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	Šafárik Universi				
Faculty: Faculty					
Course ID: CJP PFAJAKA/07	Course na	me: Academic	English		
Per week: 2 Pe	-	ours): 28			
Number of ECT	S credits: 2				
Recommended	semester/trimes	ter of the cours	se:		
Course level: I.,	II., N				
Prerequisities:					
epidemiological Presentation on Final evaluation Grading scale: A Learning outco	situation – onlin chosen topic (in - average assess A 93-100%, B 86 mes:	e) case of distance nent of test (40	in case of dist e learning - online %), essay (30%) %, D 72-78%, E e	e thorugh MS Tea and presentation	ams) (30%).
Brief outline of	the course:				
T. Armer :Camb M. McCarthy M Zemach, D.E, R Olsen, A. : Acti www.bbclearnin	nic Encounters, C pridge English for [., O'Dell F Ac umisek, L.A: Ac ve Vocabulary, Po	r Scientists, CU ademic Vocabu ademic Writing earson, 2013	lary in Use, CUP 5, Macmillan 2003		
Course languag English languag	e: e, level B2 accor	ding to CEFR.			
Notes:					
Course assessm Total number of	ent assessed student	s: 380			
А	В	С	D	Е	FX
			1	1	1
33.68	22.11	15.53	10.0	6.58	12.11
	22.11 Viktória Mária Sl		10.0	6.58	12.11

		ity in Košice			
Faculty: Faculty	y of Science				
Course ID: ÚM ALGa/10	V/ Course na	me: Algebra I			
Course type: I Recommended	ope and the met Lecture / Practice d course-load (h 3 Per study perio d: present	ours):			
Number of EC	FS credits: 7				
Recommended	semester/trimes	ster of the cours	e: 3.		
Course level: I.					
Prerequisities:					
	course completi the results from th		n view of the res	ults of the writte	en and oral final
	knowledge from	•	concerning divi le to apply it in co		•
•		-	ations, Gauss el rule.	imination. Maps	s, permutations.
Divisibility in 2 Computing with Recommended T.S Blyth, E.F. 1 K. Jänich: Linea	Z. Fields. Systen n matrices. Deter literature: Robertson: Basic ar algebra, Spring	minants, Cramer	rule. Springer Verlag, 2		s, permutations.
Divisibility in 2 Computing with Recommended T.S Blyth, E.F. 1	Z. Fields. Systen n matrices. Deter literature: Robertson: Basic ar algebra, Spring	minants, Cramer	rule. Springer Verlag, 2		s, permutations.
Divisibility in Z Computing with Recommended T.S Blyth, E.F. J K. Jänich: Linea Course languag	Z. Fields. Systen n matrices. Deter literature: Robertson: Basic ar algebra, Spring	minants, Cramer	rule. Springer Verlag, 2		s, permutations.
Divisibility in Z Computing with Recommended T.S Blyth, E.F. J K. Jänich: Linea Course languag Slovak Notes: Course assessm	Z. Fields. System n matrices. Detern literature: Robertson: Basic ar algebra, Spring ge:	minants, Cramer linear algebra, S ger Verlag, 1991	rule. Springer Verlag, 2		s, permutations.
Divisibility in Z Computing with Recommended T.S Blyth, E.F. J K. Jänich: Linea Course languag Slovak Notes: Course assessm	Z. Fields. System n matrices. Deter literature: Robertson: Basic ar algebra, Spring ge:	minants, Cramer linear algebra, S ger Verlag, 1991	rule. Springer Verlag, 2		s, permutations.
Divisibility in 2 Computing with Recommended T.S Blyth, E.F. 1 K. Jänich: Linea Course languag Slovak Notes: Course assessm Total number of	Z. Fields. System n matrices. Detern literature: Robertson: Basic ar algebra, Spring ge: nent f assessed studen	minants, Cramer linear algebra, S ger Verlag, 1991 ts: 1279	rule. Springer Verlag, 2	.001.	
Divisibility in 2 Computing with Recommended T.S Blyth, E.F. 1 K. Jänich: Linea Course languag Slovak Notes: Course assessm Total number of A 11.81 Provides: prof.	Z. Fields. System n matrices. Detern literature: Robertson: Basic ar algebra, Spring ge: nent f assessed studen B 11.65	minants, Cramer linear algebra, S ger Verlag, 1991 ts: 1279 C 19.0 rudenovská, CSc	rule. Springer Verlag, 2 D 17.9 ., RNDr. Igor Fab	E 28.3	FX 11.34
Divisibility in 2 Computing with Recommended T.S Blyth, E.F. 1 K. Jänich: Linea Course languag Slovak Notes: Course assessm Total number of A 11.81 Provides: prof. Janičková, PhD.	Z. Fields. System n matrices. Detern literature: Robertson: Basic ar algebra, Spring ge: nent f assessed studen B 11.65 RNDr. Danica St	minants, Cramer linear algebra, S ger Verlag, 1991 ts: 1279 C 19.0 rudenovská, CSc Rindošová, RNI	rule. Springer Verlag, 2 D 17.9 ., RNDr. Igor Fab	E 28.3	FX 11.34

University: P. J	~	1			
	. Šafárik Univer	sity in Košice			
Faculty: Facult					
Course ID: ÚM ALG3b/10	IV/ Course n	ame: Algebra II f	or informatician	s and physicists	
Course type: I Recommendee	cope and the me Lecture / Practic d course-load (l 2 Per study per d: present	e hours):			
Number of EC	TS credits: 7				
Recommended	semester/trime	ester of the cours	e: 4.		
Course level: I.	, II.				
Prerequisities:	ÚMV/ALGa/10)			
C onditions for Exam	course complet	tion:			
Learning outco To provide deep		on vector spaces, l	inear transforma	tions and Euclide	ean spaces.
- ·	-		and a character		
spaces. The rar tranformations, transformations of linear transfor Affine spaces, s and quadrics. Recommended A. F. Beardon:	hk of a matrix. matrices of su pregular matrice prmations. subspaces and the literature: Algebra and Geo	Linear transforma ums and compos es. Similar matrice heir positions. Euc ometry, Cambridg	tions and their r itions of linear es. Characteristic elidean spaces, th e University Pre	matrices. Operations. tranformations. vectors and char he distance of sub	ions with linear Regular linear acteristic values
spaces. The rar tranformations, transformations of linear transfor Affine spaces, s and quadrics. Recommended A. F. Beardon:	hk of a matrix. matrices of su pregular matrice ormations. subspaces and the literature: Algebra and Geo Mac Lane: A Su	Linear transforma ums and compos es. Similar matrice heir positions. Euc	tions and their r itions of linear es. Characteristic elidean spaces, th e University Pre	matrices. Operations. tranformations. vectors and char he distance of sub	ions with linear Regular linear acteristic values
spaces. The rar tranformations, transformations of linear transfor Affine spaces, s and quadrics. Recommended A. F. Beardon: G. Birkhoff, S.	hk of a matrix. matrices of su pregular matrice ormations. subspaces and the literature: Algebra and Geo Mac Lane: A Su	Linear transforma ums and compos es. Similar matrice heir positions. Euc ometry, Cambridg	tions and their r itions of linear es. Characteristic elidean spaces, th e University Pre	matrices. Operations. tranformations. vectors and char he distance of sub	ions with linear Regular linear acteristic values
spaces. The rar tranformations, transformations of linear transfor Affine spaces, s and quadrics. Recommended A. F. Beardon: G. Birkhoff, S. Course languag Slovak	hk of a matrix. matrices of su pregular matrice ormations. subspaces and the literature: Algebra and Geo Mac Lane: A Su	Linear transforma ums and compos es. Similar matrice heir positions. Euc ometry, Cambridg	tions and their r itions of linear es. Characteristic elidean spaces, th e University Pre	matrices. Operations. tranformations. vectors and char he distance of sub	ions with linear Regular linear acteristic values
spaces. The rar tranformations, transformations of linear transfor Affine spaces, s and quadrics. Recommended A. F. Beardon: G. Birkhoff, S. Course languag Slovak Notes: Course assessm	nk of a matrix. matrices of su s, regular matrice prmations. subspaces and the literature: Algebra and Gen Mac Lane: A Su ge:	Linear transforma ums and compos es. Similar matrice heir positions. Euc ometry, Cambridg urvey of Modern A	tions and their r itions of linear es. Characteristic elidean spaces, th e University Pre	matrices. Operations. tranformations. vectors and char he distance of sub	ions with linear Regular linear acteristic values
spaces. The rar tranformations, transformations of linear transfor Affine spaces, s and quadrics. Recommended A. F. Beardon: G. Birkhoff, S. Course languag Slovak Notes: Course assessm	hk of a matrix. matrices of su s, regular matrice formations. subspaces and the literature: Algebra and Geo Mac Lane: A Su ge:	Linear transforma ums and compos es. Similar matrice heir positions. Euc ometry, Cambridg urvey of Modern A	tions and their r itions of linear es. Characteristic elidean spaces, th e University Pre	matrices. Operations. tranformations. vectors and char he distance of sub	ions with linear Regular linear acteristic values
spaces. The rar tranformations, transformations of linear transfor Affine spaces, s and quadrics. Recommended A. F. Beardon: G. Birkhoff, S. Course languag Slovak Notes: Course assessm Total number of	hk of a matrix. matrices of su s, regular matrice ormations. subspaces and the literature: Algebra and Gen Mac Lane: A Su ge: hent f assessed stude	Linear transforma ums and compos es. Similar matrice heir positions. Euc ometry, Cambridg urvey of Modern A	tions and their i itions of linear es. Characteristic elidean spaces, the ge University Pre Algebra, New Yo	matrices. Operati tranformations. vectors and char he distance of sul ess, 2005 ork 1965	ions with linear Regular linear acteristic values bspaces. Conics
spaces. The rar tranformations, transformations of linear transfor Affine spaces, s and quadrics. Recommended A. F. Beardon: G. Birkhoff, S. Course languag Slovak Notes: Course assessm Total number of A 15.52 Provides: doc. 1	hk of a matrix. matrices of su s, regular matrice ormations. subspaces and the literature: Algebra and Geo Mac Lane: A Su ge: hent f assessed studen B 10.69 RNDr. Roman S	Linear transforma ums and composes. Similar matrices heir positions. Euco ometry, Cambridg urvey of Modern A nts: 290	tions and their i itions of linear es. Characteristic elidean spaces, the ge University Pre Algebra, New Yo D 18.62	matrices. Operative tranformations. vectors and char he distance of sulf ess, 2005 ork 1965 E 31.72	ions with linear Regular linear acteristic values bspaces. Conics FX 10.69
spaces. The rar tranformations, transformations of linear transfor Affine spaces, s and quadrics. Recommended A. F. Beardon: G. Birkhoff, S. Course languag Slovak Notes: Course assessm Total number of A 15.52 Provides: doc. I Janičková, PhD.	hk of a matrix. matrices of su s, regular matrice ormations. subspaces and the literature: Algebra and Geo Mac Lane: A Su ge: hent f assessed studen B 10.69 RNDr. Roman S	Linear transforma ums and composes. Similar matrices heir positions. Euc ometry, Cambridg urvey of Modern A nts: 290 C 12.76 Soták, PhD., RND	tions and their i itions of linear es. Characteristic elidean spaces, the ge University Pre Algebra, New Yo D 18.62	matrices. Operative tranformations. vectors and char he distance of sulf ess, 2005 ork 1965 E 31.72	ions with linear Regular linear acteristic values bspaces. Conics FX 10.69

	COURSE INFORMATION LETTER
University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	science
Course ID: ÚFV/ ABE/18	Course name: Analýza biofyzikálnych experimentov
Course type, scope a Course type: Lectur Recommended cou Per week: 2 / 1 Per Course method: pro	re / Practice rse-load (hours): study period: 28 / 14
Number of ECTS cr	redits: 4
Recommended seme	ester/trimester of the course: 6.
Course level: I.	
Prerequisities:	
Conditions for cours Exam.	se completion:
experiments, verifica	overview of the basic knowledge related to data evaluation in biophysical tion of hypotheses and discrimination between different models. Students will for experimental data computer processing.
knowing the uncerta discrepancy, compare 2. Checking relations uncertainties in direct differences, products variable, experimenta 3. Analysis of random the normal distribution	ntal measurements, physical units, errors and uncertainties, importance of ainties, estimating uncertainties in repeatable experiments, best estimates, ison of measured and accepted values, comparison of two measured numbers. ships with a graph, fractional uncertainties, multiplying two measured numbers, ct measurements, the square-root rule for counting experiments, sums and a and quotients, independent uncertainties in a sum, arbitrary functions of one al examples. In uncertainties, random and systematic errors, the mean and standard deviation, on, hystograms and distributions, limiting distributions, the standard deviation mit, rejection of data, Chauvenet's criterion, weighted averages, experimental

5. Covariance and correlation, covariance in error propagation, coefficient of linear correlation, quantitative significance r, autocorrelation, cross-correlation, use of correlation functions in monitoring the dynamics of individual molecules.

6. The binomial distribution, probabilities in dice throwing, definition of the binomial distribution, the Gauss distribution of random errors, testing of hypothesis, the properties of the Poisson distribution, applications, Chi squared testing, degrees of freedom and reduced chi squared, probabilities for chi squared, experimental examples, solutions, using Excel calculations.

7. Noise sources in biophysical experiments, mechanical noise, electrical noise (thermal noise, shot noise, interference), noise sources in optical imaging experiments, noise characterisctics: color, power spectrum, signal-to-noise ratio, methods for noise reduction and spectral filtration.

8. Computer processing of experimental data (Origin, Igor), the usage of fitting algorithms, statistical analysis, data plotting in graphs, 3D graphs, statistical graphs, figure preparation for publications.

9. Matlab/Octave: a tool for numerical modeling, complex data fitting with shared parameters, examples and applications.

10. Python: simple still complex tool for data analysis, large set of libraries, application examples: polynomial fitting, Fourier transformation, machine learning.

11. Data analysis in the field of spectral data, data smoothing (moving average, Savitzky-Golay filter, Fourier filter), background subtraction (high-order polynomial fitting, rolling ball algorithm, iterative methods), searching for peak position and intensity, complex spectra as a linear combination of simple contributions.

12. Image processing: using Python for image processing, imaging the relevant regions (ROI – region of interest) and firther analysis, binary thresholding, region separation by color, intensity normalization, border detection.

Recommended literature:

1. J.R. Taylor. An Introduction to Error Analysis: The Study of Uncertainties in Physical Measurements, University Science Books, 1997.

2. J. Mandel. The Statistical Analysis of Experimental Data, Dover Publications. 19643. E.J. Billo. Excel for Chemist, Wiley, 2011

Course language:

Slovak, English.

Notes:

Course assessment

Total number of assessed students: 0

A	В	С	D	Е	FX		
0.0	0.0	0.0	0.0	0.0	0.0		
Provides: doc. Mgr. Gregor Bánó, PhD., RNDr. Gabriel Žoldák, PhD.							
Date of last mo	dification: 28.06	5.2021					

University: P. J.	Šafárik Univers	ity in Košice			
Faculty: Faculty	y of Science				
Course ID: ÚF BPO/14	V/ Course na	ame: Bachelor T	hesis and its Def	ènce	
Course type, sc Course type: Recommended Per week: Per Course metho Number of EC	l course-load (h study period: d: present				
		ster of the cours	6•		
Course level: I.					
Prerequisities:					
Conditions for	-	on: led basedon subn	nitting the bache	lor thesis.	
Learning outco	mes:				
Brief outline of Presentation of professional con	the bachelor the	esis results, answ	ering questions	of the reviewer a	and members of
Recommended	literature:				
Course languag Slovak or Engli					
Notes:					
Course assessm Total number of	ent f assessed studen	ts: 44			
А	В	С	D	Е	FX
90.91	4.55	4.55	0.0	0.0	0.0
Provides:		1	1		
Date of last mo	dification: 03.05	5.2015			

University: P					
Faculty: Facult					
Course ID: ÚC BAM1/00		ame: Biochemica	l Analytical Me	thods	
Course type: Recommende	cope and the me Lecture / Practice ed course-load (h 1 Per study peri od: present	e 1ours):			
Number of EC	CTS credits: 4				
Recommended	l semester/trime	ster of the cours	e: 5.		
Course level: I	., II.			_	
Prerequisities:					
Conditions for Written examin	course complet	ion:			
Learning outco Brief outline o					
Brief outline o General princi spectroscopy. application of o Ions, electrode Recommended D. J. Holme, H S. R. Mikkelse	f the course: iples of analytic Centrifugation an electrophoresis. A s and biosensors. I literature: I. Peck: Analytica en, E. Cortón: Bio	al Biochemistry, 1 panalytical Chemi	998 stry, 2004	of biomolecules. Immunochemica	Principles and I techniques
Brief outline o General princi spectroscopy. application of o Ions, electrode Recommended D. J. Holme, H S. R. Mikkelse V. A. Gault, N.	f the course: iples of analytic Centrifugation an electrophoresis. A s and biosensors. I literature: I. Peck: Analytica en, E. Cortón: Bio . H. McClenaghar	nd separation. Cl Application of ma al Biochemistry, 1	998 stry, 2004	of biomolecules. Immunochemica	Principles and I techniques
Brief outline o General princi spectroscopy. application of o Ions, electrode Recommended D. J. Holme, H S. R. Mikkelse V. A. Gault, N. applications, 20	f the course: iples of analytic Centrifugation an electrophoresis. A s and biosensors. I literature: I. Peck: Analytica m, E. Cortón: Bio . H. McClenaghan 009	nd separation. Cl Application of ma al Biochemistry, 1 panalytical Chemi	998 stry, 2004	of biomolecules. Immunochemica	Principles and I techniques
Brief outline o General princi- spectroscopy. application of o Ions, electrode Recommended D. J. Holme, H S. R. Mikkelse V. A. Gault, N. applications, 20 Course langua	f the course: iples of analytic Centrifugation an electrophoresis. A s and biosensors. I literature: I. Peck: Analytica m, E. Cortón: Bio . H. McClenaghan 009	nd separation. Cl Application of ma al Biochemistry, 1 panalytical Chemi	998 stry, 2004	of biomolecules. Immunochemica	Principles and I techniques
Brief outline o General princi- spectroscopy. application of o Ions, electrode Recommended D. J. Holme, H S. R. Mikkelse V. A. Gault, N. applications, 20 Course langua Notes: Course assessm	f the course: iples of analytic Centrifugation an electrophoresis. A s and biosensors. I literature: I. Peck: Analytica m, E. Cortón: Bio . H. McClenaghan 009	nd separation. Cl Application of ma al Biochemistry, 1 banalytical Chemi n: Understanding	998 stry, 2004	of biomolecules. Immunochemica	Principles and I techniques
Brief outline o General princi- spectroscopy. application of o Ions, electrode Recommended D. J. Holme, H S. R. Mikkelse V. A. Gault, N. applications, 20 Course langua Notes: Course assessm	f the course: iples of analytic Centrifugation an electrophoresis. A s and biosensors. I literature: I. Peck: Analytica en, E. Cortón: Bio . H. McClenaghan 009 ge: nent	nd separation. Cl Application of ma al Biochemistry, 1 banalytical Chemi n: Understanding	998 stry, 2004	of biomolecules. Immunochemica	Principles and I techniques
Brief outline o General princi- spectroscopy. application of o Ions, electrode Recommended D. J. Holme, H S. R. Mikkelse V. A. Gault, N. applications, 20 Course langua Notes: Course assess Total number o	f the course: iples of analytic Centrifugation an electrophoresis. A s and biosensors. I literature: I. Peck: Analytica en, E. Cortón: Bio . H. McClenaghan 009 ge: nent	nd separation. Cl Application of ma al Biochemistry, 1 panalytical Chemi n: Understanding	nromatography ss spectrometry. 998 stry, 2004 Bioanalytical Cl	of biomolecules. Immunochemica hemistry: Princip	Principles and I techniques
Brief outline o General princi- spectroscopy. application of o Ions, electrode Recommended D. J. Holme, H S. R. Mikkelse V. A. Gault, N. applications, 20 Course langua Notes: Course assess Total number o A 41.54	f the course: iples of analytic Centrifugation and electrophoresis. A s and biosensors. I literature: I. Peck: Analytica en, E. Cortón: Bio . H. McClenaghar 009 age: ment of assessed studer B	nd separation. Cl Application of ma al Biochemistry, 1 banalytical Chemi n: Understanding nts: 65 C 13.85	promatography of ss spectrometry. 998 stry, 2004 Bioanalytical Cl	ef biomolecules. Immunochemica hemistry: Princip	Principles and I techniques
Brief outline o General princi- spectroscopy. application of o Ions, electrode Recommended D. J. Holme, H S. R. Mikkelsee V. A. Gault, N. applications, 20 Course langua Notes: Course assess Total number o A 41.54 Provides: RNE	f the course: iples of analytic Centrifugation electrophoresis. s and biosensors. I literature: I. Peck: I. Peck: Analytica on, E. Cortón: Bio prese B 21.54	nd separation. Cl Application of ma al Biochemistry, 1 banalytical Chemi n: Understanding nts: 65 C 13.85 bač, PhD.	promatography of ss spectrometry. 998 stry, 2004 Bioanalytical Cl	ef biomolecules. Immunochemica hemistry: Princip	Principles and I techniques

