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University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ Course name: Aktivizujúce metódy výučby chémie AMCU/15 Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 2 Per study period: 28 / 28 Course method: present **Number of ECTS credits: 5 Recommended semester/trimester of the course:** 2. Course level: II. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 44  $\mathbf{C}$ A В D Ε FX 100.0 0.0 0.0 0.0 0.0 0.0 Provides: doc. RNDr. Mária Ganajová, CSc., RNDr. Ivana Sotáková, Ph.D. Date of last modification: 03.05.2015 Approved:

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ Cou

**Course name:** Basic Toxicology

ZTOX/04

Course type, scope and the method: Course type: Lecture / Practice

Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14

Course method: present

**Number of ECTS credits: 5** 

**Recommended semester/trimester of the course:** 1.

Course level: II.

**Prerequisities:** 

## **Conditions for course completion:**

## **Learning outcomes:**

Goal of the course is to provide the students with a knowledge of types of toxic substances and their metabolism, safe and handling of toxic substances.

#### **Brief outline of the course:**

Historical aspects, types of toxic substances, types of exposure, dose-response relationship. Disposition of toxic compounds (absorption, distribution, excretion of toxic compounds). Metabolism of toxic compounds. Drugs as toxic substances, food additives and contaminants, environmental pollutans. Statement of chemistry laboratory policy. Safe and handling of toxic substances.

### **Recommended literature:**

G. F. Fuhrman: Allgemeine Toxikologie fuer Chemiker, Teubner Verlag, Stutgart 1984.

V. E. Forbes, T. L. Forbe: Ecotoxicology in Theory and Practice, Chapman&Hall, London 1994.

J. A. Timbrell: Introduction to Toxicology, Taylor&Francis, London 1994.

#### Course language:

Notes:

## Course assessment

Total number of assessed students: 320

Α	В	С	D	Е	FX
21.25	27.5	25.0	17.5	7.5	1.25

Provides: RNDr. Miroslava Matiková Maľarová, PhD.

Date of last modification: 03.05.2015

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ Cour

Course name: Biotechnology

BTC/03

Course type, scope and the method:

Course type: Lecture

Recommended course-load (hours): Per week: 3 Per study period: 42

Course method: present

**Number of ECTS credits: 5** 

**Recommended semester/trimester of the course:** 2.

Course level: II.

**Prerequisities:** 

## **Conditions for course completion:**

test

## **Learning outcomes:**

Students obtained the knowledge of basic biotechnological processes and their applications in agriculture, industry, food production and medicine.

#### **Brief outline of the course:**

Classification of biotechnology, disciplines and subjects which are involved with biotechnology. The fermentation processes, types of bioreactors, impellers, principles of microbial growth, media and substrates for fermentation processes. The bioremediation, production and application of biogas, in-vessel composting. Micro-organisms used to preparation amino acids, their fermentation preparation, isolation and possible uses. The methods of classical Plant Biotechnology. Ethanol fermentation, spirits, production of wine and beer. The biological filters, nutrient removal and the membrane bioreactors. Antibiotics.

#### **Recommended literature:**

E.M.T. El-Mansi et al. ,Fermentation microbiology ang biotechnology,second edition, 2007

Y.H. Hui, Food biochemistry & food processing, Blackwell Publishing 2006

J.E. Smith, Biotechnology, Cambridge university press 2009

## Course language:

## **Notes:**

#### Course assessment

Total number of assessed students: 114

A	В	С	D	Е	FX
50.0	20.18	16.67	7.89	5.26	0.0

Provides: RNDr. Danica Sabolová, PhD.

Date of last modification: 03.05.2015

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/

**Course name:** Chemical Engineering

ZCVU/04

Course type, scope and the method:

Course type: Lecture / Practice Recommended course-load (hours):

Per week: 2 / 1 Per study period: 28 / 14

Course method: present

**Number of ECTS credits: 5** 

Recommended semester/trimester of the course: 2., 4.

Course level: I., II., III.

