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	1 TT							
	University: P. J. Šafárik University in Košice							
Faculty: Faculty of Science								
Course ID: CJP/ PFAJAKA/07	Course name: Academic English							
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: con	ce rse-load (hours): dy period: 28							
Number of ECTS cr	edits: 2							
Recommended seme	ster/trimester of the course:							
Course level: I., II., N	1							
Prerequisities:								
1 test (10th week), no Presentation on chose Final evaluation- ave	ticipation, assignments handed in on time, 2 absences tolerated o retake.							
of their linguistic cor syntactic aspects, dev	students' language skills - reading, writing, listening, speaking, improvement npetence - students acquire knowledge of selected phonological, lexical and relopment of pragmatic competence - students can effectively use the language with focus on Academic English, level B2.							
Brief outline of the course: Formal and informal English Academic English and its specific features Key academic verbs and nouns Linking words in academic writing, writing a paragraph, word-order, topic sentences Word-formation - affixation abstract Selected aspects of English pronunciation, academic vocabulary Selected functional grammar structures - defining, classifying, epressing opinion, cause-effec paraphrasing								
T. Armer :Cambridge M. McCarthy M., O Zemach, D.E, Rumis Olsen, A. : Active Vo www.bbclearningeng	ncounters, CUP, 2002 English for Scientists, CUP 2011 Dell F Academic Vocabulary in Use, CUP 2008 ek, L.A: Academic Writing, Macmillan 2005 ocabulary, Pearson, 2013							

Course langua English langua	ge: ge, level B2 acco	rding to CEFR.							
Notes:	Notes:								
Course assessment Total number of assessed students: 400									
А	В	С	D	Е	FX				
34.75	22.0	15.75	9.5	6.25	11.75				
Provides: Mgr. Viktória Mária Slovenská									
Date of last modification: 19.09.2022									
Approved: doc	. Mgr. Daniel Jan	cura, PhD.							

University: P. J.	Šafárik Univers	ity in Košice					
Faculty: Faculty of Science							
Course ID: ÚMV/ Course name: Algebra I ALGa/10							
Recommended	ecture / Practice l course-load (h Per study peri	ours):					
Number of ECT	S credits: 7						
Recommended	semester/trimes	ster of the cours	e: 3.				
Course level: I.							
Prerequisities:							
Conditions for o According to the exam	-	on: le semester and in	n view of the res	sults of the writte	en and oral final		
theory related to	nethods of mathe	ematical thinking ster the basic con natical problems.	cepts of linear a		0		
	Z. Fields. System	ms of linear equ minants, Cramer		limination. Map	s, permutations.		
T.S Blyth, E.F. H K. Jänich: Linea	l.: Algebra a teo Robertson: Basic ar algebra, Spring	retická aritmetika i linear algebra, S ger Verlag, 1991.	pringer Verlag,				
Course languag Slovak	e:						
Notes:							
Course assessment Total number of assessed students: 1369							
А	В	С	D	Е	FX		
11.91	11.83	18.99	18.41	28.12	10.74		
Provides: prof. I Janičková, PhD.,		tudenovská, CSc. ga	, RNDr. Igor Fa	brici, Dr. rer. nat	., RNDr. Lucia		
Date of last modification: 16.04.2022							
Approved: doc.	Mgr. Daniel Jan	cura, PhD.					

		UNSE INFURI			
University: P. J.	Šafárik Univers	ity in Košice			
Faculty: Faculty	of Science				
Course ID: ÚM ALG3b/10	V/ Course na	ame: Algebra II f	or informatician	s and physicists	
Recommended	Lecture / Practice l course-load (h 2 Per study peri	e ours):			
Number of EC	FS credits: 7			_	
Recommended	semester/trimes	ster of the course	e: 4.		
Course level: I.	II.				
Prerequisities:	ÚMV/ALGa/10				
Conditions for Exam	course completi	on:			
Brief outline of Vector spaces, spaces. The ram tranformations, transformations of linear transfo Affine spaces, s and quadrics. Recommended A. F. Beardon: A	er knowledge or the course: subspaces. A ba k of a matrix. I matrices of su , regular matrices rmations. ubspaces and the literature: Algebra and Geo	n vector spaces, li sis, a dimension Linear transforma ms and compos s. Similar matrice eir positions. Euc ometry, Cambridg rvey of Modern A	and a character tions and their r itions of linear s. Characteristic clidean spaces, th e University Pre	rization of n-dim matrices. Operati tranformations. vectors and chara ne distance of sub ss, 2005	ensional vector ions with linear Regular linear acteristic values
Course languag Slovak					
Notes:	_				
Course assessm Total number of	ent assessed studen	ts: 317			
A	B	C	D	Е	FX
15.77	10.41	12.93	18.93	32.18	9.78
Provides: doc. I	RNDr. Roman So	ták, PhD., Mgr. 1	Martin Vodička	<u> </u>	l
Date of last mo	dification: 26.03	3.2020			
Approved: doc.	Mgr. Daniel Jan	cura, PhD.			
	<u> </u>	,			

University: P. J. Šaf	ărik University in Košice
Faculty: Faculty of	Science
Course ID: ÚFV/ ABE/18	Course name: Analysis of Biophysical Experiments
Course type, scope Course type: Lectu Recommended cou Per week: 2 / 1 Per Course method: pr	are / Practice arse-load (hours): r study period: 28 / 14
Number of ECTS c	redits: 4
Recommended sem	ester/trimester of the course: 6.
Course level: I.	
Prerequisities:	
Conditions for cour Oral exam where the	rse completion: e students present theoretical knowledge of topics listed in the course syllabus.
experiments, verific	: n overview of the basic knowledge related to data evaluation in biophysical ation of hypotheses and discrimination between different models. Students will for experimental data computer processing.
knowing the uncert discrepancy, compare 2. Checking relation uncertainties in dire	ental measurements, physical units, errors and uncertainties, importance of tainties, estimating uncertainties in repeatable experiments, best estimates, rison of measured and accepted values, comparison of two measured numbers, ships with a graph, fractional uncertainties, multiplying two measured numbers, ect measurements, the square-root rule for counting experiments, sums and s and quotients, independent uncertainties in a sum, arbitrary functions of one

3. Analysis of random uncertainties, random and systematic errors, the mean and standard deviation, the normal distribution, hystograms and distributions, limiting distributions, the standard deviation as 68% confidence limit, rejection of data, Chauvenet's criterion, weighted averages, experimental examples.

4. Least-squares fitting, linear data: the slope and the constant parameter, uncertainty in the measured data, experimental examples, least-squares fits to other curves e.g. polynomial or exponential functions, multiple regression, calibration curves in biophysics and biochemistry.

5. Covariance and correlation, covariance in error propagation, coefficient of linear correlation, quantitative significance r, autocorrelation, cross-correlation, use of correlation functions in monitoring the dynamics of individual molecules.

6. The binomial distribution, probabilities in dice throwing, definition of the binomial distribution, the Gauss distribution of random errors, testing of hypothesis, the properties of the Poisson distribution, applications, Chi squared testing, degrees of freedom and reduced chi squared, probabilities for chi squared, experimental examples, solutions, using Excel calculations.

7. Noise sources in biophysical experiments, mechanical noise, electrical noise (thermal noise, shot noise, interference), noise sources in optical imaging experiments, noise characterisctics: color, power spectrum, signal-to-noise ratio, methods for noise reduction and spectral filtration.

8. Computer processing of experimental data (Origin, Igor), the usage of fitting algorithms, statistical analysis, data plotting in graphs, 3D graphs, statistical graphs, figure preparation for publications.

9. Matlab/Octave: a tool for numerical modeling, complex data fitting with shared parameters, examples and applications.

10. Python: simple still complex tool for data analysis, large set of libraries, application examples: polynomial fitting, Fourier transformation, machine learning.

11. Data analysis in the field of spectral data, data smoothing (moving average, Savitzky-Golay filter, Fourier filter), background subtraction (high-order polynomial fitting, rolling ball algorithm, iterative methods), searching for peak position and intensity, complex spectra as a linear combination of simple contributions.

12. Image processing: using Python for image processing, imaging the relevant regions (ROI – region of interest) and firther analysis, binary thresholding, region separation by color, intensity normalization, border detection.

Recommended literature:

1. J.R. Taylor. An Introduction to Error Analysis: The Study of Uncertainties in Physical Measurements, University Science Books, 1997.

J. Mandel. The Statistical Analysis of Experimental Data, Dover Publications. 1964
 E.J. Billo. Excel for Chemist, Wiley, 2011

Course language:

Slovak, English.

Notes:

Course assessment

Total number of assessed students: 4

А	В	С	D	Е	FX	
75.0	25.0	0.0	0.0	0.0	0.0	
Provides: doc. Mgr. Gregor Bánó, PhD., doc. RNDr. Gabriel Žoldák, PhD.						
Date of last modification: 22.09.2021						

Approved: doc. Mgr. Daniel Jancura, PhD.

University: P. J.	Šafárik Universi	ity in Košice					
Faculty: Faculty	of Science						
Course ID: ÚFV/ Course name: Bachelor Thesis and its Defence BPO/14							
Course type, sco Course type: Recommended Per week: Per Course method	course-load (he study period:						
Number of ECT	S credits: 4						
Recommended s	semester/trimes	ter of the cours	e:				
Course level: I.							
Prerequisities:							
Conditions for c Required number	-		nitting the bachel	or thesis.			
Learning outcom	nes:						
Brief outline of Presentation of professional con	the bachelor the	sis results, answ	ering questions	of the reviewer a	and members of		
Recommended	literature:						
Course languag Slovak or Englis							
Notes:							
Course assessme Total number of		ts: 61					
А	В	С	D	Е	FX		
86.89	8.2	3.28	1.64	0.0	0.0		
Provides:				·			
Date of last mod	lification: 07.12	.2021					
Approved: doc.	Mgr. Daniel Jan	cura, PhD.					

	cience
	Course name: Biochemical Analytical Methods
BAM1/00	Course name. Dioenennear Anaryticar Wethous
Course type, scope an Course type: Lecture Recommended cour Per week: 2 / 1 Per s Course method: pres	e / Practice rse-load (hours): study period: 28 / 14
Number of ECTS cre	edits: 4
Recommended semes	ster/trimester of the course: 5.
Course level: I., II.	
Prerequisities:	
Conditions for course Absence of a maximu Exam carried out in w	-
Learning outcomes: The student will gain in analyzes in the bioc	comprehensive information about the methods and approaches that are used chemical laboratory.
 Processing and inte The effectiveness of reliability 	lytical methods in biochemistry erpretation of results of the chosen system of methods to ensure the required level of analytical or determination of biomacromolecules lytical chemistry s nethods
Recommended literat	ture: Analytical Biochemistry, 1998

Notes:

Teaching is carried out in person or, if necessary, remotely using the tool MS Teams, BigBlueButton, etc. The form of teaching is specified by the teacher at the beginning of the semester, updated continuously.

Course assessment Total number of assessed students: 85							
А	В	С	D	Е	FX		
34.12	20.0	18.82	22.35	4.71	0.0		
Provides: doc. 1	Provides: doc. RNDr. Rastislav Varhač, PhD.						
Date of last modification: 16.11.2021							
Approved: doc.	Approved: doc. Mgr. Daniel Jancura, PhD.						

	University:	ΡJ	Šafárik	University	v in Košice
I	University.	1	Salarik	Oniversity	

Faculty: Faculty of Science

Course ID: ÚCHV/	Course name: Biochemistry Practical
PBC2/99	

Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 4 Per study period: 56

Course method: present

Number of ECTS credits: 4

Recommended semester/trimester of the course: 3.

Course level: I.

Prerequisities:

Conditions for course completion:

Active participation with a maximum of one excused absence without the need for compensation. In case of excused absence from two or more practical exercises (e.g. due to illness), the student agrees with the teacher on alternative dates for practice.

Correctly prepared protocols from all completed tasks.

At least 51% of points from each of the written tests.

Learning outcomes:

To allow students to get practical experience in experimental techniques and methods, currently used in a biochemical research: pipetting, titration, UV/VIS spectrophotometry, thin layer chromatography (TLC), gel electrophoresis, isolation of macromolecules and substances from biological materials and their quantitative and qualitative determination.

Brief outline of the course:

1. Biochemistry laboratory safety rules. Basic biochemical laboratory procedures.

- 2. Qualitative tests for amino acids and proteins.
- 3. Isolation of casein from milk. Determination of protein concentration by Lowry method.

4. Determination of the iodine number by Yasud method . Soap production. Reactions with soap. Oxidation of unsaturated fatty acids.

5. Saponification number of fats and oils. Qualitative test for cholesterol: Salkowsky reaction.

6. Qualitative tests for carbohydrates. Determination of reducing carbohydrates by the Schoorl's method.

7. Determination of reducing and nonreducing carbohydrates in germinant plants.

8. Time-dependent course of enzyme-catalyzed reaction: digestion of gelatin by trypsine.

9. Determination of catalase activity and the first order rate constant. Effect of pH on alpha-amylase activity.

10. Effect of substrate concentration on initial rate of reaction, determination of Km and Vmax for urease-catalyzed hydrolysis of urea.

11. Isolation of DNA from spleen. Isolation of RNA from yeast. Qualitative tests for DNA and RNA components.

12. Determination of vitamin C concentration by 2,4-dinitrofenylhydrazine. Determination of vitamins A, B1, and C.

13. Final evaluation of students.

Recommended literature:

Sedlák, Varhač, Danko, Paulíková, Podhradský: Praktické cvičenia z biochémie, 2020, https://unibook.upjs.sk/sk/chemia/1411-prakticke-cvicenia-z-biochemie

Course language:

Slovak

Notes:

Teaching is carried out in person.

Course assessment

Total number of assessed students: 927

А	В	С	D	Е	FX
57.61	25.67	10.36	4.53	1.62	0.22

Provides: prof. RNDr. Mária Kožurková, CSc., RNDr. Nataša Tomášková, PhD., doc. RNDr. Rastislav Varhač, PhD., RNDr. Danica Sabolová, PhD.

Date of last modification: 17.08.2022

Approved: doc. Mgr. Daniel Jancura, PhD.

University: P. J. Ša	fárik Univers	ity in Košice			
Faculty: Faculty of	Science				
Course ID: ÚFV/ BCHF1/18	Course na	me: Biochemistr	ry for Physicists	Ι	
Course type, scope Course type: Lec Recommended co Per week: 3 / 2 Po Course method: p	ture / Practice ourse-load (h er study perio	ours):			
Number of ECTS	credits: 6				
Recommended ser	nester/trimes	ter of the course	e: 2.		
Course level: I.					
Prerequisities:					
Conditions for cou	irse completi	on:			
Learning outcome	s:				
Brief outline of the	e course:				
Recommended lite	erature:				
Course language:					
Notes:					
Course assessment Total number of as		ts: 12			
A	В	С	D	Е	FX
33.33	16.67	16.67	8.33	25.0	0.0
Provides: doc. RN	Dr. Erik Sedlá	ık, DrSc., doc. RN	NDr. Gabriel Žol	ldák, PhD.	
Date of last modifi	cation: 27.09	.2021			
Approved: doc. M	gr. Daniel Jan	cura, PhD.			

University: P. J. Ša	afárik Univers	ity in Košice			
Faculty: Faculty o	f Science				
Course ID: ÚFV/ BCHF2/18	Course na	me: Biochemist	ry for Physicists	II	
Course type, scop Course type: Lec Recommended co Per week: 3 / 2 P Course method:	ture / Practice ourse-load (h er study peri	ours):			
Number of ECTS	credits: 6				
Recommended ser	mester/trimes	ster of the cours	e: 3.		
Course level: I.					
Prerequisities: ÚF	V/BCHF1/18				
Conditions for cou	urse completi	on:			
Learning outcome	es:				
Brief outline of th	e course:				
Recommended lite	erature:				
Course language:					
Notes:					
Course assessmen Total number of as		ts: 7			
А	В	С	D	Е	FX
42.86	14.29	28.57	14.29	0.0	0.0
Provides: doc. RN	Dr. Erik Sedlá	ik, DrSc., doc. R	NDr. Gabriel Žol	dák, PhD.	
Date of last modif	ication: 16.12	2.2021			
Approved: doc. M	gr. Daniel Jan	cura, PhD.			

	CO	URSE INFORM	MATION LET	ΓER	
University: P. J.	Šafárik Univers	ity in Košice			
Faculty: Faculty	of Science				
Course ID: ÚCH BAC1/04	V/ Course na	me: Bioinorgani	ic Chemistry I		
Course type, sco Course type: Le Recommended Per week: 2 / 1 Course method	ecture / Practice course-load (h Per study perio	ours):			
Number of ECT	S credits: 5				
Recommended s	emester/trimes	ter of the cours	e: 3.		
Course level: I.,	II.				
Prerequisities:					
Conditions for c Test or seminar v examination	-	on:			
	edges about bio als in biology a			ecules, biomateria s, toxic metals for	, , , ,
elements, essent Oxygen carriers processes. Calciu	metalic elemen tial trace elem and oxygen tra im biominerals mistry in pharm	ents). Biocoord nsport proteins. and biomineraliz nacy, chemothera	ination compo Photochemical zation.Toxic me apy (e.g. platin	vstems (biometals, unds, bioligands. process. Catalysis tals. Application c um complexes in anches of life.	Biocatalyzers. and regulation of knowledge of
Atkins. Inorganic 2. Kaim W., Schw Life. Wiley, Chic	Atkins P. W., O c Chemistry. Ox wederski B.: Bi chester 1998.	ford University I binorganic Chem	Press, Oxford 2 listry: Inorganic	M.T., Amstrong F 006. Elements in the C OCP, Oxford 199	Chemistry of
Course language	2:				
Notes:					
Course assessme Total number of a		ts: 350			
A	В	С	D	Е	FX
42.57	27.71	18.57	6.0	4.86	0.29
Provides: doc. R	NDr. Zuzana Va	urgová, Ph.D.			

Date of last modification: 28.10.2021

Approved: doc. Mgr. Daniel Jancura, PhD.

University: P. J. Ša		ity in Košice			
Faculty: Faculty o					
Course ID: ÚFV/ BSIM1/14		me: Biomolecul	ar Simulations		
Course type, scop Course type: Lec Recommended c Per week: 2 / 2 P Course method:	ture / Practice ourse-load (h er study perio	ours):			
Number of ECTS	credits: 5				
Recommended set	mester/trimes	ster of the course	e: 6.		
Course level: I., II	•				
Prerequisities:					
Conditions for con Elaboration and pr programs on proje Q/A part.	esentation of t	the project on giv			
Learning outcome Introduction to act		ics of biomolecul	lar simulations.		
Brief outline of the Structural character as flow of biologic mechanisms. Experience force fields and Carlo methods - a approaches. Compreactions, free em approaches and he	eristics of biole cal information erimental met methods of c lgorithms and putational cha ergy evaluati	h. 3D-structure an hods of structure classical molecul paralelization. < .llenges in biomon, protein fold	d function of fo e determination ar dynamics. M i>Ab initio	oldamers. Recent v and their limitat Molecular dynam molecular dynar tions - simulatio	view on enzyme ions. Empirical ics and Monte nics and hybrid ns of chemical
Recommended lite					
Actual literature re Course language:					
Notes:					
Course assessmen Total number of as		ts: 56			
А	В	С	D	Е	FX
76.79	7.14	12.5	1.79	1.79	0.0
Provides: doc. RN	Dr. Jozef Ulič	ný, CSc.		·	
		J ,			
Date of last modif	ication: 27.03	<i>.</i>			

University: P. J. S	Safárik Universi	ty in Košice				
Faculty: Faculty	of Science					
Course ID: ÚFV/ BFSb1/18	7/ Course name: Biophysical Seminary I					
Course type, scop Course type: Pr Recommended Per week: 1 Per Course method:	actice course-load (ho study period:	ours):				
Number of ECTS	S credits: 1					
Recommended so	emester/trimes	ter of the cour	se: 3.			
Course level: I.						
Prerequisities:						
Conditions for co Independent indiv	-		e participation or	n seminars. Final	diploma thesis.	
Learning outcom Completing this s in comprehensive	eminar, the stud		-	•	oloma thesis and	
Brief outline of the Seminar on select students.		biophysical re	search and topics	related to the fir	nal theses of the	
Recommended li The literature wil		led by supervis	ors of the theses.			
Course language English language	:					
Notes:						
Course assessme Total number of a		s: 8				
A	В	С	D	Е	FX	
100.0	0.0	0.0	0.0	0.0	0.0	
Provides: doc. M	gr. Daniel Jancu	ıra, PhD.	·	<u>.</u>	•	
	fination, 12.07	2022				
Date of last modi	Incation: 12.07.	2022				

University: P. J. Ša	fárik Univers	ity in Košice			
Faculty: Faculty of	Science				
Course ID: ÚFV/ BFSb2/18	Course na	me: Biophysical	Seminary II		
Course type, scope Course type: Prac Recommended co Per week: 1 Per s Course method: p	tice ourse-load (ho tudy period:	ours):			
Number of ECTS	credits: 1				
Recommended sen	nester/trimes	ter of the course	e: 4.		
Course level: I.					
Prerequisities:					
Conditions for cou	rse completi	on:			
Learning outcome	s:				
Brief outline of the	course:				
Recommended lite	rature:				
Course language:					
Notes:					
Course assessment Total number of ass		ts: 6			
A	В	С	D	Е	FX
100.0	0.0	0.0	0.0	0.0	0.0
Provides: doc. Mgr	. Daniel Janc	ura, PhD.			
Date of last modifi	cation: 12.07	.2022			
Approved: doc. Mg	gr. Daniel Jan	cura, PhD.			

University: P. J. Ša	afárik Univers	ity in Košice			
Faculty: Faculty o	f Science				
Course ID: ÚFV/ BSSBF/18	Course na	me: Biophysics			
Course type, scop Course type: Recommended c Per week: Per st Course method:	ourse-load (h tudy period:				
Number of ECTS	credits: 4				
Recommended set	mester/trimes	ster of the cours	e:		
Course level: I.					
Prerequisities: ÚF ÚFV/EMBF2/18 a:)2 and ÚFV/BF	B1/14 and ÚFV/E	EMBF1/18 and
Conditions for co	urse completi	on:			
Learning outcome	es:				
Brief outline of th	e course:				
Recommended lit	erature:				
Course language:					
Notes:					
Course assessmen Total number of as		ts: 5			
A	В	С	D	E	FX
20.0	60.0	20.0	0.0	0.0	0.0
Provides:				•	•
Date of last modif	ication: 15.12	2.2021			
Approved: doc. M	gr. Daniel Jan	cura, PhD.			

University: P. J. Šafárik University in Košice
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Faculty: Faculty of Science

Course ID: ÚFV/	Course name: Biophysics in Biomedicine and Biotechnologies
BFBB/18	

Course type, scope and the method: Course type: Lecture Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of ECTS credits: 3

Recommended semester/trimester of the course: 2.

Course level: I.

Prerequisities:

Conditions for course completion:

Elaboration of a written essay on a tpoic related to the application of biophysics in biomedicine and biotechnologies.

Learning outcomes:

This course will provide an opportunity for students to obtain knowledge about the utility of biophysics in biomedical research and the application in biotechnology progress. Completing this course, the students should be able to evaluate importance of biophysics for development of biomedical and biotechnology industry.

Brief outline of the course:

Week 1

Ion channels in personalized medicine. Basic knowledge about the structure and function of ion channels in cells. The relationship between the development of various diseases (cardiovascular, neurodegenerative, tumor) and ion channels. Ion channels as tools for drug development. Ion channels as targets for targeted therapy.

