University:	ΡJ	Šafárik	University	in Košice
University.	1	Salarin	Oniversity	III IXOSICC

Faculty: Faculty of Science

Course ID: ÚCHV/	Course name: Inorganic Chemistry II
ACH2/03	

### **Course type, scope and the method: Course type:** Lecture / Practice

Recommended course-load (hours):

Per week: 3 / 2 Per study period: 42 / 28

Course method: present

Number of credits: 7

### **Recommended semester/trimester of the course:**

Course level: I.

Prerequisities: ÚCHV/ACH1/03 or ÚCHV/ACH1/10 or ÚCHV/ACHU/03

### **Conditions for course completion:**

Written examination at the end of the course. The final mark is given by the sum of points from seminars (max. 10 points) and 3x30 points from written test, totally 100 points. To pass it is required to obtain at least 51 points as well as 51 % of points from every partial examination.

### Learning outcomes:

Goal of the course is to provide the students with a knowledge of systematic chemistry of metallic elements.

### Brief outline of the course:

Electronic configuration, abundance, use, physical and chemical properties and reactivity of the elements of the 1st, 2nd groups, transition metal elements, elements of the 12th group, Al, Ga, In, Tl, Ge, Sn, Pb, As, Sb, Bi, Se, Te, Po, lanthanides and actinides. Binary and other compounds formed by these elements, their properties and reactivity. General properties, structure and bonding in metals, co-ordination and organometallic compounds.

#### **Recommended literature:**

1. Greenwood, N. N., Earnshaw, A: Chemistry of the Elements. Pergamon Press, Oxford, 1984 2. Shriver, D.F., Atkins, P.W., Langford, C. H.: Inorganic Chemistry. 2ndEd., Oxford University Press, Oxford, 1995

### **Course language:**

#### **Course assessment**

Total number of assessed students: 602

А	В	С	D	Е	FX
11.63	19.1	30.4	26.58	7.81	4.49

Provides: prof. RNDr. Juraj Černák, DrSc., doc. RNDr. Juraj Kuchár, PhD.

### Date of last modification: 26.02.2018

	Šafárik Univers	sity in Kosice			
Faculty: Faculty	v of Science				
Course ID: ÚCI ACHU/03	HV/ Course na	ame: Inorganic C	hemistry		
Course type, sc	-				
• 1	Lecture / Practico I course-load (h				
	Per study peri	/			
Course metho	d: present				
Number of cred	lits: 6				
Recommended	semester/trime	ster of the cours	e: 2.		
Course level: I.					
Prerequisities:	ÚCHV/VCHU/1	0 or ÚCHV/VCH	IU/14 or ÚCHV/	/VCHU/15	
<b>Conditions for</b> Test in the midd Oral examination	lle and at the end	ion: 1 of the semester.			
Learning outco Aim of the cour metallic elemen	se is to provide	the students with	a knowledge of s	systematic chemi	stry of non-
of non-metallic silicon, boron a	elements hydro and rare gases. reactivity. Met	ance, use, physica ogen, halogens, o Binary and othe tals and transitio	oxygen, sulphur, er compounds fo	nitrogen, phosp prmed by these	phorus, carbon elements, thei
<b>Recommended</b> http://kosice.upj self study)	s.sk/~vladimir.z	elenak/ACHU.ht		m the lectures as gamon Press, Oxf	a support for
Greenwood, N.	ton T., Rourke J	., Weller M., Arm	strong F.: Inorga	anic Chemistry, U	
Greenwood, N. Atkins O., Over Press, Oxford, 2	ton T., Rourke J 2006.	•	strong F.: Inorga	nnic Chemistry, U	
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Greenwood, N. Atkins O., Over Press, Oxford, 2 Course languag Course assessm	ton T., Rourke J 2006. ge: ent	., Weller M., Arm	nstrong F.: Inorga	E	
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Greenwood, N. Atkins O., Over Press, Oxford, 2 Course languag Course assessm Total number of A	ton T., Rourke J 2006. ee: ent Sassessed studer B 20.12 RNDr. Vladimír	., Weller M., Arm nts: 661 C 30.41 Zeleňák, PhD.	D	E	Jniversity FX

University: P. J.	Šafárik Univers	ity in Košice			
Faculty: Faculty	y of Science				
Course ID: ÚB ACL/03	EV/ Course na	me: Human An	atomy		
Recommended	Lecture / Practice l course-load (h 2 Per study perio	ours):			
Number of cred	lits: 5				
Recommended	semester/trimes	ster of the cours	<b>e:</b> 3.		
Course level: I.					
Prerequisities:					
<b>Conditions for</b> Written examination	-	on:			
Learning outco Anatomic system					
	inology, skeleto			nal system, resp gans, nervous syste	5 5
Anatomy in 3 V and Volume 3: N Thieme Medica	nhardt, H., Platze	e 1 : Locomotor S and Sensory Org New York, 199	System, Volume ans 3	2: Internal Organ	IS
Course languag	ge:				
Course assessm Total number of	ent fassessed studen	ts: 1652			
А	В	С	D	E	FX
4.6	16.71	27.36	25.48	22.82	3.03
Provides: RND	r. Juraj Ševc, PhI	D., RNDr. Anna	Alexovič Matia	šová, PhD.	
Date of last mo	dification: 23.02	2.2018			
Approved: Gua PhD.	ranteeprof. RND	r. Pavol Mártoni	i, PhD.Guaranto	eeprof. RNDr. Vla	dimír Zeleňák,

	P. J. Safárik	University i	n Košice				
Faculty: Facu	ulty of Scie	ence					
<b>Course ID:</b> Ú ADP/03	ÚCHV/ C	ourse name	: Porous mat	erials and the	eir applicatio	ons	
Course type, Course type Recommend Per week: 2 Course met	e: Lecture / ded course 2 / 1 Per stu	Practice -load (hours idy period:	s):				
Number of c	redits: 5						
Recommend	ed semeste	er/trimester	of the cours	<b>e:</b> 6.			
Course level:	<b>:</b> I., II., III.						
Prerequisitie	es:						
<b>Conditions fo</b> Written test i		1	d of the sem	ester.			
investigation		the students	with the mo	thode used in	n characteria	ation of man	ific surface
	e size of dif of the cou	fferent types	of porous ma	aterials.			
	e size of dif of the cou and prin y of adsorpt osity. Inorg	fferent types rse: cipal terms tion at the ga ganic materia	of porous ma associated s-solid interf als (active ca	aterials. with powde face, liquid-s urbon, metal	ers, porous olid interface oxides, zeol	solids and e. Assessmer ites, clay mi	adsorption adsorption of surfac
area and pore Brief outline Terminology Methodology area and por	e size of dif of the cou and prin y of adsorpt osity. Inorg terials) and ed literatu rol, J. Roud on, UK, 199 g, K.S.W. S	fferent types rse: cipal terms tion at the ga ganic materia l phenomeno re: querol, K. Sin 9 Sing: Adsorpt	of porous ma associated s-solid interf als (active ca on of adsorption ng: Adsorption tion, surface	aterials. with powde face, liquid-s arbon, metal ion. Applicat on by powde area and por	ers, porous olid interface oxides, zeol ion in the inc ers and porou osity, Acade	solids and e. Assessmer ites, clay mi dustry and ev s solids, Aca mic Press, L	adsorption nt of surfac inerals, new veryday life ademic ondon,,
area and pore Brief outline Terminology Methodology area and pore advanced ma Recommend 1. F. Rouquer press, Londo 2. S. J. Gregg UK, 1982.	e size of dif of the cou and prin y of adsorpt osity. Inorg terials) and ed literatu rol, J. Roud on, UK, 199 g, K.S.W. S	fferent types rse: cipal terms tion at the ga ganic materia l phenomeno re: querol, K. Sin 9 Sing: Adsorpt	of porous ma associated s-solid interf als (active ca on of adsorption ng: Adsorption tion, surface	aterials. with powde face, liquid-s arbon, metal ion. Applicat on by powde area and por	ers, porous olid interface oxides, zeol ion in the inc ers and porou osity, Acade	solids and e. Assessmer ites, clay mi dustry and ev s solids, Aca mic Press, L	adsorption nt of surfac inerals, new veryday life ademic ondon,,
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PhD.

University: P. J. Š	afárik Univers	ty in Košice			
Faculty: Faculty o	f Science				
Course ID: KPE/ ALP/06	Course na	me: Alternative	Education		
Course type, scop Course type: Pra Recommended c Per week: 2 Per Course method:	ctice ourse-load (he study period:	ours):			
Number of credits	s: 2				
Recommended se	mester/trimes	ter of the cours	<b>e:</b> 4.	_	
Course level: I.					
Prerequisities:					
Conditions for co	urse completi	on:			
Learning outcom	es:				
Brief outline of th	e course:				
Recommended lit	erature:				
Course language:					
Course assessmen Total number of as	-	s: 180			
A	В	С	D	Е	FX
66.11	30.56	0.56	1.11	0.56	1.11
Provides: Mgr. Ka	tarína Petríkov	vá, PhD.			1
Date of last modif	ication: 23.08	.2017			
<b>Approved:</b> Guaran PhD.	nteeprof. RND	r. Pavol Mártonf	i, PhD.Guarante	eprof. RNDr. Vla	dimír Zeleňák,

	University:	ΡJ	Šafárik	University	v in Košice
I	University.	1	Salarik	Oniversity	

Faculty: Faculty of Science

Course ID: ÚCHV/ Course name: Instrumental Analytical Chemistry ANCH1b/03

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14 Course method: present

Number of credits: 5

**Recommended semester/trimester of the course:** 

Course level: I.

**Prerequisities:** 

Conditions for course completion:

Test

#### Learning outcomes:

Getting a knowledge of the theoretical principles and instrumentation in analytical chemistry.

### **Brief outline of the course:**

Spectroscopic methods of analysis. Electromagnetic radiation. Basic components of spectroscopic instrumentation. Sources of energy. Detectors. Spectroscopy based on absorption. Transmittance and absorbance. Beer's Law. Limitations to Beer's Law. Ultraviolet-visible and infrared spectrophotometry. Atomic absorption spectroscopy. Spectroscopy based on emission. Molecular photoluminescence spectroscopy. Atomic emission spectroscopy. Spectroscopy based on scattering. Mass spectrometry. Electrochemical methods of analysis. Potentiometric methods of analysis. Reference electrodes. Membrane electrodes. Coulometric methods of analysis. Voltammetric methods of analysis. Chromatographic methods. General theory of column chromatography. Optimizing chromatographic separations. Gas chromatography. High-performance liquid chromatography. Ion-exchange chromatography. Supercritical fluid chromatography.

#### **Recommended literature:**

1. Labuda a kol. Analytická chémia. ISBN: 9788022742429, Vydavateľstvo: STU Bratislava, Rok vydania: 2014, Počet strán: 671

2. Christian G.D. Analytical Chemistry. John Wiley & Sons, Inc. New York – Chichester – Brisbane – Toronto – Singapore 1994.

3. Holtzclaw H.F., Jr., Robinson W.R. College Chemistry with Qualitation Analysis. D.C. Heath and Company 1988.

#### **Course language:**

# Course assessment

Total number of assessed students: 519

А	В	С	D	Е	FX
20.42	12.14	21.58	19.08	26.4	0.39

**Provides:** prof. Mgr. Vasil' Andruch, DrSc., RNDr. Rastislav Serbin, PhD., RNDr. Lívia Kocúrová, PhD., RNDr. Jana Šandrejová, PhD.

**Date of last modification:** 26.02.2018

U		URSE INFORM			_
University: P. J.		ity in Kosice			
Faculty: Faculty		A 1 4 1 4	<b>C1</b> · /		
Course ID: ÚCH ANCHU/03	tv/ Course na	me: Analytical (	Lnemistry		
Course type, sco Course type: Lo Recommended Per week: 3 / 1 Course method	ecture / Practice course-load (h Per study peri	ours):			
Number of credi	i <b>ts:</b> 6				
Recommended s	emester/trimes	ster of the cours	<b>e:</b> 3.		
Course level: I.					
Prerequisities: Ú	UCHV/VCHU/1	4 or ÚCHV/VCI	HU/15 or ÚCHV	/VCHU/10 or Ú	CHV/VACH/10
Conditions for c 3x test of analyti Examination	-				
Learning outcom Survey of basic p methods in resea	principles and ta		chemistry and a	pplications of an	alytical
treatment. Prepar Classification of of organic analys Methods of quan	of analytical choration of solution analytical reactions. titative analysis shods of analytic optical and sep	ns. Evaluation of tions. Qualitative s. General princip cal chemistry (ba	the results. analysis of cati oles of gravimetr sic principles, ir	ocedures - sampli ons and anions. I y. Volumetric ana istrumentaion and	Basic principles
D.Harvey: Mode	rn Analytical C	5		Publishing, New Y 2000.	York 1985.
Course language					
Course assessme Total number of		ts: 634			
A	В	С	D	E	FX
18.61	18.77	25.08	24.29	9.31	3.94
Provides: doc. R	NDr. Taťána Go	ondová, CSc.			
Date of last mod	ification: 26.02	2.2018			
<b>Approved:</b> Guar PhD.	anteeprof. RND	r. Pavol Mártonf	i, PhD.Guarante	eprof. RNDr. Vla	dimír Zeleňák,

University: P. J. Šafárik University in Košice
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Faculty: Faculty of Science

**Course ID:** ÚCHV/ **Course name:** Separation Methods ASM/03

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours):

Per week: 2 / 1 Per study period: 28 / 14

Course method: present

Number of credits: 5

Recommended semester/trimester of the course: 6.

Course level: I.

**Prerequisities:** (ÚCHV/ANCHU/03 or ÚCHV/ANCHE/09 or ÚCHV/ANCH1b/03) and (ÚCHV/ PAEC/03 or ÚCHV/PANCH/06 or ÚCHV/PANCHE/09 or ÚCHV/PACU/03)

**Conditions for course completion:** 

Examination

#### Learning outcomes:

Survey of basic principles, theoretical background and applications of separation methods in research and analytical practice.

#### Brief outline of the course:

Basic principles, classification, theory and applications of separation methods. Extraction - LLE, SPE, SPME. Chromatographic methods - theory, classification. Gas chromatography, retention mechanisms, stationary phases and their selection. Instrumentation, detectors in GC. Data evaluation - qualitative and quantitative analysis. High-performance liquid chromatography, principles, classification. Stationary and mobile phases in LC, instrumentation. Applications. Comparison of GC and HPLC methods.

Planar chromatographic methods - TLC, HPTLC, PC.

Electrophoretic techniques - CE, ITP, HPCE. MEKC - micellar electrokinetic capillary chromatography. Lab-on-a-Chip (LOC), TAS, electrophoresis on a chip, principles and applications.

#### **Recommended literature:**

Krupčík, J.: Separačné metódy, SVŠT CHTF, Bratislava 1983.

Skoog D. A., Leary J. J.: Principles of instrumental analysis. Saunders College Publishing, New York 1997.

Pawliszyn J., Lord H. L.: Handbook of sample preparation, Wiley 2010.

Churáček J., Jandera P.: Úvod do vysokoúčinné kapalinové chromatografie, SNTL, Praha 1984.

### **Course language:**

#### Course assessment

Total number of assessed students: 439

А	В	С	D	Е	FX
27.11	25.06	26.2	12.98	6.15	2.51

Provides: doc. RNDr. Taťána Gondová, CSc.

Date of last modification: 26.02.2018

COURSE INFORMATION L	EIIEK		
University: P. J. Šafárik University in Košice			
Faculty: Faculty of Science			
Course ID: ÚCHV/ Course name: Bioinorganic Chemistr BAC1/04	y I		
Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14 Course method: present			
Number of credits: 5			
Recommended semester/trimester of the course: 5.			
Course level: I., II.			
Prerequisities:			
<b>Conditions for course completion:</b> Test or seminar works examination			
Learning outcomes: The basic knowledges about biometal interactions with bion biocatalysis, metals in biology and medicine, metal-based dr metals in the environment.		,	· ·
<b>Brief outline of the course:</b> Metalic and non-metalic elements and their roles in biologic elements, essential trace elements). Biocoordination con Oxygen carriers and oxygen transport proteins. Photochem processes. Calcium biominerals and biomineralization. Toxic bioinorganic chemistry in pharmacy, chemotherapy (e.g. pl radiodiagnostics, mineral biotechnology, ecology and in othe	npounds, bic ical process. ( metals. Appl atinum compl	bligands. Biocatalyz Catalysis and regula lication of knowledg lexes in cancer thera	zers. tion ge of
Recommended literature: 1. Shriver D. F., Atkins P. W., Overton T. L., Rourke J.P., We Atkins. Inorganic Chemistry. Oxford University Press, Oxfo 2. Kaim W., Schwederski B.: Bioinorganic Chemistry: Inorg Life. Wiley, Chichester 1998. 3. Wilkins P. C., Wilkins R. G.: Inorganic Chemistry in Biole	rd 2006. anic Elements	s in the Chemistry of	
Course language:			
Course assessment Total number of assessed students: 243			
A B C D	I	E FX	
41.98 27.98 18.11 4.53	7	0.41	
Provides: doc. RNDr. Zuzana Vargová, Ph.D.	I	1	
Date of last modification: 26.02.2018			

University: P. J. Šafárik University in Košice         Faculty: Faculty of Science         Course ID: ÚCHV/       Course name: Fundamentals of Bioanalytical Chemistry         BACHZ/06       Course type, scope and the method:         Course type, scope and the method:       Course type: Lecture / Practice         Recommended course-load (hours):       Per week: 2 / 1 Per study period: 28 / 14         Course method: present       Number of credits: 5         Recommended semester/trimester of the course: 3.       Course level: I.         Prerequisities:       Conditions for course completion:         written test       Oral examination         Learning outcomes:       Principles and theoretical foundations the application of analytical methods in bioanalysis.         Brief outline of the course:       Introduction to Bioanalytical Chemistry. Biological samples classification. Factors that affer analytes in biological samples. Collection, transport and storage of samples, the main principl of sampling, the suppressing of undesirable phenomena. Selected methods of pretreatment biological samples. Analyzers, equipment and organization of work in a clinical laboratory. Contr and management of quality in clinical laboratory. Quality manual, calibration, control, and referem materials. Validation and Good Laboratory Practice. Buffers in bioanalysis. Enzymes in bioanalysis introduction, distribution, Mechanism of enzyme catalysis. The kinetics of enzymatic reactions with theoretical soft enzymatic reactions with theoretical soft enzymatic reactions with tester theoretical soft enzymatic reactions with tester theoretical s
Course ID: ÚCHV/ BACHZ/06       Course name: Fundamentals of Bioanalytical Chemistry BACHZ/06         Course type, scope and the method: Course type; Lecture / Practice Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14 Course method: present         Number of credits: 5         Recommended semester/trimester of the course: 3.         Course level: I.         Prerequisities:         Conditions for course completion: written test Oral examination         Learning outcomes: Principles and theoretical foundations the application of analytical methods in bioanalysis.         Brief outline of the course: Introduction to Bioanalytical Chemistry. Biological samples classification. Factors that affe analytes in biological samples. Collection, transport and storage of samples, the main principl of sampling, the suppressing of undesirable phenomena. Selected methods of pretreatment biological samples. Analyzers, equipment and organization of work in a clinical laboratory. Contr and management of quality in clinical laboratory. Quality manual, calibration, control, and referen materials. Validation and Good Laboratory Practice. Buffers in bioanalysis. Enzymes in bioanalysis
BACHZ/06       Course type, scope and the method:         Course type; Lecture / Practice       Recommended course-load (hours):         Per week: 2 / 1 Per study period: 28 / 14       Course method: present         Number of credits: 5       Recommended semester/trimester of the course: 3.         Course level: I.       Prerequisities:         Conditions for course completion:       written test         Oral examination       Learning outcomes:         Principles and theoretical foundations the application of analytical methods in bioanalysis.       Brief outline of the course:         Introduction to Bioanalytical Chemistry. Biological samples classification. Factors that affed analytes in biological samples. Collection, transport and storage of samples, the main principl of sampling, the suppressing of undesirable phenomena. Selected methods of pretreatment biological samples. Analyzers, equipment and organization of work in a clinical laboratory. Contra and management of quality in clinical laboratory. Quality manual, calibration, control, and referen materials. Validation and Good Laboratory Practice. Buffers in bioanalysis. Enzymes in bioanalysis
Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14 Course method: present Number of credits: 5 Recommended semester/trimester of the course: 3. Course level: 1. Prerequisities: Conditions for course completion: written test Oral examination Learning outcomes: Principles and theoretical foundations the application of analytical methods in bioanalysis. Brief outline of the course: Introduction to Bioanalytical Chemistry. Biological samples classification. Factors that affe analytes in biological samples. Collection, transport and storage of samples, the main principl of sampling, the suppressing of undesirable phenomena. Selected methods of pretreatment biological samples. Analyzers, equipment and organization of work in a clinical laboratory. Contr and management of quality in clinical laboratory. Quality manual, calibration, control, and referen materials. Validation and Good Laboratory Practice. Buffers in bioanalysis. Enzymes in bioanalysis
Recommended semester/trimester of the course: 3.         Course level: I.         Prerequisities:         Conditions for course completion:         written test         Oral examination         Learning outcomes:         Principles and theoretical foundations the application of analytical methods in bioanalysis.         Brief outline of the course:         Introduction to Bioanalytical Chemistry. Biological samples classification. Factors that affed analytes in biological samples. Collection, transport and storage of samples, the main principl of sampling, the suppressing of undesirable phenomena. Selected methods of pretreatment biological samples. Analyzers, equipment and organization of work in a clinical laboratory. Contra and management of quality in clinical laboratory. Quality manual, calibration, control, and referen materials. Validation and Good Laboratory Practice. Buffers in bioanalysis. Enzymes in bioanalysis
Course level: I.         Prerequisities:         Conditions for course completion:         written test         Oral examination         Learning outcomes:         Principles and theoretical foundations the application of analytical methods in bioanalysis.         Brief outline of the course:         Introduction to Bioanalytical Chemistry. Biological samples classification. Factors that affee analytes in biological samples. Collection, transport and storage of samples, the main principl of sampling, the suppressing of undesirable phenomena. Selected methods of pretreatment biological samples. Analyzers, equipment and organization of work in a clinical laboratory. Contra and management of quality in clinical laboratory. Quality manual, calibration, control, and referen materials. Validation and Good Laboratory Practice. Buffers in bioanalysis. Enzymes in bioanalysis
Prerequisities:         Conditions for course completion:         written test         Oral examination         Learning outcomes:         Principles and theoretical foundations the application of analytical methods in bioanalysis.         Brief outline of the course:         Introduction to Bioanalytical Chemistry. Biological samples classification. Factors that affer analytes in biological samples. Collection, transport and storage of samples, the main principl of sampling, the suppressing of undesirable phenomena. Selected methods of pretreatment biological samples. Analyzers, equipment and organization of work in a clinical laboratory. Contra and management of quality in clinical laboratory. Quality manual, calibration, control, and referen materials. Validation and Good Laboratory Practice. Buffers in bioanalysis. Enzymes in bioanalysis
Conditions for course completion: written test Oral examination Learning outcomes: Principles and theoretical foundations the application of analytical methods in bioanalysis. Brief outline of the course: Introduction to Bioanalytical Chemistry. Biological samples classification. Factors that affe analytes in biological samples. Collection, transport and storage of samples, the main principl of sampling, the suppressing of undesirable phenomena. Selected methods of pretreatment biological samples. Analyzers, equipment and organization of work in a clinical laboratory. Contr and management of quality in clinical laboratory. Quality manual, calibration, control, and referen materials. Validation and Good Laboratory Practice. Buffers in bioanalysis. Enzymes in bioanalys
written test Oral examination Learning outcomes: Principles and theoretical foundations the application of analytical methods in bioanalysis. Brief outline of the course: Introduction to Bioanalytical Chemistry. Biological samples classification. Factors that affe analytes in biological samples. Collection, transport and storage of samples, the main principl of sampling, the suppressing of undesirable phenomena. Selected methods of pretreatment biological samples. Analyzers, equipment and organization of work in a clinical laboratory. Contr and management of quality in clinical laboratory. Quality manual, calibration, control, and referen materials. Validation and Good Laboratory Practice. Buffers in bioanalysis. Enzymes in bioanalys
<b>Brief outline of the course:</b> Introduction to Bioanalytical Chemistry. Biological samples classification. Factors that affer analytes in biological samples. Collection, transport and storage of samples, the main principl of sampling, the suppressing of undesirable phenomena. Selected methods of pretreatment biological samples. Analyzers, equipment and organization of work in a clinical laboratory. Contr and management of quality in clinical laboratory. Quality manual, calibration, control, and referen materials. Validation and Good Laboratory Practice. Buffers in bioanalysis. Enzymes in bioanalys
Introduction to Bioanalytical Chemistry. Biological samples classification. Factors that affe analytes in biological samples. Collection, transport and storage of samples, the main principl of sampling, the suppressing of undesirable phenomena. Selected methods of pretreatment biological samples. Analyzers, equipment and organization of work in a clinical laboratory. Contr and management of quality in clinical laboratory. Quality manual, calibration, control, and referen materials. Validation and Good Laboratory Practice. Buffers in bioanalysis. Enzymes in bioanalys
one substrate, the Michaelis constant, constant specificity, lag phase, kinetics of reactions with tw substrates. Moderators of enzyme activity. Selected methods for analysis of biomolecules.
Recommended literature: 1.Mikkelsen S.R, Cortón E.: Bioanalytical Chemistry, Wiley, 2004 2.Wilson I., Bioanalytical Separations 4, (Handbook of Analytical Separations), Elsevier, 2003 3.Lee, D.C., Webb, M. Pharmaceutical Analysis, Blackwell, 2003
Course language:
Course assessment Total number of assessed students: 75
A B C D E FX
30.67 33.33 30.67 4.0 0.0 1.33
Provides: doc. RNDr. Katarína Reiffová, PhD.
Date of last modification: 26.02.2018