**Prerequisities:** 

**Conditions for course completion:** 

**Learning outcomes:** 

#### **Brief outline of the course:**

General and Inorganic Engineering; Mineral raw materials; Raw materials processing, transport and holding; Chemical reactors; Chemical metallurgy – Fe, Al, Cu working; Inorganic acids manufacture (H2SO4, HNO3, HCl, HF, H3PO4); Industrial electrochemistry; Industrial fertilizers; Silicate industry – cement manufacture, ceramics; Petrochemistry

### **Recommended literature:**

**Course language:** 

**Notes:** 

**Course assessment** 

Total number of assessed students: 20

A	В	С	D	Е	FX	N	P
20.0	60.0	15.0	5.0	0.0	0.0	0.0	0.0

Provides: doc. RNDr. Zuzana Vargová, Ph.D.

Date of last modification: 23.02.2018

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ Course name: Chemical Excursion CHE2/03 Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: Per study period: 1t Course method: present **Number of ECTS credits: 4 Recommended semester/trimester of the course:** 2. Course level: II. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 94 В  $\mathbf{C}$ A D Ε FX 89.36 10.64 0.0 0.0 0.0 0.0 Provides: doc. RNDr. Zuzana Vargová, Ph.D. Date of last modification: 03.05.2015 Approved:

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ Course name: Chemistry and Didactics of Chemistry I MSSU1/14 Course type, scope and the method: **Course type:** Recommended course-load (hours): Per week: Per study period: Course method: present **Number of ECTS credits: 2** Recommended semester/trimester of the course: Course level: II. Prerequisities: ÚCHV/VKAU/04,ÚCHV/DCH2/15 **Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 95 C Α В D Ε FX 56.84 27.37 13.68 2.11 0.0 0.0 **Provides:** Date of last modification: 03.05.2015 Approved:

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ Course name: Chemistry and Didactics of Chemistry II MSSU2/14 Course type, scope and the method: **Course type:** Recommended course-load (hours): Per week: Per study period: Course method: present **Number of ECTS credits: 2** Recommended semester/trimester of the course: Course level: II. Prerequisities: ÚCHV/VKOCH/03,ÚCHV/DCH2/15 **Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 37 C A В D Ε FX 81.08 10.81 5.41 2.7 0.0 0.0 **Provides:** Date of last modification: 03.05.2015 Approved:

University: P. J. Šafá	rik University in Košice			
Faculty: Faculty of S	cience			
Course ID: ÚCHV/ MPPc/15	Course name: Continuous	practice teaching I		
Course type, scope a Course type: Practic Recommended cour Per week: Per stud Course method: pre	ce rse-load (hours): ly period: 4t esent			
Number of ECTS cr		2		
	ster/trimester of the cours	<b>e:</b> 3.		
Course level: II.				
Prerequisities: ÚCH	V/MPPb/15 and leboÚCHV	/MPPb/03		
Conditions for cours	e completion:			
Learning outcomes:				
Brief outline of the c	ourse:			
Recommended litera	iture:			
Course language:				
Notes:				
Course assessment Total number of asse	ssed students: 108			
	abs	n		
100.0 0.0				
<b>Provides:</b> RNDr. Ivai	na Sotáková, Ph.D., doc. RN	IDr. Mária Ganajová, CSc.		
Date of last modifica	tion: 03.05.2015			
Approved:				

<b>University:</b> P. J. Šafá	rik University in Košice			
Faculty: Faculty of S	cience			
Course ID: ÚCHV/ MPPd/15	Course name: Continuous	practice teaching II		
Course type, scope a Course type: Practic Recommended cour Per week: Per stud Course method: pre	ce rse-load (hours): ly period: 6t esent			
Number of ECTS cr				
	ster/trimester of the cours	<b>e:</b> 4.		
Course level: II.				
Prerequisities: ÚCH	V/MPPc/15			
Conditions for cours	e completion:			
Learning outcomes:				
Brief outline of the c	ourse:			
Recommended litera	iture:			
Course language:				
Notes:				
Course assessment Total number of asses	ssed students: 87			
	abs	n		
100.0 0.0				
<b>Provides:</b> RNDr. Ivar	na Sotáková, Ph.D., doc. RN	Dr. Mária Ganajová, CSc.		
Date of last modifica	ition: 03.05.2015			
Approved:				

