A	В	С	D	Е	FX
57.56	36.55	5.46	0.0	0.0	0.42

Provides: RNDr. Dušan Koščík, CSc., RNDr. Ján Elečko

Date of last modification: 03.02.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/

Course name: Fundamentals of Bioanalytical Chemistry

BACHZ/06

Course type, scope and the method:

Course type: Lecture / Practice Recommended course-load (hours):

Per week: 2 / 1 Per study period: 28 / 14

Course method: present

Number of credits: 5

Recommended semester/trimester of the course: 3.

Course level: I.

Prerequisities:

Conditions for course completion:

written test

Oral examination

Learning outcomes:

Principles and theoretical foundations the application of analytical methods in bioanalysis.

Brief outline of the course:

Introduction to Bioanalytical Chemistry. Biological samples classification. Factors that affect analytes in biological samples. Collection, transport and storage of samples, the main principles of sampling, the suppressing of undesirable phenomena. Selected methods of pretreatment of biological samples. Analyzers, equipment and organization of work in a clinical laboratory. Control and management of quality in clinical laboratory. Quality manual, calibration, control, and reference materials. Validation and Good Laboratory Practice. Buffers in bioanalysis. Enzymes in bioanalysis, introduction, distribution, Mechanism of enzyme catalysis. The kinetics of enzymatic reactions with one substrate, the Michaelis constant, constant specificity, lag phase, kinetics of reactions with two substrates. Moderators of enzyme activity. Selected methods for analysis of biomolecules.

Recommended literature:

- 1. Mikkelsen S.R, Cortón E.: Bioanalytical Chemistry, Wiley, 2004
- 2. Wilson I., Bioanalytical Separations 4, (Handbook of Analytical

Separations), Elsevier, 2003

3.Lee, D.C., Webb, M. Pharmaceutical Analysis, Blackwell, 2003

Course language:

Notes:

Course assessment

Total number of assessed students: 52

A	В	С	D	Е	FX
34.62	26.92	28.85	5.77	0.0	3.85

Provides: doc. RNDr. Katarína Reiffová, PhD.

 $\textbf{Date of last modification:}\ 03.02.2014$

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚFV/ **Course name:** General Biophysics I

VBF1/08

Course type, scope and the method:

Course type: Lecture

Recommended course-load (hours): Per week: 3 Per study period: 42

Course method: present

Number of credits: 4

Recommended semester/trimester of the course: 3., 5.

Course level: I.

Prerequisities:

Conditions for course completion:

Exam.

Learning outcomes:

To provide information about the object, significance and role of biophysics in science. The main emphasis will be given on the understanding of the principles determining the structure and function of the most important biological structures (nucleis acids, proteins, biomembranes) as well as on the thermodynamics and kinetics of selected chemical and biophysical processes.

Brief outline of the course:

The definition of biophysics and its role in the science. Intra- and inter-molecular interactions in biological systems. Function and structure of the important biomacromolecules (nucleic acids, proteins, biomembranes, sugars). Conformational transitions in biopolymers: helix-coil transition in DNA, denaturation of proteins, phase transitions in biomembranes.

Thermodynamics of biological processes. Gibbs energy and chemical equilibrium, chemical potential, binding constants of the ligand-macromolecule intractions, cooperativity of the binding between biological important molecules, membrane potential.

Kinetics of the chemical and biophysical processes. The principles of chemical kinetics, enzymatic reactions, inhibition of the enzymes, membrane transport, introduction to the pharmacokinetics.

Cell biophysics. The basic bioenergetic processes, oxidative phosphorylation, photosynthesis. Mechanisms of regulations and control processes in cells-the basic principles.

Medicinal biophysics. Biophysical principles of selected diagnostic and therapeutical methods. Radiation and environmental biophysics. The influence of physico-chemical factors of the environment on the living systems.

Recommended literature:

- 1. M. B. Jackson, Molecular and cellular biophysics, Cambridge University Press, 2006.
- 2. M. Daune, Molecular biophysics Structures in motion, Oxford University Press, 2004.
- 3. R. Glaser, Biophysics, Springer Verlag, 2001.
- 4. M.V. Volkenštein, Biofizika, Nauka, Moskva 1988.
- 5. W.Hoppe and W. Lohmann, Biophysics, Springer Verlag, 1988.
- 6. D.G. Nichols and S.J. Ferguson, Bioenergetics 3, Academic Press, Elsevier Science Ltd., 2002.
- 7. D. T. Haynie, Biological thermodynamics, Cambridge University Press, 2001.

Course language:

Slovak

Notes:

Course assessment

Total number of assessed students: 123

A	В	С	D	Е	FX
19.51	28.46	24.39	16.26	11.38	0.0

Provides: doc. Mgr. Daniel Jancura, PhD.

Date of last modification: 10.02.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ Course name:

PACU/03

Course name: General Course of Analytical Chemistry - Laboratory

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 4 Per study period: 56

Course method: present

Number of credits: 4

Recommended semester/trimester of the course: 4.

Course level: L

Prerequisities: ÚCHV/ANCHU/03

Conditions for course completion:

Assessment

Learning outcomes:

Application of theoretical knowledge to analytical laboratory practise

Brief outline of the course:

Practical in qualitative and quantitative analysis. Qualitative analysis, separation by selective precipitation. Quantitative methods. Gravimetry, general principles of method. Volumetric methods. Preparation of accurate solutions. Indication of equvivalency point. Titration curves, calculations in volumetric analysis. Acidimetry, alkalimetry. Manganometry. Iodometry. Complexometry. Selected Instrumental analytical methods.

Recommended literature:

D.Harvey: Modern Analytical Chemistry. McGraw Hill, Boston, 2000.

D.A.Skoog: Principles of Instrumental Analysis. Saunders Col. Publishing, New York 1985.

E.Prichard: Quality in the Analytical Chemistry Laboratory, Wiley, 1995

Course language:

Notes:

Course assessment

Total number of assessed students: 174

Α	В	C	D	Е	FX
53.45	30.46	12.64	1.15	2.3	0.0

Provides: RNDr. Rastislav Serbin, PhD., RNDr. Jana Šandrejová, PhD., RNDr. Lívia Kocúrová, PhD.

Date of last modification: 03.02.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/

Course name: General Chemistry

VCHU/14

Course type, scope and the method: Course type: Lecture / Practice

Recommended course-load (hours): Per week: 4 / 2 Per study period: 56 / 28

Course method: present

Number of credits: 7

Recommended semester/trimester of the course: 1.

Course level: I.

Prerequisities:

Conditions for course completion:

2 written tests

Oral examination.

Learning outcomes:

To provide students with knowledge about atoms, chemical bonds, physical properties of elements and compounds.

Brief outline of the course:

Main terms used in chemistry. Atoms – models of atoms, electron configuration, chemical periodicity and its effect on the properties of elements, radioactivity. Chemical bonds and intermolecular interactions. Chemical structure and physical properties of matter. State of matter. Solutions. Chemical equilibrium. Basis of chemical thermodynamics and chemical kinetics. Classification of chemical reactions. Electrochemistry.

Recommended literature:

- 1. Atkins P., Jones L.: Chemical Principles, 2nd ed., Freeman, New York 2002.
- 2. Russel J.B.: General Chemistry, 2nd ed., McGraw Hill, London 1992.

Course language:

Notes:

Course assessment

Total number of assessed students: 43

A	В	C	D	Е	FX
20.93	27.91	32.56	13.95	4.65	0.0

Provides: doc. RNDr. Vladimír Zeleňák, PhD.

Date of last modification: 13.03.2014

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚFV/ **Course name:** General Physics I VF1a/12 Course type, scope and the method: Course type: Lecture / Practice **Recommended course-load (hours):** Per week: 4 / 2 Per study period: 56 / 28 Course method: present Number of credits: 7 Recommended semester/trimester of the course: 1. Course level: L **Prerequisities: Conditions for course completion:** Monitoring tests during the calculus lessons 1. in the 6th week 2.in the 12th week Final assessment is based on th results of: - oral examination assessment of the calculus lessons (written tests, overall performance during the lessons) **Learning outcomes:** Basic knowledge about the mechanics, molecular physics and thermodynamics. **Brief outline of the course:** Basic knowledge of the calculus, vector algebra. Standards and units. Kinematics. Dynamics. The principle of relativity in the classical mechanics. Gravitation. Mechanics of many-particle systems. The motio of rigid bodies. Deformation, elasticity. Mechanics of fluids and gases. Laws of ideal gases. Kinetic theory. The thermodynamic laws. Statistical character of the second law. Entropy. Molecular phenomena in liquids and solids. Phase transitions. **Recommended literature:** Hajko V., Daniel-Szabó J.: Základy fyziky, VEDA, Bratislava 1983. Veis Š., Maďar J., Martišovits V.: Všeobecná fyzika I., Mechanika a molekulová fyzika, ALFA Bratislava, 1987. Fuka J., Široká M.: Obecná fyzika I / skriptum /, PF Univ. Palackého, Olomouc 1983. Hlavička A., a kol.: Fyzika pre pedagogické fakulty, SPN, Praha 1971. Hajko V., a kol.: Fyzika v príkladoch, ALFA Bratislava 1983. Ilkovič D.: Fyzika, SVTL Bratislava, 1962. Slaviček V., Wagner J.: Fyzika pro chemiky, SNTL Praha 1971. Krempaský J.: Fyzika, ALFA Bratislava 1982. Course language: Slovak

Notes:

Course assessment							
Total number of assessed students: 231							
A	В	С	D	Е	FX		
25.11	16.02	20.35	12.99	18.18	7.36		

Provides: doc. RNDr. Zuzana Ješková, PhD., doc. RNDr. Alžbeta Orendáčová, DrSc., Doc. RNDr. Jozef Hanč, PhD.

Date of last modification: 18.02.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚFV/ C

Course name: General Physics II

VF1b/03

Course type, scope and the method:

Course type: Lecture / Practice

Recommended course-load (hours): Per week: 4 / 2 Per study period: 56 / 28

Course method: present

Number of credits: 7

Recommended semester/trimester of the course: 2.

Course level: I.

Prerequisities: ÚFV/VF1a/07 or ÚFV/VF1a/12

Conditions for course completion:

Test.

Oral examination.

Learning outcomes:

To obtain a general view on basic electric magnetic phenomena and ability to solve basic problems of this subject.

Brief outline of the course:

Electric field in the free space. Work of the forces in the electrostatic field. Electrostatic field and steady current. Current in electrolytes, semiconductors, gasses and vacuum. Thermoelctric effects. Magnetic field in the free space. The interaction of moving charges with the electric current. Quasi steady electric field. Electromagnetic induction. Energy of magnetic field. AC current and circuits with ac current. Multiphase AC current. Rotating magnetic field. Electric effects in the substances. Magnetic properties of the substancies. Magnetic polarization. Diamagnetism and paramagnetism, Magnetic ordering. Ferromagnetism.

