University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ Course name: Analytická chémia RANC/06 Course type, scope and the method: Course type: Lecture **Recommended course-load (hours):** Per week: Per study period: 24s Course method: present **Number of credits: 8** Recommended semester/trimester of the course: 2. Course level: N **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 3 \mathbf{C} A В D Е FX 100.0 0.0 0.0 0.0 0.0 0.0 Provides: doc. RNDr. Taťána Gondová, CSc. Date of last modification: 03.02.2014 Approved: doc. RNDr. Mária Ganajová, CSc.

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ Course name: Anorganická chémia RACH/06 Course type, scope and the method: Course type: Lecture **Recommended course-load (hours):** Per week: Per study period: 36s Course method: present Number of credits: 12 **Recommended semester/trimester of the course:** 1. Course level: N **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 3 C Α В D Е FX 0.0 0.0 100.0 0.0 0.0 0.0 Provides: prof. RNDr. Katarína Györyová, DrSc., doc. RNDr. Jozef Chomič, CSc. Date of last modification: 03.02.2014 Approved: doc. RNDr. Mária Ganajová, CSc.

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ **Course name:** Basis of Mineralogy RMIN/06 Course type, scope and the method: Course type: Lecture **Recommended course-load (hours):** Per week: Per study period: 24s Course method: present **Number of credits: 8** Recommended semester/trimester of the course: 4. Course level: N **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 9 \mathbf{C} A В D Е FX 100.0 0.0 0.0 0.0 0.0 0.0 Provides: doc. RNDr. Ivan Potočňák, PhD. Date of last modification: 03.02.2014 Approved: doc. RNDr. Mária Ganajová, CSc.

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ Course name: Biochemistry RBCH/06 Course type, scope and the method: Course type: Lecture **Recommended course-load (hours):** Per week: Per study period: 24s Course method: present Number of credits: 10 **Recommended semester/trimester of the course:** 3. Course level: N **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 9 \mathbf{C} A В D Е FX 100.0 0.0 0.0 0.0 0.0 0.0 Provides: doc. RNDr. Viktor Víglaský, PhD. Date of last modification: 03.02.2014 Approved: doc. RNDr. Mária Ganajová, CSc.

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/

Course name: Biochemistry Practical

RPBCH/06

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: Per study period: 24s

Course method: present

Number of credits: 6

Recommended semester/trimester of the course: 4.

Course level: N

Prerequisities:

Conditions for course completion:

Learning outcomes:

To allow students to get practical experience in experimental techniques and methods, currently used in a biochemical research: UV/VIS spectrophotometry, thin layer chromatography (TLC), gel electrophoresis, isolation of macromolecules and substances from biological materials and their quantitative and qualitative determination.

Brief outline of the course:

The most important biochemical laboratory methods. The qualitative tests for amino acids and proteins. Time-dependent course of enzyme-catalyzed reaction: determination of enzymatic activity, determination of the first order rate constant, calculations of math models (examples), effect of a substrate concentration on initial rate of reaction, determination of Km and Vmax for urease. Isolation and detection of nucleic acids.

Recommended literature:

Sedlák, Danko, Varhač, Paulíková, Podhradský: Practical exercises from biochemistry, 2007, http://kosice.upjs.sk/~kbch/document.php?name=pbc&lang=sk

Course language:

Notes:

Course assessment

Total number of assessed students: 3

A	В	С	D	Е	FX
100.0	0.0	0.0	0.0	0.0	0.0

Provides: doc. RNDr. Viktor Víglaský, PhD.

Date of last modification: 03.02.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/

Course name: Bioinorganic Chemistry

RBAC/06

Course type, scope and the method:

Course type: Lecture

Recommended course-load (hours): Per week: Per study period: 24s

Course method: present

Number of credits: 8

Recommended semester/trimester of the course: 5.