COURSE INFORMATION LETTER
University: P. J. Šafárik University in Košice
Faculty: Faculty of Science
Course ID: ÚCHV/ Course name: Cosmetic chemistry KC/03
Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14 Course method: present
Number of ECTS credits: 4
Recommended semester/trimester of the course: 3.
Course level: II.
Prerequisities:
Conditions for course completion: Seminar report on the selected subjects of cosmetic chemistry and its oral presentation connecte with discussion. Terminal examination by oral form.
<b>Learning outcomes:</b> The basic chemical ingredients in cosmetic products, their isolation from natural sources. The construction of some interesting groups of the organic structures and their application in cosmeti industry.
Brief outline of the course:  Skin and its components. The chemistry of lipids. Lipids, their classification (triacylglycerols glycerophospholipids and sfingophoslipids), liposomes as transport systems. Fatty acids an alcohols, natural and synthetic waxes. Surfactants, their classification. Antioxidants. Dyes, their classification, organic and inorganic dyes, natural and synthetic. Biological active compound (amino acids, peptides, proteins hydroxy acids, vitamins, polysaccharides) as the cosmeti ingredients. The chemistry of fragrances. Compounds derived from shikimic acid and mevaloni acid, their biosynthesis, Synthetic fragrances and their construction.
Recommended literature:  1. S. V. Bhat, B. A. Nagasampagi, M. Sivakumar: Chemistry of Natural Products, Springer Narosa 2005, ISBN 81-7319-481-5.  2. G. Ohloff: Scent and Fragrances, Springer-Verlag Berlín Heidelberg 1994, ISBN 3-540-57108-6.  3. D. H. Pybus, CH. S. Sell: The chemistry of fragrances, Royal Society of Chemistry 1999, ISBN 0-8540-528-7.  4. J. McMurry: Organic chemistry, Brooks/Cole, a Thomson Learning Company 2004, Sixth Eddition, ISBN 0534389996.
Course language:

**Notes:** 

Course assessment						
Total number of	f assessed studen	ts: 86				
A	В	С	D	Е	FX	
79.07 15.12 4.65 1.16 0.0 0.0						
Provides: doc.	Provides: doc. RNDr. Miroslava Martinková, PhD.					
Date of last modification: 06.02.2020						
Approved:						

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ Course name: Didaktika chémie I DCH1/15 Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 1/2 Per study period: 14/28 Course method: present **Number of ECTS credits: 4 Recommended semester/trimester of the course:** 2. Course level: II. Prerequisities: ÚCHV/SPC1a/03 **Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 116 C Α В D Ε FX 66.38 18.97 8.62 3.45 2.59 0.0 Provides: doc. RNDr. Mária Ganajová, CSc., RNDr. Ivana Sotáková, Ph.D. Date of last modification: 03.05.2015 Approved:

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ Course name: Didaktika chémie II DCH2/15 Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 1/2 Per study period: 14/28 Course method: present **Number of ECTS credits: 4 Recommended semester/trimester of the course:** 3. Course level: II. Prerequisities: ÚCHV/DCH1/15 **Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 108 C Α В D Ε FX 77.78 13.89 6.48 1.85 0.0 0.0 Provides: doc. RNDr. Mária Ganajová, CSc., RNDr. Ivana Sotáková, Ph.D. Date of last modification: 03.05.2015 Approved:

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ Course name: Digitálne technológie vo výučbe chémie DTCU/15 Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 2 Per study period: 28 / 28 Course method: present **Number of ECTS credits: 5 Recommended semester/trimester of the course:** 3. Course level: II. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 10  $\mathbf{C}$ A В D Ε FX 100.0 0.0 0.0 0.0 0.0 0.0 Provides: doc. RNDr. Mária Ganajová, CSc., RNDr. Ivana Sotáková, Ph.D. Date of last modification: 03.05.2015 Approved:

University: P. J. Šafá	rik University in Košice	,	
Faculty: Faculty of S	cience		
Course ID: ÚCHV/ DPP1/14	Course name: Diplom	a Project I	
Course type, scope a Course type: Recommended cour Per week: Per stud Course method: pre	rse-load (hours): y period:		
Number of ECTS cr	edits: 1		
Recommended seme	ster/trimester of the co	ourse: 1.	
Course level: II.			
Prerequisities:			
Conditions for cours	e completion:		
Learning outcomes:			
Brief outline of the c	ourse:		
Recommended litera	ture:		
Course language:			
Notes:			
Course assessment Total number of asse	ssed students: 64		
	abs		n
	100.0		0.0
<b>Provides:</b>		-	
Date of last modifica	tion: 03.05.2015		
Approved:			

University: P. J. Šafá	rik University in Košice		
Faculty: Faculty of S	cience		
Course ID: ÚCHV/ DPP2/14	Course name: Diplom	a Project II	
Course type, scope a Course type: Recommended cour Per week: Per stud Course method: pre	rse-load (hours): ly period:		
Number of ECTS cr	edits: 2		
Recommended seme	ster/trimester of the co	urse: 2.	
Course level: II.			
Prerequisities:			
<b>Conditions for cours</b>	se completion:		
Learning outcomes:			
Brief outline of the c	ourse:		
Recommended litera	iture:		
Course language:			
Notes:			
Course assessment Total number of asse	ssed students: 63		
	abs		n
	100.0		0.0
<b>Provides:</b>		•	
Date of last modifica	ntion: 03.05.2015		
Approved:			

University: P. J. Šafá	rik University in Košice		
Faculty: Faculty of S	cience		
Course ID: ÚCHV/ DPP3/14	Course name: Diploma	n Project III	
Course type, scope a Course type: Recommended cour Per week: Per stud Course method: pre	rse-load (hours): ly period:		
Number of ECTS cr	edits: 2		
Recommended seme	ster/trimester of the co	urse: 3.	
Course level: II.			
Prerequisities:			
<b>Conditions for cours</b>	e completion:		
Learning outcomes:			
Brief outline of the c	ourse:		
Recommended litera	iture:		
Course language:			
Notes:			
Course assessment Total number of asse	ssed students: 66		
	abs		n
	100.0		0.0
<b>Provides:</b>		<u>'</u>	
Date of last modifica	tion: 03.05.2015		
Approved:			

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ Course name: Diploma Thesis and its Defence DPOU/14 Course type, scope and the method: **Course type:** Recommended course-load (hours): Per week: Per study period: Course method: present **Number of ECTS credits: 14** Recommended semester/trimester of the course: Course level: II. Prerequisities: ÚCHV/DPP3/14 **Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 67 C Α В D Ε FX 80.6 16.42 2.99 0.0 0.0 0.0 **Provides:** Date of last modification: 03.05.2015 Approved:

University: P. J. Šafá	rik University in Košice		
Faculty: Faculty of S	cience		
Course ID: ÚCHV/ DSU1a/10	Course name: Diplomový	seminár z chémie pre XCH	
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	ce rse-load (hours): idy period: 28 esent		
Number of ECTS cr			
	ster/trimester of the cours	<b>e:</b> 2.	
Course level: II.			
Prerequisities:			
Conditions for cours	se completion:		
Learning outcomes:			
Brief outline of the c	ourse:		
Recommended litera	iture:		
Course language:			
Notes:			
Course assessment Total number of asse	ssed students: 11		
	abs	n	
	100.0	0.0	
Provides: doc. RNDr	: Mária Ganajová, CSc.		
Date of last modifica	ntion: 03.05.2015		
Approved:			

University: P. J. Šafá	rik University in Košice				
Faculty: Faculty of S	cience				
Course ID: ÚCHV/ DSU1b/10					
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	ce rse-load (hours): idy period: 28 esent				
Number of ECTS cr					
	ster/trimester of the cours	e: 3.			
Course level: II.					
Prerequisities:					
Conditions for cours	se completion:				
Learning outcomes:					
Brief outline of the c	ourse:				
Recommended litera	iture:				
Course language:					
Notes:					
Course assessment Total number of asse	ssed students: 6				
	abs	n			
100.0 0.0					
Provides: doc. RNDr	. Mária Ganajová, CSc.				
Date of last modifica	ntion: 03.05.2015				
Approved:					