Recommended literature:

I. S. Grant, W.R. Phillips, Electromagnetism, John Wiley&Sons, Ltd, England, 1990

Course language:

english

Notes:

Course assessment

Total number of assessed students: 278

A	В	С	D	Е	FX
32.01	15.47	16.55	11.87	12.23	11.87

Provides: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Adriana Zeleňáková, PhD., RNDr. Erik Čižmár, PhD.

Date of last modification: 18.02.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚFV/ Co

Course name: General Physics III

VF1c/12

Course type, scope and the method:

Course type: Lecture / Practice

Recommended course-load (hours): Per week: 4 / 2 Per study period: 56 / 28

Course method: present

Number of credits: 7

Recommended semester/trimester of the course: 3.

Course level: I.

Prerequisities: ÚFV/VF1b/03

Conditions for course completion:

Exam+ 2 succesfull test from seminars

Learning outcomes:

The objective is to acquaint the students with the basis of oscilations, waves and optics.

Brief outline of the course:

Undamped oscilations, Mathematical, Physical and Torsional pendulum, Damped oscilations, Fourier transformation, Forced oscilations. Waves, their generation, waves equation.Interference. Huyghens principle. Reflection, difraction. Doppler effect. Waves speed in materials. Acoustics. Geometrical optics. Mirrors, lens. Fotometry.

Light as electromagnetic wave. Dispersion, absorption, interference, difraction, polarization. Photon's theory of light. Law of emision and absorption, Planck's law of radiation. Lasers.

Recommended literature:

- 1. A. Hlavička et al., Fyzika pro pedagogické fakulty, SPN, 1971
- 2. R.P. Feynman et al., Feynmanove prednášky z Fyziky I,II,III, ALFA, 1985
- 3. D. Halliday et al., Fyzika-Vysokoškolská učebnice obecné fyziky, VUTIUM, 2010
- 4. J. Fuka, B. Havelka, Optika a atómová fyzika, SPN,1961
- 5. A. Štrba, Všeobecná Fyzika 3 Optika, ALFA, 1979

Course language:

slovak

Notes:

Course assessment

Total number of assessed students: 95

A	В	С	D	Е	FX
32.63	25.26	25.26	12.63	4.21	0.0

Provides: doc. RNDr. Rastislav Varga, DrSc.

Date of last modification: 18.02.2014

	COURSE INFORMATION LETTER
University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚFV/ VF1d/12	Course name: General Physics IV
Course type, scope a Course type: Lectur Recommended cour Per week: 4/2 Per Course method: pre	re / Practice rse-load (hours): study period: 56 / 28
Number of credits: 7	
Recommended seme	ster/trimester of the course: 4.
Course level: I.	
Prerequisities: ÚFV/	VF1c/08 or ÚFV/VF1c/10 or ÚFV/VF1c/12
Conditions for cours written tests exam	e completion:
	ut the atomic structure and spectra and nuclei, and elementary particles. Basic s in nuclear physics and passage of nuclear radiation through media.
Structure and models characteristics of the radioactivity. Nuclear	ourse: articles. De Broglie waves. Experimental evidence for de Broglie waves. of atoms. Atomic spectra. Magnetic properties of atoms. X-ray spectra. Basic atomic nuclei. Nuclear forces and models. Radioactivity. Applications of reactions. Elementary particles, basic properties and classification. Types of ces. Cosmic rays. Passage of particles through matter. Detectors. Accelerators.
2. Vanovič J.: Atómo 3. Griffiths D., Introd 4. Úlehla I., Suk M., 5. Síleš E., Martinská 5. Hajko V. and team 6. Nosek D., Jádra a	o moderní fyziky, Praha, 1975. vá fyzika, Bratislava, 1980. duction to Elementary Particles, WILEY, 1987. Trka Z.: Atómy, jádra, částice, Praha, 1990. a G.: Všeobecná fyzika IV, skriptá PF UPJŠ, 2. vydanie, Košice, 1992. of authors, Physics in experiments, Bratislava, 1997. částice (Řešené příklady), Matfyzpress, MFF UK, Praha 2005, vest of a century, Discoveries of modern physics in 100 episodes, Oxford,
Course language: slovak and english	

Notes:

Course assessment						
Total number of assessed students: 52						
A	В	C	D	Е	FX	
36.54	21.15	19.23	11.54	11.54	0.0	

Provides: prof. RNDr. Stanislav Vokál, DrSc., RNDr. Janka Vrláková, PhD., RNDr. Adela Kravčáková, PhD.

Date of last modification: 11.02.2014

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: KGER/ Course name: German Language for Specific Purposes - German in **OJPV1/07** Natural Sciences 1 Course type, scope and the method: Course type: Practice **Recommended course-load (hours):** Per week: 2 Per study period: 28 Course method: present Number of credits: 2 Recommended semester/trimester of the course: 4. Course level: I. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 120 C Α В D Е FX 19.17 20.83 27.5 23.33 8.33 0.83

Provides: Mgr. Eva Černáková, PhD., Dr. rer. pol. Michaela Kováčová

Date of last modification: 05.02.2014

University: P. J. Šafárik University in Košice Faculty: Faculty of Science **Course ID:** KGER/ **Course name:** Grammar in the German Language Communication NJKG/07 Course type, scope and the method: Course type: Practice **Recommended course-load (hours):** Per week: 2 Per study period: 28 Course method: present Number of credits: 2 Recommended semester/trimester of the course: Course level: I., II. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 46 \mathbf{C} Α В D Е FX 54.35 13.04 8.7 8.7 4.35 10.87 Provides: Dr. rer. pol. Michaela Kováčová Date of last modification: 05.02.2014

Page: 52

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: KFaDF/

Course name: History of Philosophy and Philosophy of Education -

DF1/05

Cultural and Socio-Anthropological Relations

Course type, scope and the method:

Course type: Lecture / Practice

Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14

Course method: present

Number of credits: 4

Recommended semester/trimester of the course: 5.

Course level: I.

Prerequisities:

Conditions for course completion:

Learning outcomes:

Brief outline of the course:

Recommended literature:

Course language:

Notes:

Course assessment

Total number of assessed students: 392

Α	В	С	D	Е	FX
63.52	16.58	10.97	5.87	2.55	0.51

Provides: doc. PhDr. Pavol Tholt, PhD., mim.prof., Doc. PhDr. Peter Nezník, CSc.

Date of last modification: 26.01.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: KFaDF/

Course name: Chapters from History of Philosophy of 19th and 20th

KDF/05

Centuries (General Introduction)

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of credits: 2

Recommended semester/trimester of the course: 6.

Course level: I., II.

Prerequisities:

Conditions for course completion:

Learning outcomes:

Brief outline of the course:

Recommended literature:

Course language:

Notes:

Course assessment

Total number of assessed students: 10

Α	В	С	D	Е	FX
50.0	20.0	10.0	0.0	10.0	10.0

Provides: doc. PhDr. Pavol Tholt, PhD., mim.prof.

Date of last modification: 26.01.2014

Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.

University: P. J. Šafárik University in Košice Faculty: Faculty of Science **Course ID:** KFaDF/ **Course name:** Chapters from Philosophy of Education FVp/04 Course type, scope and the method: Course type: Practice **Recommended course-load (hours):** Per week: 2 Per study period: 28 Course method: present Number of credits: 2 **Recommended semester/trimester of the course:** 5. Course level: I., II. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 3 \mathbf{C} A В D Е FX 100.0 0.0 0.0 0.0 0.0 0.0 Provides: doc. PhDr. Pavol Tholt, PhD., mim.prof. Date of last modification: 26.01.2014 Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ C

Course name: Chemical calculations

CHV1/99

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of credits: 2

Recommended semester/trimester of the course: 1.

Course level: I.

Prerequisities:

Conditions for course completion:

Short written tests.

Written test.

Learning outcomes:

To teach students how to calculate material balances in the systems with or without chemical processes and how to calculate examples concerning the chemical equilibrium.

Brief outline of the course:

Expression of the clear matter amount and the system composition. Stoichiometric formula. Material bilances for preparation, dissolving and mixing of solutions, and for separating of mixtures. Material bilances for combined processes. Chemical equations and material bilances in the systems with chemical processes. Acid-Base equilibrium and the pH calculations. The solubility product and solubility.

Recommended literature:

Potočňák I.: Chemické výpočty vo všeobecnej a anorganickej chémii (skriptum), PF UPJŠ, Košice, 2006.

Course language:

Notes:

Course assessment

Total number of assessed students: 865

A	В	С	D	Е	FX
16.07	19.31	25.09	23.12	15.72	0.69

Provides: RNDr. Martin Vavra, PhD., doc. RNDr. Zuzana Vargová, Ph.D., Mgr. Miroslav Almáši, PhD., RNDr. Lukáš Smolko

Date of last modification: 03.02.2014

Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ Co

Course name: Cheminformatics I

ISC1a/00

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of credits: 2

Recommended semester/trimester of the course: 1.

Course level: I.

Prerequisities:

Conditions for course completion:

seminar project

Learning outcomes:

Introductory course aimed at introducing students to the fundamental informatics techniques for chemistry-related disciplines. The class will cover a wide range of topics, including searching chemical information on internet, searching for patent information and work with the primary and secondary literature.

Brief outline of the course:

Searching, retrieving and use of the informations in chemistry. Using of "paper" resources (primary journals, Chemical Abstracts, Beilstein). Searching chemical information on Internet (Scirus, ScienceDirect, Scopus, Web of Science, Medline, NIST) and e-journals.

Recommended literature:

1. R.E. Maizell: How to find Chemical Information, John Wiley,

New York 1998

2. Internet resources for chemistry.

Course language:

slovak language and english language

Notes:

Course assessment

Total number of assessed students: 661

A	В	С	D	Е	FX
62.48	10.29	15.73	8.62	1.97	0.91

Provides: RNDr. Monika Tvrdoňová, PhD., RNDr. Ladislav Janovec, PhD.

Date of last modification: 03.02.2014

Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ C

Course name: Chemistry

CHS/03

Course type, scope and the method:

Course type:

Recommended course-load (hours):

Per week: Per study period: Course method: present

Number of credits: 0

Recommended semester/trimester of the course:

Course level: I.

Prerequisities: ÚCHV/ACHU/03 and ÚCHV/OCHU/03 and ÚCHV/ANCHU/03 and (ÚCHV/FCHU/03 or ÚCHV/FCHU/10) and ÚCHV/BCHU/03 and ÚCHV/MUSU/03

Conditions for course completion:

Learning outcomes:

Brief outline of the course:

Recommended literature:

Course language:

Notes:

Course assessment

Total number of assessed students: 175

A	В	С	D	Е	FX
22.29	25.14	24.0	17.71	8.57	2.29

Provides:

Date of last modification: 03.02.2014

Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/

Course name: Chemistry of Inorganic & Organic Compounds

CAOZ/03

Course type, scope and the method:

Course type: Lecture / Practice Recommended course-load (hours):

Per week: 2 / 1 Per study period: 28 / 14

Course method: present

Number of credits: 5

Recommended semester/trimester of the course: 4.