Week 2

Trans-disciplinary research - workflow for the identification of antivirals (high-throughput Xray crystallography of complexes, modeling, epitopes, in vitro essays, and the role of XBI). MicroCT with ROI for the study of the pathology of Covid-19-induced lung neovascularization. Experimental monitoring of non-equilibrium structural dynamics - photosynthetic reaction centers, opsins, photolysis of water for hydrogen energy.

Week 3

Application of NiR photobiostimulation in the therapy of neurodegenerative diseases - the principle of photobiostimulation, cell chromophores, application of NiR photobiostimulation in the treatment of Alzheimer's disease and Parkinson's disease

Week 4

In vitro evolution of proteins. Evolutionary techniques in the development of proteins / enzymes with new properties - interconnection of the methods of biophysics, bioinformatics, biochemistry and molecular biology. Design of mutations by methods of bioinformatics and molecular biology, selection and evolution by methods of molecular biology, analysis of properties - solubility, stability,

activity - of newly developed proteins by the methods of biochemistry and biophysics. Importance and application of proteins / enzymes with improved properties. Week 5

Carcinogenesis as an evolutionary process. According to current knowledge, cancer is the result of somatic mutations and epigenetic changes (epimutations), by which cells acquire new properties affecting the ratio between proliferation and apoptosis of individual cells, represented by the so-called the fitness of individual cells depending on the particular environment, which creates a selection pressure on the cells. Because cell properties are inherited during replication, all prerequisites for evolution are met and carcinogenesis can be studied and modeled as an evolutionary process.

Week 6

Lab-on-chip technology. Microfluidic systems. Light-controlled microrobots. Two-photon polymerization of microstructures. Construction of experimental facilities. Time-resolved laser spectroscopy of photosensitive drugs (phosphorescence, fluorescence, transient absorption). Singlet oxygen detection.

Week 7

Biophysical methods for the study of monoclonal antibodies and other therapeutic proteins. Biological function of antibodies, primary and secondary immune response, classes and subclasses of antibodies, role of individual types of antibodies, primary, secondary, tertiary and quaternary structure of antibodies, examples of therapeutic antibodies: trastuzumab and adalimumab. Preparation of antibodies, overview of monitored physicochemical properties of antibodies, critical quality attributes.

Week 8

Application of bioinformatics in biomedicine. The development of new technologies has provided us with an astonishing amount of information about the human gene, including individual differences. In addition to the natural need for database storage of information obtained in this way, there exists also a need to conceptualize procedures, analyzes and data processing in order to understand, model and address complex challenges such as pharmacogenetics and personalized medicine.

Week 9

Mitochondria in diseases and aging. The force of proton motion and the origin of life. Importance of cellular respiration for multicellular organisms. Mitochondria and the origin of the complexity of life. Mitochondrial aging theory. Mitochondria and diseases - origin, course and therapy. Week 10

Optical imaging, diagnostics and therapy in medicine. Modern trends in optical imaging methods (fluorescence and Raman imaging). Diagnosis and therapy (interaction of light with biological tissues, selected methods of medical optical diagnostics and therapy).

Weeks 11 and 12

Individual study of the selected texts about the applications of biophysics in biomedicine and biotechnologies.

Recommended literature:

1. E. Schrödinger. What is life? Cambridge University Press, 1992.

2. T. Hülswitt a R. Brinzanik. Budeme žít věčne? Kniha Zlín , 2012

3. J. Dowsett, P.A. Kenny a R.E. Johnston. The physics of diagnostic imaging. Hodder Arnold, 2006.

4. M.A. Hamblin a P. Mroz. Advances in photodynamic therapy. Artech House, 2008.

5. Súbor aktuálnych vedeckých publikácií

Course language:

Notes:					
Course assessm Total number o	nent f assessed studen	ts: 12			
А	В	С	D	Е	FX
91.67	8.33	0.0	0.0	0.0	0.0
Gabriel Žoldák,	PhD., doc. Mgr.	Daniel Jancura,	., doc. RNDr. Joz PhD., RNDr. Bra ová, PhD., doc. F	nislav Brutovský	, CSc., doc.
Date of last mo	dification: 12.07	2.2022			
· 1 1	. Mgr. Daniel Jan	D1-D			

University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	
Course ID: ÚBEV/ BS1/03	Course name: Biostatistics
Course type, scope a Course type: Lectur Recommended cour Per week: 2 / 2 Per Course method: pre	re / Practice rse-load (hours): study period: 28 / 28
Number of ECTS cro	edits: 6
Recommended seme	ster/trimester of the course: 3., 5.
Course level: I.	
Prerequisities:	
Passing the continual	on practicals, including successful solving of the assigned numerical examples.
	ts with knowledge on basic principles of statistic methods used in biology and ation in statistical evaluation of experimental results, and with the principles riments, as well.
 2.Basic principles of t and variability of data 3. Theoretical and em 4. Reliability of estim 5. Statistical sampling 6. One-way and mult 7. Regression analysis 8. Correlations. 9. Non-parametrical m 10. Design and planm 11. Aanalysis of time 12. Analysis of quality 	etical background of biostatistics. he probability theory. Descriptive statistics: variables, measures of mean value a. pirical distributions. Experimental sampling from the normal distribution. nations. Testing of hypotheses. I and IItype errors. g. Comparison of two groups. iple analysis of variance. Tests for multiple comparisons. s. methods. ing of biological experiments. series.
Snedecor, G.W., Coch	rstanding biostatistics. Mosby Year Book, 1991 rran,W.G.: Statistical methods. The Iowa state university, Ames, 1972. M.Hernandez: Biostatistics. A guide to design, analysis and dicovery.
Course language:	

Notes:					
Course assessm Total number of	nent f assessed studen	ts: 259			
А	В	С	D	Е	FX
4.63	7.72	20.08	24.71	32.82	10.04
Provides: prof. RNDr. Beňadik Šmajda, CSc.					
Date of last modification: 21.10.2021					
Approved: doc.	Approved: doc. Mgr. Daniel Jancura, PhD.				

University: P. J. Ša	afárik Univers	ity in Košice					
Faculty: Faculty of	f Science						
Course ID: ÚFV/ BFB1/14	Course name: Cell Biophysics I						
Course type, scope Course type: Lec Recommended co Per week: 3 Per s Course method: 1	ture ourse-load (h study period:	ours):					
Number of ECTS	credits: 4						
Recommended ser	nester/trimes	ster of the course	e: 5.				
Course level: I., II.							
Prerequisities:							
Conditions for cou	ırse completi	on:					
Learning outcome	es:						
Brief outline of the	e course:						
Recommended lite	erature:						
Course language:							
Notes:							
Course assessmen Total number of as		ts: 30					
A	В	С	D	Е	FX		
43.33	23.33 13.33 20.0 0.0 0.0						
Provides: RNDr. G	abriela Fabrie	ciová, PhD.					
Date of last modifi	ication: 12.07	2.2022					
Approved: doc. M	gr. Daniel Jan	cura, PhD.					

University: P. J.	Šafárik Univers	ity in Košice			
Faculty: Faculty	y of Science				
Course ID: CJP PFAJKKA/07	Course na	me: Communica	ative Competenc	e in English	
Course type: F Recommended Per week: 2 Pe	ope and the met Practice I course-load (h er study period: d: combined, pre	ours): 28			
Number of EC	FS credits: 2				
Recommended	semester/trimes	ter of the cours	e:		
Course level: I.	, II., N				
Prerequisities:					
two classes at th 2 credit tests (pr Final evaluation Final grade will FX 64 % and le Learning outco Brief outline of Recommended www.bbclearnin	ne most. resumably in wea a consists of the s be calculated as t ss. mes: the course: literature: ngenglish.com	eks 6/7 and 12/13 acores obtained fo follows: A 93-10	8) and an oral properties (50 or the 2 tests (50 0 %, B 86-92%,	nts. Students are esentation in Eng 0%) and the prese C 79-85%, D 72-7	lish. ntation (50%). 78%, E 65-71%,
McCarthy M., C Fictumova J., C Principal, 2008. Peters S., Gráf	eccarelli J., Long	g T.: Angličtina, l se. Polyglot, 200	konverzace pro j 07.	mediate. CUP, 19 pokročilé. Barrist	
Course languag English languag	ge: ge, B2 level acco	rding to CEFR			
Notes:					
Course assessm Total number of	ent f assessed studen	ts: 289			
А	В	С	D	Е	FX
44.64	20.76	17.65	7.96	6.23	2.77
Provides: Mgr.	Barbara Mitríkov	vá, Mgr. Viktória	Mária Slovensk	tá	
Date of last mo	dification: 12.02	.2023			

Approved: doc. Mgr. Daniel Jancura, PhD.

	cience
Course ID: CJP/ PFAJGA/07	Course name: Communicative Grammar in English
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: course	ce rse-load (hours): dy period: 28
Number of ECTS cr	edits: 2
Recommended seme	ster/trimester of the course:
Course level: I., II., N	۸
Prerequisities:	
by given deadlines. Powerpoint presentat Final Test - end of se Final assessment = av Grading scale: A 93- Learning outcomes: The development of so of their communic	ticipation (maximum 2 absences tolerated), homework assignments completed ion of a topic related to the study field. mester, no retake verage of test and presentation. 100%, B 86-92%, C 79-85%, D 72-78%, E 65-71%, FX 64% and less students' language skills - reading, writing, listening, speaking, improvement ative linguistic competence. Students acquire knowledge of selected
phonological, lexical	and syntactic aspects, development of pragmatic competence. Students can
efectively use the lan level B2.	and syntactic aspects, development of pragmatic competence. Students can aguage for a given purpose, with focus on Academic English and English on
efectively use the lan level B2. Brief outline of the c Selected aspects of E Word formation Contrast of tenses in The passive voice Types of Conditional Phrasal verbs and En	and syntactic aspects, development of pragmatic competence. Students can aguage for a given purpose, with focus on Academic English and English on ourse: nglish grammar and pronunciation English

Notes:					
Course assessm Total number of	ent f assessed studen	ts: 432			
А	В	С	D	Е	FX
39.81	19.91	16.2	8.1	5.79	10.19
Provides: Mgr.	Lenka Klimčáko	vá			
Date of last mo	dification: 13.09	.2022			
Approved: doc.	Mgr. Daniel Jan	cura, PhD.			

University: P. J. Šafárik University in Košice
Faculty: Faculty of Science
Course ID: KGER/ Course name: Communicative Grammar in German Language
Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: present

Number of ECTS credits: 2

Recommended semester/trimester of the course:

Course level: I., II.

Prerequisities:

Conditions for course completion:

Active participation in class and completed homework assignments. Students are allowed to miss 2 classes at the most (2x90 min.). 2 control tests during the semester. Final grade will be calculated as follows: A 93-100 %, B 86-92%, C 79-85%, D 72-78%, E 65-71%, FX 64 % and less.

Learning outcomes:

The aim of the course is to identify and eliminate the most frequent grammatical errors in oral and written communication, learning language skills of listening comprehension, speaking, reading and writing, increasing students 'language competence (acquisition of selected phonological, lexical and syntactic knowledge), development of students' pragmatic competence (acquisition of the ability to express selected language functions), development of presentation skills, etc.

Brief outline of the course:

The course is aimed at practicing and consolidating knowledge of morphology and syntax of German in order to show the context in grammar as a whole. The course is intended for students who often make grammatical errors in oral as well as written communication. Through the analysis of texts, audio recordings, tests, grammar exercises, monologic and dialogical expressions of students focused on specific grammatical structures, problematic cases are solved individually and in groups. Emphasis is placed on the balanced development of grammatical thinking in the communication process, which ultimately contributes to the development of all four language skills.

Recommended literature:

Dreyer, H. – Schmitt, R.: Lehr- und Übungsbuch der deutschen Grammatik. Hueber Verlag GmbH & Co. Ismaning, 2009.

Krüger, M.: Motive Kursbuch, Lektion 1 – 30. Huebert Verlag GmbH & Co. Ismaning, 2020. Brill, L.M. – Techmer, M.: Deutsch. Großes Übungsbuch. Wortschatz. Huebert Verlag GmbH & Co. Ismaning, 2011.

Földeak, Hans: Sag's besser!. Grammatik. Arbeitsbuch für Fortgeschrittene. Huebert Verlag GmbH & Co. Ismaning, 2001.

Geiger, S. – Dinsel, S.: Deutsch Übungsbuch Grammatik A2-B2. Huebert Verlag GmbH & Co. Ismaning, 2018.

Dittelová, E. – Zavatčanová, M.: Einführung in das Studium der deutschen Fachsprache. Košice: ES UPJŠ, 2000.

Course langua German, Slova	0						
Notes:							
Course assessm Total number of	nent of assessed studen	ts: 56					
А	В	B C D E FX					
60.71	10.71	10.71 8.93 3.57 8.93 7.14					
Provides: Mgr. Ulrika Strömplová, PhD.							
Date of last modification: 12.07.2022							
Approved: doc	. Mgr. Daniel Jan	cura, PhD.					

University: P. J. Šaf	árik University in Košice	
Faculty: Faculty of	Science	
Course ID: ÚFV/ POF1a/99	Course name: Computational Physics I	_
Course type, scope Course type: Lectu Recommended cou Per week: 2 / 1 Per Course method: pr	are / Practice arse-load (hours): r study period: 28 / 14	
Number of ECTS c	redits: 4	
Recommended sem	ester/trimester of the course: 6.	

Course level: I.

Prerequisities: ÚFV/NUM/10

Conditions for course completion:

To successfully complete the course, the student must demonstrate a sufficient degree of understanding of the principles of computer solution of some typical physical problems. The basis of continuous assessment is participation and activity in exercises and work on assignments. The course ends with a final oral exam, the completion of which is conditional on the submission of all four assignments (projects) electronically and with the attached computer program. The credit evaluation of the course takes into account the following student workload: direct teaching (2 credits) and individual work on projects (2 credits). The minimum threshold for completing the course is to obtain at least 50% of the total score, using the following rating scale: A (90-100%), B (80-89%), C (70-79%), D (60- 69%), E (50-59%), F (0-49%).

Learning outcomes:

To teach the basic principles of computer solution of some typical physical problems. The course covers both the area of deterministic methods for solving problems by ordinary and partial differential equations as well as the area of stochastic Monte Carlo simulations and thus forms the basis for further study of more advanced computer methods contained in the follow-up course Computational Physics II.

Brief outline of the course:

- 1. Introduction to dynamical systems.
- 2. Numerical solution of systems of ordinary differential equations with initial condition.
- 3. Euler's method, convergence, error estimation and order of the method. One-step methods, Tylortype and Runge-Kuta (RK2, RK4) methods.
- 4. Multistep methods, general linear method (explicit, implicit). Methods based on numerical quadrature.
- 5. Boundary value problems for ordinary differential equations.
- 6. Numerical solution of partial differential equations (PDE). Difference methods, their consistence, convergence and stability. Elliptic PDE.
- 7. Parabolic PDE, diffusion equation. Explicit and implicit methods.

8. Introduction to the Monte Carlo method. Monte Carlo integration and application in statistical physics.

9. Basics of probability theory. Monte Carlo estimate of mean and standard deviation. Central theorem of Monte Carlo sampling.

10. Simple and importance sampling. Markov chain. Perron-Frobenius theorem. Metropolis algorithm, detailed balance condition.

11. Monte Carlo simulations of lattice spin systems - application to Ising model.

12. Statistical analysis of Monte Carlo data.

Recommended literature:

Basic literature:

POZRIKIDIS, C.: Num. Comp. in Science and Engineering, Oxford Univ. Press, 2008.

GARCIA A.L.: Numerical Methods for Physics, Prentice-Hall, 1994.

LANDAU D.P., BINDER K.: A Guide to Monte Carlo Simulations in Statistical Physics,

Cambridge Univ. Press, 5-th edition, 2021.

Other literature:

BERG, B.A.: Introduction to Markov Chain Monte Carlo Simulations and Their Statistical Analysis (http://www.worldscibooks.com/etextbook/5904/5904_intro.pdf)

JANKE, W.: Monte Carlo Simulations of Spin Systems (http://www.physik.uni-leipzig.de/~janke/ Paper/spinmc.pdf)

Course language:

Notes:

Course assessment

Total number of assessed students: 130

А	В	С	D	Е	FX	Ν	Р
30.0	18.46	12.31	15.38	16.92	2.31	0.0	4.62

Provides: prof. RNDr. Milan Žukovič, PhD.

Date of last modification: 14.09.2021

Approved: doc. Mgr. Daniel Jancura, PhD.

University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚBEV/ CYT1/15	Course name: Cytology
Course type, scope a Course type: Lectur Recommended cou Per week: 3 / 2 Per Course method: pre	re / Practice rse-load (hours): study period: 42 / 28
Number of ECTS cr	redits: 6
Recommended seme	ester/trimester of the course: 1.
Course level: I.	
Prerequisities:	
Conditions for cours Practicals graduation each); Oral examinat	(without absence); Two written tests graduation (min. 70 % fruitfulness of
Learning outcomes:	

Learning outcomes:

To provide the students with knowledge of basic principles of cell microscopic and submicroscopic structure and function.

Brief outline of the course:

Lectures:

1.) Cell theory. Cell. 2.) Organization of living systems. 3.) Biological membranes. 4.) Transfer of substances across membranes. 5.) Cell wall of plant cells. 6.) Surface structures of cells. Extracellular matrix. Cell movement. 7.) Intercellular connections. 8.) Cytoskeleton. 9.) Cell nucleus. 10.) Mitochondria and cellular metabolism. 11.) Plastids and vacuoles. 12.) Ribosomes. Endoplasmic reticulum. Golgi apparatus. Lysosomes. 13.) Differentiation, aging and cell death, pathological changes in cells.

Exercises:

1.) Safety at work in a cytomorphological laboratory. Conditions for successful completion of exercises. 2.) Basics of optics. Origin and construction of the image with a magnifying glass and a microscope. 3.) Microscopic technique. 4.) Shape and size of cells. 5.) Principle of fluorescence and confocal microscopy. 6.) Control test. Vacuole. 7.) Cytoplasm movement. 8.) Nucleus and nucleolus. 9.) Cytoplasmic membrane. 10.) Osmotic processes. 11.) Cell inclusions. 12.) Cell walls of plant cells. 13.) Cell counting. Control test.

Recommended literature:

K.Kapeller, H.Strakele: Cytomorfológia. Osveta Martin, 1999

M.Babák, J.Šamaj: Cytológia. Univerzita Komenského Bratislava, 2002

Alberts B., Bray D., Johnson A., Lewis J.: Základy buněčné biologie. Espero Publishing, 2003 Campbell N. a Reece J.: Biologie. Computer Press, 2006

Kleban J., Mikeš J., Jendželovská Z., Jendželovský R., Fedoročko P.: Cytológia pracovný zošit na praktické cvičenia, 2018

Course language:

Notes:

notes.								
Course assessment Total number of assessed students: 946								
A B C D E FX								
14.16	19.77	28.54	19.87	16.6	1.06			
	Provides: doc. RNDr. Rastislav Jendželovský, PhD., RNDr. Zuzana Jendželovská, PhD., RNDr. Jana Vargová, PhD.							
Date of last modification: 08.09.2021								
Approved: doc.	Mgr. Daniel Jan	cura, PhD.						

University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: CJP/ PFAJ4/07	Course name: English Language of Natural Science
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	ce rse-load (hours): dy period: 28
Number of ECTS cr	edits: 2
Recommended seme	ster/trimester of the course: 4.
Course level: I.	
Prerequisities:	
2 classes at the most Continuous assessmen 1 credit test taken pre- 1 project (quiz on the 5 LMS quizzes (25% In order to be admitted assessment The exam test results represent the other 50 The final grade for the A 93-100, B 86-92, C	in class and completed homework assignments. Students are allowed to miss ent: esumably in weeks 6/7 topic of the student's field of study) 25% of the continuous assessment of the continuous assessment) ed to the final exam, a student has to score at least 65 % from the continuous represent 50% of the final grade for the course, continuous assessment results
in English for specific Students obtain know English, improve the	ents' language skills (speaking, writing, reading and listening comprehension) c and academic purposes and development of students' linguistic competence vledge of selected phonological, lexical and syntactic aspects of professional ir pragmatic competence - students can effectively use the language for a given presentation skills at B2 level (CEFR) with focus on terminology of natural
 6. Expressing cause a 7. Describing structure 8. Explaining process 	dying language f scientific language lemic study terminology and concepts and effect res

10. Talking about problem and solution

- 11. Referencing authors
- 12. Giving examples
- 13. Visual aids and numbers
- 14. Referencing time and place

Presentation topics related to students' study fields.

Recommended literature:

lms.upjs.sk - e-kurz Odborný anglický jazyk pre prírodné vedy.

Redman, S.: English Vocabulary in Use, Pre-intermetdiate, Intermediate. Cambridge University Press, 2003.

Armer, T.: Cambridge English for Scientists. CUP, 2011.

Wharton J.: Academic Encounters. The Natural World. CUP, 2009.

P. Fitzgerald : English for ICT studies. Garnet Publishing, 2011.

https://worldservice/learningenglish, https://spectator.sme.sk

www.isllibrary.com

linguahouse.com

Course language:

English, level B2 (CEFR)

Notes:

Course assessment

Total number of assessed students: 3056

Trovides. Mgi. L	Lenka Klimcako	va, Mgr. Viktoria	a Maria Slovensk	a			
Provides: Mgr. Lenka Klimčáková, Mgr. Viktória Mária Slovenská							
38.29	26.18	16.46	9.55	7.46	2.06		
A	В	С	D	Е	FX		

	COURSE INFORMATION LETTER
University: P. J. Šafán	rik University in Košice
Faculty: Faculty of So	cience
Course ID: ÚFV/ ZPU1/03	Course name: Essentials of UNIX Programming
Course type, scope an Course type: Lectur Recommended cour Per week: 1 / 2 Per s Course method: pre	e / Practice rse-load (hours): study period: 14 / 28
Number of ECTS cro	edits: 4
Recommended seme	ster/trimester of the course: 2.
Course level: I.	
Prerequisities:	
	e completion: I's programming skills In of the program to solve the given task at the end
1	with basic programming skills necessary for solving problems which require ric methods, simulation techniques and computer data processing.
(*,?). File ownership management. Manual 2nd week: C program GCC Compiler. Form Arithmetic operators. 3th week: Control fl operators. Loops "wh logical operators. Con 4th week: Functions functions. User defin automatic variables. 5th week: Library fu functions (cos, sin, e (rint, round, floor, cei inclusion. Bit operato 6th week: Pointers an Functions for memor	cs: Characteristics. Linux distributions. UNIX/LINUX filesystem. Wildcards and permissions. Command line. Shell. Basic LINUX commands for file pages. ning language. Source code. C language syntax. Structure of C programs. natted output (printf). Declarations and types of variables. Operator sizeof. Assignement operators. Indexed variables (arrays). Text strings as arrays. low. Control structures. Statements and blocks. Increment and decrement tile", "for" and "do while". Break and continue statements. Relational and nditional expressions. Syntax of the "switch" statement. . Declaration of function. Arguments of functions. Return of values by led functions. Scope and lifetime of variables. Storage classes - static and unctions. Header files. Mathematical library (math.h). Basic mathematical exp, log). Generator of random numbers (function rand). Rounded values il). Symbolic constants. The C preprocessor: macro substitution, conditional rs. d addresses (&). Operator of dereferencing (*). Dynamic memory allocation. ry allocation and deallocation (malloc, calloc, free). Pointers and function d input (scanf). Structures and unions. Structure FILE. Formatted writing to/

8th week: Basics of C++. OOP (Object oriented programming) paradigm. Data abstraction. Class, object. Data encapsulation. Member functions. Public and private parts of class. Difference between class and structures.

