University: P. J.	Šafárik Univers	ity in Košice			
Faculty: Faculty	of Science				
Course ID: KFa AFS/05	DF/ Course na	me: Ancient Ph	ilosophy and Pre	esent Times	
Course type, sco Course type: P Recommended Per week: 2 Pe Course method	ractice course-load (he r study period: l: present	ours):			
Number of cred					
Recommended s	semester/trimes	ter of the cours	se: 2.		
Course level: II.					
Prerequisities:					
Conditions for c	ourse completi	on:			
Learning outcor	nes:				
Brief outline of	the course:				
Recommended I	iterature:				
Course language	e:				
Course assessme Total number of		ts: 31			
A	В	С	D	Е	FX
80.65	6.45	6.45	0.0	6.45	0.0
Provides: Doc. F	hDr. Peter Nezr	úk, CSc.			
Date of last mod	lification: 31.08	.2017			
Approved: Guar	anteeprof. RND	r. Jozef Gonda,	DrSc.		

University: P. J. Šaf	arik Universi	ty in Košice			
Faculty: Faculty of	Science				
Course ID: ÚCHV/ AS1/03	Course na	ne: Asymmetri	c synthesis		
Course type, scope Course type: Lectu Recommended cou Per week: 2 / 1 Per Course method: p	ure / Practice urse-load (ho r study perio	urs):			
Number of credits:	5				
Recommended sem	ester/trimest	er of the cours	e:		
Course level: II.					
Prerequisities:					
Conditions for cou	rse completio	n:			
Learning outcomes	:				
Brief outline of the	course:				
Recommended liter	rature:				
Course language:					
Course assessment Total number of ass		s: 122			
A	В	С	D	Е	FX
73.77	14.75	6.56	2.46	2.46	0.0
Provides: prof. RNI	Dr. Jozef Gon	da, DrSc.	1	۰۱	
Date of last modific	cation: 26.02.	2018			
Approved: Guarant	eeprof. RNDr	. Jozef Gonda, I	DrSc.		

University: P. J. Šafa	árik Universi	ty in Košice			
Faculty: Faculty of S	Science				
Course ID: ÚCHV/ BOC/18	Course na	ne: Bioorganic	ká chémia		
Course type, scope a Course type: Lectu Recommended cou Per week: 2 / 1 Per Course method: pr	re / Practice rse-load (ho study perio resent	urs):			
Number of credits:	5				
Recommended sem	ester/trimest	er of the cours	se:		
Course level: II.					
Prerequisities:					
Conditions for cour	se completio	n:			
Learning outcomes	:				
Brief outline of the	course:				
Recommended liter	ature:				
Course language:					
Course assessment Total number of asse	essed students	5: 6			
A	В	С	D	Е	FX
33.33	33.33	0.0	33.33	0.0	0.0
Provides: prof. RND	Dr. Jozef Gon	da, DrSc.			
Date of last modific	ation: 02.03.	2018			
Approved: Guarante	eprof. RNDr	. Jozef Gonda.	DrSc.		

Universitare D. I. Čefárile Universitar in Kažias					
University: P. J. Šafárik University in Košice					
Faculty: Faculty of Science					
Course ID: ÚCHV/ Course name: Chemical nanotechnology CHN/09					
Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 0 Per study period: 28 / 0 Course method: present					
Number of credits: 4					
Recommended semester/trimester of the cours	e: 4.				
Course level: II.					
Prerequisities:					
Conditions for course completion:					
Learning outcomes: Students will be familiar with modern trends in the in creation and application of nanostructured mate Brief outline of the course: Modern trends in nanotechnology, in particular nate and switchable polymers, sensors and biosensors photonics.	erials and devices.				
Recommended literature: 1. Lectures handouts can be found at http://lms.u 2. Steed, J. W.; Turner, D. R. Wallace, K. J. Core nanochemistry; John Wiley & sons, Chichester 2 3. Rao, C. N. R.; Muller, A.; Cheetham, A. K. Na Weinheim 2007.	concepts in supramolecular chemistry and 007.				
Course language:					
Course assessment Total number of assessed students: 4					
abs n					
100.0 0.0					
Provides: prof. RNDr. Jozef Gonda, DrSc.					
Date of last modification: 26.02.2018					
Approved: Guaranteeprof. RNDr. Jozef Gonda, I	DrSc.				

University: P. J. Š	afárik Univers	ity in Košice			
Faculty: Faculty of	of Science				
Course ID: ÚCHV CHOZ/18	V/ Course na	me: Chémia orga	anokovových zlú	ičenín	
Course type, scop Course type: Lec Recommended c Per week: 2 / 1 P Course method:	cture / Practice course-load (he Per study perio	ours):			
Number of credit	s: 5				
Recommended se	mester/trimes	ter of the course	2.		
Course level: II.					
Prerequisities:					
Conditions for co	urse completi	on:			
Learning outcom	es:				
Brief outline of th	e course:				
Recommended lit	erature:				
Course language:					
Course assessmen Total number of a	-	ts: 12			
A	В	С	D	Е	FX
25.0	25.0	33.33	16.67	0.0	0.0
Provides: RNDr. J	lana Špaková F	Raschmanová, Ph	D.		
Date of last modif	fication: 02.03	.2018			
Approved: Guaran	nteeprof. RND	r. Jozef Gonda, E	DrSc.		

Faculty: Faculty	of Science				
Course ID: KFa DF2p/03	DF/ Course na	me: History of	Philosophy 2 (Ge	eneral Introductio	n)
Recommended	ecture / Practice course-load (h Per study peri	ours):			
Number of cred	lits: 4				
Recommended	semester/trimes	ster of the cours	se:		
Course level: I.,	II.				
Prerequisities:					
Conditions for a	course completi	on:			
Learning outco	mes:				
Brief outline of	the course:				
Recommended	literature:				
Course languag	e:				
Course assessm Total number of		ts: 738			
А	В	С	D	Е	FX
60.84	13.82	12.6	8.67	3.39	0.68
Provides: doc. P Katarína Mayerc		· · 1	· ·	Peter Nezník, CSo	c., PhDr.
Date of last mod	dification: 31.08	3.2017			
Approved: Guar		T 60 1			

University: P. J. S	Šafárik Univers	ity in Košice			
Faculty: Faculty	of Science				
Course ID: ÚCH EMDP/03	V/ Course na	me: Experiment	al Methods to M	aster's Thesis	
Course type, sco Course type: Pr Recommended Per week: 6 Per Course method	actice course-load (h r study period:	ours):			
Number of credi	ts: 6				
Recommended s	emester/trimes	ster of the cours	e:		
Course level: II.					
Prerequisities:					
Conditions for co	ourse completi	on:			
Learning outcon	nes:				
Brief outline of t	he course:				
Recommended li	iterature:				
Course language	2:				
Course assessme Total number of a		ts: 344			
A	В	С	D	Е	FX
94.77	3.2	0.58	0.58	0.87	0.0
Provides: RNDr. DrSc., doc. RNDr Antalík, DrSc., pr RNDr. Jozef Gon doc. RNDr. Taťán Oriňaková, DrSc. Vladimír Zeleňák RNDr. Miroslava Tomášková, PhD. Daniela Kladekov RNDr. Danica Sa prof. Mgr. Vasiľ A doc. Ing. Viera Vo	r. Ján Imrich, C rof. RNDr. Juraj da, DrSc., prof. na Gondová, CS , doc. RNDr. Iv a, PhD., doc. RN Matiková-Maľ ., RNDr. Andrea vá, CSc., RNDr. bolová, PhD., F Andruch, DrSc.,	Sc., doc. RNDr. j Černák, DrSc., RNDr. Andrej C Sc., doc. RNDr. M an Potočňák, Ph NDr. Viktor Vígla arová, PhD., doc a Morovská Turc Slávka Hamuľa RNDr. Zuzana Ku , prof. Dr. Yarosl	Mária Kožurková prof. RNDr. Kata Driňak, PhD., doc Airoslava Martinl D., doc. RNDr. E uský, PhD., doc. I . RNDr. Juraj Ku Mová, PhD., RNI ková, PhD., RNI udličková, PhD., av Bazeľ, DrSc.,	i, CSc., prof. Ing urína Györyová, I . RNDr. Zuzana cová, PhD., prof. rik Sedlák, PhD. RNDr. Katarína H chár, PhD., RNE Dr. Dušan Koščík Dr. Rastislav Vark RNDr. Lívia Koc	Marián DrSc., prof. Vargová, Ph.D., RNDr. Renáta , prof. RNDr. Reiffová, PhD., Dr. Nataša c, CSc., RNDr. nač, PhD., cúrová, PhD.,

Date of last modification: 26.02.2018

Approved: Guaranteeprof. RNDr. Jozef Gonda, DrSc.

University: P. J. Šafa	rik University in Košic	e					
Faculty: Faculty of S	Science						
Course ID: ÚCHV/ FAK1a/07	Course ID: ÚCHV/ Course name: Pharmacology I FAK1a/07						
Course method: pr	re / Practice rse-load (hours): study period: 28 / 28 esent						
Number of credits:	4						
Recommended sem	ester/trimester of the c	course:					
Course level: II.							
Prerequisities: ÚCH	V/FMCH/04						
Conditions for cour	se completion:						
Learning outcomes:							
Brief outline of the	course:						
Recommended liter	ature:						
Course language:							
Course assessment Total number of asse	essed students: 12						
	abs	n					
	100.0	0.0					
Provides: prof. MVI	Dr. Ján Mojžiš, DrSc.						
Date of last modific	ation: 26.02.2018						
Approved: Guarante	eprof. RNDr. Jozef Gor	nda, DrSc.					