Course level: I.

Prerequisities:

Conditions for course completion:

Inorganic part: one test in 6th week; 50 points.

Organic part: one test in 12th week; 50 points.

At least 50% of points is required from both parts.

Terminal examination by written form, 100 points; 50 points from inorganic part and 50 points from organic part.

Learning outcomes:

The main goal of this subject is to provide a general overview of inorganic and organic compound, including special materials, in regard to their properties and applications in the industry.

Brief outline of the course:

The compounds of some metallic and nonmetallic elements. Relationship between structure, properties and application. Natural and synthetic materials, nanomaterials on the basis of inorganic compounds. Overview of the organic compounds including natural and unnatural saccharides, peptides, lipid substances, nucleosides, terpenes and their industrial applications.

Recommended literature:

- 1. Ondrejovič G. a kol.: Anorganická chémia 2, STU, Bratislava 1995.
- 2. Shriver D. F., Atkins P. at al.: Inorganic Chemistry. Oxford University Press, Oxford 2006.
- 3. K. C. Timberlake: Organic and Biological Chemistry, Benjamin Cummings Publishing Company, Inc., 2002.
- 4. McMurry, J. Organic chemistry, Brooks/Cole, a Thomson Learning Company, 2004.

Course language:

Notes:

Course assessment

Total number of assessed students: 66

A	В	С	D	Е	FX
33.33	12.12	21.21	24.24	7.58	1.52

Provides: doc. RNDr. Mária Reháková, CSc., doc. RNDr. Miroslava Martinková, PhD.

 $\textbf{Date of last modification:}\ 03.02.2014$

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: R UPJŠ/ Course name: IB10 - Medzinárodný certifikát ECo-C IB10/14 Course type, scope and the method: **Course type: Recommended course-load (hours):** Per week: Per study period: Course method: present **Number of credits: 16** Recommended semester/trimester of the course: Course level: I., I.II., II. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 0 abs neabs n 0.0 0.0 0.0 **Provides:** Date of last modification: 11.08.2014 Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: R UPJŠ/ Course name: IB11 - Medzinárodný certifikát ECDL IB11/14 Course type, scope and the method: **Course type: Recommended course-load (hours):** Per week: Per study period: Course method: present Number of credits: 14 Recommended semester/trimester of the course: Course level: I., I.II., II. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 0 abs neabs n 0.0 0.0 0.0 **Provides:** Date of last modification: 11.08.2014 Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: R UPJŠ/ Course name: IB12 - Používanie, administrácia a vývoj v systéme SAP IB12/14 Course type, scope and the method: **Course type: Recommended course-load (hours):** Per week: Per study period: Course method: present **Number of credits: 54** Recommended semester/trimester of the course: Course level: I., I.II., II. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 0 abs neabs n 0.0 0.0 0.0 **Provides:** Date of last modification: 11.08.2014 Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: R UPJŠ/ Course name: IB1 - Etika v biomedicínskych vedách pre zdravotnícku prax IB1/14 Course type, scope and the method: **Course type: Recommended course-load (hours):** Per week: Per study period: Course method: present **Number of credits: 16** Recommended semester/trimester of the course: Course level: I., I.II., II. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 0 abs neabs n 0.0 0.0 0.0 **Provides:** Date of last modification: 11.08.2014

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: R UPJŠ/ | Course name: IB2 - Právne minimum – súkromnoprávne aspekty IB2/14 Course type, scope and the method: **Course type: Recommended course-load (hours):** Per week: Per study period: Course method: present **Number of credits: 16** Recommended semester/trimester of the course: Course level: I., I.II., II. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 0 abs neabs n 0.0 0.0 0.0 **Provides:** Date of last modification: 11.08.2014 Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: R UPJŠ/ | Course name: IB3 - Právne minimum – verejnoprávne aspekty IB3/14 Course type, scope and the method: **Course type: Recommended course-load (hours):** Per week: Per study period: Course method: present **Number of credits: 16** Recommended semester/trimester of the course: Course level: I., I.II., II. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 0 abs neabs n 0.0 0.0 0.0 **Provides:** Date of last modification: 11.08.2014 Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: R UPJŠ/ | Course name: IB4 - Projektový manažment IB4/14 Course type, scope and the method: **Course type: Recommended course-load (hours):** Per week: Per study period: Course method: present Number of credits: 20 Recommended semester/trimester of the course: Course level: I., I.II., II. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 0 abs neabs n 0.0 0.0 0.0 **Provides:** Date of last modification: 11.08.2014 Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: R UPJŠ/ Course name: IB5 - Manažérska ekonomika IB5/14 Course type, scope and the method: **Course type: Recommended course-load (hours):** Per week: Per study period: Course method: present **Number of credits: 16** Recommended semester/trimester of the course: Course level: I., I.II., II. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 0 abs neabs n 0.0 0.0 0.0 **Provides:** Date of last modification: 11.08.2014 Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: R UPJŠ/ Course name: IB6 - Riešenie konfliktných a krízových situácií v školskej IB6/14 praxi Course type, scope and the method: **Course type: Recommended course-load (hours):** Per week: Per study period: Course method: present **Number of credits: 16** Recommended semester/trimester of the course: Course level: I., I.II., II. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 0 abs neabs n 0.0 0.0 0.0 **Provides:** Date of last modification: 11.08.2014

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: R UPJŠ/ Course name: IB7 - Štatistika pre prax IB7/14 Course type, scope and the method: **Course type:** Recommended course-load (hours): Per week: Per study period: Course method: present **Number of credits: 16** Recommended semester/trimester of the course: Course level: I., I.II., II. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 0 abs neabs n 0.0 0.0 0.0 **Provides:** Date of last modification: 11.08.2014 Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: R UPJŠ/ | Course name: IB8 - Environmentálne aspekty záťaže životného prostredia IB8/14 Course type, scope and the method: **Course type: Recommended course-load (hours):** Per week: Per study period: Course method: present **Number of credits: 16** Recommended semester/trimester of the course: Course level: I., I.II., II. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 0 abs neabs n 0.0 0.0 0.0 **Provides:** Date of last modification: 11.08.2014 Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: R UPJŠ/ Course name: IB9 - Medzinárodný certifikát TOEFL IB9/14 Course type, scope and the method: **Course type: Recommended course-load (hours):** Per week: Per study period: Course method: present Number of credits: 17 Recommended semester/trimester of the course: Course level: I., I.II., II. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 0 abs neabs n 0.0 0.0 0.0 **Provides:** Date of last modification: 11.08.2014 Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: KFaDF/ **Course name:** Idea Humanitas 1 (General Introduction) IH1/03 Course type, scope and the method: Course type: Practice **Recommended course-load (hours):** Per week: 2 Per study period: 28 Course method: present Number of credits: 2 Recommended semester/trimester of the course: 6. Course level: I., II. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 9 \mathbf{C} Α В D Е FX 55.56 11.11 0.0 11.11 22.22 0.0 Provides: Doc. PhDr. Peter Nezník, CSc. Date of last modification: 26.01.2014 Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚINF/

Course name: Information and Communication Technologies

IKTP/10

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of credits: 2

Recommended semester/trimester of the course: 4.

Course level: I.

Prerequisities:

Conditions for course completion:

Learning outcomes:

Brief outline of the course:

Text processing using a word processor.

Processing and evaluation of information using a spreadsheet.

Search, retrieval and exchange of information via the Internet.

Creating presentations.

Recommended literature:

- 1. Franců, M: Jak zvládnout testy ECDL. Praha : Computer Press. 2007. 160 s. ISBN 978-80-251-1485-8
- 2. Jančařík, A. et al.: S počítačem do Evropy ECDL. 2. vydanie. Praha : Computer Press, 2007. 152 s. ISBN 80-251-1844-3
- 3. Kolektív autorov: Sylabus ECDL verzia 5.0. [on-line] [citované 9.2.2010]. Dostupné na internete: http://www.ecdl.sk/buxus/docs//interne_informacie/Sylabus_V5.0/20090630ECDL-Sylabus_V50_SK-V01_FIN.pdf
- 4. Kalakay, R. et al: Informačné a komunikačné technológie prezenčný kurz. [on-line] [citované 9.2.2010]. Dostupné na internete: http://moodle.science.upjs.sk/course/view.php?id=90

Course language:

Notes:

Course assessment

Total number of assessed students: 970

A	В	С	D	Е	FX
66.8	17.63	6.91	3.51	1.75	3.4

Provides: Mgr. Alexander Szabari, PhD., RNDr. Jozef Studenovský, CSc., RNDr. Zuzana Bednárová, PhD., doc. Ing. Štefánia Gallová, CSc.

Date of last modification: 03.02.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚINF/

Course name: Information and Communication Technologies - distance

IKTD/10

learning

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of credits: 2

Recommended semester/trimester of the course: 4.

Course level: I.

Prerequisities:

Conditions for course completion:

Learning outcomes:

Brief outline of the course:

Text processing using a word processor.

Processing and evaluation of information using a spreadsheet.

Search, retrieval and exchange of information via the Internet.

Creating presentations.

Recommended literature:

- 1. Franců, M: Jak zvládnout testy ECDL. Praha : Computer Press. 2007. 160 s. ISBN 978-80-251-1485-8
- 2. Jančařík, A. et al.: S počítačem do Evropy ECDL. 2. vydanie. Praha : Computer Press, 2007. 152 s. ISBN 80-251-1844-3
- 3. Kolektív autorov: Sylabus ECDL verzia 5.0. [on-line] [citované 9.2.2010]. Dostupné na internete: http://www.ecdl.sk/buxus/docs//interne_informacie/Sylabus_V5.0/20090630ECDL-Sylabus_V50_SK-V01_FIN.pdf
- 4. Kalakay, R. et al: Informačné a komunikačné technológie prezenčný kurz. [on-line] [citované 9.2.2010]. Dostupné na internete: http://moodle.science.upjs.sk/course/view.php?id=90

Course language:

Notes:

Course assessment

Total number of assessed students: 113

A	В	С	D	Е	FX
75.22	7.08	3.54	0.0	3.54	10.62

Provides: doc. Ing. Štefánia Gallová, CSc., RNDr. Jozef Studenovský, CSc., RNDr. Zuzana Bednárová, PhD.

Date of last modification: 03.02.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/

Course name: Inorganic Chemistry

ACHU/03

Course type, scope and the method:

Course type: Lecture / Practice

Recommended course-load (hours): Per week: 3 / 1 Per study period: 42 / 14

Course method: present

Number of credits: 6

Recommended semester/trimester of the course: 2.

Course level: I.

Prerequisities: ÚCHV/VCHU/10 or ÚCHV/VCHU/14

Conditions for course completion:

Test in the middle and at the end of the semester.

Oral examination.