	CO	URSE INFORM	MATION LETT	ER	
University: P. J.	Šafárik Univers	ity in Košice			
Faculty: Faculty	of Science				
Course ID: ÚCH BCHU/03	IV/ Course na	me: Biochemist	ry		
Course type, sco Course type: Le Recommended Per week: 3 Per Course method	ecture course-load (h r study period:	ours):			
Number of credi	its: 5				
Recommended s	semester/trimes	ster of the cours	<b>e:</b> 5.		
Course level: I.					
Prerequisities: Ú	JCHV/VCHU/1	0 or ÚCHV/VCI	HU/15 or ÚCHV	/VACH/10 or ÚC	CHV/VCHU/14
<b>Conditions for c</b> test + oral exami	-	on:			
Learning outcom The aim of bioch basis of their mo	nemistry teachin	• 1	•	ield of living org	ganisms on the
Brief outline of t 1. Protein Structu 2. DNA and RNA 3. Enzymes: Bas 4. Carbohydrates 5. Lipids and Cel 6. Metabolis: Bas 7. Glycolysis and 8. The Citric Aci 9. Oxidative Pho 10. The Calvine 11. Fatty Acids M 12. DNA Replica 13. Protein Synth Recommended II	ure and Function A and the Flow tic Concepts and s (Monosacchari Ils Membranes, sic Concepts an d Gluconeogene id Cycle and Gly osphorylation, The Cycle and the P Metabolism, Ure ation, Transcript hesis & Degrada	of Genetic Inform l Kinetics, Cataly ides, Disaccharid Membrane Chan d Design, Signal sis, Glycogen M yoxylate Cycle he Light Reaction entose Phosphate ca Cycle tion (RNA Synth	mation, Exploring tic Strategies and les, Polysacchario inels and Pumps -Transduction Pa etabolism ns of Photosyntes e Pathway essis)	d Regulatory Stra des – Functions a thways sis	-
Kecommended I Škárka: Biochém Voet a Voetová: I Stryer, L.: Bioch	nia. Alfa, 1992 Biochemie. Vict			, 1988	
Course language	e:				
Course assessme					
Total number of			D	<b>F</b>	EV
A	В	С	D	E	FX

18.01

17.47

20.81

21.54

19.28

2.9

Provides: doc. RNDr. Erik Sedlák, PhD., RNDr. Nataša Tomášková, PhD.

Date of last modification: 26.02.2018

University: P. J.		ity in Kosice			
Faculty: Faculty					
<b>Course ID:</b> ÚBE BDD/05	EV/ Course na	me: Biology of	Children and Ad	olescents	
Course type, sco Course type: L Recommended Per week: 2 / 0 Course method	ecture / Practice course-load (h Per study perio	ours):			
Number of cred	its: 2				
Recommended s	semester/trimes	ter of the cours	se: 4., 6.		
Course level: I.					
Prerequisities:					
<b>Conditions for c</b> Written test	ourse completi	on:			
The aim of the su development. It children and ado <b>Brief outline of t</b> Human ontogen circulatory, resp system. Nervous population and e	is neccessary for lescents linked t the course: esis. Postnatal iratory, gastroir s system. Age s	to development. A development. A testinal and ur	Age specific feat	ological characte tures of skeletal Reproductive sys	and muscalar, tem. Endocrine
Recommended I Drobný I., Drobi 2000 Lipková V.: Som Malá H., Klemei	<b>iterature:</b> ná M.: Biológia natický a fyziolo	gický vývoj diet	2at'a. Osveta Brat	islava, 1980	ava, PdF UK,
Course language	e:				
Course assessme Total number of		ts: 1402			
A	В	С	D	Е	FX
30.53	22.97	17.9	18.12	9.91	0.57
Provides: doc. R	NDr. Monika K	assayová, CSc.			
Date of last mod	lification: 21.08	.2017			
Approved: Guar PhD.	anteeprof. RND	r. Pavol Márton	fi, PhD.Guarante	eprof. RNDr. Vla	adimír Zeleňák,

University: P. J. Šafá	rik University in Košice	
Faculty: Faculty of S	cience	
<b>Course ID:</b> ÚBEV/ BKP/14	Course name: Bachelor P	roject
Course type, scope a Course type: Recommended cou Per week: Per stud Course method: pre	rse-load (hours): ly period: esent	
Number of credits: 2		
Recommended seme	ster/trimester of the cours	se: 5.
Course level: I.		
Prerequisities:		
<b>Conditions for cours</b> Submission of the ba supervisor.	-	of the project and acceptance of its content by the
Learning outcomes:		
Brief outline of the c	ourse:	
<b>Recommended litera</b> 1. Scientific papers rector UPJS in Košic	elated to the topic of the bac	chelor project. 2. Directive No. 1/2011 of the
Course language:		
<b>Course assessment</b> Total number of asse	ssed students: 71	
	abs	n
	100.0	0.0
Provides:		
Date of last modifica	tion: 23.02.2018	
Approved: Guarantee PhD.	eprof. RNDr. Pavol Márton	fi, PhD.Guaranteeprof. RNDr. Vladimír Zeleňák,

University: P. J. Šafá	rik University in Košice	
Faculty: Faculty of S	cience	
<b>Course ID:</b> ÚCHV/ BKP/14	Course name: Bachelor I	roject
Course type, scope a Course type: Recommended cou Per week: Per stud Course method: pre	rse-load (hours): ly period: esent	
Number of credits: 2		
Recommended seme	ster/trimester of the cour	se: 5.
Course level: I.		
Prerequisities:		
supervisor.	1	of the project and acceptance of its content by the
Learning outcomes:		
Brief outline of the c	ourse:	
	<b>iture:</b> elated to the topic of the ba 11 of the rector UPJS in K	1 0
Course language:		
<b>Course assessment</b> Total number of asse	ssed students: 40	
	abs	n
	100.0	0.0
Provides:		
Date of last modifica	tion: 26.02.2018	
Approved: Guarantee PhD.	eprof. RNDr. Pavol Mártor	fi, PhD.Guaranteeprof. RNDr. Vladimír Zeleňák,

University: P. J.	. Šafárik Univers	sity in Košice			
Faculty: Faculty	y of Science				
Course ID: ÚB BO1/03	EV/ Course n	ame: Botany I			
Recommended	Lecture / Practico d course-load (h 2 Per study peri	e 10urs):			
Number of crea	lits: 5				
Recommended	semester/trime	ster of the cours	<b>e:</b> 3.		
Course level: I.					
Prerequisities:					
Conditions for	course complet	ion:			
Learning outco	<b>mes:</b> biology of lower	r plants.			
Heterocontophy Chlorophyta). Labyrinthulomy Ascomycota, Ba Literature: Deacon, J.W. (1 <b>Recommended</b> Bačkor, M.: Zál Deacon, J.W. (1 Van den Hoek, 9	vta, Haptophyta Slime moulds vcota). Fungi (C asidiomycota). I 998) Modern M literature: clady systému ni 998) Modern M C. a kol. 1995: A	, Cryptophyta, (Plasmodiophoro	Dinophyta, Eug omycota, Dicty ochytriomycota, tes. ell Science Ltd. inice, riasy a sliz ell Science Ltd. ction to phycolo	glenophyta, Chl vosteliomycota, , Chytridiomyco zovky). UPJŠ, Ka gy,	a, Rhodophyta, orarachniophyta, Acrasiomycota, ta, Zygomycota, ošice 2002;
Course languag	ge:				
Course assessm Total number of	ent f assessed studer	nts: 1656			
А	В	C	D	Е	FX
13.41	19.26	25.24	20.23	19.2	2.66
Provides: prof.	RNDr. Martin B	ačkor, DrSc., RN	Dr. Michal Gog	ga, PhD.	
Date of last mo	dification: 23.02	2.2018			
Approved: Gua	ranteeprof. RNI	Dr. Pavol Mártoní	i, PhD.Guarante	eeprof. RNDr. Vl	adimír Zeleňák,

University: P. J.	. Šafárik Univer	sity in Košice			
Faculty: Faculty	y of Science				
<b>Course ID:</b> ÚB BO1/15	EV/ Course n	ame: Botany I			
Recommended	Lecture / Practic l course-load (l 2 Per study per	e hours):			
Number of crea	lits: 4				
Recommended	semester/trime	ester of the cours	se: 3.		
Course level: I.					
Prerequisities:					
Conditions for	course complet	tion:			
Learning outco		r plants.			
Heterocontophy Chlorophyta). Labyrinthulomy Ascomycota, Ba Literature: Deacon, J.W. (1 <b>Recommended</b> Bačkor, M.: Zál Deacon, J.W. (1 Van den Hoek,	vta, Haptophyta Slime moulds vcota). Fungi (( asidiomycota). I 998) Modern M literature: clady systému n 998) Modern M C. a kol. 1995: 4	algae (Cyanophy a, Cryptophyta, s(Plasmodiophoro Domycota, Hyph Lichens. Bryophy Iycology. Blackw ižších rastlín I. (s Iycology. Blackw Algae, an introduc evolúcia nižších	Dinophyta, Eug omycota, Dicty lochytriomycota, rtes. rell Science Ltd. sinice, riasy a sliz rell Science Ltd. ction to phycolog	glenophyta, Chlo osteliomycota, Chytridiomycot zovky). UPJŠ, Ko	orarachniophyta, Acrasiomycota, ta, Zygomycota,
Course languag	ge:				
Course assessm Total number of		nts: 214			
А	В	C	D	E	FX
26.64	16.82	23.83	18.69	11.21	2.8
Provides: prof.	RNDr. Martin E	Bačkor, DrSc., RN	Dr. Michal Gog	a, PhD.	
Date of last mo	dification: 23.0	2.2018			
Approved: Gua					

Faculty: Faculty of So Course ID: ÚBEV/ BOT1/03 Course type, scope an Course type: Lecture Recommended cour Per week: 2 / 2 Per s Course method: pres Number of credits: 5 Recommended semes Course level: I. Prerequisities: ÚBEV Conditions for course Practical and theoretic Learning outcomes: To obtain of survey in Brief outline of the co History and present t cladistics and molecu plants. Gymnosperms Evolution and general and Caryophyllid clad	Course na nd the mether e / Practice rse-load (ho study perior sent ster/trimes //TCB1/03 e completion cal exam. n knowledge ourse: time of plan	ours): od: 28 / 28 ter of the cours on: e and methods in nt systematics.	n systematics of Approaches to p		
BOT1/03 Course type, scope an Course type: Lecture Recommended cour Per week: 2 / 2 Per s Course method: pres Number of credits: 5 Recommended semes Course level: I. Prerequisities: ÚBEV Conditions for course Practical and theoretic Learning outcomes: To obtain of survey in Brief outline of the co History and present t cladistics and molecu plants. Gymnosperms Evolution and general	nd the mether e / Practice se-load (ho study period sent ster/trimes //TCB1/03 e completion cal exam. h knowledge ourse: time of plan alar taxonor	hod: burs): bd: 28 / 28 ter of the cours bn: e and methods in nt systematics.	n systematics of Approaches to p		
Course type: Lecture Recommended cour Per week: 2 / 2 Per s Course method: pres Number of credits: 5 Recommended semes Course level: I. Prerequisities: ÚBEV Conditions for course Practical and theoretic Learning outcomes: To obtain of survey in Brief outline of the co History and present t cladistics and molecu plants. Gymnosperms Evolution and general	e / Practice se-load (ho study perior sent ster/trimes //TCB1/03 e completion cal exam. h knowledge ourse: time of plan alar taxonor	ours): od: 28 / 28 ter of the cours on: e and methods in nt systematics.	n systematics of Approaches to p		
Recommended semes Course level: I. Prerequisities: ÚBEV Conditions for course Practical and theoretic Learning outcomes: To obtain of survey in Brief outline of the co History and present t cladistics and molecu plants. Gymnosperms Evolution and general	ster/trimes //TCB1/03 e completion cal exam. n knowledge ourse: time of plan alar taxonor	on: e and methods in nt systematics.	n systematics of Approaches to p		n Dringinlag of
Course level: I. Prerequisities: ÚBEV Conditions for course Practical and theoretic Learning outcomes: To obtain of survey in Brief outline of the co History and present t cladistics and molecu plants. Gymnosperms Evolution and general	//TCB1/03 e completion cal exam. n knowledge ourse: time of plan alar taxonor	on: e and methods in nt systematics.	n systematics of Approaches to p		n Drinciples of
Prerequisities: ÚBEV Conditions for course Practical and theoretic Learning outcomes: To obtain of survey in Brief outline of the co History and present t cladistics and molecu plants. Gymnosperms Evolution and general	e completion cal exam. In knowledge ourse: time of plan alar taxonor	e and methods in the systematics.	Approaches to p		n Dringinlag of
Conditions for course Practical and theoretic Learning outcomes: To obtain of survey in Brief outline of the co History and present t cladistics and molecu plants. Gymnosperms Evolution and general	e completion cal exam. In knowledge ourse: time of plan alar taxonor	e and methods in the systematics.	Approaches to p		n Dringinlag of
Practical and theoretic Learning outcomes: To obtain of survey in Brief outline of the co History and present t cladistics and molecu plants. Gymnosperms Evolution and general	a knowledge ourse: time of plan	e and methods in the systematics.	Approaches to p		n Dringinlag of
To obtain of survey in Brief outline of the co History and present t cladistics and molecu plants. Gymnosperms Evolution and general	ourse: time of plan	nt systematics.	Approaches to p		n Dringinlag of
History and present t cladistics and molecu plants. Gymnosperms Evolution and general	time of plan alar taxonor	•		plant classification	n Dringinlag of
Practices are devoted of ferns and allies fr conifers. Selected fam Cyperaceae, Poaceae, Fabaceae, Rosaceae, Apiaceae, Asteraceae	description de. Rosid an to study o om Palaeoz ilies of angl Ranuncula Betulacea	evolution: cycad h. Basal clades and asterid clades of the most import zoic age. Tropid iosperms. ( <i>M ceae, Papaverac ae, Brassicacea</i>	ds, ginkgos, con nd Magnoliid cla s of tricolpates. ortant families of cal a subtropical fagnoliaceae, Ar ceae, Caryophyll ae, Boraginacea	cophytes, ferns a ifers, gnetophytes de. Monocots. "B f tracheophytes. f flora. Ferns. Pr aceae, Liliaceae, <i>A</i> aceae, Euphorbia e, Plantaginace	and allies. Seed s. Angiosperms. casal tricolpates" Fossil evidence actical study of Amaryllidaceae, ceae, Violaceae, ae, Lamiaceae,
Recommended litera Mártonfi P.: Systemat Mártonfi P.: Systemat Judd W. S., Campbell A phylogenetic Appro Dostál J., Červenka M Course language:	ika cievnat ika cievnat Ch. S., Ke bach, 2nd e	ých rastlín ES llogg E. A. & S d Sinauer Ass	5 UPJŠ, Košice, 2 tevens P. F., Don sociates, Sunderla	2003. oghue M. J.: Plar and, 2002.	nt Systematics.
<b>Course assessment</b> Total number of asses	sed student	s: 1439			
A	В	С	D	Е	FX
10.35 1	12.44	17.37	19.81	24.67	15.36

Date of last modification: 23.02.2018

			MATION LET I		
University: P. J.	Šafárik Univers	ity in Košice			
Faculty: Faculty	of Science				
Course ID: ÚB BOT1/15	EV/ Course na	me: Botany II			
Course type: I Recommended	ope and the met Lecture / Practice l course-load (h 2 Per study perio d: present	ours):			
Number of cred					
Recommended	semester/trimes	ter of the cours	<b>e:</b> 2.		
Course level: I.					
Prerequisities:	ÚBEV/TCB1/03				
<b>Conditions for</b> Practical and the	<b>course completi</b> eoretical exam.	on:			
<b>Learning outco</b> To obtain of sur	mes: vey in knowledg	e and methods in	n systematics of	tracheophytes.	
cladistics and m plants. Gymnos Evolution and g and Caryophylli Practices are de of ferns and all conifers. Selecte Cyperaceae, Pos Fabaceae, Ros	nolecular taxono perms and their eneral description id clade. Rosid at evoted to study of lies from Palaeo ed families of ang aceae, Ranuncula aceae, Betulace	my. Tracheophy evolution: cycac n. Basal clades and asterid clades of the most impo- zoic age. Tropic iosperms. ( <i>Maceae, Papaverac ae, Brassicacea</i>	tes, clades of ly ls, ginkgos, coni nd Magnoliid clau of tricolpates. ortant families o cal a subtropical lagnoliaceae, Ara eae, Caryophylla ae, Boraginacea	plant classificatio cophytes, ferns a fers, gnetophytes de. Monocots. "B f tracheophytes. flora. Ferns. Pra aceae, Liliaceae, A aceae, Euphorbiad e, Plantaginacea tification accordin	and allies. Seed s. Angiosperms. asal tricolpates" Fossil evidence actical study of Amaryllidaceae, ceae, Violaceae, ae, Lamiaceae,
Mártonfi P.: Sys Judd W. S., Can A phylogenetic	stematika cievnat stematika cievnat npbell Ch. S., Ke Approach, 2nd e nka M.: Veľký k	ých rastlín ES llogg E. A. & St d Sinauer Ass	UPJŠ, Košice, 2 evens P. F., Done ociates, Sunderla	oghue M. J.: Plan	-
Course assessm					
	assessed studen	ts: 263			
А	В	С	D	E	FX
14.45	13.31	26.62	22.81	14.07	8.75
Provides: prof	RNDr. Pavol Má	rtonfi, PhD., Mg	r. Vladislav Kola	arčik PhD	

Date of last modification: 23.02.2018

University: P. J.	Šafárik Univers	ity in Košice			
Faculty: Faculty	of Science				
<b>Course ID:</b> ÚCH BPO/14	IV/ Course na	me: Bachelor T	Thesis and its De	fence	
Course type, sco Course type: Recommended Per week: Per Course method	course-load (h study period:				
Number of credi	its: 4				
Recommended s	emester/trimes	ter of the cours	se:		
Course level: I.					
Prerequisities:					
Conditions for c	ourse completi	on:			
Learning outcom	nes:				
Brief outline of t Oral presentation the state examina	n of the thesis re	esults. Answerii	ng questions of t	the thesis oponent	or members of
Recommended l	iterature:				
<b>Course language</b> slovak	2:				
Course assessme Total number of		ts: 154			
Α	В	С	D	E	FX
87.01	9.09	1.95	1.95	0.0	0.0
Provides:			1	-	
Date of last mod	ification: 26.02	.2018			
<b>Approved:</b> Guar PhD.	anteeprof. RND	r. Pavol Márton	fi, PhD.Guarante	eeprof. RNDr. Vla	dimír Zeleňák,

University: P. J. Š	afárik Univers	ity in Košice			
Faculty: Faculty of	of Science				
<b>Course ID:</b> ÚBEV BPO/14	V/ Course na	me: Bachelor Tl	nesis and its Def	ence	
Course type, scop Course type: Recommended o Per week: Per s Course method:	course-load (h tudy period:				
Number of credit	s: 4				
Recommended se	emester/trimes	ter of the cours	e:		
Course level: I.					
Prerequisities:					
Conditions for co	urse completi	on:			
Learning outcom	es:				
Brief outline of th	ne course:				
Recommended lit	terature:				
Course language:					
Course assessmen Total number of a	-	ts: 172			
A	В	С	D	Е	FX
50.58 26.16 16.86 4.65 1.74 0.0					
Provides:				·	
Date of last modi	fication: 23.02	.2018			
<b>Approved:</b> Guara PhD.	nteeprof. RND	r. Pavol Mártonf	i, PhD.Guarantee	eprof. RNDr. Vla	dimír Zeleňák,