University: P. J. Ša	afárik Univers	ity in Košice			
Faculty: Faculty of	f Science				
Course ID: ÚCHV FAK1b/07	// Course na	me: Pharmacolo	gy II		
Course type, scope Course type: Lec Recommended co Per week: 2 / 2 P Course method:	ture / Practice ourse-load (he er study perio	ours):			
Number of credits	s : 6				
Recommended ser	mester/trimes	ter of the cours	e:		
Course level: II.					
Prerequisities: ÚC	CHV/FAK1a/0	7			
Conditions for cou	urse completi	on:			
Learning outcome	es:				
Brief outline of th	e course:				
Recommended lite	erature:				
Course language:					
Course assessmen Total number of as		ts: 9			
A	В	С	D	Е	FX
0.0	11.11	33.33	11.11	44.44	0.0
Provides: prof. MV	VDr. Ján Mojž	iš, DrSc.		·	
Date of last modif	ication: 26.02	.2018			
Approved: Guaran	teeprof. RND	r. Jozef Gonda, I	DrSc.		

University: P. J. Ša	fárik Universi	ty in Košice			
Faculty: Faculty of	Science				
Course ID: ÚCHV FMCH/18	/ Course na	me: Farmaceuti	cká chémia		
Course type, scope Course type: Lect Recommended co Per week: 2 / 1 Pe Course method: p	cure / Practice purse-load (ho er study perio	ours):			
Number of credits	: 5				
Recommended sen	nester/trimes	ter of the cours	e:		
Course level: II.					
Prerequisities:					
Conditions for cou	rse completio	on:			
Learning outcome	s:				
Brief outline of the	e course:				
Recommended lite	rature:				
Course language:					
Course assessment Total number of ass		s: 7			
A	В	С	D	Е	FX
28.57	42.86	14.29	14.29	0.0	0.0
Provides: RNDr. M	lariana Budov	ská, PhD.			
Date of last modifi	cation: 02.03	.2018			
Approved: Guarant	teeprof. RND	. Jozef Gonda. I	DrSc.		

University: P. J. Š	Šafárik Univers	ity in Košice			
Faculty: Faculty	of Science				
Course ID: ÚCH HZ1/00	V/ Course na	ame: Heterocycli	c compounds		
Course type, sco Course type: Le Recommended Per week: 2 / 1 1 Course methods	cture / Practice course-load (h Per study peri	e ours):			
Number of credit	ts: 4				
Recommended so	emester/trimes	ster of the cours	e:		
Course level: II.					
Prerequisities:					
Conditions for co Two tests at semi Written exam	-	on:			
Learning outcom Goal of the subje synthesis, chemic	ct is to afford the			· 1	gnificance,
Brief outline of the Preparation and p and non-aromatic synthesis. Natural heterocycles and	propertieis of v c compounds, l compounds co	including their lontaining heteroc	piological prope	rties and applica	tion in organic
Recommended li 1. Gilchrist T.L.: 2. Eichler T., Hau Application. Seco	Heterocyclic C ptmann S.: Th	e Chemistry of H	eterocycles. Stru		, Synthesis and
Course language Slovak	:				
Course assessme Total number of a		ts: 131			
A	В	С	D	Е	FX
56.49	28.24	10.69	3.82	0.76	0.0
	Maniana Da 1	uská PhD			
Provides: RNDr.	Mariana Budo	vska, i iiD.			
Date of last modi		-			

University: P. J. Š	Šafárik Universi	ty in Košice			
Faculty: Faculty	of Science				
Course ID: KFaE IH2/03	DF/ Course na	me: Idea Huma	nitas 2 (General]	Introduction)	
Course type, scop Course type: Pra Recommended Per week: 2 Per Course method:	actice course-load (ho study period:	ours):			
Number of credit	ts: 2				
Recommended so	emester/trimes	ter of the cours	se: 3.		
Course level: II.					
Prerequisities:					
Conditions for co	ourse completio	on:			
Learning outcom	ies:				
Brief outline of t	he course:				
Recommended li	terature:				
Course language	:				
Course assessme Total number of a	-	s: 8			
A	В	С	D	Е	FX
87.5	12.5	0.0	0.0	0.0	0.0
Provides: Doc. Pl	hDr. Peter Nezn	ík, CSc.		·	
Date of last modi	fication: 31.08	2017			
Approved: Guara	inteeprof. RNDi	. Jozef Gonda,	DrSc.		

	CU	OURSE INFORM	MATION LET I	EK	
University: P. J.	Šafárik Univers	ity in Košice			
Faculty: Faculty	of Science				
Course ID: ÚCH KC/03	IV/ Course na	ame: Cosmetic c	hemistry		
Course type, sco Course type: Lo Recommended Per week: 2 / 1 Course method	ecture / Practice course-load (h Per study perio	ours):			
Number of credi	its: 4				
Recommended s	semester/trimes	ster of the cours	e:		
Course level: II.					
Prerequisities:					
Conditions for c Seminar report o with discussion.	on the selected s	ubjects of cosme	-	its oral presenta	tion connected
Learning outcom The basic chemic construction of s industry.	cal ingredients i	-	,		
Brief outline of the Skin and its corrigly cerophosphol alcohols, natural classification, or (amino acids, pp) ingredients. The acid, their biosym	mponents. The ipids and sfing and synthetic rganic and inor peptides, protein chemistry of fr	gophoslipids), lip waxes. Surfactar ganic dyes, natu ns hydroxy acic agrances. Compo	posomes as transits, their classific ral and synthetic ls, vitamins, pol punds derived fro	sport systems. I cation. Antioxida c. Biological act lysaccharides) a om shikimic acid	Fatty acids and nts. Dyes, their ive compounds s the cosmetic
Recommended I 1. S. V. Bhat, B. Narosa 2005, ISI 2. G. Ohloff: Sce 3-540-57108-6. 3. D. H. Pybus, C ISBN 0-8540-52 4. J. McMurry: C Eddition, ISBN 0	A. Nagasampag BN 81-7319-48 ent and Fragrand CH. S. Sell: The 8-7. Organic chemist	1-5. ces, Springer-Ver chemistry of fra	lag Berlín Heide grances, Royal S	lberg 1994, ISBN ociety of Chemi	N stry 1999,
Course language					
Course assessme					
Total number of		· · · · · · · · · · · · · · · · · · ·	D		
A	В	С	D	E	FX

79.07

15.12

4.65

1.16

0.0

0.0

Provides: doc. RNDr. Miroslava Martinková, PhD.

Date of last modification: 26.02.2018

Approved: Guaranteeprof. RNDr. Jozef Gonda, DrSc.

University: P. J. Š	Safárik Universi	ty in Košice			
Faculty: Faculty	of Science				
Course ID: KFaE KDF/05		me: Chapters fr General Introdu	•	nilosophy of 19th	and 20th
Course type, scop Course type: Pra Recommended Per week: 2 Per Course method:	actice course-load (ho study period:	ours):			
Number of credit	ts: 2				
Recommended se	emester/trimes	ter of the cours	e: 2.		
Course level: II.					
Prerequisities:					
Conditions for co	ourse completio	on:			
Learning outcom	ies:				
Brief outline of t	he course:				
Recommended li	terature:				
Course language	:				
Course assessme Total number of a		s: 10			
A	В	С	D	E	FX
50.0	20.0	10.0	0.0	10.0	10.0
Provides: doc. Ph	Dr. Pavol Thol	, PhD., mim. pr	of.		1
Date of last modi	fication: 31.08	.2017			
Approved: Guara	inteeprof. RND	. Jozef Gonda,	DrSc.		

University: P. J. Ša	ıfárik Universi	ty in Košice					
Faculty: Faculty of	f Science						
Course ID: KPPaPZ/KK/07							
Course type, scope Course type: Prac Recommended co Per week: 2 Per s Course method:	ctice ourse-load (ho study period:	ours):					
Number of credits	:2						
Recommended ser	nester/trimes	ter of the course: 3.					
Course level: II.							
Prerequisities:							
Conditions for cou	irse completio	on:					
Learning outcome	S:						
Brief outline of th	e course:						
Recommended lite	erature:						
Course language:							
Course assessmen Total number of as		s: 281					
abs		n	Z				
98.22		1.78	0.0				
Provides: Mgr. On	drej Kalina, Pł	D., Mgr. Lucia Hricová, PhD.					
Date of last modif	ication: 21.08.	2017					
Approved: Guaran	teeprof. RNDr	: Jozef Gonda, DrSc.					

Faculty: Faculty of Science

Course ID: ÚCHV/	Course name: Quantum Chemistry
KOC1/01	

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours):

Per week: 3 / 1 Per study period: 42 / 14

Course method: present

Number of credits: 5

Recommended semester/trimester of the course:

Course level: II.

Prerequisities:

Conditions for course completion:

Activity within practice will be evaluated. Two written tests will be realized in 7-th and 14-th week, resp. during the term of the course.

The examination will consist of written and verbal test. Continuous evaluation will be also taken into account.

Learning outcomes:

Students will intensify their knowledge in the field of valence-bond based on molecular orbital theory (MO) and self-reliant perform basic quantum chemical calculations (molecular geometry optimization, transition states, vibrational analysis, etc.).

Brief outline of the course:

Development of valence-bond theory. Time-independent Schrodinger equation. Basic approximations in molecular orbital valence-bond theory. Variant methods of calculation in the framework of molecular orbital valence-bond theory. Chemical reactivity. Potential energy hypersurfaces of molecules. Reaction coordinate. Calculation of the absolute and relative equilibrium and rate constants, resp. in gas phase. Solvatation energy calculation.

Recommended literature:

1. Jensen F.: Introduction to Computational Chemistry, Wiley, 2000.

- 2. Leach A. R.: Molecular Modelling, Addison Wesley Longman Ltd. 1998.
- 3. Náray-Szabó G., Surján P. R., Ángyán J. G.: Applied Quantum

Chemistry, Akadémia Kiadó, Budapest, 1987.

Course language:

slovak language and english language

Course assessment

Total number of assessed students: 28

А	В	С	D	Е	FX
82.14	14.29	3.57	0.0	0.0	0.0

Provides: RNDr. Ladislav Janovec, PhD.