9th week: Constructor and destructor. Dynamic allocation and deallocation of memory by operators new and delete. Operator overloading. Polymorfism and inheritance.

10th week: Memory Layout of a Process in Linux. Monitoring of runing processes (commands ps, top). Filesystem /proc. Process priorities and scheduling. Signals. Running, stopping and ending processes in background. Ignoring hangup signal by command nohup. Commands bg, fg, jobs. Delayed start of processes - commands at, atq and atrm.

11th week: The Linux programming Interface. System calls. Process identifier (PID), function getpid. Signals -fundamental concepts. Interprocess communication via signals. Signal mask. Commands "kill" and "raise". Changing signal dispositions. Designing signal handlers, commands signal and sigaction. System data types.

12th week: Time Functions: time a gettimeofday. Time-conversion functions. Structures timeval, timespec and tm. Real vs. CPU time. Sleepers, functions sleep, usleep. Interval timers. Nanosecond timers and sleepers.

13th week: Multithreading (API Pthread). Thread ID. Thread creation and termination. Threads synchronization - mutexes. Communication between threads. Thread signal mask. Thread timers via signals.

Recommended literature:

William E. Shotts, Jr., The Linux Command Line: A Complete Introduction, No Starch Press, 2012

Kernighan, B. W., Ritchie, D. M., C programming language, 2nd edition, Prentice Hall PTR, 1988

Stroustrup, B., The C++ Programming Language, Pearson Education, 2013

Kerrisk, M, The Linux Programming Interface: A Linux and UNIX System Programming Handbook, No Starch Press, 2010

Course language:

Notes:

Course assessment

Total number of assessed students: 170

А	В	С	D	Е	FX	
56.47	16.47	20.0	3.53	3.53	0.0	
Provides: RNDr. Branislav Brutovský, CSc.						

Date of last modification: 20.09.2021

University: P. J. Ša	fárik Univers	ity in Košice			
Faculty: Faculty of	Science				
Course ID: ÚFV/ EMBF1/18	Course na	me: Experiment	al Methods of B	iophysics I	
Course type, scope Course type: Lect Recommended co Per week: 2 Per se Course method: p	ure urse-load (h tudy period: present	ours):			
Number of ECTS of					
Recommended sen	nester/trimes	ter of the course	e: 4.		
Course level: I.					
Prerequisities:					
Conditions for cou	rse completi	on:			
Learning outcomes	s:				
Brief outline of the	course:				
Recommended lite	rature:				
Course language:					
Notes:					
Course assessment Total number of ass		ts: 5			
А	В	С	D	E	FX
20.0	40.0	40.0	0.0	0.0	0.0
Provides: prof. RN	Dr. Pavol Mi	škovský, DrSc.			1
Date of last modified	cation: 30.03	.2022			
Approved: doc. Mg	gr. Daniel Jan	cura, PhD.			

University: P. J.	Šafárik Univers	ity in Košice			
Faculty: Faculty	of Science				
Course ID: ÚFV EMBF2/18	/ Course na	me: Experimenta	al Methods of B	Biophysics II	
Course type, sco Course type: Le Recommended Per week: 2 Per Course method	ecture course-load (h r study period:	ours):			
Number of ECT	S credits: 3				
Recommended s	emester/trimes	ster of the course	e: 4.		
Course level: I.					
Prerequisities:					
Conditions for co	ourse completi	on:			
Learning outcon	nes:				
Brief outline of t	he course:				
Recommended li	iterature:				
Course language					
Notes:					
Course assessme Total number of a		ts: 7			
A	В	С	D	Е	FX
42.86	14.29	42.86	0.0	0.0	0.0
Provides: doc. R Fabián, CSc., doc			Gabriela Fabric	iová, PhD., RND	r. Marián
Date of last mod	ification: 30.11	.2021			
Approved: doc. 1	Mgr. Daniel Jan	cura, PhD.			

University: P	J	Šafárik	University	in Košice
Chiver Stey . 1		Suluin	Oniversity	

Faculty: Faculty of Science

Course ID: ÚFV/	Course name: Experimental Methods of Biophysics III
EMBF3/18	

Course type, scope and the method: Course type: Lecture

Recommended course-load (hours):

Per week: 2 Per study period: 28

Course method: present

Number of ECTS credits: 3

Recommended semester/trimester of the course: 5.

Course level: I.

Prerequisities:

Conditions for course completion:

Oral exam where the students present theoretical knowledge of topics listed in the course syllabus.

Learning outcomes:

Students will gain basic knowledge on confocal fluorescence microscopy, time-resolved fluorescence microscopy (FLIM / PLIM), high resolution microscopy, flow cytometry. Students will also learn the methods of optical imaging used in clinical practice, the basics of Lab on chip technology and the principles of building experimental equipment. Students will get a theoretical basis, which they will be able to utilize for the experiments using the listed methods.

Brief outline of the course:

1. Confocal fluorescence microscopy:

Principles of confocal fluorescence microscopy, preparation of samples for vital staining and for immunofluorescence, detection of cell organelles, localization and distribution of proteins in cells, colocalization of dyes, analysis of fluorescence image.

2. Time-resolved fluorescence microscopy (FLIM):

Principle of time-resolved fluorescence microscopy, time and frequency domain FLIM detection, Förster resonance energy transfer (FRET), the influence of the environment on the detection of fluorescent probes, measurement of the level of oxidative stress in cells.

3. Time-resolved phosphorescence microscopy (PLIM):

Principles of time-resolved phosphorescence microscopy, the influence of the environment on the detection of phosphorescent probes, quenching of phosphorescence, detection of temperature changes and oxygenation in cells and tissues. Possibilities of sensor application for detection of oxygen and oxygen deficiency in vitro and in vivo.

4. High resolution microscopy:

Principle of selected microscopic approaches for detection of proteins and molecules in cells at high resolution: structured illumination microscopy (SIM), stimulated emission depletion (STED), photo-activated localization microscopy (PALM), stochastic optical reconstruction microscopy (STORM).

5. Flow cytometry:

Principles of flow cytometry, use of fluorescence in cell counting and subsequent analysis of specific factors, cell cycle, changes caused by oxidative stress, separation of cells based on vital staining and immunolabeling, detection of apoptosis and necrosis in the cell population.

6. Optical imaging techniques used in clinical practice:

Examples of bioimaging through: autofluorescence, second harmonic generation (SHG), fluorescence of clinically approved contrast molecules, optical coherence tomography (OCT). 7. Advanced imaging methods:

Electron microscopy, transmission electron microscope, scanning electron microscope. Atomic force microscopy (AFM).

8. Lab on chip technology:

Advantages of LOC technologies. Overview of microfluidic systems and their applications in molecular and cell biology. Instrumentation and preparation of microfluidic devices.

Development of experimental equipment:

9. Electrical signals:

Excitation and detection of electrical signals. Arbitrary function generator, digital oscilloscope and multimeter, laboratory card. Voltage sources. Weak signal measurements: lock-in amplifier. Shielding and grounding of devices. Basics of PID control.

10. Sources of optical radiation:

Classic sources of optical radiation: gas discharge lamps, fluorescent lamps, light bulbs. LEDs. Pulsed LED emitters. Lasers. Selection of lasers based on laser radiation parameters: wavelength, power, beam quality, polarization. Pulsed laser radiation sources.

11. Detection of optical radiation:

Basic terms, flux, irradiance, light intensity. Optical detectors, photodiodes, avalanche photodiodes, pin diodes, photomultipliers, thermal detectors. CCD cameras, CMOS cameras.

12. Building experimental apparatuses:

Basic principles of planning experimental equipment. Detection sensitivity, noise sources. Mechanical and thermal stability. Overview of optical and optomechanical components.

Recommended literature:

1. Wolfgang Becker: The bh TCSPC Handbook Seventh Edition, Becker & Hickl GmbH 2017;

- 2. Guy Cox: Optical Imaging Techniques in Cell Biology, Taylor & Frances;
- 3. Howard M. Shapiro: Practical Flow Cytometry Fourth edition, 2003;

4. Nikolas Long and Wing-Tak Wong: The chemistry of molecular imaging, Wiley 2014

Course language:

Slovak language, English.

Notes:

INOTES:						
Course assessm Total number of	nent f assessed studen	ts: 5				
А	В	С	D	Е	FX	
100.0	0.0	0.0	0.0	0.0	0.0	
Provides: doc. Mgr. Gregor Bánó, PhD., RNDr. Zuzana Naďová, PhD., RNDr. Veronika Huntošová, PhD.						
Date of last mo	dification: 22.09	0.2021				

University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚBEV/ ETB1/99	Course name: Experimental techniques in Biology
Course type, scope a Course type: Practic Recommended cour Per week: 4 Per stu Course method: pre	ce rse-load (hours): dy period: 56
Number of ECTS cr	edits: 4
Recommended seme	ster/trimester of the course: 4., 6.
Course level: I.	
Prerequisities: ÚBEV	V/CYT1/15
Conditions for cours active participation o	e completion: n practicals/seminars, exam
 Brief outline of the c Course manageme Molecular markers Molecular cytolog Work in aseptic con MTT test. Laboratory animal Manipulation with Animal dissection; Surgery in experim Fluorescence and c Fieldwork; Botan Use of scientific l 	nt. Laboratory safety. s; Polymerase chain reaction (PCR), RT-PCR, Real-Time PCR). y – Flow cytometry – principles and application in cell biology. nditions (in vitro); work with cell lines – subculturing, staining, cell counting, s, strains and inbreed lines; Breeding and manipulation with animals. laboratory animals; Behavioural tests. ; Anatomy of animals.
13. Conditional Gene Recommended litera Zutphen, L. F. M., Ba Elsevier, Amsterdam	aumans, V., Beynen, A. C.: Principles of Laboratory Animal Science.
Course language: English for Erasmus : Notes:	students

Notes:

Course assessment Total number of assessed students: 235								
A B C D E FX								
56.17	12.34	12.34	4.26	13.62	1.28			
Provides: RNDr. Ján Košuth, PhD., RNDr. Anna Alexovič Matiašová, PhD., RNDr. Terézia Kisková, PhD., Mgr. Vladislav Kolarčik, PhD., doc. RNDr. Juraj Ševc, PhD., doc. RNDr. Rastislav Jendželovský, PhD., RNDr. Natália Pipová, PhD., doc. RNDr. Monika Kassayová, CSc., RNDr. Jana Vargová, PhD.								
Date of last mo	dification: 15.10	0.2021		_				

Course type, scope and the method: Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 4 / 2 Per study period: 56 / 28 Course method: present Number of ECTS credits: 7 Recommended semester/trimester of the course: 1. Course level: I. Prerequisities: ÚCHV/CHV1/99 Conditions for course completion: Written test in the middle and the end of the semester followed by the oral examination. Active participation on seminars. Learning outcomes: To provide students with knowledge of atoms and molecules their electronic structure, theories of chemical bonds, physical and chemical properties of elements and compounds as well as their periodicity. Brief outline of the course: Main terms used in chemistry. Atoms – models of atoms, electron configuration, chemical periodicity and its effect on the properties of elements, radioactivity. Chemical bonds and intermolecular interactions. Chemical structure and physical properties of matter. State of matter. Solutions. Chemical equilibrium. Basis of chemical thermodynamics and chemical kinetics. Classification of chemical reactions. Electrochemistry. Recommended literature: 1. Atkins P., Jones L.: Chemical Principles, 2nd ed., Freeman, New York 2002. 2. Russel J.B.: General Chemistry, 2nd ed., McGraw Hill, London 1992. Course language:		CO	UKSE INFORM	AATION LET T	EK	
Course ID: ÚCHV/ VCHU/15 Course name: General Chemistry VCHU/15 Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 4 / 2 Per study period: 56 / 28 Course method: present State of the course in the second of the course in the second of the course in the second course in the second course in the second course in the middle and the end of the semester followed by the oral examination. Active participation on seminars. Course level: I. Prerequisities: ÚCHV/CHV1/99 Conditions for course completion: Written test in the middle and the end of the semester followed by the oral examination. Active participation on seminars. Learning outcomes: To provide students with knowledge of atoms and molecules their electronic structure, theories of chemical bonds, physical and chemical properties of elements and compounds as well as their periodicity. Stef outline of the course: Main terms used in chemistry. Atoms – models of atoms, electron configuration, chemical periodicity and its effect on the properties of elements, radioactivity. Chemical bonds and thermodynamics and chemical kinetics. Classification of chemical reactions. Electrochemistry. Recommended literature: 1. Atkins P., Jones L.: Chemical Principles, 2nd ed., Freeman, New York 2002. 2. Russel J.B.: General Chemistry, 2nd ed., McGraw Hill, London 1992. Course assessment Total number of assessed students: 310 A B C D E FX 23.87 29.03 28.39 11.61 7.1 0.0 Prov	University: P. J.	Šafárik Univers	ity in Košice			
VCHU/15 Course type: Lecture / Practice Recommended course-load (hours): Per week: 4 / 2 Per study period: 56 / 28 Course method: present Number of ECTS credits: 7 Recommended semester/trimester of the course: 1. Course level: I. Prerequisities: ÚCHV/CHV1/99 Conditions for course completion: Written test in the middle and the end of the semester followed by the oral examination. Active participation on seminars. Learning outcomes: To provide students with knowledge of atoms and molecules their electronic structure, theories of chemical bonds, physical and chemical properties of elements and compounds as well as their periodicity. Brief outline of the course: Main terms used in chemistry. Atoms – models of atoms, electron configuration, chemical periodicity and its effect on the properties of elements, radioactivity. Chemical bonds and intermolecular interactions. Chemical structure and physical properties of matter. State of matter. Solutions. Chemical Principles, 2nd ed., Freeman, New York 2002. 2. Russel J.B.: General Chemistry, 2nd ed., McGraw Hill, London 1992. Course assessment Total number of assessed students: 310 A B C D E FX 23.87 29.03 28.39 11.61 7.1 0.0 </td <td>Faculty: Faculty</td> <td>of Science</td> <td></td> <td></td> <td></td> <td></td>	Faculty: Faculty	of Science				
Course type: Lecture / Practice Recommended course-load (hours): Per week: 4 / 2 Per study period: 56 / 28 Course method: present Number of ECTS credits: 7 Recommended semester/trimester of the course: 1. Course level: I. Prerequisities: ÚCHV/CHV1/99 Conditions for course completion: Written test in the middle and the end of the semester followed by the oral examination. Active participation on seminars. Learning outcomes: To provide students with knowledge of atoms and molecules their electronic structure, theories of chemical bonds, physical and chemical properties of elements and compounds as well as their periodicity. Brief outline of the course: Main terms used in chemistry. Atoms – models of atoms, electron configuration, chemical periodicity and its effect on the properties of elements, radioactivity. Chemical bonds and intermolecular interactions. Chemical structure and physical properties of matter. State of matter. State of matter. State of matter. Solutions. Chemical reactions. Electrochemistry. Recommended literature: 1. Atkins P., Jones L.: Chemical Principles, 2nd ed., Freeman, New York 2002. 2. Russel J.B.: General Chemistry. 2nd ed., McGraw Hill, London 1992. Course assessment Total number of assessed students: 310 A B C D E <td< td=""><td>Course ID: ÚCH VCHU/15</td><td>V/ Course na</td><td>ame: General Ch</td><td>emistry</td><td></td><td></td></td<>	Course ID: ÚCH VCHU/15	V/ Course na	ame: General Ch	emistry		
Recommended semester/trimester of the course: 1. Course level: I. Prerequisities: ÚCHV/CHV1/99 Conditions for course completion: Written test in the middle and the end of the semester followed by the oral examination. Active participation on seminars. Learning outcomes: To provide students with knowledge of atoms and molecules their electronic structure, theories of chemical bonds, physical and chemical properties of elements and compounds as well as their periodicity. Brief outline of the course: Main terms used in chemistry. Atoms – models of atoms, electron configuration, chemical periodicity and its effect on the properties of elements, radioactivity. Chemical bonds and intermolecular interactions. Chemical structure and physical properties of matter. State of matter. Solutions. Chemical reactions. Electrochemistry. Recommended literature: 1. Atkins P, Jones L.: Chemical Principles, 2nd ed., Freeman, New York 2002. 2. Russel J.B.: General Chemistry, 2nd ed., McGraw Hill, London 1992. Course assessment Total number of assessed students: 310 A B C D E FX Qa.839 11.61 7.1 0.0 Provides: prof. RNDr. Vladimír Zeleňák, DrSc. <td>Course type: Le Recommended Per week: 4 / 2</td> <td>ecture / Practice course-load (h Per study peri</td> <td>ours):</td> <td></td> <td></td> <td></td>	Course type: Le Recommended Per week: 4 / 2	ecture / Practice course-load (h Per study peri	ours):			
Course level: I. Prerequisities: ÚCHV/CHV1/99 Conditions for course completion: Written test in the middle and the end of the semester followed by the oral examination. Active participation on seminars. Learning outcomes: To provide students with knowledge of atoms and molecules their electronic structure, theories of chemical bonds, physical and chemical properties of elements and compounds as well as their periodicity. Brief outline of the course: Main terms used in chemistry. Atoms – models of atoms, electron configuration, chemical periodicity and its effect on the properties of elements, radioactivity. Chemical bonds and intermolecular interactions. Chemical structure and physical properties of matter. State of matter. Solutions. Chemical equilibrium. Basis of chemical thermodynamics and chemical kinetics. Classification of chemical reactions. Electrochemistry. Recommended literature: 1. Atkins P., Jones L.: Chemical Principles, 2nd ed., Freeman, New York 2002. 2. Russel J.B.: General Chemistry, 2nd ed., McGraw Hill, London 1992. Course language: Notes: Course assessment Total number of assessed students: 310 A B C A B C A B C P A B C D A B C D	Number of ECT	S credits: 7				
Prerequisities: ÚCHV/CHV1/99 Conditions for course completion: Written test in the middle and the end of the semester followed by the oral examination. Active participation on seminars. Learning outcomes: To provide students with knowledge of atoms and molecules their electronic structure, theories of chemical bonds, physical and chemical properties of elements and compounds as well as their periodicity. Brief outline of the course: Main terms used in chemistry. Atoms – models of atoms, electron configuration, chemical periodicity and its effect on the properties of elements, radioactivity. Chemical bonds and intermolecular interactions. Chemical structure and physical properties of matter. State of matter. Solutions. Chemical equilibrium. Basis of chemical thermodynamics and chemical kinetics. Classification of chemical reactions. Electrochemistry. Recommended literature: 1. Atkins P., Jones L.: Chemical Principles, 2nd ed., Freeman, New York 2002. 2. Russel J.B.: General Chemistry, 2nd ed., McGraw Hill, London 1992. Course language: Notes: Course assessment Total number of assessed students: 310 A B C A B C A B C D Y 23.87 29.03 28.39 Y 23.87 29.03 28.39 Provides: prof. RN	Recommended s	emester/trimes	ster of the cours	e: 1.		
Conditions for course completion: Written test in the middle and the end of the semester followed by the oral examination. Active participation on seminars. Learning outcomes: To provide students with knowledge of atoms and molecules their electronic structure, theories of chemical bonds, physical and chemical properties of elements and compounds as well as their periodicity. Brief outline of the course: Main terms used in chemistry. Atoms – models of atoms, electron configuration, chemical periodicity and its effect on the properties of elements, radioactivity. Chemical bonds and intermolecular interactions. Chemical structure and physical properties of matter. State of matter. State of matter. Solutions. Chemical equilibrium. Basis of chemical thermodynamics and chemical kinetics. Classification of chemical reactions. Electrochemistry. Recommended literature: 1. Atkins P., Jones L.: Chemical Principles, 2nd ed., Freeman, New York 2002. 2. Russel J.B.: General Chemistry, 2nd ed., McGraw Hill, London 1992. Course language: Notes: Course assessment Total number of assessed students: 310 A B C D E A B C 23.87 29.03 28.39 11.61 7.1 0.0 Provides: prof. RNDr. Vladimír Zeleňák, DrSc. Date of last modification: 07.02.2022 <td>Course level: I.</td> <td>,</td> <td></td> <td></td> <td></td> <td></td>	Course level: I.	,				
Written test in the middle and the end of the semester followed by the oral examination. Active participation on seminars. Learning outcomes: To provide students with knowledge of atoms and molecules their electronic structure, theories of chemical bonds, physical and chemical properties of elements and compounds as well as their periodicity. Brief outline of the course: Main terms used in chemistry. Atoms – models of atoms, electron configuration, chemical periodicity and its effect on the properties of elements, radioactivity. Chemical bonds and intermolecular interactions. Chemical structure and physical properties of matter. State of matter. Solutions. Chemical equilibrium. Basis of chemical thermodynamics and chemical kinetics. Classification of chemical reactions. Electrochemistry. Recommended literature: 1. Atkins P., Jones L.: Chemical Principles, 2nd ed., Freeman, New York 2002. 2. Russel J.B.: General Chemistry, 2nd ed., McGraw Hill, London 1992. Course language: Notes: C D E FX 23.87 29.03 28.39 11.61 7.1 0.0 Provides: prof. RNDr. Vladimir Zeleňák, DrSc. D E FX Date of last modification: 07.02.2022 D E FX	Prerequisities: Ú	CHV/CHV1/99)			
To provide students with knowledge of atoms and molecules their electronic structure, theories of chemical bonds, physical and chemical properties of elements and compounds as well as their periodicity. Brief outline of the course: Main terms used in chemistry. Atoms – models of atoms, electron configuration, chemical periodicity and its effect on the properties of elements, radioactivity. Chemical bonds and intermolecular interactions. Chemical structure and physical properties of matter. State of matter. Solutions. Chemical equilibrium. Basis of chemical thermodynamics and chemical kinetics. Classification of chemical reactions. Electrochemistry. Recommended literature: 1. Atkins P., Jones L.: Chemical Principles, 2nd ed., Freeman, New York 2002. 2. Russel J.B.: General Chemistry, 2nd ed., McGraw Hill, London 1992. Course language: Notes: Course assessment Total number of assessed students: 310 A B C D E FX 23.87 29.03 28.39 11.61 7.1 0.0 Provides: prof. RNDr. Vladimír Zeleňák, DrSc. Date of last modification: 07.02.2022	Written test in th	e middle and t		mester followed	by the oral exam	ination. Active
Main terms used in chemistry. Atoms – models of atoms, electron configuration, chemical periodicity and its effect on the properties of elements, radioactivity. Chemical bonds and intermolecular interactions. Chemical structure and physical properties of matter. State of matter. Solutions. Chemical equilibrium. Basis of chemical thermodynamics and chemical kinetics. Classification of chemical reactions. Electrochemistry. Recommended literature: 1. Atkins P., Jones L.: Chemical Principles, 2nd ed., Freeman, New York 2002. 2. Russel J.B.: General Chemistry, 2nd ed., McGraw Hill, London 1992. Course language: Notes: Course assessment Total number of assessed students: 310 A B C D E FX 23.87 29.03 28.39 11.61 7.1 0.0 Provides: prof. RNDr. Vladimír Zeleňák, DrSc. Date of last modification: 07.02.2022	To provide stude	ents with know	•			,
1. Atkins P., Jones L.: Chemical Principles, 2nd ed., Freeman, New York 2002. 2. Russel J.B.: General Chemistry, 2nd ed., McGraw Hill, London 1992. Course language: Notes: Course assessment Total number of assessed students: 310 A B C D E FX 23.87 29.03 28.39 11.61 7.1 0.0 Provides: prof. RNDr. Vladimír Zeleňák, DrSc. Date of last modification: 07.02.2022	Main terms used periodicity and intermolecular in Solutions. Chem	d in chemistry its effect on t iteractions. Che nical equilibrium	he properties of mical structure a m. Basis of che	f elements, radi nd physical prop emical thermody	oactivity. Chemi perties of matter.	ical bonds and State of matter.
Notes:Course assessment Total number of assessed students: 310ABCDEFX23.8729.0328.3911.617.10.0Provides: prof. RNDr. Vladimír Zeleňák, DrSc.Date of last modification: 07.02.2022	1. Atkins P., Jone	es L.: Chemical				
Course assessmentTotal number of assessed students: 310ABCDEFX23.8729.0328.3911.617.10.0Provides: prof. RNDr. Vladimír Zeleňák, DrSc.Date of last modification: 07.02.2022	Course language					
A B C D E FX 23.87 29.03 28.39 11.61 7.1 0.0 Provides: prof. RNDr. Vladimír Zeleňák, DrSc. Date of last modification: 07.02.2022 Date of last modification: 07.02.2022	Notes:					
23.87 29.03 28.39 11.61 7.1 0.0 Provides: prof. RNDr. Vladimír Zeleňák, DrSc. Date of last modification: 07.02.2022		-	ts: 310			
Provides: prof. RNDr. Vladimír Zeleňák, DrSc. Date of last modification: 07.02.2022	A	В	С	D	Е	FX
Date of last modification: 07.02.2022	23.87	29.03	28.39	11.61	7.1	0.0
	Provides: prof. R	NDr. Vladimír	Zeleňák, DrSc.		<u> </u>	
Approved: doc. Mgr. Daniel Jancura. PhD.	Date of last mod	ification: 07.02	2.2022			
11 ···································	Approved: doc. N	Mgr. Daniel Jan	cura, PhD.			