University: P. J.					
Faculty: Faculty					
<b>Course ID:</b> ÚBE BS1/03	V/ Course na	me: Biostatistic	S		
Course type, sco Course type: La Recommended Per week: 2 / 2 Course method	ecture / Practice course-load (h Per study perio	ours):			
Number of credi	<b>ts:</b> 6				
Recommended s	emester/trimes	ster of the cours	<b>e:</b> 3., 5.		
Course level: I.					
Prerequisities:					
<b>Conditions for c</b> Recognition. Recognition.	ourse completi	on:			
Learning outcom To provide the st and their scope o	udents with kno	owledge on basic	principles of sta	tistic methods us	ed in biology
Brief outline of t Sources and the Descriptive statis empirical distribution One-way and mut Correlations. No	oretical backgro stics: variables, utions. Experim iltiple analysis n-parametrical i	measures of me ental sampling f of variance. Test	an value and var from normal dist	riability of data. ributions. Testing omparisons. Regr	Theoretical and g of hypotheses ression analysis
Recommended I Hassard, T. H.: U Snedecor,G.W., O R.Forthofer, E.S.	Understanding b Cochran,W.G.: S	Statistical metho	ds. The Iowa stat	te university, Am	es, 1972.
Course language					
<b>Course assessme</b> Total number of		ts: 181			
A	В	С	D	Е	FX
3.31	8.84	16.57	21.55	35.36	14.36
Provides: prof R	NDr. Beňadik Š	Šmajda, CSc.			
rovides. prof. is					
Date of last mod	ification: 23.02	2.2018			

University: P. J. Šafa	árik Univers	sity in Košice			
Faculty: Faculty of S	Science				
<b>Course ID:</b> ÚBEV/ BSB/15	Course na	ame: Biology			
Course type, scope a Course type: Recommended cou Per week: Per stud Course method: pr	ırse-load (h dy period:				
Number of credits:	1				
Recommended sem	ester/trime	ster of the cours	e:		
Course level: I.					
Prerequisities:					
Conditions for cour	se completi	ion:			
Learning outcomes	:				
Brief outline of the	course:				
<b>Recommended</b> liter	ature:				
Course language:					
<b>Course assessment</b> Total number of asse	essed studer	nts: 143			
А	В	C	D	Е	FX
22.38	21.68	26.57	18.18	11.19	0.0
Provides:		·		I	
Date of last modific	ation: 23.02	2.2018			
<b>Approved:</b> Guarante PhD.	eprof. RND	Dr. Pavol Mártonf	i, PhD.Guarantee	eprof. RNDr. Vla	dimír Zeleňák,

	. Šafárik Univers				
Faculty: Facult	y of Science				
Course ID: ÚC CHV1/99	HV/ Course na	me: Chemical o	calculations		
Course type: l Recommende	d course-load (h er study period:	ours):			
Number of cree	lits: 2			_	
Recommended	semester/trimes	ster of the cours	se: 1.		
Course level: I.					
Prerequisities:					
<b>Conditions for</b> Short written te Written test.	<b>course completi</b> sts.	on:			
			•	ns with or withou l equilibrium.	t chemical
Material bilance Material bilance	the clear matter es for preparation es for combined p	, dissolving and processes. Chem	mixing of solution ical equations and	osition. Stoichion ns, and for separat d material bilance culations. The so	ing of mixtures in the system
Recommended		vo všeobecnej a	anorganickej cho	émii (skriptum), F	PF UPJŠ.
Potočňák I.: Ch Košice, 2006.					
Košice, 2006.	ge:				
Košice, 2006. Course languaş Course assessm	ient	ts: 1240			
Košice, 2006. Course languaş Course assessm		ts: 1240 C	D	E	FX
Košice, 2006. Course languaş Course assessm Total number o	ent f assessed studen		D 20.56	E 13.95	
Košice, 2006. Course languag Course assessm Total number o A 20.56	f assessed studen	C 24.44	20.56	13.95	FX
Košice, 2006. Course languag Course assessm Total number o A 20.56 Provides: RND	f assessed studen B 19.68	C 24.44 PhD., RNDr. Mi	20.56	13.95	FX

University: P. J.	Šafárik Univers	ity in Košice			
Faculty: Faculty	y of Science				
Course ID: ÚB CYT1/15	EV/ Course name: Cytology				
Recommended	Lecture / Practice l course-load (h 2 Per study peri	ours):			
Number of cred	lits: 6				
Recommended	semester/trimes	ster of the cours	e: 1.		
Course level: I.					
Prerequisities:					
<b>Conditions for</b> Practicals gradu each); Oral exam	ation (without al		itten tests gradua	ation (min. 70 % f	fruitfulness of
			principles of cel	Il microscopic and	1
plant and anim	system organiza al cells. Micros		scopic and mol	son of prokaryotic ecular structure a	
Recommended Alberts, B., Bra New York, Lond	y, D., Lewis, J. e	t al.: Molecular	Biology of the C	ell. Garland Publ	ishing Inc.,
Course languag	ge:				
Course assessm Total number of	ent assessed studen	ts: 3862			
А	В	С	D	Е	FX
5.85	15.64	24.52	23.05	26.05	4.89
Provides: RND	r. Rastislav Jend	želovský, PhD., I	RNDr. Zuzana Je	endželovská, PhD	
Date of last mo	dification: 23.02	2.2018			
<b>Approved:</b> Gua PhD.	ranteeprof. RND	r. Pavol Márton	fi, PhD.Guarante	eprof. RNDr. Vla	dimír Zeleňák,

University: P. J.	Šafárik Univers	ity in Košice			
Faculty: Faculty	of Science				
<b>Course ID:</b> KFal DF2p/03	DF/ Course na	me: History of I	Philosophy 2 (G	eneral Introductio	n)
Course type, sco Course type: Lo Recommended Per week: 2 / 1 Course method	ecture / Practice course-load (h Per study perio	ours):			
Number of credi	its: 4				
Recommended s	emester/trimes	ter of the cours	<b>e:</b> 6.		
Course level: I.,	II.				
Prerequisities:					
Conditions for c	ourse completi	on:			
Learning outcon	nes:				
Brief outline of t	the course:				
Recommended l	iterature:				
Course language	e:				
Course assessme Total number of		ts: 738			
A	В	С	D	Е	FX
60.84	13.82	12.6	8.67	3.39	0.68
Katarína Mayerov	vá, PhD., doc. N	Igr. Róbert Stojk	· ·	Peter Nezník, CSc	c., PhDr.
Date of last mod	ification: 31.08	3.2017			
Approved: Guar PhD.	anteeprof. RND	r. Pavol Mártonf	i, PhD.Guarante	eeprof. RNDr. Vla	dimír Zeleňák,

Faculty: Faculty of S	
<b>Course ID:</b> ÚMV/ DGS/15	Course name: Students` Digital Literacy
Course type, scope a Course type: Practi- Recommended cou Per week: 2 Per stu Course method: practice	ce rse-load (hours): Idy period: 28
Number of credits: 2	2
Recommended seme	ester/trimester of the course: 1.
Course level: I.	
Prerequisities:	
Conditions for cours continuous assessme	-
Learning outcomes: To acquire an overvie competencies with en To acquire basic digi	ew of the current possibilities of digital technology to develop skills and nphasis on the area of communication, social interaction and personal. tal skills for working with advanced technologies (mobile phone, tablet,
Learning outcomes: To acquire an overvie competencies with en To acquire basic digi laptop, social media, technologies for bette	ew of the current possibilities of digital technology to develop skills and mphasis on the area of communication, social interaction and personal.
Learning outcomes: To acquire an overvie competencies with en To acquire basic digi laptop, social media, technologies for bette lifelong learning and Brief outline of the of Introduction to the pr online information so books). Tools for co and visualization. To Google Drive, Youtu collaborative activiti	ew of the current possibilities of digital technology to develop skills and nphasis on the area of communication, social interaction and personal. tal skills for working with advanced technologies (mobile phone, tablet, online webtechnologies). To understand the value of existing advanced er and more effective learning, work and active life in higher education, further career prospects.
Learning outcomes: To acquire an overvie competencies with en To acquire basic digi laptop, social media, technologies for bette lifelong learning and <b>Brief outline of the o</b> Introduction to the prion online information so books). Tools for co and visualization. The Google Drive, Youtu collaborative activitie evaluation of digital <b>Recommended litera</b> 1. Bruff, D. (2009). The environments. San Fri 2. Byrne, R. (2012).	ew of the current possibilities of digital technology to develop skills and nphasis on the area of communication, social interaction and personal. tal skills for working with advanced technologies (mobile phone, tablet, online webtechnologies). To understand the value of existing advanced er and more effective learning, work and active life in higher education, further career prospects. <b>course:</b> roblems of current, commonly available digital technology. Tools for access to purce (mobile applications for access to information systems, databases, data llecting, generating direct information and data and its subsequent analysis ools for providing and sharing of electronic content (cloud technology be, Google+, Skydrive, Dropbox). Tools for communication, discussion and es. Legal work with digital technologies and resources, plagiarism, critical resources. Security, privacy, digital ethics and etiquette, digital citizenship. <b>ature:</b> Feaching with classroom response systems: Creating active learning rancisco: Jossey-Bass. Google Drive and Docs for Teachers. Free Tech for Teachers. 2). What the Plus! Google+ for the Rest of Us. Amazon igital Services. ell Phones in the Classroom: A Practical Guide for Educators. International

**Course assessment** Total number of assessed students: 147

96.6 3.4	abs	n
	96.6	3.4

**Provides:** doc. RNDr. Stanislav Lukáč, PhD., doc. RNDr. Jozef Hanč, PhD., doc. RNDr. Ľubomír Šnajder, PhD.

Date of last modification: 23.08.2017

University: P. J. Šafári	k University in Košice
Faculty: Faculty of Sci	
Course ID: ÚINF/ C EDS/15	Course name: Educational software
Course type, scope an Course type: Lecture Recommended cours Per week: 0 / 2 Per st Course method: pres	/ Practice ee-load (hours): tudy period: 0 / 28
Number of credits: 2	
Recommended semest	ter/trimester of the course: 5.
Course level: I.	
Prerequisities:	
<ul> <li>b) Multimedia education</li> <li>c) Interactive education</li> <li>d) Methodological guid chosen school subject.</li> <li>2 Creation and present</li> </ul> Learning outcomes: <ol> <li>To acquire an overvit</li> <li>To gain or enhance being an overvit</li> <li>presentation software concept maps,</li> <li>b) programs for creation</li> <li>c) simulation and moded</li> <li>d) selected subject-oried</li> </ol>	n assignments: nt (with custom graphics) onal presentation (with pictures, animations and sounds) nal quiz (with several types of quiz items) dance on the use of interactive applications in teaching selected topic of ation of final project on the use of educational software in education. iew of the educational software types and its exploitation in education. basic skills in working with: re, programs for creation and editing images, animations, diagrams, sounds, on of quizes, questionnaires, voting,
Brief outline of the co Educational software t for creation of teaching	ypes. Onlilne educational sources and tools. Multimedia processing. Tools
Košice : Ústav informá 2. Moderná didaktická [et al.] ; recenzenti Vili 9788080861353 (brož. 3. Web, Multimédiá / M	sť učiteľa : učebný materiál- modul 1 / Rastislav Adámek [et al.] ácií a prognóz školstva, 2009 80 s ISBN 9788080861193(brož.). technika v práci učiteľa : učebný materiál modul 2 / Rastislav Adámek iam Fedák, Anton Lavrin Košice : Elfa, 2010 200 s ISBN
Course language:	
Notes:	

Content of lessons will be flexibly adapted to the field of study of learners. Language learners will be able to work more with pictures and sounds, physicists with simulation programs, mathematicians with mathematical software, etc.

Course assessn							
Total number o	f assessed studen	ts: 30					
А	В	С	D	E	FX		
63.33 20.0 13.33 0.0 3.33 0.0							
Provides: doc.	RNDr. Ľubomír S	Šnajder, PhD.			•		
Date of last mo	dification: 23.08	3.2017					
<b>Approved:</b> Gua PhD.	aranteeprof. RND	r. Pavol Mártonf	i, PhD.Guarante	eprof. RNDr. Vla	dimír Zeleňák,		

University:	ΡJ	Šafárik	University	in Košice
omversiey.	1.0.	Suluin	Oniversity	

Faculty: Faculty of Science

**Course ID:** ÚCHV/ **Course name:** Physical Chemistry II FCH1b/10

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours):

Per week: 3 / 2 Per study period: 42 / 28

Course method: present

Number of credits: 6

**Recommended semester/trimester of the course:** 

Course level: I.

Prerequisities: ÚCHV/FCH1a/03 or ÚCHV/FCHU/10

#### **Conditions for course completion:**

Two partial tests from computational seminars in 6th and 12th week of semester. Examination.

#### Learning outcomes:

Understandable explain to students the principles of chemical kinetics of processes, to elucidate the kinetics and mechanism of some reactions. To analyse particularly the equilibrium and kinetics of electrode processes.

#### Brief outline of the course:

Electrochemistry. Equilibrium homogeneous processesn electrolyte solutions. Charge transfer in electrolyte solutions. Nonequilibrium homogeneous processes. Transport processes in electrolyte solutions. Conductance and molar conductivity. Hindering effects. Transport numbers. Equilibrium in heterogeneous electrochemical systems. Pocesses on charged interfaces. Electrochemical cells and fuel cells. Classification of electrode types. Concentration cells. Electrolysis. Electrochemical power sources. Potentiometry. Electrical double layer. Surface tension.

Chemical kinetics. Homogeneous processes. Reaction rate. Reaction order. Classification of chemical reactions. Elementary chemical reactions. Mechanism and kinetics equations of complicated chemical processes. Methods of rate low determination. Theory of chemical kinetics. Ttemperature dependence of reaction rates. Collision theory. Activated complex theory. Chain reactions. Structure and rate lows of chain reactions. Explosion. Polymerisation reactions. Photochemical reactions. Catalysis. Theory of homogeneous catalysis. Chemical oscillation reactions. Heterogeneous processes. Difusion. Physical and chemical adsorption. Adsorption and diffusion. Processes in heterogeneous electrochemical systems. Electrode kinetics, activation and diffusive mechanism of charge transfer.

Application of theoretical relationships on the solving of concrete problems and on the calculation of examples during seminars.

#### **Recommended literature:**

T. Engel, P. Reid : Physical Chemistry, Pearson Educat. Inc., San Francisco 2006 P.W. Atkins : Physical Chemistry,Oxford University Presss, Oxford 1986, 1990, 1994, 1998 W.J. Moore : Physical Chemistry,Longman, London 1972 and newer editions

#### **Course language:**

Course assessm Total number of	nent f assessed studen	ts: 530			
А	В	С	D	Е	FX
15.66	18.68	22.26	18.87	20.57	3.96
1	RNDr. Renáta O ka Gorejová, Mg	, , ,		ancová, RNDr. (	Indrej Petruš,
Date of last mo	dification: 26.02	2.2018			

Chiver Sity • 1. J.	Šafárik Univers	ity in Košice			
Faculty: Faculty	y of Science				
<b>Course ID:</b> ÚC FCHU/10	HV/ Course na	ame: Physical Ch	emistry		
Course type: I Recommended	ope and the met Lecture / Practice d course-load (h 2 Per study peri d: present	e ours):			
Number of crea	lits: 6				
Recommended	semester/trimes	ster of the course	e: 4.		
Course level: I.					
Prerequisities:	ÚCHV/VCHU/1	4 or ÚCHV/VCH	IU/10 or ÚCHV	//VACH/10 or ÚC	CHV/VCHU/15
	<b>course completi</b> s from computat				
<b>Learning outco</b> To provide the s		ic knowledge of	physical chemis	stry.	
equilibria and electrolytes. El	concepts of the diagrams, laws ectrochemistry:	for ideal gas ar	nd reals gases, trodics. Electro	y, chemical equi liquids, solution odes and electro	ns, solutions o
	d: Physical Chen ysical Chemistry	nistry, Pearson Ec y, Oxford Univers y, Longman, Lond	ity Presss, Oxfo	ord 1986, 1990, 19	996
	J				
W.J. Moore: Ph					
W.J. Moore: Ph Course languag Course assessm	ge:	.ts: 256			
W.J. Moore: Ph Course languag Course assessm	ge:	ts: 256 C	D	Е	FX
W.J. Moore: Ph Course languag Course assessm Total number of	ge: hent f assessed studen	т — т	D 17.19	E 13.28	FX 3.91
W.J. Moore: Ph Course languag Course assessm Total number of A 31.64 Provides: prof.	ge: f assessed studen B 19.53 RNDr. Renáta O	C 14.45	17.19		3.91
W.J. Moore: Ph Course languag Course assessm Total number of A 31.64 Provides: prof. Ján Macko, PhD	ge: f assessed studen B 19.53 RNDr. Renáta O	C 14.45 riňaková, DrSc., J Šišoláková, PhD.	17.19	13.28	3.91

University: P. J.	Šafárik Univers	sity in Košice						
Faculty: Faculty	y of Science							
<b>Course ID:</b> ÚB FG1/03	D: ÚBEV/ Course name: Phytogeography							
Recommended	Lecture / Practice l course-load (h Per study peri	e ours):						
Number of cred	lits: 5							
Recommended	semester/trimes	ster of the cours	se:					
Course level: I.,	, II.							
Prerequisities:								
<b>Conditions for</b> Written work. Exam.	course completi	ion:						
<b>Learning outco</b> To obtain theore		al knowledge fr	om phytogeogra	phy.				
ages. Postglacia geography: from Geographical on Practices: Field	l evolution of S n tropical rainf rigin of cultivate	lovak vegetation orests to tundra d plants. ng of maps. Ph	n. Regional phy s. Changes of e	esis since paleozo togeography of E earth vegetation l division of Slo	arth. Vegetation and their study			
<b>Recommended</b> Hendrych R.: Fy Brown J. H., Lo	ytogeografie S			es, Sunderland, 1	998.			
Course languag	ge:							
Course assessm Total number of	ent fassessed studen	nts: 349						
А	В	С	D	E	FX			
38.97	22.35	21.49	8.02	8.31	0.86			
Provides: prof.	RNDr. Pavol Má	artonfi, PhD., Mg	gr. Vladislav Ko	larčik, PhD.				
Date of last mo	dification: 23.02	2.2018						
Approved: Gua	ranteeprof. RNE	Pr. Pavol Márton	fi PhD Guarante	eenrof RNDr VI	adimír Zeleňák			

		ity in Košice			
Faculty: Faculty	of Science				
<b>Course ID:</b> ÚFV FPCh/08	Course na	me: Physics for	Chemists		
Course type, sco Course type: Le Recommended Per week: 2 / 2 Course method	cture / Practice course-load (he Per study perio	ours):			
Number of credi	<b>ts:</b> 6				
Recommended s	emester/trimes	ter of the cours	<b>e:</b> 1.		
Course level: I.					
Prerequisities:					
<b>Conditions for co</b> Test-papers (2). Exam.	ourse completi	on:			
Learning outcon Completing the c understand their	ourse students v	•	ge of fundamenta	al physical laws a	and will
Brief outline of t Kinematics and d The kinetic theor liquids. Mechani current. Magnetic	ynamics of mas y of gases and cal properties c	the foundations	of thermodynan	nics. Structure ar	nd properties of
Recommended li 1. V. Hajko, J. Da 2. Š. Veis, J. Mad Bratislava, 1978. 3. P. Čičmanec: V 4. R.P. Feynman, Bratislava, 1985. 5. V. Hajko a kol	miel-Szabó: Zá 'ar, V. Martišov 'šeobecná fyzik R.B. Leighton,	ič: Všeobecná fy a 2, Elektrina a M. Sands: Feyn	yzika 1, Mechani magnetizmus. Al manove prednášl	ka a molekulová lfa, Bratislava, 19	980.
<b>Course language</b> Slovak language.					
<b>Course assessme</b> Total number of a		ts: 496			
A	В	С	D	Е	FX
19.56	29.23	29.03	14.11	7.86	0.2
Provides: doc. M	gr. Gregor Bán	ó, PhD., RNDr. Z	Zuzana Jurašekov	vá, PhD.	