Learning outcomes:

Aim of the course is to provide the students with a knowledge of systematic chemistry of non-metallic elements

Brief outline of the course:

Electronic configuration, abundance, use, physical and chemical properties, preparation, reactivity of non-metallic elements hydrogen, halogens, oxygen, sulphur, nitrogen, phosphorus, carbon, silicon, boron and rare gases. Binary and other compounds formed by these elements, their properties and reactivity. Metals and transition elements. Abudance, properties, reactivity, important compounds.

Recommended literature:

http://kosice.upjs.sk/~vladimir.zelenak/ACHU.htm (ppt slides from the lectures as a support for self study)

Greenwood, N. N., Earnshaw, A: Chemistry of the Elements. Pergamon Press, Oxford, 1984 Atkins O., Overton T., Rourke J., Weller M., Armstrong F.: Inorganic Chemistry, University Press, Oxford, 2006.

Course language:

Notes:

Course assessment

Total number of assessed students: 530

A	В	С	D	Е	FX
9.81	17.74	30.19	27.17	10.94	4.15

Provides: doc. RNDr. Vladimír Zeleňák, PhD.

Date of last modification: 03.02.2014

	COURSE INFORMATION LETTER
University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚCHV/ UECH/03	Course name: Introduction to Environmental Chemistry
Course type, scope a Course type: Lectur Recommended cou Per week: 2 / 1 Per Course method: pre	re / Practice rse-load (hours): study period: 28 / 14
Number of credits: 5)
Recommended seme	ester/trimester of the course: 3.
Course level: I., II.	
Prerequisities:	
Conditions for cours Oral examination	se completion:
Learning outcomes: Introduction to topics protection.	s in environmental chemistry and basic procedures applied for environmental
atmosphere. Energy photoprocesses in the environmental pollut Environmental chem metals). Environmentals their protection. The utilization. Energy at	
Oxford University Pr 2. R.A. Bailey, H.M. Academic Press, San 3. G. Schwedt: The E 4. R.N. Reeve, J.D. E 5. G. Burton, J. Holn London 1994 6. www	, Stephen J. Duffy: Environmental Chemistry - A Global Perspective, ress, Oxford 2003 Clark, J.P. Ferris, S. Krause, R.L. Strong: Chemistry of the Environment,
Course language:	

Notes:

Course assessment					
Total number o	Total number of assessed students: 189				
A	В	С	D	Е	FX
48.15	19.05	16.93	9.52	5.82	0.53

Provides: RNDr. Andrea Straková Fedorková, PhD.

Date of last modification: 03.02.2014

	COURSE INFORMATION LETTER
University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚFV/ UVF/05	Course name: Introduction to General Physics
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	ce rse-load (hours): dy period: 28
Number of credits: 2	
Recommended seme	ster/trimester of the course: 1.
Course level: I.	
Prerequisities:	
Conditions for cours Active presentation of Solved assignments Positive results at two	luring the lessons twice a year
gained with the help inevitable precondition	nding of the key concepts of the topics of Mechanics and Molecular Physics of problem solving, physical experiments and multimedial support that is on for the further study at University level. At the end of this course the student with the courses proceeding from the course General Physics I.
Physics. The content school experiments, The aim is to help st	course: course subject to the course General physics 1 - Mechanics and Molecular involves key concepts in mechanics and molecular physics with the help of interactive multimedial teaching materials and physical tasks and problems. The conceptual understaning of the University course content.
2. Pizzo, J.: Interactiv 3. Cunningham, J, Ho 4. Halliday D., Resni VUTIUM, Brno, 200 5. Walker, J.: The Fly	nonstration Experiments in Physics, AAPT, 2003 we Physics demonstration, AAPT, 2001 err, N.: Hands on Physics Activities, Jossey-Bass A Wiley Imprint, 1994 ck R., Walker J.: Fyzika. Část 1- 5., Vysokoškolská učebnica fyziky,
Course language:	

Slovak

Notes:

Course assessment					
Total number of assessed students: 188					
A	В	C	D	Е	FX
38.3	16.49	23.4	13.83	7.45	0.53

Provides: doc. RNDr. Zuzana Ješková, PhD., doc. RNDr. Marián Kireš, PhD., Doc. RNDr. Jozef Hanč, PhD., RNDr. Brigita Balogová

Date of last modification: 18.02.2014

	COURSE INFORMATION LETTER
University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚFV/ UVF2/07	Course name: Introduction to General Physics II
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	ce rse-load (hours): dy period: 28
Number of credits: 2	·
Recommended seme	ster/trimester of the course: 2.
Course level: I.	
Prerequisities:	
Active presentations Solved assignments Postive results at two	duringf the lessons twice a year
the help of problem precondition for the	nding of the key concepts of the topics of Electricity and Magnetism with solving, physical experiments and multimedial support that is inevitable further study at University level. At the end of the course the studnet will be the courses, proceeding from the course General physics II.
The content involves interactive multimed students to overcome	ourse: cortive subject to the course General Physics 2 - Electricity and Magnetism. key concepts of electricity and magntism with the help of school experiments, ial teaching materials and physical tasks and problems. The aim is to help e difficulties connected with knowledge gained during the previous study al understanding of the University course content.
2. Pizzo, J.: Interactiv 3. Cunningham, J, He 4. Halliday D., Resni VUTIUM, Brno, 200	constration Experiments in Physics, AAPT, 2003 We Physics demonstration, AAPT, 2001 err, N.: Hands on Physics Activities, Jossey-Bass A Wiley Imprint, 1994 ck R., Walker J.: Fyzika. Část 1- 5., Vysokoškolská učebnica fyziky,
Course language: Slovak	

Notes:

Course assessment					
Total number of assessed students: 158					
Α	В	C	D	Е	FX
41.77	16.46	22.15	8.86	10.76	0.0

Provides: doc. RNDr. Zuzana Ješková, PhD.

Date of last modification: 18.02.2014

University: P. J. Šafá	rik University in Košice					
Faculty: Faculty of S	cience					
Course ID: ÚFV/ ZMF/07	· · · · · · · · · · · · · · · · · · ·					
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	ce rse-load (hours): idy period: 28					
Number of credits: 2	<u> </u>					
Recommended seme	ster/trimester of the course: 1.					
Course level: I.						
Prerequisities:						
	se completion: ng semester, solving of three sets of problems, active participation. of active participation during semester, completed sets of problems					
	standing and mastery of basic mathematical concepts and skills ntial and integral calculus and ordinary differential equations tory physics course.					
Brief outline of the c	ourse:					
Molecular Physics ar The content deals wit and integral calculus After the course stud fields, the function of	es basic mathematical background to general physics courses: Mechanics & ad Electricity & magnetism. In understanding the basic concepts of vector algebra and analysis, differential and differential equations. In dent should be familiar with the concepts: vector, scalar, vector and scalar of one variable, derivative, integral, differential equation; to be able to interpret a phenomena and acquire basic mathematical skills related to these concepts					
2. Stewart, J., Calcul	et al, Applied Calculus, 4th ed., John Wiley & Sons, 2010 us: early transcendentals, 6th ed., Brooks Cole, 2008 glom, I.M.: Higher Math for Beginners (Mostly Physicists and Engineers),					
Course language: Slovak						

Notes:

Course assessment					
Total number of assessed students: 166					
Α	В	C	D	Е	FX
38.55	18.67	21.08	11.45	10.24	0.0

Provides: Doc. RNDr. Jozef Hanč, PhD.

Date of last modification: 18.02.2014

University: P. J. Šafá	University: P. J. Šafárik University in Košice				
Faculty: Faculty of S	cience				
Course ID: Dek. PF UPJŠ/USPV/13	Course ID: Dek. PF Course name: Introduction to Study of Sciences JPJŠ/USPV/13				
Course type: Lectur Recommended cour Per week: Per stud	Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: Per study period: 12s / 3d Course method: present				
Number of credits: 2					
Recommended seme	ster/trimester of the cours	e: 1.			
Course level: I.					
Prerequisities:					
Conditions for cours	e completion:				
Learning outcomes:					
Brief outline of the c	ourse:				
Recommended litera	ture:				
Course language:					
Notes:					
Course assessment Total number of asses	ssed students: 539				
	abs	n			
	95.18 4.82				
Provides: doc. RNDr. Mária Kožurková, CSc., prof. RNDr. Katarína Cechlárová, DrSc., prof. RNDr. Beňadik Šmajda, CSc., prof. Mgr. Jaroslav Hofierka, PhD., doc. RNDr. Ivan Žežula, CSc., doc. RNDr. Vladimír Zeleňák, PhD., Doc. RNDr. Jozef Hanč, PhD., RNDr. Ondrej Krídlo, PhD., Mgr. Vladislav Kolarčik, PhD., RNDr. Janetta Nestorová-Dická, PhD.					
Date of last modification: 17.02.2014					
Approved: prof. RNI	Dr. Peter Kollár, DrSc., doc.	RNDr. Vladimír Zeleňák, PhD.			

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚMV/ Cours

Course name: Mathematics I for chemists

MTCa/13

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 2 Per study period: 28 / 28

Course method: present

Number of credits: 6

Recommended semester/trimester of the course: 1.

Course level: I.

Prerequisities:

Conditions for course completion:

According to the results from the semester and in view of the results of the written final test.

Learning outcomes:

To obtain basic knowledge on functions of one variable and their properties; to be able to apply the theory in concrete excercises.

Brief outline of the course:

Functions, basic properties. Elementary functions. Continuous functions. Limits. Derivation and its geometric aplications. Theorems about continuous functions. Behaviour of functions. Indefinite integrals, basic methods of integration. Definite integral and its applications.

Recommended literature:

S. Lang: A First Course in Calculus, Springer Verlag, 1998

Course language:

Slovak

Notes:

Course assessment

Total number of assessed students: 393

A	В	С	D	Е	FX
9.67	10.43	18.07	19.85	26.21	15.78

Provides: doc. RNDr. Roman Soták, PhD., RNDr. Jana Borzová, Mgr. Timea Gábová, RNDr. Eva Oceľáková

Date of last modification: 14.02.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚMV/ Course

Course name: Mathematics I for physicists

MTFa/13

Course type, scope and the method: Course type: Lecture / Practice

Recommended course-load (hours): Per week: 2 / 2 Per study period: 28 / 28

Course method: present

Number of credits: 6

Recommended semester/trimester of the course: 1.

Course level: I.

Prerequisities:

Conditions for course completion:

Two written tests and one homework with excercises from the whole semester. The final evaluation is given according to the results from the semester and in view of the results of the written final test.

Learning outcomes:

To obtain basic knowledge on functions of one variable and their properties; to be able to apply the theory in concrete excercises.

Brief outline of the course:

Functions, basic properties. Elementary functions. Continuous functions. Limits. Derivation and its geometric aplications. Theorems about continuous functions. Behaviour of functions. Indefinite integrals, basic methods of integration. Definite integral and its applications.

Recommended literature:

S. Lang: A First Course in Calculus, Springer Verlag, 1998

Course language:

Slovak

Notes:

Course assessment

Total number of assessed students: 304

A	В	С	D	Е	FX
7.57	8.22	13.82	19.41	30.26	20.72

Provides: doc. RNDr. Roman Soták, PhD.