University: P. J. Šafá	rik University in Košice			
Faculty: Faculty of S	cience			
Course ID: ÚFV/ VF1a/12				
Course type, scope a Course type: Lectur Recommended cour Per week: 4 / 2 Per Course method: pre	re / Practice rse-load (hours): study period: 56 / 28			
Number of ECTS cr	edits: 7			
Recommended seme	ster/trimester of the course: 1.			
Course level: I.				
Prerequisities:				
 -participation in class -active participation a -submitting all the ass -tests during the seme -project group work a Final assessment: -final oral examination Conditions for success -participation in lesson 	s of assessment during the semester ses in accordance with study regulations and teacher's instructions at seminars and exercises signments in accordance with teacher's instruction ester and its successful presentation and defence			
physics and thermody	urse student masters basic knowledge connected with mechanics, molecular ynamics. Student will be able to solve various problems connected with the oply gained knowledge in different situations.			
 Mechanics of parti Gravitational field. Work, power and e Mechanics of syste Mechanics of rigid Mechanics of elast Mechanics of fluid 	of the calculus, vector algebra. Standards and units. cle. emergy. em of particles. l body. ic body. is. ur physics. Structure and properties of gases. dynamics. ermal expansion.			

Recommended literature:

CUMMINGS, Karen, LAWS, Priscilla, REDISH, Edward, COONEY, Patrick: Understanding Physics, John Wiley & Sons, 2004

Course language:

English

Notes:

Course assessment

Total number of assessed students: 324

А	В	С	D	Е	FX
23.77	16.36	21.3	14.51	16.05	8.02
Provides: doc.]	Provides: doc. RNDr. Zuzana Ješková, PhD.				

Date of last modification: 15.09.2021

University: P. J. Šafa Faculty: Faculty of S	Science	
Course ID: ÚFV/	Course name: General Physics II	
VF1b/03		
Course type, scope	and the method:	
Course type: Lectu		
Recommended cou		
	r study period: 56 / 28	
Course method: pr	resent	
Number of ECTS c	redits: 7	
Recommended sem	ester/trimester of the course: 2.	
Course level: I.		
Prerequisities: ÚFV	//VF1a/12	
Conditions for cour	1	
To successfully com	plete the course (presence, if necessary distance), the student must demonstrate	
sufficient understand	ding of the basic concepts and laws of electromagnetism, so that it is possibl	
to continue the study	y of general physics III, IV and the discipline of electromagnetic field theory.	
-	ridual laws of electricity and magnetism and their generalization in the form o	
-	s is required. Knowledge of these laws in nature and in practical use is required	
-	it is adequate skills in solving the problems of electricity and magnetism.	
-	kes into account the scope of teaching (4 hours of lectures, 2 hours of numerica	
	, self-study (1 credit), evaluation (2 credits) and the fact that it is a basic subject	
	chelor's state exam. The minimum limit for successful completion of the cours	
-	ts from the subsequent point evaluation, while it is necessary to obtain at least	
-		
50% of points from		
	s maximum number of 20 points (usually 2 written tests of 10 points each, th	
	at least 5 points from each test)	
level of at least 50%	naximum of 80 points (answer to three questions, each of which must reach	
Rating scale	·).	
A 100-91		
B 90-81		
C 80-71		
D 70-61		
E 60-50		
Fx 49-0		
Learning outcomes:		
U		
After completing lea	ctures and exercises, the student will have sufficient knowledge of the basic	
After completing led of electricity and ma	ctures and exercises, the student will have sufficient knowledge of the basic agnetism and will be able to solve numerical problems of electromagnetism lequate knowledge about electromagnetic phenomena in nature and the use o	

electromagnetic phenomena in technical applications.

Brief outline of the course:

1. Week: Electrostatic field in vacuum. Culomb's law. Electric field. Electric dipole. Flux of electric field. Gauss' law.

2. Week: Work of forces in the electrostatic field. Potential. Relationship between electric fiel and electric potential. Potential and its measurement. Capacity of conductor and conductor system. Energy of electrostatic field.

3. Week: Stationary electric field and steady electric current. Ohm's law. Superconductivity. Equation of continuity of electric current. Electrical circuits with steady voltage. Kirchhoff's laws and their application. Work, power, energy and efficiency of the source of electromotive voltage.

4. Week: Electric current in electrolytes, semiconductors, gases and in vacuum. Thermoelectric phenomena and their use.

5. Week: Origin, properties and basic quantities of a stationary magnetic field in vacuum. Biot-Savart law and its application. Magnetic flux density.

6. Week: Interactions of a magnetic field with moving electrically charged particles and with electric currents. Ampere's law. Interaction between current conductors. Definition of ampere as current unit. Lorentz force.

7. Week: Quasi-stationary electric field. Capacitor charging and discharging process (R-C circuit). The phenomenon of electromagnetic induction. Faraday's law. Phenomenon of self-induction and mutual inductance, mutual inductance. Potential of magnetic field.

8. Week: Transient in the R-L circuit. Energy of magnetic field. Energy conservation law. Magnetic dipole. Alternating currents and basic circuits of alternating electric current. RLC circuit

9. Week: Serial and parallel resonance. Multiphase currents. Rotating magnetic field. Formation of multiphase currents. Electric motor. Power of alternating electric current.

10. Week: Electrical phenomena in the material environment. Dielectric polarization, mechanisms. Electric field in dielectric. Interaction of electric charges stored in a dielectric. Gauss' law. Polarization vector and electrical induction vector and their mutual relationship. Linear and nonlinear dielectrics.

11. Week: Magnetic properties of substances. Elementary magnetic field of an atom. Magnetic state of substances. Magnetic polarization. Diamagnetism and paramagnetism. Arranged magnetic structure. Ferromagnets.

12. Week: Unsteady electromagnetic field. Maxwell's equations.

Recommended literature:

T. Matsushita: Electricity and Magnetism, Springer, 2017

Course language:

english

Notes:

Presence form represents a standart form for the course, if a need arises, the course is performed using MS Teams.

Course assessment

Total number of assessed students: 365

А	В	С	D	Е	FX
34.79	15.07	15.62	11.51	9.86	13.15

Provides: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Adriana Zeleňáková, PhD., doc. RNDr. Erik Čižmár, PhD.

Date of last modification: 10.02.2023

University: P. J. S	Šafárik Univers	ity in Košice			
Faculty: Faculty	of Science				
Course ID: ÚFV/ VF1c/12	V/ Course name: General Physics III				
Course type, sco Course type: Le Recommended Per week: 4 / 2 Course method	ecture / Practice course-load (h Per study perio	ours):			
Number of ECTS	S credits: 7				
Recommended se	emester/trimes	ster of the cours	e: 3.		
Course level: I.					
Prerequisities: Ú	FV/VF1b/03 or	r ÚFV/VFM1b/1	5		
Conditions for co Written test (2x) Oral examination	from seminars		ter.		
Learning outcom The objective is t		students with the	basis of oscilat	ions, waves and op	ptics.
Fourier transform Huyghens princip Geometrical optic Light as electron	ations, Mathem nation, Forced o ple. Reflection, cs. Mirrors, len magnetic wave	oscilations. Wave difraction. Dop s. Fotometry. e. Dispersion, al	es, their generat pler effect. Wav bsorption, inter	pendulum, Damp ion, waves equation res speed in mater ference, difraction s law of radiation.	on.Interference. ials. Acoustics. n, polarization.
Recommended li 1. A. Hlavička et 2. R.P. Feynman 3. D. Halliday et 4. J. Fuka, B. Hay 5. A. Štrba, Všeo	al., Fyzika pro et al., Feynman al.,Fyzika-Vyso velka, Optika a	ove prednášky z okoškolská učebi atómová fyzika,	Fyziky I,II,III, nice obecné fyzi SPN,1961		10
Course language 1. slovak 2. english	:				
Notes:					
Course assessme Total number of a		ts: 147			
A	В	С	D	Е	FX
31.29	26.53	25.85	11.56	4.76	0.0
Provides: doc. R	NDr. Ján Füzer,	PhD.			

Date of last modification: 10.09.2021

University: P. J. Šafárik University	in Košice
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Faculty: Faculty of Science

Course ID: ÚFV/	Course name: General Physics IV
VF1d/12	

Course type, scope and the method:

Course type: Lecture / Practice

Recommended course-load (hours): Per week: 4 / 2 **Per study period:** 56 / 28

Course method: present

Number of ECTS credits: 7

Recommended semester/trimester of the course: 6.

Course level: I.

Prerequisities: ÚFV/VF1c/10 or ÚFV/VF1c/12 or ÚFV/VF1c/22

Conditions for course completion:

- active participation in lectures and excersises

- submission of solved tasks

- 2x test

- an exam

Credit evaluation of the subject: direct teaching and consultations (2credits), self-study

(1credit), practical activities- solved tasks (2redits), evaluation (2credits), a total of 7credits. Minimum limit for completion of the course is to obtain at least 51% of the total evaluation.

Learning outcomes:

Basic knowledge about the atomic structure and spectra and nuclei, and elementary particles. Basic experimental methods in nuclear physics and passage of nuclear radiation through media.

Brief outline of the course:

1.-6. week Atomic Physics - A.Kravčáková (P):

Corpuscular-wave dualism: De Broglie waves. Experimental confirmation of de Broglie's hypothesis. Uncertainty principle.

Atom structure: Atomic hypothesis. Rutherford's experiment. Bohr model of the atom.

Hydrogen radiation spectra. Combination principle. Quantum mechanical description of a hydrogen atom.

Electron shell: Spectra of hydrogen type atoms. Experimental verification of the existence of discrete levels of atoms (Franck-Hertz experiment). Angulat momentum of electron motion. Stern-Gerlach experiment. Quantum states of electrons. Atoms with more electrons. Alkali metal spectra. Total angular momentum of an atom. Magnetic momentum of an atom. An atom in an external magnetic and electric field. Zeeman's phenomenon. Selection rules. Pauli's principle. Periodic table of elements. X-ray spectra.

Molecules: Ion and covalent coupling, spectra of molecules.

7.-12. week Nuclear Physics - J.Vrláková (P):

Basic characteristics of atomic nuclei: Mass and electric charge. Radius of the atomic nucleus. Binding energy. Spin and magnetic momentum of the nucleus. Quadrupole momentum. Parity.

Nuclear forces and models of atomic nuclei: Properties of nuclear forces. Meson theory of nuclear forces. Models of atomic nuclei (droplet, layer and generalized model).

Radioactive radiation: Basic laws of radioactive decay. Law of decay. Alpha decay. Beta decay. Processes taking place in the nucleus during beta conversion. Neutrino existence hypothesis. Fermi's theory. Internal conversion. Gamma radiation.

Nuclear reactions: Basic terms and definitions. Classification of nuclear reactions. Conservation laws. Effective cross section. Mechanisms of nuclear reactions. Basic types of reactions. Breit-Wigner formula. Reactions with neutrons. Fission of atomic nuclei. Mechanism of fission. Nuclear reactor. Thermonuclear reactions.

Week 13 Subnuclear physics - A.Kravčáková (P):

Elementary particles: The concept of an elementary particle. Basic characteristics of particles. Conservation laws. Types of interactions. Antiparticles. Classification of elementary particles. Strange particles. Resonances. Quark model of hadrons.

Cosmic radiation: Primary and secondary components. Elementary particles and cosmology.

Week 14 Experimental methods - A.Kravčáková (P):

Passage of radiation through matter: The passage of heavy charged particles, electrons and gamma radiation through the matter.

Detectors: Basic characteristics of detectors. Volt-ampere characteristic. Gas detectors. Ionization chambers and Geiger-Müller computer. Scintillation, Cherenkov and semiconductor detectors. Track detectors.

Particle accelerators: Linear accelerator. Cyclic accelerators. Colliders.

Recommended literature:

1. Beiser A., Úvod do moderní fyziky, Praha, 1975.

2. Úlehla I., Suk M., Trka Z.: Atómy, jádra, částice, Praha, 1990.

3. Síleš E., Martinská G.: Všeobecná fyzika IV, skriptá PF UPJŠ, 2. vydanie, Košice, 1992.

4. Vrláková J., Kravčáková A., Vokál S.: Zbierka príkladov z atómovej a jadrovej fyziky, skriptá PF UPJŠ, Košice, 2016.

5. Hajko V. and team of authors, Physics in experiments, Bratislava, 1997.

6. Nosek D., Jádra a částice (Řešené příklady), Matfyzpress, MFF UK, Praha 2005,

7. Kravčáková A., Vokál S., Vrláková J., Všeobecná fyzika IV, 1.časť Atómová fyzika, skriptá PF UPJŠ, Košice, 2020.

8. Yang F., Hamilton J.H., Modern Atomic and Nuclear Physics, WSC Singapore, 2010.

Course language:

slovak and english

Notes:

Course assessment

Total number of assessed students: 108

А	В	С	D	Е	FX
39.81	29.63	12.04	9.26	9.26	0.0

Provides: doc. RNDr. Adela Kravčáková, PhD., doc. RNDr. Janka Vrláková, PhD., RNDr. Zuzana Paulínyová, PhD.

Date of last modification: 16.09.2021

University: P. J.	Šafárik Univers	ity in Košice			
Faculty: Faculty	of Science				
Course ID: ÚFV UPF1/12	Course name: Introduction to Computational Physics				
Course type, sco Course type: La Recommended Per week: 2 / 1 Course method	ecture / Practice course-load (h Per study perio	ours):			
Number of ECT	S credits: 4				
Recommended s	emester/trimes	ster of the cour	se: 1.		
Course level: I.					
Prerequisities:					
Conditions for c Elaboration of m Exam and discus	icroreferat on g	iven topics.	he given project.		
processes in cor	ecture is to pro-	puters, as well	as to provide l	background of the ess conventional physical processes	possibilities to
point of view. Ph . Computer mode	es utilised in con ysical limits of eling and physic mative methods	current compute cal reality. Com	er technologies (l putational compl	ional processes / t Moore, Amdahl la exity and paraleli cal processors, D	sm. Distributed
Recommended la Actual literature		turer.			
Course language	2:				
Notes:					
Course assessme Total number of	-	ts: 45			
A	В	С	D	Е	FX
88.89	8.89	0.0	0.0	2.22	0.0
Provides: doc. R	NDr. Jozef Ulič	ný, CSc.			
Date of last mod	ification: 22.09	0.2021			

Faculty: Faculty of S	cience
Course ID: ÚFV/ UVF/05	Course name: Introduction to General Physics
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	ce rse-load (hours): Idy period: 28
Number of ECTS cr	edits: 2
Recommended seme	ster/trimester of the course: 1.
Course level: I.	
Prerequisities:	
-active participation a -submitting all the as -tests during the seme Final assessment: -based on assessment Conditions for succes -participation in lesso	
Learning outcomes: By the end of the comphysics and thermod	urse student is able to solve problems connected with mechanics, molecula ynamics. In solving problems student is able to apply digital tools for dat surement and computer modelling and data processing and their analysis.
 and Thermodynamic connected with the for 1. Kinematics and d Equation of motion. 2. Gravitational field. 3. Work, power and e 4. Rotational motion. 5. Law of momentum 6. Deformation. Hool 7. Fluid mechanics. 8. Gases. Ideal gas la 	liary subject to the course General physics 1 - Mechanics, Molecular Physic s aimed to development of conceptual understanding and problem solvin ollowing areas: lynamics of motion along a line and two-dimensional motion of particle . Projectile motion. energy. Law of energy conservation. . Equation of rotational motion. n conservation and angular momentum conservation. k's law.

11. Liquids. Surface tension.

12. Changes of state.

Recommended literature:

CUMMINGS, Karen, LAWS, Priscilla, REDISH, Edward, COONEY, Patrick: Understanding Physics, John Wiley & Sons, 2004

Course language:

English

Notes:

Course assessment

А	В	С	D	Е	FX
37.31	20.49	24.16	12.84	4.89	0.31
Provides: doc.	RNDr. Zuzana Je	šková, PhD.			

Date of last modification: 15.09.2021

Faculty: Faculty of Seculty of Seculty of Seculty Security Seculty Seculty Security Securit	cience
Course ID: ÚFV/ UVF2/07	Course name: Introduction to General Physics II
Course type, scope an Course type: Practic Recommended cour Per week: 2 Per stue Course method: pre	ce rse-load (hours): dy period: 28
Number of ECTS cro	edits: 2
Recommended seme	ster/trimester of the course: 2.
Course level: I.	
Prerequisities:	
-participation in class -active participation a -submitting all the ass -tests during the seme -based on assessment Conditions for succes -participation in lesso	a of assessment during the semester bes in accordance with study regulations and teacher's instructions at seminars and exercises signments in accordance with teacher's instruction ester Final assessment: during the semester asful completion of the course: ons in accordance with the study regulations and teacher's instructions higher than 50 % in assessment during the semester and in final assessment
-	rse student is able to solve problems and explain phemomena and experiments and areas of Electricity and Magnetism.
to development of co areas: 1. Electric field. Coul 2. Work, electric pote 3. Electric capacitanc 4. Electric current. Ol 5. Work and power. E 6. Magnetic field.	iary subject to the course General physics 2 - Electricity and Magnetism aimed onceptual understanding and problem solving connected with the following lomb's law. ential energy, electric potential. e and capacitors. hm's law, Kirchhoff's laws. Energy and efficiency of sources of electromotive force n magnetic field and electric charge. ena in RC circuit.

CUMMINGS,	uo. Electricity ar Karen, LAWS, Pr Viley & Sons, 20	riscilla, REDISH		IEY, Patrick: Un	derstanding	
Course langua English	ge:					
Notes:						
Course assessm Total number o	nent f assessed studen	ts: 270				
А	В	С	D	Е	FX	
<u>39.26</u> 22.59 20.74 8.15 9.26 0.0						
Provides: doc.	RNDr. Zuzana Je	šková, PhD.				

Date of last modification: 15.09.2021

University: P. J. Š	afárik Univers	ity in Košice			
Faculty: Faculty o	f Science				
Course ID: ÚFV/ ZMF/17	Course na	me: Introduction	n to Mathematics	for Physicists	
Course type, scop Course type: Lec Recommended c Per week: 1 / 2 P Course method:	cture / Practice ourse-load (h er study perio	ours):			
Number of ECTS	credits: 3				
Recommended se	mester/trimes	ster of the cours	e: 1.		
Course level: I.					
Prerequisities:					
Conditions for co	urse completi	on:			
Learning outcom	es:				
Brief outline of th	e course:				
Recommended lit	erature:				
Course language:					
Notes:					
Course assessmen Total number of as		ts: 287			
A	В	С	D	Е	FX
40.77	21.25	18.47	10.45	8.71	0.35
Provides: RNDr. 7	Tomáš Lučivja	nský, PhD., doc.	RNDr. Jozef Ha	nč, PhD.	
Date of last modif	fication: 16.11	.2021			
Approved: doc. M	Igr. Daniel Jan	cura, PhD.			

University: P. J. Šafárik University in Košice							
Faculty: Faculty of Science							
Course ID: Dek. PF Course name: Introduction to Study of Sciences UPJŠ/USPV/13							
Course type: Lectur Recommended cour Per week: Per stud	Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: Per study period: 12s / 3d Course method: present						
Number of ECTS cr							
Recommended seme	ster/trimester of the cours	e: 1.					
Course level: I.							
Prerequisities:	Prerequisities:						
Conditions for cours	e completion:						
Learning outcomes:							
Brief outline of the c	ourse:						
Recommended litera	iture:						
Course language:							
Notes:							
Course assessment Total number of asses	Course assessment Total number of assessed students: 2012						
	abs n						
88.37 11.63							
Provides: doc. RNDr	Provides: doc. RNDr. Marián Kireš, PhD.						
Date of last modification: 30.08.2022							
Approved: doc. Mgr.	Daniel Jancura, PhD.						

Faculty: Faculty of S	rik University in Košice
	cience
Course ID: ÚMV/ UAD/10	Course name: Introduction to data analysis
Course type, scope a Course type: Lectur Recommended cour Per week: 1 / 1 Per Course method: pre	re / Practice rse-load (hours): study period: 14 / 14
Number of ECTS cr	edits: 2
Recommended seme	ster/trimester of the course: 1.
Course level: I.	
Prerequisities:	
Oral presentation of t At least 50% must be	dual project work (20p). the individual project work (5p). e obtained from each part. $0\% A; \ge 80\% B; \ge 70\% C; \ge 60\% D; \ge 50\% E; < 50\% FX.$
understand its import To understand element	burpose of statistical data analysis, its methods and statistical thinking and stance for science and practical life. Intary statistical concepts. In handling real data using spreadsheet Excel and statistical software R.
Brief outline of the c	0 01
 Introduction (the b statistics) Collecting Data (ty Handling Data (v skewness and kurtosi Relationships in data 	
 Introduction (the b statistics) Collecting Data (ty Handling Data (v skewness and kurtosi Relationships in da Statistical inference Recommended literat Anděl, J.: Statistical Rossman, A.J. et at 2009 Utts, J.M.: Seeing Utts, J.M., Heckard 	course: basic philosophy and aim of statistical data analysis, descriptive and inductive wypes of data, random sample, randomized experiment) wisualization, summarizing – measures of center, measures of variability, is, empirical rule) - 5 weeks ata (introduction to regression and correlation) - 4 weeks the (elementary view into estimation and testing hypothesis) - 2 weeks
 Introduction (the b statistics) Collecting Data (ty 3. Handling Data (v skewness and kurtosi 4. Relationships in da 5. Statistical inference Recommended litera 1. Anděl, J.: Statistical 2. Rossman, A.J. et a 2009 Utts, J.M.: Seeing 4. Utts, J.M., Heckard 5. Zvára, K., Štěpán, 	course: asic philosophy and aim of statistical data analysis, descriptive and inductive wpes of data, random sample, randomized experiment) visualization, summarizing – measures of center, measures of variability, is, empirical rule) - 5 weeks ata (introduction to regression and correlation) - 4 weeks te (elementary view into estimation and testing hypothesis) - 2 weeks the (elementary view into estimation and testing hypothesis) - 2 weeks the true: ké metody, Matfyzpress, Praha, 1998 (in Czech) 1.: Workshop Statistics: Discovery with Data and Fathom, 3rd ed. Wiley, Through Statistics, 4th ed., Thomson Brooks/Cole, Belmont, 2014 d R.F.: Mind on Statistics, 6th ed. Thomson Brooks/Cole, Belmont, 2021

Course assessment Total number of assessed students: 390							
A B C D E FX							
37.44	37.44 25.13 26.41 10.0 0.51 0.51						
Provides: doc. 1	RNDr. Martina H	ančová, PhD.					
Date of last modification: 13.09.2021							
Approved: doc.	Mgr. Daniel Jan	cura, PhD.					