University: P. J	. Šafárik Univers	sity in Košice			
Faculty: Facult	y of Science				
Course ID: ÚC PBC2/99	HV/ Course na	ame: Biochemist	ry Practical		
Course type: I Recommended	d course-load (h er study period:	ours):			
Number of EC	ΓS credits: 4				
Recommended	semester/trimes	ster of the cours	e: 3.		
Course level: I.					
Prerequisities:					
Learning outco To allow studen used in a bioche electrophoresis, quantitative and Brief outline of The most imp	nts to get practic emical research: U , isolation of mail d qualitative deter 7 the course: ortant biochemic	al experience in UV/VIS spectrop cromolecules and rmination.	hotometry, thin d substances fro nethods. The q	echniques and me layer chromatogr om biological ma qualitative tests f	aphy (TLC), generation in the second
activity, determ effect of a subs	ination of the fi	rst order rate co on on initial rate	nstant, calculati	ction: determination ions of math moo termination of Kr	dels (examples)
, , ,		· •		ses from biochem	istry, 2007,
Course languag	ge:				
Notes:					
Course assessm Total number of	ent f assessed studen	ts: 847			
А	В	С	D	Е	FX
57.62	25.62	10.27	4.6	1.65	0.24

University: P. J. Ša	fárik Univers	ity in Košice			
Faculty: Faculty of	Science				
Course ID: ÚFV/ BCHF1/18	Course na	ame: Biochémia	pre fyzikov I		
Course type, scope Course type: Lect Recommended co Per week: 3 / 2 Po Course method: p	ture / Practice ourse-load (h er study peri	e ours):			
Number of ECTS	credits: 6				
Recommended sen	nester/trimes	ster of the cours	e: 2.		
Course level: I.					
Prerequisities:					
Conditions for cou	rse completi	on:			
Learning outcome	s:				
Brief outline of the	e course:				
Recommended lite	erature:				
Course language:					
Notes:					
Course assessment Total number of as		ts: 7			
A	В	С	D	Е	FX
57.14	0.0	14.29	0.0	28.57	0.0
Provides: doc. RNI	Dr. Erik Sedlå	ák, DrSc., RNDr.	Gabriel Žoldák	PhD.	
Date of last modifi	cation: 01.10).2018			
Approved:					

University: P. J. Š	Safárik Univers	ity in Košice			
Faculty: Faculty	of Science				
Course ID: ÚFV/ BCHF2/18	Course na	me: Biochémia	pre fyzikov II		
Course type, scop Course type: Le Recommended Per week: 3 / 2 1 Course method:	cture / Practice course-load (h Per study peri	ours):			
Number of ECTS	S credits: 6				
Recommended se	emester/trimes	ster of the cours	e: 3.		
Course level: I.					
Prerequisities: Ú	FV/BCHF1/18				
Conditions for co	ourse completi	on:			
Learning outcom	ies:				
Brief outline of tl	he course:				
Recommended li	terature:				
Course language	•				
Notes:					
Course assessme Total number of a	-	ts: 4			
A	В	С	D	Е	FX
50.0	0.0	50.0	0.0	0.0	0.0
Provides: doc. RN	NDr. Erik Sedlá	ik, DrSc., RNDr.	Gabriel Žoldák,	PhD.	
Date of last modi	fication: 01.10	0.2018			
Approved:					

University: P. J. Š	afárik Universi	ty in Košice					
Faculty: Faculty of	of Science						
Course ID: ÚFV/ BSSBF/18	Course name: Biofyzika						
Course type, scop Course type: Recommended o Per week: Per s Course method:	course-load (ho tudy period:						
Number of ECTS	S credits: 4						
Recommended se	emester/trimes	ter of the cours	e:				
Course level: I.							
Prerequisities: Ú EMBF2/18,ÚFV/I		JFV/FCH1/02,Ú	FV/BFB1/14,Ú	FV/EMBF1/18,Ú	FV/		
Conditions for co	ourse completio	on:					
Learning outcom	les:						
Brief outline of th	ne course:						
Recommended li	terature:						
Course language							
Notes:							
Course assessmen Total number of a		s: 0					
A	В	С	D	Е	FX		
0.0	0.0	0.0	0.0	0.0	0.0		
Provides:	L			•	•		
Date of last modi	fication: 19.10	.2018					
Approved:							

University: P. J. Ša	afárik Univers	ity in Košice				
Faculty: Faculty o	f Science					
Course ID: ÚFV/ BFBB/18	Course name: Biofyzika v biomedicíne a biotechnológiách					
Course type, scop Course type: Lec Recommended c Per week: 2 Per Course method:	cture ourse-load (h study period:	ours):				
Number of ECTS	credits: 3					
Recommended set	mester/trimes	ster of the cours	e: 2.			
Course level: I.						
Prerequisities:						
Conditions for co	urse completi	on:				
Learning outcome	es:					
Brief outline of th	e course:					
Recommended lit	erature:					
Course language:						
Notes:						
Course assessmen Total number of as		ts: 5				
A	В	С	D	E	FX	
100.0	0.0	0.0	0.0	0.0	0.0	
Provides: doc. RN Žoldák, PhD., doc. Fabriciová, PhD., c	Mgr. Gregor	Bánó, PhD., RNI	Dr. Branislav Br	utovský, CSc., R		
Date of last modif	ication: 05.10	0.2018				
Approved:						

University: P. J. Ša	afárik Univers	ity in Košice			
Faculty: Faculty of	f Science				
Course ID: ÚFV/ BFSb1/18	Course na	me: Biofyzikáln	y seminár I		
Course type, scope Course type: Prac Recommended co Per week: 1 Per s Course method: p	ctice ourse-load (h study period:	ours):			
Number of ECTS	credits: 1				
Recommended ser	nester/trimes	ter of the cours	e: 3.		
Course level: I.					
Prerequisities:					
Conditions for cou	ırse completi	on:			
Learning outcome	es:				
Brief outline of the	e course:				
Recommended lite	erature:				
Course language:					
Notes:					
Course assessmen Total number of as		ts: 4			
A	В	С	D	Е	FX
100.0	0.0	0.0	0.0	0.0	0.0
Provides: doc. Mg	r. Daniel Janc	ura, PhD.			
Date of last modif	ication: 01.10	0.2018			
Approved:					

University: P. J. Ša	afárik Univers	ity in Košice			
Faculty: Faculty of	f Science				
Course ID: ÚFV/ BFSb2/18	Course na	me: Biofyzikálr	y seminár II		
Course type, scope Course type: Prace Recommended co Per week: 1 Per s Course method: p	ctice ourse-load (h study period:	ours):			
Number of ECTS	credits: 1				
Recommended ser	nester/trimes	ster of the cours	e: 4.		
Course level: I.					
Prerequisities:					
Conditions for cou	ırse completi	on:			
Learning outcome	es:				
Brief outline of the	e course:				
Recommended lite	erature:				
Course language:					
Notes:					
Course assessmen Total number of as		ts: 4			
A	В	С	D	Е	FX
100.0	0.0	0.0	0.0	0.0	0.0
Provides: doc. Mg	r. Daniel Janc	ura, PhD.			
Date of last modif	ication: 01.10	0.2018			
Approved:					

University: P. J. Šafärik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ Course name: Bioinorganic Chemistry I BAC1/04 Course type, scope and the method: Course type, scope and the method: Course type. Course type, scope and the method: Course type. Course type, scope and the method: Course type. Course type. Per study period: 28 / 14 Course method: Present Number of ECTS credits: 5 Recommended semester/trimester of the course: 3. Course level: 1, II. Prerequisities: Conditions for course completion: Test or seminar works examination Carning outcomes: The basic knowledges about biometal interactions with biomolecules, biomaterials, biominerals, biocatalysis, metals in biology and medicine, metal-based drugs, toxic metals for biosystems and metals in the environment. Brief outline of the course: Metalic and non-metalic elements and their roles in biological systems (biometals, bulk biological elements, essential trace elements). Biocoordination compounds, bioligands. Biocatalyzers. Oxygen carriers and oxygen transport proteins. Photochemical process. Catalysis of no regulation processes. Calcium biominerals and biomineralization. Toxic metals. Application of meediadion processes. Calcium biominerals and biomineralization. Toxic metals. Application of negulation processes. Calcium biomin		CO	URSE INFORM	MATION LET	ΓER	
Course ID: ÚCHV/ BACI/04 Course name: Bioinorganic Chemistry I BACI/04 Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14 Course method: present Number of ECTS credits: 5 Recommended semester/trimester of the course: 3. Course level: I., II. Prerequisities: Contract on seminar works examination Contract on the course: 3. Learning outcomes: The basic knowledges about biometal interactions with biomolecules, biomaterials, biominerals, bioectalysis, metals in biology and medicine, metal-based drugs, toxic metals for biosystems and metals in the environment. Brief outline of the course: Metalic and non-metalic elements and their roles in biological systems (biometals, bulk biological elements, essential trace elements). Biocoordination compounds, bioligands. Biocalalyzers. Oxygen carriers and oxygen transport proteins. Photochemical process. Catalysis and regulation processes. Calcium biominerals and biomineralization. Toxic metals. Application of knowledge of bioinorganic chemistry in pharmacy, chemotherapy (e.g. platinum complexes in cancer therapy) radiodiagnostics, mineral biotechnology, ceology and in other branches of life. Recommended literature: 1. Shriver D. F., Atkins P. W., Overton T. L., Rourke J.P., Weller M.T., Amstrong F.A.: Shiver & Atkins. Inorganic Chemistry. Oxford University Press, Oxford 2006. 2. Kaim W., Schwederski B.: Bioinorganic Chemistry in Biology. OCP, Oxford 1997. Course language: Notes: Course language: <td>University: P. J.</td> <td>Šafárik Univers</td> <td>ity in Košice</td> <td></td> <td></td> <td></td>	University: P. J.	Šafárik Univers	ity in Košice			
BAC1/04 Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per weck: 2 / 1 Per study period: 28 / 14 Per weck: 2 / 1 Per study period: 28 / 14 Course method: present Number of ECTS credits: 5 Recommended semester/trimester of the course: 3. Course level: L, II. Prerequisities: Conditions for course completion: Test or seminar works examination Learning outcomes: The basic knowledges about biometal interactions with biomolecules, biomaterials, biominerals, biocatalysis, metals in biology and medicine, metal-based drugs, toxic metals for biosystems and metals in the environment. Brief outline of the course: Metalic and non-metalic elements and their roles in biological systems (biometals, bulk biological elements, essential trace elements). Biocoordination compounds, bioligands. Biocatalyzers. Oxygen carriers and oxygen transport proteins. Photochemical process. Catalysis and regulation processes. Calcium biominerals and biomineralization.Toxic metals. Aplication of knowledge of bioinorganic chemistry in pharmacy, chemotherapy (e.g. platinum complexes in cancer therapy) radiodiagnostics, mineral biotechnology, ecology and in other branches of life. Recommended literature: 1. Kniver D. F., Atkins P. W., Overton T. L., Rourke J.P., Weller M.T., Amstrong F.A.: Shiver & Atkins. Inorganic Chemistry. Oxford University Press, Oxford 2006. 1. Shriver D. F., Atkins P. W., Overton T. L., Rourke J.P., Weller M.T., Amstrong F.A.: Shiver & Atkins. Inorganic Chemistry	Faculty: Faculty	of Science				
Course type: Lecture / Practice Recommended course-load (hours): Per weck: 2 / 1 Per study period: 28 / 14 Course method: present Number of ECTS credits: 5 Recommended semester/trimester of the course: 3. Course level: L, II. Prerequisities: Conditions for course completion: Test or seminar works examination Learning outcomes: The basic knowledges about biometal interactions with biomolecules, biomaterials, biominerals, biocatalysis, metals in biology and medicine, metal-based drugs, toxic metals for biosystems and metals in the environment. Brief outline of the course: Metalic and non-metalic elements and their roles in biological systems (biometals, bulk biological elements, essential trace elements). Biocoordination compounds, bioligands. Biocatalyzers. Oxygen carriers and oxygen transport proteins. Photochemical process. Catalysis and regulation processes. Calcium biominerals and biomineralization. Toxic metals. Application of knowledge of bioinorganic chemistry in pharmacy, chemotherapy (e.g. platinum complexes in cancer therapy) radiodiagnostics, mineral biotechnology, ecology and in other branches of life. Recommended literature: 1. Shriver D. F., Atkins P. W., Overton T. L., Rourke J.P., Weller M.T., Amstrong F.A.; Shiver & Atkins. Inorganic Chemistry. Oxford University Press, Oxford 2006. 2. Kaim W., Schwederski B.: Bioinorganic Chemistry: Inorganic Elements in the Chemistry of Life. Wiley, Chichester 1998.		IV/ Course na	me: Bioinorgani	ic Chemistry I		
Recommended semester/trimester of the course: 3. Course level: 1., II. Prerequisities: Conditions for course completion: Test or seminar works examination Learning outcomes: The basic knowledges about biometal interactions with biomolecules, biomaterials, biominerals, biocatalysis, metals in biology and medicine, metal-based drugs, toxic metals for biosystems and metals in the environment. Brief outline of the course: Metalic and non-metalic elements and their roles in biological systems (biometals, bulk biological elements, essential trace elements). Biocoordination compounds, bioligands. Biocatalyzers. Oxygen cariers and oxygen transport proteins. Photochemical process. Catalysis and regulation processes. Calcium biominerals and biomineralization. Toxic metals. Application of knowledge of bioinorganic chemistry in pharmacy, chemotherapy (e.g. platinum complexes in cancer therapy) radiodiagnostics, mineral biotechnology, ecology and in other branches of life. Recommended literature: 1. Shriver D. F., Atkins P. W., Overton T. L., Rourke J.P., Weller M.T., Amstrong F.A.: Shiver & Atkins. Inorganic Chemistry: Inorganic Elements in the Chemistry of Life. Wiley, Chichester 1998. 3. Wilkins P. C., Wilkins R. G.: Inorganic Chemistry in Biology. OCP, Oxford 1997. Course language: Notes: Course assessment <	Course type: La Recommended Per week: 2 / 1	ecture / Practice course-load (h Per study perio	ours):			
Course level: 1., II. Prerequisities: Conditions for course completion: Test or seminar works examination Learning outcomes: The basic knowledges about biometal interactions with biomolecules, biomaterials, biominerals, biocatalysis, metals in biology and medicine, metal-based drugs, toxic metals for biosystems and metals in the environment. Brief outline of the course: Metalic and non-metalic elements and their roles in biological systems (biometals, bulk biological elements, essential trace elements). Biocoordination compounds, bioligands. Biocatalyzers. Oxygen carriers and oxygen transport proteins. Photochemical process. Catalysis and regulation processes. Calcium biominerals and biomineralization. Toxic metals. Application of knowledge of bioinorganic chemistry in pharmacy, chemotherapy (e.g. platinum complexes in cancer therapy) radiodiagnostics, mineral biotechnology, ecology and in other branches of life. Recommended literature: 1. Shriver D. F., Atkins P. W., Overton T. L., Rourke J.P., Weller M.T., Amstrong F.A.: Shiver & Atkins. Inorganic Chemistry. Oxford University Press, Oxford 2006. 2. Kaim W., Schwederski B.: Bioinorganic Chemistry in Biology. OCP, Oxford 1997. Course language: Notes: Course assessment Total number of assessed students: 304 A B	Number of ECT	S credits: 5				
Prerequisities: Conditions for course completion: Test or seminar works examination Learning outcomes: The basic knowledges about biometal interactions with biomolecules, biomaterials, biominerals, biocatalysis, metals in biology and medicine, metal-based drugs, toxic metals for biosystems and metals in the environment. Brief outline of the course: Metalic and non-metalic elements and their roles in biological systems (biometals, bulk biological elements, essential trace elements). Biocoordination compounds, bioligands. Biocatalyzers. Oxygen carriers and oxygen transport proteins. Photochemical process. Catalysis and regulation processes. Calcium biominerals and biomineralization. Toxic metals. Application of knowledge of bioionoganic chemistry in pharmacy, chemotherapy (e.g. platinum complexes in cancer therapy) radiodiagnostics, mineral biotechnology, ecology and in other branches of life. Recommended literature: 1. Shriver D. F., Atkins P. W., Overton T. L., Rourke J.P., Weller M.T., Amstrong F.A.: Shiver & Atkins. Inorganic Chemistry. Oxford University Press, Oxford 2006. 2. Kaim W., Schwederski B.: Bioinorganic Chemistry: Inorganic Elements in the Chemistry of Life. Wiley, Chichester 1998. 3. Wilkins P. C., Wilkins R. G.: Inorganic Chemistry in Biology. OCP, Oxford 1997. Course language: Notes: C D E FX <td>Recommended s</td> <td>semester/trimes</td> <td>ster of the cours</td> <td>e: 3.</td> <td></td> <td></td>	Recommended s	semester/trimes	ster of the cours	e: 3.		
Conditions for course completion: Test or seminar works examination Learning outcomes: The basic knowledges about biometal interactions with biomolecules, biomaterials, biominerals, biocatalysis, metals in biology and medicine, metal-based drugs, toxic metals for biosystems and metals in the environment. Brief outline of the course: Metalic and non-metalic elements and their roles in biological systems (biometals, bulk biological elements, essential trace elements). Biocoordination compounds, bioligands. Biocatalyzers. Oxygen carriers and oxygen transport proteins. Photochemical process. Catalysis and regulation processes. Calcium biominerals and biomineralization. Toxic metals. Application of knowledge of bioinorganic chemistry in pharmacy, chemotherapy (e.g. platinum complexes in cancer therapy) radiodiagnostics, mineral biotechnology, ecology and in other branches of life. Recommended literature: 1. L, Rourke J.P., Weller M.T., Amstrong F.A.: Shiver & Atkins. Inorganic Chemistry: Oxford University Press, Oxford 2006. 2. Kaim W., Schwederski B.: Bioinorganic Chemistry: Inorganic Elements in the Chemistry of Life. Wiley, Chichester 1998. 3. Wilkins P. C., Wilkins R. G.: Inorganic Chemistry in Biology. OCP, Oxford 1997. Course language: Notes: A B C D E FX	Course level: I.,	II.				
Test or seminar works examination Learning outcomes: The basic knowledges about biometal interactions with biomolecules, biomaterials, biominerals, biocatalysis, metals in biology and medicine, metal-based drugs, toxic metals for biosystems and metals in the environment. Brief outline of the course: Metalic and non-metalic elements and their roles in biological systems (biometals, bulk biological elements, essential trace elements). Biocoordination compounds, bioligands. Biocatalyzers. Oxygen carriers and oxygen transport proteins. Photochemical process. Catalysis and regulation processes. Calcium biominerals and biomineralization. Toxic metals. Application of knowledge of bioinorganic chemistry in pharmacy, chemotherapy (e.g. platinum complexes in cancer therapy) radiodiagnostics, mineral biotechnology, ecology and in other branches of life. Recommended literature: 1. Shriver D. F., Atkins P. W., Overton T. L., Rourke J.P., Weller M.T., Amstrong F.A.: Shiver & Atkins. Inorganic Chemistry. Oxford University Press, Oxford 2006. 2. Kaim W, Schwederski B.: Bioinorganic Chemistry: Inorganic Elements in the Chemistry of Life. Wiley, Chichester 1998. 3. Wilkins P. C., Wilkins R. G.: Inorganic Chemistry in Biology. OCP, Oxford 1997. Course language: Notes: Caurse assessment Total number of assessed students: 304	Prerequisities:					
The basic knowledges about biometal interactions with biomolecules, biomaterials, biominerals, biocatalysis, metals in biology and medicine, metal-based drugs, toxic metals for biosystems and metals in the environment. Brief outline of the course: Metalic and non-metalic elements and their roles in biological systems (biometals, bulk biological elements, essential trace elements). Biocoordination compounds, bioligands. Biocatalyzers. Oxygen carriers and oxygen transport proteins. Photochemical process. Catalysis and regulation processes. Calcium biominerals and biomineralization. Toxic metals. Application of knowledge of bioinorganic chemistry in pharmacy, chemotherapy (e.g. platinum complexes in cancer therapy) radiodiagnostics, mineral biotechnology, ecology and in other branches of life. Recommended literature: 1. Shriver D. F., Atkins P. W., Overton T. L., Rourke J.P., Weller M.T., Amstrong F.A.: Shiver & Atkins. Inorganic Chemistry. Oxford University Press, Oxford 2006. 2. Kaim W., Schwederski B.: Bioinorganic Chemistry: Inorganic Elements in the Chemistry of Life. Wiley, Chichester 1998. 3. Wilkins P. C., Wilkins R. G.: Inorganic Chemistry in Biology. OCP, Oxford 1997. Course language: Notes: Course assessment Total number of assessed students: 304 A B C D E FX	Test or seminar v	1	on:			
Metalic and non-metalic elements and their roles in biological systems (biometals, bulk biological elements, essential trace elements). Biocoordination compounds, bioligands. Biocatalyzers. Oxygen carriers and oxygen transport proteins. Photochemical process. Catalysis and regulation processes. Calcium biominerals and biomineralization.Toxic metals. Application of knowledge of bioinorganic chemistry in pharmacy, chemotherapy (e.g. platinum complexes in cancer therapy) radiodiagnostics, mineral biotechnology, ecology and in other branches of life. Recommended literature: 1. Shriver D. F., Atkins P. W., Overton T. L., Rourke J.P., Weller M.T., Amstrong F.A.: Shiver & Atkins. Inorganic Chemistry. Oxford University Press, Oxford 2006. 2. Kaim W., Schwederski B.: Bioinorganic Chemistry: Inorganic Elements in the Chemistry of Life. Wiley, Chichester 1998. 3. Wilkins P. C., Wilkins R. G.: Inorganic Chemistry in Biology. OCP, Oxford 1997. Course language: Notes: A B C D E FX	The basic knowl biocatalysis, met	ledges about bio tals in biology a			,	, ,
1. Shriver D. F., Atkins P. W., Overton T. L., Rourke J.P., Weller M.T., Amstrong F.A.: Shiver & Atkins. Inorganic Chemistry. Oxford University Press, Oxford 2006. 2. Kaim W., Schwederski B.: Bioinorganic Chemistry: Inorganic Elements in the Chemistry of Life. Wiley, Chichester 1998. 3. Wilkins P. C., Wilkins R. G.: Inorganic Chemistry in Biology. OCP, Oxford 1997. Course language: Notes: Ourse assessment Total number of assessed students: 304 A B C D E FX	Metalic and non- elements, essen Oxygen carriers processes. Calcin bioinorganic che	metalic elemen tial trace elem and oxygen tra um biominerals emistry in pharm	ents). Biocoord nsport proteins. and biomineraliz nacy, chemothera	ination compo Photochemical zation.Toxic me apy (e.g. platinu	unds, bioligands. process. Catalysis tals. Application of um complexes in	Biocatalyzers. and regulation of knowledge of
Notes: Course assessment Total number of assessed students: 304 A B C D E FX	 Shriver D. F., Atkins. Inorganie Kaim W., Schulter, Chie 	Atkins P. W., O c Chemistry. Ox wederski B.: Bio chester 1998.	ford University	Press, Oxford 20 histry: Inorganic	006. Elements in the C	Chemistry of
Course assessmentTotal number of assessed students: 304ABCDEFX	Course language	e:				
Total number of assessed students: 304ABCDEFX	Notes:					
			ts: 304			
	A	В	С	D	Е	FX
41.12 28.29 18.75 5.92 5.59 0.33	41.12	28.29	18.75	5.92	5.59	0.33
Provides: doc. RNDr. Zuzana Vargová, Ph.D.	Provides: doc. R	NDr. Zuzana Va	argová, Ph.D.			