Date of last modification: 14.02.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚMV/ **Course name:** Mathematics II for physicists

MTFb/13

Course type, scope and the method:

Course type: Lecture / Practice

Recommended course-load (hours): Per week: 2 / 2 Per study period: 28 / 28

Course method: present

Number of credits: 5

Recommended semester/trimester of the course: 2.

Course level: I.

Prerequisities: ÚMV/MTFa/13

Conditions for course completion:

Two written tests and one homework with excercises from the whole semester, final test. According to the results from the semester and in view of the results of the written final test.

Learning outcomes:

To develop acquired knowledge of mathematical analysis with knowledge on linear algebra and functions of more variables. To learn to solve basic types of differential equations and know how to use them to model real-world phenomena. To learn to solve problems about infinite series.

Brief outline of the course:

System of linear algebraic equations, determinants. Functions of more variables, continuity and limits, partial derivations, local extremes of functions of two variables. Some types of differential equations. Series, functional series, Taylor and MacLaurin series.

Recommended literature:

- 1. S. Lang: A First Course in Calculus, Springer Verlag, 1998
- 2. Huťka V., Benko E., Ďurikovič V.: Matematika, Alfa, Bratislava 1991.
- 3. Došlá, Z.: Matematika pro chemiky, 1.díl. Masarykova univerzita, Brno, 2010.

Course language:

Slovak

Notes:

Course assessment

Total number of assessed students: 171

Α	В	С	D	Е	FX
9.94	15.79	12.28	26.32	30.41	5.26

Provides: doc. RNDr. Stanislav Lukáč, PhD.

Date of last modification: 14.02.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚFV/ Course name: Methods of Data Processing in Physics

SDF1/99

Course type, scope and the method:

Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14

Course method: present

Number of credits: 4

Recommended semester/trimester of the course: 3.

Course level: I.

Prerequisities:

Conditions for course completion:

Five tasks in Matlab/Octave.

Exam interview - 60%, tasks - 40%.

Learning outcomes:

Methods of data processing in physics.

Brief outline of the course:

- 1. Numerical methods.
- 2. Regression analysis.
- 3. Computational physics.

Recommended literature:

Buchanan J. L., Turner P. R.: Numerical Methods and Analysis. McGraw-Hill, Inc., New York, 1992. Siegel A. F.: Statistics and Data Analysis. An Introduction. J. Wiley&Sons, NY, 1988.

Course language:

slovak, basics of english

Notes:

Course assessment

Total number of assessed students: 113

A	В	С	D	Е	FX
23.01	27.43	20.35	14.16	3.54	11.5

Provides: RNDr. Erik Čižmár, PhD.

Date of last modification: 18.02.2014

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚFV/ Course name: Modern Didactical Technics MDT06/06 Course type, scope and the method: Course type: Practice **Recommended course-load (hours):** Per week: 2 Per study period: 28 Course method: present Number of credits: 3 **Recommended semester/trimester of the course:** 5. Course level: I., II. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 76 \mathbf{C} A В D Е FX 97.37 1.32 0.0 0.0 1.32 0.0 Provides: doc. RNDr. Marián Kireš, PhD. Date of last modification: 18.02.2014 Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚFV/ | Course name: Modern Trends in Physics

MTF/07

Course type, scope and the method:

Course type: Lecture

Recommended course-load (hours): Per week: 3 Per study period: 42

Course method: present

Number of credits: 4

Recommended semester/trimester of the course: 4.

Course level: I.

Prerequisities:

Conditions for course completion:

test

test

Learning outcomes:

Presentation of scientific goals and experimental facilities on the Institute of Physics. Discussion of new trends in physics of micro-world, astrophysics, biophysics and physics of condensed matter.

Brief outline of the course:

The present state of the micro-world physics – fundamental particles and the interaction forces. Theoretical description of the micro-world – the Standard Model. Experimental tests of the Standard Model - the discovery of neutral currents and intermediate W+-, Z0 bosons. Heavy ion collisions and the search for new state of matter - quark gluon plasma - on the most powerful accelerators RHIC (Relativistic Heavy Ion Collider), Brookhaven National Laboratory) , USA and on the constructed LHC (Large Hadron Collider), CERN, Geneva. Big Bang and the quark gluon plasma. Some open questions – search for Higgs boson, responsible for the mass of fundamental particles and quark gluon plasma in laboratory conditions.

Practical activities – demonstration of the knowledge from lectures at identification of the real Z0 decay events in experimental data from the LEP accelerator, CERN, Swizterland.

New trends in astrophysical investigation: Solar system planets and exoplanets; cataclysmic variables, blazers and polars; black holes; quasars and active galactic nuclei, clusters of galaxies and web structure of Universe; gravitational lensing, dark matter and dark energy; gamma ray bursts. Topical problems in biophysics

Low temperatures as a tool for the study of physical properties of matter. Non-Fermi liquid materials... Geometrically frustrated systems. Quantum tunneling in molecular magnets. Application of quantum magnets. Excursion in the Centre of Excellence of Low Temperature Physics.

Soft magnetic nanostructure materials prepared by milling and alloying: magnetic properties of small particles, magnetization processes, domain structure, milling and alloying.

Recommended literature:

S. Chikazumi: Physics of Magnetism, J. Willey and Sons, Inc. New York, London, Sydney, 1997.

C. Suryanarayana, Progress in Materials Science 46 (2001), 1-184

F. Close: The Cosmic Onion, 1990

Cindy Schwarz: A Tour of the Subatomic Zoo, 1997

Frank Close, Michael Marten, Christine Sutton: The Particle Odyssey-

A Journey to the Heart of Matter, 2002

http://vk.upjs.sk/~epog/2006/

Scientific journals

Course language:

english

Notes:

Course assessment

Total number of assessed students: 53

abs	n
100.0	0.0

Provides: Dr.h.c. prof. RNDr. Alexander Feher, DrSc.

Date of last modification: 18.02.2014

University: P. J. Šafá	rik University in Košice		
Faculty: Faculty of S	cience		
Course ID: ÚTVŠ/ NJ//13	Course name: Naval Yach	ting	
Course type, scope a Course type: Practic Recommended cou Per week: 36 Per st Course method: pre	ce rse-load (hours): sudy period: 504 esent		
Number of credits: 2			
	ster/trimester of the cours	e: 	
Course level: I., II.			
Prerequisities:			
Conditions for cours	se completion:		
Learning outcomes:			
Brief outline of the c	ourse:		
Recommended litera	nture:		
Course language:			
Notes:			
Course assessment Total number of asse	ssed students: 2		
	abs	n	
	100.0	0.0	
Provides: doc. Mgr. l	Rastislav Feč, PhD.		
Date of last modifica	ation: 15.01.2014		
Approved: prof. RNI	Dr. Peter Kollár, DrSc., doc.	RNDr. Vladimír Zeleňák, PhD.	

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/

Course name: Organic chemistry

OCHU/03

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 3 / 1 Per study period: 42 / 14

Course method: present

Number of credits: 6

Recommended semester/trimester of the course: 2.

Course level: L

Prerequisities: ÚCHV/VCHU/15 or ÚCHV/VCHU/14 or ÚCHV/VCHU/10 or ÚCHV/VACH/10

Conditions for course completion:

Two tests at lecture in 7 and 14th week. Test max 50 points. At least 25 points required.

Written exam, 100 points. At least 49% of points required.

Final evaluation: A 90-100 pts, B 80-89 pts, C 70-79 pts, D 60-69 pts, E 50-59 pts, FX 0-49 pts

Learning outcomes:

Basic organic chemistry course.

Brief outline of the course:

Chemical bonding Hybridization and Bonding Covalent bonds Double bonds and Triple Bonds Structural Formulas of Organic Molecules Polar Covalent Bonds and Electronegativity Constitutional Isomers Alkenes Electrophilic Additions Strong Brønsted Acids Lewis Acids (non-Proton Electrophiles) Electrophilic Halogen Reagents Other Electrophilic Reagents Reduction Oxidation Radical Additions Allylic Substitution Alkynes Addition Reactions Hydrogenation Electrophiles Hydration & Tautomerism Hydroboration Nucleophilile Addition & Reduction Acidity of Terminal Alkynes (Substitution of H) Alkyl Halides General Reactivity Substitution(of X) SN2 Mechanism SN1 Mechanism Elimination (of HX) Summary of Substitution vs. Elimination Substitution by Metals Elimination Reactions of Dihalides Alcohols Reactions of Alcohols Substitution of the Hydroxyl H Substitution of the Hydroxyl Group Elimination of Water Oxidation of Alcohols Reactions of Phenols Acidity of Phenols Ring Substitution of Phenols Oxidation to Quinones Aromatic compounds Electrophilic Substitution A Substitution Mechanism Reactions of Substituted Benzenes Reaction Characteristics Reactions of Disubstituted Rings Reactions of Substituent Groups Nucleophilic Substitution, Elimination & Addition Reactions Amines Basicity of Nitrogen Compounds Acidity of Nitrogen Compounds Important Reagent Bases Reactions of Amines Electrophilic Substitution at Nitrogen Preparation of 1°-Amines Preparation of 2° & 3°-Amines Reactions with Nitrous Acid Reactions of Aryl Diazonium Intermediates Elimination Reactions of Amines Oxidation States of Nitrogen Basic information: Aldehydes & Ketones Carboxylic Acids Carboxylic Derivatives Natural products

Recommended literature:

- 1. on-line ppt presentation in MOODLE, moodle science.upjs.sk
- 2. Organic Chemistry, Clayden, Greeves Warren & Wothers, Oxford University Press, 2010
- 3. Organic Chemistry, Solomon, Willey, 2009

Course language:							
Notes:							
Course assessment Total number of assessed students: 556							
A B C D E FX							
3.6	8.27	14.75	20.86	47.84	4.68		

Provides: prof. RNDr. Jozef Gonda, DrSc., RNDr. Slávka Hamul'aková, PhD.

Date of last modification: 03.02.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/

Course name: Organic chemistry - Lab.

POCHU/03

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 4 Per study period: 56

Course method: present

Number of credits: 5

Recommended semester/trimester of the course: 3.

Course level: L

Prerequisities: ÚCHV/OCHU/03

Conditions for course completion:

Two tests 2x25 p., twelve reports 12x2 p., laboratory skills 12 p., short quizzes and questions 14 p. A 100 p. in total.

Grades: A: 91-100b, B: 81-90b, C: 71-80b, D: 61-70b, E: 51-60b, Fx: 0-50b.

Based on continuous evaluation.

Learning outcomes:

Students will become familiar with the basic isolation and purification methods used in a synthetic laboratory. Students should master basic laboratory technique and be able to apply the theoretical knowledge from the basic course of organic chemistry in simple synthetic projects.

Brief outline of the course:

Preparation, isolation, purification and identification of organic compounds. The emphasis is on gaining the experimental skills in synthesis of organic compounds, distillation, extraction, crystallization, sublimation and thin-layer chromatography.