Date of last modification: 25.09.2017

Approved: Guaranteeprof. RNDr. Jozef Gonda, DrSc.

		ity in Košice			
Faculty: Facult	y of Science				
Course ID: ÚC KOR1/00	HV/ Course na	me: Organic rea	ction kinetics		
Course type: l Recommende	ope and the met Lecture / Practice d course-load (h 1 Per study perio d: present	ours):			
Number of crea	dits: 4				
Recommended	semester/trimes	ster of the course	e:		
Course level: II	· ·				
Prerequisities:					
Work at semina reactions. Terminal exami	ination consists o	on: Calculations of k f responding 3 th tical solutions of	emes and 3 exe	, I	
for kinetic meas	surements of mai	odology of the ki n types of chemic and thermodynan	cal reactions. Le	earning of measu	rements and
for kinetic meas calculations of chemical exper- organic reaction Brief outline of The importance equations. Meth kinetic equation conditions on the	surements of mai the basic kinetic iments and the us is. The course: e of kinetics and hods used at mea as and rate consta he reaction rate. D	n types of chemic and thermodynam se of these data for d mechanisms of asuring of the rea nts. Main stages a Determination of t	cal reactions. Le nic parameters to or determination f organic reaction action rates. Par at solving of kin he kinetic equat	earning of measur using examples fit of the mechanist cons. Rate constant ticular steps of on the problems. Efficient and rate constant ion and rate constant	rements and rom real ms of the ants and kinetic determination of ffects of reaction stants. Reactions,
for kinetic meas calculations of chemical exper- organic reaction Brief outline of The importance equations. Meth kinetic equation conditions on th kinetic equation reactions. Paral	surements of mai the basic kinetic iments and the us is. The course: e of kinetics and hods used at mea hods used at mea hods and rate consta he reaction rate. D ns, and rate con ilel reactions. Co	n types of chemic and thermodynam se of these data for d mechanisms or asuring of the rea nts. Main stages a	cal reactions. Le nic parameters to or determination f organic reaction action rates. Par at solving of kin he kinetic equat rst, pseudo-firs ons. Activation	earning of measur using examples find of the mechanist ions. Rate constant ticular steps of of the problems. En- tion and rate constant t, and second of energy and entropy	rements and rom real ms of the ants and kinetic determination of ffects of reaction tants. Reactions, rder. Reversible opy. Acido-basic
for kinetic meas calculations of chemical exper- organic reaction Brief outline of The importance equations. Meth kinetic equation conditions on th kinetic equation reactions. Paral catalysis. Isotop	surements of mai the basic kinetic iments and the us is. The course: e of kinetics and hods used at mea as and rate constance reaction rate. D ns, and rate con ilel reactions. Co pic effects. Influe	n types of chemic and thermodynam se of these data for d mechanisms of asuring of the rea nts. Main stages a Determination of t instants of the fir insecutive reaction	cal reactions. Le nic parameters to or determination f organic reaction action rates. Par at solving of kin he kinetic equat rst, pseudo-firs ons. Activation	earning of measur using examples find of the mechanist ions. Rate constant ticular steps of of the problems. En- tion and rate constant t, and second of energy and entropy	rements and rom real ms of the ants and kinetic determination of ffects of reaction tants. Reactions, rder. Reversible opy. Acido-basic
for kinetic meas calculations of chemical exper- organic reaction Brief outline of The importance equations. Meth kinetic equation conditions on th kinetic equation reactions. Paral catalysis. Isotop relationships. Recommended	surements of mai the basic kinetic iments and the us is. The course: e of kinetics and hods used at mea as and rate consta ne reaction rate. D ns, and rate con lel reactions. Co pic effects. Influe literature:	n types of chemic and thermodynam se of these data for d mechanisms of asuring of the rea nts. Main stages a Determination of t instants of the fir insecutive reaction	cal reactions. Le nic parameters to or determination f organic reaction action rates. Par at solving of kin he kinetic equat rst, pseudo-firs ons. Activation	earning of measur using examples find of the mechanist ions. Rate constant ticular steps of of the problems. En- tion and rate constant t, and second of energy and entropy	rements and rom real ms of the ants and kinetic determination of ffects of reaction tants. Reactions, rder. Reversible opy. Acido-basic
for kinetic meas calculations of chemical exper- organic reaction Brief outline of The importance equations. Meth kinetic equation conditions on th kinetic equation reactions. Paral catalysis. Isotop relationships. Recommended Course languag Notes:	surements of mai the basic kinetic iments and the us is. The course: e of kinetics and hods used at mea as and rate consta he reaction rate. D ns, and rate con lel reactions. Co bic effects. Influe literature: ge:	n types of chemic and thermodynam se of these data for d mechanisms of asuring of the rea nts. Main stages a Determination of t instants of the fir insecutive reaction	cal reactions. Le nic parameters o or determination f organic reactin action rates. Par at solving of kin he kinetic equat rst, pseudo-firs ons. Activation im on the chem	earning of measur using examples find of the mechanist ions. Rate constant ticular steps of of the problems. En- tion and rate constant t, and second of energy and entropy	rements and rom real ms of the ants and kinetic determination of ffects of reaction tants. Reactions, rder. Reversible opy. Acido-basic
for kinetic meas calculations of chemical exper- organic reaction Brief outline of The importance equations. Meth kinetic equation conditions on th kinetic equation reactions. Paral catalysis. Isotop relationships. Recommended Course languag Notes: Advanced know	surements of mai the basic kinetic iments and the us is. The course: e of kinetics and hods used at mea hods used at mea ho	n types of chemic and thermodynam se of these data for d mechanisms of asuring of the rea nts. Main stages a Determination of t instants of the fir insecutive reaction ence of the mediu	cal reactions. Le nic parameters o or determination f organic reactin action rates. Par at solving of kin he kinetic equat rst, pseudo-firs ons. Activation im on the chem	earning of measur using examples find of the mechanist ions. Rate constant ticular steps of of the problems. En- tion and rate constant t, and second of energy and entropy	rements and rom real ms of the ants and kinetic determination of ffects of reaction tants. Reactions, rder. Reversible opy. Acido-basic
for kinetic meas calculations of chemical exper- organic reaction Brief outline of The importance equations. Meth kinetic equation conditions on th kinetic equation reactions. Paral catalysis. Isotop relationships. Recommended Course languag Notes: Advanced know	surements of mai the basic kinetic iments and the us is. The course: e of kinetics and hods used at mea hods used hods used hods used hods hods used hods hods used hods hods hods used hods	n types of chemic and thermodynam se of these data for d mechanisms of asuring of the rea nts. Main stages a Determination of t instants of the fir insecutive reaction ence of the mediu	cal reactions. Le nic parameters o or determination f organic reactin action rates. Par at solving of kin he kinetic equat rst, pseudo-firs ons. Activation im on the chem	earning of measur using examples find of the mechanist ions. Rate constant ticular steps of of the problems. En- tion and rate constant t, and second of energy and entropy	rements and rom real ms of the ants and kinetic determination of ffects of reaction tants. Reactions, rder. Reversible opy. Acido-basic

Provides: doc. RNDr. Ján Imrich, CSc.

Date of last modification: 26.02.2018

Approved: Guaranteeprof. RNDr. Jozef Gonda, DrSc.

University: P. J. Safái	rik University in Košice	
Faculty: Faculty of S	cience	
Course ID: ÚTVŠ/ KP/12	Course name: Survival Co	purse
Course type, scope a Course type: Practic Recommended cour Per week: Per stud Course method: pre	ce r se-load (hours): y period: 36s	
Number of credits: 2	,	
Recommended seme	ster/trimester of the cours	e:
Course level: I., II.		
Prerequisities:		
Conditions for cours Conditions for course Attendance Final assessment: cor	1	ks within the course
conditions as they wi and demanding situat	Il obtain theoretical knowled ions connected with surviva work and students will lear	afe stay and movement in extreme natural dge and practical skills to solve the extraordinary and minimization of damage to health. The m how to manage and face the situations that
Brief outline of the c Brief outline of the co Lectures: 1. Principles of behav	ourse:	
 Preparation and lea Objective and subj Principles of hygie Exercises: Movement in terra 	ective danger in mountains ne and prevention of damag in, orientation and navigatic rovised overnight stay	ent and stay in unknown mountains ge to health in extreme conditions on in terrain (compasses, GPS)
 Preparation and lea Objective and subj Principles of hygie Exercises: Movement in terra Preparation of imp 	ective danger in mountains ne and prevention of damag in, orientation and navigatio rovised overnight stay d food preparation.	e to health in extreme conditions
 Preparation and lea Objective and subj Principles of hygie Exercises: Movement in terra Preparation of imp Water treatment an 	ective danger in mountains ne and prevention of damag in, orientation and navigatio rovised overnight stay d food preparation.	e to health in extreme conditions
 Preparation and lea Objective and subj Principles of hygie Exercises: Movement in terra Preparation of imp Water treatment an Recommended litera 	ective danger in mountains ne and prevention of damag in, orientation and navigatic rovised overnight stay d food preparation. ture:	e to health in extreme conditions
 Preparation and lea Objective and subj Principles of hygie Exercises: Movement in terra Preparation of imp Water treatment an Recommended litera Course language: Course assessment 	ective danger in mountains ne and prevention of damag in, orientation and navigatic rovised overnight stay d food preparation. ture:	e to health in extreme conditions

Provides: MUDr. Peter Dombrovský, Mgr. Marek Valanský

Date of last modification: 18.08.2017

Approved: Guaranteeprof. RNDr. Jozef Gonda, DrSc.