University: P. J.	Šafárik Univers	ity in Košice			
Faculty: Faculty	y of Science				
<b>Course ID:</b> ÚB FR1/10	EV/ <b>Course n</b>	ame: Plant Physic	ology		
Recommended	Lecture / Practice I course-load (h B Per study peri	e ours):			
Number of cred	lits: 6				
Recommended	semester/trime	ster of the cours	e: 4.		
Course level: I.					
Prerequisities:	ÚBEV/VB1/01				
Conditions for Active participa	<b>course complet</b> tion on practical				
<b>Learning outco</b> Overview of all		ological processe	s in plant organ	isms.	
heterotrophy, m hormones, phot Lab practicals: Separation of as of cytokinins. O fructose. Measu Kjeldahl metho	etabolism of mac oreceptors, dorm Measurements ssimilation pigm Qualitative and our urements of resp d. Qualitative an ocyanins at diff	eronutrients, secon ancy, germination of water potent ents by TLC. Qua quantitative analy iration by selection alyses of protein	ndary metabolism n, flowering, pla ial, Quantitative antitative analys yses of sugars. I ive electrode. M as. Activity of s	rt, respiration, lip m, growth and dev ant movements, s e analyses of nu- es of chlorophyll HPLC separation feasurement of to ome enzymes in ca level by dist	velopment, plant tress physiology atrients in dust. a and b. Biotest of glucose and otal nitrogen by potato and pea.
<b>Recommended</b> Hopkins W.G. H		troduction to plan	t physiology. 31	rd ed., Wiley, Nev	w York 2004
Course languag	ge:				
Course assessm Total number of	ent f assessed studen	ts: 1641			
А	В	С	D	Е	FX
14.69	12.8	15.42	13.77	23.71	19.62
Provides: Mgr.	Silvia Gajdošova	á, Ph.D., doc. RN	Dr. Peter Pal'ov	e-Balang, PhD.	
Date of last mo	dification: 23.02	2.2018			
<b>Approved:</b> Gua PhD.	ranteeprof. RND	r. Pavol Mártonf	i, PhD.Guarante	eprof. RNDr. Vla	adimír Zeleňák,

University: P. J					
Faculty: Facult	y of Science				
<b>Course ID:</b> ÚB FZ1/10	EV/ Course na	<b>me:</b> Animal Ph	ysiology		
• 1 /	ope and the met				
	Lecture / Practice				
	d course-load (h 3 Per study perio				
Course metho	• 1				
Number of cree	dits: 7				
Recommended	semester/trimes	ster of the cours	se: 6.		
Course level: I.					
Prerequisities:	ÚBEV/HIS1/03	or ÚBEV/HISE	l/04 or ÚBEV/HI	IS1/15 or ÚBEV	/HISE1/15
	<b>course completi</b> from practicals ar		ion		
Learning outco To provide stud and man. Brief outline of	lents with basic k	nowledge about	physiological pro	ocesses in organi	sms of animals
To provide stud and man. <b>Brief outline of</b> The physiolog circulatory phy metabolism an Physiology of th neurophysiolog CNS. Association	the course: y of blood and siology. Physiology of he endocrine secret y. Functions of n tive functions of of tion and active m	d hemopoietic ogy of the gastro of nutrition. Wa etion. Physiolog neurons and neur CNS. Functions	physiological pro- organs. Physiologintestinal tract. Tater and minerary of reproduction. conal networks. So of the vegetative rsiology. Sensory	ogy of respiration The functions of I household of Physiology of ex Sensory and moto Physiology and moto	ion. Heart an liver. Energeti the organism ccretion.Genera pric functions c
To provide stud and man. <b>Brief outline of</b> The physiolog circulatory phy metabolism an Physiology of the neurophysiolog CNS. Association muscle contract <b>Recommended</b> Ganong, W. F.: Varder, A. J., Sta 1990 Schmidt, R. F., R.W.Hill, R.Wy	the course: y of blood and siology. Physiolo ad physiology of the endocrine secre- ty. Functions of n twe functions of of tion and active m literature: Review of medic herman, J. H., Lu Thews, G.: Hum yse, M.Anderson	d hemopoietic ogy of the gastro of nutrition. Wa etion. Physiology neurons and neur CNS. Functions otion. Work phy cal physiology. Functions, D. S.: The an Physiology, S	organs. Physiolo pintestinal tract. T ater and minera y of reproduction. conal networks. S of the vegetative	ogy of respirations of The functions of I household of Physiology of ex Bensory and moto e nervous system physiology pleton & Langer, body functions,	ion. Heart an liver. Energeti the organism ccretion.Genera oric functions c n. Physiology c 1993
To provide stud and man. <b>Brief outline of</b> The physiolog circulatory phy metabolism an Physiology of the neurophysiolog CNS. Association muscle contract <b>Recommended</b> Ganong, W. F.: Varder, A. J., Sta 1990 Schmidt, R. F., R.W.Hill, R.Wy <b>Course languag</b>	the course: y of blood and siology. Physiolo d physiology on e endocrine secre- ty. Functions of n twe functions of of tion and active m literature: Review of medic herman, J. H., Lu Thews, G.: Hum yse, M.Anderson ge:	d hemopoietic ogy of the gastro of nutrition. Wa etion. Physiology neurons and neur CNS. Functions otion. Work phy cal physiology. Functions, D. S.: The an Physiology, S	organs. Physiolo pintestinal tract. Tater and minera y of reproduction. conal networks. S of the vegetative siology. Sensory Prentice-Hall, App e mechanisms of Springer-Verlag, 1	ogy of respirations of The functions of I household of Physiology of ex Bensory and moto e nervous system physiology pleton & Langer, body functions,	ion. Heart an liver. Energeti the organism ccretion.Genera oric functions c n. Physiology c 1993
To provide stud and man. <b>Brief outline of</b> The physiolog circulatory phy metabolism an Physiology of the neurophysiolog CNS. Association muscle contract <b>Recommended</b> Ganong, W. F.: Varder, A. J., Station 1990 Schmidt, R. F., R.W.Hill, R.Wy <b>Course languag</b>	the course: y of blood and siology. Physiolo d physiology on e endocrine secre- ty. Functions of n twe functions of of tion and active m literature: Review of medic herman, J. H., Lu Thews, G.: Hum yse, M.Anderson ge:	d hemopoietic ogy of the gastro of nutrition. Wa etion. Physiolog neurons and neur CNS. Functions otion. Work phy cal physiology. F an Physiology, S : Animal Physio	organs. Physiolo pintestinal tract. Tater and minera y of reproduction. conal networks. S of the vegetative siology. Sensory Prentice-Hall, App e mechanisms of Springer-Verlag, 1	ogy of respirations of The functions of I household of Physiology of ex Bensory and moto e nervous system physiology pleton & Langer, body functions,	ion. Heart an liver. Energeti the organism ccretion.Genera oric functions c n. Physiology c 1993
To provide stud and man. <b>Brief outline of</b> The physiolog circulatory phy metabolism an Physiology of the neurophysiolog CNS. Association muscle contract <b>Recommended</b> Ganong, W. F.: Varder, A. J., Station 1990 Schmidt, R. F., R.W.Hill, R.Wy <b>Course languag</b>	the course: y of blood and siology. Physiology of the endocrine secre- ty. Functions of main two functions of main tion and active main literature: Review of medic herman, J. H., Lu Thews, G.: Hum yse, M.Anderson ge: hent	d hemopoietic ogy of the gastro of nutrition. Wa etion. Physiolog neurons and neur CNS. Functions otion. Work phy cal physiology. F an Physiology, S : Animal Physio	organs. Physiolo pintestinal tract. Tater and minera y of reproduction. conal networks. S of the vegetative siology. Sensory Prentice-Hall, App e mechanisms of Springer-Verlag, 1	ogy of respirations of The functions of I household of Physiology of ex Bensory and moto e nervous system physiology pleton & Langer, body functions,	ion. Heart an liver. Energeti the organism ccretion.Genera oric functions c n. Physiology c 1993

**Date of last modification:** 23.02.2018

University: P. J. Š	Šafárik Univers	ity in Košice			
Faculty: Faculty	of Science				
<b>Course ID:</b> ÚBE GE1/10	V/ Course na	me: Genetics			
Course type, scop Course type: Le Recommended Per week: 3 / 3 ] Course method:	cture / Practice course-load (h Per study peri	ours):			
Number of credit	ts: 7				
Recommended se	emester/trimes	ster of the cours	<b>e:</b> 5.		
Course level: I.					
Prerequisities: Ú	BEV/MB1/01	or ÚBEV/MOB1	/03 or ÚBEV/M	OB1/15	
Conditions for co	ourse completi	on:			
Learning outcom	ies:				
Brief outline of t	he course:				
Recommended li	terature:				
Course language	:				
<b>Course assessme</b> Total number of a	-	ts: 1275			
A	В	С	D	Е	FX
18.75	16.08	15.84	14.12	18.75	16.47
Provides: prof. R	NDr. Eva Čellá	rová, DrSc., RN	Dr. Katarína Bru	ňáková, PhD.	
Date of last modi	fication: 23.02	2.2018			
<b>Approved:</b> Guara PhD.	inteeprof. RND	r. Pavol Mártonf	i, PhD.Guarantee	eprof. RNDr. Vla	dimír Zeleňák,

University: P. J.	Šafárik Univers	ity in Košice			
Faculty: Faculty	of Science				
<b>Course ID:</b> ÚBI HISE1/15	EV/ Course na	me: Histology			
Recommended	Lecture / Practice l course-load (h 2 Per study perio	ours):			
Number of cred	lits: 6			_	
Recommended	semester/trimes	ster of the cours	<b>e:</b> 2.		
Course level: I.					
Prerequisities:	ÚBEV/CYT1/02	or ÚBEV/CYT1	/15		
<b>Conditions for</b> Oral examination	-	on:			
<b>Learning outco</b> To provide the s		owledge of basic	morphology of ti	issues of animals	
hemopoiesis. C system. Digestiv	irculatory system	tive tissue. Carti n. Lymphoid sys ry system. Femal	stem. Endocrine	system.Integume	ent. Respiratory
1997 Juanqueira, L.C Apleton & Lang	iatt, J.L.: Color 7 ., Carneiro, J., K ge, 1992	Fexbook of Histo elley, R.O.: Basi ina: Histology, Li	c Histology. Pren	ntice Hall Interna	tional Inc.,
Course languag	ge:				
Course assessm Total number of	ent assessed studen	ts: 1838			
А	В	С	D	Е	FX
17.68	13.71	15.78	12.57	25.46	14.8
Provides: doc. F Matiašová, PhD.		axnerová, CSc., l	RNDr. Juraj Ševc	, PhD., RNDr. A	nna Alexovič
Date of last mo	dification: 23.02	2.2018			
<b>Approved:</b> Gua PhD.	ranteeprof. RND	r. Pavol Mártonf	i, PhD.Guarantee	eprof. RNDr. Vla	dimír Zeleňák,

University: P. J. Ša	fárik Univers	ity in Košice			
Faculty: Faculty of	Science				
<b>Course ID:</b> KPE/ INP/17	Course na	me: Inclusive P	edagogy		
Course type, scope Course type: Prac Recommended co Per week: 2 Per s Course method: p	tice ourse-load (h tudy period:	ours):			
Number of credits	: 2				
Recommended sen	nester/trimes	ter of the cours	<b>e:</b> 5.		
Course level: I.					
Prerequisities:					
Conditions for cou	rse completi	on:			
Learning outcome	s:				
Brief outline of the	e course:				
<b>Recommended</b> lite	rature:				
Course language:					
<b>Course assessment</b> Total number of ass		ts: 0			
A	В	С	D	Е	FX
0.0	0.0	0.0	0.0	0.0	0.0
Provides:					
Date of last modifi	cation: 05.02	.2018			
<b>Approved:</b> Guaran <sup>®</sup> PhD.	teeprof. RND	r. Pavol Mártonf	i, PhD.Guarantee	eprof. RNDr. Vla	dimír Zeleňák,

Es aveltave Es avelte		sity in Košice			
Faculty: Facult					
<b>Course ID:</b> ÚC ISC1a/00	HV/ Course n	ame: Cheminfor	matics I		
	Practice d course-load (l er study period	hours):			
Number of crea	dits: 2				
Recommended	semester/trime	ester of the cours	se: 1.		
Course level: I.					
Prerequisities:					
Conditions for seminar project	-	tion:			
chemical inform secondary litera <b>Brief outline of</b> Searching, retri- journals, Chem	nation on internet ture. <b>The course:</b> eving and use of nical Abstracts,	The class will cov et, searching for p The informations Beilstein).Searc Science, Medline	in chemistry. Us hing chemical in	n and work with ing of "paper" res nformation on I	the primary ar
Recommended 1. R.E. Maizell New York 1998 2. Internet resou	literature: How to find Cl urces for chemis	nemical Informat			
Course language slovak language	ge: e and english lar	nguage			
Course assessme Total number of	ent f assessed studer	nts: 787			
А	В	С	D	Е	FX
68.49	8.64	13.21	7.24	1.65	0.76
Provides: RND	r. Monika Tvrdo	oňová, PhD., RNI	Dr. Ladislav Jano	ovec, PhD.	
Date of last mo	dification: 26.0	2.2018			

Faculty: Faculty of Course ID: ÚCHY KCHU/03		mar Coordinati			
KCHU/03 Course type, scop	V/ Course na	mar Coordinati			
• • • •		me: Coordinati	on Chemistry		
Recommended c Per week: 2 / 1 F Course method:	cture / Practice course-load (ho Per study perio	ours):			
Number of credit	s: 4				
Recommended se	mester/trimes	ter of the cour	se: 5.		
Course level: I.					
Prerequisities: Ú	CHV/ACHU/03	3			
<b>Conditions for co</b> Final written exan	-	)n:			
Learning outcom The student acquir and properties of a compounds. Brief outline of th Definition and no	res basic knowl coordination co ne course: menclature of o	coordination co	ell as about the ch mpounds. Centra	nemical bonding i	n coordination
numbers. Isomeria		and stability of	of coordination c	compounds, chem	ical bonding i
Recommended lit J. Ribas: Coordina J. C. Huheey, E. A G. A. Lawrance: I	terature: ation Chemistry A. Keiter, R. L.	Keiter: Inorgan	ic Chemistry, Ha	per Collins, New	York, 1993.
Course language:					
Course assessmer Total number of a		s: 63			
A	В	С	D	Е	FX
55.56	22.22	15.87	3.17	3.17	0.0
/	NDr. Juraj Čern		<u>.</u>	·	

University: P. J. Safái	rik University in Košice				
Faculty: Faculty of S	cience				
Course ID: ÚTVŠ/ Course name: Survival Course KP/12					
Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: Per study period: 36s Course method: present					
Number of credits: 2					
Recommended seme	ster/trimester of the cours	e:			
Course level: I., II.					
Prerequisities:					
<b>Conditions for cours</b> Conditions for course Attendance Final assessment: cor	1	ks within the course			
conditions as they wi and demanding situat	Il obtain theoretical knowled ions connected with surviva work and students will lear	afe stay and movement in extreme natural dge and practical skills to solve the extraordinary and minimization of damage to health. The m how to manage and face the situations that			
<b>Brief outline of the c</b> Brief outline of the co Lectures: 1. Principles of behav	ourse:				
<ol> <li>Preparation and lea</li> <li>Objective and subj</li> <li>Principles of hygie</li> <li>Exercises:</li> <li>Movement in terra</li> </ol>	ective danger in mountains ne and prevention of damag in, orientation and navigatic rovised overnight stay	ent and stay in unknown mountains ge to health in extreme conditions on in terrain (compasses, GPS)			
<ol> <li>Preparation and lea</li> <li>Objective and subj</li> <li>Principles of hygie</li> <li>Exercises:         <ol> <li>Movement in terra</li> <li>Preparation of imp</li> </ol> </li> </ol>	ective danger in mountains ne and prevention of damag in, orientation and navigatio rovised overnight stay d food preparation.	e to health in extreme conditions			
<ol> <li>Preparation and lea</li> <li>Objective and subj</li> <li>Principles of hygie</li> <li>Exercises:         <ol> <li>Movement in terra</li> <li>Preparation of imp</li> <li>Water treatment an</li> </ol> </li> </ol>	ective danger in mountains ne and prevention of damag in, orientation and navigatio rovised overnight stay d food preparation.	e to health in extreme conditions			
<ol> <li>Preparation and lea</li> <li>Objective and subj</li> <li>Principles of hygie</li> <li>Exercises:         <ol> <li>Movement in terra</li> <li>Preparation of imp</li> <li>Water treatment an</li> </ol> </li> <li>Recommended litera</li> </ol>	ective danger in mountains ne and prevention of damag in, orientation and navigatic rovised overnight stay d food preparation. ture:	e to health in extreme conditions			
<ol> <li>Preparation and lea</li> <li>Objective and subj</li> <li>Principles of hygie</li> <li>Exercises:         <ol> <li>Movement in terra</li> <li>Preparation of imp</li> <li>Water treatment an</li> </ol> </li> <li>Recommended litera</li> <li>Course language:</li> <li>Course assessment</li> </ol>	ective danger in mountains ne and prevention of damag in, orientation and navigatic rovised overnight stay d food preparation. ture:	e to health in extreme conditions			

Provides: MUDr. Peter Dombrovský, Mgr. Marek Valanský

Date of last modification: 18.08.2017

University: P. J. Šafá	rik University in Košice	
Faculty: Faculty of S	cience	
<b>Course ID:</b> ÚTVŠ/ LKSp/13	Course name: Summer Co	ourse-Rafting of TISA River
Course type, scope a Course type: Practic Recommended cour Per week: Per stud Course method: pre	ce <b>rse-load (hours):</b> l <b>y period:</b> 36s	
Number of credits: 2		
Recommended seme	ster/trimester of the cours	e:
Course level: I., II.		
Prerequisities:		
<b>Conditions for course</b> Conditions for course Attendance Final assessment: Ra	-	attended/not attended)
Learning outcomes: Learning outcomes: Students have knowle	edge of rafts (canoe) and the	eir control on waterway.
5. Canoe lifting and c	ourse: iculty of waterways iting ning using an empty canoe carrying n the water without a shore be out of the water	contact
Recommended litera	iture:	
Course language:		
<b>Course assessment</b> Total number of asses	ssed students: 142	
	abs	n
	41.55	58.45

Provides: Mgr. Peter Bakalár, PhD.

Date of last modification: 18.08.2017

University: P. J.	Šafárik Univers	sity in Košice			
Faculty: Faculty	of Science				
Course ID: ÚBE MB1/01	COURSE N	ame: Molecular I	Biology		
Course type, sco Course type: La Recommended Per week: 3 Per Course method	ecture course-load (h r study period:	ours):			
Number of credi	its: 4				
Recommended s	semester/trime	ster of the cours	<b>e:</b> 4.		
Course level: I.					
Prerequisities:					
<b>Conditions for c</b> Oral examination	-	ion:			
Learning outcom To provide the st expression and d Brief outline of t	udents with known with known with known with known with a second se	owledge of molec	cular basis of inl	heritance and con	trol of gene
replication and re	epair, transcript		n. Prokaryotic a	Iolecular mechar nd eukaryotic gen ycle.	
Freeman and Co	more, D., Berk mpany, New Yo	ork, 1995		y. Sci. Amer. Boo shers Inc., New Y	
Course language	e:				
Course assessme Total number of		its: 920			
Α	В	С	D	E	FX
6.63	10.87	16.96	18.26	34.02	13.26
Provides: doc. R	NDr. Peter Pris	taš, CSc.			1
Date of last mod	lification: 23.02	2.2018			
Approved: Guar	anteeprof. RND	Dr. Pavol Mártonf	i PhD Guarante	eprof RNDr Vla	adimír Zeleňák

University: P. J.	Šafárik Univers	ity in Košice			
Faculty: Faculty	of Science				
<b>Course ID:</b> ÚCF MIN1/14	IV/ Course na	me: Basis of Mi	neralogy		
Recommended	ecture / Practice course-load (h Per study peri	ours):			
Number of cred	its: 5				
Recommended s	semester/trimes	ter of the cours	e: 4.		
Course level: I.					
<b>Prerequisities:</b> U ÚCHV/VACH/1	ÚCHV/VCH/03 0 or ÚCHV/CH	or ÚCHV/VCH/ G/09 or ÚCHV/Z	10 or ÚCHV/V( CF/03	CHU/10 or ÚCHV	V/ZAC2/10 or
	heoretical knowl	edge and recogn		tional oral examin	ation.
students with pro Brief outline of Basic terms and crystallography: cells and their pa types of bonds a	e beauty of natur operties of usual the course: definitions, ori characteristic p trameters, crysta and structures an	minerals and to gin of minerals in roperties of cryst llographic system d their effect on	in nature. Basis tals, crystallogra tals with example the properties o	from mineralogy. minerals. of morphologica aphic laws, crysta s of minerals. Cryst f minerals. Physic netic and systema	I and structural I structure, unit stallochemistry: cal properties of
V. Radzo: Miner	l <b>iterature:</b> eralógia. Elfa, s. ralógia, Alfa Bra	r.o. Košice, 2001 tislava, 1987.			
<b>Course languag</b> Slovak	e:				
Course assessm Total number of		ts: 63			
А	В	С	D	Е	FX
84.13	11.11	1.59	1.59	0.0	1.59
Provides: doc. R	NDr. Ivan Poto	čňák, PhD.			
Date of last mod	lification: 26.02	.2018			
<b>Approved:</b> Guar PhD.	canteeprof. RND	r. Pavol Mártonf	i, PhD.Guarante	eeprof. RNDr. Vla	dimír Zeleňák,

University: P. J. Šaf	árik Universi	ty in Košice			
Faculty: Faculty of	Science				
Course ID: ÚBEV/ MKV/15	Course na	me: Mikrobioló	gia a základy vi	rológie	
Course type, scope Course type: Lectu Recommended cou Per week: 2 / 2 Per Course method: pr	ure / Practice urse-load (he r study perio	ours):			
Number of credits:	5				
Recommended sem	ester/trimes	ter of the cours	e:		
Course level: I.					
Prerequisities:					
Conditions for cour Attendance of practi examination	-		examinations d	uring semester, fi	nal oral
Learning outcomes Students will obtain their cytology, physi basic methods for st	a basic infor iology, genet	ics, ecology, cla	ssification, and i	-	-
<b>Brief outline of the</b> Viruses, prokaryotic classification. The in	and eukaryo				netics, ecology
Recommended liter	ature:				
Course language:					
Course assessment Total number of asse	essed student	s: 1339			
A	В	С	D	Е	FX
21.73	12.85	18.15	19.94	22.78	4.56
<b>Provides:</b> doc. RND Maliničová, PhD.	r. Peter Prist	aš, CSc., RNDr.	Mariana Kolesá	rová, PhD., RND	r. Lenka
Date of last modific	ation: 23.02	.2018			
<b>Approved:</b> Guarante PhD.	eeprof. RND	r. Pavol Mártonf	i, PhD.Guarante	eprof. RNDr. Vla	dimír Zeleňák,

University: P. J. Ša	afárik Univers	ity in Košice			
Faculty: Faculty o	f Science				
Course ID: KPE/ MMKV/17	Course na	me: Multicultur	alism and Multic	cultural Education	1
Course type, scop Course type: Pra Recommended c Per week: 2 Per Course method:	ctice ourse-load (h study period:	ours):			
Number of credits	s: 2				
Recommended set	mester/trimes	ter of the cours	<b>e:</b> 4.		
Course level: I.					
Prerequisities:					
Conditions for co	urse completi	on:			
Learning outcome	es:				
Brief outline of th	e course:				
Recommended lit	erature:				
Course language:					
Course assessmen Total number of as	-	ts: 48			
A	В	С	D	E	FX
31.25	27.08	37.5	2.08	2.08	0.0
Provides: PaedDr.	Janka Ferenco	ová, PhD.		·	
Date of last modif	ication: 05.02	.2018			
<b>Approved:</b> Guarar PhD.	nteeprof. RND	r. Pavol Mártonf	i, PhD.Guarante	eprof. RNDr. Vla	dimír Zeleňák,

Fooulty Fooulty of S	
Faculty: Faculty of S	cience
<b>Course ID:</b> ÚMV/ MTB/13	Course name: Mathematics for biologists
Course type, scope a Course type: Lectur Recommended cour Per week: 2 / 2 Per Course method: pre	re / Practice rse-load (hours): study period: 28 / 28
Number of credits: 5	5
Recommended seme	ster/trimester of the course: 4.
Course level: I.	
Prerequisities:	
Conditions for cours	se completion:
	mathematics, mathematical problem solving strategies and their applications n biology and other sciences.
<ol> <li>Systems of linear elimination)</li> <li>Functions (monot functions and their propriets)</li> <li>Combinatorics (bininclusion-exclusion propriets)</li> <li>Sequences and servering the sequences and servering sequences (sum, polynomial, analysis)</li> <li>Integrals (indefinite definite integral)</li> </ol>	inomial theorem, combinations and permutations without / with repetition, brinciple) ies (monotonicity and boundedness, recurrent sequence, geometric series) equence, limit of function, convergence, divergence, methods for computing product, quotient and chain rule, derivatives of elementary functions, Taylor of functions) ie integral, integration methods: by substitution, by parts, by partial fractions; ttial equations (first order separable ODE, first order linear ODE)

A B C D E FX							
7.85	11.21	16.82	19.63	33.08	11.4		
Provides: RND	Provides: RNDr. Igor Fabrici, Dr. rer. nat., RNDr. Anton Hovana						
Date of last mo	Date of last modification: 27.02.2018						
Approved: Guaranteeprof. RNDr. Pavol Mártonfi, PhD.Guaranteeprof. RNDr. Vladimír Zeleňák, PhD.							

University: P. J. Šafárik University in Košice	University: P J Šafárik Unive	ersity in Košice
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Faculty: Faculty of Science

**Course ID:** ÚCHV/ **Course name:** Structure determination - spectroscopic methods MUSU/15

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 3 Per study period: 28 / 42

Course method: present

Number of credits: 7

Recommended semester/trimester of the course: 6.