Recommended literature:

- 1. Handout with experimental procedures http://kekule.science.upjs.sk/pochu.
- 2. Organic chemistry lectures.

Course language:

Notes:

Course assessment

Total number of assessed students: 282

A	В	С	D	Е	FX
33.33	26.95	21.99	12.06	5.67	0.0

Provides: RNDr. Jana Špaková Raschmanová, PhD., RNDr. Dávid Maliňák, PhD., RNDr. Slávka Hamuľaková, PhD., RNDr. Martin Walko, PhD., RNDr. Zuzana Kudličková, PhD., RNDr. Mária Vilková, PhD., RNDr. Ladislav Janovec, PhD., RNDr. Ján Elečko

Date of last modification: 03.02.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/

Course name: Physical Chemistry

FCHU/10

Course type, scope and the method:

Course type: Lecture / Practice Recommended course-load (hours): Per week: 3 / 2 Per study period: 42 / 28

Course method: present

Number of credits: 6

Recommended semester/trimester of the course: 4.

Course level: I.

Prerequisities: ÚCHV/VCHU/14 or ÚCHV/VCHU/10 or ÚCHV/VACH/10

Conditions for course completion:

Written test Examination

Learning outcomes:

To provide the students with basic knowledge of physical chemistry.

Brief outline of the course:

Fundamental concepts of thermodynamics, thermochemistry, chemical equilibrium, phase equilibria and diagrams, laws for ideal gas and reals gases, liquids, solutions, solutions of electrolytes. Electrochemistry: ionics and electrodics. Electrodes and electrochemical cells, corrosion. Chemical kinetics, catalysis. Adsorption.

Recommended literature:

T. Engel, P. Reid: Physical Chemistry, Pearson Educat. Inc., San Francisco 2006

P.W. Atkins: Physical Chemistry, Oxford University Presss, Oxford 1986, 1990, 1996

W.J. Moore: Physical Chemistry, Longman, London 1972 and newer editions

Course language:

Notes:

Course assessment

Total number of assessed students: 131

A	В	C	D	Е	FX
29.77	15.27	13.74	18.32	16.79	6.11

Provides: RNDr. Andrea Morovská Turoňová, PhD., Mgr. Ján Macko

Date of last modification: 11.02.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚFV/

Course name: Physics

ZVF/03

Course type, scope and the method:

Course type:

Recommended course-load (hours):

Per week: Per study period: Course method: present

Number of credits: 0

Recommended semester/trimester of the course:

Course level: I.

Prerequisities: ((ÚFV/VF1a/07 or ÚFV/VF1a/12) or ÚFV/VF1b/03 and (ÚFV/VF1c/08 or ÚFV/VF1c/10 or ÚFV/VF1c/12) and (ÚFV/VF1d/08 or ÚFV/VF1d/12) and ÚFV/TMEU/03 and ÚFV/TEP1/03 and

Conditions for course completion:

Learning outcomes:

Brief outline of the course:

Recommended literature:

Course language:

Notes:

Course assessment

Total number of assessed students: 69

A	В	С	D	Е	FX
17.39	21.74	36.23	14.49	5.8	4.35

Provides:

Date of last modification: 18.02.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚFV/ Course name: Physics

FDE/07

Course name: Physics in Demonstration Experiments

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of credits: 3

Recommended semester/trimester of the course: 3.

Course level: I.

Prerequisities:

Conditions for course completion:

Learning outcomes:

The goal of the course is to get better the understanding of basic physical concepts and phenomena through demonstrational physical experiments.

Brief outline of the course:

The course is aimed at the conceptual understanding of basic physical concepts and phenomena with the help of selected demonstrational experiments. The experiments concern the content of the subject Introductory physics and their realization is based on students' active participation.

Recommended literature:

- 1. D.Halliday, R.Resnick, J.Walker: Fyzika, VUTIUM, Brno, 2000
- 2.K.Cummings, P.W.Law, E.F.Redish, P.J.Cooney: Understanding Physics,

John Wiley & Sons, Inc., 2004

- 3.P.G.Hewitt: Conceptual Physics, tenth edition, Pearson, Addison Wesley, 2006
- 4.Ľ.Onderová, M.Kireš, Z.Ješková, J.Degro: Praktikum školských pokusov II, PF UPJŠ, 2004

Course language:

Notes:

Course assessment

Total number of assessed students: 66

Α	В	C	D	Е	FX
50.0	25.76	24.24	0.0	0.0	0.0

Provides: doc. RNDr. Zuzana Ješková, PhD., doc. RNDr. Marián Kireš, PhD., RNDr. Ľudmila Onderová, PhD.

Date of last modification: 18.02.2014

Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚFV/ Cou

Course name: Physics Practical I

ZFP1a/03

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 3 Per study period: 42

Course method: present

Number of credits: 3

Recommended semester/trimester of the course: 2.

Course level: I.

Prerequisities:

Conditions for course completion:

The active work during semester and hand in all reports.

Vindication of reports.

Learning outcomes:

Developing proper laboratory habits, skills and verify their theoretical knowledge.

Brief outline of the course:

The goal of this laboratory exercises is to familiarize the students with measurement methods, with kinds and calculus of mistakes, with measured results processing, and with presentation of results. The students gain practical skills, and verify their theoretical knowledge of first semester introductory physics course. They develop proper laboratory habits.

Laboratory assignment:

- 1. Density measurements of liquids and solids.
- 2. Radius measurements of spherical cap. Measurements of surface using planimeter.
- 3. Gravitational acceleration measurements using mathematical and physical pendulum.
- 4. Moment of inertia measurement using physical and torsion pendulum.
- 5. Measurements of Young's modulus.
- 6. Measurement of coefficient of viscosity.
- 7. Measurement of the speed of sound.
- 8. Measurements of general gas constant and Boltzmann constant.
- 9. Measurements of thermal expansivity of air.
- 10. Measurements of thermal capacity of matter.
- 11. Measurement of the surface tension.

Recommended literature:

Degro, J., Ješková, Z., Onderová, Ľ., Kireš, M.: Základné fyzikálne praktikum I. (Basic physical measurements I), Ed. PF UPJŠ Košice 2007.

Standards STN ISO 31. Slovenský inštitút normalizácie v Bratislave (Slovak institute of technical standards in Bratislava),1997.

Ješková, Z.: Computer based experiments in thermodynamics using IP COACH,ed. PF UPJŠ in Košice, 2004.

Course language:

english

Notes:

Course assessment

Total number of assessed students: 174

A	В	С	D	Е	FX
54.02	27.01	12.64	5.17	1.15	0.0

Provides: doc. RNDr. Adriana Zeleňáková, PhD., doc. RNDr. Zuzana Ješková, PhD., doc. RNDr. Marián Kireš, PhD., RNDr. Ľudmila Onderová, PhD., RNDr. Marcela Kajňaková, PhD.

Date of last modification: 18.02.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚFV/

Course name: Physics Practical II

ZFP1b/03

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 3 Per study period: 42

Course method: present

Number of credits: 3

Recommended semester/trimester of the course: 3.

Course level: L

Prerequisities: ÚFV/ZFP1a/03

Conditions for course completion:

Meausirning of experimental tasks, their appreciation in the form of a written report, defending. Further evaluation is also a good theoretical preparation for the measurement of the task.

Learning outcomes:

The objectives of the laboratory are:

- a. To gain some physical inside into some of the concepts presented in the lectures.
- b. To gain some practice in data collection, analysis and interpretation of resumance.
- c. To gain experience and report writing presentation and results.

Brief outline of the course:

Students on practical exercises are working in pairs experimental tasks in the field of electrical, electromagnetic and magnetic properties of matters.

Recommended literature:

Tumanski S, Handbook of magnetic measurements, CRC press, 2011.

Fiorillo F, Characterization and Measurement of Magnetic Materials, Elsevier, 2004.

Course language:

Slovak

Notes:

Course assessment

Total number of assessed students: 150

A	В	C	D	Е	FX
64.0	19.33	14.0	2.0	0.0	0.67

Provides: doc. RNDr. Adriana Zeleňáková, PhD., doc. RNDr. Ján Füzer, PhD.

Date of last modification: 18.02.2014

Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/

Course name: Porous materials and their applications

ADP/03

Course type, scope and the method:

Course type: Lecture / Practice

Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14

Course method: present

Number of credits: 5

Recommended semester/trimester of the course: 6.

Course level: I., II., III.

Prerequisities:

Conditions for course completion:

Written test in the middle and the end of the semester.

Learning outcomes:

To make the acquaintance of various types of advanced porous solids and basic methods for their investigation. To gen up the students with the methods used in characterisation of specific surface area and pore size of different types of porous materials.

Brief outline of the course:

Terminology and principal terms associated with powders, porous solids and adsorption. Methodology of adsorption at the gas-solid interface, liquid-solid interface. Assessment of surface area and porosity. Inorganic materials (active carbon, metal oxides, zeolites, clay minerals, new advanced materials) and phenomenon of adsorption. Application in the industry and everyday life.

Recommended literature:

- 1. F. Rouquerol, J. Rouquerol, K. Sing: Adsorption by powders and porous solids, Academic press, London, UK, 1999
- 2. S. J. Gregg, K.S.W. Sing: Adsorption, surface area and porosity, Academic Press, London,, UK. 1982.
- 3. V. Zeleňák: Adsorption and porosity of solid substances, internal study text, PF UPJŠ, 2007.

Course language:

Notes:

Course assessment

Total number of assessed students: 49

A	В	C	D	Е	FX	N	P
81.63	10.2	4.08	0.0	0.0	0.0	0.0	4.08

Provides: doc. RNDr. Vladimír Zeleňák, PhD.

Date of last modification: 03.02.2014

Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚFV/

Course name: Practical Course III

ZPJF/03

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 3 Per study period: 42

Course method: present

Number of credits: 3

Recommended semester/trimester of the course: 5.

Course level: I.

Prerequisities:

Conditions for course completion:

Learning outcomes:

Practice in nuclear physics.

Brief outline of the course:

Geiger-Müller counter. Analysing power of coincidence circuit by random coincidences. Statistic distribution of measured quantities. Measurement time scale selection. Absorption of beta rays. Backward scattering of beta rays. Gamma rays absorption. Scintillation gamma spectrometer. Determination of 60Co preparat activity using beta-gamma coincidences. Emulsion detector. Dosimetry measurements. Semiconductor detector. Franck Hertz experiment.

Recommended literature:

1. J.Vrláková, S.Vokál: Základné fyzikálne praktikum, skriptá PF UPJŠ, Košice, 2012, dostupné na

http://www.upjs.sk/public/media/5596/Zakladne-fyzikalne-praktikum-III.pdf

Course language:

Notes:

Course assessment

Total number of assessed students: 121

A	В	С	D	Е	FX
64.46	26.45	6.61	0.83	1.65	0.0

Provides: RNDr. Janka Vrláková, PhD., RNDr. Adela Kravčáková, PhD., RNDr. Igor Parnahaj, PhD., RNDr. Zuzana Fecková

Date of last modification: 11.02.2014

Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/

Course name: Practical from Inorganic Chemistry

PACHU/03

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 4 Per study period: 56

Course method: present

Number of credits: 4

Recommended semester/trimester of the course: 2.