University: P. J. Šafá	rik University in Košice	
Faculty: Faculty of S	cience	
Course ID: ÚTVŠ/ LKSp/13	Course name: Summer Co	ourse-Rafting of TISA River
Course type, scope a Course type: Practic Recommended cour Per week: Per stud Course method: pre	ce rse-load (hours): l y period: 36s	
Number of credits: 2		
Recommended seme	ster/trimester of the cours	e:
Course level: I., II.		
Prerequisities:		
Conditions for course Conditions for course Attendance Final assessment: Ra	-	attended/not attended)
Learning outcomes: Learning outcomes: Students have knowle	edge of rafts (canoe) and the	eir control on waterway.
5. Canoe lifting and c	ourse: iculty of waterways iting ning using an empty canoe carrying n the water without a shore be out of the water	contact
Recommended litera	iture:	
Course language:		
Course assessment Total number of asses	ssed students: 142	
	abs	n
	41.55	58.45

Provides: Mgr. Peter Bakalár, PhD.

Date of last modification: 18.08.2017

Approved: Guaranteeprof. RNDr. Jozef Gonda, DrSc.

University: P. J.	Šafárik Univers	sity in Košice			
Faculty: Faculty	of Science				
Course ID: ÚCH MM1/00	HV/ Course na	ame: Molecular n	nodeling		
Course type, sco Course type: L Recommended Per week: 1 / 3 Course method	ecture / Practice course-load (h Per study peri	e ours):			
Number of cred	its: 4				
Recommended	semester/trime	ster of the course	2 •		
Course level: II.					
Prerequisities:					
Conditions for c Verbal examinat	-				
chemistry using studies of the str	theory necessar specialized soft ucture and elect	y for the realisation ware packages. S tronic properties of structural aspects	tudents will be a of the smalll and	able to perform th I middle-sized mo	neoretical
minimum energ Methods in mole	isualization of y structure. The ecular mechanic	chemical structu coretical studies of s and semi-empir dynamics. Confor	of reaction mechical methods. A	hanisms and cher b initio and DFT	mical reactions.
2. JENSEN, Fra	rew R.: Molecu nk: An Introduc	lar Modelling: Pr tion to Computati RCHEM, GAME	ional Chemistry	•	
Course languag slovak language		guage			
Course assessm Total number of		nts: 62			
А	В	C	D	E	FX
77.42	22.58	0.0	0.0	0.0	0.0
Provides: RNDr	. Ladislav Janov	vec, PhD.		•	
Date of last mod	lification: 26.02	2.2018			
Approved: Guai	anteeprof. RNE	Dr. Jozef Gonda, E	DrSc.	-	

	CO	OURSE INFORM	IATION LETT	'ER	
University: P. J. Šafá	rik Univers	ity in Košice			
Faculty: Faculty of S	science				
Course ID: ÚCHV/ MSM1/00	Course na	ame: Modern syn	thetic methods		
Course type, scope a Course type: Lectur Recommended cou Per week: 3 / 1 Per Course method: pro	re / Practice rse-load (h study perie	e ours):			
Number of credits: (5				
Recommended seme	ester/trimes	ster of the course	2:		
Course level: II.					
Prerequisities:					
Conditions for course Seminar written disc	-		n by written forr	n.	
Learning outcomes: Understanding of mo		ods in the synthes	is of organic cor	npounds.	
Its purpose is to comolecules that play retrosynthetic analys oxidation, reduction,	important is of simple	roles in moderr e organic molecul	organic synth es, asymmetric	esis. The conce	pt of synthons,
Recommended litera 1. T. W. Green, P. G. Sons, Inc. 1999, ISB 2. B. M. Trost, I. Fle Oxford 1991. 3. B. Carruthers, I. C University Press 200 4. G. S. Zweifel, M. NY, ISBN: 0-7167-7 5. J. Fuhrhop, G. Per	M. Wuts: P N: 0-471-22 ming I.: Con Coldham: Mo 4, UK, ISB H. Nantz: M 266-3.	2057-4. mprehensive orga odern methods of N: 0-521-77097-1 Aodern Organic S	unic synthesis, E Forganic synthes I. ynthesis, W. H.	ds. Vol. 1-9. Perg sism 4th edition, Freeman and Co	gamon Press, Cambridge
Course language:					
Course assessment Total number of asse	ssed studen	ts: 125			
A	В	С	D	Е	FX
56.0	20.8	13.6	8.0	1.6	0.0
Provides: prof. RND	r. Jozef Goi	nda, DrSc., doc. F	RNDr. Miroslava	Martinková, Ph	D.
Date of last modifica	ation: 26.02	2.2018			

Easulter E. 14		sity in Košice			
racuity: Facult	y of Science				
Course ID: ÚC NCH/03	CHV/ Course n	ame: Neurochem	istry		
Course type: Recommende	cope and the me Lecture / Practic d course-load (I 1 Per study per od: present	e 1ours):			
Number of cre	dits: 5				
Recommended	semester/trime	ester of the cours	e:		
Course level: I	I				
Prerequisities:					
Seminar report		ion: subjects of neuroc ination by oral fo		oral presentation	connected
Learning outco Explanation of		principles of the	chemical transm	ission between n	erve cells.
bilayer, memb cellular signal (glutamate, asp	anatomy, charac rane proteins. M ing. Neurotransi partate, GABA, g	teristics of the ne lembrane transpo mitters - acetylcl lycine). Neuropep ger hypothesis (cA	rt and ion chan noline, catechola ptides - neuroper	nels. Synaptic tra amines, serotonin tide functions an	ansmission and n, amino acids
	S. Siegel, R. W. ular, and medicir	Albers, D. L Pric nal neurobiology,		• •	
S. T. Brady, G. molecular, cell	S. Siegel, R. W. ular, and medicir 2-374947-5			• •	
S. T. Brady, G. molecular, cell ISBN: 978-0-1 Course langua Course assessm	S. Siegel, R. W. ular, and medicir 2-374947-5 ge:	nal neurobiology,		• •	
S. T. Brady, G. molecular, cell ISBN: 978-0-1 Course langua Course assessm	S. Siegel, R. W. ular, and medicir 2-374947-5 ge: nent	nal neurobiology,		• •	
S. T. Brady, G. molecular, cell ISBN: 978-0-1 Course langua Course assessm Total number c	S. Siegel, R. W. ular, and medicir 2-374947-5 ge: nent of assessed studer	nal neurobiology,	eighth edition, A	cademic Press 20	012, UK,
S. T. Brady, G. molecular, cell ISBN: 978-0-1 Course langua Course assess Total number c A 55.37	S. Siegel, R. W. ular, and medicir 2-374947-5 ge: nent of assessed studer B 19.83	nts: 121	D 7.44	E	012, UK, FX
S. T. Brady, G. molecular, cell ISBN: 978-0-1 Course langua Course assesses Total number of A 55.37 Provides: doc.	S. Siegel, R. W. ular, and medicir 2-374947-5 ge: nent of assessed studer B 19.83	nts: 121 C 15.7 a Martinková, Phl	D 7.44	E	012, UK, FX

University: P. J. Šafárik University in Košic

Faculty: Faculty of Science

Course ID: ÚCHV/	Course name: 1D & 2D NMR Spectroscopy
NMR1/00	

Course type, scope and the method: Course type: Lecture / Practice

Recommended course-load (hours):

Per week: 2 / 3 Per study period: 28 / 42

Course method: present

Number of credits: 6

Recommended semester/trimester of the course:

Course level: II.

Prerequisities:

Conditions for course completion:

Active student's work at seminars and individual homework, written examinations in 7th and 14th semestral week.

Terminal examination in written form (4 exercises from combined applications of 1D a 2D NMR and other spectral methods) and oral form (3 themes) joining theoretical knowledge with a practical solution of selected NMR problems and exercises.

Learning outcomes:

Students will learn how to analyze structure and properties of organic, inorganic and biomolecular compounds by 1D and 2D proton and carbon NMR spectra, quantitative NMR analysis, and practical applications in various fields of science and technology.

Brief outline of the course:

Theoretical principles of nuclear magnetic resonance (NMR), basic NMR pulse techniques and Fourier transformation, NMR spectrometers, description of NMR by vector models. Parameters of one- (1D) and two-dimensional (2D) NMR spectra, practical application of 1H and 13C NMR spectra and basic correlated 2D spectra for structure and stereochemical arrangement, elucidation of reaction mechanisms, molecular dynamics, physico-chemical properties and quantitative analysis of chemical compounds.

Recommended literature:

1. Friebolin H.: Basic One- and Two-Dimensional NMR Spectrocopy, 5. Ed., Wiley, 2010.

2. T. D. W. Claridge: High-Resolution NMR Techniques in Organic Chemistry, Elsevier, 1999.

3. Atta-ur-Rahman, M. I. Choudhary: Solving Problems with NMR spectroscopy, Academic Press 1996.

4. H.-O. Kalinowski, S. Berger, S. Braun: Carbon-13 NMR Spectroscopy. Wiley, New York 1988.

5. A. E. Derome: Modern NMR Techniques for Chemistry Research. Pergamon Press, Oxford 1987.

6. E. Pretsch, B. Buhlmann, C. Affolter: Structure Determination of Organic Compounds. Tables of Spectral Data. Springer Verlag, Berlin 2000.

7. E. Breitmaier: Structure Elucidation by NMR in Organic Chemistry: A Practical Guide, 3rd Ed., Wiley, 2002.

8. E. Breitmaier, W. Voelter: Carbon-13 NMR Spectroscopy. VCH Weinheim, 1990.

Course language:							
Course assessm Total number of	nent f assessed studen	ts: 160					
А	В	С	D	Е	FX		
38.75	25.0	23.75	10.63	1.88	0.0		
Provides: doc.]	RNDr. Ján Imrich	n, CSc.			•		
Date of last mo	dification: 26.02	2.2018					
Approved: Gua	aranteeprof. RND	r. Jozef Gonda, I	DrSc.				