Course level: I.

Prerequisities: ÚCHV/ACHU/03 and ÚCHV/ANCHU/03 and ÚCHV/OCHU/03

**Conditions for course completion:** 

Learning outcomes:

#### **Brief outline of the course:**

Fundamentals of molecular spectroscopy and magnetic properties study, as powerful tools for structure determination in chemistry. Those are ultraviolet, visible, infrared and Raman spectroscopy, mass spectrometry and methods based on magnetic resonance (1H NMR, 13C NMR).

#### **Recommended literature:**

L.G.Wade, Jr.: Organic Chemistry. Prentice Hall International, Inc. Englewood Cliffs, New Yersey 1995.

#### **Course language:**

#### **Course assessment**

Total number of assessed students: 96

А	В	С	D	Е	FX
10.42	34.38	33.33	18.75	3.13	0.0

**Provides:** doc. RNDr. Ján Imrich, CSc., RNDr. Monika Tvrdoňová, PhD., doc. RNDr. Juraj Kuchár, PhD.

Date of last modification: 26.02.2018

University: P. J.	Šafárik Univers	ity in Košice			
Faculty: Faculty	y of Science				
<b>Course ID:</b> KG NJKG/07	ER/ Course na	me: Communica	ative Grammar in	n German Langua	ige
Course type: F Recommended	l course-load (h er study period:	ours):			
Number of crea	lits: 2				
Recommended	semester/trimes	ter of the cours	e:		
Course level: I.,	, II.				
Prerequisities:					
Conditions for	course completi	on:			
Learning outco	mes:				
Brief outline of	the course:				
Recommended	literature:				
Course languag	ge:				
Course assessm Total number of	ent f assessed studen	ts: 48			
А	В	С	D	Е	FX
54.17	12.5	10.42	4.17	10.42	8.33
Provides: PaedI	Dr. Ingrid Puchal	ová, PhD., Mgr.	Barbora Moloká	čová	
Date of last mo	dification: 25.08	.2017			
<b>Approved:</b> Gua PhD.	ranteeprof. RND	r. Pavol Mártonf	i, PhD.Guarante	eprof. RNDr. Vla	dimír Zeleňák,

University: P.	J Šafárik	University in	Košice
University. 1.	J. Darank	Oniversity in	RUSICC

Faculty: Faculty of Science

Course ID: ÚCHV/	Course name: Organic chemistry II
OCH1b/03	

Course type, scope and the method: Course type: Lecture / Practice

**Recommended course-load (hours): Per week:** 3 / 2 **Per study period:** 42 / 28

Course method: present

Number of credits: 7

**Recommended semester/trimester of the course:** 

Course level: I.

**Prerequisities:** 

#### **Conditions for course completion:**

Two tests at lecture in 7 and 14th week. Test max 50 points. At least 25 points required. Written exam, 100 points. At least 49% of points required.

Final evaluation: A 90-100 pts, B 80-89 pts, C 70-79 pts, D 60-69 pts, E 50-59 pts, FX 0-49 pts

#### Learning outcomes:

Second part of two-semester organic chemistry course.

#### Brief outline of the course:

Reaction Mechanisms, Mechanisms of Organic Reactions, Reactive Intermediates, Ionic Reactions Radical Reactions Bond Energy Reaction Energetics Activation Energy Reaction Rates and Kinetics Thermodynamic and Chemical Stability Aromaticity Benzene and Other Aromatic Compounds Fused Benzene Ring Compounds Other Aromatic Systems Factors Required for Aromaticity Stereoisomers Chirality and Symmetry Enantiomorphism Polarimetry Optical Activity Designating the Configuration of Stereogenic Centers The Sequence Rule for Assignment of Configurations to Stereogenic Carbons Compounds Having Two or More Stereogenic Centers Stereogenic Nitrogen Fischer Projection Formulas Aldehydes & Ketones Natural Products Synthetic Preparation Properties of Aldehydes & Ketones Reversible Addition Reactions Hydration & Hemiacetal Formation Acetal Formation Imine Formation Enamine Formation Cyanohydrin Formation Irreversible Addition Reactions Complex Metal Hydrides Organometallic Reagents Carbonyl Group Modification Wolff-Kishner Reduction Clemmensen Reduction Hydrogenolysis of Thioacetals Oxidations Reactions at the a-Carbon Mechanism of Electrophilic a-Substitution The Aldol Reaction Ambident Enolate Anions Alkylation of Enolate Anions Carboxylic Acids Natural Products Related Derivatives Preparation of Carboxylic Acids Reactions of Carboxylic Acids Salt Formation Substitution of Hydroxyl Hydrogen Substitution of the Hydroxyl Group Reduction & Oxidation Carboxylic Derivatives Reactions of Carboxylic Acid Derivatives Acyl Group Substitution Mechanism Reduction Catalytic Reduction Metal Hydride Reduction Diborane Reduction Reaction with Organometallic Reagents Reactions at the a Carbon Acidity of a C-H The Claisen Condensation Synthesis Applications Carbohydrates Glucose The Structure and Configuration of Glucose Anomeric Forms of Monosaccharides Glycosides Disaccharides Polysaccharides Lipids Fatty Acids Soaps & Detergents Fats & Oils Nucleic Acids Alkaloids Terpenes

**Recommended literature:** 

- 1. on-line moodle.science.upjs.sk
- 2. Organic Chemistry, Clayden, Greeves Warren & Wothers, Oxford University Press, 2010
- 3. Organic Chemistry, Solomon, Willey, 2009

### **Course language:**

# Course assessment

Total number of assessed students: 571

А	В	С	D	Е	FX
11.91	11.38	16.99	22.07	34.68	2.98

Provides: prof. RNDr. Jozef Gonda, DrSc., doc. RNDr. Miroslava Martinková, PhD.

**Date of last modification:** 26.02.2018

University: P. J. Šafárik University in Košice
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Faculty: Faculty of Science

Course ID: ÚCHV/	Course name: Organic chemistry
OCHU/03	

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours):

Per week: 3 / 1 Per study period: 42 / 14

Course method: present

Number of credits: 6

#### Recommended semester/trimester of the course: 2.

Course level: I.

Prerequisities: ÚCHV/VCHU/15 or ÚCHV/VCHU/14 or ÚCHV/VCHU/10 or ÚCHV/VACH/10

#### **Conditions for course completion:**

Two tests at lecture in 7 and 14th week. Test max 50 points. At least 25 points required. Written exam, 100 points. At least 49% of points required.

Final evaluation: A 90-100 pts, B 80-89 pts, C 70-79 pts, D 60-69 pts, E 50-59 pts, FX 0-49 pts

#### Learning outcomes:

Basic organic chemistry course.

#### Brief outline of the course:

Chemical bonding Hybridization and Bonding Covalent bonds Double bonds and Triple Bonds Structural Formulas of Organic Molecules Polar Covalent Bonds and Electronegativity Constitutional Isomers Alkenes Electrophilic Additions Strong Brønsted Acids Lewis Acids (non-Proton Electrophiles) Electrophilic Halogen Reagents Other Electrophilic Reagents Reduction Oxidation Radical Additions Allylic Substitution Alkynes Addition Reactions Hydrogenation Electrophiles Hydration & Tautomerism Hydroboration Nucleophilile Addition & Reduction Acidity of Terminal Alkynes (Substitution of H) Alkyl Halides General Reactivity Substitution(of X) SN2 Mechanism SN1 Mechanism Elimination (of HX) Summary of Substitution vs. Elimination Substitution by Metals Elimination Reactions of Dihalides Alcohols Reactions of Alcohols Substitution of the Hydroxyl H Substitution of the Hydroxyl Group Elimination of Water Oxidation of Alcohols Reactions of Phenols Acidity of Phenols Ring Substitution of Phenols Oxidation to Quinones Aromatic compounds Electrophilic Substitution A Substitution Mechanism Reactions of Substituted Benzenes Reaction Characteristics Reactions of Disubstituted Rings Reactions of Substituent Groups Nucleophilic Substitution, Elimination & Addition Reactions Amines Basicity of Nitrogen Compounds Acidity of Nitrogen Compounds Important Reagent Bases Reactions of Amines Electrophilic Substitution at Nitrogen Preparation of 1°-Amines Preparation of 2° & 3°-Amines Reactions with Nitrous Acid Reactions of Aryl Diazonium Intermediates Elimination Reactions of Amines Oxidation States of Nitrogen Basic information: Aldehydes & Ketones Carboxylic Acids Carboxylic Derivatives Natural products

#### **Recommended literature:**

1. on-line ppt presentation in MOODLE, moodle science.upjs.sk

- 2. Organic Chemistry, Clayden, Greeves Warren & Wothers, Oxford University Press, 2010
- 3. Organic Chemistry, Solomon, Willey, 2009

Course languag	ge:				
<b>Course assessm</b> Total number o	nent f assessed studen	ts: 700			
А	В	С	D	Е	FX
3.14	7.43	13.43	22.43	47.57	6.0
Provides: prof.	RNDr. Jozef Goi	nda, DrSc., RND	r. Slávka Hamuľ	aková, PhD.	
Date of last mo	dification: 26.02	2.2018			
<b>Approved:</b> Gua PhD.	aranteeprof. RND	r. Pavol Mártonf	i, PhD.Guarante	eprof. RNDr. Vla	dimír Zeleňák,

University: P. J.	Šafárik Universi	ty in Košice			
Faculty: Faculty	of Science				
Course ID: KGE OJPV1/07	ER/ Course na	me: Specialised	German Langua	ge - Natural Scie	ences 1
	ractice course-load (he r study period:	ours):			
Number of cred	its: 2				
Recommended s	semester/trimes	ter of the cours	e: 4.		
Course level: I.					
Prerequisities:					
Conditions for <b>c</b>	course completio	on:			
Learning outcom	mes:				
Brief outline of	the course:				
Recommended l	literature:				
Course languag	e:				
Course assessme Total number of	ent assessed student	s: 136			
A	В	С	D	Е	FX
21.32	22.79	25.0	22.06	8.09	0.74
Provides: Mgr. A	Andreas Schiestl				
Date of last mod	lification: 25.08	.2017			
<b>Approved:</b> Guar PhD.	anteeprof. RND	r. Pavol Márton	ñ, PhD.Guarantee	eprof. RNDr. Vla	ıdimír Zeleňál

University: P. J.	Šafárik Univers	ity in Košice			
Faculty: Faculty	of Science				
<b>Course ID:</b> KPE/ OLŠ/15	Course na	ame: School Adr	ninistration and	Legislation	
Course type, sco Course type: Pr Recommended Per week: 2 Per Course method	actice course-load (h study period:	ours):			
Number of credi	ts: 2				
Recommended s	emester/trimes	ster of the cours	<b>e:</b> 3., 5.		
Course level: I.					
Prerequisities:					
Conditions for co	ourse completi	ion:			
Learning outcon	nes:				
Brief outline of t	he course:				
Recommended li	iterature:				
Course language	•				
<b>Course assessme</b> Total number of a	-	ıts: 168			
A	В	С	D	E	FX
35.71	30.36	22.02	8.33	2.98	0.6
Provides: PaedD	r. Renáta Orosc	ová, PhD.	1	·	
Date of last mod	ification: 23.08	3.2017			
<b>Approved:</b> Guara PhD.	anteeprof. RND	Pr. Pavol Mártonf	i, PhD.Guarante	eprof. RNDr. Vla	dimír Zeleňák,

University: P. J. Šaf	árik University in Košice	
Faculty: Faculty of	Science	
<b>Course ID:</b> KOP/ OPaPDV/14	<b>Course name:</b> Civil Law a	and Intellectual Property Rights
Course type, scope Course type: Lectu Recommended cou Per week: 2 Per st Course method: pr	ire irse-load (hours): udy period: 28	
Number of credits:	4	
Recommended sem	ester/trimester of the cours	<b>e:</b> 3., 5.
Course level: I., N		
Prerequisities:		
Conditions for cour	se completion:	
Learning outcomes	•	
Brief outline of the	course:	
Recommended liter	rature:	
Course language:		
<b>Course assessment</b> Total number of ass	essed students: 67	
	abs	n
	94.03	5.97
Provides: doc. JUD	: Renáta Bačárová, PhD., LI	M., prof. JUDr. Peter Vojčík, CSc.
Date of last modific	ation: 18.01.2018	
Approved: Guarante PhD.	eeprof. RNDr. Pavol Márton	ň, PhD.Guaranteeprof. RNDr. Vladimír Zeleňák,

University:	ΡI	Šafárik	University	in Košice
University.	F. J	. Salalik	University	III KUSICE

Faculty: Faculty of Science

Course ID: ÚCHV/	Course name: Practical from Inorganic Chemistry
PACHU/03	

Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 4 Per study period: 56

Course method: present

Number of credits: 4

Recommended semester/trimester of the course: 2.

Course level: I.

Prerequisities: ÚCHV/VCHU/14 or ÚCHV/VCHU/15 or ÚCHV/VCHU/10 or ÚCHV/VACH/10

Conditions for course completion:

#### Learning outcomes:

The practical acquirements at preparation and study of inorganic compounds and their physicochemical properties by common laboratory techniques.

#### Brief outline of the course:

The utilization of common laboratory techniques for preparation of elements (H2, O2, Cu, Ni), oxides(CO2, Al2O3·xH2O), nitrides(Mg3N2), acids (HNO3, H3BO3), salts((NH4)2SO4, KMnO4), binary salts(NH4)Fe(SO4)2·12H2O), halides (CuCl, CuCl2·2H2O, SnI4, CuBr2) and coordination compounds ([Cr2(CH3COO)4(H2O)2], [CoCl2(en)2]Cl, [Cu(NH3)4]SO4·H2O, K3[Al(C2O4)3]·3H2O).

#### **Recommended literature:**

Z. Vargová, J. Kuchár: Praktikum z anorganickej chémie, Košice, 2008
M. Reháková, M. Dzurillová, V. Zeleňák, V. Urvichiarová: Laboratórna technika, PF UPJŠ, Košice, 1999

#### **Course language:**

#### **Course assessment**

Total number of assessed students: 481

А	В	С	D	Е	FX
51.77	27.03	15.18	2.7	2.29	1.04

**Provides:** doc. RNDr. Juraj Kuchár, PhD., RNDr. Martin Vavra, PhD., RNDr. Miroslava Matiková-Maľarová, PhD.

Date of last modification: 26.02.2018

Faculty Faculty		ity in Košice			
raculty. Faculty	of Science				
<b>Course ID:</b> ÚCH PACU/03	IV/ Course na	ame: General Cou	urse of Analytic	cal Chemistry - La	boratory
Course type, sco Course type: Pr Recommended Per week: 4 Per Course method	ractice course-load (h r study period:	ours):			
Number of credi	its: 4				
Recommended s	semester/trimes	ster of the course	e <b>:</b> 4.		
Course level: I.					
Prerequisities: Ú	JCHV/ANCHU	/03			
Conditions for c Assessment	ourse completi	on:			
Learning outcom Application of th		ledge to analytica	l laboratory pra	actise	
_	-	thods. Gravimeti	ry, general pr	analysis, separation inciples of metho	-
methods. Prepa curves, calculation	ons in volumet		limetry, alkalin	netry. Manganome	ooint. Titratio
methods. Prepa curves, calculation Complexometry. <b>Recommended I</b> D.Harvey: Mode D.A.Skoog: Prin	ons in volumet Selected Instru iterature: ern Analytical C ciples of Instru	ric analysis. Acic mental analytical hemistry. McGra	limetry, alkalin methods. w Hill, Boston, Saunders Col. 1	netry. Manganome , 2000. Publishing, New Y	ooint. Titratio etry. Iodometry
methods. Prepa curves, calculation Complexometry. <b>Recommended I</b> D.Harvey: Mode D.A.Skoog: Prin E.Prichard: Qual	ons in volumet Selected Instru iterature: ern Analytical C ciples of Instru- ity in the Analy	ric analysis. Acid mental analytical hemistry. McGra mental Analysis.	limetry, alkalin methods. w Hill, Boston, Saunders Col. 1	netry. Manganome , 2000. Publishing, New Y	ooint. Titratio etry. Iodometry
methods. Prepa curves, calculation Complexometry. <b>Recommended I</b> D.Harvey: Mode D.A.Skoog: Prin E.Prichard: Qual <b>Course language</b>	ons in volumet Selected Instru iterature: ern Analytical C ciples of Instru- ity in the Analy e: ent	ric analysis. Acic mental analytical hemistry. McGra mental Analysis. tical Chemistry I	limetry, alkalin methods. w Hill, Boston, Saunders Col. 1	netry. Manganome , 2000. Publishing, New Y	ooint. Titratio etry. Iodometry
methods. Prepa curves, calculation Complexometry. Recommended I D.Harvey: Mode D.A.Skoog: Prin E.Prichard: Qual Course language Course assessme	ons in volumet Selected Instru iterature: ern Analytical C ciples of Instru- ity in the Analy e: ent	ric analysis. Acic mental analytical hemistry. McGra mental Analysis. tical Chemistry I	limetry, alkalin methods. w Hill, Boston, Saunders Col. 1	netry. Manganome , 2000. Publishing, New Y	ooint. Titratio etry. Iodometry
methods. Prepa curves, calculati Complexometry. <b>Recommended I</b> D.Harvey: Mode D.A.Skoog: Prin E.Prichard: Qual <b>Course language</b> <b>Course assessme</b> Total number of	ons in volumet Selected Instru iterature: ern Analytical C ciples of Instru- ity in the Analy e: ent assessed studen	ric analysis. Acid mental analytical hemistry. McGra mental Analysis. tical Chemistry I ts: 286	limetry, alkalin methods. w Hill, Boston, Saunders Col. I aboratory, Wil	netry. Manganome , 2000. Publishing, New Y ey, 1995	ooint. Titratio etry. Iodometry York 1985.
methods. Prepa curves, calculation Complexometry. Recommended I D.Harvey: Mode D.A.Skoog: Prin E.Prichard: Qual Course language Course assessme Total number of A 56.99 Provides: doc. In	ons in volumet Selected Instru iterature: ern Analytical C ciples of Instru- ity in the Analy e: ent assessed studen B 28.32 ng. Viera Vojtek	ric analysis. Acid mental analytical hemistry. McGra mental Analysis. rtical Chemistry I ts: 286 C 11.54 ová, PhD., RNDr	limetry, alkalin methods. w Hill, Boston, Saunders Col. I aboratory, Wil D 1.4	netry. Manganome , 2000. Publishing, New Y ey, 1995 E	ooint. Titratio etry. Iodometry York 1985. FX 0.0
methods. Prepa curves, calculation Complexometry. <b>Recommended I</b> D.Harvey: Mode D.A.Skoog: Prin E.Prichard: Qual <b>Course language</b> <b>Course assessme</b> Total number of A 56.99	ons in volumet Selected Instru iterature: ern Analytical C ciples of Instru- ity in the Analy e: ent assessed studen B 28.32 ng. Viera Vojtek a Šandrejová, P	ric analysis. Acid mental analytical hemistry. McGra mental Analysis. rtical Chemistry I ts: 286 C 11.54 ová, PhD., RNDr hD.	limetry, alkalin methods. w Hill, Boston, Saunders Col. I aboratory, Wil D 1.4	netry. Manganome , 2000. Publishing, New Y ey, 1995 E 1.75	ooint. Titratio etry. Iodometry York 1985. FX 0.0