Course level: L

Prerequisities: ÚCHV/VCHU/14 or ÚCHV/VCHU/15 or ÚCHV/VCHU/10 or ÚCHV/VACH/10

Conditions for course completion:

Learning outcomes:

The practical acquirements at preparation and study of inorganic compounds and their physicochemical properties by common laboratory techniques.

Brief outline of the course:

The utilization of common laboratory techniques for preparation of elements (H2, O2, Cu, Ni), oxides(CO2, Al2O3·xH2O), nitrides(Mg3N2), acids (HNO3, H3BO3), salts((NH4)2SO4, KMnO4), binary salts(NH4)Fe(SO4)2·12H2O), halides (CuCl, CuCl2·2H2O, SnI4, CuBr2) and coordination compounds ([Cr2(CH3COO)4(H2O)2], [CoCl2(en)2]Cl, [Cu(NH3)4]SO4·H2O, K3[Al(C2O4)3]·3H2O).

Recommended literature:

Z. Vargová, J. Kuchár: Praktikum z anorganickej chémie, Košice, 2008

M. Reháková, M. Dzurillová, V. Zeleňák, V. Urvichiarová: Laboratórna technika, PF UPJŠ, Košice, 1999

Course language:

Notes:

Course assessment

Total number of assessed students: 337

A	В	С	D	Е	FX
44.81	28.49	18.99	3.26	2.97	1.48

Provides: RNDr. Juraj Kuchár, PhD., RNDr. Lukáš Smolko, Mgr. Romana Smolková, RNDr. Martin Vavra, PhD., Mgr. Veronika Farkasová

Date of last modification: 03.02.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/

Course name: Practical in Physical Chemistry

PFCU/03

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 3 Per study period: 42

Course method: present

Number of credits: 4

Recommended semester/trimester of the course: 5.

Course level: I., II.

Prerequisities:

Conditions for course completion:

Approved laboratory reports

Assessment

Learning outcomes:

Theoretical principles, description of each technique and appropriate physical chemistry experiments.

Brief outline of the course:

Experimental verification of theoretical knowledge on thermodynamics, thermochemistry, chemical equilibria (determination of enthalpy, phase diagrams), colligative properties (cryoscopy, ebulioscopy), adsorption.

Experimental verification of theoretical knowledge on electrochemistry (conductivity, dissociation constants, activity coefficients, electromotive force of galvanic cell, Daniell cell, potentials, polarography) and chemical kinetics (determination of rate constants).

Recommended literature:

B.P. Levitt: Findlay's Practical Physical Chemistry, Longman, London 1973

W.J. Moore: Physical Chemistry, Longman, London 1972

P.W. Atkins: Physical Chemistry, Oxford University Press, Oxford, New York 2002

Course language:

Notes:

Course assessment

Total number of assessed students: 211

A	В	С	D	Е	FX
68.25	23.7	7.11	0.95	0.0	0.0

Provides: RNDr. František Kal'avský, RNDr. Andrea Morovská Turoňová, PhD.

Date of last modification: 03.02.2014

Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.

Page: 110

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID:

Course name: Psychology and Health Psychology (Bc. study)

KPPaPZ/PPZBc/12

Course type, scope and the method:

Course type: Lecture / Practice

Recommended course-load (hours): Per week: 1 / 2 Per study period: 14 / 28

Course method: present

Number of credits: 4

Recommended semester/trimester of the course: 5.

Course level: I.

Prerequisities:

Conditions for course completion:

Learning outcomes:

Brief outline of the course:

Recommended literature:

Course language:

Notes:

Course assessment

Total number of assessed students: 92

Α	В	С	D	Е	FX
14.13	20.65	25.0	28.26	11.96	0.0

Provides: PhDr. Anna Janovská, PhD., PhDr. Karolína Barinková, PhD., Mgr. Lucia Hricová

Date of last modification: 04.02.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚFV/

Course name: Quantum Mechanics I.

KVM/08

Course type, scope and the method:

Course type: Lecture / Practice

Recommended course-load (hours): Per week: 3 / 2 Per study period: 42 / 28

Course method: present

Number of credits: 7

Recommended semester/trimester of the course: 5.

Course level: I.

Prerequisities:

Conditions for course completion:

Learning outcomes:

To become familiar with elementary principles of quantum mechanics and to illustrate its possible applications on selected examples.

Brief outline of the course:

A subject matter, experimental and theoretical foundations of quantum mechanics (QM). Basic axioms of QM. Schrödinger equation and its solution for a square potential well, harmonic oscillator and spherically symmetric potentials. Tunnel effect and over-barrier reflection. Spin and Pauli matrices. Systems of identical particles, bosons, fermions and Pauli exclusion principle.

Recommended literature:

- 1. Ľ. Tóth, M. Tóthová, Kvantová a štatistická fyzika I, Rektorát Univerzity P. J. Šafárika, 1982. (in Slovak language)
- 2. Ľ. Skála, Úvod do kvantovej mechaniky, Academia, Praha, 2005. (in Czech language)
- 3. J. Pišút, L. Gomolčák, Úvod do kvantovej mechaniky, Bratislava 1983. (in Slovak language)
- 4. W. Greiner, Quantum Mechanics, 4th edition, Springer, Berlin, 2000.
- 5. A. C. Philips, Introduction to Quantum Mechanics, Wiley, Weinheim, 2003.
- 6. D. J. Griffiths, Introduction to Quantum Mechanics, Prentice Hall, New Jersey, 1995.

Course language:

EN - english

Notes:

Course assessment

Total number of assessed students: 90

A	В	С	D	Е	FX
22.22	16.67	17.78	15.56	21.11	6.67

Provides: doc. RNDr. Jozef Strečka, PhD.

Date of last modification: 31.01.2014

University: P. J. Šafá	University: P. J. Šafárik University in Košice				
Faculty: Faculty of S	cience				
Course ID: ÚTVŠ/ ÚTVŠ/CM/13					
Course type, scope a Course type: Practic Recommended cour Per week: 36 Per st Course method: pre	ce rse-load (hours): tudy period: 504 esent				
Number of credits: 2					
Recommended seme	ester/trimester of the cours	e:			
Course level: I., II.					
Prerequisities:					
Conditions for cours	se completion:				
Learning outcomes:					
Brief outline of the c	course:				
Recommended litera	ature:				
Course language:					
Notes:					
Course assessment Total number of asse	ssed students: 7				
	abs n				
	57.14	42.86			
Provides: Mgr. Alena	a Buková, PhD., Mgr. Agata	Horbacz, PhD.			
Date of last modifica	ntion: 15.01.2014				
Approved: prof RNI	Dr Peter Kollár DrSc doc	RNDr. Vladimír Zeleňák, PhD.			

University: P. J. Šafárik University in Košice				
Faculty: Faculty of Science				
Course ID: ÚCHV/ Course name: Separation Methods ASM/03				
Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14 Course method: present				
Number of credits: 5				
Recommended semester/trimester of the course: 6.				
Course level: I.				
Prerequisities: (ÚCHV/ANCHU/03 or ÚCHV/ANCHE/09 or ÚCHV/ANCH1b/03) and (ÚCHV/PAEC/03 or ÚCHV/PANCH/06 or ÚCHV/PANCHE/09 or ÚCHV/PACU/03)				
Conditions for course completion: Examination				
Learning outcomes: Survey of basic principles, theoretical background and applications of separation methods in research and analytical practice.				
Brief outline of the course: Basic principles, classification, theory and applications of separation methods. Extraction - LLE, SPE, SPME. Chromatographic methods - theory, classification. Gas chromatography, retention mechanisms, stationary phases and their selection. Instrumentation, detectors in GC. Data evaluation - qualitative and quantitative analysis. High-performance liquid chromatography, principles, classification. Stationary and mobile phases in LC, instrumentation. Applications. Comparison of GC and HPLC methods. Planar chromatographic methods - TLC, HPTLC, PC. Electrophoretic techniques - CE, ITP, HPCE. MEKC - micellar electrokinetic capillary chromatography. Lab-on-a-Chip (LOC), TAS, electrophoresis on a chip, principles and applications.				
Recommended literature: Krupčík, J.: Separačné metódy, SVŠT CHTF, Bratislava 1983. Skoog D. A., Leary J. J.: Principles of instrumental analysis. Saunders College Publishing, New York 1997. Pawliszyn J., Lord H. L.: Handbook of sample preparation, Wiley 2010. Churáček J., Jandera P.: Úvod do vysokoúčinné kapalinové chromatografie, SNTL, Praha 1984.				
Course language:				

Notes:

	Course assessment Total number of assessed students: 392					
A	B	C	D	Е	FX	
28.32	25.51	25.0	11.99	6.38	2.81	

Provides: doc. RNDr. Taťána Gondová, CSc.

Date of last modification: 03.02.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚFV/

Course name: Solid State Physics

FKS/03

Course type, scope and the method:

Course type: Lecture

Recommended course-load (hours): Per week: 3 Per study period: 42

Course method: present

Number of credits: 5

Recommended semester/trimester of the course: 6.

Course level: L

Prerequisities: ÚFV/KVM/08

Conditions for course completion:

oral examination

Learning outcomes:

A general introductory course in solid state physics and material science. A student will be able to understand basic theoretical concepts used in solid state physics and interpret selected experimental data. She/he will also be able to adopt simpler experimental techniques.

Brief outline of the course:

Crystal structures and methods of structure analysis. Defects in crystalline solids. Chemical bonding in solids. Thermal properties of crystal lattice. "Free" electrons in metals. The electronic band structure of solids. Transport phenomena in metals and semiconductors. Superconductivity and superfluidity. Magnetic properties of solids. New problems of condensed matter physics.

Recommended literature:

H. Ibach, H. Lüth: Solid-State Physics. Springer - Verlag, Berlin, 1993.

Ch. Kittel: Introduction to Solid State Physics. John Wiley & Sons, Inc. 1976.

Course language:

Notes:

Course assessment

Total number of assessed students: 125

A	В	С	D	Е	FX
17.6	31.2	17.6	16.8	16.8	0.0

Provides: Dr.h.c. prof. RNDr. Alexander Feher, DrSc., prof. Ing. Martin Orendáč, CSc.

Date of last modification: 18.02.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚTVŠ/ | Course n

Course name: Sports Activities I.

TVa/11

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of credits: 2

Recommended semester/trimester of the course: 1.

Course level: I., I.II., II.