Course level: N

Prerequisities:

Conditions for course completion:

Learning outcomes:

Brief outline of the course:

Content: Metalic and non-metalic elements and their roles in biological systems(biometals, bulk biological elements, essential trace elements). Biocoordination compounds, bioligands. Biocatalyzers. Oxygen carriers and oxygen transport proteins. Photochemical process. Catalysis and regulation of bioenergetic processes by the alkaline earth metal ions. Calcium biominerals and biomineralization. Toxic metals. Application of knowledge of bioinorganic chemistry in pharmacy, chemotherapy (e.g. platinum complexes in cancer therapy)radiodiagnostics, mineral biotechnology and in other branches of life.

Recommended literature:

Recommended reading: Kaim W., Schwederski B.: Bioinorganic Chemistry: Inorganic Elements in the Chemistry of Life. Wiley, Chichester 1998.

Wilkins P. C., Wilkins R. G.: Inorganic Chemistry in Biology. OCP, Oxford 1997.

Course language:

Notes:

Course assessment

Total number of assessed students: 0

A	В	C	D	Е	FX
0.0	0.0	0.0	0.0	0.0	0.0

Provides: doc. RNDr. Mária Reháková, CSc.

Date of last modification: 03.02.2014

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ Course name: Bioorganická chémia RBOC/06 Course type, scope and the method: Course type: Lecture **Recommended course-load (hours):** Per week: Per study period: 24s Course method: present **Number of credits: 8 Recommended semester/trimester of the course:** 5. Course level: N **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 1 \mathbf{C} A В D Е FX 100.0 0.0 0.0 0.0 0.0 0.0 Provides: prof. RNDr. Jozef Gonda, DrSc. Date of last modification: 03.02.2014 Approved: doc. RNDr. Mária Ganajová, CSc.

COURSE INFORMATION LETTER							
University: P. J. Šafárik University in Košice							
Faculty: Faculty of S	Faculty: Faculty of Science						
Course ID: ÚCHV/ RECH/06	Course name: Electrochemical Methods						
Course type, scope a Course type: Lectur Recommended cour Per week: Per stud Course method: pre	re rse-load (hours): ly period: 24s						
Number of credits: 8	3						
Recommended seme	ster/trimester of the course: 4.						
Course level: N							
Prerequisities:							
Conditions for cours	se completion:						
Learning outcomes:							
of practice, electrocland diffusion, Cottrel backround, examples voltammetry, pulse te square - wave voltam voltammetry, adsorpt analysis), working eleglassy carbon electrodering-disk electrode, ul of ion selective electrodes detectors in flow syst	canalytical methods for environmental control and protection, requirements themical cells, electrode potential, mass transfer by convection, migration all equation, direct current voltametry and polarography(principle, theoretical sof practical application). TAST polarography and voltametry, staircase echniques: normal pulse and differential pulse voltammetry and polarography, metry and polarography, AC polarography and voltammetry, anodic stripping ive(or accumulation) voltammetry (applications in clinical and environmental ectrodes in voltammetry: stationary mercury electrode, mercury film electrode, de, carbon paste electrode,metallic electrodes, rotating disk electrode, rotating ltramicroelectrodes, chemically modified electrodes, potentiometry, principles trodes, glass electrodes, ISE with solid and liquid membranes, biocatalytic strodes, glass electrodes, biamperometric stripping analysis, electroanalytical ems, amperometric titrations, biamperometric and bipotentiometric titrations, lyanostatic coulometry.						
J. Wang: Analytical F Electroanalytical Met York 1987 A.J. Bard, L.R. Faulk T. Riley, A. Watson: I Chichester 1987	alytical Methods, Springer Vrlg., Heidelberg 2002, ISBN 3-540-42449-3 Electrochemistry, VCH Publ., New York 1994,2000 R. Kalvoda (Ed.): thods in Chemical and Environmental Analysis, Plenum Publ. Corp., New there: Electrochemical Methods, Jofn Wiley and Sons, New York 1980 Polarography and Other Voltametric methods, John Wiley and Sons, malysis, VCH Publ. Inc., Deerfield Beach 19858						
Course language:							

Notes:

Course assessment Total number of assessed students: 0						
A	В	С	D	Е	FX	
0.0 0.0 0.0 0.0						
Duavidan dan DNDr Kvistarlava Marlavčavá CCa						

Provides: doc. RNDr. Kvetoslava Markušová, CSc.