Course ID: ÚCHV/ PBCHU/15       Course name: Biochemistry Practical         Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 4 Per study period: 56 Course method: present         Number of credits: 4         Recommended semester/trimester of the course: 6.         Course level: I.         Prerequisities: ÚCHV/BCHU/03         Conditions for course completion: Protocols + 75 % continuous evaluation.         Learning outcomes:         Brief outline of the course: The most important biochemical laboratory methods. The qualitative tests and proteins. Time-dependent course of enzyme-catalyzed reaction: determina activity, determination of the first order rate constant, calculations of math m effect of a substrate concentration on initial rate of reaction, determination of I urease. Isolation and detection of nucleic acids.         Recommended literature: http://kosice.upjs.sk/~kbch/         Course language:         Course language:         Course language:         Course language:         A       B       C       D       E         67.37       26.32       4.21       1.05       1.05         Provides: doc. RNDr. Mária Kožurková, CSc., RNDr. Nataša Tomášková, PhD.       Varhač, PhD., RNDr. Danica Sabolová, PhD., RNDr. Eva Konkoľová, PhD.		y of Science				
Course type: Practice         Recommended course-load (hours):         Per week: 4 Per study period: 56         Course method: present         Number of credits: 4         Recommended semester/trimester of the course: 6.         Course level: I.         Prerequisities: ÚCHV/BCHU/03         Conditions for course completion:         Protocols + 75 % continuous evaluation.         Learning outcomes:         Brief outline of the course:         The most important biochemical laboratory methods. The qualitative tests and proteins. Time-dependent course of enzyme-catalyzed reaction: determina activity, determination of the first order rate constant, calculations of math m effect of a substrate concentration on initial rate of reaction, determination of I urease. Isolation and detection of nucleic acids.         Recommended literature:         http://kosice.upjs.sk/~kbch/         Course language:         Course assessment         Total number of assessed students: 95         A       B       C       D       E         67.37       26.32       4.21       1.05       1.05         Provides: doc. RNDr. Mária Kožurková, CSc., RNDr. Nataša Tomášková, PhD.			me: Biochemist	ry Practical		
Recommended semester/trimester of the course: 6.         Course level: I.         Prerequisities: ÚCHV/BCHU/03         Conditions for course completion: Protocols + 75 % continuous evaluation.         Learning outcomes:         Brief outline of the course:         The most important biochemical laboratory methods. The qualitative tests and proteins. Time-dependent course of enzyme-catalyzed reaction: determina activity, determination of the first order rate constant, calculations of math m effect of a substrate concentration on initial rate of reaction, determination of I urease. Isolation and detection of nucleic acids.         Recommended literature: http://kosice.upjs.sk/~kbch/         Course language:         Course assessment         Total number of assessed students: 95         A       B       C       D       E         67.37       26.32       4.21       1.05       1.05         Provides: doc. RNDr. Mária Kožurková, CSc., RNDr. Nataša Tomášková, PhD.	Course type: F Recommended Per week: 4 Pe	Practice I course-load (her er study period:	ours):			
Course level: I.         Prerequisities: ÚCHV/BCHU/03         Conditions for course completion: Protocols + 75 % continuous evaluation.         Learning outcomes:         Brief outline of the course:         Brief outline of the course:         The most important biochemical laboratory methods. The qualitative tests and proteins. Time-dependent course of enzyme-catalyzed reaction: determina activity, determination of the first order rate constant, calculations of math m effect of a substrate concentration on initial rate of reaction, determination of I urease. Isolation and detection of nucleic acids.         Recommended literature:         http://kosice.upjs.sk/~kbch/         Course language:         Course assessment         Total number of assessed students: 95         A       B       C       D       E         67.37       26.32       4.21       1.05       1.05         Provides: doc. RNDr. Mária Kožurková, CSc., RNDr. Nataša Tomášková, PhD.	Number of cred	lits: 4				
Prerequisities: ÚCHV/BCHU/03         Conditions for course completion: Protocols + 75 % continuous evaluation.         Learning outcomes:         Brief outline of the course: The most important biochemical laboratory methods. The qualitative tests and proteins. Time-dependent course of enzyme-catalyzed reaction: determina activity, determination of the first order rate constant, calculations of math m effect of a substrate concentration on initial rate of reaction, determination of I urease. Isolation and detection of nucleic acids.         Recommended literature: http://kosice.upjs.sk/~kbch/         Course language:         Course language:         A       B       C       D       E         67.37       26.32       4.21       1.05       1.05         Provides: doc. RNDr. Mária Kožurková, CSc., RNDr. Nataša Tomášková, PhD.	Recommended	semester/trimes	ster of the cours	<b>e:</b> 6.		
Conditions for course completion: Protocols + 75 % continuous evaluation.         Learning outcomes:         Brief outline of the course: The most important biochemical laboratory methods. The qualitative tests and proteins. Time-dependent course of enzyme-catalyzed reaction: determina activity, determination of the first order rate constant, calculations of math m effect of a substrate concentration on initial rate of reaction, determination of I urease. Isolation and detection of nucleic acids.         Recommended literature: http://kosice.upjs.sk/~kbch/         Course language:         Course assessment         Total number of assessed students: 95         A       B       C       D       E         67.37       26.32       4.21       1.05         Provides: doc. RNDr. Mária Kožurková, CSc., RNDr. Nataša Tomášková, PhD.	Course level: I.					
Protocols + 75 % continuous evaluation.         Learning outcomes:         Brief outline of the course:         The most important biochemical laboratory methods. The qualitative tests and proteins. Time-dependent course of enzyme-catalyzed reaction: determina activity, determination of the first order rate constant, calculations of math m effect of a substrate concentration on initial rate of reaction, determination of I urease. Isolation and detection of nucleic acids.         Recommended literature:         http://kosice.upjs.sk/~kbch/         Course language:         Course assessment         Total number of assessed students: 95         A       B       C       D       E         67.37       26.32       4.21       1.05       1.05         Provides: doc. RNDr. Mária Kožurková, CSc., RNDr. Nataša Tomášková, PhD.	Prerequisities:	ÚCHV/BCHU/0	3			
Brief outline of the course:         The most important biochemical laboratory methods. The qualitative tests and proteins. Time-dependent course of enzyme-catalyzed reaction: determina activity, determination of the first order rate constant, calculations of math m effect of a substrate concentration on initial rate of reaction, determination of I urease. Isolation and detection of nucleic acids.         Recommended literature:       Iterature:         http://kosice.upjs.sk/~kbch/       Volume assessment         Course language:         A       B       C       D       E         67.37       26.32       4.21       1.05       1.05         Provides: doc. RNDr. Mária Kožurková, CSc., RNDr. Nataša Tomášková, PhD.		-				
The most important biochemical laboratory methods. The qualitative tests and proteins. Time-dependent course of enzyme-catalyzed reaction: determination activity, determination of the first order rate constant, calculations of math meffect of a substrate concentration on initial rate of reaction, determination of I urease. Isolation and detection of nucleic acids.          Recommended literature:       http://kosice.upjs.sk/~kbch/         Course language:       Course assessment         Total number of assessed students: 95       D       E         67.37       26.32       4.21       1.05       1.05         Provides: doc. RNDr. Mária Kožurková, CSc., RNDr. Nataša Tomášková, PhD.	Learning outco	mes:				
http://kosice.upjs.sk/~kbch/ Course language: Course assessment Total number of assessed students: 95          A       B       C       D       E         67.37       26.32       4.21       1.05       1.05         Provides: doc. RNDr. Mária Kožurková, CSc., RNDr. Nataša Tomášková, PhD.	and proteins Ti	· · · · ·	C		tion: datarminati	
Course language:Course assessmentTotal number of assessed students: 95ABCDE67.3726.324.211.051.05Provides: doc. RNDr. Mária Kožurková, CSc., RNDr. Nataša Tomášková, PhD.	activity, determ effect of a subst urease. Isolatior	ination of the fit trate concentration and detection o	rst order rate co on on initial rate	nstant, calculati	ions of math mo	ion of enzymation of enzymation of enzymation of enzymatic dels (examples)
Total number of assessed students: 95ABCDE67.3726.324.211.051.05Provides: doc. RNDr. Mária Kožurková, CSc., RNDr. Nataša Tomášková, PhD.	activity, determ effect of a subst urease. Isolatior Recommended	ination of the fin trate concentration and detection of <b>literature:</b>	rst order rate co on on initial rate	nstant, calculati	ions of math mo	ion of enzymation of enzymation of enzymation of enzymatic dels (examples)
67.3726.324.211.051.05Provides: doc. RNDr. Mária Kožurková, CSc., RNDr. Nataša Tomášková, PhD.	activity, determ effect of a subst urease. Isolatior <b>Recommended</b> http://kosice.upj	ination of the fin trate concentration and detection or <b>literature:</b> js.sk/~kbch/	rst order rate co on on initial rate	nstant, calculati	ions of math mo	ion of enzymation of enzymation of enzymation of enzymatic dels (examples)
Provides: doc. RNDr. Mária Kožurková, CSc., RNDr. Nataša Tomášková, PhD.	activity, determ effect of a subst urease. Isolation <b>Recommended</b> http://kosice.upj Course languag	ination of the fin trate concentration and detection of <b>literature:</b> js.sk/~kbch/ ge: ment	rst order rate co on on initial rate f nucleic acids.	nstant, calculati	ions of math mo	ion of enzymation of enzymation of enzymation of enzymatic dels (examples)
	activity, determ effect of a subst urease. Isolation <b>Recommended</b> http://kosice.upj Course languag Course assessm Total number of	ination of the fin trate concentration and detection of <b>literature:</b> js.sk/~kbch/ ge: tent f assessed studen	rst order rate co on on initial rate f nucleic acids. ts: 95	nstant, calculati	ions of math mo termination of K	ion of enzymation of enzymation of enzymation of enzymatic dels (examples)
	activity, determ effect of a subst urease. Isolation Recommended http://kosice.upj Course languag Course assessm Total number of A	ination of the fin trate concentration and detection of <b>literature:</b> js.sk/~kbch/ ge: ment f assessed studen B	rst order rate co on on initial rate f nucleic acids. ts: 95	nstant, calculati of reaction, det D	ions of math mo termination of K	ion of enzymati dels (examples) m and Vmax fo
Date of last modification: 26.02.2018	activity, determ effect of a subst urease. Isolation Recommended http://kosice.upj Course languag Course assessm Total number of A 67.37 Provides: doc. F	ination of the fin trate concentration and detection of <b>literature:</b> js.sk/~kbch/ ge: lent f assessed studen B 26.32 RNDr. Mária Kož	rst order rate co on on initial rate f nucleic acids. ts: 95 C 4.21 žurková, CSc., R	D 1.05 NDr. Nataša To	E 1.05 mášková, PhD.,	ion of enzymati odels (examples) m and Vmax fo FX 0.0

University: P. J.	. Šafárik Univers	sity in Košice						
Faculty: Faculty	y of Science							
Course ID: ÚC PCH1/00	Course ID: ÚCHV/ Course name: Food chemistry PCH1/00							
Recommended	Lecture / Practico d course-load (h l Per study peri	e iours):						
Number of cred	lits: 4							
Recommended	semester/trime	ster of the cours	e: 5.					
Course level: I.,	, II.							
Prerequisities:								
Conditions for	course complet	ion:						
importance and Brief outline of The main catego carbohydrates.	the course: ories of substance Water, minerals,	ons and knowledg es in food during es in the most imp low concentration	processing and ortant group of for anorganic com	storage. ood. Aminoacids pounds, vitamins	, proteins, lipids			
	- · ·	tives. Chemical r	eactions in dairy	products.				
Recommended								
Course languag Course assessm Total number of		nts: 256						
А	В	С	D	Е	FX			
60.55	33.98	5.08	0.0	0.0	0.39			
00.55					0.57			
Provides: RND	r. Ján Elečko, Ph	nD.			0.37			

PhD.

University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
<b>Course ID:</b> CJP/ PFAJ4/07	Course name: English Language of Natural Science
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	ce rse-load (hours): dy period: 28
Number of credits: 2	
Recommended seme	ster/trimester of the course: 4.
Course level: I.	
Prerequisities:	
2 classes at the most. Continuous assessme in English.	n class and completed homework assignments. Students are allowed to miss nt: 2 credit tests (presumably in weeks 6 and 13) and academic presentation
credit tests. The exam test results results represent the o The final grade for th	ed to the final exam, a student has to score at least 65 % as a sum of both represent 50% of the final grade for the course, continuous assessment other 50% of the final grade. e course will be calculated as follows: C 79-85, D 72-78, E 65-71, FX 64 and less.
comprehension) in En competence (familiar improvement of stude	ents' language skills (speaking, writing, reading and listening nglish for specific purposes and development of students' language ization with selected phonological, lexical and syntactic phenomena), ents' pragmatic competence (familiarization with selected language vement of presentation skills at B2 level (CEFR) with focus on terminology science.
Veda a výskum. Odbo Planéta Zem. Naša sl Zemetrasenia, Sopečn Svetové oceány a ľad Životné prostredie a s Počasie a klíma. ANGLICKÝ JAZYK Veda a výskum. Odbo	A PRE GEOGRAFOV: or geografia. nečná sústava. ná činnosť. lovce. geografia. A PRE EKOLÓGOV: or ekológia. nečistenie a dôsledky. netrasenia.

Globálne otepľovanie a dôsledky. Ľadovce. Počasie a klíma. Búrky, hurikány, tsunami. Život na Zemi. Ohrozené rastlinné a živočíšne druhy. ANGLICKÝ JAZYK PRE BIOLÓGOV: veda a výskum, odbor biológia. morfológia rastlín, koreň. stonka, list. rozmnožovanie rastlín, kvet. biológia človeka - telesné sústavy. slovná zásoba z oblasti botanickej a zoologickej nomenklatúry. ANGLICKÝ JAZYK PRE MATEMATIKOV: Veda a výskum, odbor matematika. čísla a tvary v matematike. Elementárna algebra. Elementárna geometria. Výpočty v matematike. Pytagoras, Pytagorova veta. Grafy a diagramy. Štatistika. ANGLICKÝ JAZYK PRE FYZIKOV Veda a výskum, odbor fyzika. Atómy a molekuly. Hmota a jej premeny. Elektrina, jej využitie. Zvuka, jeho prenos. Svetlo. Solárny systém. Matematické operácie. ANGLICKÝ JAZYK PRE CHEMIKOV: Veda a výskum, odbor chémia. História, Každodenná chémia. Laboratórium a jeho vybavenie. Periodická tabuľka. Hmota a jej premeny. Životné prostredie a chémia. ANGLICKÝ JAZYK PRE INFORMATIKOV: Veda a výskum, informatika. Život s počítačom. Typický PC. Zdravie a bezpečnosť, ergonomika. Programovanie. Emailovanie. Cybercrime. Trendy budúcnosti.

#### **Recommended literature:**

study materials provided by the course instructor Royds-Irmak, D.E. Beginning Scientific English. Nelson, 1975. Velebná, B. English for Chemists. ffweb.ff.upjs.sk/vyuka// Redman, S.: English Vocabulary in Use, Pre-intermetdiate, Intermediate. Cambridge University Press, 2003. Powel, M.: Dynamic Presentations. CUP, 2010.

Armer, T.: Cambridge English for Scientists. CUP, 2011.

Wharton J.: Academic Encounters. The Natural World. CUP, 2009.

Murphy, R.: English Grammar in Use. Cambridge University Press, 1994.

Redman, S.: English Vocabulary in Use, Pre-intermetdiate, Intermediate. Cambridge University Press, 2003.

P. Fitzgerald : English for ICT studies. Garnet Publishing, 2011.

https://worldservice/learningenglish, https://spectator.sme.sk

#### **Course language:**

#### **Course assessment**

Total number of assessed students: 2443

Α	B	C	D	E	FX
	D		D	L	171
34.55	25.83	17.6	10.89	8.8	2.33

Provides: Mgr. Zuzana Naďová, Mgr. Lenka Klimčáková

**Date of last modification:** 06.02.2018

**Approved:** Guaranteeprof. RNDr. Pavol Mártonfi, PhD.Guaranteeprof. RNDr. Vladimír Zeleňák, PhD.

University: P. J.	Šafárik Univers	ity in Košice					
Faculty: Faculty	of Science						
Course ID: CJP/     Course name: Academic English       PFAJAKA/07							
Per week: 2 Pe	-	ours): 28					
Number of cred	lits: 2						
Recommended	semester/trimes	ter of the course	<b>)</b> •				
Course level: I.,	II., N						
Prerequisities:							
and 12th/13th wassessment of te	reek), no retake. sts and presentat 1%, FX 64% and	Minipresentation tion. Grading sca	on chosen topic	tolerated. 2 tests c. Final evaluatio B 86-92%, C 79-	n- average		
Brief outline of							
T. Armer :Camb M. McCarthy M Zemach, D.E, R Olsen, A. : Acti www.bbclearnin	nic Encounters, ( oridge English fo I., O'Dell F Ac umisek, L.A: Ac ve Vocabulary, P ogenglish.com	r Scientists, CUP ademic Vocabula ademic Writing,	ry in Use, CUP Macmillan 200				
<b>Course languag</b> English languag	e: e, level B2 acco	rding to CEFR.					
Course assessm Total number of	ent `assessed studen	ts: 344					
А	В	С	D	Е	FX		
30.81	23.55	15.99	11.05	7.27	11.34		
Provides: Mgr. 2	Zuzana Naďová						
Date of last mo	dification: 06.02	.2018					
Approved: Guar PhD.	ranteeprof. RND	r. Pavol Mártonfi	, PhD.Guarante	eprof. RNDr. Vla	adimír Zeleňák,		

University: P. J.	Šafárik Univers	ity in Košice			
Faculty: Faculty	y of Science				
<b>Course ID:</b> CJP PFAJGA/07	Course na	ame: Communic	ative Grammar i	n English	
Course type: F Recommended Per week: 2 Pe	ope and the met Practice I course-load (h er study period: d: combined, pre	<b>ours):</b> 28			
Number of cred	lits: 2				
Recommended	semester/trimes	ster of the cours	e:		
Course level: I.,	, II., N				
Prerequisities:					
86-92%, C 79-8 Learning outco	5%, D 72-78%, mes:	E 65-71%, FX 6		Grading scale: A g	95-100%, B
Brief outline of	the course:				
McCarthy, O'De Alexander L.G. Jones I Comm Vince M.: Macr www.bbclearnin	ematic Vocabular ell: English Voca : Longman Engli nunicative Gram nillan Grammar ngenglish.com	ry, Fragment, 199 bulary in Use, 19 ish Grammar, Lo mar Practice, CU in Context, Mac ise, Polyglot, 200	994 ngman, 1988 JP, 1992 millan, 2008		
Course languag	ge:				
Course assessm	ent fassessed studen	ts: 394			
A	B	C	D	Е	FX
39.34	18.53	17.01	8.88	6.09	10.15
Provides: Mgr.	Lenka Klimčáko	vá		<u> </u>	<u>I</u>
Date of last mo					
			ñ, PhD.Guarante	eprof. RNDr. Vla	adimír Zeleňá

	COURSE INFORMATION LETTER
University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
<b>Course ID:</b> CJP/ PFAJKKA/07	Course name: Communicative Competence in English
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: con	ce rse-load (hours): dy period: 28
Number of credits: 2	
Recommended seme	ster/trimester of the course:
Course level: I., II., N	1
Prerequisities:	
two classes at the mo 2 credit tests (presum on selected topics.	ably in weeks 6/7 and 12/13) and short academic presentations in English alculated as follows: A 93-100 %, B 86-92%, C 79-85%, D 72-78%, E
situáciách. Zdokonale vecnej kompetencie, výpovede, efektívne výpovede. Precvičova oslovenie), informatí časových vzťahov), r a hodnotiacich (napr. budovania praktickej požiadavkám a kritér jazykov.	používanie svojich teoretických vedomostí v praktických komunikačných enie jazykových vedomostí a zručností študenta, rečovej, pragmatickej a predovšetkým zlepšujú komunikáciu, schopnosť prijímať a formulovať vyjadrovať svoje myšlienky ako aj orientovať sa v obsahovom pláne anie rečových intencií kontaktných (napr. pozdravy, oslovenia, pozvanie, vnych (napr. získavanie a podávanie informácií, vyjadrenie priestorových a egulačných (napr. prosba, poďakovanie, zákaz, pochvala, súhlas, nesúhlas) vyjadrenie vlastného názoru, stanoviska, želania, emócií). Výsledkom jazykovej kompetencie majú byť vedomosti a zručnosti zodpovedajúce iám dokumentu Spoločný európsky referenčný rámec pre vyučovanie
<b>Brief outline of the c</b> Rodina, jej formy a p Vyjadrovanie pocitov Dom, bývanie a budú Formy a dialekty v au Život v meste a na vi Kolokácie a idiomy	roblémy z a dojmov icnosť iglickom jazyku

Kolokácie a idiomy, zaužívané slovné spojenia

Prázdniny a sviatky vo svete

Životné prostredie a ekológia Výnimky zo slovosledu Frázové slovesá a ich použitie

Charakteristiky neformálneho diškurzu

#### **Recommended literature:**

www.bbclearningenglish.com

McCarthy M., O'Dell F.: English Vocabulary in Use, Upper-Intermediate. CUP, 1994. Misztal M.: Thematic Vocabulary. SPN, 1998.

Fictumova J., Ceccarelli J., Long T.: Angličtina, konverzace pro pokročilé. Barrister and Principal, 2008.

Peters S., Gráf T.: Time to practise. Polyglot, 2007.

Jones L.: Communicative Grammar Practice. CUP, 1985.

Alexander L.G.: Longman English Grammar. Longman, 1988.

#### **Course language:**

English language, B2 level according to CEFR

#### **Course assessment**

Total number of assessed students: 220

А	В	С	D	Е	FX
36.36	21.82	20.45	10.45	7.27	3.64

Provides: Mgr. Zuzana Naďová

Date of last modification: 06.02.2018

**Approved:** Guaranteeprof. RNDr. Pavol Mártonfi, PhD.Guaranteeprof. RNDr. Vladimír Zeleňák, PhD.

University: P. J.	. Šafárik Univers	ity in Košice			
Faculty: Faculty	y of Science				
<b>Course ID:</b> ÚC PFCU/03	HV/ Course na	me: Practical in	Physical Chem	istry	
Course type: I Recommended	d course-load (h er study period:	ours):			
Number of crea	lits: 4				
Recommended	semester/trimes	ter of the course	e: 5.		
Course level: I.	, II.				
Prerequisities:					
<b>Conditions for</b> Approved labor Assessment.	course completi atory reports.	on:			
Learning outco Theoretical prin experiments.		on of each technic	que and appropri	riate physical cher	nistry
chemical equilil ebulioscopy), ad Experimental ve constants, activ	verification of bria (determination dsorption. erification of theory vity coefficients,	on of enthalpy, ph pretical knowledg	nase diagrams), ge on electroche force of galvan	ermodynamics, th colligative proper emistry (conductiv nic cell, Daniell o tants).	ties (cryoscopy, ity, dissociation
W.J. Moore: Ph	dlay´s Practical F ysical Chemistry	Physical Chemistr , Longman, Lond , Oxford Univers	lon 1972	ondon 1973 rd, New York 200	2
Course languag	ge:				
Course assessm Total number of	ent f assessed studen	ts: 288			
А	В	С	D	Е	FX
70.83	22.92	5.21	0.69	0.35	0.0
Provides: RND	r. František Kaľa	vský, RNDr. And	lrea Morovská '	Turoňová, PhD.	
Date of last mo	dification: 26.02	.2018			
<b>Approved:</b> Gua PhD.	ranteeprof. RND	r. Pavol Mártonfi	i, PhD.Guarante	eeprof. RNDr. Vla	dimír Zeleňák,

University: P. J.	Šafárik Univers	ity in Košice					
Faculty: Faculty	of Science						
<b>Course ID:</b> KPPaPZ/PKŽ/15		Course name: Psychology of Everyday Life					
Course type, sco Course type: P Recommended Per week: 2 Pe Course methoo	ractice course-load (h r study period:	ours):					
Number of cred	its: 2						
Recommended s	semester/trimes	ter of the cours	<b>e:</b> 3.				
Course level: I.							
Prerequisities:							
Conditions for <b>c</b>	ourse completi	on:					
Learning outcom	mes:						
Brief outline of	the course:						
Recommended	literature:						
Course languag	e:						
Course assessme Total number of		ts: 116					
А	В	С	D	Е	FX		
43.1	14.66	30.17	8.62	2.59	0.86		
Provides: Mgr. (	Ondrej Kalina, P	hD.	1				
Date of last mod	lification: 21.08	.2017					
<b>Approved:</b> Guar PhD.	anteeprof. RND	r. Pavol Mártonf	i, PhD.Guarante	eprof. RNDr. Vla	dimír Zeleňák		

University: P. J. Šat	fárik Univers	ity in Košice					
Faculty: Faculty of	Science						
<b>Course ID:</b> ÚBEV/ PMZ/10	ÚBEV/ Course name: Comparative Animal Morphology						
Course type, scope Course type: Lect Recommended co Per week: 2 / 1 Pe Course method: p	ure / Practice urse-load (h r study perie	ours):					
Number of credits:	4						
Recommended sem	nester/trimes	ter of the cours	<b>e:</b> 1.				
Course level: I.							
Prerequisities:							
<b>Conditions for cou</b> Lectures and practic examination.	-		g of some parts	of animal body or	it derivates,		
Learning outcomes	S:						
Brief outline of the	course:						
Recommended liter Kardong, K. V., 200 Hill, New York. Pough, F. H., Janis, edition. Ruppert, E. E., Fox approach. Belmont,	)2: Vertebrate Ch. M., Heis , R. S., & Ba	ser, J. B., 2008: V rnes, R. D., 2004	/ertebrate Life.	Prentice Hall, Inc.	., 752 pp. 8th		
Course language:							
<b>Course assessment</b> Total number of ass		ts: 1782					
A	В	С	D	E	FX		
16.11	18.35	24.75	22.78	12.74	5.27		
Provides: RNDr. A	ndrej Mock, I	PhD., RNDr. And	drea Parimucho	vá, PhD.	<u>ــــــــــــــــــــــــــــــــــــ</u>		
Date of last modifie	cation: 23.02	.2018					
<b>Approved:</b> Guarant PhD.	eeprof. RND	r. Pavol Mártonf	i, PhD.Guarante	eeprof. RNDr. Vla	dimír Zeleňák,		