Date of last modification: 03.05.2015

	COURSE INFOR		. 1718				
University: P. J. Šafárik	University in Košice						
Faculty: Faculty of Scien	nce						
Course ID: ÚFV/ Co BSIM1/14							
Course type, scope and Course type: Lecture / Recommended course- Per week: 2 / 2 Per stu Course method: presen	Practice ·load (hours): dy period: 28 / 28						
Number of ECTS credit	t s: 5						
Recommended semester	r/trimester of the cours	se: 6.					
Course level: I., II.							
Prerequisities:							
Conditions for course co Elaboration and presenta programs on project give Q/A part.	tion of the project on give	-	_	-			
Learning outcomes: Introduction to actual pro	oblematics of biomolect	ılar simulations.					
Brief outline of the cour Structural characteristics as flow of biological info mechanisms. Experiment force fields and metho Carlo methods - algorith approaches. Computation reactions, free energy approaches and heuristic	s of biological polymers ormation. 3D-structure a ntal methods of structur ds of classical molecu- nms and paralelization. onal challenges in bion evaluation, protein fol-	nd function of for re determination lar dynamics. M <i>Ab initio</i>	ldamers. Recent v and their limitati Aolecular dynam molecular dynam tions - simulation	view on enzyme ions. Empirical ics and Monte nics and hybrid ns of chemical			
Recommended literatur							
Actual literature recomm	iended by lecturer.						
Course language:							
Notes:							
Course assessment Total number of assessed	l students: 46						
A B	C	D	E	FX			
76.09 8.1	7 10.87	2.17	2.17	0.0			
Provides: doc. RNDr. Jos	zef Uličný, CSc.		<u>.</u>				
Date of last modification	n: 27.03.2020						

E14 E14		sity in Košice			
Faculty: Faculty		.			
Course ID: ÚBE BS1/03	Course n	ame: Biostatistic	2S		
Course type, sco Course type: L Recommended Per week: 2 / 2 Course method	ecture / Practice course-load (h Per study peri	e 1ours):			
Number of ECT	'S credits: 6				
Recommended s	semester/trime	ster of the cours	se: 3., 5.		
Course level: I.					
Prerequisities:					
Conditions for c Written test after Final test (soluti	the 7th week.		owledge)		
their scope of ap of the design of Brief outline of	udents with kno oplication in sta experiments, as the course:	tistical evaluations well.	n of experimenta	tistic methods use al results, and wit	th the principle
empirical distrib One-way and m	utions. Experin ultiple analysis	nental sampling of variance. Tes	from normal dist ts for multiple co	riability of data. tributions. Testing omparisons. Regr f quantitative data	g of hypotheses ression analysis
Recommended Hassard, T. H.: U Snedecor,G.W.,	iterature: Jnderstanding b Cochran,W.G.: .Lee, M.Hernar	biostatistics. Mos Statistical metho	by Year Book, 1 ods. The Iowa sta		es, 1972.
Course languag	e:				
Notes:					
Course assessme		nts: 212			
Total number of	ubbebbed bruder				
Total number of A	B	С	D	Е	FX
i i		C 16.98	D 25.0	E 33.02	FX 12.26
A	B 8.49	16.98			

University: P. J.	Šafárik Univers	ity in Košice			
Faculty: Faculty	of Science				
Course ID: ÚFV BFB1/14	// Course na	me: Cell Bioph	ysics I		
	ecture l course-load (h er study period:	ours):			
Number of ECT	S credits: 4				
Recommended	semester/trimes	ter of the cours	se: 5.		
Course level: I.,	II.				
Prerequisities:					
Conditions for Participation in	course completi problem solutior		t the lectures. Ex	kam.	
-		•	nowledge about	the mechanisms of	of processes that
Thermodynamic membrane prote signals through Metabolic signa	es and active n sins. Oxidative pl synapses. Muscl l pathways: Ger	nembrane transphosphorylation. e contraction. neral description	oort. Classificat Photosynthesis. of signal pathw	is of biologica ion and properti Action potential. vays in cells. Extr nd their role in sig	es of transport Transmission of racellular signal
 van Winkle I. Stein W. D.: 6 Glaser R.: Big Pollard T. D., 		e transport, Aca s, and pumps, A er-Verlag, Heide : Cell biology, S	demic Press, San cademic Press, S lberg 1999 Saunders, Philado	San Diego 1990 elphia 2004	ork 1988
Course languag Slovak	e:				
Notes:					
Course assessm Total number of	ent assessed studen	ts: 159			
А	В	С	D	Е	FX
22.64	25.79	18.24	23.9	8.18	1.26
Provides: RND	: Gabriela Fabric	ciová, PhD.		•	•

Date of last modification: 03.05.2015

cience
Course name: Communicative Competence in English
nd the method: ce rse-load (hours): dy period: 28 mbined, present
edits: 2
ster/trimester of the course:
1
e completion: n class and completed homework assignments. Students are allowed to miss st. Teams), in case of an improved epidemiological situation = on-site teaching. ably in weeks 6/7 and 12/13) and a short oral presentation in English. en online (MS Teams) during online teaching and in class in case of on-site be sent to the course instructor as a video recording.

Final evaluation consists of the scores obtained for the 2 tests (70%) and the presentation (30%). Final grade will be calculated as follows: A 93-100 %, B 86-92%, C 79-85%, D 72-78%, E 65-71%, FX 64 % and less.

Learning outcomes:

Uplatnenie a aktívne používanie svojich teoretických vedomostí v praktických komunikačných situáciách. Zdokonalenie jazykových vedomostí a zručností študenta, rečovej, pragmatickej a vecnej kompetencie, predovšetkým zlepšujú komunikáciu, schopnosť prijímať a formulovať výpovede, efektívne vyjadrovať svoje myšlienky ako aj orientovať sa v obsahovom pláne výpovede. Precvičovanie rečových intencií kontaktných (napr. pozdravy, oslovenia, pozvanie, oslovenie), informatívnych (napr. získavanie a podávanie informácií, vyjadrenie priestorových a časových vzťahov), regulačných (napr. prosba, poďakovanie, zákaz, pochvala, súhlas, nesúhlas) a hodnotiacich (napr. vyjadrenie vlastného názoru, stanoviska, želania, emócií). Výsledkom budovania praktickej jazykovej kompetencie majú byť vedomosti a zručnosti zodpovedajúce požiadavkám a kritériám dokumentu Spoločný európsky referenčný rámec pre vyučovanie jazykov.

Brief outline of the course:

Rodina, jej formy a problémy Vyjadrovanie pocitov a dojmov Dom, bývanie a budúcnosť Formy a dialekty v anglickom jazyku Život v meste a na vidieku Kolokácie a idiomy, zaužívané slovné spojenia Prázdniny a sviatky vo svete

Životné prostredie a ekológia Výnimky zo slovosledu	
Frázové slovesá a ich použitie	
Charakteristiky neformálneho diškurzu	
Recommended literature:	
www.bbclearningenglish.com	
McCarthy M., O'Dell F.: English Vocabulary in Use, Upper-Intermediate. CUP, 1994.	
Misztal M.: Thematic Vocabulary. SPN, 1998.	
Fictumova J., Ceccarelli J., Long T.: Angličtina, konverzace pro pokročilé. Barrister and	
Principal, 2008.	
Peters S., Gráf T.: Time to practise. Polyglot, 2007.	
Jones L.: Communicative Grammar Practice. CUP, 1985.	
Alexander L.G.: Longman English Grammar. Longman, 1988.	
Course language: English language, B2 level according to CEFR	
Notes:	
Course assessment Total number of assessed students: 260	
A B C D E FX	
40.38 22.31 18.85 8.85 6.54 3.08	
Provides: Mgr. Barbara Mitríková, Mgr. Zuzana Naďová	
Date of last modification: 11.02.2021	
Approved:	

University: P. J.	Šafárik Univers	ity in Košice			
Faculty: Faculty	of Science				
Course ID: CJP PFAJGA/07	/ Course na	me: Communica	ative Grammar in	n English	
Course type: F Recommended Per week: 2 Pe	ope and the met Practice I course-load (h er study period: d: combined, pre	ours): 28			
Number of ECT	FS credits: 2				
Recommended	semester/trimes	ster of the cours	e:		
Course level: I.,	II., N				
Prerequisities:					
week), no retak	te. Final evaluati 5%, D 72-78%,		essment of tests	tted). 2 test (5th/ s. Grading scale:	
Brief outline of					
McCarthy, O'De C. Oxengen, C.	nillan Grammar ell: English Voca Latham-Koenig: ematic Vocabular	in Context, Macı bulary in Use, C New English Fi y, Fragment, 199	UP, 1994 le Advanced, Ox	xford 2010	
Course languag					
Notes:					
Course assessm Total number of	ent fassessed studen	ts: 406			
	D	С	D	Е	FX
A	В				17
ĺ	B 18.97	16.75	8.62	5.91	10.1
A 39.66			8.62	5.91	
A 39.66 Provides: Mgr. 1	18.97	vá	8.62	5.91	

University: P. J. Šaf	ärik Univers	ity in Košice			
Faculty: Faculty of	Science				
Course ID: KGER/ NJKG/07	Course na	me: Communica	tive Grammar i	in German Langua	ige
Course type, scope Course type: Pract Recommended cou Per week: 2 Per st Course method: pr	ice 1rse-load (h udy period:	ours):			
Number of ECTS c	redits: 2				
Recommended sem	ester/trimes	ster of the course	2.		
Course level: I., II.					
Prerequisities:					
Conditions for cour	·se completi	on:			
Learning outcomes	:				
Brief outline of the	course:				
Recommended liter	ature:				
Course language:					
Notes:					
Course assessment Total number of ass	essed studen	ts: 54			
A	В	С	D	Е	FX
59.26	11.11	9.26	3.7	9.26	7.41
Provides: Mgr. Blan	ka Jenčíkov	á			
Date of last modific	ation: 03.05	5.2015			
Approved:					

University: P. J. Šafárik University in Košice
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Faculty: Faculty of Science

Course ID: ÚFV/	Course name: Computational Physics I
POF1a/99	

Course type, scope and the method:

Course type: Lecture / Practice

Recommended course-load (hours): Per week: 2 / 1 **Per study period:** 28 / 14

Course method: present

Number of ECTS credits: 4

Recommended semester/trimester of the course: 6.

Course level: I.

Prerequisities: ÚFV/NUM/10

Conditions for course completion:

Continuous evaluation is based on students' presence and activity in the classroom and work on assignments. Examination and all assignments submitted electronically with the attached computer code.

Learning outcomes:

To teach students to use computer as a tool of modeling of physical reality. To present basic deterministic and stochastic approaches to solving mathematical models.

Brief outline of the course:

1. Introduction to dynamical systems.

2. Numerical solution of systems of ordinary differential equations with initial condition.

3. Euler's method, convergence, error estimation and order of the method. One-step methods, Tylortype and Runge-Kuta (RK2, RK4) methods.

4. Multistep methods, general linear method (explicit, implicit). Methods based on numerical quadrature.

5. Boundary value problems for ordinary differential equations.

6. Numerical solution of partial differential equations (PDE). Difference methods, their consistence, convergence and stability. Elliptic PDE.

7. Parabolic PDE, diffusion equation. Explicit and implicit methods.

8. Introduction to the Monte Carlo method. Monte Carlo integration and application in statistical physics.

9. Basics of probability theory. Monte Carlo estimate of mean and standard deviation. Central theorem of Monte Carlo sampling.

10. Simple and importance sampling. Markov chain. Perron-Frobenius theorem. Metropolis algorithm, detailed balance condition.

11. Monte Carlo simulations of lattice spin systems - application to Ising model.

12. Statistical analysis of Monte Carlo data.

Recommended literature:

Basic literature:

- C. Pozrikidis: Num. Comp. in Science and Engineering, Oxford Univ. Press, 2008.

- A.L. Garcia: Numerical Methods for Physics, Prentice-Hall, 1994.

- D. P. Landau, K. Binder: A Guide to Monte Carlo Simulations in Statistical Physics, Cambridge Univ. Press, 2021.

Other literature:

- B. A. Berg: Introduction to Markov Chain Monte Carlo Simulations and Their Statistical Analysis (http://www.worldscibooks.com/etextbook/5904/5904_intro.pdf)

- W. Janke: Monte Carlo Simulations of Spin Systems (http://www.physik.uni-leipzig.de/~janke/ Paper/spinmc.pdf)

Course language:

Notes:

Course asso Total numb	essment er of assesse	d students: 1	19				
А	В	С	D	Е	FX	Ν	Р
31.93	17.65	12.61	16.81	13.45	2.52	0.0	5.04

Provides: prof. RNDr. Milan Žukovič, PhD.

Date of last modification: 30.06.2021

Course language:

Notes:

Course assessm Total number of	ent f assessed studen	ts: 754				
А	В	С	D	Е	FX	
11.54 19.89 32.63 20.03 15.25 0.66						
Provides: doc. RNDr. Rastislav Jendželovský, PhD., RNDr. Zuzana Jendželovská, PhD., RNDr. Jana Vargová, PhD.						
Date of last modification: 16.07.2021						
Approved:						

	COURSE INFORMATION LETTER
University: P. J. Šafá	arik University in Košice
Faculty: Faculty of S	Science
Course ID: CJP/ PFAJ4/07	Course name: English Language of Natural Science
Course type, scope a Course type: Practi Recommended cou Per week: 2 Per stu Course method: pr	ce irse-load (hours): idy period: 28
Number of ECTS cr	redits: 2
Recommended seme	ester/trimester of the course: 4.
Course level: I.	
Prerequisities:	
Active participation is classes at the most (i Continuous assessme 13) and academic pro In order to be admit credit tests. The exam test results represent the other 5 The final grade for the	se completion: y (Online through MS teams) - based on the sylabus in class and completed homework assignments. Students are allowed to miss 2 n case of online form - not attending online class/ assignments not handed in) ent: 2 credit tests taken thorugh MS Teams online(presumably in weeks 6 and esentation in English given through MS Teams online. ted to the final exam, a student has to score at least 65 % as a sum of both a represent 50% of the final grade for the course, continuous assessment results 0% of the final grade. he course will be calculated as follows: C 79-85, D 72-78, E 65-71, FX 64 and less.
in English for specifi with selected phonol competence (familia	lents' language skills (speaking, writing, reading and listening comprehension) c purposes and development of students' language competence (familiarization ogical, lexical and syntactic phenomena), improvement of students' pragmatic rization with selected language functions) and improvement of presentation EFR) with focus on terminology of English for natural science.
 6. Expressing cause a 7. Describing structure 8. Explaining procession 	idying language f scientific language demic study e c terminology and concepts and effect tres ses s, structures and concepts oblem and solution

12. Giving examples	1				
13. Visual aids and nu					
14. Referencing time	-	1 1 11			
Presentation topics rel	lated to stuc	lents study field	S.		
Recommended literat	ture:				
study materials provid	led by the c	ourse instructor			
Redman, S.: English V	Vocabulary	in Use, Pre-inter	rmetdiate, Intern	nediate. Cambrid	ge University
Press, 2003.					
Armer, T.: Cambridge	•		·		
Wharton J.: Academic					
Murphy, R.: English C		-	-	ss, 1994.	
P. Fitzgerald : English			-		
https://worldservice/le	earningengl	ish, https://spect	ator.sme.sk		
www.isllibrary.com					
Course language:	·				
Notes:					
Course assessment				-	
Total number of asses	sed student	s: 2744			
А	В	С	D	Е	FX
38.16	25.4	16.65	9.73	7.87	2.19
Provides: Mgr. Lenka	Klimčákov	á, Mgr. Viktória	n Mária Slovensk	ká, Mgr. Zuzana	Naďová
Date of last modificat	tion: 14.02.	2021			
Approved:					

University. F. J	. Šafárik Univers	ity in Kosice			
Faculty: Facult	y of Science				
Course ID: ÚF ZPU1/03	V/ Course na	me: Essentials of	of UNIX Progran	nming	
Course type: I Recommended	ope and the met Lecture / Practice d course-load (h 2 Per study perio d: present	ours):			
Number of EC	ΓS credits: 4				
Recommended	semester/trimes	ster of the cours	e: 2.		
Course level: I.					
Prerequisities:					
monitoring of s	course completi tudent's program ogram to solve th	ming skills			
	lents with basic p			solving problem puter data proces	
output. Redirect commands. Pro The C program Types of variable and program stu The C++ program Data encapsulat Component pro	ork in Unix type tion of input and cess managemen ning language: pu les. Operators and ructure. Pointers amming languag tion. Polymorphi gramming philos	output. Comman t. Remote termin rogramming envi- d expressions. An and arrays. Struc e. Object oriente sm. Constructor ophy. Make, RC	d line, command hal. ironment in UND rithmetic operation tures. Standard 1 ed programming. and destructor. S, profilers, debu	stems. Terminal. I interpreters and X. Compilers. C la ons. Control struc ibrary. Header fil Data abstraction uggers. Utilisatior ng (LAPACK, M	formats of basic anguage syntax. tures. Functions les. Object. Class. n and creation of
Kernighan, B. V	thew, N., Beginn W., Ritchie, D. M	., The C Program	amming, Comput nming Language , Addison-Wesle	, Prentice Hall, 1	978
				-	
Course languag	ge:				
	ge:				
Notes: Course assessm	ent	ts: 148			
Course languag Notes: Course assessm Total number of A		ts: 148 C	D	E	FX

Provides: RNDr. Branislav Brutovský, CSc.