University: P. J. Ša	fárik Universi	ity in Košice			
Faculty: Faculty of	Science				
Course ID: ÚCHV NMRP/14	Course na	me: NMR prakt	ikum		
Course type, scope Course type: Prac Recommended co Per week: 3 Per s Course method: p	tice ourse-load (ho tudy period:	ours):			
Number of credits:	: 6				
Recommended sen	nester/trimes	ter of the cours	e:		
Course level: II.					
Prerequisities:					
Conditions for cou	rse completio	on:			
Learning outcome	s:				
Brief outline of the	course:				
Recommended lite	rature:				
Course language:					
Course assessment Total number of ass		s: 46			
А	В	С	D	Е	FX
100.0 0.0 0.0 0.0 0.0 0.0					
Provides: RNDr. M	lária Vilková,	PhD.			
Date of last modifi	cation: 26.02	.2018			
Approved: Guarant	teeprof. RND	r. Jozef Gonda, 1	DrSc.		

University: P. J. Šafá	rik University in Košice				
Faculty: Faculty of S	cience				
Course ID: ÚCHV/ Course name: Organic chemistry OCHST/15					
Course type, scope a Course type: Recommended cour Per week: Per stud Course method: pre	rse-load (hours): ly period:				
Number of credits: 4	•				
Recommended seme	ster/trimester of the course:				
Course level: II.					
Prerequisities:					
Conditions for cours	e completion:				

Learning outcomes:

Brief outline of the course:

Reaction Mechanisms, Mechanisms of Organic Reactions Reactive Intermediates Ionic Reactions Radical Reactions Bond Energy Reaction Energetics Activation Energy Reaction Rates and Kinetics Thermodynamic and Chemical Stability Aromaticity Benzene and Other Aromatic Compounds Fused Benzene Ring Compounds Other Aromatic Systems Factors Required for Aromaticity Stereoisomers Chirality and Symmetry Enantiomorphism Polarimetry Optical Activity Designating the Configuration of Stereogenic Centers The Sequence Rule for Assignment of Configurations to Stereogenic Carbons Compounds Having Two or More Stereogenic Centers Stereogenic Nitrogen Fischer Projection Formulas Alkenes Electrophilic Additions Strong Brønsted Acids Lewis Acids (non-Proton Electrophiles) Electrophilic Halogen Reagents Other Electrophilic Reagents Reduction Oxidation Radical Additions Allylic Substitution Alkynes Addition Reactions Hydrogenation Electrophiles Hydration & Tautomerism Hydroboration Nucleophilile Addition & Reduction Acidity of Terminal Alkynes (Substitution of H) Alkyl Halides General Reactivity Substitution(of X) SN2 Mechanism SN1 Mechanism Elimination (of HX) Summary of Substitution vs. Elimination Substitution by Metals Elimination Reactions of Dihalides Alcohols Reactions of Alcohols Substitution of the Hydroxyl H Substitution of the Hydroxyl Group Elimination of Water Oxidation of Alcohols Reactions of Phenols Acidity of Phenols Ring Substitution of Phenols Oxidation to Quinones Aromatic compounds Electrophilic Substitution A Substitution Mechanism Reactions of Substituted Benzenes Reaction Characteristics Reactions of Disubstituted Rings Reactions of Substituent Groups Nucleophilic Substitution, Elimination & Addition Reactions Amines Basicity of Nitrogen Compounds Acidity of Nitrogen Compounds Important Reagent Bases Reactions of Amines Electrophilic Substitution at Nitrogen Preparation of 1°-Amines Preparation of 2° & 3°-Amines Reactions with Nitrous Acid Reactions of Aryl Diazonium Intermediates Elimination Reactions of Amines Oxidation States of Nitrogen Basic information: Aldehydes & Ketones Carboxylic Acids Carboxylic Derivatives Natural products, Saccharides, Aminoacids, Biologically active compounds Aldehydes & Ketones Natural Products Synthetic Preparation Properties of Aldehydes & Ketones Reversible Addition Reactions Hydration & Hemiacetal Formation Acetal Formation Imine Formation Enamine Formation Cyanohydrin Formation Irreversible Addition Reactions Complex Metal Hydrides Organometallic Reagents Carbonyl Group Modification Wolff-Kishner Reduction Clemmensen Reduction Hydrogenolysis of Thioacetals Oxidations Reactions at the a-Carbon Mechanism of Electrophilic a-Substitution The Aldol Reaction Ambident Enolate Anions Alkylation of Enolate Anions Carboxylic Acids Natural Products Related Derivatives Physical Properties Preparation of Carboxylic Acids Reactions of Carboxylic Acids Salt Formation Substitution of Hydroxyl Hydrogen Substitution of the Hydroxyl Group Reduction & Oxidation Carboxylic Derivatives Reactions of Carboxylic Acid Derivatives Acyl Group Substitution Mechanism Reduction Catalytic Reduction Metal Hydride Reduction Diborane Reduction Reaction with Organometallic Reagents Reactions at the a Carbon Acidity of a C-H The Claisen Condensation Synthesis Applications Carbohydrates Glucose The Structure and Configuration of Glucose Anomeric Forms of Monosaccharides Glycosides Disaccharides Polysaccharides Lipids Fatty Acids Soaps & Detergents Fats & Oils Waxes Phospholipids Prostaglandins Terpenes Proteins and Amino Acids a-Amino Acids Reactions of Amino Acids Synthesis of Amino Acids Peptides & Proteins The Primary Structure of Peptides Secondary & Tertiary Structure of Large Peptides and Proteins Peptide Synthesis Nucleic Acids The Primary Structure of DNA The Secondary & Tertiary Structures of DNA RNA and Protein Synthesis

Recommended literature:

Course language:

Course assessment

Total number of assessed students: 46

Iotal number of assessed students: 46							
Α	В	С	D	Е	FX		
63.04 30.43 4.35 0.0 2.17 0.0							
Provides:							
Date of last modification: 26.02.2018							
Approved: Guaranteeprof. RNDr. Jozef Gonda, DrSc.							

University: P. J. Šaf	čárik Universi	ty in Košice			
Faculty: Faculty of	Science				
Course ID: ÚCHV/ ODPFC/01	Course na	ne: Defence of	Diploma Thesis		
Course type, scope Course type: Recommended cou Per week: Per stu Course method: p	urse-load (ho dy period: resent				
Number of credits:	_				
Recommended sem	ester/trimest	er of the cours	e:		
Course level: II.					
Prerequisities:					
Conditions for cou	rse completio	n:			
Learning outcomes	:				
Brief outline of the	course:				
Recommended liter	rature:				
Course language:					
Course assessment Total number of ass	essed student	s: 36			
A	В	С	D	Е	FX
88.89 8.33 2.78 0.0 0.0 0.0					
Provides:	I			·	
Date of last modific	cation: 21.09.	2017			
Approved: Guarant	eeprof. RNDr	Jozef Gonda I	DrSc.		

University. 1. J. k	Salarik Univers	sity in Košice			
Faculty: Faculty	of Science				
Course ID: ÚCH OS/03	V/ Course na	ame: Organic syr	nthesis		
Course type, scop Course type: Le Recommended Per week: 2 / 1 1 Course method	cture / Practice course-load (h Per study peri	e Iours):			
Number of credi	ts: 5				
Recommended se	emester/trime	ster of the cours	e:		
Course level: II.					
Prerequisities:					
Midterm exam. Presentation of a Final written examine	m	hesis.			
Learning outcom The aim is to bec compounds, their	ome familiar w	-		-	-
Brief outline of the Retrosynthetic and backbone using or bonds. Synthesis and substitutions molecules and national substitutions and substitutions and substitutions and substitutions and substitutions molecules and national substitutions and substitutions are substitutions and substitutions are substitutions and substitutions are substitutions are substitutions.	nalysis of org rganometallic of cyclic mole s. Protecting g	compounds and er cules. Functional groups and speci	nolates. Reaction group manipula	is resulting in creation using oxidation	tion of multiple
Recommended li 1. Lecture handor id=386 2. Carruthers W., University Press, 3. Hanson, J. R.:	uts and semina Coldham I.: M 2005	Iodern Methods c	of Organic Synth	esis, Fourth Editi	on, Cambridge
Course language	:				
Course assessme Total number of a		nts: 156			
А	В	С	D	E	FX
51.92	30.13	12.18	3.21	2.56	0.0
	Ián Eločko. Dh	 D			
Provides: RNDr.	Jan Elecko, Fi				
Provides: RNDr. Date of last modi					

University: P. J.	Šafárik Univers	ity in Košice					
Faculty: Faculty	of Science						
Course ID: ÚCH PCH1/00	IV/ Course name: Food chemistry						
Recommended	ecture / Practice l course-load (h Per study peri	e ours):					
Number of cred	lits: 4						
Recommended	semester/trime	ster of the cours	e: 3.				
Course level: I.,	II.						
Prerequisities:							
Conditions for a	course completi	on:					
	cieve informatio	ns and knowledg es in food during		cal substances in storage.	food, their		
carbohydrates. V	ories of substance Water, minerals,	1	n anorganic con	food. Aminoacids npounds, vitamins y products.			
Recommended	literature:						
Course languag	je:						
Course assessm Total number of		ts: 256					
А	В	С	D	Е	FX		
60.55	33.98	5.08	0.0	0.0	0.39		
Provides: RNDr	: Ján Elečko, Ph	D.	1	1	,		
Date of last mod	dification: 26.02	2.2018					
	(r. Jozef Gonda, I	2 2				

University: P. J.	Šafárik Univers	ity in Košice			
Faculty: Faculty	of Science				
Course ID: Dek. UPJŠ/PPZ/13	PF Course na on a Labor		Development a	nd Key Competer	nces for Success
Course type, sco Course type: Pr Recommended Per week: Per Course method	cactice course-load (h study period: 1	ours):			
Number of credi	its: 2				
Recommended s	emester/trimes	ter of the cours	se: 1., 3.		
Course level: II.					
Prerequisities:					
Conditions for c	ourse completi	on:			
Learning outcon	nes:				
Brief outline of t	the course:				
Recommended l	iterature:				
Course language	2.				
Course assessme Total number of	-	ts: 39			
A	В	С	D	E	FX
100.0	0.0	0.0	0.0	0.0	0.0
Provides: RNDr.	Peter Stefányi,	PhD.			·
Date of last mod	ification: 19.02	.2018			
Approved: Guara	anteeprof. RND	r. Jozef Gonda,	DrSc.		