Date of last modification: 03.02.2014

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ Course name: Chémia a didaktika chémie RCHDCH/04 Course type, scope and the method: **Course type: Recommended course-load (hours):** Per week: Per study period: Course method: present Number of credits: 0 Recommended semester/trimester of the course: Course level: N Prerequisities: ÚCHV/RACH/06 and ÚCHV/ROCH/06 and ÚCHV/RDCH2/06 and ÚCHV/ RDCH1/06 **Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes: Course assessment** Total number of assessed students: 9 \mathbf{C} В Ε FX Α D 44.44 11.11 33.33 11.11 0.0 0.0 **Provides:** Date of last modification: 03.02.2014

Page: 10

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ Course name: Methodology of Chemistry Teaching I RDCH1/06 Course type, scope and the method: Course type: Lecture **Recommended course-load (hours):** Per week: Per study period: 24s Course method: present **Number of credits: 9** Recommended semester/trimester of the course: 4. Course level: N **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 3 \mathbf{C} A В D Е FX 100.0 0.0 0.0 0.0 0.0 0.0 Provides: doc. RNDr. Mária Ganajová, CSc. Date of last modification: 03.02.2014 Approved: doc. RNDr. Mária Ganajová, CSc.

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ Course name: Methodology of Chemistry Teaching II RDCH2/06 Course type, scope and the method: Course type: Lecture **Recommended course-load (hours):** Per week: Per study period: 24s Course method: present **Number of credits: 9 Recommended semester/trimester of the course:** 5. Course level: N **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 9 \mathbf{C} A В D Е FX 88.89 11.11 0.0 0.0 0.0 0.0 Provides: doc. RNDr. Mária Ganajová, CSc. Date of last modification: 03.02.2014 Approved: doc. RNDr. Mária Ganajová, CSc.

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ Course name: Metódy určovania štruktúry, spektrálne metódy MOSU/06 Course type, scope and the method: Course type: Lecture **Recommended course-load (hours):** Per week: Per study period: 24s Course method: present **Number of credits: 8** Recommended semester/trimester of the course: 4. Course level: N **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 0 \mathbf{C} Α В D Е FX 0.0 0.0 0.0 0.0 0.0 0.0 Provides: doc. RNDr. Ján Imrich, CSc., doc. RNDr. Jozef Chomič, CSc. Date of last modification: 03.02.2014 Approved: doc. RNDr. Mária Ganajová, CSc.

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ Course name: Obhajoba záverečnej práce ROZP/12 Course type, scope and the method: **Course type:** Recommended course-load (hours): Per week: Per study period: Course method: present Number of credits: 0 Recommended semester/trimester of the course: Course level: N **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 0 \mathbf{C} Α В D Ε FX 0.0 0.0 0.0 0.0 0.0 0.0 **Provides:** Date of last modification: 03.02.2014 Approved: doc. RNDr. Mária Ganajová, CSc.

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ Course name: Organická chémia ROCH/06 Course type, scope and the method: Course type: Lecture **Recommended course-load (hours):** Per week: Per study period: 36s Course method: present Number of credits: 12 Recommended semester/trimester of the course: 2. Course level: N **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 3 \mathbf{C} A В D Е FX 100.0 0.0 0.0 0.0 0.0 0.0 Provides: prof. RNDr. Jozef Gonda, DrSc. Date of last modification: 03.02.2014 Approved: doc. RNDr. Mária Ganajová, CSc.