Date of last modification: 03.05.2015

Approved:

	CO	UNSE INFUR			
University: P. J. Š	afárik Univers	ity in Košice			
Faculty: Faculty o	of Science				
Course ID: ÚBEV ETB1/99	7/ Course na	me: Experiment	al techniques in	Biology	
Course type, scop Course type: Pra Recommended c Per week: 4 Per Course method:	ctice ourse-load (h study period:	ours):			
Number of ECTS	credits: 4				
Recommended se	mester/trimes	ster of the cours	e: 4., 6.		
Course level: I.					
Prerequisities: ÚI	BEV/CYT1/15				
Conditions for co	urse completi	on:			
Learning outcom To provide the stu		knowledge of ba	asic experimenta	l techniques in bio	ology.
Brief outline of th Manipulation with research methods.		nimals. Narcotiz	ing of the anim	als. Operating tec	hniques. Basic
Recommended lit Zutphen, L. F. M., Elsevier, Amsterd	Baumans, V.,	Beynen, A. C.: I	Principles of Lal	poratory Animal S	cience.
Course language:					
Notes:	,				
Course assessmen Total number of as	-	ts: 201			
A	В	С	D	E	FX
48.76	14.43	14.43	4.98	15.92	1.49
Provides: RNDr. J Kisková, PhD., Mg Jendželovský, PhD Jana Vargová, PhD	gr. Vladislav K)., RNDr. Natá	Colarčik, PhD., do	oc. RNDr. Juraj	Ševc, PhD., doc. I	RNDr. Rastislav
Date of last modif	fication: 07.02	2.2017			
Approved:					
r F					

University: P. J. Ša	fárik Univers	ity in Košice				
Faculty: Faculty of	Science					
Course ID: ÚFV/ EMBF1/18	FV/ Course name: Experimentálne metódy biofyziky I					
Course type, scope Course type: Lect Recommended co Per week: 2 Per st Course method: p	ure urse-load (h tudy period: resent	ours):				
Number of ECTS of						
Recommended sem	ester/trimes	ster of the cours	e: 3.	=		
Course level: I.						
Prerequisities:						
Conditions for cou	rse completi	on:				
Learning outcomes	5:					
Brief outline of the	course:					
Recommended lite	rature:					
Course language:						
Notes:						
Course assessment Total number of ass	essed studen	ts: 4				
А	В	С	D	Е	FX	
25.0	50.0	25.0	0.0	0.0	0.0	
Provides: prof. RN	Dr. Pavol Mi	škovský, DrSc.			1	
Date of last modifie	cation: 05.10	0.2018				
Approved:	,			-		

University: P. J. Š	afárik Universi	ty in Košice			
Faculty: Faculty of	of Science				
Course ID: ÚFV/ EMBF2/18	Course na	me: Experiment	álne metódy bio	fyziky II	
Course type, scop Course type: Lea Recommended o Per week: 2 Per Course method:	cture course-load (ho study period:	ours):			
Number of ECTS	credits: 3				
Recommended se	mester/trimes	ter of the cours	e: 4.		
Course level: I.					
Prerequisities:					
Conditions for co	urse completio	on:			
Learning outcom	es:				
Brief outline of th	e course:				
Recommended lit	erature:				
Course language:					
Notes:					
Course assessmer Total number of a		s: 4			
A	В	С	D	Е	FX
50.0	25.0	25.0	0.0	0.0	0.0
Provides: doc. RN CSc., RNDr. Gabr			Gabriel Žoldák,	PhD., RNDr. Ma	arián Fabián,
Date of last modi	fication: 01.10	.2018			
Approved:					

University: P. J. Ša	fárik Universit	y in Košice			
Faculty: Faculty of	f Science				
Course ID: ÚFV/ EMBF3/18	Course nar	ne: Experiment	álne metódy bio	fyziky III	
Course type, scope Course type: Lec Recommended co Per week: 2 Per s Course method: 1	ture ourse-load (ho study period: 2	urs):			
Number of ECTS	credits: 3				
Recommended ser	nester/trimest	er of the cours	e: 5.		
Course level: I.					
Prerequisities:					
Conditions for cou	ırse completio	n:			
Learning outcome	es:				
Brief outline of the	e course:				
Recommended lite	erature:				
Course language:					
Notes:					
Course assessmen Total number of as		s: 0			
А	В	С	D	Е	FX
0.0	0.0	0.0	0.0	0.0	0.0
Provides: doc. Mg Huntošová, PhD.	r. Gregor Bánó	, PhD., RNDr. Z	Zuzana Naďová,	PhD., RNDr. Ver	onika
Date of last modif	ication: 01.07.	2021		_	
Approved:					

University: P. J.	Šafárik Univers	sity in Košice					
Faculty: Faculty	of Science						
Course ID: ÚCH VCHU/15	Course ID: ÚCHV/ Course name: General Chemistry VCHU/15						
Course type, sco Course type: La Recommended Per week: 4 / 2 Course method	ecture / Practice course-load (h Per study peri	e ours):					
Number of ECT	S credits: 7						
Recommended s	emester/trime	ster of the cours	e: 1.				
Course level: I.							
Prerequisities: Ú	JCHV/CHV1/9	9					
Conditions for c Written test in th Oral examination	e middle and th		ester.				
Learning outcom To provide stude chemical bonds,	nts with knowl	-		r electronic struc	ture, theories of		
Brief outline of t Main terms use periodicity and intermolecular in Solutions. Chem Classification of	d in chemistry its effect on nteractions. Che nical equilibriu	the properties o emical structure a m. Basis of che	f elements, radi and physical prop emical thermody	oactivity. Chemperties of matter.	ical bonds and State of matter.		
Recommended In 1. Atkins P., Jone 2. Russel J.B.: G	es L.: Chemical	1 /	, , ,				
Course language	2.						
Notes:							
Course assessme Total number of		nts: 245					
A	В	С	D	Е	FX		
20.41	28.57	31.43	12.24	7.35	0.0		
Provides: prof. R	NDr. Vladimír	Zeleňák, DrSc.					
Provides: prof. R Date of last mod					<u> </u>		

University: P. J. Šafár	rik University in Košice					
Faculty: Faculty of S	cience					
Course ID: ÚFV/Course name: General Physics IVF1a/12						
Course type, scope a Course type: Lectur Recommended cour Per week: 4 / 2 Per Course method: pre	e / Practice rse-load (hours): study period: 56 / 28					
Number of ECTS cr	edits: 7					
Recommended seme	ster/trimester of the course: 1.					
Course level: I.						
Prerequisities:						
1. in the 6th week 2.in the 12th week Final assessment is ba - oral examination	ng the calculus lessons					
Learning outcomes: Basic knowledge abo	ut the mechanics, molecular physics and thermodynamics.					
principle of relativity The motio of rigid bo gases. Kinetic theory	ourse: he calculus, vector algebra. Standards and units. Kinematics. Dynamics. The in the classical mechanics. Gravitation. Mechanics of many-particle systems. odies. Deformation, elasticity. Mechanics of fluids and gases. Laws of ideal . The thermodynamic laws. Statistical character of the second law. Entropy. a in liquids and solids. Phase transitions.					
Veis Š., Maďar J., Ma Bratislava, 1987. Fuka J., Široká M.: O Hlavička A., a kol.: F Hajko V., a kol.:Fyzik Ilkovič D.: Fyzika, S Slaviček V., Wagner J	hture: bó J.: Základy fyziky, VEDA, Bratislava 1983. artišovits V.: Všeobecná fyzika I., Mechanika a molekulová fyzika, ALFA obecná fyzika I / skriptum /, PF Univ. Palackého, Olomouc 1983. Gyzika pre pedagogické fakulty, SPN, Praha 1971. ka v príkladoch, ALFA Bratislava 1983. VTL Bratislava, 1962. J.: Fyzika pro chemiky, SNTL Praha 1971. a, ALFA Bratislava 1982.					
Course language: Slovak						

Course assessm Total number of	nent f assessed studen	ts: 289				
А	В	С	D	Е	FX	
25.26	15.92	19.72	14.88	15.92	8.3	
Provides: doc. RNDr. Zuzana Ješková, PhD.						
Date of last modification: 14.06.2021						
Approved:						

University: P. J.	Šafárik Univers	ity in Košice			
Faculty: Faculty	of Science				
Course ID: ÚFV VF1b/03	// Course na	me: General Ph	ysics II		
Recommended	ecture / Practice course-load (h Per study perio	ours):			
Number of ECT	S credits: 7				
Recommended :	semester/trimes	ster of the cours	se: 2.		
Course level: I.					
Prerequisities: U	ÚFV/VF1a/12				
Conditions for c Two written dist Distance oral ex	ance tests.	on:			
Learning outcom To obtain a gene of this subject.		c electric magne	tic phenomena a	nd ability to solve	basic problems
steady current. C Magnetic field in steady electric fi with ac current.	the free space. V Current in electron in the free space. field. Electromag Multiphase AC of ties of the substa	olytes, semicond The interaction netic induction. current. Rotating ancies. Magnetic	uctors, gasses an of moving charg Energy of magne g magnetic field.	static field. Electron ad vacuum. Therm es with the electri etic field. AC curr Electric effects in iamagnetism and	noelctric effects. c current. Quasi rent and circuits the substances.
Recommended I. S. Grant, W.R		omagnetism, Jol	nn Wiley&Sons,	Ltd, England, 199	90
Course languag english	e:				
Notes:					
Course assessm Total number of		ts: 329			
А	В	С	D	Е	FX
34.65	16.11	15.2	11.85	10.94	11.25
<u> </u>					•
Provides: prof. I Erik Čižmár, PhI		lár, DrSc., doc.	RNDr. Adriana Z	Zeleňáková, PhD.,	, doc. RNDr.

Approved:

University: P. J.	Šafárik Univers	sity in Košice				
Faculty: Faculty	of Science					
Course ID: ÚFV VF1c/12	5					
Recommended	ecture / Practice course-load (h Per study per	e iours):				
Number of ECT	S credits: 7					
Recommended	semester/trime	ster of the cours	e: 3.			
Course level: I.						
Prerequisities:	ÚFV/VF1b/03					
Conditions for Exam+ 2 succes						
Learning outco The objective is		students with the	basis of oscilation	ons, waves and o	ptics.	
Huyghens princ Geometrical opt Light as electro	iple. Reflection ics. Mirrors, ler omagnetic wav	, difraction. Dop as. Fotometry. e. Dispersion, a	pler effect. Wave bsorption, interf	on, waves equati es speed in mater Ference, difractio law of radiation.	rials. Acoustics.	
Recommended 1. A. Hlavička e 2. R.P. Feynmar 3. D. Halliday e 4. J. Fuka, B. H	literature: et al., Fyzika pro a et al., Feynman t al.,Fyzika-Vys avelka, Optika a	pedagogické fak nove prednášky z	culty, SPN, 1971 Fyziky I,II,III, A nice obecné fyzik SPN,1961			
Course languag slovak	e:					
Notes:						
Course assessm Total number of		nts: 139				
А	В	C	D	Е	FX	
30.22	26.62	26.62	11.51	5.04	0.0	
Provides: doc. F	RNDr. Ján Füzer	, PhD.	1		1	
Date of last mo	lification: 03.0	5.2015				

Approved:

Course type, scope and the method: Course type; Lecture / Practice Recommended course-load (hours): Per week: 4 / 2 Per study period: 56 / 28 Course method: present Number of ECTS credits: 7 Recommended semester/trimester of the course: 6. Course level: I. Prerequisities: ÚFV/VF1c/10 and leboÚFV/VF1c/12 Conditions for course completion: written tests exam Learning outcomes: Basic knowledge about the atomic structure and spectra and nuclei, and elementary particles. Basi experimental methods in nuclear physics and passage of nuclear radiation through media. Brief outline of the course: Wave character of particles. De Broglie waves. Experimental evidence for de Broglie waves Structure and models of atoms. Atomic spectra. Magnetic properties of atoms. X-ray spectra. Basi characteristics of the atomic nuclei. Nuclear forces and models. Radioactivity. Applications of radioactivity. Nuclear racations. Elementary particles, basic properties and classification. Types of interactions. Resonances. Cosmic rays. Passage of particles through matter. Detectors. Accelerator Recommended literature: 1. 1. Beiser A., Úvod do moderní fyziky, Praha, 1975. 2. Úlehla I., Suk M., Trka Z.: Atómy, jádra, částice, Praha, 1990. 3. Síleš E., Martinská G.: Všeobecná fyzika IV, skriptá PF UPJŠ, Z. vydanie, Košice, 1992. 4. Vrláková	II. · · · · · · ·	
Course ID: ÚFV/ VF1d/12 Course name: General Physics IV Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 4 / 2 Per study period: 56 / 28 Course method: present Number of ECTS credits: 7 Recommended semester/trimester of the course: 6. Course level: I. Prerequisities: ÚFV/VF1e/10 and leboÚFV/VF1c/12 Conditions for course completion: written tests exam Learning outcomes: Basic knowledge about the atomic structure and spectra and nuclei, and elementary particles. Basi experimental methods in nuclear physics and passage of nuclear radiation through media. Brief outline of the course: Wave character of particles. De Broglie waves. Experimental evidence for de Broglie waves Structure and models of atoms. Atomic spectra. Magnetic properties of atoms. X-ray spectra. Basi characteristics of the atomic nuclei. Nuclear forces and models. Radioactivity. Applications of radioactivity. Nuclear reactions. Elementary particles, basic properties and classification. Types of interactions. Resonances. Cosmic rays. Passage of particles through matter. Detectors. Accelerator Recommended literature: 1. Beiser A., Úvod do moderní fyziky, Praha, 1975. 2. Ulchla L., Suk M., Trka Z.: Atómy, jádra, částice, Praha, 1990. 3. Síleš E., Martinská G.: Všcobecná fyzika IV, skriptá PF UPJŠ, 2. vydanic, Košice, 1992. 4. Vrláková J., Kravčáková A., Vokál S.: Zbierka prikladov z atómovej a jadrovej fyziky, skriptá PF UPJŠ, Košice, 2016. 5. Hajko V. and team of authors, Physics in experiments, Bratislava, 1997. 6. Nosek D., Jádra a částice (Résené příklady), Matfyzpress, MFF UK, Praha 2005, 7. Kravčáková A., Vokál S., Vrláková J., Všcobecná fyzika IV, 1. časť Atómová	-	
VF1d/12 Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 4 / 2 Per study period: 56 / 28 Course method: present Number of ECTS credits: 7 Recommended semester/trimester of the course: 6. Course level: I. Prerequisities: ÚFV/VF1c/10 and leboÚFV/VF1c/12 Conditions for course completion: written tests exam Learning outcomes: Basic knowledge about the atomic structure and spectra and nuclei, and elementary particles. Basic experimental methods in nuclear physics and passage of nuclear radiation through media. Brief outline of the course: Wave character of particles. De Broglie waves. Experimental evidence for de Broglie wave Structure and models of atoms. Atomic spectra. Magnetic properties of atoms. X-ray spectra. Basi characteristics of the atomic nuclei. Nuclear forces and models. Radioactivity. Applications of radioactivity. Nuclear reactions. Elementary particles, basic properties and classification. Types of interactions. Resonances. Cosmic rays. Passage of particles through matter. Detectors. Accelerator Recommended literature: 1. 1. Beiser A., Úvod do moderní fyziky, Praha, 1975. 2. Úlehla I., Suk M., Trka Z.: Atómy, jádra, částice, Praha, 1990. 3. Sílés		
Course type: Lecture / Practice Recommended course-load (hours): Per week: 4 / 2 Per study period: 56 / 28 Course method: present Number of ECTS credits: 7 Recommended semester/trimester of the course: 6. Course level: 1. Prerequisities: ÚFV/VF1c/10 and leboÚFV/VF1c/12 Conditions for course completion: written tests exam Learning outcomes: Basic knowledge about the atomic structure and spectra and nuclei, and elementary particles. Basi experimental methods in nuclear physics and passage of nuclear radiation through media. Brief outline of the course: Wave character of particles. De Broglie waves. Experimental evidence for de Broglie waves Structure and models of atoms. Atomic spectra. Magnetic properties of atoms. X-ray spectra. Basi characteristics of the atomic nuclei. Nuclear forces and models. Radioactivity. Applications of radioactivity. Nuclear reactions. Elementary particles, basic properties and classification. Types of interactions. Resonances. Cosmic rays. Passage of particles through matter. Detectors. Accelerator Recommended literature: 1. Beiser A., Úvod do moderni fyziky, Praha, 1975. 2. Ulehla I., Suk M., Trka Z.: Atómy, jádra, částice, Praha, 1990. 3. Sileš E., Martinská G.: Všeobecná fyzika IV, skriptá PF UPJŠ, 2. vydanie, Košice, 1992. 4. Vrláková J., Kravčáková A., Vokál S.: Zbierka príkladov z atómovej a jadrovej fyziky, skriptá PF UPJŠ, Košice, 2016. 5. Hajko V. and team of authors, Physics in experiments, Bratislava, 1997. 6. Nosek D., Jádra a částice (Řešené příklady), Matfyzpress, MFF UK, Praha 2005, 7. Kravčáková A., Vokál S., Vrláková J., Všeobecná fyzika IV, 1.časť Atómová fyzika, skriptá PI UPJŠ, Košice, 2020. 8. Yang F., Hamilton J.H., Modern Atomic and Nuclear Physics, WSC Singapore, 2010. Course language:	Course ID: UFV/ VF1d/12	Course name: General Physics IV
Recommended semester/trimester of the course: 6. Course level: I. Prerequisities: ÚFV/VF1c/10 and leboÚFV/VF1c/12 Conditions for course completion: written tests exam Learning outcomes: Basic knowledge about the atomic structure and spectra and nuclei, and elementary particles. Basi experimental methods in nuclear physics and passage of nuclear radiation through media. Brief outline of the course: Wave character of particles. De Broglie waves. Experimental evidence for de Broglie wave Structure and models of atoms. Atomic spectra. Magnetic properties of atoms. X-ray spectra. Basi character of the atomic nuclei. Nuclear forces and models. Radioactivity. Applications of radioactivity. Nuclear reactions. Elementary particles, basic properties and classification. Types of interactions. Resonances. Cosmic rays. Passage of particles through matter. Detectors. Accelerator Recommended literature: 1. Beiser A., Úvod do moderní fyziky, Praha, 1975. 2. Úlehla I., Suk M., Trka Z.: Atómy, jádra, částice, Praha, 1990. 3. Síleš E., Martinská G.: Všeobecná fyzika IV, skriptá PF UPJŠ, 2. vydanie, Košice, 1992. 4. Vrláková J., Kravčáková A., Vokál S.: Zbierka prikladov z atómovej a jadrovej fyziky, skriptá PF UPJŠ, Košice, 2016. 5. Hajko V. and team of authors, Physics in experiments, Bratislava, 1997. 6. Nosek D., Jádra a částice (Réšené příklady), Matfyzpress, MFF UK, Praha 2005, 7. Kravčáková A., Vokál S., Vrláková J., Všeobecná fyzika IV, 1.časť Atómová fyzika, skriptá P UPJŠ, Košice, 2020. 8. Yang F., Hamilton J.H., Modern Atomic and Nuclear Physics, WSC Singapore, 2010. Course language:	Course type: Lectu Recommended cou Per week: 4 / 2 Per	re / Practice arse-load (hours): r study period: 56 / 28
Course level: I. Prerequisities: ÚFV/VF1c/10 and leboÚFV/VF1c/12 Conditions for course completion: written tests exam Learning outcomes: Basic knowledge about the atomic structure and spectra and nuclei, and elementary particles. Basi experimental methods in nuclear physics and passage of nuclear radiation through media. Brief outline of the course: Wave character of particles. De Broglie waves. Experimental evidence for de Broglie waves Structure and models of atoms. Atomic spectra. Magnetic properties of atoms. X-ray spectra. Basi characteristics of the atomic nuclei. Nuclear forces and models. Radioactivity. Applications of radioactivity. Nuclear reactions. Elementary particles, basic properties and classification. Typeso interactions. Resonances. Cosmic rays. Passage of particles through matter. Detectors. Accelerator Recommended literature: 1. Beiser A., Úvod do moderní fyziky, Praha, 1975. 2. Úlehla I., Suk M., Trka Z.: Atómy, jádra, částice, Praha, 1990. 3. Síleš E., Martinská G.: Všeobecná fyzika IV, skriptá PF UPJŠ, 2. vydanie, Košice, 1992. 4. Vrláková J., Kravčáková A., Vokál S.: Zbierka prikladov z atómovej a jadrovej fyziky, skriptá PF UPJŠ, Košice, 2016. 5. Hajko V. and team of authors, Physics in experiments, Bratislava, 1997. 6. Nosek D., Jádra a částice (Řešené příklady), Matfyzpress, MFF UK, Praha 2005, 7. Kravčáková A., Vokál S., Vrláková J., Všeobecná fyzika IV, 1.časť Atómová fyzika, skriptá PI UPJŠ, Košice, 2020. 8. Yang F., Hamilton J.H., Modern Atomic and Nuclear Physics, WSC Singapore, 2010. Course language:	Number of ECTS cr	redits: 7
 Prerequisities: ÚFV/VF1c/10 and leboÚFV/VF1c/12 Conditions for course completion: written tests exam Learning outcomes: Basic knowledge about the atomic structure and spectra and nuclei, and elementary particles. Basi experimental methods in nuclear physics and passage of nuclear radiation through media. Brief outline of the course: Wave character of particles. De Broglie waves. Experimental evidence for de Broglie waves structure and models of atoms. Atomic spectra. Magnetic properties of atoms. X-ray spectra. Basi characteristics of the atomic nuclei. Nuclear forces and models. Radioactivity. Applications of radioactivity. Nuclear reactions. Elementary particles, basic properties and classification. Types of interactions. Resonances. Cosmic rays. Passage of particles through matter. Detectors. Accelerator Recommended literature: Beiser A., Úvod do moderní fyziky, Praha, 1975. Úlehla I., Suk M., Trka Z.: Atómy, jádra, částice, Praha, 1990. Síleš E., Martinská G.: Všeobecná fyzika IV, skriptá PF UPJŠ, 2. vydanie, Košice, 1992. Vrláková J., Kravčáková A., Vokál S.: Zbierka príkladov z atómovej a jadrovej fyziky, skriptá PF UPJŠ, Košice, 2016. Hajko V. and team of authors, Physics in experiments, Bratislava, 1997. Nosek D., Jádra a částice (Řešené příklady), Matfyzpress, MFF UK, Praha 2005, 7. Kravčáková A., Vokál S., Vrláková J., Všcobecná fyzika IV, 1.časť Atómová fyzika, skriptá PI UPJŠ, Košice, 2020. Yang F., Hamilton J.H., Modern Atomic and Nuclear Physics, WSC Singapore, 2010. 	Recommended seme	ester/trimester of the course: 6.
 Conditions for course completion: written tests exam Learning outcomes: Basic knowledge about the atomic structure and spectra and nuclei, and elementary particles. Basi experimental methods in nuclear physics and passage of nuclear radiation through media. Brief outline of the course: Wave character of particles. De Broglie waves. Experimental evidence for de Broglie wave Structure and models of atoms. Atomic spectra. Magnetic properties of atoms. X-ray spectra. Basi characteristics of the atomic nuclei. Nuclear forces and models. Radioactivity. Applications of radioactivity. Nuclear reactions. Elementary particles, basic properties and classification. Types of interactions. Resonances. Cosmic rays. Passage of particles through matter. Detectors. Accelerator Recommended literature: 1. Beiser A., Úvod do moderní fyziky, Praha, 1975. 2. Úlehla I., Suk M., Trka Z.: Atómy, jádra, částice, Praha, 1990. 3. Síleš E., Martinská G.: Všeobecná fyzika IV, skriptá PF UPJŠ, 2. vydanie, Košice, 1992. 4. Vrláková J., Kravčáková A., Vokál S.: Zbierka príkladov z atómovej a jadrovej fyziky, skriptá PF UPJŠ, Košice, 2016. 5. Hajko V. and team of authors, Physics in experiments, Bratislava, 1997. 6. Nosek D., Jádra a částice (Řešené příklady), Matfyzpress, MFF UK, Praha 2005, 7. Kravčáková A., Vokál S., Vrláková J., Všeobecná fyzika IV, 1.časť Atómová fyzika, skriptá PI UPJŠ, Košice, 2020. 8. Yang F., Hamilton J.H., Modern Atomic and Nuclear Physics, WSC Singapore, 2010. Course language: 	Course level: I.	
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 Basic knowledge about the atomic structure and spectra and nuclei, and elementary particles. Basic experimental methods in nuclear physics and passage of nuclear radiation through media. Brief outline of the course: Wave character of particles. De Broglie waves. Experimental evidence for de Broglie waves Structure and models of atoms. Atomic spectra. Magnetic properties of atoms. X-ray spectra. Basic characteristics of the atomic nuclei. Nuclear forces and models. Radioactivity. Applications of radioactivity. Nuclear reactions. Elementary particles, basic properties and classification. Types of interactions. Resonances. Cosmic rays. Passage of particles through matter. Detectors. Accelerator Recommended literature: Beiser A., Úvod do moderní fyziky, Praha, 1975. Úlehla I., Suk M., Trka Z.: Atómy, jádra, částice, Praha, 1990. Síleš E., Martinská G.: Všeobecná fyzika IV, skriptá PF UPJŠ, 2. vydanie, Košice, 1992. Vrláková J., Kravčáková A., Vokál S.: Zbierka príkladov z atómovej a jadrovej fyziky, skriptá PF UPJŠ, Košice, 2016. Hajko V. and team of authors, Physics in experiments, Bratislava, 1997. Nosek D., Jádra a částice (Řešené příklady), Matfyzpress, MFF UK, Praha 2005, 7. Kravčáková A., Vokál S., Vrláková J., Všeobecná fyzika IV, 1.časť Atómová fyzika, skriptá Pf UPJŠ, Košice, 2020. Yang F., Hamilton J.H., Modern Atomic and Nuclear Physics, WSC Singapore, 2010. Course language: 	written tests	se completion:
 Wave character of particles. De Broglie waves. Experimental evidence for de Broglie wave Structure and models of atoms. Atomic spectra. Magnetic properties of atoms. X-ray spectra. Basi characteristics of the atomic nuclei. Nuclear forces and models. Radioactivity. Applications of radioactivity. Nuclear reactions. Elementary particles, basic properties and classification. Types of interactions. Resonances. Cosmic rays. Passage of particles through matter. Detectors. Accelerator Recommended literature: Beiser A., Úvod do moderní fyziky, Praha, 1975. Úlehla I., Suk M., Trka Z.: Atómy, jádra, částice, Praha, 1990. Síleš E., Martinská G.: Všeobecná fyzika IV, skriptá PF UPJŠ, 2. vydanie, Košice, 1992. Vrláková J., Kravčáková A., Vokál S.: Zbierka príkladov z atómovej a jadrovej fyziky, skriptá PF UPJŠ, Košice, 2016. Hajko V. and team of authors, Physics in experiments, Bratislava, 1997. Nosek D., Jádra a částice (Řešené příklady), Matfyzpress, MFF UK, Praha 2005, Kravčáková A., Vokál S., Vrláková J., Všeobecná fyzika IV, 1.časť Atómová fyzika, skriptá PI UPJŠ, Košice, 2020. Yang F., Hamilton J.H., Modern Atomic and Nuclear Physics, WSC Singapore, 2010. 	•	out the atomic structure and spectra and nuclei, and elementary particles. Basic
 Beiser A., Úvod do moderní fyziky, Praha, 1975. Úlehla I., Suk M., Trka Z.: Atómy, jádra, částice, Praha, 1990. Síleš E., Martinská G.: Všeobecná fyzika IV, skriptá PF UPJŠ, 2. vydanie, Košice, 1992. Vrláková J., Kravčáková A., Vokál S.: Zbierka príkladov z atómovej a jadrovej fyziky, skriptá PF UPJŠ, Košice, 2016. Hajko V. and team of authors, Physics in experiments, Bratislava, 1997. Nosek D., Jádra a částice (Řešené příklady), Matfyzpress, MFF UK, Praha 2005, Kravčáková A., Vokál S., Vrláková J., Všeobecná fyzika IV, 1.časť Atómová fyzika, skriptá PI UPJŠ, Košice, 2020. Yang F., Hamilton J.H., Modern Atomic and Nuclear Physics, WSC Singapore, 2010. 	Wave character of p Structure and models characteristics of the radioactivity. Nuclea	particles. De Broglie waves. Experimental evidence for de Broglie waves. s of atoms. Atomic spectra. Magnetic properties of atoms. X-ray spectra. Basic e atomic nuclei. Nuclear forces and models. Radioactivity. Applications of ar reactions. Elementary particles, basic properties and classification. Types of
Course language:	 Beiser A., Úvod d Úlehla I., Suk M., Síleš E., Martinska Vrláková J., Kravá PF UPJŠ, Košice, 20 Hajko V. and team Nosek D., Jádra a Kravčáková A., W UPJŠ, Košice, 2020. 	o moderní fyziky, Praha, 1975. Trka Z.: Atómy, jádra, částice, Praha, 1990. á G.: Všeobecná fyzika IV, skriptá PF UPJŠ, 2. vydanie, Košice, 1992. čáková A., Vokál S.: Zbierka príkladov z atómovej a jadrovej fyziky, skriptá 016. n of authors, Physics in experiments, Bratislava, 1997. částice (Řešené příklady), Matfyzpress, MFF UK, Praha 2005, okál S., Vrláková J., Všeobecná fyzika IV, 1.časť Atómová fyzika, skriptá PF
	Course language: slovak and english	
	Notes:	

Course assessm	nent					
Total number of	f assessed studen	ts: 91				
А	В	С	D	Е	FX	
38.46 28.57 13.19 9.89 9.89 0.0						
Provides: prof. Adela Kravčáko		Vokál, DrSc., do	oc. RNDr. Janka V	√rláková, PhD., c	loc. RNDr.	
Date of last mo	dification: 05.08	3.2021				
Approved:						

University: P. J.	Šafárik Univers	sity in Košice				
Faculty: Faculty	of Science					
Course ID: ÚFV UPF1/12	V/ Course name: Introduction to Computational Physics					
Course type, sco Course type: L Recommended Per week: 2 / 1 Course method	ecture / Practice course-load (h Per study peri	e Iours):				
Number of ECT	S credits: 4					
Recommended s	semester/trime	ster of the cours	se: 1.			
Course level: I.						
Prerequisities:						
Conditions for c Elaboration of m Exam and discus	icroreferat on g	given topics.	ne given project.			
processes in con	ecture is to pro nventional com	puters, as well	as to provide le	background of th ess conventional physical processes	possibilities to	
point of view. Ph . Computer mod	es utilised in con nysical limits of eling and physi rnative methods	current computed cal reality. Comp	er technologies (Note that the putational complete the putational complete the putation of the	ional processes / t Moore, Amdahl la exity and paralel cal processors, D	aws ism. Distributed	
Recommended I Actual literature		cturer.				
Course language	e:					
Notes:						
Course assessme Total number of		nts: 40				
А	В	C	D	Е	FX	
90.0		0.0	0.0			
90.0	7.5	0.0	0.0	2.5	0.0	
90.0 Provides: doc. R			0.0	2.5	0.0	
	NDr. Jozef Ulid	čný, CSc.	0.0	2.5	0.0	

University: P. J. Ša	afárik Univers	ity in Košice				
Faculty: Faculty o	f Science					
Course ID: ÚFV/ ZMF/17Course name: Introduction to Mathematics for Physicists						
Course type, scop Course type: Lec Recommended co Per week: 1 / 2 P Course method:	ture / Practice ourse-load (h er study perio	ours):				
Number of ECTS						
Recommended ser	mester/trimes	ster of the cours	e: 1.			
Course level: I.						
Prerequisities:						
Conditions for cou	urse completi	on:				
Learning outcome	es:					
Brief outline of th	e course:					
Recommended lite	erature:					
Course language:						
Notes:						
Course assessmen Total number of as		ts: 264				
Α	В	С	D	Е	FX	
40.53	21.97	17.42	10.98	9.09	0.0	
Provides: RNDr. T	Comáš Lučivja	nský, PhD., doc.	RNDr. Jozef Ha	nč, PhD.		
Date of last modif	ication: 14.09	0.2017				
Approved:	,					

University: P. J. Šafárik University in Košice					
Faculty: Faculty of S	cience				
Course ID: Dek. PF Course name: Introduction to Study of Sciences UPJŠ/USPV/13 Introduction to Study of Sciences					
Course type, scope a Course type: Lectur Recommended cour Per week: Per stud Course method: pre	re / Practice r se-load (hours): y period: 12s / 3d				
Number of ECTS cr	edits: 2				
Recommended seme	ster/trimester of the cours	e: 1			
Course level: I.					
Prerequisities:					
Conditions for cours	e completion:				
Learning outcomes:					
Brief outline of the c	ourse:				
Recommended litera	ture:				
Course language:					
Notes:					
Course assessment Total number of asses	ssed students: 1734				
	abs n				
86.51 13.49					
Provides: doc. RNDr	. Marián Kireš, PhD.				
Date of last modifica	tion: 25.09.2019				
Approved:					

University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚMV/ UAD/10	Course name: Introduction to data analysis
Course type, scope a Course type: Lectur Recommended cou Per week: 1 / 1 Per Course method: pre	re / Practice rse-load (hours): study period: 14 / 14
Number of ECTS cr	edits: 2
Recommended seme	ster/trimester of the course: 1.
Course level: I.	
Prerequisities:	
Conditions for cours Test and individual p Oral presentation of t	•
understand its import To understand element	burpose of statistical data analysis, its methods and statistical thinking and stance for science and practical life. Intary statistical concepts. In handling real data using spreadsheet Excel and statistical software R.
statistics) 2. Collecting Data (ty 3. Handling Data (v skewness and kurtosi	course: basic philosophy and aim of statistical data analysis, descriptive and inductive ypes of data, random sample, randomized experiment) visualization, summarizing – measures of center, measures of variability, is, relationships in data – introduction to regression and correlation) be (elementary view into estimation and testing hypothesis)
 Rossman, A.J. et a 2009 Utts, J.M.: Seeing Utts, J.M., Heckar 	hture: ké metody, Matfyzpress, Praha, 1998 (in Czech) l.: Workshop Statistics: Discovery with Data and Fathom, 3rd ed. Wiley, Through Statistics, 4th ed., Thomson Brooks/Cole, Belmont, 2014 d R.F.: Mind on Statistics, 5th ed. Thomson Brooks/Cole, Belmont, 2014 J.: Pravděpodobnost a matematická statistika, Matfyzpress, Praha, 2001 (in
Course language: Slovak	
Notes:	

Notes:

Course assessment Total number of assessed students: 328							
A B C D E FX							
33.54	25.3	28.96	11.28	0.61	0.3		
Provides: RND	Provides: RNDr. Martina Hančová, PhD.						
Date of last modification: 18.09.2020							
Approved:							

University: P. J. Ša	fárik Universit	y in Košice			
Faculty: Faculty of	Science				
Course ID: ÚFV/ LTV/18	Course nar	ne: Laboratórn	a technika a výpo	očty	
Course type, scope Course type: Prac Recommended co Per week: 2 Per s Course method: p	tice wrse-load (ho tudy period: 2	urs):			
Number of ECTS	credits: 2				
Recommended sen	nester/trimest	er of the cours	se: 2.		
Course level: I.					
Prerequisities:					
Conditions for cou	rse completio	n:			
Learning outcome	s:				
Brief outline of the	e course:				
Recommended lite	rature:				
Course language:					
Notes:					
Course assessment Total number of ass		s: 9			
A	В	С	D	Е	FX
77.78	22.22	0.0	0.0	0.0	0.0
Provides: RNDr. Z	uzana Jurašeko	ová, PhD.			1
Date of last modifi	cation: 19.10.	2018			
Approved:	,				

University: P. J.	Šafárik Univers	sity in Košice			
Faculty: Faculty	of Science				
Course ID: ÚFV MFY/12	Course na	ame: Mathematic	al Physics		
Course type, sco Course type: L Recommended Per week: 3 / 1 Course method	ecture / Practice course-load (h Per study peri	e ours):			
Number of ECT	S credits: 6				
Recommended s	emester/trime	ster of the cours	e: 4.		
Course level: I.					
Prerequisities: (JMV/FRPb/19				
Conditions for c	ourse completi	ion:			
	course is to con	tinue in the study need branches of		l analysis with en	nphasize on the
equation. Proper coefficients. Sol functions. Special function function.	ties of Legendre lution of Lapla s: Hermite's po	e's polynomials. (ace's equation in	Derators in curv spherical coor herre's polynomi	Solution of Legen vilinear coordinate dinates. Propertie fals, Bessel's func- rential equations.	es. Lamé's es of spherical
Tai L. Chow : M J. Goldberg, M.	H. Rowland : F athematical Me Potter : Differer	thods for Physicintial Equations.	sts.	oundary Value Pro s, Harbourt Acade	
Course language	5 .				
Notes:					
Course assessme Total number of		ıts: 69			
A	В	С	D	E	FX
26.09	18.84	13.04	11.59	30.43	0.0
Provides: RNDr	Tomáš Lučivja	nský, PhD., RNE	Dr. Marián Jurčiš	in, PhD.	