	COURSE INFORMATION LETTER
University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚCHV/ UECH/03	Course name: Introduction to Environmental Chemistry
Course type, scope a Course type: Lectur Recommended cou Per week: 2 / 1 Per Course method: pre	re / Practice rse-load (hours): study period: 28 / 14
Number of ECTS cr	edits: 5
Recommended seme	ster/trimester of the course: 1., 3.
Course level: I., II.	
Prerequisities:	
Conditions for cours Oral examination	se completion:
Learning outcomes: Introduction to topics protection.	s in environmental chemistry and basic procedures applied for environmental
atmosphere. Energy photoprocesses in the environmental pollut Environmental chem metals). Environmentals their protection. The utilization. Energy at	
Oxford University Pr 2. R.A. Bailey, H.M. Academic Press, San 3. G. Schwedt: The F 4. R.N. Reeve, J.D. F	, Stephen J. Duffy: Environmental Chemistry - A Global Perspective, ress, Oxford 2003 Clark, J.P. Ferris, S. Krause, R.L. Strong: Chemistry of the Environment,
Course language:	

**Notes:** 

Course assessment Total number of assessed students: 216					
A	В	С	D	Е	FX
49.54 20.83 15.28 8.33 6.02 0.0					
D 1 1	D 1 1 DND 4 1 C/ 1 / E 1 1 / DID				

**Provides:** doc. RNDr. Andrea Straková Fedorková, PhD.

 $\textbf{Date of last modification:}\ 20.09.2017$ 

<b>University:</b> P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚCHV/ FUMCH1/03	Course name: Introduction to Material Chemistry
Course type, scope a Course type: Lectur Recommended cour Per week: 2 / 1 Per Course method: pre	re / Practice rse-load (hours): study period: 28 / 14
Number of ECTS cr	edits: 5
Recommended seme	ster/trimester of the course: 1., 3.
Course level: I., II.	
Prerequisities:	
Conditions for cours Seminar work. Examination.	e completion:
Learning outcomes: To present the diffe properties.	rent types of functional materials, their atomic structure and mechanical
engineering. Material bonding. Amorphous Crystal lattice defects Deformations and fail Intermediary phases. Phase identification in Steel. Light metals. Materials. Ceramic to Glass. Building binder	es. Materials and human being. Participation of natural science in material l'revolutions. Classification of materials. Atomic structure and interatomic and crystalline materials. Mechanics of materials. Imperfections in solids. Point defects. Line defects. Dislocations. Diffusion. Diffusion mechanisms. ilures, re-crystallization. Deformations. Plastic deformations. Solid solutions. Phases in ceramic systems. Phase transformations. Crystallization of metals. methods. Stress and strain. Structure of metallic and ceramic materials. Alloys. Metallic glasses. Gold. Inorganic non-metallic materials. Ceramic construction pols. Bio-ceramics. Ceramics in cosmos. High-temperature superconductors. ers. Polymers. Essence of polymers. Thermoplastics. Reactoplastics. Polymer l properties of polymers. Natural materials. Wood. Bones. Teeth. Conchs and
2001. Brian S. Mitchell: Ar Materials Engineers,	undamentals of Materials Science and Engineering, John Wiley & Sons,  Introduction to Materials Engineering and Science: For Chemical and
2004. Course language:	

**Notes:** 

Course assessment					
Total number o	f assessed studen	ts: 77			
A	В	C	D	Е	FX
89.61	9.09	0.0	0.0	0.0	1.3
Provides: prof. RNDr. Renáta Oriňaková, DrSc.					
Date of last modification: 20.09.2017					
Approved:					