University: P. J. Šafárik University in Košice					
Faculty: Faculty of Science					
Course ID: ÚCHV/ RPP/07	Course name: Pedagogic	ká prax			
Course type, scope a Course type: Practic Recommended cou Per week: Per stud Course method: pre	ce rse-load (hours): ly period: 20s esent				
Number of credits: 6					
Recommended seme	ster/trimester of the cour	se: 5.			
Course level: N					
Prerequisities:					
Conditions for cours	se completion:				
Learning outcomes:					
Brief outline of the c	ourse:				
Recommended litera	nture:				
Course language:					
Notes:					
Course assessment Total number of asse	ssed students: 9				
abs n					
100.0 0.0					
Provides: PhDr. Silvi	a Kontírová, PhD., Mgr. M	ária Sarková, PhD.			
Date of last modifica	ntion: 03.02.2014				
Approved: doc. RNDr. Mária Ganajová, CSc.					

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ **Course name:** Physical Chemistry RFCH/06 Course type, scope and the method: Course type: Lecture **Recommended course-load (hours):** Per week: Per study period: 24s Course method: present **Number of credits: 8** Recommended semester/trimester of the course: 1. Course level: N **Prerequisities: Conditions for course completion:** Examination **Learning outcomes:** To provide the students with basic knowledge of physical chemistry. **Brief outline of the course:** State of aggregation, laws for ideal and real gases, liquids and solids - characteristics and properties. Principles of thermodynamics, thermodynamic equilibrium, characteristic thermodynamic changes, heat, work, internal energy, enthalpy, entropy, 1st, 2nd and 3rd law of thermodynamics, Gibbs energy. Thermochemistry, heat of reaction, 1st and 2nd thermometric laws, enthalpy of formation, enthalpy of combustion, calorimetry. Phase equilibria, Gibbs' phase rule, phase diagrams for 1-, 2- and 3-componental systems, colligative properties, activity. Adsorption, adsorption isotherms. Diffusion. Chemical equilibrium, van't Hoff's reaction isotherm, isobar and isochore, influence of temperature and pressure on chemical equilibrium. Electrochemistry. Conductivity of electrolytes, utilization, Faraday's law, strong electrolytes - theory, activity coefficients, ionic strength. Weak electrolytes, theories of acids and bases, buffer solutions, hydrolysis of salts. Galvanic cells, electromotive force of cells, Nernst equation, electrodes of 1st and 2nd kind, redox electrodes, Peters equation, standard electrode potentials, potentiometric measurments, ion selective electrodes. Electrode processes, polarization of electrodes, concentration cells, corrosion of metals and passivity. Principles of polarography and voltammetric methods. Chemical kinetics - reaction types and mechanism, reaction rate, molecularity and order of reaction, rate laws for 1st and 2nd order reactions, reaction mechanisms, reaction rate theories, temperature dependence of rate constants. Catalysis - homogeneous and heterogeneous, acidobasic catalysis, enzyme catalysis. Colloids classification, preparation, stability, optical properties, dialysis. **Recommended literature:** T. Engel, P. Reid: Physical Chemistry, Pearson Educat. Inc., San Francisco, 2006 P.W. Atkins: Physical Chemistry, Oxford University Presss, Oxford, 1986, 1990, 1996 W.J. Moore: Physical Chemistry, Longman, London, 1972 and newer editions

Course language:

Notes:

	Course assessment							
Total number of assessed students: 3								
	A	В	C	D	Е	FX		
	0.0	0.0	100.0	0.0	0.0	0.0		

Provides: doc. RNDr. Kvetoslava Markušová, CSc., doc. RNDr. Renáta Oriňáková, PhD., RNDr. Daniela Kladeková, CSc.

Date of last modification: 03.02.2014

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/

Course name: Practical in Physical Chemistry

RPFC/06

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: Per study period: 12s

Course method: present

Number of credits: 3

Recommended semester/trimester of the course: 2.

Course level: N

Prerequisities:

Conditions for course completion:

Approved laboratory reports

Assessment

Learning outcomes:

Theoretical principles, description of each technique and appropriate physical chemistry experiments.