Approved:

University: P. J. S	Safárik Universi	ty in Košice			
Faculty: Faculty	of Science				
Course ID: ÚMV MTFa/15	Course na	me: Mathematic	es I for physicists	3	
Course type, sco Course type: Le Recommended Per week: 2 / 2 Course method	cture / Practice course-load (ho Per study perio	ours):			
Number of ECTS	S credits: 5				
Recommended se	emester/trimes	ter of the cours	e: 1.		
Course level: I.					
Prerequisities:					
Conditions for co Two written tests is given according Learning outcom To obtain basic k the theory in cond	and one homew g to the results fr nes: nowledge on fu	rork with excercing the semester and the semiconductive semicondu	r and in view of the	he results of the v	vritten final test
Brief outline of t Functions, basic its geometric apli integrals, basic m	properties. Elen cations. Theoren ethods of integr	ms about continu	uous functions. B	Behaviour of func	
Recommended li S. Lang: A First (lus. Springer Ve	rlag, 1998		
Course language Slovak					
Notes:					
Course assessme Total number of a		s: 20			
A	В	С	D	Е	FX
30.0	25.0	30.0	10.0	5.0	0.0
Provides: Mgr. K	atarína Lučivja	nská, PhD., Mgr	. Barbora Klemo	vá, Mgr. Diana F	lačková
Date of last modi	6 4 ² 02 05	2015			
Date of last mou	iication: 03.05	.2013			

University: P. J.	Šafárik Univers	sity in Košice			
Faculty: Faculty	of Science				
Course ID: ÚM MTFb/15	V/ Course na	ame: Mathemati	cs II for physicis	ts	
Course type, sco Course type: L Recommended Per week: 2 / 2 Course method	ecture / Practice course-load (h Per study peri	e ours):			
Number of ECT	S credits: 4				
Recommended	semester/trime	ster of the cours	se: 2.		
Course level: I.					
Prerequisities: U	ÚMV/MTFa/15				
	s and one home	work with excert		nole semester, fin ults of the writter	
functions of mo	uired knowledge re variables. To	learn to solve ba	sic types of diffe	knowledge on lir erential equations oblems about inf	s and know how
limits, partial de	r algebraic equa rivations, local		tions of two var	f more variables ables. Some type	
2. Huťka V., Ber	rst Course in Ca 1ko E., Ďurikov	lculus, Springer ič V.: Matematik emiky, 1.díl. Ma	a, Alfa, Bratislav		
J. DUSIA, Z. IVIA	tematika pro ch	5,			
Course languag Slovak					
Course languag					
Course languag Slovak	e: ent				
Course languag Slovak Notes: Course assessm	e: ent		D	E	FX
Course languag Slovak Notes: Course assessm Total number of	e: ent assessed studen	uts: 16			FX 0.0
Course languag Slovak Notes: Course assessm Total number of A 43.75	e: ent assessed studen B 25.0	tts: 16 C 25.0	D 6.25	E 0.0	0.0
Course languag Slovak Notes: Course assessme Total number of A	e: ent assessed studen B 25.0 2NDr. Stanislav	tts: 16 C 25.0 Lukáč, PhD., Mg	D 6.25	E 0.0	0.0

University:	ΡJ	Šafárik	University	in Košice
Chiver Sity.	1.0.	Suluin	Oniversity	

Faculty: Faculty of Science

Course ID: ÚFV/	Course name: Methods of Structural Analysis
MSA1/03	

Course type, scope and the method:

Course type: Lecture / Practice

Recommended course-load (hours): Per week: 3 / 2 **Per study period:** 42 / 28

Course method: present

Number of ECTS credits: 7

Recommended semester/trimester of the course: 6.

Course level: I., II., III.

Prerequisities:

Conditions for course completion:

Elaboration of practical projects on electron microscopy and XRD diffractometry topics (75%) and final test with oral examination (25%)

Learning outcomes:

The course is oriented on modern methods of structural analysis of metals. Main topics are: optic microscopy, electron microscopy (TEM, SEM), electron microprobe analysis and X-ray diffractometry.

Brief outline of the course:

Optic microscopy. Electron microscopy: Electron beam instruments, Electron optics, Electron lences and deflection systems, Transmission electron microscopy - principle and construction. Electron – specimen interactions. Electron diffraction. Kikuchy lines. Scanning electron microscopy – principle and cnstrucion. Scanning transmission electron microscopy. High Voltage electron microscopy. Electron microscopy. Electron microscopy. Convergent beam diffraction.

X-ray diffractometry: Scattering of x-rays, Neutrons and neutron scattering, CW - diffractometer, Ewald's sphere, Diffraction on powder samples, The main characteristics of powder diffraction pattern, Structure factor, Ocupation factor, Atomic displacement factor, Peak intensity, shape and symmetry, Sherrer equation. Peak profile, Rietweld method. Qualitative phase analysis, parameters of elementary cell, Profile analysis of diffraction peak and interpretation of profile analysis.

Recommended literature:

1. P. Sovák et al, Vybrané moderné metódy štruktúrnej analýzy kovov, VŠ učebné texty, UPJŠ, 2007

P.W. Hawkes, J.C.H Spence, Science of Microscopy, Springer, ISBN10: 0-387-25296-7, 2007
 C. B. Carter, J. B. Williams, Transmission electron microscopy, ISBN 978-0-387-76500-6, 2012

4.Structure Determination from Powder Diffraction Data, Edited by W.I.F. David, K. Shankland, L.B. McCusker, C. Bärlocher, Oxford University Press, 2006

Course language:

1. English

Note

Notes:							
Course assessment Total number of assessed students: 86							
А	В	С	D	Е	FX	Ν	Р
39.53	22.09	8.14	1.16	0.0	0.0	0.0	29.07
Provides: prof. RNDr. Pavol Sovák, CSc., doc. Ing. Karel Saksl, DrSc., Ing. Vladimír Girman, PhD.							
Date of last modification: 28.06.2021							
Approved:							

University: P. J.	. Šafárik Univers	ity in Košice					
Faculty: Faculty	y of Science						
Course ID: ÚBEV/ MKV/15Course name: Mikrobiológia a základy virológie							
Course type: I Recommended	ope and the met Lecture / Practice d course-load (h 2 Per study perio d: present	ours):					
Number of EC	FS credits: 5						
Recommended	semester/trimes	ster of the cours	e: 5.				
Course level: I.							
Prerequisities:	ÚBEV/CYT1/15						
	course completi practicals (at le		ritten examinati	ons during sem	ester, final oral		
their cytology, p	btain a basic info	tics, ecology, clas	ssification, and i	and eukaryotic r mportance . Infor			
				gy, physiology, go l environment.	enetics, ecology,		
Recommended	literature:						
Course languag	ge:						
Notes:							
Course assessm Total number of	ent f assessed studen	ts: 1406					
А	В	С	D	Е	FX		
22.4	13.58	18.28	19.63	21.76	4.34		
	RNDr. Peter Prist D., RNDr. Lenka		· · · · · · · · · · · · · · · · · · ·	PhD., RNDr. Ma	iriana		
Date of last mo	dification: 02.02	2.2021					
Approved:							
	<u> </u>						

University: P. J. Ša	fárik Univers	ity in Košice			
Faculty: Faculty of	f Science				
Course ID: ÚFV/ MTBF/18	Course name: Moderné trendy v biofyzike				
Course type, scope Course type: Lec Recommended co Per week: 2 Per s Course method: 1	ture ourse-load (h study period:	ours):			
Number of ECTS	credits: 2				
Recommended ser	nester/trimes	ster of the cours	e: 1.		
Course level: I.					
Prerequisities:					
Conditions for cou	ırse completi	on:			
Learning outcome	es:				
Brief outline of the	e course:				
Recommended lite	erature:				
Course language:					
Notes:					
Course assessmen Total number of as		ts: 8			
A	В	С	D	E	FX
62.5	37.5	0.0	0.0	0.0	0.0
Provides: doc. Mg Žoldák, PhD., prof. Jurašeková, PhD., c	RNDr. Pavol	Miškovský, DrS	Sc., doc. Mgr. Gr	regor Bánó, PhD	
Date of last modif	ication: 05.10	0.2018			
Approved:					-

		ity in Košice			
Faculty: Facul	ty of Science				
Course ID: ÚF MBF1/14	V / Course na	me: Molecular E	Biophysics I		
Course type: Recommende	ed course-load (h Per study period:	ours):			
Number of EC	CTS credits: 4				
Recommended	l semester/trimes	ster of the cours	e: 4.		
Course level:]	., II.				
Prerequisities					
Conditions for Exam.	course completi	on:			
-	omes: oleting the course f the biological ma	-	knowledge abou	t the structure ar	nd principles of
chain, radius o proteins. Struc biopolymers: p Recommendeo 1. C.R.Cantor, 1980.	P.R.Schimmel, B	ture and properties of saccharides. , hydration of pro- iophysical Chem	ies of nucleic ad Structure and poteins, hydration istry Part I-III, F	cids. Structure an properties of lipid of nucleic acids. reeman and Co.,	d properties of s. Hydration of San Francisco,
3. H.Frauenfel	Fabián, Vybrané k der, J.Disenhofer, University Press	P.G.Wolyns, Sim , 1999.	plicity and Corr	plexity in Proteir	
,	Aolecular biophys		51		
,					
4. M. Daune, M Course langua Slovak					
4. M. Daune, M Course langua Slovak Notes: Course assessi	ge:	ts: 28			
4. M. Daune, M Course langua Slovak Notes: Course assessi	nent	ts: 28 C	D	E	FX
4. M. Daune, M Course langua Slovak Notes: Course assessi Total number of	nent of assessed studen				FX 0.0
4. M. Daune, M Course langua Slovak Notes: Course assessi Total number of A 60.71	nent of assessed studen B	C 7.14	D	E	

Approved:

University: P. J. Ša	fárik Univers	ity in Košice			
Faculty: Faculty of	Science				
Course ID: ÚFV/ MBB1/18	Course na	me: Molekulová	a bunková biol	ógia I	
Course type, scope Course type: Lect Recommended co Per week: 2 / 2 Pe Course method: p	ture / Practice ourse-load (h er study perio	ours):			
Number of ECTS	credits: 5				
Recommended sen	nester/trimes	ster of the cours	e: 6.		
Course level: I.					
Prerequisities:					
Conditions for cou	rse completi	on:			
Learning outcome	s:				
Brief outline of the	e course:				
Recommended lite	rature:				
Course language:					
Notes:					
Course assessment Total number of as		ts: 0			
A	В	С	D	E	FX
0.0	0.0	0.0	0.0	0.0	0.0
Provides: doc. RNI	Dr. Katarína Š	Štroffeková, PhD	., RNDr. Zuzana	n Naďová, PhD.	1
Date of last modifi	cation: 01.10	0.2018			
Approved:					

University: P. J. Šafári	k University in Košice
Faculty: Faculty of Sc	ience
Course ID: ÚFV/ NUM/10	Course name: Numerical Methods
Course type, scope an Course type: Lecture Recommended cours Per week: 2 / 1 Per s Course method: pres	e / Practice se-load (hours): tudy period: 28 / 14
Number of ECTS cree	dits: 4
Recommended semes	ter/trimester of the course: 3.
Course level: I.	
Prerequisities:	
	completion: is based on students' activity in the classroom and work on assignments. written test and all worked assignments submitted electronically with the
1	vith basic numerical methods of calculus and algebra, which are necessary rse of computational physics.
 Approximation of fu Interpolation of fund Approximation by tr Solution of nonlinea Numerical methods Solution of systems Solution of systems Numerical integration Numerical different Eigenvalues and eit The complete prob 	tion of problems and errors of numerical solution. unctions. ctions. rigonometric polynomials. Fast Fourier analysis. ar equations, convergence conditions and error estimation of the methods. for solving nonlinear equations. of linear equations - direct methods. of linear equations - iterative methods. on (quadrature) of functions. tiation of functions. genvectors of a matrix - partial problem. lem of eigenvalues.
2008. Other literature: - R.W. Hamming: Num	ure: rical Computation in Science and Engineering, Oxford University Press, nerical Methods for Scientists and Engineers, Dover, 1973. cal Methods for Physics, Prentice-Hall, 1994.
Course language:	
Notes:	

Course assessment Total number of assessed students: 130							
A B C D E FX							
15.38	16.92	25.38	22.31	15.38	4.62		
Provides: prof. RNDr. Milan Žukovič, PhD.							
Date of last modification: 01.07.2021							
Approved:							

University: P. J.	Šafárik Univers	ity in Košice					
Faculty: Faculty	of Science						
Course ID: ÚCH OCHB/10	e ;						
Course type, sco Course type: L Recommended Per week: 3 / 1 Course method	ecture / Practice course-load (h Per study perio	ours):					
Number of ECT	S credits: 5						
Recommended s	semester/trimes	ster of the cours	e: 2.				
Course level: I.							
Prerequisities: Ú	JCHV/VACH/10	0					
Conditions for c	ourse completi	on:					
Learning outcor	nes:						
Brief outline of t	the course:						
3. Organic Chem	esentation in MO histry, Clayden, histry, Solomon, émia, Pavol Zah	Greeves Warren Willey, 2009. radník, Mária Mé	& Wothers, Oxf ečiarová, Peter N	ord University Pr Magdolen, Univer			
Course language	e:						
Notes:							
Course assessme Total number of		ts: 259					
A	В	С	D	Е	FX		
22.78	22.39	31.66	18.15	4.63	0.39		
Provides: RNDr. Mária Vilková, P		aková, PhD., doc	. RNDr. Mirosla	ava Martinková, F	hD., RNDr.		
Date of last mod	lification: 30.08	3.2021					
Approved:							

University: P. J. Ša	fárik University in Košice				
Faculty: Faculty of	Science				
Course ID: ÚFV/ FCH1/02	Course name: Physical Chemistry for Biological Sciences				
	ure / Practice urse-load (hours): r study period: 42 / 28				
Number of ECTS	credits: 6				
Recommended sem	nester/trimester of the course: 3.				
Course level: I., II.					
Prerequisities:					
Conditions for cou	rse completion:				

Test

Exam

Learning outcomes:

The introduction into the fundamental knowledge of selected parts of physical chemistry with emphasis on the utilization of these knowledges for the study of physico-chemical properties of biomacromolecules and biological systems.

Brief outline of the course:

Description of macroscopic systems, energy and 1. law of thermodynamics, entropy and 2. law of thermodynamics, Gibbs energy and equilibrium state, chemical potential, binding constants of the ligand-macromolecule interactions, biophysical applications of the thermodynamics. Solutions, electrolytic solutions, electrochemical equilibrium, electrodes, electrochemical potential. Statistical thermodynamics: the interpretation of energy, heat, entropy and information; the partition functions, biological applications of statistical thermodynamics, the conformational transitions in proteins and nucleic acids. Chemical reactions, chemical and biochemical kinetics, dynamics of the chemical reactions, kinetics of the enzymatical reactions, inhibition of the enzymes. Transport processes, molecular diffusion, membrane transport and its significance for the biological organisms.

Recommended literature:

1. P. Atkins and J. de Paula. Atkins's Physical Chemistry (9th Edition), Oxford University Press, 2010.

2. P. Atkins. Fyzikálna chémia (slovenský preklad 6. vydania), STU Bratislava, 1999.

3. P. Atkins, J. De Paula. Fyzikální chemie (český preklad 9. vydania), VŠCHT Praha, 2013

4. R.Chang. Physical Chemistry for the Biosciences, University Science Book, 2006.

5. D. Eisenberg and D. Crothers. Physical Chemistry with Applications to the Life Sciences, Benjamin/Cummings, 1979.

6. K. van Holde, W. Johnson and P. Ho. Principles of Physical Biochemistry, Prentice Hall, 1988.

7. D.T. Haynie. Biological Thermodynamics (2nd Edition), Cambridge University Press, 2008.

8. A.P.H. Peters. Concise Chemical Thermodynamics (3rd Edition), CRC Press, Taylor & Francis Group, 2010.

9. I. Tinoco, jr., K. Sauer, J.C. Wang, J.C. Puglisi, G. Harbison and D.Rovnyak.

Physical Chemistry – Principles and Applications in Biological Sciences (5th Edition), Pearson, 2014.

10. A. Cooksy. Physical Chemistry- Thermodynamics, Statistical Mechanics, and Kinetics, Pearson, 2014.

Course languag	ge:					
Notes:						
Course assessment Total number of assessed students: 100						
А	В	С	D	Е	FX	
18.0	29.0	31.0	11.0	11.0	0.0	
Provides: doc. 1	Mgr. Daniel Janci	ura, PhD.				
Date of last mo	dification: 03.05	.2015				
Approved:						

University: P. J. Šafá	irik University in Košice
Faculty: Faculty of S	Science
Course ID: ÚFV/ ZFP1a/03	Course name: Physics Practical I
Course type, scope a Course type: Practi Recommended cou Per week: 3 Per stu Course method: pro	ce rse-load (hours): ıdy period: 42
Number of ECTS cr	redits: 3
Recommended seme	ester/trimester of the course: 2.
Course level: I.	
Prerequisities:	
Conditions for cours The active work duri Vindication of report	ing semester and hand in all reports.
Learning outcomes: Developing proper la	aboratory habits, skills and verify their theoretical knowledge.
with kinds and calcures with kinds and calcures results. The students introductory physics Laboratory assignment 1. Density measurem 2. Radius measurem 2. Radius measurem 3. Gravitational acceleration and physical pendulu 4. Moment of inertia pendulum. 5. Measurements of 2. Measurement of the measurement of the measurement of the measurements of 2. Measurements of 3. Measurements of 4. Meas	oratory exercises is to familiarize the students with measurement methods, alus of mistakes, with measured results processing, and with presentation of gain practical skills, and verify their theoretical knowledge of first semester course. They develop proper laboratory habits. ent: hents of liquids and solids. ents of spherical cap. Measurements of eter. leration measurements using mathematical im. measurement using physical and torsion Young's modulus. oefficient of viscosity. he speed of sound. general gas constant and Boltzmann constant. thermal expansivity of air. f thermal capacity of matter. the surface tension.
measurements I), Ed	 C., Onderová, Ľ., Kireš, M.: Základné fyzikálne praktikum I. (Basic physical PF UPJŠ Košice 2007. 31. Slovenský inštitút normalizácie v Bratislave (Slovak institute of technical

Ješková, Z.: Computer based experiments in thermodynamics using IP COACH,ed. PF UPJŠ in Košice, 2004.

Course language english	ge:						
Notes:							
Course assessm Total number o	nent f assessed studen	ts: 256					
А	В	С	D	Е	FX		
56.25	25.78	13.67	3.52	0.78	0.0		
	Provides: doc. RNDr. Adriana Zeleňáková, PhD., doc. RNDr. Marián Kireš, PhD., doc. RNDr. Jár. Füzer, PhD., doc. RNDr. Jozef Hanč, PhD.						
Date of last mo	dification: 29.03	.2020					
Approved:							

University: P. J.	Šafárik Univers	ity in Košice				
Faculty: Faculty	of Science					
Course ID: ÚFV ZFP1b/03						
Course type, sco Course type: P Recommended Per week: 3 Pe Course method	ractice course-load (h r study period:	ours):				
Number of ECT	'S credits: 3					
Recommended s	semester/trimes	ster of the cours	e: 3.			
Course level: I.						
Prerequisities: (JFV/ZFP1a/03					
	xperimental tasl	on: ks, their apprecia l theoretical prep				
b. To gain somec. To gain experiBrief outline of aStudents on prace	physical inside practice in data ience and report the course: ctical exercises	are working in p	sis and interpret tion and results.	ation of resuman	ce.	
Recommended I Tumanski S, Hai	literature: ndbook of magn	netic measurement	nts, CRC press, 2		04.	
Course language Slovak	e:					
Notes:						
Course assessme Total number of						
A	В	С	D	E	FX	
		10.44	1.00	0.0		
64.98	20.74	12.44	1.38	0.0	0.46	
	,				0.46	
64.98 Provides: doc. R Date of last mod	NDr. Adriana Z	l Zeleňáková, PhD.			0.46	

University: P. J.	Šafárik Universi	ity in Košice			
Faculty: Faculty	of Science				
Course ID: ÚFV ZFP1c/14	// Course na	me: Physics Pra	actical III		
	ractice course-load (he r study period:	ours):			
Number of ECT	S credits: 3				
Recommended	semester/trimes	ter of the cours	se: 4.		
Course level: I.					
Prerequisities:					
	f experimental ta	sks, their evalua		of a written report preparation for t	
practice in data	nysical inside inte	ysis and interpre-		d in the lectures. ance. c. To gain	-
sound. Refractiv	dulum. Composi	focal length. In	terference. Diffra	lations. Resonanc action. Diffractio	-
2006 P. Kollár a kol. Z	vá, Z., Onderová,	ne praktikum II,	PF UPJŠ Košice	e praktikum I, PF e, 2006	UPJŠ Košice,
Course languag slovak or englis					
Notes:					
Course assessm Total number of	ent assessed student	ts: 68			
А	В	С	D	E	FX
70.59	16.18	5.88	2.94	4.41	0.0
Provides: doc F			L	<u> </u>	l
I TOVIUCS. UOC. I	RNDr. Marián Ki	reš, PhD., doc. I	RNDr. Ján Füzer,	, PhD.	