•	y of Science				
Course ID: ÚC POCHU/15	HV/ Course na	ame: Organic cho	emistry - Lab.		
Course type: Recommende	d course-load (h er study period:	ours):			
Number of cree	dits: 4				
Recommended	semester/trimes	ster of the cours	<b>e:</b> 3.		
Course level: I.					
Prerequisities:	ÚCHV/OCHU/0	03			
p. A 100 p. in to Grades: A: 91-3	100b, B: 81-90b, nuous evaluation	C: 71-80b, D: 61			-
Students will be	ecome familiar w	vith the basic isol	ation and munific	ation mathada w	1 •
synthetic labora theoretical know <b>Brief outline of</b> Preparation, iso	atory. Students sh wledge from the <b>the course:</b> plation, purificat	basic course of o	c laboratory tech rganic chemistry cation of organi	in simple synthe	le to apply the etic projects. Γhe emphasis i
synthetic labora theoretical know <b>Brief outline of</b> Preparation, iso on gaining the	atory. Students sh wledge from the the course:	ould master basi basic course of o ion and identific kills in synthesis	c laboratory tecl rganic chemistry cation of organic of organic cor	in simple synthe	le to apply the etic projects. Γhe emphasis i
synthetic labora theoretical know Brief outline of Preparation, iso on gaining the crystallization, Recommended	atory. Students sh wledge from the <b>the course:</b> plation, purificat experimental sk sublimation and <b>literature:</b> h experimental pu	ould master basi basic course of o ion and identific cills in synthesis thin-layer chrom	c laboratory tech rganic chemistry cation of organic of organic cor atography.	in simple synthe in simple synthe ic compounds. The npounds, distilla	le to apply the etic projects. Γhe emphasis i
synthetic labora theoretical know <b>Brief outline of</b> Preparation, iso on gaining the crystallization, <b>Recommended</b> 1. Handout with	atory. Students sh wledge from the <b>the course:</b> plation, purificat experimental sk sublimation and <b>literature:</b> h experimental punistry lectures.	ould master basi basic course of o ion and identific cills in synthesis thin-layer chrom	c laboratory tech rganic chemistry cation of organic of organic cor atography.	in simple synthe in simple synthe ic compounds. The npounds, distilla	le to apply the etic projects. Γhe emphasis i
synthetic labora theoretical know <b>Brief outline of</b> Preparation, iso on gaining the crystallization, <b>Recommended</b> 1. Handout with 2. Organic cher <b>Course languag</b>	atory. Students sh wledge from the <b>the course:</b> olation, purificat experimental sk sublimation and <b>literature:</b> h experimental punistry lectures. ge:	ould master basi basic course of o ion and identific cills in synthesis thin-layer chrom rocedures http://k	c laboratory tech rganic chemistry cation of organic of organic cor atography.	in simple synthe in simple synthe ic compounds. The npounds, distilla	le to apply the etic projects. Γhe emphasis i
synthetic labora theoretical know <b>Brief outline of</b> Preparation, iso on gaining the crystallization, <b>Recommended</b> 1. Handout with 2. Organic cher <b>Course languag</b>	atory. Students sh wledge from the <b>C the course:</b> olation, purificat experimental sk sublimation and <b>literature:</b> h experimental punistry lectures. ge: nent	ould master basi basic course of o ion and identific cills in synthesis thin-layer chrom rocedures http://k	c laboratory tech rganic chemistry cation of organic of organic cor atography.	in simple synthe in simple synthe ic compounds. The npounds, distilla	le to apply the etic projects. Γhe emphasis i
synthetic labora theoretical know <b>Brief outline of</b> Preparation, iso on gaining the crystallization, <b>Recommended</b> 1. Handout with 2. Organic cher <b>Course languag</b> <b>Course assessm</b> Total number o	atory. Students sh wledge from the sthe course: olation, purificat experimental sk sublimation and literature: h experimental punistry lectures. ge: nent f assessed studen	iould master basi basic course of o ion and identific tills in synthesis thin-layer chrom rocedures http://k	c laboratory tecl rganic chemistry cation of organic of organic cor atography.	nique and be ab in simple synth- ic compounds. T npounds, distilla pjs.sk/pochu.	le to apply the etic projects. The emphasis i ition, extraction
synthetic labora theoretical know <b>Brief outline of</b> Preparation, iso on gaining the crystallization, <b>Recommended</b> 1. Handout with 2. Organic cher <b>Course languag</b> <b>Course assessm</b> Total number o A 54.55 <b>Provides:</b> RND	atory. Students sh wledge from the <b>C the course:</b> olation, purificat experimental sk sublimation and <b>literature:</b> h experimental punistry lectures. ge: nent f assessed studen B	ts: 110 tts: 11	c laboratory tech rganic chemistry cation of organic of organic cor atography. cekule.science.up D 7.27 Dr. Mária Vilko	in simple and be abiated in simple synthematic compounds. The synthematic compounds, distillated by the synthematic compounds, distillated by the synthematic compounds. The synthematic compounds is the synthematic compounds are synthematic compounds. The synthematic compounds is the synthematic compounds is the synthematic compounds. The synthematic compounds is the synthematic compounds is the synthematic compounds is the synthematic compounds. The synthematic compounds is the synthematic compounds is the synthematic compounds is the synthematic compounds. The synthematic compounds is the synthematic compounds is the synthematic compounds is the synthematic compounds. The synthematic compounds is the synthematic compounds is the synthematic compounds is the synthematic compounds is the synthematic compounds. The synthematic compounds is the synthematic compound compounds is the synthematic compound compound	le to apply the etic projects. The emphasis i ition, extraction FX 0.0 Ladislav
synthetic labora theoretical know <b>Brief outline of</b> Preparation, iso on gaining the crystallization, <b>Recommended</b> 1. Handout with 2. Organic cher <b>Course languag</b> <b>Course assessm</b> Total number o A 54.55 <b>Provides:</b> RND Janovec, PhD., 1	atory. Students sh wledge from the <b>C the course:</b> olation, purificat experimental sk sublimation and <b>literature:</b> h experimental punistry lectures. <b>ge:</b> <b>nent</b> f assessed studen B 27.27 r. Slávka Hamul'	ts: 110 tts: 11	c laboratory tech rganic chemistry cation of organic of organic cor atography. cekule.science.up D 7.27 Dr. Mária Vilko	in simple and be abiated in simple synthematic compounds. The synthematic compounds, distillated by the synthematic compounds, distillated by the synthematic compounds. The synthematic compounds is the synthematic compounds are synthematic compounds. The synthematic compounds is the synthematic compounds is the synthematic compounds. The synthematic compounds is the synthematic compounds is the synthematic compounds is the synthematic compounds. The synthematic compounds is the synthematic compounds is the synthematic compounds is the synthematic compounds. The synthematic compounds is the synthematic compounds is the synthematic compounds is the synthematic compounds. The synthematic compounds is the synthematic compounds is the synthematic compounds is the synthematic compounds is the synthematic compounds. The synthematic compounds is the synthematic compound compounds is the synthematic compound compound	le to apply the etic projects. The emphasis i ition, extraction FX 0.0 Ladislav

University: P. J.	Šafárik Univers	ity in Košice				
Faculty: Faculty	of Science					
<b>Course ID:</b> KPPaPZ/PP/15	Course name: Positive Psychology					
	ractice   course-load (h er study period:	ours):				
Number of cred	lits: 2					
Recommended	semester/trimes	ster of the cours	<b>e:</b> 4., 6.			
Course level: I.						
Prerequisities:						
Conditions for a	course completi	on:				
Learning outco	mes:					
Brief outline of	the course:					
Recommended	literature:					
Course languag	e:					
<b>Course assessm</b> Total number of	ent `assessed studen	ts: 165				
А	В	С	D	Е	FX	
97.58	1.21	0.61	0.0	0.61	0.0	
Provides: Mgr. 3	Jozef Benka, Phl	D. et PhD.				
Date of last mod	dification: 21.08	3.2017				
<b>Approved:</b> Guar PhD.	ranteeprof. RND	r. Pavol Mártonf	i, PhD.Guarante	eprof. RNDr. Vla	dimír Zeleňák,	

University: P. J. Š	afárik Univers	ity in Košice					
Faculty: Faculty o	f Science						
<b>Course ID:</b> KPPaPZ/PUDB/15		Course name: Drug Addiction Prevention in University Students					
Course type, scop Course type: Pra Recommended c Per week: 2 Per Course method:	ctice ourse-load (h study period:	ours):					
Number of credits	s: 2						
Recommended set	mester/trimes	ter of the cours	<b>e:</b> 3., 5.				
Course level: I.							
Prerequisities:							
Conditions for co	urse completi	on:					
Learning outcom	es:						
Brief outline of th	e course:						
Recommended lit	erature:						
Course language:							
Course assessmen Total number of as		ts: 256					
A	В	С	D	E	FX		
77.34	20.31	2.34	0.0	0.0	0.0		
Provides: prof. Ph	Dr. Ol'ga Oros	ová, CSc., Mgr.	Marta Dobrowo	lska Kulanová, P	hD.		
Date of last modif	fication: 21.08	.2017					
<b>Approved:</b> Guaran PhD.	nteeprof. RND	r. Pavol Mártoni	ñ, PhD.Guarante	eprof. RNDr. Vla	dimír Zeleňák,		

University: P. J. Ša	afárik Univers	ity in Košice					
Faculty: Faculty of	f Science						
<b>Course ID:</b> KPE/ Pg/15	Course na	Course name: Pedagogy					
Course type, scope Course type: Lec Recommended co Per week: 2 Per s Course method: 1	ture ourse-load (h study period:	ours):					
Number of credits	: 2			_			
Recommended ser	nester/trimes	ster of the cours	<b>e:</b> 3., 5.				
Course level: I.							
Prerequisities:							
Conditions for cou	ırse completi	on:					
Learning outcome	es:						
Brief outline of the	e course:						
Recommended lite	erature:						
Course language:							
<b>Course assessmen</b> Total number of as		ts: 406					
A	В	С	D	Е	FX		
20.94	18.97	26.11	19.46	13.55	0.99		
Provides: Mgr. Kat	tarína Petríko	vá, PhD.	1	·	1		
Date of last modifi	ication: 23.08	3.2017					
<b>Approved:</b> Guaran PhD.	teeprof. RND	r. Pavol Mártonf	ñ, PhD.Guarante	eprof. RNDr. Vla	dimír Zeleňák,		

University: P. J.	Šafárik Univers	ity in Košice			
Faculty: Faculty	of Science				
<b>Course ID:</b> KPPaPZ/Ps/15	Course na	me: Psychology	r		
	ecture course-load (h r study period:	ours):			
Number of cred	its: 2				
Recommended s	semester/trimes	ster of the cours	<b>e:</b> 1., 3., 5.		
Course level: I.					
Prerequisities:					
Conditions for a	ourse completi	on:			
Learning outcom	mes:				
Brief outline of	the course:				
Recommended	literature:				
Course languag	e:				
Course assessme Total number of		ts: 318			
А	В	С	D	Е	FX
16.04	11.01	24.53	23.9	20.75	3.77
<b>Provides:</b> prof. I et PhD.	PhDr. Ol'ga Oros	sová, CSc., PhDr	. Anna Janovská	PhD., Mgr. Joze	ef Benka, PhD.
Date of last mod	lification: 21.08	3.2017			
<b>Approved:</b> Guar PhD.	anteeprof. RND	r. Pavol Mártonf	i, PhD.Guarantee	eprof. RNDr. Vla	adimír Zeleňák,

University: P. J.	Šafárik Univers	ity in Košice					
Faculty: Faculty	of Science						
<b>Course ID:</b> ÚCI SCHM/14	IV/   Course name: Chemistry						
Course type, sco Course type: Recommended Per week: Per Course method	l course-load (h study period:						
Number of cred	lits: 1						
Recommended	semester/trimes	ster of the cours	e:				
Course level: I.							
Prerequisities: ( ACHU/03 and Ú OCHU/03	UCHV/VCHU/ CHV/BCHU/03	10 or UCHV/VC and ÚCHV/FCI	HU/14 or UCHV HU/10 and ÚCH	//VCHU/15) and V/ANCHU/03 ar	l UCHV/ nd ÚCHV/		
Conditions for	course completi	on:					
Learning outco	mes:						
Brief outline of	the course:						
Recommended	literature:						
Course languag	je:						
Course assessm Total number of	ent assessed studen	ts: 86					
А	В	С	D	Е	FX		
23.26	38.37	22.09	10.47	5.81	0.0		
Provides:			1	·			
Date of last mo	dification: 26.02	2.2018					
<b>Approved:</b> Guar PhD.	ranteeprof. RND	r. Pavol Mártonf	i, PhD.Guarante	eprof. RNDr. Vla	adimír Zeleňák		

University: P. J. Š	afárik Univers	ity in Košice			
Faculty: Faculty o	f Science				
Course ID: KPO/ SPKVV/15	Course na	me: Social and	Political Context	of Education	
Course type, scop Course type: Lec Recommended c Per week: 2 Per Course method:	cture ourse-load (h study period:	ours):			
Number of credits	s: 2				
Recommended set	mester/trimes	ster of the cours	<b>e:</b> 4., 6.		
Course level: I.					
Prerequisities:					
Conditions for co	urse completi	on:			
Learning outcome	es:				
Brief outline of th	e course:				
Recommended lit	erature:				
Course language:					
<b>Course assessmen</b> Total number of as	-	ts: 11			
A	В	С	D	Е	FX
9.09	0.0	45.45	36.36	9.09	0.0
Provides: Dr.h.c. p	orof. PhDr. Ma	urcela Gbúrová, (	CSc.	·	
Date of last modif	<b>ication:</b> 23.08	3.2017			
<b>Approved:</b> Guaran PhD.	nteeprof. RND	r. Pavol Mártonf	i, PhD.Guarantee	eprof. RNDr. Vla	dimír Zeleňák,

University: P. J. Šafa	árik Univers	ity in Košice			
Faculty: Faculty of S	Science				
<b>Course ID:</b> ÚBEV/ SVK/01	Course na	me: Student Sci	entific Conference	се	
Course type, scope Course type: Recommended cou Per week: Per stu Course method: pr	ırse-load (h dy period:				
Number of credits:	4				
Recommended sem	ester/trimes	ter of the cours	<b>e:</b> 4., 6.		
Course level: I., II.					
Prerequisities:					
Conditions for cour	se completi	on:			
Learning outcomes					
Brief outline of the	course:				
<b>Recommended liter</b>	ature:				
Course language:					
<b>Course assessment</b> Total number of asse	essed studen	ts: 258			
A	В	С	D	Е	FX
100.0	0.0	0.0	0.0	0.0	0.0
Provides:			1	1	1
Date of last modific	ation: 23.02	.2018			
<b>Approved:</b> Guarante PhD.	eprof. RND	r. Pavol Mártonf	i, PhD.Guarantee	eprof. RNDr. Vla	dimír Zeleňák,

University: P. J. Š	Safárik Univers	ity in Košice			
Faculty: Faculty	of Science				
Course ID: ÚCH SVK/00	V/ Course na	ame: Students Sc	eientific Conferen	nce (Presentation	)
Course type, scop Course type: Recommended Per week: Per s Course method:	course-load (h study period:				
Number of credit	ts: 4				
Recommended se	emester/trimes	ster of the cours	e:		
Course level: I., I	I.				
Prerequisities:					
Conditions for co	ourse completi	on:			
Learning outcom	ies:				
Brief outline of tl	he course:				
Recommended li	terature:				
Course language	•				
<b>Course assessmen</b> Total number of a	-	ts: 35			
A	В	С	D	Е	FX
100.0	0.0	0.0	0.0	0.0	0.0
Provides:				·	
Date of last modi	fication: 26.02	2.2018			
<b>Approved:</b> Guara PhD.	nteeprof. RND	r. Pavol Mártonf	i, PhD.Guarante	eprof. RNDr. Vla	dimír Zeleňák,

University: P. J. Šafá	rik University in Košice						
Faculty: Faculty of S	Science						
<b>Course ID:</b> ÚBEV/ TCB1/03	Course name: Fieldworks from Botany						
Course type, scope a Course type: Practi Recommended cou Per week: Per stud Course method: pr Number of credits: 2	ce rse-load (hours): ly period: 5d esent						
		se: 2.					
Course level: I.							
Prerequisities:							
Conditions for cours	se completion:						
	r identification and determin	nation of common central-europaean plants.					
Brief outline of the of Plant identification in		etermination. Floristic records.					
Kubát K. (ed.): Klíč Marhold K. a Hindál vascular and vascula	M.: Veľký kľúč na určovani ke květeně České republiky	a vyšších rastlín Slovenska. Checklist of non- , Bratislava 1998.					
Course language:							
<b>Course assessment</b> Total number of asse	essed students: 1090						
	abs	n					
	99.91	0.09					
1	r. Pavol Mártonfi, PhD., pro	of. RNDr. Martin Bačkor, DrSc., Mgr. Vladislav					
Kolarčik, PhD.							

Approved: Guaranteeprof. RNDr. Pavol Mártonfi, PhD.Guaranteeprof. RNDr. Vladimír Zeleňák, PhD.

University: P. J. Šafá	rik University in Košice					
Faculty: Faculty of S	Science					
Course ID: ÚBEV/ TCZ/03	Course name: Fieldwork from zoology					
Course type, scope a Course type: Practi Recommended cou Per week: Per stuc Course method: pro-	ce rse-load (hours): ly period: 5d					
Number of credits: 2	2					
Recommended seme	ester/trimester of the cours	<b>e:</b> 4.				
Course level: I.						
Prerequisities:						
Conditions for cours	se completion:					
Learning outcomes: Practical observation	of morphology of vertebrat	es.				
	ogenetic relationships of ve	ertebrate. Review of important groups of fishes, vation, and laboratory work.				
<b>Recommended liter</b>	ature:					
Course language:						
<b>Course assessment</b> Total number of asse	ssed students: 790					
	abs	n				
	99.24 0.76					
<b>Provides:</b> RNDr. Pet PhD.	er Ľuptáčik, PhD., doc. RNI	Dr. Ľubomír Panigaj, CSc., RNDr. Andrej Mock,				
Date of last modification	ation: 23.02.2018					
Approved: Guarante PhD.	eprof. RNDr. Pavol Mártonf	i, PhD.Guaranteeprof. RNDr. Vladimír Zeleňák,				

University: P. J. Ša	afárik Univers	ity in Košice					
Faculty: Faculty of	f Science						
Course ID: KPE/ TVE/08	Course na	Course name: Theory of Education					
Course type, scope Course type: Prac Recommended co Per week: 2 Per s Course method:	ctice ourse-load (h study period:	ours):					
Number of credits	: 2						
Recommended ser	mester/trimes	ter of the cours	<b>e:</b> 4., 6.				
Course level: I.							
Prerequisities:							
Conditions for cou	urse completi	on:					
Learning outcome	es:						
Brief outline of th	e course:						
Recommended lite	erature:						
Course language:							
<b>Course assessmen</b> Total number of as		ts: 378					
A	В	С	D	Е	FX		
27.25	36.77	23.81	7.41	1.85	2.91		
Provides: Mgr. Ka	tarína Petríko	vá, PhD.		1			
Date of last modif	ication: 23.08	.2017					
<b>Approved:</b> Guaran PhD.	nteeprof. RND	r. Pavol Mártonf	i, PhD.Guarantee	eprof. RNDr. Vla	dimír Zeleňák,		

University	P. J. Šafái	rik University i	n Košice				
Faculty: Fa	aculty of So	cience					
<b>Course ID</b> TVa/11	: ÚTVŠ/	Course name	: Sports Acti	vities I.			
Course ty Recomme Per week	pe: Practic nded cour	rse-load (hours dy period: 28					
Number of	credits: 2						
Recommer	ided seme	ster/trimester	of the cours	<b>e:</b> 1.			
Course lev	el: I., I.II.,	II.					
Prerequisi	ties:						
Conditions	for course	e completion: completion: articipation in c	classes.				
relationshi	physical co p of studen	ondition and pe ts to the selecto			1		g the
University floorball, y tennis, spo In the first and particu physical co Last but no means of a In addition physical co the premise	ne of the co optional su provides f yoga, pilate rts for unfi two semes larities of i ondition, co ot least, the special pro- to these se lucation tra- es of the fac	burse: ubject, the Inst for students the es, swimming, t persons, stree sters of the firs ndividual sport oordination abi important role ogram of medic sports, the Inst inings with an a culty or Univers	e following s body-buildin tball, tennis, it level of ed ts, motor skil ilities, physic of sports act cal physical o itute offers to attractive pro	ports activiti ag, indoor for and volleyba ucation stude ls, game activities cal performativities is to e education to for those wh gram and org	ies: aerobics, otball, self-de all. ents will mas vities, they wince, and mot eliminate swin influence and o are interest ganises variou	basketball, efence and l ster basic ch ill improve l tor performa mming illite mitigate ur ted winter a us competitio	badminton karate, table aracteristics evel of their ince fitness gracy and by fitness. and summer ons, either a
Recommer	nded litera	ture:					
Course lan	guage:						
	essment						
Course ass		sed studenter 1	1672				
Course ass		abs-B	1672 abs-C	abs-D	abs-E	n	neabs

**Provides:** Mgr. Peter Bakalár, PhD., Mgr. Dana Dračková, PhD., Mgr. Agata Horbacz, PhD., Mgr. Dávid Kaško, Mgr. Zuzana Küchelová, PhD., doc. PaedDr. Ivan Uher, PhD., Mgr. Marek Valanský, prof. RNDr. Stanislav Vokál, DrSc., Mgr. Marcel Čurgali, Ing. Iveta Cimboláková, PhD.

Date of last modification: 18.08.2017

Approved: Guaranteeprof. RNDr. Pavol Mártonfi, PhD.Guaranteeprof. RNDr. Vladimír Zeleňák, PhD.