University: P. J. Ša	fárik Univers	ity in Košice			
Faculty: Faculty of	Science				
Course ID: KPPaPZ/PPZMg/12		me: Psychology	and Health Psyc	chology (Master's	s Study)
Course type, scope Course type: Lec Recommended co Per week: 1 / 2 Po Course method: 1	ture / Practice ourse-load (he er study perio	ours):			
Number of credits	: 4				
Recommended ser	nester/trimes	ter of the cours	e:		
Course level: II.					
Prerequisities:					
Conditions for cou	Irse completi	on:			
Learning outcome	s:				
Brief outline of the	e course:				
Recommended lite	erature:				
Course language:					
Course assessmen Total number of as	-	ts: 226			
А	В	С	D	Е	FX
19.47	25.22	25.66	13.27	15.93	0.44
Provides: PhDr. An	nna Janovská,	PhD., Mgr. Luc	ia Hricová, PhD.		
Date of last modifi	cation: 21.08	.2017			
Approved: Guaran	teeprof. RND	r. Jozef Gonda, 1	DrSc.		

University: P. J. Šat	fárik Universi	ty in Košice			
Faculty: Faculty of	Science				
Course ID: ÚCHV/ PRL/18	Course na	me: Chémia prí	rodných látok		
Course type, scope Course type: Lect Recommended co Per week: 2 / 1 Pe Course method: p	ure / Practice urse-load (ho r study perio	ours):			
Number of credits:	5				
Recommended sem	nester/trimest	er of the cours	e:		
Course level: II.					
Prerequisities:					
Conditions for cou	rse completio	n:			
Learning outcomes	S:				
Brief outline of the	course:				
Recommended lite	rature:				
Course language:					
Course assessment Total number of ass		s: 4			
A	В	С	D	Е	FX
75.0	25.0	0.0	0.0	0.0	0.0
Provides: doc. RNI	Dr. Miroslava	Martinková, Ph	D.	·	•
Date of last modified	cation: 02.03.	2018			
Approved: Guarant	eeprof. RNDr	Jozef Gonda, I	DrSc.		

vý projekt
urse:
pratory, evaluation of results, discussion, results
degree . Evaluation of results and verbal presentation
n
0.0
orof. RNDr. Renáta Oriňaková, DrSc., RNDr. Andrea a Straková Fedorková, PhD., prof. RNDr. Jozef

University: P. J. Šaf	ărik Universit	y in Košice			
Faculty: Faculty of	Science				
Course ID: ÚCHV/ SEM1a/00	Course nan	ne: Diploma w	ork seminar		
Course type, scope Course type: Pract Recommended cou Per week: 2 Per st Course method: pr	tice urse-load (hou udy period: 2	ırs):			
Number of credits:	2				
Recommended sem	ester/trimeste	er of the cours	e:		
Course level: II.					
Prerequisities:					
Conditions for cour	rse completion	n:			
Learning outcomes	•				
Brief outline of the	course:				
Recommended liter	rature:				
Course language:					
Course assessment Total number of ass	essed students	: 107			
A	В	С	D	Е	FX
100.0	0.0	0.0	0.0	0.0	0.0
Provides: prof. RNI	Dr. Jozef Gond	a, DrSc.		·	
Date of last modific	cation: 26.02.2	2018			
Approved: Guarante	eeprof. RNDr.	Jozef Gonda.	DrSc.		

University: P. J. Ša	fárik Universit	y in Košice			
Faculty: Faculty of	Science				
Course ID: ÚCHV/ SEM1b/00	Course nan	ne: Diploma w	ork seminar		
Course type, scope Course type: Prac Recommended co Per week: 2 Per st Course method: p	tice urse-load (hou tudy period: 2	urs):			
Number of credits:	: 2				
Recommended sem	nester/trimeste	er of the cours	e:		
Course level: II.					
Prerequisities: ÚC	HV/SEM1a/00				
Conditions for cou	rse completio	n:			
Learning outcomes	5:				
Brief outline of the	course:				
Recommended lite	rature:				
Course language:					
Course assessment Total number of ass		: 94			
A	В	С	D	Е	FX
100.0	0.0	0.0	0.0	0.0	0.0
Provides: prof. RN	Dr. Jozef Gond	la, DrSc.	1		
Date of last modified	cation: 26.02.2	2018			
Approved: Guarant	teeprof. RNDr.	Jozef Gonda.	DrSc.		

University: P. J. Šafá	rik University in Košice			
Faculty: Faculty of S	cience			
Course ID: ÚCHV/ SEP1/15	Course name: Semestrálny	y projekt 1		
Course type, scope a Course type: Practic Recommended cour Per week: 6 Per stu Course method: pre	ce rse-load (hours): dy period: 84			
Number of credits: 4				
Recommended seme	ster/trimester of the cours	e:		
Course level: II.				
Prerequisities:				
Conditions for cours Notification any thes work with master deg	is adversed by Department of	of Physical Chemistry. Semester experimental		
Learning outcomes: Semester scientific th	esis.			
Brief outline of the c Experimental work in and discussion about	research field for master deg	gree . Evaluation of results and verbal presentation		
Recommended litera Recent journal refere				
Course language:				
Course assessment Total number of asses	ssed students: 38			
	abs	n		
	97.37 2.63			
Morovská Turoňová, Martinková, PhD., pro Patrik Olekšák, RND Imrich, CSc., RNDr. 1	PhD., doc. RNDr. Andrea S of. RNDr. Jozef Gonda, DrS Kvetoslava Stanková, PhD	. RNDr. Renáta Oriňaková, DrSc., RNDr. Andrea traková Fedorková, PhD., doc. RNDr. Miroslava c., RNDr. Monika Tvrdoňová, PhD., RNDr. D., RNDr. Ján Elečko, PhD., doc. RNDr. Ján NDr. Martin Walko, PhD., RNDr. Ladislav		
Date of last modifica	tion: 21.00.2017			
Date of fast mounted				

University: P. J. Šafá	rik University in Košice	
Faculty: Faculty of S	cience	
Course ID: ÚCHV/ SEP2/15	Course name: Semestrálny	y projekt 2
Course type, scope a Course type: Practic Recommended cour Per week: 6 Per stu Course method: pre	ce rse-load (hours): dy period: 84	
Number of credits: 6	5	
Recommended seme	ster/trimester of the cours	e:
Course level: II.		
Prerequisities:		
Conditions for cours Notification any thes work with master deg	is adversed by Department of	of Physical Chemistry. Semester experimental
Learning outcomes: Semester scientific th	nesis.	
Brief outline of the c Experimental work in and discussion about	research field for master dea	gree . Evaluation of results and verbal presentation
Recommended litera	nture:	
Course language:		
Course assessment Total number of asse	ssed students: 39	
	abs	n
	100.0	0.0
Morovská Turoňová, Martinková, PhD., pro Kvetoslava Stanková,	PhD., doc. RNDr. Andrea S of. RNDr. Jozef Gonda, DrS PhD., RNDr. Ján Elečko, P	E. RNDr. Renáta Oriňaková, DrSc., RNDr. Andrea traková Fedorková, PhD., doc. RNDr. Miroslava c., RNDr. Monika Tvrdoňová, PhD., RNDr. hD., RNDr. Mariana Budovská, PhD., RNDr. , PhD., RNDr. Ladislav Janovec, Ph.D.
Date of last modifica	tion, 21,00,2017	
Date of last mounica	LION: 21.09.2017	

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ Course name: Supramolecular chemistry SMCH/03 Course name: Supramolecular chemistry Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14 Course method: present Number of credits: 4 Recommended semester/trimester of the course: Course: Course level: II. Prerequisities:	
Course ID: ÚCHV/ SMCH/03 Course name: Supramolecular chemistry Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14 Course method: present Number of credits: 4 Recommended semester/trimester of the course: Course level: II.	
SMCH/03 Image: Constant of the method: Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14 Course method: present Image: Course method: Number of credits: 4 Image: Course level: II.	
Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14 Course method: present Number of credits: 4 Recommended semester/trimester of the course: Course level: II.	
Recommended semester/trimester of the course: Course level: II.	
Course level: II.	
Prerequisities:	
Conditions for course completion: Presentation of a chosen topic. Final written exam. Learning outcomes:	
Brief outline of the course:	
Recommended literature:1. Lecture handouts can be found at http://lms.upjs.sk/course/view.php?id=3852. J.W.Steed and J.L.Atwood, Supramolecular chemistry, Wiley : Chichester, 2000.3. F.Vogtle, Supramolecular chemistry: an introduction, Wiley : Chichester, 1991.	
Course language:	
Course assessment Total number of assessed students: 67	
A B C D E FX	ζ
62.69 22.39 11.94 1.49 1.49 0.0)
Provides: RNDr. Martin Walko, PhD.	
Date of last modification: 26.02.2018	
Approved: Guaranteeprof. RNDr. Jozef Gonda, DrSc.	

University: P. J. Šafá	rik University in	Košice	
Faculty: Faculty of S	cience		
Course ID: KPPaPZ/SPVKE/07	Course name: Situations	Social-Psychological Tr	raining of Coping with Critical Life
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	ce rse-load (hours) dy period: 28	:	
Number of credits: 2			
Recommended seme	ster/trimester o	f the course: 2.	
Course level: II.			
Prerequisities:			
Conditions for cours	e completion:		
Learning outcomes:			
Brief outline of the c	ourse:		
Recommended litera	iture:		
Course language:			
Course assessment Total number of asse	ssed students: 12	.6	
abs		n	Z
97.62		2.38	0.0
Provides: Mgr. Ondro	ej Kalina, PhD.		
Date of last modifica	tion: 21.08.2017	7	
Approved: Guarantee	eprof. RNDr. Joz	ef Gonda, DrSc.	