University: P. J. Šafá	rik University in Košice				
Faculty: Faculty of S	cience				
Course ID: ÚCHV/ MPPb/15	Course name: Scheduled 1	Course name: Scheduled practice teaching			
Course type, scope a Course type: Practic Recommended cour Per week: Per stud Course method: pre	ce rse-load (hours): y period: 36s				
Number of ECTS cr	edits: 1				
Recommended seme	ster/trimester of the cours	e: 2.			
Course level: II.					
Prerequisities: KPE/	MPPa/15,KPE/PDU/15,(KP	PaPZ/PaSPP/09 and leboKPPaPZ/PPgU/15)			
<b>Conditions for cours</b>	e completion:				
Learning outcomes:					
Brief outline of the c	ourse:				
Recommended litera	iture:				
Course language:					
Notes:					
Course assessment Total number of asses	ssed students: 274				
	abs n				
100.0 0.0					
<b>Provides:</b> RNDr. Ivan	na Sotáková, Ph.D., doc. RN	Dr. Mária Ganajová, CSc.			
Date of last modifica	tion: 03.05.2015				
Approved:					

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ | Course name: Selected Topics in Inorganic Chemistry

VKAU/04

Course type, scope and the method:

Course type: Lecture / Practice

Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14

Course method: present

**Number of ECTS credits: 5** 

Recommended semester/trimester of the course: 3.

Course level: II.

**Prerequisities:** 

**Conditions for course completion:** 

**Learning outcomes:** 

**Brief outline of the course:** 

#### **Recommended literature:**

Greenwood, N.N., Earnshaw, A.: Chemistry of the elements I and II, Pergamon Press N.Y., 1993. C. N. R. Rao, A. Muller, A. K. Cheetham: The Chemistry of Nanomaterials (Vol. 1,2), Wiley-VCH,2006.

Atkins O., Overton T., Rourke J., Weller M., Armstrong F.: Inorganic Chemistry, University Press, Oxford, 2006.

## Course language:

**Notes:** 

### **Course assessment**

Total number of assessed students: 75

A	В	С	D	Е	FX
45.33	29.33	20.0	2.67	2.67	0.0

Provides: prof. RNDr. Vladimír Zeleňák, DrSc.

Date of last modification: 03.05.2015

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ Course name: Selected topics in organic chemistry VKOCH/03 Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14 Course method: present **Number of ECTS credits: 5 Recommended semester/trimester of the course:** 3. Course level: II. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 108 C Α В D Ε FX 35.19 25.0 20.37 13.89 5.56 0.0 Provides: doc. RNDr. Ján Imrich, CSc. Date of last modification: 03.05.2015 Approved:

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ | Course name: Special practising the school experiments I

SPC1a/03

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 4 Per study period: 56

Course method: present

**Number of ECTS credits: 5** 

Recommended semester/trimester of the course: 1.

Course level: II.

## **Prerequisities:**

## **Conditions for course completion:**

Continuous checking of theoretical preparation, development of report and presentation. Semestral test

### **Learning outcomes:**

The aim of this subject is learn of basic experimental skillfulness in techniques in school experiment with accent on safety and health protections of students at scholar experimental work.

#### **Brief outline of the course:**

Selection and arrangement of chemical experiments as the demonstrative experiments, or pupils 'experiments to themes basic laws of chemistry, determination of constant physicochemical, factors influence speed of chemical reaction, experiments from electrochemistry, creating gases; preparation works characters of quantitative, interesting experiments of everyday life.

#### **Recommended literature:**

- 1. Ganajová, M., Dzurillová, M. 2005: Školské pokusy z chémie I. UPJŠ v Košiciach, Prírodovedecká fakulta, 140 s. ISBN 80-7097-617-9
- 2. Ganajová, M. 2005: Chemické experimenty s vybranými produktami z obchodu. UPJŠ v Košiciach, Prírodovedecká fakulta, 110 s. ISBN 80-7097-611-X
- 3. Tomeček,O.: Školská experimentálna semimikrosúprava. Učebné pomôcky Banská Bystrica 1980
- 4. The primary and secondary textbook of chemistry
- 5. http://kekule.science.upjs.sk (ŠIS)

#### Course language:

#### Notes:

#### Course assessment

Total number of assessed students: 282

A	В	С	D	Е	FX
66.67	25.53	6.74	1.06	0.0	0.0

Provides: doc. RNDr. Mária Ganajová, CSc., RNDr. Ivana Sotáková, Ph.D.