University: P. J. Ša	fárik Univers	ity in Košice			
Faculty: Faculty of	Science				
Course ID: ÚFV/ PEMBF1/18	Course na	me: Praktikum l	k experimentáln	ym metódam biof	ŷyziky I
Course type, scope Course type: Prac Recommended co Per week: 2 Per s Course method: p	etice ourse-load (h tudy period:	ours):			
Number of ECTS	credits: 2				
Recommended ser	nester/trimes	ster of the cours	e: 4.		
Course level: I.					
Prerequisities: ÚF	V/EMBF1/18				
Conditions for cou	rse completi	on:			
Learning outcome	s:				
Brief outline of the	e course:				
Recommended lite	erature:				
Course language:					
Notes:					
Course assessment Total number of as	-	ts: 4			
A	В	С	D	Е	FX
100.0	0.0	0.0	0.0	0.0	0.0
Provides: RNDr. G	abriela Fabrie	ciová, PhD.		1	1
Date of last modifi	cation: 01.10	0.2018			
Approved:	,				

University: P. J. Ša	afárik Universi	ty in Košice				
Faculty: Faculty o	f Science					
Course ID: ÚFV/ PEMBF2/18	V/ Course name: Praktikum k experimentálnym metódam biofyziky II					
Course type, scop Course type: Pra Recommended c Per week: 2 Per Course method:	ctice ourse-load (ho study period: 1	ours):				
Number of ECTS	credits: 2					
Recommended set	mester/trimest	ter of the cours	e: 5.			
Course level: I.						
Prerequisities: ÚF	V/EMBF2/18					
Conditions for co	urse completio	on:				
Learning outcome	es:					
Brief outline of th	e course:					
Recommended lit	erature:					
Course language:						
Notes:						
Course assessmen Total number of as		s: 0				
A	В	С	D	Е	FX	
0.0	0.0	0.0	0.0	0.0	0.0	
Provides: doc. RN Žoldák, PhD.	Dr. Erik Sedlál	k, DrSc., RNDr.	Gabriela Fabric	iová, PhD., RND	r. Gabriel	
Date of last modif	ication: 01.10.	2018				
Approved:						

University: P. J. Š	afárik Universi	ty in Košice				
Faculty: Faculty of	of Science					
Course ID: ÚFV/ PEMBF3/18	V/ Course name: Praktikum k experimentálnym metódam biofyziky III					
Course type, scop Course type: Pra Recommended c Per week: 2 Per Course method:	nctice course-load (ho study period: 2	urs):				
Number of ECTS	credits: 2					
Recommended se	emester/trimest	er of the cours	se: 6.			
Course level: I.						
Prerequisities: Úl	FV/EMBF3/18					
Conditions for co	urse completio	n:				
Learning outcom	es:					
Brief outline of th	ne course:					
Recommended lit	terature:					
Course language:	;					
Notes:						
Course assessmer Total number of a		s: 0				
Α	В	С	D	Е	FX	
0.0	0.0	0.0	0.0	0.0	0.0	
Provides: doc. Mg Huntošová, PhD.	gr. Gregor Bánó	, PhD., RNDr.	Zuzana Naďová,	PhD., RNDr. Ver	onika	
Date of last modi	fication: 01.07.	2021		_		
Approved:						

University: P. J.	Šafárik Univers	sity in Košice					
Faculty: Faculty	of Science						
Course ID: ÚFV KVM I/11							
Course type, sco Course type: L Recommended Per week: 4 / 2 Course method	ecture / Practico course-load (h Per study peri	e 1ours):					
Number of ECT	S credits: 8						
Recommended s	semester/trime	ster of the cours	e: 3.				
Course level: I.							
Prerequisities:							
Conditions for c	ourse complet	ion:					
Learning outcourse of the second seco	liar with elemer	<i>y</i> 1 1	f quantum mech	anics and to illust	trate its possible		
axioms of QM. S and spherically	chrödinger equi symmetric pote	ation and its solut entials. Tunnel e	ion for a square p ffect and over-b	uantum mechani ootential well, har arrier reflection. uli exclusion prin	monic oscillator Spin and Paul		
(in Slovak langu 2. Ľ. Skála, Úvo 3. J. Pišút, L. Go 4. W. Greiner, Q 5. A. C. Philips,	óthová, Kvanto age) d do kvantovej omolčák, Úvod uantum Mechar Introduction to	mechaniky, Acad do kvantovej me nics, 4th edition, Quantum Mecha	lemia, Praha, 200 chaniky, Bratisla Springer, Berlin, mics, Wiley, Wei		guage) ak language)		
Course languag EN - english	e:						
Notes:							
Course assessme Total number of		nts: 85					
A	В	C	D	Е	FX		
27.06	18.82	21.18	9.41	17.65	5.88		
1					5.00		
Provides: doc. R	NDr. Jozef Stre	ečka, PhD.			5.00		

		sity in Košice			
Faculty: Faculty	of Science				
Course ID: ÚFV KVM II/08	Course n	ame: Quantum M	lechanics II.		
Course type, sco Course type: Le Recommended Per week: 3 / 1 Course method	ecture / Practic course-load (l Per study per	e hours):			
Number of ECT					
Recommended s	emester/trime	ester of the cours	e: 4.		
Course level: I.	,			_	
Prerequisities: Ú	JFV/KVM/08 a	and leboÚFV/KV	M I/11		
Conditions for c	ourse complet	tion:			
	iar with the app	proximate method ny-particle quant	-	chanics and their	applications b
The stationary a	nd non-station	ary perturbation	• I		2
The stationary a discrete, continuadiabatic and has electric and mag method and its a	and non-station nous and discr rmonic perturb gnetic field, State applications. M	ary perturbation ete-continuous er pations. Anharmor ark effect, normal lany-particle quar gen molecule. Ha	ergy spectrum. nic oscillator. Th and anomalous tum-mechanical	The special cas e hydrogen atom Zeeman effect. I systems, atoms	es of constant in the externa Ritz variationa
discrete, continu adiabatic and har electric and mag method and its a The helium atom Recommended li 1. V. Ilkovič, Kva 2. J. Pišút, L. Go 3. W. Greiner, Qu 4. D. J. Griffiths,	and non-station ious and discr rmonic perturb gnetic field, Sta applications. Ma and the hydro iterature: antová teória II molčák, Úvod uantum Mecha , Introduction t	ete-continuous er pations. Anharmor ark effect, normal lany-particle quar	ergy spectrum. hic oscillator. Th and anomalous htum-mechanical rtree and Hartree Košice, 1989. (i haniky, Bratislay Springer, Berlin,	The special cas e hydrogen atom Zeeman effect. I systems, atoms -Fok method. n Slovak) /a 1983. (in Slova 2000.	ak)
The stationary a discrete, continu adiabatic and han electric and mag method and its a The helium atom Recommended li 1. V. Ilkovič, Kva 2. J. Pišút, L. Go 3. W. Greiner, Qu	and non-station ious and discr rmonic perturb gnetic field, Sta applications. Ma and the hydro iterature: antová teória II molčák, Úvod uantum Mecha , Introduction t	ete-continuous er pations. Anharmor ark effect, normal lany-particle quar gen molecule. Ha I, Scriptum UPJŠ, do kvantovej mec nics, 4th edition, S	ergy spectrum. hic oscillator. Th and anomalous htum-mechanical rtree and Hartree Košice, 1989. (i haniky, Bratislay Springer, Berlin,	The special cas e hydrogen atom Zeeman effect. I systems, atoms -Fok method. n Slovak) /a 1983. (in Slova 2000.	ak)
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University: P. J. Šafá	rik University in Košice					
Faculty: Faculty of Science						
Course ID: ÚTVŠ/ ÚTVŠ/CM/13	Course name: Seaside Ae	robic Exercise				
Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: Per study period: 36s Course method: combined, present						
Number of ECTS cr	edits: 2					
Recommended seme	ster/trimester of the cours	e:				
Course level: I., II.						
Prerequisities:						
Conditions for course Conditions for course Attendance	-					
Learning outcomes: Learning outcomes:						
Learning outcomes: Students will be pro conditions actively a Students will acquire	and their skills in work and practical experience in org the stay and to create positive	ssibilities how to spend leisure time in seaside d communication with clients will be improved. canising the cultural and art-oriented events, with e experiences for visitors.				
Learning outcomes: Students will be pro- conditions actively a Students will acquire the aim to improve th Brief outline of the c Brief outline of the c 1. Basics of seaside a 2. Morning exercises 3. Pilates and its appl 4. Exercises for the s 5. Yoga basics 6. Sport as a part of 1 7. Application of pro- (children, young peop 8. Application of sea	and their skills in work and e practical experience in org the stay and to create positive course: therobics lication in seaside conditions pine eisure time jects of productive spending ple, elderly) side cultural and art-oriented	d communication with clients will be improved. anising the cultural and art-oriented events, with e experiences for visitors.				
Learning outcomes: Students will be pro- conditions actively a Students will acquire the aim to improve th Brief outline of the c Brief outline of the c 1. Basics of seaside a 2. Morning exercises 3. Pilates and its appl 4. Exercises for the s 5. Yoga basics 6. Sport as a part of 1 7. Application of pro- (children, young peop 8. Application of sea	and their skills in work and e practical experience in org the stay and to create positive course: therobics lication in seaside conditions pine eisure time jects of productive spending ple, elderly) side cultural and art-oriented	d communication with clients will be improved. anising the cultural and art-oriented events, with e experiences for visitors.				
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Learning outcomes: Students will be pro- conditions actively a Students will acquire the aim to improve th Brief outline of the c Brief outline of the c I. Basics of seaside a 2. Morning exercises 3. Pilates and its appl 4. Exercises for the s 5. Yoga basics 6. Sport as a part of I 7. Application of pro- (children, young peop 8. Application of sea Recommended litera Course language: Notes: Course assessment	and their skills in work and e practical experience in org he stay and to create positive course: herobics lication in seaside conditions pine eisure time jects of productive spending ple, elderly) side cultural and art-oriented nture:	d communication with clients will be improved. anising the cultural and art-oriented events, with e experiences for visitors.				

Provides: Mgr. Agata Horbacz, PhD.

Date of last modification: 15.03.2019

University: P. J. Šaf	árik University in Košice				
Faculty: Faculty of	Science				
Course ID: ÚFV/ SPBFb1/18Course name: Semestrálna práca I					
Course type, scope Course type: Recommended cou Per week: Per stu Course method: pr	urse-load (hours): dy period: resent				
Number of ECTS c					
	ester/trimester of the cou	rse: 5.			
Course level: I.					
Prerequisities:					
Conditions for cour	se completion:				
Learning outcomes	:				
Brief outline of the	course:				
Recommended liter	ature:				
Course language:					
Notes:					
Course assessment Total number of ass	essed students: 0				
	abs n				
	0.0	0.0			
Provides:		· · · · · · · · · · · · · · · · · · ·			
Date of last modific	ation: 01.10.2018				
Approved:					

University: P. J. Šaf	árik University in Košice				
Faculty: Faculty of	Science				
Course ID: ÚFV/ SPBFb2/18	Course name: Semestrál	na práca II			
Course type, scope Course type: Recommended cou Per week: Per stu Course method: p	urse-load (hours): dy period: resent				
Number of ECTS c					
	ester/trimester of the cour	rse: 6.			
Course level: I.					
Prerequisities:					
Conditions for cour	se completion:				
Learning outcomes	:				
Brief outline of the	course:				
Recommended liter	ature:				
Course language:					
Notes:					
Course assessment Total number of ass	essed students: 0				
	abs n				
	0.0	0.0			
Provides:					
Date of last modific	ation: 01.10.2018				
Approved:			_		

University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚTVŠ/ TVa/11	Course name: Sports Activities I.
Course type, scope a Course type: Practi- Recommended cou Per week: 2 Per stu Course method: co	ce rse-load (hours): idy period: 28 mbined, present
Number of ECTS cr	edits: 2
Recommended seme	ester/trimester of the course: 1.
Course level: I., I.II.,	II.
Prerequisities:	
Conditions for cours Min. 80% of active p	se completion: participation in classes.
They have a great in	their forms prepare university students for their professional and personal life pact on physical fitness and performance. Specialization in sports activitie strengthen their relationship towards the selected sport in which they also
University provides badminton, body forr indoor football, S-M In the first two seme and particularities of physical condition, c Last but not least, the	

In addition to these sports, the Institute offers for those who are interested winter and summer physical education trainings with an attractive program and organises various competitions, either at the premises of the faculty or University or competitions with national or international participation.

Recommended literature:

Course language:

Notes:

Course ass Total numb	essment per of assesse	d students: 1	2859					
abs abs-A abs-B abs-C abs-D abs-E n neabs								
87.01	87.01 0.08 0.0 0.0 0.0 0.04 8.1 4.77							
doc. PaedD	r. Ivan Uher,	PhD., prof. l	RNDr. Stanis	d Kaško, PhI slav Vokál, D Richard Mel	orSc., Mgr. M	arcel Čurgal	li, Mgr.	
Date of last	t modificatio	on: 13.05.202	21					
Approved:								

Faculty: Fa	culty of Sc	eience					
Course ID: TVb/11	ÚTVŠ/	Course name	: Sports Acti	vities II.			
Course ty Recomme Per week:	pe: Practic nded cour 2 Per stud	nd the method e se-load (hour ly period: 28 abined, presen	s):				
Number of	ECTS cre	dits: 2					
Recommen	ded semes	ter/trimester	of the cours	se: 2.		-	
Course leve	el: I., I.II.,	II.					
Prerequisit	ies:						
		e completion: classes - min.	80%.				
They have	a great im	their forms pre pact on physic	1	5	-	-	
improve.		_	r relationshi	p towards th	e selected s	-	
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improve. Brief outlin Within the University badminton, indoor foot In the first and particul physical co Last but no means of a In addition physical ed the premise	ne of the co optional su provides body form ball, S-M s two semes larities of in ordition, co t least, the special pro- to these s ucation trai es of the fac	burse: abject, the Inst for students t a, bouldering, f systems, step a ters of the first adividual sport bordination ab- important role ogram of media ports, the Inst anings with an a ulty or University	itute of Phys he following loorball, yog erobics, tabl st level of ed ts, motor skil ilities, physic of sports ac cal physical itute offers attractive pro	sical Education g sports action ga, power yog e tennis, tenno lucation study ls, game action cal performativities is to e education to for those who ogram and org	on and Sport ivities: aerob ga, pilates, sw his, volleybal ents will mas vities, they w nce, and mo eliminate swi influence and to are interes ganises variou	port in whic s of Pavol Jo bics, aikido, vimming, boo l and chess. ster basic cha vill improve lo tor performa imming illite d mitigate un sted winter a us competitio	h they also ozef Šafárik basketball, dy-building, aracteristics evel of their ince fitness. racy and by ifitness. and summer ons, either at
improve. Brief outlin Within the University badminton, indoor foot In the first and particul physical co Last but no means of a In addition physical ed the premise Recommen	ne of the co optional su provides body form ball, S-M s two semes larities of in ordition, co t least, the special pro- to these s ucation trai es of the fac	burse: abject, the Inst for students t a, bouldering, f systems, step a ters of the first adividual sport bordination ab- important role ogram of media ports, the Inst anings with an a ulty or University	itute of Phys he following loorball, yog erobics, tabl st level of ed ts, motor skil ilities, physic of sports ac cal physical itute offers attractive pro	sical Education g sports action ga, power yog e tennis, tenno lucation study ls, game action cal performativities is to e education to for those who ogram and org	on and Sport ivities: aerob ga, pilates, sw his, volleybal ents will mas vities, they w nce, and mo eliminate swi influence and to are interes ganises variou	port in whic s of Pavol Jo bics, aikido, vimming, boo l and chess. ster basic cha vill improve lo tor performa imming illite d mitigate un sted winter a us competitio	h they also ozef Šafárik basketball, dy-building, aracteristics evel of their ince fitness. racy and by ifitness. and summer ons, either at
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improve. Brief outlin Within the University badminton, indoor foot In the first and particul physical co Last but no means of a In addition physical edi the premise Recommen Course lang Notes: Course asso	ne of the co optional su provides body form ball, S-M s two semes larities of in ondition, co t least, the special pro- to these s ucation traises of the fac ded literat guage:	burse: abject, the Inst for students t a, bouldering, f systems, step a ters of the first adividual sport bordination ab- important role ogram of media ports, the Inst anings with an a ulty or University	itute of Phys he following loorball, yog erobics, tabl st level of ed ts, motor skil ilities, physic of sports ac cal physical itute offers attractive pro sity or compe	sical Education g sports action ga, power yog e tennis, tenre lucation study ls, game action cal performativities is to e education to for those who ogram and org	on and Sport ivities: aerob ga, pilates, sw his, volleybal ents will mas vities, they w nce, and mo eliminate swi influence and to are interes ganises variou	port in whic s of Pavol Jo bics, aikido, vimming, boo l and chess. ster basic cha vill improve lo tor performa imming illite d mitigate un sted winter a us competitio	h they also ozef Šafárik basketball, dy-building, aracteristics evel of their ince fitness. racy and by ifitness. and summer ons, either at

Provides: Mgr. Agata Horbacz, PhD., Mgr. Dávid Kaško, PhD., Mgr. Zuzana Küchelová, PhD., doc. PaedDr. Ivan Uher, PhD., prof. RNDr. Stanislav Vokál, DrSc., Mgr. Marcel Čurgali, Mgr. Patrik Berta, Mgr. Ladislav Kručanica, PhD., Bc. Richard Melichar, Mgr. Petra Tomková, PhD.