University: P. J. Šafán	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚFV/ LTV/18	Course name: Laboratory techniques
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	ce rse-load (hours): dy period: 28
Number of ECTS cr	edits: 2
Recommended seme	ster/trimester of the course: 2.
Course level: I.	
Prerequisities:	
	n exam during the semester - approximately in the 5th week of the semester oretical-computational part of the course)
	e student will get knowledge and first experiences of safe and efficient work cal, optical spectroscopy) laboratory.
Brief outline of the c	ourse:
the fundamentals of	d requirements for successful completion of the course. Introduction to laboratory work and safety, chemical and general safety. Introduction and on of the laboratories at the Department of Biophysics and Center for sciences.
Composition of subst molecular weights, p formula, mass and m solution.	ances and solutions: basic characteristics of solutions. Chemical formula and ercentage composition from formulas, from empirical formula to molecular lass fraction, molar weight, molar volume, molarity, the concentration of a
concentration, the con	ons: solubility of substances, solution and its concentration, mass/volume neentration of a solution in %, molar concentration, mole-mass relationships , concentration units – ppm, ppb.
	ns: diluting and mixing solutions.
	atory safety rules and guidelines.
laboratory dryer, Mil	of small laboratory equipment/instruments: automatic pipettes, centrifuge, li-Q ultrapure water system. Laboratory digester. Care and safe laboratory e – handling, cleaning and storing.

Week 7

Preparation of solutions: Analytical balances and proper weighing practices; working safely with solvents, the storage and disposal of chemicals, solvents, stock solutions and chemical waste. What is Parafilm?

Week 8

Preparing buffer solutions. pH determination, acidity and alkalinity. Working principle of pH meter - calibration and working demonstration. Working with acids and bases.

Week 9

Introduction to spectroscopy. The light. Spectroscopic experiment. Spectroscopic techniques. Jablonski diagram. UV-Vis absorption spectrophotometry. Chromophore. Lambert-Beer's law. Absorption spectrum. Absorption spectrophotometer. Fluorescence spectroscopy. Fluorophore. Excitation and emission spectra. Characteristics of fluorescence spectra. Fluorescence quenching. Week 10

Introduction to spectrophotometry: working with spectroscopic equipment - preparation of solutions of selected molecules at different pH and measurement of their UV-Vis absorption spectra. Week 11

Introduction to spectrophotometry: working with spectroscopic equipment - measurement of fluorescence spectra of the selected molecules at different pH.

Week 12/13

Data collection, processing and analysis. Preparing a Practical/Scientific laboratory report. Evaluation of Laboratory reports.

Keeping the laboratory environment clean and safe.

Recommended literature:

Course language:

0 0	e:						
Notes:							
Course assessme Total number of		ts: 10					
А	B C D E FX						
80.0	20.0	0.0	0.0	0.0	0.0		
Provides: RNDr.	Provides: RNDr. Zuzana Jurašeková, PhD.						
Date of last modification: 21.09.2021							
Approved: doc.	Mgr. Daniel Jan	cura, PhD.					

University: P. J. Ša	fárik Univers	ity in Košice			
Faculty: Faculty of	Science				
Course ID: ÚFV/ MFY/12	Course na	me: Mathematic	al Physics		
Course type, scope Course type: Lect Recommended co Per week: 3 / 1 Pe Course method: p	ure / Practice urse-load (h er study perio	ours):			
Number of ECTS	credits: 6				
Recommended sen	nester/trimes	ster of the course	e: 4.		
Course level: I.					
Prerequisities: ÚM	V/FRPb/19				
Conditions for cou	rse completi	on:			
Learning outcome	s:				
Brief outline of the	e course:				
Recommended lite	rature:				
Course language:					
Notes:					
Course assessment Total number of ass		ts: 83			
A	В	С	D	Е	FX
25.3	16.87	13.25	10.84	30.12	3.61
Provides: RNDr. To	omáš Lučivja	nský, PhD.		·4	
Date of last modifi	cation: 16.11	.2021			
Approved: doc. Mg	gr. Daniel Jan	cura, PhD.			

	COURSE INFORMATION LETTER
University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚMV/ MTFa/15	Course name: Mathematics I for physicists
Course type, scope a Course type: Lectur Recommended cour Per week: 2 / 2 Per Course method: pre	re / Practice rse-load (hours): study period: 28 / 28
Number of ECTS cr	edits: 5
Recommended seme	ster/trimester of the course: 1.
Course level: I.	
Prerequisities:	
terms and the ability is according to the re During the semester, a (together 50 points). may write the exam. number of 30 points. 59-50-D, 49-40-E. If	The completion: It is necessary to demonstrate the acquirement of basic mathematical to solve problems from selected thematic units. The evaluation of the subject esults from the semester and in view of the results of the written final test. students write tests at all seminars (together 20 points) and two extensive tests It is necessary to obtain at least 28 points during the semester. Then students To pass the exam, it is necessary to obtain at least 12 points from the maximum The scale for student evaluation is as follows: 100-80-A, 79-70-B, 69-60-C, a student does not achieve the required minimal number of points from the and during the semester (together 28 points), he/she is evaluated by FX.
equations and inequ	e course, the student can use basic mathematical terms, can solve various nations, and is acquainted with basic mathematical knowledge from the ral calculus, and is able to apply the theory in concrete excercises.
functions. Composition Week 7-14: Limit of	of function. Domain and range of functions. Elementary functions. Inverse
D. Studenovská, T. N odbory, UPJŠ 2006 D. Studenovská, T. N	nture: covič: Matematika, Alfa, Bratislava 1991 Iadaras, S. Mockovčiak: Zbierka úloh z matematiky pre nematematické Iadaras: Matematika pre nematematické odbory, UPJŠ 2006 rse in Calculus, Springer Verlag, 1998
Course language: Slovak	
Notes:	

Course assessment Total number of assessed students: 54								
A B C D E FX								
25.93	12.96	29.63	16.67	11.11	3.7			
Provides: RND	r. Jana Borzová,	PhD., RNDr. Baı	rbora Klemová, F	RNDr. Diana Trel	lová			
Date of last modification: 18.04.2022								
Approved: doc.	Approved: doc. Mgr. Daniel Jancura, PhD.							

University: P. J. Šafá						
University: P. J. Šafárik University in Košice						
Faculty: Faculty of Science						
Course ID: ÚMV/ MTFb/15	Course name: Mathematics II for physicists					
Course method: pre	re / Practice rse-load (hours): study period: 28 / 28 esent					
Number of ECTS credits: 4						
Recommended semester/trimester of the course: 2.						
Course level: I.						
Prerequisities: ÚMV/MTFa/15						
of a function, different skills associated with Mastering standard p	al variables, mastering the definitions of limit of function, partial derivation ential of a function, local and global extrema of a function and acquiring the their use in calculations focused mainly on functions of two variables. procedures for solving basic types of ordinary differential equations of the 1st g the concept of infinite series and acquiring skills to use the basic criteria of					

E: 42 p. - 48 p.

Learning outcomes:

The student should be able to explain the basic concepts and gain skills in using standard procedures for solving systems of linear equations using matrices and determinants. The student will expand his knowledge of the function of one variable and master the concept of a function of several variables, and will be able to explain the definitions of function limit, partial derivation of a function, differential of a function, local and global extrema of a function and acquire knowledge and skills oriented mainly on the functions of two variables. The student will learn standard procedures for solving basic types of ordinary differential equations of the 1st order. He will be able to use the

acquired knowledge about solving differential equations in modeling and solving problems derived from real situations. The student will gain skills to use the basic criteria of convergence of number series when deciding on the convergence or divergence of number series.

The student will be able to use the acquired knowledge and skills in creating a mathematical model and will learn to effectively use the commands of the mathematical program Maple for routine calculations and visualization for solving created model.

Brief outline of the course:

1. - 3. Systems of linear equations, matrices, determinants.

4. - 7. Functions of several variables, continuity and limit, partial derivatives, differential, local and global extrema of a function of two variables.

8. - 11. Modeling of relations between quantities using differential equations. Methods for solving ordinary differential equations of the 1st order.

12. - 13. Sequences, infinite number series, convergence criteria of infinite number series, infinite functional series, Taylor series.

Recommended literature:

1. Hughes-Hallett, D., et al.: Applied Calculus. John Wiley & Sons, Inc., 2010.

2. Rogers, R., C.: The Calculus of Several Variables. 2011.

Course language:

Slovak

Notes:

Course assessment

Total number of assessed students: 20

A	D	C	D	Е	БУ
A	D	C	D	E	ГЛ
50.0	20.0	20.0	5.0	5.0	0.0

Provides: doc. RNDr. Stanislav Lukáč, PhD., RNDr. Stanislav Basarik, Mgr. Barbora Hennelová

Date of last modification: 17.09.2021

University: P. J. Šaf	fárik University in Košice		
Faculty: Faculty of	Science		
Course ID: ÚFV/ MSA1/03Course name: Methods of Structural Analysis			
Course type, scope Course type: Lectu Recommended cou Per week: 3 / 2 Per Course method: p	ure / Practice urse-load (hours): r study period: 42 / 28		
Number of ECTS c	eredits: 7		
Recommended sem	nester/trimester of the course: 6.		
Course level: I., II.,	III.		
Prerequisities:			

Conditions for course completion:

For successful completing of the subject student has to show after taking exam adequate knowledge from the area using sophisticated research infrastructure for structural analysis of solids. Content of the subject is focused study of structure analytical methods as TEM, SEM, STEM and X-ray techniques. Credits evaluation takes into account taking part at the lectures - 3credits, study of recommended literature -1credit, working out of experimental protocol from OM and EM -2 credidts and study of recommended literature -2 credits, 2 credits – project, 1 credit – study for 2 written tests (EM and X-ray) - 1 credit. Minimal value to obtain evaluation for is reach 50% of each evaluation (tests and project) points. Point ratio protocol/test EM/TEST X-ray is 40/30/30.

Learning outcomes:

Student due to lecrures and experimental work after final exam demonstrates that he/she meets expectations according to the standards of the subject, which is predicted by short content andreferences. Student is able to use modern methods for structural analysis of metals. He has experiences with optic microscopy, electron microscopy (TEM, SEM, STEM), electron microprobe analysis and X-ray diffractometry.

Brief outline of the course:

Optic microscopy. Electron microscopy: Electron beam instruments, Electron optics, Electron lences and deflection systems, Transmission electron microscopy - principle and construction. Electron – specimen interactions. Electron diffraction. Kikuchy lines. Scanning electron microscopy – principle and cnstrucion. Scanning transmission electron microscopy. High Voltage electron microscopy. Electron microscopy. Electron microscopy. Convergent beam diffraction.

X-ray diffractometry: Scattering of x-rays, Neutrons and neutron scattering, CW - diffractometer, Ewald's sphere, Diffraction on powder samples, The main characteristics of powder diffraction pattern, Structure factor, Ocupation factor, Atomic displacement factor, Peak intensity, shape and symmetry, Sherrer equation. Peak profile, Rietweld method. Qualitative phase analysis, parameters of elementary cell, Profile analysis of diffraction peak and interpretation of profile analysis.

Recommended literature:

1. P.W. Hawkes, J.C.H. Spence, Science of Microscopy, Springer, 2007, ISBN: 10:0-387-25296-7.

2. Vitalij Pecharsky, Peter Zavalij, Fundamentals of Powder Diffraction and Structural characterization of Materials, Publisher: Springer (March 3, 2005)

ISBN-10: 0387241477, ISBN-13: 978-0387241470

3. Jens Als-Nielsen, Des McMorrow, Elements of Modern X-ray Physics, Publisher: Wiley; 2 edition (April 4, 2011),ISBN-10: 0470973943, ISBN-13: 978-0470973943.

4. Current Publications in the field of TEM, REM, X-ray

5. M.D. Graef, M.E. Henry, Structure of Materials, Cambridge Univ. Press, 2012, ISBN:978-1-107-00587-7.

6. S. Amelinckx, D. Dyck, et al, Electron Microscopy - Principle and Fundamentals, VCH, 1997, ISBN: 3-527-29479-1.

Course language:

1. English

Notes:

Lectures can be done at presence form or online using MS Teams. Education form is updated at the begining of the subject. All ppt presentations are accesible in LMS UPJŠ.

Course assessment

Total number of assessed students: 93

А	В	С	D	Е	FX	Ν	Р
38.71	21.51	7.53	1.08	0.0	0.0	0.0	31.18

Provides: prof. RNDr. Pavol Sovák, CSc., doc. Ing. Karel Saksl, DrSc., Ing. Vladimír Girman, PhD.

Date of last modification: 21.09.2021

University: P I	. Šafárik Univers	ity in Košice			
Faculty: Faculty					
Course ID: ÚB MKV/15		me: Microbiolo	gy and basics of	virology	
Course type: I Recommende	cope and the met Lecture / Practice d course-load (h 2 Per study peri d: present	ours):			
Number of EC	TS credits: 5				
Recommended	semester/trimes	ster of the cours	e: 5.		
Course level: I.					
Prerequisities:	ÚBEV/CYT1/15				
	course completi practicals (at le		ritten examinati	ons during sem	ester, final oral
their cytology, p	btain a basic info	tics, ecology, clas	ssification, and in	and eukaryotic 1 mportance . Infor	
				gy, physiology, ga l environment.	enetics, ecology,
Recommended	literature:				
Course languag	ge:				
Notes:					
Course assessm Total number of	nent f assessed studen	ts: 1464			
А	В	С	D	Е	FX
23.5	13.52	18.24	19.26	21.24	4.23
	RNDr. Peter Pris D., RNDr. Lenka			PhD., RNDr. Ma	iriana
Date of last mo	dification: 10.12	2.2021			
Approved: doc.	. Mgr. Daniel Jan	cura, PhD.			
		-			

University: P. J. Šafa	árik University in Košice
Faculty: Faculty of S	Science
Course ID: ÚFV/ MTBF/18	Course name: Modern Trends in Biophysics
Course type, scope a Course type: Lectu Recommended cou Per week: 2 Per stu Course method: pr	re irse-load (hours): udy period: 28
Number of ECTS ci	redits: 2

Recommended semester/trimester of the course: 1.

Course level: I.

Prerequisities:

Conditions for course completion:

Elaboration of a written essay on a selected topic from the course Modern trends in Biophysics.

Learning outcomes:

This course will provide an opportunity for students to be informed about the actual trends in biophysical research in the world as well about the research performed at the Department of Biophysics, Faculty of Science of UPJS and Center for Interdisciplinary Biosciences TIP-UPJS. Completing this course, the students should be able to understand texts from popular scientific literature about biophysics or bio-sciences in general.

Brief outline of the course:

Areas of interest in biophysics and its importance and position in science.

The structure of biophysics. Characterization of molecular, cellular, medical, environmental and radiation biophysics. Scientific disciplines related to biophysics. The future of biophysics. Lasers in life sciences.

History of laser development. Basic properties and principle of operation of lasers. Different types of lasers. Applications of lasers in biophysical experiments. Diagnostics and manipulation of samples using lasers, modern imaging methods, laser spectroscopy. Laser applications in clinical practice.

Raman spectroscopy and its application in biophysics.

Interaction of matter and light. Methods of optical spectroscopy (vibrational spectroscopy, Raman effect, mutual relation of Raman and infrared spectroscopy). Surface-enhanced Raman scattering. Raman macro- and micro-spectroscopy and imaging. Overview of the use of Raman spectroscopy with emphasis on biophysical applications (Raman spectra of proteins and other biomacromolecules, Raman cell imaging).

PickMolTM nanotechnology based on surface-enhanced Raman spectroscopy for water and food purity screening.

This technology detects whether water or any food matrix is contaminated with persistent organic pollutants (POPs)/pesticides/drugs. The patented PickMoITM technology has recently been verified by a certified laboratory and can be adapted for any organic molecule, which means its great potential for application in other areas, e.g. pharmaceutical and chemical industry, security and sports (doping control). PickMoITM technology is: highly sensitive (ppb concentration level),

selective (detection of specific molecules), efficient (up to 90% cost savings), fast (10 minutes per analysis), portable with immediate on-site analysis.

Methods of studying ligand-macromolecule interaction.

Use of optical spectroscopy methods in the study of ligand-macromolecule interactions: UV-vis absorption spectroscopy, fluorescence spectroscopy, Raman spectroscopy techniques (classical, surface-enhanced, DCDRS). Their advantages and limits. Use of thermodynamic methods: differential scanning calorimetry (study of the stability of the ligand-macromolecule complex) and isothermal titration calorimetry (direct measurement of thermodynamic parameters associated with the formation of complexes).

Scientific superhub European XFEL and biophysics.

X-ray optics and imaging techniques in nano- and mesoscopic scale. Serial protein crystallography and imaging of biological particles (viruses, supramolecular complexes and nature-inspired nanotechnologies). Structural dynamics of biomolecules - dynamic mixing and pump-probe experiments from Terraherz to the hard X-ray region. Mapping the electronic structure of molecules and clusters and their induced dynamics. Chemical imaging. Supercomputer clusters, protein engineering and production of protein nanocrystals, digital micro- and nano-fluidics, 3D printing and additive technologies. Complementary and supporting technologies - optical superresolution techniques, cryolelectron microscopy and electron diffraction. Diffraction limited X-ray sources in the future and atomistic cell mapping. "State-of-the-art" application - demos with commentary. The ecosystem of scientific super hubs as user equipment and opportunities to use EuXFEL. Protein evolution "in a test tube".

Principle of evolutionary methods for the development of proteins and enzymes. Examples of evolutionary methods - display technologies: ribosome display and yeast display. Application of display technology in the development of new or improved protein / enzyme properties for pharmaceutical and biotechnological applications.

Photobiostimulation

Principles of photobiostimulation, cell chromophores, application of therapeutic applications of photobiostimulation on absorption spectra of cell chromophores, perspective applications of photobiostimulation.

Metabolic flows in cells.

The importance of cell metabolism for the therapy of diseases, the principles of measuring metabolic flows, possible applications of methodology for the research in the field of cancer and neurodegenerative diseases.

Modern techniques for the study of individual molecules.

Force spectroscopy of proteins and nucleic acids (AFM, use of optical tweezers), determination of nanomechanical properties of biomolecules, overview of time trajectory analyzes of individual molecules, overview of applications of methods for studying properties of single (molecule) biomolecules: acoustic force spectroscopy, magnetic tweezers and mass photometry. Bioenergetics.

Central concept of bioenergetics - chemiosmotic theory. The main sources of energy in living organisms. Processes in biological systems in which energy is consumed. Mitochondria - structure and basic functions. Respiratory chain in mitochondria. Respiratory chain components. Mechanism of electron transport in the respiratory chain.

Recommended literature:

1. R. Glaser. Biophysics (2nd Edition), Springer-Verlach Berlin, 2012.

- 2. M.B. Jackson. Molecular and cellular biophysics, Cambridge University Press, 2006.
- 3. D.J. Dowsett, P.A. Kenny and R.E. Johnston. The physics of diagnostic imaging, Hodder Arnold, 2006.

4. M.R. Hamblin and P. Mroz. Advances in photodynamic therapy, Artech House,

2008.

4. D. Nicholls and S. Fergusson. Bioenergetics 4, Academic Press, 2013.

5. A.D.N.J. de Grey. The mitochondrial free radical theory of aging, R.G. Landis Company, 1999.

6. N. Lane. Síla, sexualita, sebevražda. Mitochondrie a smysl života, Academia, 2012.

Course language:

English language

Notes:

Course assessment

А	В	С	D	Е	FX
76.92	23.08	0.0	0.0	0.0	0.0

Provides: doc. Mgr. Daniel Jancura, PhD., doc. RNDr. Katarína Štroffeková, PhD., doc. RNDr. Gabriel Žoldák, PhD., prof. RNDr. Pavol Miškovský, DrSc., doc. Mgr. Gregor Bánó, PhD., RNDr. Zuzana Jurašeková, PhD., doc. RNDr. Jozef Uličný, CSc., doc. RNDr. Erik Sedlák, DrSc.

Date of last modification: 29.09.2022

University: P. J. Ša	ıfárik Universi	ty in Košice			
Faculty: Faculty of	f Science				
Course ID: ÚFV/ MBF1/14	Course na	me: Molecular]	Biophysics I		
Course type, scope Course type: Lect Recommended co Per week: 2 Per s Course method: p	ture ourse-load (ho study period:	ours):			
Number of ECTS	credits: 4				
Recommended sen	nester/trimes	ter of the cours	e: 4.		
Course level: I., II.					
Prerequisities:					
Conditions for cou	irse completio	on:			
Learning outcome	s:				
Brief outline of the	e course:				
Recommended lite	erature:				
Course language:					
Notes:					
Course assessment Total number of as		s: 32			
A	В	С	D	Е	FX
59.38	28.13	9.38	0.0	3.13	0.0
Provides: RNDr. G	abriela Fabric	iová, PhD.			
Date of last modifi	ication: 24.11	.2021			
Approved: doc. Ma	gr. Daniel Jan	cura, PhD.			

University: P. J. Ša	fárik Univers	ity in Košice			
Faculty: Faculty of	Science				
Course ID: ÚFV/ MBB1/18	Course na	me: Molecular a	nd cell biology		
Course type, scope Course type: Lect Recommended co Per week: 2 / 2 Po Course method: p	ture / Practice ourse-load (h er study perio	ours):			
Number of ECTS	credits: 5				
Recommended sen	nester/trimes	ster of the course	e: 6.		
Course level: I.					
Prerequisities:					
Conditions for cou	rse completi	on:			
Learning outcome	s:				
Brief outline of the	e course:				
Recommended lite	erature:				
Course language:					
Notes:					
Course assessment Total number of as	-	ts: 6			
А	В	С	D	Е	FX
83.33	16.67	0.0	0.0	0.0	0.0
Provides: doc. RNI	Dr. Katarína Š	Stroffeková, PhD.	, RNDr. Zuzana	Naďová, PhD.	1
Date of last modifi	cation: 12.07	2.2022			
Approved: doc. M	gr. Daniel Jan	cura, PhD.			

Faculty: Faculty of S	cience
Course ID: ÚFV/ NUM/10	Course name: Numerical Methods
Course type, scope a Course type: Lectur Recommended cour Per week: 2 / 1 Per Course method: pre	re / Practice rse-load (hours): study period: 28 / 14
Number of ECTS cr	edits: 4
Recommended seme	ster/trimester of the course: 3.
Course level: I.	
Prerequisities:	
algebra, which are n evaluation is particip obtaining credits is p electronically and wi into account the follo projects (2 credits). T	bility to apply the basic numerical methods of mathematical analysis and necessary for subsequent courses in computational physics. The basis o bation and activity in exercises and work on assignments. The condition for bassing 2 written tests at seminars and submitting 4 assignments (projects the attached computer program. The credit evaluation of the course takes owing student workload: direct teaching (2 credits) and individual work of the minimum threshold for completing the course is to obtain at least 50% of the following rating scale: A (90-100%), B (80-89%), C (70-79%), D (60 (0-49%).
for the next course of functions, solve syst	with the basic numerical methods of mathematical analysis and algebra needed computational physics. The student will learn to approximate and interpolate ems of linear and nonlinear equations, numerically derive and integrate o es and eigenvectors of matrices.
 Approximation of Interpolation of function Approximation by Solution of nonline Numerical method Solution of system Solution of system Numerical integrat Numerical different 	ution of problems and errors of numerical solution. functions.