Date of last modification: 03.05.2015	
Approved:	

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ | Course name: Special practising the school experiments II

SPC1b/03

Course type, scope and the method:

**Course type:** Practice

Recommended course-load (hours): Per week: 3 Per study period: 42

Course method: present

**Number of ECTS credits: 3** 

Recommended semester/trimester of the course: 2.

Course level: II.

**Prerequisities:** 

## **Conditions for course completion:**

The knowledge of the reaction mechanism of the main tests of several organic compounds derivatives and the ability of their laboratory realization are required.

Written tests: at least 51% from each one is required.

### **Learning outcomes:**

The students will become familiar with the basic laboratory skills and techniques that they can apply in demonstrating experiments in their future career as a teacher. The rules of healthy and safety laboratory work are emphasised.

#### **Brief outline of the course:**

Qualitative analysis of organic compounds

Alkanes - preparation of methane

Alkenes - preparation and addition reactions of ethene, addition reactions of  $\beta$ -carotene

Alkynes - preparation of acetylene and studying of its reaction

Aromatic hydrocarbons and their derivatives – preparation and their characteristic reactions

Halogenoderivatives – preparation of chloroethane and iodoform

Hydroxoderivatives and ethers – properties and reactivity - methanol, ethanol, glycerol, preparation of sodium ethanolate, phenols, characteristic properties of diethylether

Carbonyl compounds - preparation and their reactions

Carboxylic acids and their derivatives – preparation and properties

Natural compounds – carbohydrates, proteins, amino acids, lipids

Column chromatography -acetylation reaction of ferrocene - its preparation and separation of the obtained products by column chromatography

Isolation of the fragrant components using steam distillation

Everyday life chemistry

#### Recommended literature:

- 1. Smik, L., Merva, L., Brutovská, A: Technika a didaktika školských pokusov, Vyd.Rektorát UPJŠ,Košice,1988
- 2. Smik, L. a kol.: Špeciálna didaktika chémie II., Vyd. Rektorát UPJŠ, Košice, 1984
- 3. Internal studying material Špeciálne praktikum školských pokusov z organickej chémie

Course languag	~					
Notes:						
Course assessn Total number o	nent f assessed studen	nts: 276				
A	В	С	D	Е	FX	
44.2	28.26 17.03 7.25 3.26 0.0					
Provides: RND	r. Jana Špaková l	Raschmanová, P	hD., RNDr. Ján E	lečko, PhD.		
Date of last mo	odification: 05.02	2.2020				
Approved:						

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ | Course name: Stereochemistry of Inorganic Compounds

SAZ1/15

Course type, scope and the method:

**Course type:** Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

**Number of ECTS credits: 3** 

#### Recommended semester/trimester of the course:

Course level: II.

**Prerequisities:** 

### **Conditions for course completion:**

## **Learning outcomes:**

#### **Brief outline of the course:**

Symmetry, elements of symmetry, point groups, symmetrical properties of orbitals and bonds. Principles of stereochemistry, VSEPR, configuration of molecules, polyhedra, regular and semiregular polyhedra. Valence shells with 4–12 electron pairs, geometry of molecules and periodic system.

### **Recommended literature:**

Kepert, D. L.: Inorganic Stereochemistry. Springer-Verlag, Berlin, 1982.

Kettle, S. F. A.: Symmetry and Structure. John Wiley & Sons, New York, 1985.

## Course language:

#### **Notes:**

#### Course assessment

Total number of assessed students: 22

A	В	С	D	Е	FX
59.09	13.64	18.18	9.09	0.0	0.0

Provides: prof. RNDr. Vladimír Zeleňák, DrSc.