Brief outline of the course:

Experimental verification of theoretical knowledge on thermodynamics, thermochemistry, chemical equilibria (determination of enthalpy, phase diagrams), colligative properties (cryoscopy, ebulioscopy), adsorption.

Experimental verification of theoretical knowledge on electrochemistry (conductivity, dissociation constants, activity coefficients, electromotive force of galvanic cell, Daniell cell, potentials, polarography) and chemical kinetics (determination of rate constants).

Recommended literature:

B.P. Levitt: Findlay's Practical Physical Chemistry, Longman, London, 1973

W.J. Moore: Physical Chemistry, Longman, London, 1972

P.W. Atkins: Physical Chemistry, Oxford University Press, Oxford, New York, 2002

Course language:

Notes:

Course assessment

Total number of assessed students: 0

A	В	С	D	Е	FX
0.0	0.0	0.0	0.0	0.0	0.0

Provides: RNDr. František Kal'avský, RNDr. Andrea Morovská Turoňová, PhD.

Date of last modification: 03.02.2014

Approved: doc. RNDr. Mária Ganajová, CSc.

Page: 19

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ Course name: Praktikum z analytickej chémie RPAC/06 Course type, scope and the method: Course type: Practice **Recommended course-load (hours):** Per week: Per study period: 12s Course method: present Number of credits: 3 Recommended semester/trimester of the course: 3. Course level: N **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 0 \mathbf{C} Α В D Е FX 0.0 0.0 0.0 0.0 0.0 0.0 Provides: doc. RNDr. Taťána Gondová, CSc. Date of last modification: 03.02.2014 Approved: doc. RNDr. Mária Ganajová, CSc.

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ Course name: Praktikum z anorganickej chémie RPACH/06 Course type, scope and the method: Course type: Practice **Recommended course-load (hours):** Per week: Per study period: 24s Course method: present **Number of credits:** 6 Recommended semester/trimester of the course: 2. Course level: N **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 0 \mathbf{C} Α В D Е FX 0.0 0.0 0.0 0.0 0.0 0.0 Provides: doc. RNDr. Jozef Chomič, CSc. Date of last modification: 03.02.2014 Approved: doc. RNDr. Mária Ganajová, CSc.

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ Course name: Praktikum z organickej chémie RPOC/06 Course type, scope and the method: Course type: Practice **Recommended course-load (hours):** Per week: Per study period: 24s Course method: present **Number of credits:** 6 **Recommended semester/trimester of the course:** 3. Course level: N **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 0 \mathbf{C} Α В D Е FX 0.0 0.0 0.0 0.0 0.0 0.0 Provides: doc. RNDr. Miroslava Martinková, PhD. Date of last modification: 03.02.2014 Approved: doc. RNDr. Mária Ganajová, CSc.

Page: 22

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ Course name: Všeobecná chémia+chemické výpočty RVCH/06 Course type, scope and the method: Course type: Lecture **Recommended course-load (hours):** Per week: Per study period: 36s Course method: present Number of credits: 12 **Recommended semester/trimester of the course:** 1. Course level: N **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 3 \mathbf{C} A В D Е FX 100.0 0.0 0.0 0.0 0.0 0.0 Provides: prof. RNDr. Katarína Györyová, DrSc., doc. RNDr. Jozef Chomič, CSc. Date of last modification: 03.02.2014 Approved: doc. RNDr. Mária Ganajová, CSc.

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ Course name: Záverečná práca RSZP1/00 Course type, scope and the method: **Course type:** Recommended course-load (hours): Per week: Per study period: Course method: present Number of credits: 10 **Recommended semester/trimester of the course:** 5. Course level: N **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 9 \mathbf{C} Α В D Ε FX 100.0 0.0 0.0 0.0 0.0 0.0 **Provides:** Date of last modification: 03.02.2014 Approved: doc. RNDr. Mária Ganajová, CSc.