Basic literature:

POZRIKIDIS, C.: Numerical Computation in Science and Engineering, Oxford University Press, 2008.

Other literature:

HAMMING, R.W.: Numerical Methods for Scientists and Engineers, Dover, 1973.

GARCIA, A.L.: Numerical Methods for Physics, Prentice-Hall, 1994.

Notes:

Course assessment

А	В	С	D	Е	FX		
14.63	15.85	23.17	24.39	17.68	4.27		

Provides: prof. RNDr. Milan Žukovič, PhD.

Date of last modification: 14.09.2021

University: P. J. Šafárik University in Koš	ice
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Faculty: Faculty of Science

Course ID: ÚCHV/	Course name: Organic Chemistry
OCHB/10	

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours):

Per week: 3 / 1 **Per study period:** 42 / 14

Course method: present

Number of ECTS credits: 5

Recommended semester/trimester of the course: 2.

Course level: I.

Prerequisities: ÚCHV/VACH/10

Conditions for course completion:

1.Participation in seminars (also applies to the online form of teaching): justified non-participation of the student in two seminars will be excused by the teacher; longer-term justified non-participation of the student in seminars must be proven by the student's mastery of the subject matter in an alternative form determined by the teacher (e.g. preparation of assignments and others...).

2. Activity at seminars (also applies to the online form of teaching) - theoretical preparation of students is required for all seminars.

3.Short written examinations at seminars (max. 50b). Credit slips in the 7th and 14th week with a total sum of 100b. To pass the E rating, it is necessary to obtain 25.5b from each test.

4. The exam is a form of test. A minimum of 51 points is required to pass the exam. The final grade is calculated as the average of the evaluation of papers in seminars, credit papers and the exam itself. Final Grade: A: 91-100b, B: 81-90b, C: 71-80b, D: 61-70b, E: 51-60b, FX: 0-50b.

Learning outcomes:

After completing the course, the student, based on the study of common and different features of compounds, should be able to assess the properties of a given type of compound from the structure and name the corresponding type of compound based on nomenclature principles. From the acquired knowledge about the structure and properties of the relevant types of hydrocarbon compounds, the student should be able to independently derive the mechanisms of individual reactions.

Brief outline of the course:

Recommended literature:

- 1. Online ppt presentations in the system MOODLE na moodle science.upjs.sk
- 2. Organic chemistry, Clayden, Greeves Warren & Wothers, Oxford university Press, 2010.
- 3. Organická chémia, John McMurry, Vysoké učení technické v Brne, 2007, VUTIUM, ISBN: 978-80-214-3291-8 (VUT v Brne).
- 4. Organická chémia, Pavol Zahradník, Mária Mečiarová, Peter Magdolen, Univerzita

Komenského v Bratislave, 2019, ISBN: 978-80-223-4589-7.

Course language:

anglický

Notes:

Teaching is carried out in person or, if necessary, online, using the MS Teams tool. The form of teaching is specified by the teacher at the beginning of the semester, updated continuously.

0 1	5	e	8	× 1	5		
Course assessment Total number of assessed students: 298							
A B C D E FX							
20.13	20.81	32.89	19.13	6.71	0.34		
Provides: RNDr. Slávka Hamuľaková, PhD., doc. RNDr. Miroslava Martinková, PhD., doc. RNDr. Mária Vilková, PhD.							
Date of last modification: 15.08.2022							
Approved: doc.	Approved: doc. Mgr. Daniel Jancura, PhD.						

University: P. J. Šafárik University in Košice						
Faculty: Faculty of Science						
Faculty: Faculty of S						
Course ID: ÚFV/ FCH1/02	Course name: Physical Chemistry for Biological Sciences					
Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 3 / 2 Per study period: 42 / 28 Course method: present						
Number of ECTS credits: 6						
Recommended semester/trimester of the course: 3.						
Course level: I., II.	Course level: I., II.					
Prerequisities:						
Conditions for cours Test	Conditions for course completion:					

Exam

During an exam, a student should demonstrate his/her ability to solve theoretical exercises from the selected parts of the Physical chemistry for biological sciences. Morever, the students should be able to manifest theoretical knowledge from the chapters which are present in the brief outline of the course.

Learning outcomes:

The introduction into the fundamental knowledge of selected parts of physical chemistry with emphasis on the utilization of these knowledge for the study of physico-chemical properties of biomacromolecules and biological systems. After completing the course, the students should understand physico-chemical mechanisms of many biological processes.

Brief outline of the course:

Week 1

Physical chemistry - areas of research, importance for science, definition. Thermodynamics - definition, areas of interest. Thermodynamic system. Properties of thermodynamic system. Basic thermodynamic quantities (pressure, volume, temperature, internal energy). Zero law of thermodynamics. Ideal gas. Equation of state of an ideal gas. Gas mixtures - Dalton's law. Real gas. Van der Waals equation of state.

Week 2

1st law of thermodynamics. Internal energy, work, heat. Mathematical formulation of the 1st law of thermodynamics. Enthalpy. Heat capacity. Relationship between heat capacities at constant pressure and volume. Isothermal expansion of an ideal gas. Work in reversible and irreversible isothermal expansion. Adiabatic expansion of an ideal gas. Exothermic and endothermic reactions and processes. Standard state of substances. Hess's law.

Week 3

Examples of spontaneous processes in nature. Definitions of the 2nd law of thermodynamics (Kelvin, Celsius). Entropy - introduction of the term. Thermodynamic definition of entropy. Entropy as a state function. Carnot cycle. Efficiency of a heat engine. Claussius inequality. Entropy

of isothermal expansion, gas mixing, melting and evaporation processes. Dependence of entropy on temperature. Nernst's heat theorem. 3rd law of thermodynamics. Week 4

Entropy as a property determining the spontaneity of processes. Criteria of process spontaneity at constant volume and constant pressure. Helmoltz and Gibbs free energy. Properties of Helmoltz energy. Properties of Gibbs energy. Standard Gibbs energy of a chemical reaction. Dependence of Gibbs energy on temperature - Gibbs-Helmoltz equation. Dependence of Gibbs energy on pressure for solids, liquids and gases. Simple mixtures. Partial molar volume. Partial molar Gibbs energy, chemical potential.

Week 5

Chemical potential in a liquid. Raoult's law, the ideal solution. Henry's law, ideally diluted solution. Mixing solutions, ideal solutions. Residual functions and regular solutions. Colligative properties. Increasing the boiling point and decreasing the melting point of the liquid in which the soluble chemical compound is located. Osmosis. Solvent activity, soluble substance activity. Week 6

Chemical equilibrium. Gibbs energy of a chemical reaction. Chemical equilibrium in an ideal gas. Equilibrium constant of chemical reaction. Temperature dependence of the equilibrium constant van't Hoff's equation. Stability of protein structure. Thermal denaturation of proteins. Van't Hoff enthalpy of protein denaturation. Chemical denaturation of proteins. Physiological consequences of incorrectly folded proteins.

Week 7

Examples of molecular associations and their significance for biological systems. Dissociation and association binding constants. Determination of dissociation binding constant - Langmuir isotherm. Cooperativity in ligand-macromolecule interactions. Cooperativity - simultaneous ligand binding, Hill's equation. Cooperativity - gradual binding of ligands. Allosteric interactions. Qualitative description of the Monod - Wyman - Changeaux model for cooperative binding of ligands to macromolecules. Experimental methods used to study the ligand - macromolecule interactions.

Week 8 Chemical and biochemical kinetics - basic definitions. Rates of chemical reactions. Rate constant. Order of chemical reaction. First order reactions. Second order reactions. Consecutive reactions. Determination of the rate law. Reverse chemical reactions. Relaxation processes. Temperature dependence of rate constants - Arrhenius equation. Experimental techniques used to determine the rates of chemical reactions. Transition state theory - Eyring's theory. Week 9

Enzymes - characterization and classification. Equilibrium model of enzyme kinetics. Steady state model of enzyme kinetics. Experimental determination of maximum rate and Michaelis-Menten constant in enzymatic reactions. Deviations from Michaelis-Menten kinetics. Enzyme inhibition. Reversible inhibition. Competitive, non-competitive and uncompetitive inhibition. Week 10

Kinetics of photophysical and photochemical processes. Jablonski diagram. Fluorescence, phosphorescence. Quantum yields of photophysical processes. Quenching of the excited states of molecules by external factors. Fluorescence quenching. Stern-Volmer equation. Förster resonance energy transfer (FRET). Biological application of FRET.

Week 11

Electrochemical reactions. Electrochemical cell. Standard redox potentials. Relationship between Gibbs energy change and electrochemical potential. Temperature dependence of electrochemical potential. Use of electrochemical cells. Determination of redox potential. Ionic electrochemical gradient. Proton motive force. Nernst potential. Introduction to the respiratory chain in mitochondria.

Week 12

Acids and bases. Acid-base properties of water. pH - measurement of environmental acidity. Dissociation of acids and bases - acid-base equilibrium. Henderson - Hasselbalch equation. Buffers.

Recommended literature:

	literature:						
1. P. Atkins and	J. de Paula. Atk	ins's Physical Ch	nemistry (9th Edi	tion), Oxford			
University Pres	· ·						
2. P. Atkins. Fyzikálna chémia (slovenský preklad 6. vydania), STU Bratislava, 1999.							
3. P. Atkins, J. De Paula. Fyzikální chemie (český preklad 9. vydania), VŠCHT Praha,							
2013							
	· ·		ces, University S)6.		
-		•	stry with Applica	ations to the Life			
, 5	min/Cummings,						
	e, W. Johnson and	d P. Ho. Principle	es of Physical Bio	ochemistry, Prent	ice		
Hall, 1988.	D: 1 · 1 m	1		• • • •	D		
	Biological Theri	modynamics (2nd	d Edition), Camb	ridge University	Press,		
2008.	Canaiga Chami	a al Thanna a dama	union (2nd Edition) CDC Dragg Ta	vilan Pr		
		ical Thermodyna	mics (3rd Edition	I), CRC Press, Ta	lylor a		
Francis Group,		Inna I.C. Dualia	, G. Harbison and	d D D overvalz			
	,	0, 0	s in Biological So		(on)		
Pearson, 2014.	istry – Fincipies	and Application	s in Diological Sc	Jences (Jui Editi	.011),		
· · ·	Physical Chemis	try- Thermodyna	amics, Statistical	Mechanics and			
Kinetics, Pearso		ary mermodyne	unites, Statistical	inteenanies, and			
,							
Course languag							
English languag							
Notes:							
Notes: Course assessm		ts: 112					
Notes: Course assessm	nent	ts: 112 C	D	E	FX		
Notes: Course assessm Total number of	ent f assessed studen		D 11.61	E 9.82	FX 0.0		
Notes: Course assessm Total number of A 19.64	tent f assessed studen B	C 30.36					
Notes: Course assessm Total number of A 19.64 Provides: doc. N	f assessed studen B 28.57	C 30.36 ura, PhD.					

University: P. J. Šafá	arik University in Košice
Faculty: Faculty of S	Science
Course ID: ÚFV/ ZFP1a/03	Course name: Physics Practical I
Course type, scope a Course type: Practi Recommended cou Per week: 3 Per stu Course method: pro	ce rse-load (hours): ıdy period: 42
Number of ECTS cr	redits: 3
Recommended seme	ester/trimester of the course: 2.
Course level: I.	
Prerequisities:	-
Conditions for cours The active work duri Vindication of report	ing semester and hand in all reports.
Learning outcomes: Developing proper la	aboratory habits, skills and verify their theoretical knowledge.
 with kinds and calcures with kinds and calcures results. The students introductory physics Laboratory assignment 1. Density measurement 2. Radius measurement 2. Radius measurement 3. Gravitational acceleration of the strategy of	oratory exercises is to familiarize the students with measurement methods, alus of mistakes, with measured results processing, and with presentation of gain practical skills, and verify their theoretical knowledge of first semester course. They develop proper laboratory habits. ent: hents of liquids and solids. ents of spherical cap. Measurements of leter. heration measurements using mathematical im. measurement using physical and torsion Young's modulus. oefficient of viscosity. he speed of sound. general gas constant and Boltzmann constant. thermal expansivity of air. f thermal capacity of matter.
measurements I), Ed	 C., Onderová, Ľ., Kireš, M.: Základné fyzikálne praktikum I. (Basic physical PF UPJŠ Košice 2007. 31. Slovenský inštitút normalizácie v Bratislave (Slovak institute of technical

Ješková, Z.: Computer based experiments in thermodynamics using IP COACH,ed. PF UPJŠ in Košice, 2004.

Course language english	ge:					
Notes:						
Course assessm Total number o	nent f assessed studen	ts: 275				
А	В	С	D	Е	FX	
57.45	25.82	12.73	3.27	0.73	0.0	
Provides: doc. RNDr. Adriana Zeleňáková, PhD., doc. RNDr. Marián Kireš, PhD., doc. RNDr. Jár Füzer, PhD., doc. RNDr. Jozef Hanč, PhD.						
Date of last modification: 29.03.2020						
Approved: doc.	. Mgr. Daniel Jan	cura, PhD.				

University: P. J.	Šafárik University in Košice	

Faculty: Faculty of Science

Course ID: ÚFV/	Course name: Physics Practical II
ZFP1b/03	

Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 3 Per study period: 42

Course method: present

Number of ECTS credits: 3

Recommended semester/trimester of the course: 3.

Course level: I.

Prerequisities: ÚFV/ZFP1a/03

Conditions for course completion:

To successfully complete the course, the student must measure at least 11 experimental tasks, process and analyze the measured results and evaluate the experimental results in the form of a protocol.

The condition for the implementation of another experimental task is the submission of a protocol from the previous exercise.

The condition for the implementation of the practical task is sufficient theoretical training at home. If the student is not ready for the task in advance, the teacher can send him home and the student must replace the exercise at another time.

The credit evaluation of the course takes into account the following student workload:

1 credit: self-study of recommended literature and subsequent direct teaching

1 credits: realization of experimental exercise and subsequent defense of measuring procedure - it is obligatory to complete all practical tasks in the semester,

1 credit: elaboration and submission of protocols from measurements, which are evaluated.

Learning outcomes:

By completing the course, the student will get acquainted with selected physical experiments in the field of electricity and magnetism and supplement the theoretical knowledge acquired in the course General Physics in a practical way.

The result of education is:

a) Complementing and summarizing knowledge and experimental skills in the field of electricity and magnetism.

b) Gaining practical experience with recording, analysis and interpretation of experimental data from practical measurements.

c) Gaining experience with the presentation of experimental results in the form of a measurement protocol.

Brief outline of the course:

Students on practical exercises are working in pairs experimental tasks in the field of electrical, electromagnetic and magnetic properties of matters.

1. Electrical Resistivity

2. Self - and Mutual Inductance and Capacity

- 3. Serial and Parallel Resonance
- 4. Thermal Dependence of Selected Electrical Phenomena in Solids
- 5. The Characteristics of Semiconductor Diod
- 6. The Characteristics of Semiconductor Bipolar Transistor
- 7. Magnetic Hysteresis
- 8. Hall Constant Measurements
- 9. Measurements of Horizontal Component of Earth Magnetic Field
- 10. Measuring characteristics of switching components
- 11. Measuring the properties of optoelectronic components
- 12. Electric current in liquids and electrolysis

Recommended literature:

- 1. Tumanski S, Handbook of magnetic measurements, CRC press, 2011.
- 2. Fiorillo F, Characterization and Measurement of Magnetic Materials, Elsevier, 2004.

Course language:

english

Notes:

Teaching is carried out in person. If necessary, part of the teaching can be realized remotely using the MS Teams or BBB tool. At the beginning of the semester, the teacher sets the conditions for completing and mastering the course.

Course assessment

Total number of assessed students: 249

А	В	С	D	Е	FX
66.27	19.68	12.05	1.61	0.0	0.4

Provides: doc. RNDr. Adriana Zeleňáková, PhD., doc. RNDr. Ján Füzer, PhD.

Date of last modification: 30.09.2021

University: P. J. Ša	fárik Univers	ity in Košice			
Faculty: Faculty of	Science				
Course ID: ÚFV/ ZFP1c/14	Course na	me: Physics Pra	ctical III		
Course type, scope Course type: Prac Recommended co Per week: 3 Per s Course method: p	tice urse-load (he tudy period:	ours):			
Number of ECTS	credits: 3				
Recommended sen	nester/trimes	ter of the cours	e: 4.		
Course level: I.					
Prerequisities:					
Conditions for cou Measurements of ex defended. As a part of the task.	xperimental ta	isks, their evalua			
Learning outcome To gain some physic practice in data co report writing prese	ical inside int llection, anal	ysis and interpre			-
Brief outline of the Oscilations. Pendul sound. Refractive i of waves. Polarizat	um. Compos ndex. Lense's	focal length. In	terference. Diffra		-
Recommended lite Degro,J., Ješková, 2006 P. Kollár a kol. Zák J. Brož Základy fys	Z., Onderová, ladné fyzikál	ne praktikum II,	PF UPJŠ Košice	-	UPJŠ Košice,
Course language: slovak, english					
Notes:					
Course assessment Total number of ass		ts: 94			
A	В	С	D	Е	FX
68.09	19.15	7.45	2.13	3.19	0.0
Provides: doc. RNI	Dr. Marián Ki	reš, PhD., doc. F	NDr. Ján Füzer.	PhD.	
Date of last modifi			- 7		
		.2022			

University: P. J. Ša	fárik Univers	ity in Košice				
Faculty: Faculty of	Science					
Course ID: ÚFV/ PEMBF1/18	Course name: Practical exercises in experimental methods of biophysics I					
Course type, scope Course type: Prac Recommended co Per week: 2 Per s Course method: p	tice ourse-load (ho tudy period:	ours):				
Number of ECTS	credits: 2					
Recommended sen	nester/trimes	ter of the cours	e: 4.			
Course level: I.						
Prerequisities: ÚF	V/EMBF1/18					
Conditions for cou	rse completi	on:				
Learning outcome	s:					
Brief outline of the	e course:					
Recommended lite	rature:					
Course language:						
Notes:						
Course assessment Total number of ass		ts: 6				
А	В	С	D	Е	FX	
66.67	33.33	0.0	0.0	0.0	0.0	
Provides: RNDr. G	abriela Fabric	ciová, PhD.				
Date of last modifi	cation: 25.11	.2021				
Approved: doc. Mg	gr. Daniel Jan	cura, PhD.		_		

University: P. J. Ša	fárik Univers	ity in Košice			
Faculty: Faculty of Science					
Course ID: ÚFV/ Course name: Practical exercises in experimental methods of biophysics II PEMBF2/18					
Course type, scope Course type: Prac Recommended co Per week: 2 Per s Course method: p	etice ourse-load (h tudy period:	ours):			
Number of ECTS	credits: 2				
Recommended sen	nester/trimes	ster of the course	e: 5.		
Course level: I.					
Prerequisities: ÚF	V/EMBF2/18				
Conditions for course completion:					
Learning outcomes:					
Brief outline of the	e course:				
Recommended lite	erature:				
Course language:					
Notes:					
Course assessment Total number of as		ts: 5			
A	В	С	D	Е	FX
100.0	0.0	0.0	0.0	0.0	0.0
Provides: doc. RN	Dr. Erik Sedlá	ık, DrSc., RNDr.	Gabriela Fabrici	ová, PhD.	
Date of last modification: 30.11.2021					
Approved: doc. Mgr. Daniel Jancura, PhD.					

	University:	ΡJ	Šafárik	University	in Košice
I	University.	1	Juliant	Oniversity	

Faculty: Faculty of Science

Course ID: ÚFV/	Course name: Practical exercises in experimental methods of biophysics III
PEMBF3/18	

Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: present

Number of ECTS credits: 2

Recommended semester/trimester of the course: 6.

Course level: I.

Prerequisities: ÚFV/EMBF3/18

Conditions for course completion:

Elaboration of protocols.

Learning outcomes:

The students gain experimental skills in the methods of confocal fluorescence microscopy, time-resolved fluorescence microscopy (FLIM / PLIM), flow cytometry, microfluidics and the development of experimental apparatuses in the field of biophotonics.

Brief outline of the course:

1-2. Sample preparation for confocal fluorescence microscopy and vital fluorescence imaging in cells.

3-4. Time-resolved fluorescence microscopy (FLIM) - sample preparation and fluorescence lifetime imaging in cells.

5. Immuno-fluorescence imaging - sample preparation for confocal fluorescence microscopy and imaging of selected proteins in cells.

6. Flow cytometry - detection of fluorescent labels in living cells.

7. Setting up the function generator and the digital oscilloscope (the influence of input resistance, triggering and averaging of time courses). Noise monitoring, optimization of shielding and grounding of devices. Measurement using a laboratory card, evaluation of spectral noise density.

8. Safety training for working with optical radiation in practice. Safe work with lasers, setting the beam path through a cuvette. Measuring fluorescence spectra using a spectrograph.

9. Measurement of weak optical signals. Comparison of different types of photodiodes. Connection and setting of the lock-in amplifier when measuring with a chopped laser beam.

10. Temperature regulation and measurement. Connection and adjustment of a PID temperature regulator using a Peltier cell in combination with a thermocouple.

11. Preparation of a simple microfluidic system. Channel design. Mold preparation. Fabrication of a microfluidic system using PDMS.

12. Setting up the imaging system using a CMOS camera. Measurement of fluid flow rate in a microfluidic system based on video recording.

Recommended literature:

1. Wolfgang Becker: The bh TCSPC Handbook Seventh Edition, Becker & Hickl GmbH 2017;

2. Guy Cox: Optical Imaging Techniques in Cell Biology, Taylor & Frances;

I	3. Howard M. Shapiro: Practical Flow Cytometry Fourth edition, 2003;
l	4 Nilvalag Long and Wing Tale Wang, The chamistry of malacular imaging Wiley 20

4. Nikolas Long and Wing-Tak Wong: Th	ne chemistry of molecular imaging, Wiley 2014
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Course language: Slovac language, English.

Notes:

1100050					
Course assessment					
Total number o	f assessed studen	ts: 5			
А	В	С	D	Е	FX
100.0	0.0	0.0	0.0	0.0	0.0
Provides: doc. Mgr. Gregor Bánó, PhD., RNDr. Zuzana Naďová, PhD., RNDr. Veronika Huntošová, PhD.					
Date of last mo	dification: 22.09	9.2021			

University: P. J. Šaf	ărik University in Košice		
Faculty: Faculty of	Science		
Course ID: ÚFV/ KVM I/11			
Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 4 / 2 Per study period: 56 / 28 Course method: present			
Number of ECTS c	redits: 8		
Recommended sem	ester/trimester of the course: 3.		
Course level: I.			
Prerequisities:			
Conditions for cour	rse completion:		

To successfully complete the course, the student must demonstrate sufficient understanding of the basic terms, concepts and applications of quantum physics. Knowledge of basic concepts of quantum physics is required at the level of their mathematical definition, as well as their physical content and specific applications. During the semester, the student must continuously master the content of the curriculum so that he can actively and creatively use the acquired knowledge in solving specific computational tasks during the exercises and pass continuous written tests taken into account in the overall evaluation of the subject. The condition for obtaining credits is passing 2 continuous written tests in exercises and an oral exam, which consists of one more demanding calculation task and theoretical questions. The credit evaluation of the course takes into account the following student workload: direct teaching (3 credits), self-study (2 credits), individual consultations (1 credit) and assessment (2 credits). The minimum threshold for completing the course is to obtain at least 50% of the total score, using the following rating scale: A (90-100%), B (80-89%), C (70-79%), D (60- 69%), E (50-59%), F (0-49%).