University:	ΡJ	Šafárik	University	in Košice
omversiey.	1.0.	Suluin	Oniversity	

Faculty: Faculty of Science

Course ID: ÚCHV/ **Course name:** Structure and Reactivity in Organic Chemistry STRE/09

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours):

Per week: 2 / 1 Per study period: 28 / 14

Course method: present

Number of credits: 4

Recommended semester/trimester of the course:

Course level: II.

Prerequisities:

Conditions for course completion:

Tests: in 6th week (50 points) and in 12th week (50 points). At least 50% of points required from both. Terminal examination by written form, 100 points (2 x 50 points).

Learning outcomes:

This module aims to give an understanding of the major principles involved in organic chemistry - covering the fundamentals of bonding, structure and stereochemistry, leading to a description of the types of reaction and reactivity of the various structural types.

Brief outline of the course:

1. Bonding: atomic structure - the chemical bond, the periodic table, valence electrons, Lewis structures, conventions for drawing structures, atomic orbital theory, molecular orbital theory; covalent bonding – bonding in hydrocarbons, bonding in compounds containing heteroatoms, bonding in common functional groups, electronic effects, steric effects.

2. Structure: configuration – geometrical isomerism, optical isomerism, representations of stereoisomers, molecules with one stereogenic centre, molecules with more than one stereogenic centre, asymmetric heteroatoms; conformations – representations of conformers.

3. Reactivity: thermodynamics – Gibbs energy, enthalpy, entropy, chemical equilibrium; kinetics – rates of reaction, activation energy; classes of reaction mechanism – polar, radical, pericyclic, ligand coupling mechanisms, selectivity of reactions, solvents in organic chemistry.

4. Intermediates: carbocations, carbanions, radicals, carbenes, benzynes, ketenes.

5. Acidity a basicity: Lowry-Bronsted acid-base theory, organic acidity, organic basicity.

6. Nucleophilic Substitution: the SN1 reaction, the SN2 reaction, factors affecting reactions.

7. Electrophilic addition reactions, the energy profile of the reaction, the addition of HX to alkenes, Markovnikov's rule, the stereochemistry of electrophilic addition reactions, addition X2 to alkenes, hydration, hydroxymerkuration, hydroboration, addition of carbenes, addition of polyenes, nucleophilic addition reactions, nucleophilic addition to carbonyl compounds, addition water, addition of alcohols, addition of carbanions, the addition of organometallic reagents, addition of amines, conjugated additions, radical addition reactions.

8. Elimination reaction, E1, E2, E1cB, dehydration, dehydrohalogenation, dehalogenation, dehydrogenation.

9. The electrophilic aromatic substitution, halogenation, nitration, sulfonation, Friedel-Crafts alkylation, acylation, towards the impact of multiple groups, nucleophilic aromatic substitution, addition-elimination mechanism, benzynic mechanism, radical substitution of aromatics.

10. Nucleophilic substitution of sp2 carbon, tetrahedral mechanism, addition-elimination mechanism, the elimination-addition mechanism, the types of nucleophilic acyl substitution, nucleophilic acyl substitution of carboxylic acids and their derivatives.

11. Radical reactions, radical substitution, the radical addition, homolytic cleavage of σ -bond photochemical cleavage of π -bonds, one electron oxidation or reduction, cykloaromatiztion.

12. Pericyclic reaction types of pericyclic reactions: electrocyclic reactions, cycloaddition, sigmatropic rearrangements, ene reactions, Woodward-Hoffman rules.

7. Addition reactions

8. Elimination reactions

9. Aromatic substitution

10. Addition-elimination reactions

11. Radical reactions

12. Pericyclic reactions

Recommended literature:

1. Structure and Reactivity in Organic Chemistry, Mark G. Moloney, ISBN: 978-1-4051-1451-6, 318 pages, 2008, Wiley-Blackwell

2. Organic Chemistry: Structure and Reactivity, Seyhan N. Ege, ISBN-10: 0395902231, 1148 pages, 1998, Houghton Mifflin College Div.

Course language:

Slovak language and english language.

Course assessment

Total number of assessed students: 63

А	В	С	D	Е	FX
33.33	38.1	20.63	6.35	1.59	0.0

Provides: RNDr. Slávka Hamuľaková, PhD., RNDr. Mária Vilková, PhD.

Date of last modification: 26.02.2018

University: P. J. Š	afárik Univers	ity in Košice			
Faculty: Faculty c	of Science				
Course ID: ÚCHV SVK1/00	// Course na	me: Students S	cientific Conferen	nce (Presentation))
Course type, scop Course type: Recommended c Per week: Per s Course method:	ourse-load (he tudy period:				
Number of credit	s: 4				
Recommended se	mester/trimes	ter of the cours	se:		
Course level: II.					
Prerequisities:					
Conditions for co	urse completi	on:			
Learning outcom	es:				
Brief outline of th	e course:				
Recommended lit	erature:				
Course language:					
Course assessmen Total number of a	-	ts: 223			
A	В	С	D	Е	FX
100.0	0.0	0.0	0.0	0.0	0.0
Provides: prof. RN Yaroslav Bazel', D	•	riňak, PhD., pro	f. RNDr. Renáta	Oriňaková, DrSc.	., prof. Dr.
Date of last modif	fication: 25.09	.2017			
Approved: Guara	nteeprof. RND	r. Jozef Gonda,	DrSc.		

University: P. J. Šafárik University in Košice							
Faculty: Faculty of Science							
Course ID: ÚCHV/ Course name: Toxikológia organických látok TOXOL/18							
Course type, sco Course type: Le Recommended Per week: 2 / 1 Course method	ecture / Practice course-load (he Per study perio	ours):					
Number of credi	ts: 5						
Recommended s	emester/trimes	ter of the course	2.				
Course level: II.							
Prerequisities:							
Conditions for c	ourse completi	on:					
Learning outcon	nes:						
Brief outline of t	he course:						
Recommended li	iterature:						
Course language	2:						
Course assessment Total number of assessed students: 5							
A	В	С	D	Е	FX		
60.0	0.0	20.0	0.0	20.0	0.0		
Provides: doc. R	NDr. Miroslava	Martinková, PhI).	· · · ·			
Date of last mod	ification: 02.03	.2018					
Approved: Guara	Approved: Guaranteeprof. RNDr. Jozef Gonda, DrSc.						

University	P. J. Šafái	rik University i	n Košice								
Faculty: Fa	aculty of So	cience									
Course ID TVa/11	Course ID: ÚTVŠ/ Course name: Sports Activities I. Va/11										
Course ty Recomme Per week	pe: Practic nded cour	se-load (hours) dy period: 28									
Number of	credits: 2										
Recommer	ided seme	ster/trimester	of the cours	e: 1.							
Course lev	el: I., I.II.,	II.									
Prerequisi	ties:										
Conditions	for course	e completion: completion: articipation in c	classes.								
relationshi	physical co p of studen	ondition and perturbed to the selected			1		g the				
University floorball, y tennis, spo In the first and particu physical co Last but no means of a In addition physical co the premise	ne of the co optional su provides f yoga, pilate rts for unfi two semes larities of i ondition, co ot least, the special pro- to these se lucation tra- es of the fac	burse: bubject, the Inst for students the es, swimming, t persons, stree sters of the firs ndividual sport bordination abi important role ogram of medic sports, the Inst inings with an a culty or Univers	e following s body-buildin tball, tennis, it level of ed ts, motor skil ilities, physic of sports act cal physical o itute offers to attractive pro	ports activiti ag, indoor for and volleyba ucation stude ls, game activities cal performativities is to e education to for those wh gram and org	ies: aerobics, otball, self-de all. ents will mas vities, they wince, and mot eliminate swin influence and o are interest ganises variou	basketball, efence and l ster basic ch ill improve l tor performa mming illite mitigate ur ted winter a us competitio	badminton karate, table aracteristics evel of their ince fitness gracy and by fitness. and summer ons, either a				
Recommen	nded litera	ture:									
Course lan	guage:										
	essment										
Course ass		sed students [,] 1	Total number of assessed students: 11672 abs abs-A abs-C abs-D abs-E n neabs								
Course ass		abs-B	1672 abs-C	abs-D	abs-E	n	neabs				

Provides: Mgr. Peter Bakalár, PhD., Mgr. Dana Dračková, PhD., Mgr. Agata Horbacz, PhD., Mgr. Dávid Kaško, Mgr. Zuzana Küchelová, PhD., doc. PaedDr. Ivan Uher, PhD., Mgr. Marek Valanský, prof. RNDr. Stanislav Vokál, DrSc., Mgr. Marcel Čurgali, Ing. Iveta Cimboláková, PhD.

Date of last modification: 18.08.2017

	COUR	RSE INFORM	MATION LI	ETTER				
University: P. J. Šaf	árik University	in Košice						
Faculty: Faculty of	Science							
Course ID: ÚTVŠ/ TVb/11	1							
Course type, scope Course type: Pract Recommended cou Per week: 2 Per st Course method: pu	ice 1 rse-load (hour 1dy period: 28							
Number of credits:	2							
Recommended sem	ester/trimester	of the cours	e: 2.					
Course level: I., I.II	, II.							
Prerequisities:								
Conditions for cour Conditions for cours Final assessment an	se completion:		ses - min. 759	<i>%</i> .				
Learning outcomes: Learning outcomes: Increasing physical relationship of stude	condition and p			-		g the		
Brief outline of the Brief outline of the Within the optional University provides floorball, yoga, pila tennis, sports for un In the first two sem and particularities of physical condition, Last but not least, th means of a special p In addition to these physical education to the premises of the fa	course: subject, the Inst for students the tes, swimming, fit persons, streed esters of the first individual sport coordination ab e important role rogram of medit sports, the Inst rainings with an aculty or Univer	e following s body-buildir etball, tennis, st level of ed ts, motor skil ilities, physic e of sports ac cal physical titute offers attractive pro	sports activiting, indoor for and volleyba ucation study ls, game activities is to ever cal performativities is to ever education to for those who gram and org	ies: aerobics otball, self-d all. ents will ma- vities, they w nce, and mo eliminate swi influence an o are interes ganises vario	, basketball, lefence and l ster basic ch vill improve l tor performa imming illite d mitigate un sted winter a us competitio	badminton, karate, table aracteristics evel of their ance fitness. eracy and by hfitness. and summer ons, either at		
Recommended liter	ature:							
Course language:								
Course assessment								
Total number of ass abs abs-A		10971 abs-C	abs-D	abs-E	n	neabs		
			1		n 10.12			
85.37 0.57	0.02	0.0	0.0	0.05	10.13	3.86		

Provides: Mgr. Peter Bakalár, PhD., Mgr. Dana Dračková, PhD., Mgr. Agata Horbacz, PhD., Mgr. Dávid Kaško, Mgr. Zuzana Küchelová, PhD., doc. PaedDr. Ivan Uher, PhD., Mgr. Marek Valanský, prof. RNDr. Stanislav Vokál, DrSc., Mgr. Marcel Čurgali, Ing. Iveta Cimboláková, PhD.