	COUR	RSE INFORM	MATION LI	ETTER		
University: P. J. Šaf	árik University	in Košice				
Faculty: Faculty of	Science					
Course ID: ÚTVŠ/ TVb/11	Course name	: Sports Acti	vities II.			
Course type, scope Course type: Pract Recommended cou Per week: 2 Per st Course method: pu	ice 1 <b>rse-load (hour</b> 1dy period: 28					
Number of credits:	2					
Recommended sem	ester/trimester	of the cours	e: 2.			
Course level: I., I.II	, II.					
Prerequisities:						
<b>Conditions for cour</b> Conditions for cours Final assessment an	se completion:		ses - min. 759	%.		
Learning outcomes: Learning outcomes: Increasing physical relationship of stude	condition and p			-		g the
Brief outline of the Brief outline of the Within the optional University provides floorball, yoga, pila tennis, sports for un In the first two sem and particularities of physical condition, Last but not least, th means of a special p In addition to these physical education to the premises of the fa	course: subject, the Inst for students the tes, swimming, fit persons, streed esters of the first individual sport coordination ab e important role rogram of medit sports, the Inst rainings with an aculty or Univer	e following s body-buildir etball, tennis, st level of ed ts, motor skil ilities, physic e of sports ac cal physical titute offers attractive pro	sports activiting, indoor for and volleyba ucation study ls, game activities is to ever tivities is to ever education to for those who gram and org	ies: aerobics otball, self-d all. ents will ma- vities, they w nce, and mo eliminate swi influence an o are interes ganises vario	, basketball, lefence and l ster basic ch vill improve l tor performa imming illite d mitigate un sted winter a us competitio	badminton, karate, table aracteristics evel of their ance fitness. eracy and by hfitness. and summer ons, either at
<b>Recommended</b> liter	ature:					
Course language:						
Course assessment						
Total number of ass abs abs-A		10971 abs-C	abs-D	abs-E	n	neabs
			1		n 10.12	
85.37 0.57	0.02	0.0	0.0	0.05	10.13	3.86

**Provides:** Mgr. Peter Bakalár, PhD., Mgr. Dana Dračková, PhD., Mgr. Agata Horbacz, PhD., Mgr. Dávid Kaško, Mgr. Zuzana Küchelová, PhD., doc. PaedDr. Ivan Uher, PhD., Mgr. Marek Valanský, prof. RNDr. Stanislav Vokál, DrSc., Mgr. Marcel Čurgali, Ing. Iveta Cimboláková, PhD.

Date of last modification: 18.08.2017

Approved: Guaranteeprof. RNDr. Pavol Mártonfi, PhD.Guaranteeprof. RNDr. Vladimír Zeleňák, PhD.

University: P.	J. Šafárik	University i	n Košice				
Faculty: Facul	ty of Scie	ence					
<b>Course ID:</b> ÚT TVc/11	ΓVŠ/ C	ourse name:	Sports Acti	vities III.			
Course type, s Course type: Recommende Per week: 2 I Course methe	Practice ed course Per study	e-load (hours period: 28					
Number of cre	edits: 2						
Recommended	d semeste	er/trimester	of the cours	<b>e:</b> 3.			
Course level:	I., I.II., II.						
Prerequisities	:						
Conditions for	r course c	completion:					
Learning outc	omes:						
Brief outline o	of the cou	rse:					
Recommended	d literatu	re:					
Course langua	nge:						
Course assess Total number of		ed students: 6	910				
abs	abs-A	abs-B	abs-C	abs-D	abs-E	n	neabs
89.84	0.04	0.0	0.0	0.0	0.03	4.23	5.86
<b>Provides:</b> Mgr Horbacz, PhD. Mgr. Marek Va	, Mgr. Dá	vid Kaško, N	Agr. Zuzana	Küchelová, I	PhD., doc. Pa	edDr. Ivan U	Jher, PhD.,
Date of last m	odificatio	on: 18.08.201	17				
Approved: Gu	aranteepr	of. RNDr. Pa	vol Mártonf	i, PhD.Guara	anteeprof. RI	NDr. Vladim	ír Zeleňák,

PhD.

University:	P. J. Šafárik	University i	n Košice				
Faculty: Fac	culty of Scie	ence					
Course ID: TVd/11	ÚTVŠ/ C	ourse name:	Sports Acti	vities IV.			
Per week: 2 Course me	e: Practice ded course 2 Per study thod: prese	e-load (hours v period: 28					
Number of o	credits: 2						
Recommend	led semeste	er/trimester	of the cours	<b>e:</b> 4.			
Course level	<b>I:</b> I., I.II., II	•					
Prerequisiti	es:						
Conditions f	for course	completion:					
Learning ou	itcomes:						
Brief outline	e of the cou	irse:					
Recommend	led literatu	ire:					
Course lang	uage:						
Course asses Total numbe		ed students: 5	045				
abs	abs-A	abs-B	abs-C	abs-D	abs-E	n	neabs
85.09	0.3	0.04	0.0	0.0	0.0	6.82	7.75
Provides: M Horbacz, Phl Mgr. Marek	D., Mgr. Dá	ivid Kaško, N	Agr. Zuzana	Küchelová, I	PhD., doc. Pa	edDr. Ivan U	Jher, PhD.,
Date of last	modificatio	on: 18.08.201	17				
Approved: (	Guaranteep	rof. RNDr. Pa	avol Mártonf	i, PhD.Guara	anteeprof. RI	NDr. Vladim	ír Zeleňák,

PhD.

University: P.	J. Šafárik	University in	Košice
Chiver Sity 11.	J. Dururin	Oniversity in	

Faculty: Faculty of Science

**Course ID:** ÚCHV/ **Course name:** Introduction to Environmental Chemistry UECH/03

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14

Number of credits: 5

Course method: present

**Recommended semester/trimester of the course:** 3.

Course level: I., II.

**Prerequisities:** 

**Conditions for course completion:** 

Oral examination

#### Learning outcomes:

Introduction to topics in environmental chemistry and basic procedures applied for environmental protection.

#### **Brief outline of the course:**

Introduction to Environmental Chemistry

Chemical aspects of pollution and environmental problems. Composition and behavior of the atmosphere. Energy balance of the Earth and climate changes. Principles of photochemistry, photoprocesses in the atmosphere. Petroleum, hydrocarbons and coal (characteristics, sources and environmental pollution). Soaps, polymers and synthetic surfactants. Haloorganics and pesticides. Environmental chemistry of some important elements (C, N, S, P, halogens, biologically important metals ...). Environmental chemistry in aqueous media. Aqueous systems, parameters, cycles and their protection. The Earth's crust (rocks, minerals, soils). Natural and artificial radioactivity, utilization. Energy and energy sources (fossil fuels, nuclear, geothermal, solar energy, wind and water energy). Solid waste disposal and recycling.

#### **Recommended literature:**

1. Gary W. van Loon, Stephen J. Duffy : Environmental Chemistry - A Global Perspective, Oxford University Press, Oxford 2003

2. R.A. Bailey, H.M. Clark, J.P. Ferris, S. Krause, R.L. Strong : Chemistry of the Environment, Academic Press, San Diego 2002

- 3. G. Schwedt: The Essential Guide to Environmental Chemistry, Wiley and Sons, London 2001
- 4. R.N. Reeve, J.D. Barnes: General Environmental Chemistry, Wiley, London 1994

5. G. Burton, J. Holman, G. Pilling, D. Waddington: Chemical Storylines, Heinemann, Oxford, London 1994

6. www

#### Course language:

#### **Course assessment**

Total number of assessed students: 209

А	В	С	D	Е	FX
48.8	20.57	15.79	8.61	6.22	0.0
Provides: doc. 1	RNDr. Andrea St	raková Fedorkov	rá, PhD.		
Date of last mo	dification: 21.09	0.2017			
<b>Approved:</b> Gua PhD.	ranteeprof. RND	r. Pavol Mártonf	i, PhD.Guarantee	eprof. RNDr. Vla	dimír Zeleňák,

University: P. J. Šafá	rik University in Ko	šice
Faculty: Faculty of S	cience	
Course ID: Dek. PF UPJŠ/USPV/13	Course name: Intr	oduction to Study of Sciences
Course type, scope a Course type: Lectur Recommended cou Per week: Per stud Course method: pre	re / Practice rse-load (hours): ly period: 12s / 3d	
Number of credits: 2	2	
Recommended seme	ster/trimester of th	e course: 1.
Course level: I.		
Prerequisities:		
Conditions for cours	e completion:	
Learning outcomes:		
Brief outline of the c	ourse:	
Recommended litera	iture:	
Course language:		
<b>Course assessment</b> Total number of asse	ssed students: 1356	
	abs	n
	88.86	11.14
Provides:		
Date of last modifica	tion: 19.02.2018	
<b>Approved:</b> Guarantee PhD.	eprof. RNDr. Pavol	Mártonfi, PhD.Guaranteeprof. RNDr. Vladimír Zeleňák,

University: P. J. S	Šafárik Univers	ity in Košice			
Faculty: Faculty	of Science				
Course ID: ÚBE VB1/01	V/ Course na	me: General bo	tany		
Course type, sco					
Course type: Le Recommended					
Per week: 3 / 2	```	,			
Course method	v 1				
Number of credi	<b>ts:</b> 6				
Recommended se	emester/trimes	ster of the cours	se: 2.		
Course level: I.					
Prerequisities: Ú	BEV/CYT1/02	or ÚBEV/CYT	1/15		
Conditions for co	ourse completi	on:			
Learning outcom This subject enab to enhance studer Brief outline of t	les to understant's ability to de		1	,	•
The structure and and organization, that are necessar organs and functi	Plant reproduce y for understar	ction and ground ding of relation	ling in embryold	ogy. Basic inform	nation and terms
Recommended li	terature:				
Course language	:				
<b>Course assessme</b> Total number of a		ts: 1777			
A	В	С	D	E	FX
18.91	22.62	27.41	18.01	9.23	3.83
Provides: prof. R	NDr. Pavol Má	rtonfi, PhD., Mg	gr. Vladislav Kol	arčik, PhD.	<u>.</u>
Date of last modi	ification: 23.02	2.2018			
<b>Approved:</b> Guara PhD.	inteeprof. RND	r. Pavol Márton	fi, PhD.Guarante	eeprof. RNDr. Vla	adimír Zeleňák,

University: P. J.	. Šafárik Univers	sity in Košice			
Faculty: Faculty	y of Science				
Course ID: ÚC VCHU/15	HV/ Course na	ame: General Ch	emistry		
Course type: I Recommended	ope and the met Lecture / Practice d course-load (h 2 Per study peri d: present	e ours):			
Number of crea	lits: 7				
Recommended	semester/trimes	ster of the cours	<b>e:</b> 1.		
Course level: I.					
Prerequisities:	ÚCHV/CHV1/9	9			
		on: e end of the seme	ester.		
-	ents with knowle	edge of atoms and ties of elements		r electronic struc	ture, theories of
periodicity and intermolecular Solutions. Cher	ed in chemistry its effect on t interactions. Che mical equilibriu	t. Atoms – mod the properties of mical structure a m. Basis of che ons. Electrochem	f elements, radi and physical prop emical thermody	oactivity. Chem perties of matter.	ical bonds and State of matter.
	nes L.: Chemical	Principles, 2nd e ry, 2nd ed., McG			
Course languag	ge:				
Course assessm Total number of	ent f assessed studen	its: 150			
А	В	С	D	Е	FX
16.0	24.0	39.33	12.67	8.0	0.0
Provides: prof.	RNDr. Vladimír	Zeleňák, PhD.		<u>.                                    </u>	1
Date of last mo	dification: 26.02	2.2018			
<b>Approved:</b> Gua PhD.	ranteeprof. RND	r. Pavol Mártonf	i, PhD.Guarantee	eprof. RNDr. Vla	adimír Zeleňák,

University: P. J. Ša	fárik Univers	ity in Košice			
<b>Faculty:</b> Faculty of					
Course ID: ÚBEV/ VEK1/03		me: Introduction	n to Ecology		
Course type, scope Course type: Lect Recommended co Per week: 2 Per st Course method: p Number of credits:	ure urse-load (h tudy period: resent	ours):			
Recommended sem		ster of the cours	e:		
Course level: I., II.					
Prerequisities:					
Conditions for cou	rse completi	on:			
Learning outcomes Fundamental param		ations in ecologi	cal science.		
Brief outline of the Ecological factors a on individuals (mo ecosystems (impact	and relations rphological a	adaptations, beha	vioral reactions		•
Recommended lite Begon, M., Harper, Blackwell Sci. Publ	J. L., Towns	end, C. L.: Ecolo	ogy: individuals,	populations, and	communities.
Course language:					
<b>Course assessment</b> Total number of ass	essed studen	ts: 1522			
A	В	С	D	Е	FX
19.58	15.37	24.9	18.46	12.75	8.94
Provides: RNDr. N	atália Raschr	nanová, PhD.		·	
Date of last modified	cation: 23.02	2.2018			
<b>Approved:</b> Guarant PhD.	eeprof. RND	r. Pavol Mártonf	i, PhD.Guarante	eprof. RNDr. Vla	dimír Zeleňák,

University: P. J.	Šafárik Univers	ity in Košice			
Faculty: Faculty	of Science				
<b>Course ID:</b> KFa VKFV/07	DF/ Course na Introductio		ppics in Philosop	hy of Education (	General
Course type, sco Course type: Recommended Per week: Per Course methoo	- l course-load (h r study period:				
Number of cred	lits: 2				
Recommended	semester/trimes	ster of the cours	e: 3., 5.		
Course level: I.					
Prerequisities: 1	KFaDF/DF1/05				
Conditions for o	course completi	on:			
Learning outco	mes:				
Brief outline of	the course:				
Recommended	literature:				
Course languag	je:				
Course assessm Total number of	ent assessed studen	ts: 0			
А	В	С	D	E	FX
0.0	0.0	0.0	0.0	0.0	0.0
Provides: doc. P	PhDr. Pavol Thol	t, PhD., mim. pr	of.		
Date of last mod	dification: 23.08	3.2017			
<b>Approved:</b> Guar PhD.	ranteeprof. RND	r. Pavol Márton	ñ, PhD.Guarante	eprof. RNDr. Vla	ıdimír Zeleňák,

University: P. J.	Šafárik Univers	ity in Košice			
Faculty: Faculty	of Science				
<b>Course ID:</b> ÚBI ZO1/03	EV/ Course na	ame: Zoology I			
Recommended	ecture / Practice course-load (h Per study peri	ours):			
Number of cred	lits: 5				
Recommended	semester/trimes	ster of the cours	e: 3.		
Course level: I.					
Prerequisities: 1	ÚBEV/PMZ/10				
Conditions for a	course completi	on:			
Learning outco Basis of Inverte Phylogenetic rel	brata taxonomy-	Importance and	function of chos	en individual taxo	ons.
• • •	hology and deve elminthes, Nem			vertebrates – espe a, Arthropoda, 1	•
Recommended	literature:				
Course languag	e:				
Course assessm Total number of		ts: 1043			
А	В	С	D	Е	FX
7.96	15.34	21.67	20.42	25.02	9.59
Provides: doc. F	NDr. Ľubomír I	Panigaj, CSc., RN	NDr. Peter Ľuptá	čik, PhD.	
Date of last mod	dification: 23.02	2.2018			
<b>Approved:</b> Guar PhD.	ranteeprof. RND	r. Pavol Mártonf	i, PhD.Guarantee	eprof. RNDr. Vla	dimír Zeleňák,

University: P. J.	Šafárik Univers	ity in Košice			
Faculty: Faculty	of Science				
Course ID: ÚB ZO1/15	EV/ Course na	ame: Zoology I			
Course type, sc	ope and the me	thod:			
Course type: I	Lecture / Practice	2			
	l course-load (h	/			
	Per study peri	od: 28 / 28			
Course metho					
Number of cred	lits: 4				
Recommended	semester/trimes	ster of the cours	se: 3.		
Course level: I.					
Prerequisities:	ÚBEV/PMZ/10				
Conditions for	course completi	on:			
chosen individu	al taxons. Phylo	including taxono genetic relations		ozoa. Importance	and function o
<b>2</b> / 1	hology and deve elminthes, Nem	1 1	• 1	vertebrates – espe la, Arthropoda,	
Recommended	literature:				
Course languag	ge:				
Course assessm Total number of	ent assessed studen	ts: 208			
А	В	С	D	E	FX
6.73	19.23	20.19	26.44	19.71	7.69
0.75					
	RNDr. Ľubomír I	Panigaj, CSc., R	NDr. Peter Ľuptá	čik, PhD.	
		0.54	NDr. Peter Ľuptá	čik, PhD.	

PhD.

Faculty: Faculty	· · · · · · · · · · · · · · · · · · ·				
Course ID: ÚB ZOG1/03	EV/ Course na	ame: Zoogeograp	bhy		
Course type: I Recommended	ope and the met Lecture / Practice I course-load (h 2 Per study perio d: present	ours):			
Number of cred	lits: 6				
Recommended	semester/trimes	ster of the cours	e:		
Course level: I.,	, II.				
Prerequisities:					
	en test.	to selected topic.			
Learning outco	mos.				
The main goal of animals on the I	of the subject is t	o get knowledge phic regionalizat history.			
The main goal of animals on the I on the faunal dia <b>Brief outline of</b> This course will processes that in information on interaction with distributions. Th	of the subject is the Earth, zoogeograstribution in the Earth the course: <b>the course:</b> I review our curnfluence distribution the historical and the new ironmental the course will employed by the subject of the the course will employed by the subject of the the course will employed by the the the course will employed by the	phic regionalizat	ion of the Earth's ng of the pattern and their attribut y, genetics, and inental drift, cli ve and analytical	s surface and hu s of animal dist es. Zoogeograp physiology of a mate) in regula approaches use	man influence ribution and the hy will integrate nimals and their ting geographic ful in hypothesis
The main goal of animals on the I on the faunal dif <b>Brief outline of</b> This course will processes that in information on interaction with distributions. The testing in zooge conservation). <b>Recommended</b> Buchar, J., 1983 Darlington, P.J., Lomolino M.V.,	of the subject is the Earth, zoogeograsstribution in the Earth zoogeograsstribution in the Earth zoogeograms and the course of the course will employ and will <b>literature:</b> B: Zoogeografie., 1998: Zoogeografie, Brown J.H., Rice	phic regionalizat history. rent understandir tions of species d current ecolog processes (cont phasize descripti l illustrate applie	ion of the Earth's ng of the pattern and their attribut y, genetics, and inental drift, cli ve and analytical d aspects of zoos aphical distributi Biogeography. S	s surface and hu s of animal dist res. Zoogeograp physiology of a mate) in regula approaches use geography (e.g.	man influence ribution and the hy will integrate nimals and their ting geographic ful in hypothesis refuge design in Krieger, USA es, 1-845
The main goal of animals on the I on the faunal dif <b>Brief outline of</b> This course will processes that in information on interaction with distributions. The testing in zooge conservation). <b>Recommended</b> Buchar, J., 1983 Darlington, P.J., Lomolino M.V., Plesník, P., Zatk	of the subject is the Earth, zoogeograsstribution in the Earth zoogeograsstribution in the Earth zoogeograms and the course of the course will emplement and the historical and the expression of the course will emplement and the course will emplem	phic regionalizat history. rent understandir ttions of species d current ecolog processes (cont phasize descripti l illustrate applie SPN Praha raphy: The geogr ddle B. R., 2005:	ion of the Earth's ng of the pattern and their attribut y, genetics, and inental drift, cli ve and analytical d aspects of zoos aphical distributi Biogeography. S	s surface and hu s of animal dist res. Zoogeograp physiology of a mate) in regula approaches use geography (e.g.	man influence ribution and the hy will integrate nimals and their ting geographic ful in hypothesis refuge design in Krieger, USA es, 1-845
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Provides: prof. RNDr. Ľubomír Kováč, CSc.

Date of last modification: 23.02.2018

Approved: Guaranteeprof. RNDr. Pavol Mártonfi, PhD.Guaranteeprof. RNDr. Vladimír Zeleňák, PhD.

University: P. J.	Šafárik Univers	ity in Košice				
Faculty: Faculty of Science						
Course ID: ÚBI ZOO1/03						
Recommended	Lecture / Practice l course-load (h 2 Per study peri	ours):				
Number of cred	lits: 5					
Recommended semester/trimester of the course: 4.						
Course level: I.						
Prerequisities:	ÚBEV/PMZ/10					
Conditions for	course completi	on:				
Learning outco Fundamental in:		onomy and morp	hology of verteb	orates		
•		-	rtebrate. Reviev	v of important g	roups of fishes,	
Recommended	literature:					
Course languag	je:					
Course assessm Total number of	ent assessed studen	ts: 910				
А	В	С	D	Е	FX	
22.75	27.36	19.23	16.48	10.11	4.07	
Provides: doc. F	RNDr. Marcel Ul	nrin, PhD., RND	: Peter Ľuptáčik	, PhD.	l	
Date of last mo	dification: 23.02	2.2018				
<b>Approved:</b> Gua PhD.	ranteeprof. RND	r. Pavol Mártonf	i, PhD.Guarante	eprof. RNDr. Vla	dimír Zeleňák,	

	ee	UNSE INFORM				
University: P. J.	Šafárik Univers	sity in Košice				
Faculty: Faculty	of Science					
Course ID: ÚB ZOO1/15	EV/ Course name: Zoology II					
Recommended	Lecture / Practice l course-load (h 2 Per study peri	e ours):				
Number of cred	lits: 4					
Recommended	semester/trimes	ster of the cours	<b>e:</b> 4.			
Course level: I.						
Prerequisities:	ÚBEV/PMZ/10					
Conditions for	course completi	on:				
Learning outco Fundamental in		onomy and morp	bhology of verte	brates		
•		-	ertebrate. Review	w of important g	roups of fishes	
Recommended	literature:					
Course languag	ge:					
Course assessm Total number of	ent fassessed studen	ıts: 142				
А	В	С	D	Е	FX	
0.7	19.72	17.61	21.13	26.06	14.79	
Provides: doc. F	RNDr. Marcel Ul	hrin, PhD., RND	r. Peter Ľuptáčik	, PhD.	1	
Date of last mo	dification: 23.02	2.2018	•			
<b>Approved:</b> Gua PhD.	ranteeprof. RND	r. Pavol Mártonf	i, PhD.Guarante	eprof. RNDr. Vla	dimír Zeleňák,	

University: P. J. Šafá	University: P. J. Šafárik University in Košice					
Faculty: Faculty of Science						
<b>Course ID:</b> ÚTVŠ/ ÚTVŠ/CM/13	Course name: Seaside Aer	obic Exercise				
Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: Per study period: 36s Course method: present						
Number of credits: 2						
Recommended semester/trimester of the course:						
Course level: I., II.						
Prerequisities:						
Conditions for course completion: Conditions for course completion: Attendance						
Learning outcomes: Learning outcomes: Students will be provided an overview of possibilities how to spend leisure time in seaside conditions actively and their skills in work and communication with clients will be improved. Students will acquire practical experience in organising the cultural and art-oriented events, with the aim to improve the stay and to create positive experiences for visitors.						
<ul> <li>Brief outline of the course:</li> <li>Brief outline of the course:</li> <li>1. Basics of seaside aerobics</li> <li>2. Morning exercises</li> <li>3. Pilates and its application in seaside conditions</li> <li>4. Exercises for the spine</li> <li>5. Yoga basics</li> <li>6. Sport as a part of leisure time</li> <li>7. Application of projects of productive spending of leisure time for different age and social groups (children, young people, elderly)</li> <li>8. Application of seaside cultural and art-oriented activities in leisure time</li> </ul>						
Recommended literature:						
Course language:						
Course assessment Total number of assessed students: 33						
	abs	n				
	12.12 87.88					
Provides: Mgr. Alena Buková, PhD., Mgr. Agata Horbacz, PhD.						
Date of last modifica	Date of last modification: 18.08.2017					

**Approved:** Guaranteeprof. RNDr. Pavol Mártonfi, PhD.Guaranteeprof. RNDr. Vladimír Zeleňák, PhD.