Learning outcomes:

After completing lectures and exercises, the student will have sufficient physical skills,

knowledge and mathematical apparatus enabling independent solution of a wide range

traditional and current scientific problems in quantum physics. At the same time, he will gain an overview of the applications of quantum physics in various areas of physics - nuclear physics, condensed matter physics, statistical physics, quantum information theory, etc.

Brief outline of the course:

1. Subject matter, experimental and theoretical foundations of quantum mechanics (QM).

2. Wave formulation of QM. Postulate about wave function, superposition principle and postulate about operators.

3. Eigenvalues and eigenfunctions of operators. Measurement of quantities and reduction of wave function.

4. Time-independent and time-dependent Schrödinger equation. Ehrenfest equations and integrals of motion. Continuity equation.

5. Matrix formulation of QM, Dirac symbolics, calculation of mean values and density matrix.

6. Current immeasurability of physical quantities, Heisenberg uncertainty relations.

7. Solution of the Schrödinger equation for a particle in an infinitely deep potential well and a particle in a finite potential well. Bound and scattering states.

8. Passage of a particle through a potential barrier: tunneling and barrier reflection.

9. Solution of Schrödinger equation for linear harmonic oscillator.

10. Particle motion in the central potential field, angular part of the Schrödinger equation.

11. Particle motion in the central potential field, radial part of the Schrödinger equation. Hydrogen atom.

12. Electron spin, Pauli matrix. Principle of indistinguishability of identical particles, fermions and bosons. Pauli's exclusion principle.

13. Paradoxes and modern problems of QM. Quantum entanglement, nonlocality, computing, cryptography and teleportation.

Recommended literature:

1. Ľ. Tóth, M. Tóthová, Kvantová a štatistická fyzika I, Rektorát Univerzity P. J. Šafárika, 1982. (in Slovak language)

2. Ľ. Skála, Úvod do kvantovej mechaniky, Academia, Praha, 2005. (in Czech language)

3. J. Pišút, L. Gomolčák, Úvod do kvantovej mechaniky, Bratislava 1983. (in Slovak language)

4. W. Greiner, Quantum Mechanics, 4th edition, Springer, Berlin, 2000.

5. A. C. Philips, Introduction to Quantum Mechanics, Wiley, Weinheim, 2003.

6. D. J. Griffiths, Introduction to Quantum Mechanics, Prentice Hall, New Jersey, 1995.

7. G. Auletta, M. Fortunato, G. Parisi, Quantum Mechanics, Cambridge University Press, Cambridge, 2009.

Course language:

Notes:

Course assessment

Total number of assessed students: 97

А	В	С	D	Е	FX
26.8	19.59	19.59	10.31	18.56	5.15

Provides: doc. RNDr. Jozef Strečka, PhD.

Date of last modification: 19.09.2021

University: P. J. Šat	fárik University in Košice		
Faculty: Faculty of Science			
Course ID: ÚFV/ KVM II/08Course name: Quantum Mechanics II.			
Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 3 / 1 Per study period: 42 / 14 Course method: present			
Number of ECTS credits: 6			
Recommended semester/trimester of the course: 4.			
Course level: I.			

Prerequisities: ÚFV/KVM/08 or ÚFV/KVM I/11

Conditions for course completion:

To successfully complete the course, the student must demonstrate sufficient understanding of the basics terms, concepts and applications of quantum physics. Knowledge of basic concepts is required from quantum physics at the level of their mathematical definition as well as their physical content and concrete applications. During the semester, the student must continuously master the content of the curriculum in order to gain the acquired knowledge in order to actively and creatively use them in solving specific tasks during the exercises and complete the continuous written test taken into account in the overall evaluation of the subject. The condition for obtaining credits is passing 1 continuous written test in exercises and an oral exam, which consists of one more complex computational task and theoretical questions. The credit evaluation of the course takes into account the following student workload: direct teaching (2 credits), self-study (2 credits), individual consultations (1 credit) and assessment (1 credit). Minimum threshold for passing the subject is to obtain at least 50% of the total score, while it is used the following rating scale: A (90-100%), B (80-89%), C (70-79%), D (60-69%), E (50-59%), F (0-49%).

Learning outcomes:

After completing lectures and exercises, the student will have sufficient physical skills, knowledge and mathematical apparatus enabling independent solution of a wide range traditional and current scientific problems in quantum physics using approximate methods. At the same time, he will gain an overview of the applications of quantum physics in various fields of

same time, he will gain an overview of the applications of quantum physics in various fields of physics such as atomic and nuclear physics, condensed matter physics, statistical physics, quantum theory of magnetism, etc.

Brief outline of the course:

1. Stationary perturbation theory for non-degenerate quantum-mechanical systems with discrete energy spectrum.

2. Stationary perturbation theory for degenerate quantum-mechanical systems with discrete energy spectrum. Zeeman and Stark effects.

3. Stationary perturbation theory for two-level quantum-mechanical systems with two closelyspaced energy levels: crossing and selfavoided crossing of energy levels.

4. Ritz's variational method. Bound state of a quantum-mechanical system with attractive potential.

5. Applications of Ritz's variation method in finding the ground state of quantum spin models.

6. Nonstationary perturbation theory for non-degenerate quantum-mechanical systems with discrete energy spectrum. A special case of constant, adiabatic and short-rapid perturbation.

7. Nonstationary perturbation theory for quantum-mechanical systems with discrete-continuous energy spectrum. Harmonic perturbation and Fermi's golden rule.

8. Quantum-mechanical solution of the time-indepedent Schrődinger equation for the helium atom using stationary perturbation theory. Orthohelium and parahelium.

9. Quantum-mechanical solution of the time-indepedent Schrödinger equation for a hydrogen molecule using stationary perturbation theory. Heitler-London theory of valence bonds.

10. Quantum-mechanical solution of the time-indepedent Schrödinger equation for a hydrogen molecule using the Ritz variational method. LCAO method.

11. Hartree and Hartree-Fock method for multielectron atoms.

Recommended literature:

1. V. Ilkovič, Kvantová teória II, Scriptum UPJŠ, Košice, 1989. (in Slovak)

- 2. J. Pišút, L. Gomolčák, Úvod do kvantovej mechaniky, Bratislava 1983. (in Slovak)
- 3. W. Greiner, Quantum Mechanics, 4th edition, Springer, Berlin, 2000.
- 4. D. J. Griffiths, Introduction to Quantum Mechanics, Prentice Hall, New Jersey, 1995.

5. G. Auletta, M. Fortunato, G. Parisi, Quantum Mechanics, Cambridge University Press, Cambridge, 2009.

Course language:

EN - english

Notes:

Course assessment

Total number of assessed students: 119

А	В	С	D	Е	FX
30.25	14.29	16.81	15.13	19.33	4.2

Provides: doc. RNDr. Jozef Strečka, PhD.

Date of last modification: 19.09.2021

Faculty Faculty of S				
Faculty: Faculty of Science				
Course ID: ÚTVŠ/ ÚTVŠ/CM/13	Course name: Seaside Aerobic Exercise			
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	ce rse-load (hours): Idy period: 28			
Number of ECTS cr	redits: 2			
Recommended seme	ester/trimester of the course:			
Course level: I., II.				
Prerequisities:				
- active participation	sful course completion: in line with the study rule of procedure and course guidelines ice of all tasks- aerobics, water exercise, yoga, Pilates and others			
course syllabus and r Performance standard Upon completion of r - perform basic aerob - conduct verbal and	rates relevant knowledge and skills in the field, which content is defined in the recommended literature. d: the course students are able to meet the performance standard and: bics steps and basics of health exercises, non-verbal communication with clients during exercise, ge the process of physical recreation in leisure time			
Brief outline of the c				

 ŽECHOVSKÁ, I., MILEROVÁ, H., NOVOTNÁ, V. Aqua-fitness. Praha: Grada. 136 s. EVANS, M., HUDSON, J., TUCKER, P. 2001. Umění harmonie: meditace, jóga, tai-či, strečink. 192 s. JARKOVSKÁ, H., JARKOVSKÁ, M. 2005. Posilováni s vlastním tělem 417 krát jinak. Praha: Grada. 209 s. KOVAŘÍKOVÁ, K. 2017. Aerobik a fitness. Karolium, 130 s. 			
Course language: Slovak language			
Notes:			
Course assessment Total number of assessed students: 54			
abs	n		
11.11 88.89			
Provides: Mgr. Agata Dorota Horbacz, PhD.			
Date of last modification: 29.03.2022			
Approved: doc. Mgr. Daniel Jancura, PhD.			

University: P. J. Šafárik University in Košice		
Faculty: Faculty of Science		
Course ID: ÚFV/ SPBFb1/18	Course name: Semestral thesis I	
Course type, scope and the method: Course type: Recommended course-load (hours): Per week: Per study period: Course method: present Number of ECTS credits: 4		
Recommended semester/trimester of the course: 5.		
Course level: I.		
Prerequisities:		
tasks set by the pro required level. The	rse completion: ng the course, requires the student to demonstrate adequate level of the assigned ject leader at the beginning of the semester to the required extent and at the assignments are formulated by the teacher at the beginning of the semester, usually the supervisor of the final thesis. Tasks include e.g. study of literature	

in the field, mastering the operation of experimental equipment, sample preparation technology, preparation and implementation of the experiment, processing of the obtained data, or collaborating during the preparation of a scientific publication. Credit evaluation takes into account the time requirements of the student when working on a semester project in the range of 50 hours per semester. Individual activities of the student are evaluated by the project leader, the overall work of the student is evaluated by points on a point scale of 0 - 100 points. The minimum threshold for obtaining a rating is 50% of the rating scale, which is determined as follows: A 100-91% B 90-81% C 80-71% D 70-61% E 60-50% Fx 49-0%.

Learning outcomes:

After completing the course, the student will acquire knowledge and skills associated with scientific work in the field of biophysics. By actively participating in individual research teams, students will extend their knowledge in the relevant part of biophysics, acquire experimental skills in operating contemporary scientific equipment, study of the literature will improve their language skills. Data processing resp. the creation of original software will improve their computer skills.

Brief outline of the course:

Program for semestral project is prepared individually for each student by supervisor of the project at the beginning of each semester and can be focused on search in literature for a selected area of research, preparation of experiment and its performing, creation of software for data acquisition and analysis, collaboration during preparation of manuscript, presentation of the obtained results for department audience. Supervisor of the project will specify the topic of the project.

Recommended literature:

Selected scientific journals and books.

Course language:

Notes:

Subject Semester work I is realized in attendance taught online using software MS Teams, which a adverse conditions and also allows to meet the re	allows to maintain contact with students even in			
Course assessment				
Total number of assessed students: 5				
abs	n			
100.0	0.0			
Provides:	·			
Date of last modification: 25.02.2022				

University: P. J. Šafárik University in Košice Faculty: Faculty of Science		
Course type, scope and the method: Course type: Recommended course-load (hours): Per week: Per study period: Course method: present Number of ECTS credits: 6		
Recommended semester/trimester of the course: 6.		
Course level: I.		
Prerequisities:		
-	rse completion: ng the course, requires the student to demonstrate adequate level of the assigned ject leader at the beginning of the semester to the required extent and at the	

tasks set by the project leader at the beginning of the semester to the required extent and at the required level. The assignments are formulated by the teacher at the beginning of the semester, the project leader is usually the supervisor of the final thesis. Tasks include e.g. study of literature in the field, mastering the operation of experimental equipment, sample preparation technology, preparation and implementation of the experiment, processing of the obtained data, or collaborating during the preparation of a scientific publication. Credit evaluation takes into account the time requirements of the student when working on a semester project in the range of 50 hours per semester. Individual activities of the student are evaluated by the project leader, the overall work of the student is evaluated by points on a point scale of 0 - 100 points. The minimum threshold for obtaining a rating is 50% of the rating scale, which is determined as follows: A 100-91% B 90-81% C 80-71% D 70-61% E 60-50% Fx 49-0%.

Learning outcomes:

After completing the course, the student will acquire knowledge and skills associated with scientific work in the field of biophysics. By actively participating in individual research teams, students will extend their knowledge in the relevant part of biophysics, acquire experimental skills in operating contemporary scientific equipment, study of the literature will improve their language skills. Data processing resp. the creation of original software will improve their computer skills.

Brief outline of the course:

Program for semestral project is prepared individually for each student by supervisor of the project at the beginning of each semester and can be focused on search in literature for a selected area of research, preparation of experiment and its performing, creation of software for data acquisition and analysis, collaboration during preparation of manuscript, presentation of the obtained results for department audience. Supervisor of the project will specify the topic of the project.

Recommended literature:

Selected scientific journals and books.

Course language:

Notes:

Subject Semester work I is realized in attendance form. If necessary (e.g. Covid pandemic) it is taught online using software MS Teams, which allows to maintain contact with students even in adverse conditions and also allows to meet the requirements of the subject.					
Course assessment Total number of assessed students: 5					
abs	n				
100.0	0.0				
Provides:	·				
Date of last modification: 30.03.2022					

University: P. J. Šafárik University in Košice		
Faculty: Faculty of Science		
Course ID: ÚTVŠ/ TVa/11	Course name: Sports Activities I.	
Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: present		
Number of ECTS credits: 2		
Recommended semester/trimester of the course: 1.		
Course level: I., I.II., II.		
Prerequisities:		

Conditions for course completion:

Min. 80% of active participation in classes.

Learning outcomes:

Sports activities in all their forms prepare university students for their professional and personal life. They have a great impact on physical fitness and performance. Specialization in sports activities enables students to strengthen their relationship towards the selected sport in which they also improve.

Brief outline of the course:

Brief outline of the course:

Within the optional subject, the Institute of Physical Education and Sports of Pavol Jozef Šafárik University provides for students the following sports activities: aerobics, aikido, basketball, badminton, body form, bouldering, floorball, yoga, power yoga, pilates, swimming, body-building, indoor football, S-M systems, step aerobics, table tennis, tennis, volleyball and chess.

In the first two semesters of the first level of education students will master basic characteristics and particularities of individual sports, motor skills, game activities, they will improve level of their physical condition, coordination abilities, physical performance, and motor performance fitness. Last but not least, the important role of sports activities is to eliminate swimming illiteracy and by means of a special program of medical physical education to influence and mitigate unfitness. In addition to these sports, the Institute offers for those who are interested winter and summer physical education trainings with an attractive program and organises various competitions, either at the premises of the faculty or University or competitions with national or international participation.

Recommended literature:

BENCE, M. et al. 2005. Plávanie. Banská Bystrica: FHV UMB. 198s. ISBN 80-8083-140-8. [online] Dostupné na: https://www.ff.umb.sk/app/cmsFile.php?disposition=a&ID=571 BUZKOVÁ, K. 2006. Fitness jóga, harmonické cvičení těla I duše. Praha: Grada. ISBN 8024715252.

JARKOVSKÁ, H, JARKOVSKÁ, M. 2005. Posilování s vlastním tělem 417 krát jinak. Praha: Grada. ISBN 9788024757308.

KAČÁNI, L. 2002. Futbal:Tréning hrou. Bratislava: Peter Mačura – PEEM. 278s. ISBN 8089197027.

KRESTA, J. 2009. Futsal.Praha: Grada Publishing, a.s. 112s. ISBN 9788024725345. LAWRENCE, G. 2019. Power jóga nejen pro sportovce. Brno: CPress. ISBN 9788026427902. SNER, Wolfgang. 2004. Posilování ve fitness. České Budějovice: Kopp. ISBN 8072322141. STACKEOVÁ, D. 2014. Fitness programy z pohledu kinantropologie. Praha: Galén. ISBN 9788074921155.

VOMÁČKO, S. BOŠTÍKOVÁ, S. 2003. Lezení na umělých stěnách. Praha: Grada. 129s. ISBN 8024721743.

Course language:

Slovak language

Notes:

Course assessment

Total number of assessed students: 14548

abs	abs-A	abs-B	abs-C	abs-D	abs-E	n	neabs
86.46	0.07	0.0	0.0	0.0	0.05	8.41	5.02

Provides: Mgr. Agata Dorota Horbacz, PhD., Mgr. Dávid Kaško, PhD., Mgr. Zuzana Küchelová, PhD., doc. PaedDr. Ivan Uher, PhD., MPH, prof. RNDr. Stanislav Vokál, DrSc., Mgr. Marcel Čurgali, Mgr. Patrik Berta, Mgr. Ladislav Kručanica, PhD., Mgr. Richard Melichar, Mgr. Petra Tomková, PhD., MUDr. Peter Dombrovský

Date of last modification: 29.03.2022

	COURSE INFORMATION LETTER				
University: P. J. Šafá	irik University in Košice				
Faculty: Faculty of S	beience				
Course ID: ÚTVŠ/ Course name: Sports Activities II. TVb/11					
Course type, scope a Course type: Practi Recommended cou Per week: 2 Per stu Course method: pro	ce rse-load (hours): ıdy period: 28				
Number of ECTS cr	redits: 2				
Recommended seme	ester/trimester of the course: 2.				
Course level: I., I.II.,	, II.				
Prerequisities:					
Conditions for course active participation in	se completion: n classes - min. 80%.				
They have a great in	I their forms prepare university students for their professional and personal life. npact on physical fitness and performance. Specialization in sports activities strengthen their relationship towards the selected sport in which they also				
University provides badminton, body form indoor football, S-M In the first two seme and particularities of physical condition, of Last but not least, the means of a special pr In addition to these physical education tra the premises of the fa	subject, the Institute of Physical Education and Sports of Pavol Jozef Šafárik for students the following sports activities: aerobics, aikido, basketball, m, bouldering, floorball, yoga, power yoga, pilates, swimming, body-building, systems, step aerobics, table tennis, tennis, volleyball and chess. esters of the first level of education students will master basic characteristics individual sports, motor skills, game activities, they will improve level of their coordination abilities, physical performance, and motor performance fitness. e important role of sports activities is to eliminate swimming illiteracy and by rogram of medical physical education to influence and mitigate unfitness. sports, the Institute offers for those who are interested winter and summer ainings with an attractive program and organises various competitions, either at culty or University or competitions with national or international participation.				
[online] Dostupné na	ature: 005. Plávanie. Banská Bystrica: FHV UMB. 198s. ISBN 80-8083-140-8. a: https://www.ff.umb.sk/app/cmsFile.php?disposition=a&ID=571 6. Fitness jóga, harmonické cvičení těla I duše. Praha: Grada. ISBN				

JARKOVSKÁ, H, JARKOVSKÁ, M. 2005. Posilování s vlastním tělem 417 krát jinak. Praha: Grada. ISBN 9788024757308.

KAČÁNI, L. 2002. Futbal:Tréning hrou. Bratislava: Peter Mačura – PEEM. 278s. ISBN 8089197027.

KRESTA, J. 2009. Futsal.Praha: Grada Publishing, a.s. 112s. ISBN 9788024725345.

LAWRENCE, G. 2019. Power jóga nejen pro sportovce. Brno: CPress. ISBN 9788026427902. SNER, Wolfgang. 2004. Posilování ve fitness. České Budějovice: Kopp. ISBN 8072322141. STACKEOVÁ, D. 2014. Fitness programy z pohledu kinantropologie. Praha: Galén. ISBN 9788074921155.

VOMÁČKO, S. BOŠTÍKOVÁ, S. 2003. Lezení na umělých stěnách. Praha: Grada. 129s. ISBN 8024721743.

Course language:

Slovak language

Notes:

Course assessment

Total number of assessed students: 13211

abs	abs-A	abs-B	abs-C	abs-D	abs-E	n	neabs
84.35	0.51	0.02	0.0	0.0	0.05	10.78	4.29

Provides: Mgr. Agata Dorota Horbacz, PhD., Mgr. Dávid Kaško, PhD., Mgr. Zuzana Küchelová, PhD., doc. PaedDr. Ivan Uher, PhD., MPH, prof. RNDr. Stanislav Vokál, DrSc., Mgr. Marcel Čurgali, Mgr. Patrik Berta, Mgr. Ladislav Kručanica, PhD., Mgr. Richard Melichar, Mgr. Petra Tomková, PhD., MUDr. Peter Dombrovský

Date of last modification: 29.03.2022

-	árik University in Košice					
Faculty: Faculty of Science						
C ourse ID: ÚTVŠ/ ГVc/11	1					
Course type, scope a Course type: Practi Recommended cou Per week: 2 Per stu Course method: pro	ice irse-load (hours): udy period: 28					
Number of ECTS cr	redits: 2					
Recommended seme	ester/trimester of the course: 3.					
Course level: I., I.II.,	, II.					
Prerequisities:						
Conditions for cours min. 80% of active p Learning outcomes:	participation in classes					
Sports activities in all They have a great in	I their forms prepare university students for their professional and personal life mpact on physical fitness and performance. Specialization in sports activities strengthen their relationship towards the selected sport in which they also					
University provides badminton, body form indoor football, S-M In the first two seme and particularities of physical condition, c	course: subject, the Institute of Physical Education and Sports of Pavol Jozef Šafárik for students the following sports activities: aerobics, aikido, basketball m, bouldering, floorball, yoga, power yoga, pilates, swimming, body-building systems, step aerobics, table tennis, tennis, volleyball and chess. esters of the first level of education students will master basic characteristics individual sports, motor skills, game activities, they will improve level of their coordination abilities, physical performance, and motor performance fitness e important role of sports activities is to eliminate swimming illiteracy and by					

BUZKOVÁ, K. 2006. Fitness jóga, harmonické cvičení těla I duše. Praha: Grada. ISBN 8024715252.

JARKOVSKÁ, H, JARKOVSKÁ, M. 2005. Posilování s vlastním tělem 417 krát jinak. Praha: Grada. ISBN 9788024757308.

KAČÁNI, L. 2002. Futbal:Tréning hrou. Bratislava: Peter Mačura – PEEM. 278s. ISBN 8089197027.

KRESTA, J. 2009. Futsal.Praha: Grada Publishing, a.s. 112s. ISBN 9788024725345.

LAWRENCE, G. 2019. Power jóga nejen pro sportovce. Brno: CPress. ISBN 9788026427902. SNER, Wolfgang. 2004. Posilování ve fitness. České Budějovice: Kopp. ISBN 8072322141. STACKEOVÁ, D. 2014. Fitness programy z pohledu kinantropologie. Praha: Galén. ISBN 9788074921155.

VOMÁČKO, S. BOŠTÍKOVÁ, S. 2003. Lezení na umělých stěnách. Praha: Grada. 129s. ISBN 8024721743.