Date of last modification: 18.08.2017

University:	P. J. Šafárik	University i	n Košice					
Faculty: Faculty of Science								
Course ID: TVc/11	ÚTVŠ/ C	Course name: Sports Activities III.						
Per week: 2 Course me	e: Practice ded course 2 Per study thod: prese	e-load (hours period: 28						
Number of o	credits: 2							
Recommend	led semeste	er/trimester	of the cours	e: 3.				
Course level	: I., I.II., II							
Prerequisiti	es:							
Conditions f	for course of	completion:						
Learning ou	tcomes:							
Brief outline	e of the cou	irse:						
Recommend	led literatu	re:						
Course lang	uage:							
Course assessment Total number of assessed students: 6910								
abs	abs-A	abs-B	abs-C	abs-D	abs-E	n	neabs	
89.84	0.04	0.0	0.0	0.0	0.03	4.23	5.86	
Provides: Mgr. Marcel Čurgali, Mgr. Peter Bakalár, PhD., Mgr. Dana Dračková, PhD., Mgr. Agata Horbacz, PhD., Mgr. Dávid Kaško, Mgr. Zuzana Küchelová, PhD., doc. PaedDr. Ivan Uher, PhD., Mgr. Marek Valanský, prof. RNDr. Stanislav Vokál, DrSc., Ing. Iveta Cimboláková, PhD.								
Date of last	modificatio	on: 18.08.201	7					
Approved: (Guaranteen	of. RNDr. Jo	zef Gonda.	DrSc.				

University: F	P. J. Šafárik	University in	n Košice					
Faculty: Faculty of Science								
Course ID: U TVd/11	ΊΤ V Š/ C	Course name: Sports Activities IV.						
Per week: 2 Course met	e: Practice ded course Per study hod: prese	e-load (hours) period: 28						
Number of c	redits: 2							
Recommend	ed semeste	er/trimester	of the cours	e: 4.				
Course level	: I., I.II., II.							
Prerequisitie	es:							
Conditions f	or course o	completion:						
Learning out	tcomes:							
Brief outline	of the cou	rse:						
Recommend	ed literatu	re:						
Course lang	lage:							
Course asses		ed students: 5	045					
abs	abs-A	abs-B	abs-C	abs-D	abs-E	n	neabs	
85.09	0.3	0.04	0.0	0.0	0.0	6.82	7.75	
Provides: Mgr. Marcel Čurgali, Mgr. Peter Bakalár, PhD., Mgr. Dana Dračková, PhD., Mgr. Agata Horbacz, PhD., Mgr. Dávid Kaško, Mgr. Zuzana Küchelová, PhD., doc. PaedDr. Ivan Uher, PhD., Mgr. Marek Valanský, prof. RNDr. Stanislav Vokál, DrSc., Ing. Iveta Cimboláková, PhD.						Uher, PhD.,		
Date of last r	nodificatio	on: 18.08.201	7					
Approved: C	Juaranteepr	of. RNDr. Jo	zef Gonda. 1	DrSc.				

University: P. J. Ša	afárik Universi	ty in Košice						
Faculty: Faculty of	f Science							
Course ID: KPPaPZ/UPR/03								
Course type, scope Course type: Prac Recommended co Per week: 2 Per s Course method:	ctice ourse-load (ho study period: 2	ours):						
Number of credits	: 2							
Recommended ser	mester/trimest	er of the cours	se: 4.					
Course level: II.								
Prerequisities:								
Conditions for cou	urse completio	n:						
Learning outcome	es:							
Brief outline of th	e course:							
Recommended lite	erature:							
Course language:								
Course assessmen Total number of as		s: 49						
A	В	С	D	E	FX			
85.71	4.08	2.04	2.04	2.04	4.08			
Provides: Mgr. On	drej Kalina, Ph	D.	·	·				
Date of last modif	ication: 21.08.	2017						
Approved: Guaran	teeprof. RNDr	. Jozef Gonda.	DrSc.					

University: P. J. Šafá	rik University in Košice					
Faculty: Faculty of S	Faculty: Faculty of Science					
Course ID: ÚCHV/ USOL/09	5 8 5					
Course type, scope a Course type: Lectur Recommended cou Per week: 0 / 2 Per Course method: pre	re / Practice rse-load (hours): study period: 0 / 28					
Number of credits: 3	3					
Recommended seme	ster/trimester of the cours	e:				
Course level: II.						
Prerequisities:						
Conditions for cours Test: 7. and 14. week	e completion: - structure determination of	unknown compounds				
Learning outcomes: The main goal of the problems.	subject is to have the ability	to solve the complex assignment NMR				
TOCSY experiments	ical shifts. ects: Spin-spin coupling - ho).	omonuclear experiments (1D and 2D COSY and				
	ects: NOE (1D and 2D NOE relation experiments - HSQC					
5. Strategies for assig	gning resonances to atom wi	thin a molecule.				
 6. Strategies for elucit 7. FID processing - M 	idating unknown molecular s	structures.				
1 0	al data for publication.					
Recommended litera	nture:					
Course language:						
Course assessment Total number of asse	ssed students: 81					
	abs	n				
	100.0	0.0				
Provides: RNDr. Má	ria Vilková, PhD.					
Date of last modifica	tion: 26.02.2018					
Approved: Guarante	eprof. RNDr. Jozef Gonda, I	DrSc.				
	- ,					

Faculty: Faculty							
	y of Science						
Course ID: ÚCHV/ Course name: Basic cheminformatics tools CCI/04							
Course type, sc Course type: I Recommended Per week: 2 / 1 Course metho	Lecture / Practi l course-load (l Per study pe	ce (hours):					
Number of crea	lits: 2						
Recommended	semester/trim	ester of the cours	e:				
Course level: II	•						
Prerequisities:							
Conditions for 3 individual pro	1	etion:					
representation a	nd use of chen d computation	es. The class will on nical structure info , and handling of la	rmation, comput	er-aided drug des	sign, 3D		
Representing 2 Representing 31 systems, Electro	D structures, D structures, 31 onic laboratory	2D chemical dat D visualization & c notebooks, Chem hemical information	omputation, Lab	oratory informati	on managemen		
Representing 2 Representing 3 systems, Electro web service tech Recommended Johann Gasteige Weinheim, 2003 Andrew Leach Publishers, Dor	D structures, 3 D structures, 3 D onic laboratory hnologies for c literature: er & Thomas E & Valerie Gille drecht, NL, 20	D visualization & c notebooks, Chem hemical information Engel (eds.), Chem- et, An Introduction	omputation, Laborical informatics	oratory informati software develop Fextbook. Wiley-	on managemen oment,Emerging VCH,		
Representing 2 Representing 3 systems, Electro web service tech Recommended Johann Gasteige Weinheim, 2003 Andrew Leach	D structures, D structures, 31 Donic laboratory hnologies for c literature: er & Thomas E & Valerie Gille drecht, NL, 200 ge:	D visualization & c notebooks, Chem hemical information Engel (eds.), Chem et, An Introduction 03.	omputation, Laborical informatics	oratory informati software develop Fextbook. Wiley-	on managemen oment,Emerging VCH,		
Representing 2 Representing 3I systems, Electro web service tech Recommended Johann Gasteige Weinheim, 2003 Andrew Leach o Publishers, Dor Course language	D structures, 3D D structures, 3D Donic laboratory hnologies for c literature: er & Thomas E & Valerie Gille drecht, NL, 200 ge: e and english la ment	D visualization & c notebooks, Chem hemical information Engel (eds.), Chem- et, An Introduction 03.	omputation, Laborical informatics	oratory informati software develop Fextbook. Wiley-	on managemen oment,Emerging VCH,		
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University: P. J. Šafárik University in Košice							
Faculty: Faculty of S	Faculty: Faculty of Science						
Course ID: ÚTVŠ/ ÚTVŠ/CM/13							
Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: Per study period: 36s Course method: present							
Number of credits: 2							
Recommended seme	ster/trimester of the course	2:					
Course level: I., II.							
Prerequisities:							
Conditions for course Conditions for course Attendance	-						
Learning outcomes: Learning outcomes: Students will be provided an overview of possibilities how to spend leisure time in seaside conditions actively and their skills in work and communication with clients will be improved. Students will acquire practical experience in organising the cultural and art-oriented events, with the aim to improve the stay and to create positive experiences for visitors.							
 Brief outline of the course: Brief outline of the course: Basics of seaside aerobics Morning exercises Pilates and its application in seaside conditions Exercises for the spine Yoga basics Sport as a part of leisure time Application of projects of productive spending of leisure time for different age and social groups (children, young people, elderly) Application of seaside cultural and art-oriented activities in leisure time 							
Recommended litera	ture:						
Course language:							
Course assessment Total number of asses	ssed students: 33						
	abs	n					
	12.12	87.88					
Provides: Mgr. Alena	Buková, PhD., Mgr. Agata	Horbacz, PhD.					
Date of last modifica	tion: 18.08.2017						