Date of last modification: 13.05.2021

Faculty. F							
racuity. I	aculty of Sc	ience					
Course ID TVc/11	: ÚTVŠ/	Course name	: Sports Acti	vities III.			
Course ty Recommo Per week	pe: Practice ended cours : 2 Per stud	d the method se-load (hours y period: 28 bined, present	s):				
Number o	f ECTS cree	dits: 2					
Recomme	nded semest	ter/trimester	of the cours	e: 3.			
Course lev	r el: I., I.II., I	I.					
Prerequisi	ties:						
		completion: ticipation in c	lasses				
They have	vities in all t a great imp	heir forms pre pact on physic rengthen their	al fitness an	d performan	ce. Specializ	ation in spor	rts activities
Brief outli	ne of the co	ursa					
University badminton indoor foo In the first and particu physical co Last but no means of a In addition physical co	provides f body form, tball, S-M s two semest alarities of in ondition, co of least, the in special pro- n to these sp ducation train	bject, the Inst or students the bouldering, f ystems, step a dividual sport ordination abia mportant role gram of medic ports, the Inst nings with an a alty or Univers	he following loorball, yog erobics, table t level of ed s, motor skil ilities, physic of sports ac cal physical itute offers	g sports acti a, power yog e tennis, tenr ucation stud- ls, game acti cal performa tivities is to e education to for those wh gram and org	ivities: aerob ga, pilates, sw his, volleybal ents will mas vities, they w nce, and mo eliminate swi influence and to are interes ganises variou	bics, aikido, vimming, boo l and chess. ster basic ch vill improve l tor performa imming illite d mitigate ur sted winter a us competitio	basketball dy-building aracteristics evel of their ance fitness eracy and by offitness. and summer ons, either a
University badminton indoor foo In the first and particu physical co Last but no means of a In addition physical eo the premise	provides f body form, tball, S-M s two semest alarities of in ondition, co of least, the in special pro- n to these sp ducation train	bject, the Inst or students the bouldering, f ystems, step a ers of the first idividual sport ordination abid mportant role gram of medic ports, the Inst nings with an a alty or Univers	he following loorball, yog erobics, table t level of ed s, motor skil ilities, physic of sports ac cal physical itute offers	g sports acti a, power yog e tennis, tenr ucation stud- ls, game acti cal performa tivities is to e education to for those wh gram and org	ivities: aerob ga, pilates, sw his, volleybal ents will mas vities, they w nce, and mo eliminate swi influence and to are interes ganises variou	bics, aikido, vimming, boo l and chess. ster basic ch vill improve l tor performa imming illite d mitigate ur sted winter a us competitio	basketball dy-building aracteristics evel of their ance fitness eracy and by fitness. and summer ons, either a
University badminton indoor foo In the first and particu physical co Last but no means of a In addition physical eo the premise Recommen	provides f body form, tball, S-M s two semest alarities of in ondition, co ot least, the in special pro- n to these sp ducation train es of the fact nded literat	bject, the Inst or students the bouldering, f ystems, step a ers of the first idividual sport ordination abid mportant role gram of medic ports, the Inst nings with an a alty or Univers	he following loorball, yog erobics, table t level of ed s, motor skil ilities, physic of sports ac cal physical itute offers	g sports acti a, power yog e tennis, tenr ucation stud- ls, game acti cal performa tivities is to e education to for those wh gram and org	ivities: aerob ga, pilates, sw his, volleybal ents will mas vities, they w nce, and mo eliminate swi influence and to are interes ganises variou	bics, aikido, vimming, boo l and chess. ster basic ch vill improve l tor performa imming illite d mitigate ur sted winter a us competitio	basketball dy-building aracteristics evel of their ance fitness eracy and by fitness. and summer ons, either a
University badminton indoor foo In the first and particu physical co Last but no means of a In addition physical eo the premise Recommen Course lar Notes:	provides f body form, tball, S-M s two semest alarities of in ondition, co ot least, the in special pro- n to these sp ducation train es of the fact nded literat	bject, the Inst or students the bouldering, f ystems, step a ers of the first idividual sport ordination abid mportant role gram of medic ports, the Inst nings with an a alty or Univers	he following loorball, yog erobics, table t level of ed s, motor skil ilities, physic of sports ac cal physical itute offers	g sports acti a, power yog e tennis, tenr ucation stud- ls, game acti cal performa tivities is to e education to for those wh gram and org	ivities: aerob ga, pilates, sw his, volleybal ents will mas vities, they w nce, and mo eliminate swi influence and to are interes ganises variou	bics, aikido, vimming, boo l and chess. ster basic ch vill improve l tor performa imming illite d mitigate ur sted winter a us competitio	basketball dy-building aracteristics evel of their ance fitness eracy and by fitness. and summer ons, either a
University badminton indoor foo In the first and particu physical cu Last but no means of a In addition physical ec the premise Recommen Course lar Notes:	provides f body form tball, S-M s two semest alarities of in ondition, co ot least, the in special pro- n to these sp ducation train es of the fact nded literat nguage:	bject, the Inst or students the bouldering, f ystems, step a ters of the first individual sport ordination abia mportant role gram of medic ports, the Inst nings with an a alty or Universe ure:	he following loorball, yog erobics, table t level of ed is, motor skil lities, physic of sports ac cal physical itute offers attractive pro-	g sports acti a, power yog e tennis, tenr ucation stud- ls, game acti cal performa tivities is to e education to for those wh gram and org	ivities: aerob ga, pilates, sw his, volleybal ents will mas vities, they w nce, and mo eliminate swi influence and to are interes ganises variou	bics, aikido, vimming, boo l and chess. ster basic ch vill improve l tor performa imming illite d mitigate ur sted winter a us competitio	basketball dy-building aracteristic evel of thei ance fitness eracy and by fitness. and summe ons, either a
University badminton indoor foo In the first and particu physical co Last but no means of a In addition physical ec the premise Recommen Course lar Notes:	provides f body form tball, S-M s two semest alarities of in ondition, co ot least, the in special pro- n to these sp ducation train es of the fact nded literat nguage:	bject, the Inst or students the bouldering, f ystems, step a ers of the first idividual sport ordination abid mportant role gram of medic ports, the Inst nings with an a alty or Univers	he following loorball, yog erobics, table t level of ed is, motor skil lities, physic of sports ac cal physical itute offers attractive pro-	g sports acti a, power yog e tennis, tenr ucation stud- ls, game acti cal performa tivities is to e education to for those wh gram and org	ivities: aerob ga, pilates, sw his, volleybal ents will mas vities, they w nce, and mo eliminate swi influence and to are interes ganises variou	bics, aikido, vimming, boo l and chess. ster basic ch vill improve l tor performa imming illite d mitigate ur sted winter a us competitio	basketball dy-building aracteristics evel of their ance fitness eracy and by fitness. and summer ons, either a

Provides: Mgr. Marcel Čurgali, Mgr. Agata Horbacz, PhD., Mgr. Dávid Kaško, PhD., Mgr. Zuzana Küchelová, PhD., doc. PaedDr. Ivan Uher, PhD., prof. RNDr. Stanislav Vokál, DrSc., Mgr. Patrik Berta, Mgr. Ladislav Kručanica, PhD., Bc. Richard Melichar, Mgr. Petra Tomková, PhD.

Date of last modification: 13.05.2021

E14 E			n Košice				
raculty: Fa	culty of Scie	ence					
Course ID: TVd/11	ÚTVŠ/ C	Course name:	Sports Acti	ivities IV.			
Course ty Recomme Per week:	pe: Practice nded course 2 Per study	I the method e-load (hours y period: 28 pined, present	5):				
Number of	ECTS cred	its: 2					
Recommen	ded semeste	er/trimester	of the cours	se: 4.			
Course leve	el: I., I.II., II	•					
Prerequisit	ies:						
		completion: ticipation in c	lasses				
They have	vities in all th a great impa	act on physic	al fitness an	d performan	for their profestice. Specialization of the selected special special special spectral spectra	ation in spor	rts activities
Within the University badminton, indoor foot In the first and particu physical co Last but no means of a In addition physical ed the premise	provides for body form, ball, S-M sy two semeste larities of incondition, coo t least, the in special prog to these sp ucation train es of the facu	pject, the Instruction students the bouldering, first stems, step are ers of the first dividual sport ordination abi important role gram of medic orts, the Inst lings with an a lty or Univers	he following loorball, yog erobics, tabl t level of ed s, motor skil lities, physi- of sports ac cal physical itute offers	g sports acti ga, power yog e tennis, tenr lucation stude lls, game actir cal performa tivities is to e education to for those wh ogram and org	on and Sports ivities: aerob ga, pilates, sw nis, volleyball ents will mas vities, they w ince, and mot eliminate swi influence and to are interes ganises variou national or int	bics, aikido, rimming, boo l and chess. ster basic ch ill improve l tor performa mming illite d mitigate ur ted winter a us competitio	basketball, dy-building, aracteristics evel of their unce fitness. eracy and by fitness. and summer ons, either at
Recommen	ded literatu	ire:					
Course lan	guage:						
Notes:							
	· · · · · · · · · · · · · · · · · · ·						
Course ass		nd studentes 5	125				
		ed students: 5 abs-B	125 abs-C	abs-D	abs-E	n	neabs

Provides: Mgr. Marcel Čurgali, Mgr. Agata Horbacz, PhD., Mgr. Dávid Kaško, PhD., Mgr. Zuzana Küchelová, PhD., doc. PaedDr. Ivan Uher, PhD., prof. RNDr. Stanislav Vokál, DrSc., Mgr. Patrik Berta, Mgr. Ladislav Kručanica, PhD., Bc. Richard Melichar, Mgr. Petra Tomková, PhD.

Date of last modification: 13.05.2021

University: P. J. Š	Šafárik Universi	ty in Košice			
Faculty: Faculty	of Science				
Course ID: ÚFV/ SVK/13	Course na	me: Student Sci	entific Conferen	ce	
Course type, sco Course type: Recommended Per week: Per s Course method	course-load (ho study period: : present				
Number of ECTS					
Recommended se		ter of the cours	e:		
Course level: I., I	II				
Prerequisities:					
Conditions for co	ourse completio	on:			
Learning outcom	nes:				
Brief outline of t	he course:				
Recommended li	terature:				
Course language	:				
Notes:					
Course assessme Total number of a	-	s: 50			
Α	В	С	D	Е	FX
100.0	0.0	0.0	0.0	0.0	0.0
Provides:	L				
Date of last modi	ification:				
Approved:					

University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚTVŠ/ LKSp/13	Course name: Summer Course-Rafting of TISA River
Course type, scope a Course type: Practic Recommended cour Per week: Per stud Course method: pre	ce rse-load (hours): ly period: 36s
Number of ECTS cr	edits: 2
Recommended seme	ster/trimester of the course:
Course level: I., II.	
Prerequisities:	
Conditions for course Conditions for course Attendance Final assessment: Ra	■
Learning outcomes: Learning outcomes: Students have knowled	edge of rafts (canoe) and their control on waterway.
5. Canoe lifting and o	burse: ficulty of waterways fting ning using an empty canoe earrying n the water without a shore contact be out of the water
Recommended litera	ture:
Course language:	
Notes:	

Course assessment Total number of assessed students: 153	
abs	n
45.75	54.25
Provides: Mgr. Dávid Kaško, PhD.	
Date of last modification: 18.03.2019	
Approved:	

University: P. J. Šafá	
Faculty: Faculty of S	
Course ID: ÚTVŠ/ KP/12	Course name: Survival Course
Course type, scope a Course type: Practic Recommended cou Per week: Per stud Course method: cou	ce rse-load (hours): ly period: 36s
Number of ECTS cr	edits: 2
Recommended seme	ester/trimester of the course:
Course level: I., II.	
Prerequisities:	
Conditions for course Conditions for course Attendance Final assessment: con	•
Learning outcomes:	
Students will be far conditions as they wi and demanding situa	miliarized with principles of safe stay and movement in extreme natural ill obtain theoretical knowledge and practical skills to solve the extraordinary ations connected with survival and minimization of damage to health. The n work and students will learn how to manage and face the situations that of obstacles.
Students will be far conditions as they wi and demanding situa course develops tear require overcoming of Brief outline of the c Brief outline of the c Lectures: 1. Principles of behav 2. Preparation and lea 3. Objective and subj 4. Principles of hygic Exercises: 1. Movement in terra	ill obtain theoretical knowledge and practical skills to solve the extraordinary ations connected with survival and minimization of damage to health. The n work and students will learn how to manage and face the situations that of obstacles. course: ourse: viour and safety for movement and stay in unknown mountains adership of tour jective danger in mountains ene and prevention of damage to health in extreme conditions in, orientation and navigation in terrain (compasses, GPS) provised overnight stay
Students will be far conditions as they wi and demanding situa course develops tear require overcoming of Brief outline of the c Brief outline of the c Lectures: 1. Principles of behav 2. Preparation and lea 3. Objective and subj 4. Principles of hygie Exercises: 1. Movement in terra 2. Preparation of imp	ill obtain theoretical knowledge and practical skills to solve the extraordinary ations connected with survival and minimization of damage to health. The n work and students will learn how to manage and face the situations that of obstacles. Fourse: viour and safety for movement and stay in unknown mountains adership of tour jective danger in mountains ene and prevention of damage to health in extreme conditions in, orientation and navigation in terrain (compasses, GPS) provised overnight stay ad food preparation.
Students will be far conditions as they wi and demanding situa course develops tear require overcoming of Brief outline of the c Brief outline of the c Lectures: 1. Principles of behav 2. Preparation and lea 3. Objective and subj 4. Principles of hygie Exercises: 1. Movement in terra 2. Preparation of imp 3. Water treatment ar	ill obtain theoretical knowledge and practical skills to solve the extraordinary ations connected with survival and minimization of damage to health. The n work and students will learn how to manage and face the situations that of obstacles. course: viour and safety for movement and stay in unknown mountains adership of tour jective danger in mountains ene and prevention of damage to health in extreme conditions in, orientation and navigation in terrain (compasses, GPS) provised overnight stay

Course assessment Total number of assessed students: 393	
abs	n
44.53	55.47
Provides: MUDr. Peter Dombrovský, Mgr. Ladis	lav Kručanica, PhD.
Date of last modification: 15.03.2019	
Approved:	

University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	Science
Course ID: ÚFV/ MSB/10	Course name: System Biology Modeling
Course type, scope a Course type: Lectur Recommended cou Per week: 2 / 0 Per Course method: pro	re / Practice rse-load (hours): study period: 28 / 0
Number of ECTS cr	redits: 3
Recommended seme	ester/trimester of the course: 5.
Course level: I.	
Prerequisities:	
Conditions for cours Solving intermediate Exam.	se completion: motivating challenges given at the lectures.
Learning outcomes: To provide an overv field of systems biolo	iew of the computational techniques and achievable results in the emerging
and Anfinsen princip procedures and their Biological polymers Biological databases as an example of nor Molecular interaction approaches. Stochas	modeling. Physical structure of biopolymers. Foldamers, Levinthal paradox le. Essentials of molecular modeling and molecular simulations. Examples of
Recommended litera Alon, Uri. *An Intro	
Bioinformatics*. 2nd	duction to Systems Biology: Design Principles of Biological Circuits*. 1st ll/CRC, 2006. lm, and Laurie J. Heyer. *Discovering Genomics, Proteomics and l ed. Benjamin Cummings, 2006. m. *The Sugar Code: Fundamentals of Glycosciences*. Wiley-VCH, 2009.
Campbell, A. Malcol Bioinformatics*. 2nd	Ill/CRC, 2006. Im, and Laurie J. Heyer. *Discovering Genomics, Proteomics and I ed. Benjamin Cummings, 2006.

Course assessment Total number of assessed students: 200								
А	В	С	D	Е	FX			
92.0	6.0	2.0	0.0	0.0	0.0			
Provides: doc.]	RNDr. Jozef Ulič	ný, CSc.						
Date of last mo	dification: 03.05	5.2015						
Approved:								

University: P. J.	Šafárik Univers	ity in Košice			
Faculty: Faculty	of Science				
Course ID: ÚFV TME1/03	// Course na	ame: Theoretical	Mechanics		
Recommended	ecture / Practice course-load (h Per study peri	e ours):			
Number of ECT	S credits: 6				
Recommended	semester/trimes	ster of the cours	se: 1., 3.		
Course level: I.					
Prerequisities:	ÚFV/VF1a/12				
Conditions for Two tests to dea Examination.	-				
Learning outco To acquaint stud		ples of the theore	etical mechanics		
Lagrange's equ dynamics of rigi	ystems with co ations of motio d bodies. Euler's	n. Hamilton's p equations. Conti	rinciple. Hamilt	ork and d'Alem on's equations. I s. Deformation an eal and viscous fl	Kinematics and d stress tensors.
 2. Taylor T.T.: M 3. Strelkov S.P.: 4. Greiner W.: C 5. Goldstein H.: 	: Methods of An Iechanics: Class Mechanics, Min Classical Mechan Classical Mechan	sical and Quantum Publishers, Mos nics, Springer-Ve anics, Addison-V	m, Pergamon Pre scow, 1985. rlag, Berlin, 201 Wesley, London,		
Course languag 1. Slovak, 2. English	e:				
Notes:					
Course assessm Total number of	ent `assessed studen	its: 175			
А	В	С	D	E	FX
30.86	12.57	15.43	17.71	10.86	12.57
Provides: prof.	RNDr. Michal Ja	aščur, CSc.			

Date of last modification: 27.09.2016

University: P. J.	Šafárik Univer	sity in Košice						
Faculty: Faculty	y of Science							
Course ID: ÚF TEP1/03	ÚFV/ Course name: Theory of the Electromagnetic Field							
Recommended	Lecture / Practic l course-load (l l Per study per	e hours):						
Number of EC	FS credits: 5							
Recommended	semester/trime	ester of the cours	se: 4.					
Course level: I.								
Prerequisities:	ÚFV/VFM1b/1	5 and leboÚFV/V	/F1b/03					
Conditions for Two tests to dea Examination.	-	tion: tasks theory of the	e electromagnetic	e field.				
Learning outco To acquaint stud		iples of a theory	of the electromag	netic field.				
Static magnetic	ons in vacuum. field. Maxwell	Scalar and vector equations in maction of electromag	roscopic media. (
2. Rao N.N.: Ba 3. Greiner W.: C	Classical Elect sic Electromage Classical Electro	rodynamics, John netics with Appli odynamics, Spring	cations, Prentice-	Hall, New Jersey	r, 1972.			
Course languag 1. Slovak, 2. English	ge:							
Notes:								
Course assessm Total number of		nts: 302						
А	В	С	D	Е	FX			
27.48	8.61	17.55	22.19	15.89	8.28			
			1	•				
Provides: doc. 1	RNDr. Jozef Str	ecka, PhD.						
Provides: doc. H Date of last mo								

Faculty: Faculty of		-							
racuity. racuity 0	f Science								
Course ID: ÚFV/ TDF1/99	Course na	Course name: Thermodynamics and Statistical Physics							
Course type, scope Course type: Lec Recommended co Per week: 4 / 2 Po Course method: 1	ture / Practice ourse-load (he er study perio	ours):							
Number of ECTS	credits: 7								
Recommended ser	mester/trimes	ter of the cours	e: 6.						
Course level: I.									
Prerequisities:									
Conditions for cou	urse completi	on:							
Learning outcome	es:								
State parameters.E Absolute tempera	ture and ent				atrix.Statistica				
ensebles.Bose and Literature: P.T.Lan L.D.Landau,and E. Pergamon Press,O:	dsberg,Therm .M.Lifshitz,St	· ·	science,1961.						
Literature: P.T.Lan L.D.Landau,and E	dsberg, Therm .M.Lifshitz, St xford, 1977.	· ·	science,1961.						
Literature: P.T.Lan L.D.Landau,and E. Pergamon Press,O:	dsberg, Therm .M.Lifshitz, St xford, 1977.	· ·	science,1961.						
Literature: P.T.Lan L.D.Landau,and E. Pergamon Press,O: Recommended lite	dsberg, Therm .M.Lifshitz, St xford, 1977.	· ·	science,1961.						
Literature: P.T.Lan L.D.Landau,and E. Pergamon Press,O: Recommended lite Course language:	dsberg, Therm .M.Lifshitz, St xford, 1977. erature: t	atisticalphysics,	science,1961.						
Literature: P.T.Lan L.D.Landau,and E. Pergamon Press,O: Recommended lite Course language: Notes: Course assessmen	dsberg, Therm .M.Lifshitz, St xford, 1977. erature: t	atisticalphysics,	D	E	FX				
Literature: P.T.Lan L.D.Landau,and E. Pergamon Press,O: Recommended lite Course language: Notes: Course assessmen Total number of as	dsberg, Therm .M.Lifshitz, St xford, 1977. erature: t ssessed studen	atisticalphysics, ts: 164		E 1.83	FX 0.0				
Literature: P.T.Lan L.D.Landau,and E. Pergamon Press,O: Recommended lite Course language: Notes: Course assessmen Total number of as A	t B 16.46	atisticalphysics, ts: 164 C 19.51	D						
Literature: P.T.Lan L.D.Landau,and E. Pergamon Press,O: Recommended lite Course language: Notes: Course assessmen Total number of as A 56.1	t sessed studen B 16.46	atisticalphysics, ts: 164 C 19.51 ščur, CSc.	D						