Prerequisities:

Conditions for course completion:

Learning outcomes:

Brief outline of the course:

Recommended literature:

Course language:

Notes:

Course assessment

Total number of assessed students: 7160

abs	n	neabs
88.42	7.82	3.76

Provides: PaedDr. Imrich Staško, doc. PhDr. Ivan Šulc, CSc., doc. Mgr. Rastislav Feč, PhD., Mgr. Ivan Matúš, PhD., Mgr. Zuzana Küchelová, Mgr. Peter Bakalár, PhD., doc. PaedDr. Ivan Uher, PhD., PaedDr. Milena Švedová, PhD., Mgr. Agata Horbacz, PhD., Mgr. Marek Valanský, Mgr. Dávid Kaško

Date of last modification: 15.01.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚTVŠ/ | Course

Course name: Sports Activities II.

TVb/11

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of credits: 2

Recommended semester/trimester of the course: 2.

Course level: I., I.II., II.

Prerequisities:

Conditions for course completion:

Learning outcomes:

Brief outline of the course:

Recommended literature:

Course language:

Notes:

Course assessment

Total number of assessed students: 6364

abs	n	neabs
84.95	11.06	3.99

Provides: PaedDr. Imrich Staško, doc. Mgr. Rastislav Feč, PhD., doc. PhDr. Ivan Šulc, CSc., Mgr. Ivan Matúš, PhD., Mgr. Zuzana Küchelová, doc. PaedDr. Ivan Uher, PhD., Mgr. Peter Bakalár, PhD., PaedDr. Milena Švedová, PhD., Mgr. Agata Horbacz, PhD., Mgr. Marek Valanský, Mgr. Dávid Kaško

Date of last modification: 15.01.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚTVŠ/ | Course name: Sports Activities III.

TVc/11

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of credits: 2

Recommended semester/trimester of the course: 3.

Course level: I., I.II., II.

Prerequisities:

Conditions for course completion:

Learning outcomes:

Brief outline of the course:

Recommended literature:

Course language:

Notes:

Course assessment

Total number of assessed students: 4191

abs	n	neabs
89.91	4.72	5.37

Provides: PaedDr. Imrich Staško, doc. Mgr. Rastislav Feč, PhD., doc. PhDr. Ivan Šulc, CSc., Mgr. Ivan Matúš, PhD., Mgr. Zuzana Küchelová, doc. PaedDr. Ivan Uher, PhD., PaedDr. Milena Švedová, PhD., Mgr. Peter Bakalár, PhD., Mgr. Agata Horbacz, PhD., Mgr. Marek Valanský, Mgr. Dávid Kaško

Date of last modification: 15.01.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚTVŠ/ Course

Course name: Sports Activities IV.

TVd/11

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of credits: 2

Recommended semester/trimester of the course: 4.

Course level: I., I.II., II.

Prerequisities:

Conditions for course completion:

Learning outcomes:

Brief outline of the course:

Recommended literature:

Course language:

Notes:

Course assessment

Total number of assessed students: 3363

abs	n	neabs
86.14	6.78	7.08

Provides: PaedDr. Imrich Staško, doc. Mgr. Rastislav Feč, PhD., doc. PhDr. Ivan Šulc, CSc., Mgr. Ivan Matúš, PhD., Mgr. Zuzana Küchelová, PaedDr. Milena Švedová, PhD., Mgr. Peter Bakalár, PhD., doc. PaedDr. Ivan Uher, PhD., Mgr. Agata Horbacz, PhD., Mgr. Marek Valanský, Mgr. Dávid Kaško

Date of last modification: 15.01.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚFV/ **Course name:** Statistical Physics

STA1N/08

Course type, scope and the method: Course type: Lecture / Practice

Recommended course-load (hours): Per week: 2 / 2 Per study period: 28 / 28

Course method: present

Number of credits: 6

Recommended semester/trimester of the course: 6.

Course level: I.

Prerequisities: ÚFV/KVM/08

Conditions for course completion:

Learning outcomes:

Brief outline of the course:

Recommended literature:

Course language:

Notes:

Course assessment

Total number of assessed students: 65

A	В	С	D	Е	FX
43.08	24.62	23.08	6.15	3.08	0.0

Provides: prof. RNDr. Michal Jaščur, CSc., RNDr. Jana Čisárová, PhD.

Date of last modification: 31.01.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/

Course name: Structure determination - spectroscopic methods

MUSU/03

Course type, scope and the method:

Course type: Lecture / Practice

Recommended course-load (hours): Per week: 2 / 3 Per study period: 28 / 42

Course method: present

Number of credits: 8

Recommended semester/trimester of the course: 6.

Course level: I.

Prerequisities: ÚCHV/ACHU/03 and ÚCHV/ANCHU/03 and ÚCHV/OCHU/03

Conditions for course completion:

Learning outcomes:

Brief outline of the course:

Fundamentals of molecular spectroscopy and magnetic properties study, as powerful tools for structure determination in chemistry. Those are ultraviolet, visible, infrared and Raman spectroscopy, mass spectrometry and methods based on magnetic resonance (1H NMR, 13C NMR).

Recommended literature:

1. M. Hesse, H. Meier, B. Zeeh: Spectroscopic Methods in Organic Chemistry. Thieme, NY 1997. 2. L.G.Wade, Jr.: Organic Chemistry. Prentice Hall International, Inc. Englewood Cliffs, New Yersey 1995.

Course language:

Notes:

Course assessment

Total number of assessed students: 231

A	В	С	D	Е	FX
20.35	26.41	30.3	17.75	5.19	0.0

Provides: doc. RNDr. Ján Imrich, CSc., RNDr. Jana Špaková Raschmanová, PhD., RNDr. Monika Tvrdoňová, PhD., RNDr. Juraj Kuchár, PhD.

Date of last modification: 03.02.2014

University: P. J. Šafárik University in Košice					
Faculty: Faculty of Science					
Course ID: ÚTVŠ/ LKSp//13	Course name: Summer Course-Rafting of TISA River				
Course type, scope a Course type: Practic Recommended course week: 36 Per st Course method: pre	ce rse-load (hours): udy period: 504				
Number of credits: 2					
Recommended seme	ster/trimester of the co	irse:			
Course level: I., II.					
Prerequisities:					
Conditions for cours	e completion:				
Learning outcomes:					
Brief outline of the c	ourse:				
Recommended litera	nture:				
Course language:					
Notes:					
Course assessment Total number of asse	ssed students: 63				
abs n					
41.27 58.73					
Provides: Mgr. Peter	Bakalár, PhD.				
Date of last modifica	tion: 15.01.2014				
Approved: prof. RNI	Dr. Peter Kollár, DrSc., d	oc. RNDr. Vladimír Zeleňák, PhD.			

University: P. J. Šafárik University in Košice						
Faculty: Faculty of S	cience					
Course ID: ÚTVŠ/ KP/12	ΓVŠ/ Course name: Survival Course					
Course type, scope a Course type: Practic Recommended cour Per week: 36 Per st Course method: pre	ce rse-load (hours): udy period: 504 esent					
Number of credits: 2		,				
Recommended seme	ster/trimester of the cou	rse:				
Course level: I., II.						
Prerequisities:						
Conditions for cours	e completion:					
Learning outcomes:	Learning outcomes:					
Brief outline of the c	ourse:					
Recommended litera	ture:					
Course language:						
Notes:						
Course assessment Total number of asses	ssed students: 185					
abs n						
41.62 58.38						
Provides: Mgr. Marel	k Valanský	-				
Date of last modifica	tion: 15.01.2014					
Approved: prof. RNI	Dr. Peter Kollár, DrSc., do	c. RNDr. Vladimír Zeleňák, PhD.				

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚFV/

Course name: Theoretical Mechanics

TMEU/03

Course type, scope and the method:

Course type: Lecture / Practice

Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14

Course method: present

Number of credits: 4

Recommended semester/trimester of the course: 3.

Course level: I.

Prerequisities: ÚFV/VF1a/12

Conditions for course completion:

Two tests to deal with specific tasks mechanics.

Final examination.

Learning outcomes:

To acquaint students with principles of the theoretical mechanics.

Brief outline of the course:

Mechanics of particle system with constraints. Principle of virtual work and d'Alembert's principle. Lagrange's function and Lagrange's equations of motion. Hamilton's principle, Hamilton's function and Hamilton's canonical equations of motion. Mechanics of rigid body. Kinematics and dynamics of rigid body.

Recommended literature:

- 1. Meirovitch L.: Methods of Analytical dynamics, McGraw-Hill, New York, 1970.
- 2. Taylor T.T.: Mechanics: Classical and Quantum, Pergamon Press, Oxford, 1976.
- 3. Strelkov S.P.: Mechanics, Mir Publishers, Moscow, 1985.
- 4. Greiner W.: Classical Mechanics, Springer-Verlag, Berlin, 2010.
- 5. Goldstein H.: Classical Mechanics, Addison-Wesley, London, 1970.
- 6. Barger V., Olsson M.: Classical Mechanics: A Modern Perspective, McGraw-Hill, London, 1973.

Course language:

Slovak

Notes:

Course assessment

Total number of assessed students: 134

A	В	С	D	Е	FX
15.67	18.66	14.93	21.64	24.63	4.48

Provides: prof. RNDr. Andrej Bobák, DrSc., RNDr. Michal Borovský

Date of last modification: 31.01.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚFV/

Course name: Theory of the Electromagnetic Field

TEP1/03

Course type, scope and the method:

Course type: Lecture / Practice

Recommended course-load (hours): Per week: 3 / 1 Per study period: 42 / 14

Course method: present

Number of credits: 5

Recommended semester/trimester of the course: 4.

Course level: I.

Prerequisities: ÚFV/VF1b/03

Conditions for course completion:

Two tests to deal with specific tasks theory of the electromagnetic field.

Examination.

Learning outcomes:

To acquaint students with principles of a theory of the electromagnetic field.

Brief outline of the course:

Maxwell equations in vacuum. Scalar and vector potentials. Conservation laws. Electrostatic field. Static magnetic field. Maxwell equations in macroscopic media. Quasistatic electromagnetic field. Electromagnetic waves. Radiation of electromagnetic waves.

Recommended literature:

- 1. Jackson J.D.: Classical Electrodynamics, John Wiley, New York, 1975.
- 2. Rao N.N.: Basic Electromagnetics with Applications, Prentice-Hall, New Jersey, 1972.
- 3. Greiner W.: Classical Electrodynamics, Springer-Verlag, New York, 1998.

Course language:

- 1. Slovak,
- 2. English

Notes:

Course assessment

Total number of assessed students: 238

A	В	C	D	Е	FX
26.89	7.56	17.65	23.11	16.39	8.4

Provides: prof. RNDr. Andrej Bobák, DrSc., RNDr. Tomáš Lučivjanský, PhD.

Date of last modification: 31.01.2014

Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.

Page: 128

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚTVŠ/ Course name: Winter Ski Training Course ZKLS//13 Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 36 Per study period: 504 Course method: present Number of credits: 2 Recommended semester/trimester of the course: Course level: I., II. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 59 abs n 25.42 74.58 Provides: PaedDr. Imrich Staško, doc. PhDr. Ivan Šulc, CSc. Date of last modification: 15.01.2014 Approved: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Vladimír Zeleňák, PhD.