Course language:

Slovak language

Notes:

Course assessment

Total number of assessed students: 8879

abs	abs-A	abs-B	abs-C	abs-D	abs-E	n	neabs
88.62	0.07	0.01	0.0	0.0	0.02	4.25	7.03

Provides: Mgr. Marcel Čurgali, Mgr. Agata Dorota Horbacz, PhD., Mgr. Dávid Kaško, PhD., Mgr. Zuzana Küchelová, PhD., doc. PaedDr. Ivan Uher, PhD., MPH, prof. RNDr. Stanislav Vokál, DrSc., Mgr. Patrik Berta, Mgr. Ladislav Kručanica, PhD., Mgr. Richard Melichar, Mgr. Petra Tomková, PhD., MUDr. Peter Dombrovský

Date of last modification: 29.03.2022

University: P I Safe	ărik University in Košice
Faculty: Faculty of S	
Course ID: ÚTVŠ/ TVd/11	
Course type, scope a Course type: Pract Recommended cou Per week: 2 Per st Course method: pr Number of ECTS c	ice urse-load (hours): udy period: 28 resent
Recommended sem	ester/trimester of the course: 4.
Course level: I., I.II.	., II.
Prerequisities:	
Conditions for cour min. 80% of active p	rse completion: participation in classes
They have a great in	: Il their forms prepare university students for their professional and personal life mpact on physical fitness and performance. Specialization in sports activities strengthen their relationship towards the selected sport in which they also
University provides badminton, body for indoor football, S-M In the first two seme	course: subject, the Institute of Physical Education and Sports of Pavol Jozef Šafárik for students the following sports activities: aerobics, aikido, basketball m, bouldering, floorball, yoga, power yoga, pilates, swimming, body-building I systems, step aerobics, table tennis, tennis, volleyball and chess. esters of the first level of education students will master basic characteristics findividual sports, motor skills, game activities, they will improve level of their

[online] Dostupné na: https://www.ff.umb.sk/app/cmsFile.php?disposition=a&ID=571 BUZKOVÁ, K. 2006. Fitness jóga, harmonické cvičení těla I duše. Praha: Grada. ISBN 8024715252.

JARKOVSKÁ, H, JARKOVSKÁ, M. 2005. Posilování s vlastním tělem 417 krát jinak. Praha: Grada. ISBN 9788024757308.

KAČÁNI, L. 2002. Futbal:Tréning hrou. Bratislava: Peter Mačura – PEEM. 278s. ISBN 8089197027.

KRESTA, J. 2009. Futsal.Praha: Grada Publishing, a.s. 112s. ISBN 9788024725345.

LAWRENCE, G. 2019. Power jóga nejen pro sportovce. Brno: CPress. ISBN 9788026427902. SNER, Wolfgang. 2004. Posilování ve fitness. České Budějovice: Kopp. ISBN 8072322141. STACKEOVÁ, D. 2014. Fitness programy z pohledu kinantropologie. Praha: Galén. ISBN 9788074921155.

VOMÁČKO, S. BOŠTÍKOVÁ, S. 2003. Lezení na umělých stěnách. Praha: Grada. 129s. ISBN 8024721743.

Course language:

Slovak language

Notes:

Course assessment

Total number of assessed students: 5628

abs	abs-A	abs-B	abs-C	abs-D	abs-E	n	neabs
82.66	0.28	0.04	0.0	0.0	0.0	8.05	8.97

Provides: Mgr. Marcel Čurgali, Mgr. Agata Dorota Horbacz, PhD., Mgr. Dávid Kaško, PhD., Mgr. Zuzana Küchelová, PhD., doc. PaedDr. Ivan Uher, PhD., MPH, prof. RNDr. Stanislav Vokál, DrSc., Mgr. Patrik Berta, Mgr. Ladislav Kručanica, PhD., Mgr. Richard Melichar, Mgr. Petra Tomková, PhD., MUDr. Peter Dombrovský

Date of last modification: 29.03.2022

University: P. J. Šaf	University: P. J. Šafárik University in Košice				
Faculty: Faculty of Science					
Course ID: ÚFV/ Course name: Student Scientific Conference SVK/13					
Course type, scope Course type: Recommended cou Per week: Per stu Course method: p	urse-load (hours): dy period:				
Number of ECTS c	redits: 4				
Recommended sem	ester/trimester of the cour	se:			
Course level: I., II.					
Prerequisities:					
Conditions for cour	Conditions for course completion:				
Learning outcomes	:				
Brief outline of the	course:				
Recommended liter	ature:				
Course language:					
Notes:					
Course assessment Total number of ass	essed students: 18				
	abs n				
100.0 0.0					
Provides:					
Date of last modific	cation: 30.11.2021				
Approved: doc. Mg	r. Daniel Jancura, PhD.				

University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚFV/ DGS/21	Course name: Students` Digital Literacy
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	ce rse-load (hours): dy period: 28
Number of ECTS cro	edits: 2
Recommended seme	ster/trimester of the course: 1.
Course level: I.	
Prerequisities:	
 Practical ongoing a Active participation 	e completion: based on ongoing assessment: assignments and their defense (at least 50% needed) on during face-to-face contact learning in classical or virtual classroom (3 nd during online learning (no absence, uploading all individual ongoing
digital technologies (1. according to the cu	btain and know to apply basic knowledge and skills in working with current mobile phone, tablet, laptop, web technologies): rrent European framework for the Digital competence DigComp and ECDL e effective learning, work and active life in higher education, later lifelong areer prospects.
 modern web browset security, privacy, res 0305. Search, collect scanning, audio record digital notebooks (C evaluation of digital 0608. Editing and card cloud and interactive (text and spreadsheet work with pdf document (Kami, Google bookset 09 10. Organization modern LMS and cle (Google Classroom, Interaction) time management (C 	skills, DigComp framework, ECDL er and its personalization sponsible use of DT etion and evaluation of digital content ording and speech resolution, optical resolution (OCR) Google keep, Evernote, Onenote) resources (Google forms and sections) reating digital content e documents editors - Google, Microsoft, Jupyter) ments, e-books and videos 5, Screencasting) n, protection and sharing of digital content oud storage Microsoft team, Google Drive, Dropbox)

- collaborative interactive whiteboards (Jamboard, Whiteboard)

- online presentations and online meetings

(Google presentations, Powerpoint, Google meet, Microsoft teams)

Recommended literature:

1. Carretero Gomez, S., Vuorikari, R. and Punie, Y., DigComp 2.1: The Digital Competence Framework for Citizens with eight proficiency levels and examples of use, Luxembourg, 2017, ISBN 978-92-79-68006-9, https://www.ecdl.sk/

2. Bruff, D. (2019). Intentional Tech: Principles to Guide the Use of Educational Technology in College Teaching (1st edition). Morgantown: West Virginia University Press.

3. Baker, Y. (2020). Microsoft Teams for Education. Amazon Digital Services.

4. Miller, H. (2021). Google Classroom + Google Apps: 2021 Edition. Brentford: Orion Edition Limited.

Course language:

slovak

Not

Notes:					
Course assessment Total number of assessed students: 81					
A	В	С	D	E	FX
45.68	3.7	7.41	0.0	43.21	0.0
Provides: doc.	RNDr. Jozef Han	č, PhD.		· · ·	
Date of last modification: 26.01.2022					
Approved: doc	. Mgr. Daniel Jan	cura, PhD.			

University: P. J. Šafá	rik University in Košice					
Faculty: Faculty of S						
Course ID: ÚTVŠ/ Course name: Summer Course-Rafting of TISA River LKSp/13						
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	ce rse-load (hours): dy period: 28					
Number of ECTS cr	edits: 2					
Recommended seme	ster/trimester of the course:					
Course level: I., II.						
Prerequisities:						
- active participation	sful course completion: in line with the study rule of procedure and course guidelines ce of all tasks: carrying a canoe, entering and exiting a canoe, righting a canoe,					
course syllabus and r Performance standard Upon completion of t - implement the acqu - implement basic ski - determine the right	the course students are able to meet the performance standard and: ired knowledge in different situations and practice, ills to manipulate a canoe on a waterway,					
5. Canoe lifting and c	ourse: iculty of waterways iting ning using an empty canoe carrying n the water without a shore contact be out of the water					

11 Consisting					
11. Capsizing 12. Commands					
Recommended literature: 1. JUNGER, J. et al. Turistika a športy v prírode 8080680973. Internetové zdroje: 1. STEJSKAL, T. Vodná turistika. Prešov: PU v Dostupné na: https://ulozto.sk/tamhle/UkyxQ2IV ZGDjBGR2AQtkAzVkAzLkLJWuLwWxZ2ukH	Prešove. 1999. YF8qh/name/Nahrane-7-5-2021-v-14-46-39#!				
Course language: Slovak language					
Notes:					
Course assessment Total number of assessed students: 209	1				
abs	n				
37.32 62.68					
Provides: Mgr. Dávid Kaško, PhD.	1				
Date of last modification: 29.03.2022					
Approved: doc. Mgr. Daniel Jancura, PhD.					

University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚTVŠ/ KP/12	Course name: Survival Course
Course type, scope a Course type: Practic Recommended cou Per week: 2 Per stu Course method: pre	ce rse-load (hours): dy period: 28
Number of ECTS cr	
Recommended seme	ster/trimester of the course:
Course level: I., II.	
Prerequisities:	
- active participation	se completion: sful course completion: in line with the study rule of procedure and course guidelines, ce of all the tasks defined in the course syllabus
course syllabus and r Performance standard Upon completion of r - acquire knowledge - obtain theoretical kn connected with survi - be able to resist a environment, - be able implement children and youth w	the course students are able to meet the performance standard and should: about safe stay and movement in natural environment, nowledge and practical skills to solve extraordinary and demanding situations val and minimization of damage to health, nd face situations related to overcoming barriers and obstacles in natural the acquired knowledge as an instructor during summer sport camps for ithin recreational sport.
 Preparation and gu Objective and subj Principles of hygic Fire building Movement in the u Shelters Food preparation a Rappelling, Tyroliz 	ourse: uct and safety in the movement in unfamiliar natural environment idance of a hike tour ective danger in the mountains ene and prevention of damage to health in extreme conditions unfamiliar terrain, orientation and navigation and water filtering

Recommended literature:

1. JUNGER, J. et al. Turistika a športy v prírode. Prešov: Fakulta humanitných a prírodných vied PU v Prešove. 2002. 267s. ISBN 80-8068-097-3.

n

53.99

PAVLÍČEK, J. Člověk v drsné přírodě. 3. vyd. Praha: Práh. 2002. ISBN 8072520598.
 WISEMAN, J. SAS: příručka jak přežít. Praha: Svojtka & Co. 2004. 566s. ISBN 8072372807.

Course language:

Slovak language

Notes:

Course assessment

Total number of assessed students: 439

abs 46.01

Provides: Mgr. Ladislav Kručanica, PhD.

Date of last modification: 16.05.2023

	COURSE INFORMATION LETTER
University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚFV/ MSB/10	Course name: System Biology Modeling
Course type, scope a Course type: Lectur Recommended cour Per week: 2 / 0 Per Course method: pre	re / Practice rse-load (hours): study period: 28 / 0
Number of ECTS cr	edits: 3
Recommended seme	ester/trimester of the course: 5.
Course level: I.	
Prerequisities:	
Conditions for cours Solving intermediate Exam.	se completion: motivating challenges given at the lectures.
Learning outcomes: To provide an overvi field of systems biolo	iew of the computational techniques and achievable results in the emerging
and Anfinsen princip procedures and their Biological polymers Biological databases as an example of non Molecular interactio approaches. Stochas	modeling. Physical structure of biopolymers. Foldamers, Levinthal paradox le. Essentials of molecular modeling and molecular simulations. Examples of
ed. Chapman and Ha Campbell, A. Malcol Bioinformatics*. 2nd	duction to Systems Biology: Design Principles of Biological Circuits*. 1st
Course language:	

Course assessm Total number of	ent f assessed studen	ts: 215				
А	В	С	D	Е	FX	
91.16	6.51	1.86	0.47	0.0	0.0	
Provides: doc. 1	RNDr. Jozef Ulič	ný, CSc.		<u>.</u>		
Date of last mo	Date of last modification: 08.09.2021					
Approved: doc.	Mgr. Daniel Jan	cura, PhD.				

University: P. J. Šafa	árik University in Košice	
Faculty: Faculty of S	Science	
Course ID: ÚFV/ TME1/03	Course name: Theoretical Mechanics	
Course type, scope a Course type: Lectu Recommended cou Per week: 3 / 2 Per Course method: pr	are / Practice arse-load (hours): • study period: 42 / 28	
Number of ECTS c	redits: 6	
Recommended sem	ester/trimester of the course: 1., 3.	
Course level: I.		
Prerequisities: ÚFV	/VF1a/12	

Conditions for course completion:

To successfully complete the course, the student must demonstrate sufficient understanding of all basic concepts and applications of theoretical mechanics. Knowledge of basic concepts at the level of their mathematical definition is required, as well as their physical content and principled applications. The student must be able to actively master the content of the curriculum continuously during the semester, so that he can actively and creatively use the acquired knowledge in solving specific problems in exercises and independent homework. In addition to direct participation in teaching, the student is obliged to independently study professional topics assigned by the teacher and also to develop and present two home assignments. The condition for obtaining credits is, in addition to participation in teaching, also the successful completion of the 3rd written tests from exercises and lectures and the elaboration of home assignments. The minimum limit for passing the exam is to obtain 51% of the total score, which takes into account all required activities with relevant weight.

Rating scale: A - 91% -100% points, B - 81% -90% points, C - 71% -80% points, D - 61% -70% points, E - 51% -60% points.

Learning outcomes:

The lecture on Theoretical Mechanics is the first lecture of an extensive university course in theoretical physics, where the student gets acquainted with fundamental theoretical concepts (e.g., generalized coordinates, velocities and momentum, phase space, Hamiltonian Lagrangian ...), which constitute the basis for understanding advanced theoretical methods of advanced courses such as quantum mechanics, statistical physics and quantum field theory. For this reason, attending this lecture is essential for all physics students. In addition to deep physical knowledge, students will also gain a practical experience in solving complex problems of mechanics of systems of mass points, continuum mechanics, hydrostatics and hydrodynamics.

Brief outline of the course:

1. Dynamics of a free system of mass points.

2. Motion of a constrained system of mass points. Constrains and their classification. The principle of virtual work and search for equilibrium positions.

3. D'Alembert's principle. Lagrange equations of the first kind. Generalized coordinates and generalized forces.

4. Lagrange equations of the second kind and generalized potential.

5. Basic properties of Lagrange equations. First integrals of equations of motion: Integral of energy and generalized momentum.

6. Integral principles. Variation of functions and integrals. Hamilton's principle.

7. Hamilton's function. Hamilton's canonical equations.

8. Mechanics of a perfectly rigid body. Position of a rigid body in space, independent coordinates. The speed of the points of a rigid body.

9. Center of gravity, linear and angular momentums of a rigid body. Tensor of inertia.

10. Kinetic energy of a rigid body. Euler angles and Euler kinematic equations. Euler's equations of motion of a perfectly rigid body.

11. Basic concepts of continuum mechanics. Vector and tensor of deformation. Stress vector and stress tensor. Equilibrium conditions and equations of motion of a continuum. Generalized Hooke's law. Waves in an elastic environment.

12. Mechanics of fluids. Conditions of a hydrostatic equilibrium. Continuity equation. Euler's equations of motion of an ideal fluid.

Recommended literature:

1. Meirovitch L.: Methods of Analytical dynamics, McGraw-Hill, New York, 1970.

- 2. Taylor T.T.: Mechanics: Classical and Quantum, Pergamon Press, Oxford, 1976.
- 3. Strelkov S.P.: Mechanics, Mir Publishers, Moscow, 1985.
- 4. Greiner W.: Classical Mechanics, Springer-Verlag, Berlin, 2010.
- 5. Goldstein H.: Classical Mechanics, Addison-Wesley, London, 1970.

6. Barger V., Olsson M.: Classical Mechanics: A Modern Perspective, McGraw-Hill, London, 1973.

Course language:

- 1. Slovak,
- 2. English

Notes:

Course assessment

А	В	С	D	Е	FX
33.16	12.95	16.06	16.06	9.84	11.92

Provides: prof. RNDr. Michal Jaščur, CSc.

Date of last modification: 01.10.2021

University: P. J. Šaf	ărik University in Košice
Faculty: Faculty of	Science
Course ID: ÚFV/ TEP1/03	Course name: Theory of the Electromagnetic Field
Course type, scope Course type: Lectu Recommended cou Per week: 3 / 1 Per Course method: pr	ure / Practice urse-load (hours): r study period: 42 / 14
Number of ECTS c	redits: 5
Recommended sem	ester/trimester of the course: 4.
Course level. I	

Course level: I.

Prerequisities: ÚFV/VFM1b/15 or ÚFV/VF1b/03

Conditions for course completion:

To successfully complete the course, the student must demonstrate sufficient understanding of the basics terms, concepts and applications of electromagnetic field theory. Knowledge of basic concepts is required at the level of their mathematical definition, as well as their physical content and specific applications. During the semester, the student must continuously master the content of the curriculum so that he can actively and creatively use the acquired knowledge in solving specific tasks during the exercises and pass continuous written tests taken into account in the overall evaluation of the subject. The condition for obtaining credits is passing 2 continuous written tests in exercises and an oral exam, which consists of theoretical questions covering the entire scope of the course. The credit evaluation of the course takes into account the following student workload: direct teaching (2 credits), self-study (1 credit), individual consultations (1 credit) and assessment (1 credit). The minimum threshold for completing the course is to obtain at least 50% of the total score, using the following rating scale: A (90-100%), B (80-89%), C (70-79%), D (60- 69%), E (50-59%), F (0-49%).

Learning outcomes:

After completing lectures and exercises, the student will have sufficient physical skills, knowledge and mathematical apparatus enabling independent solution of a wide range scientific problems in electromagnetic field theory. The student also gets an overview of applications of electromagnetic field theory in various fields of physics such as electricity, magnetism, optics, etc.

Brief outline of the course:

1. Charge density and current density. Continuity equation. Definition of electromagnetic field.

2. System of Maxwell's equations in vacuum: differential formulation of Gauss' law of electrostatics, law of total current. The absence of magnetic monopoles and the law of electromagnetic induction.

3. Scalar and vector potential, gauge transformation. Wave equations for potentials. Energy conservation law in electromagnetic field theory: Poynting vector.

4. Conservation law of momentum of electromagnetic field: Maxwell's stress tensor.

5. Electrostatic field in vacuum and its potential. Potential of charges distributed in space and on surfaces. Boundary conditions on a charged area.

6. Multipole development of charge system potential. Electrostatic field energy. Electrostatic potential energy of a charge system and its multipole development in an external electric field.

7. Dielectric polarization. Vector of electrical induction, dielectric susceptibility and permittivity. Electrostatic field induced by a system of free charges in a dielectric, boundary conditions at the interface of two dielectrics.

8. Magnetic fields of stationary currents in vacuum; Biot-Savart law.

9. Stationary magnetic field of closed elementary current system, magnetic moment. Magnetization of magnets, magnets in the magnetic field of stationary currents.

10. Magnetic field strength, magnetic susceptibility and permeability. Magnetic field of a system of conductive currents in magnetics, boundary conditions at the interface of two magnets.

11. System of Maxwell's equations in the material environment and the conservation law of electromagnetic field energy. Quasi-stationary electromagnetic field.

12. Electromagnetic waves in homogeneous non-conductive medium, plane electromagnetic wave. Monochromatic plane wave and its polarization.

13. Refraction and reflection of a plane monochromatic wave at the interface of two media.

Recommended literature:

Kvasnica J.: Teorie elektromagnetického pole. Academia Praha, 1985.

Bobák A.: Teória elektromagnetického polľa, UPJŠ Košice, 2002.

Bobák A., Vargová E.: Zbierka riešených úloh z elektromagnetického poľa, UPJŠ Košice, 2001. Greiner W.: Classical Electrodynamics, Springer-Verlag, New York, 1998.

Course language:

1. Slovak,

2. English

Notes:

i tores:					
Course assessm	nent				
Total number of	f assessed studen	ts: 330			
А	В	С	D	Е	FX
26.97	8.79	18.18	21.21	16.67	8.18
Provides: doc. RNDr. Jozef Strečka, PhD., RNDr. Marek Semjan					
		2021			

Date of last modification: 19.09.2021

Faculty: Faculty		sity in Košice			
	of Science				
Course ID: ÚFV TDF1/99	Course n	ame: Thermodyn	amics and Statist	tical Physics	
Course type, sco Course type: Le Recommended Per week: 4 / 2 Course method	ecture / Practico course-load (h Per study peri	e 1ours):			
Number of ECT					
Recommended s	emester/trime	ster of the cours	e: 6.		
Course level: I.					
Prerequisities:					
Conditions for c	ourse complet	ion:			
Learning outcon	nes:				
Absolute tempe ensebles.Bose an	rature and er ad Fermi gases. andsberg,Thern E.M.Lifshitz,S	perature. The pric htropy. Phase spa nodynamics, Inter tatistical physics,	ce.Liouville the		natrix.Statistical
Absolute tempe ensebles.Bose an Literature: P.T.La L.D.Landau,and	rature and er ad Fermi gases. andsberg,Thern E.M.Lifshitz,S Oxford,1977.	ntropy.Phase spa	ce.Liouville the		natrix.Statistical
Absolute tempe ensebles.Bose an Literature: P.T.La L.D.Landau,and Pergamon Press,	rature and er ad Fermi gases. andsberg,Thern E.M.Lifshitz,S Oxford,1977. iterature:	ntropy.Phase spa	ce.Liouville the		natrix.Statistical
Absolute tempe ensebles.Bose an Literature: P.T.La L.D.Landau,and Pergamon Press, Recommended li	rature and er ad Fermi gases. andsberg,Thern E.M.Lifshitz,S Oxford,1977. iterature:	ntropy.Phase spa	ce.Liouville the		natrix.Statistical
Absolute tempe ensebles.Bose an Literature: P.T.La L.D.Landau,and Pergamon Press, Recommended la Course language	rature and er id Fermi gases. andsberg, Therm E.M.Lifshitz, S Oxford, 1977. iterature: e:	ntropy.Phase spa nodynamics,Inter tatisticalphysics,	ce.Liouville the		natrix.Statistical
Absolute tempe ensebles.Bose an Literature: P.T.La L.D.Landau,and Pergamon Press, Recommended li Course language Notes: Course assessme	rature and er id Fermi gases. andsberg, Therm E.M.Lifshitz, S Oxford, 1977. iterature: e:	ntropy.Phase spa nodynamics,Inter tatisticalphysics,	ce.Liouville the		natrix.Statistical
Absolute tempe ensebles.Bose an Literature: P.T.La L.D.Landau,and Pergamon Press, Recommended la Course language Notes: Course assessme Total number of a	rature and er ad Fermi gases. andsberg, Thern E.M.Lifshitz, S Oxford, 1977. iterature: e: ent assessed studer	ntropy.Phase spa nodynamics,Inter tatisticalphysics, nts: 167	ce.Liouville the science,1961.	orem.Density n	
Absolute tempe ensebles.Bose an Literature: P.T.La L.D.Landau,and Pergamon Press, Recommended li Course language Notes: Course assessme Total number of a A	rature and er d Fermi gases. andsberg, Therm E.M.Lifshitz, S Oxford, 1977. iterature: ent assessed studer B 16.17	ntropy.Phase spa nodynamics,Inter tatisticalphysics, nts: 167 C 19.76	ce.Liouville the science,1961.	orem.Density n	FX
Absolute tempe ensebles.Bose an Literature: P.T.La L.D.Landau,and Pergamon Press, Recommended li Course language Notes: Course assessme Total number of a A 55.69	rature and er d Fermi gases. andsberg, Therm E.M.Lifshitz, S Oxford, 1977. iterature: ent assessed studer B 16.17 NDr. Michal Ja	ntropy.Phase spa nodynamics,Inter tatisticalphysics, nts: 167 C 19.76 aščur, CSc.	ce.Liouville the science,1961.	orem.Density n	FX