Date of last modification: 03.05.2015

	COURSE INFORMATION LETTER
University: P. J. Šafár	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚCHV/ STA1/03	Course name: Structure Analysis
Course type, scope a Course type: Lectur Recommended cour Per week: 2 / 2 Per Course method: pre	re / Practice rse-load (hours): study period: 28 / 28
Number of ECTS cro	edits: 6
Recommended seme	ster/trimester of the course:
Course level: II.	
Prerequisities:	
Conditions for cours 2 written tests. 30 % The final examination final tests.	n is in a written form. The final mark is based on the results from current and
diffraction methods u	view about the symmetry at the micro- and macrostructure level and about used for the crystal structure determination and they will learn how to use the structure analysis in their own work.
of the diffraction expe	nicrostructure symmetry, individual work with space groups. Theoretical basis eriment. Practical aspects of crystal structure solution. Processing the results of neoretical basis, practical aspects and possibilities of X-ray powder diffraction
Clegg, W. et al.: Crys Hahn, T.: Internationa Stout, G.H. & Jensen	ructure determination, 2nd edition. Springer 2004. stal structure analysis. Principles and practice. Oxford University Press 2009. al tables for crystallography, Vol. A. Kluwer Academic Publishers 2002. , L.H.: X-ray Structure Determination. Macmillan Publishing Co., Inc. 1968. der, L.E.: X-Ray diffraction procedures for polycrystalline and amorphous
Course language: Slovak and English	

**Notes:** 

Course assessment									
Total number of assessed students: 119									
A	В	С	D	Е	FX				
28.57	16.81	26.05	19.33	8.4	0.84				
Provides: doc. RNDr. Ivan Potočňák, PhD.									
Date of last modification: 03.05.2015									
Approved:									

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ Course name: Vybrané kapitoly z chémie

VKCH/10

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours):

Per week: 2 / 1 Per study period: 28 / 14

Course method: present

**Number of ECTS credits: 4** 

Recommended semester/trimester of the course: 1.

Course level: II.

**Prerequisities:** 

#### **Conditions for course completion:**

Terminal examination by written form.

## **Learning outcomes:**

Organic chemistry:

The general review on the basic chemistry of saccharides, lipids, amino acids and peptides.

Inorganic chemstry:

To get acquaintance of the students with the stereochemistry of inorganic compounds, methods of the study and its influence on the properties of the compounds. Moreover to get acquintance of the students with actual direction of inorganic chemistry in the area of nanomaterials.

## **Brief outline of the course:**

Organic chemistry:

Nomenclature of monosaccharides, their stereochemistry (the Fischer projection, the Haworth projection, conformation of sugars). Monosaccharide derivatives. Ascending reactions. Oligosaccharides and polysaccharides.

Lipids, their structure and classification. Groups of lipids. Triacylglycerols, glycerophospholipids sfingophospholipids, glycosphingolipids.

Amino acids, their nomenclature, classification and stereochemistry. Synthesis of amino acids. Nonribosomal construction of peptides.

Inorganic chemistry:

Symmetry, elements of symmetry, point groups, symmetrical properties of orbitals and bonds. Principles of stereochemistry, VSEPR, configuration of molecules, polyhedra, regular and semiregular polyhedra, the use of concept of symmetry in IR and UV-VIS spectroscopy. Nanochemistry - definition, bonds in nanoparticles and nanopowders, interactions between nanoparticles. Unique properties of nanomaterials, new methods of the synthesis of nanomaterials.

#### **Recommended literature:**

- J. McMurry: Organic chemistry, Brooks/Cole, a Thomson Learning Company 2004, Sixth Eddition, ISBN 0534389996.
- J. Chomič: Stereochemistry of inorganic compounds, UPJŠ Košice, 1988.
- K. J. Klabunde, R. M. Richards: Nanoscale Materials in Chemistry, Wiley-CH, 2009.

Course language:  Notes:  Course assessment  Total number of assessed students: 217														
									A	В	С	D	Е	FX
									24.42	28.57	32.72	11.98	1.84	0.46

**Provides:** prof. RNDr. Mária Kožurková, CSc., prof. RNDr. Vladimír Zeleňák, DrSc., doc. RNDr. Miroslava Martinková, PhD.

**Date of last modification:** 29.08.2021