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74. Thermodynamics and Statistical physics	

University: P. J. Šafá	rik University in Košice					
Faculty: Faculty of Science						
Course ID: CJP/ PFAJAKA/07	Course name: Academic English					
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	nd the method: ce rse-load (hours): dy period: 28 esent					
Number of ECTS cr	edits: 2					
Recommended seme	ster/trimester of the course:					
Course level: I.						
Prerequisities:						
Conditions for cours Active classroom par 1 test (13th week), no Presentation on chose Final evaluation- ave Grading scale: A 93-	ticipation, assignments handed in on time, 2 absences tolerated o retake. en topic rage assessment of test (50%), and presentation (50%). 100%, B 86-92%, C 79-85%, D 72-78%, E 65-71%, FX 64% and less					
Learning outcomes: The development of so of their linguistic cor syntactic aspects, dev for a given purpose, v	students' language skills - reading, writing, listening, speaking, improvement npetence - students acquire knowledge of selected phonological, lexical and relopment of pragmatic competence - students can effectively use the language with focus on Academic English, level B2.					
Brief outline of the c Formal and informal Academic English an Key academic verbs a Linking words in aca Word-formation - aff abstract Selected aspects of E Selected functional a paraphrasing	ourse: English Id its specific features and nouns demic writing, writing a paragraph, word-order, topic sentences ixation nglish pronunciation, academic vocabulary grammar structures - defining, classifying, epressing opinion, cause-effect,					
Recommended litera Seal B.: Academic En T. Armer :Cambridge M. McCarthy M., O' Zemach, D.E, Rumis Olsen, A. : Active Vo www.bbclearningeng Cambridge Academic	ncounters, CUP, 2002 English for Scientists, CUP 2011 Dell F Academic Vocabulary in Use, CUP 2008 ek, L.A: Academic Writing, Macmillan 2005 ocabulary, Pearson, 2013 lish.com c Content Dictionary, CUP, 2009					

Course language: English language, level B2 according to CEFR.						
Notes:						
Course assessment Total number of assessed students: 416						
А	В	С	D	Е	FX	
36.54	21.63	15.14	9.38	6.01	11.3	
Provides: Mgr. Viktória Mária Slovenská						
Date of last modification: 20.09.2023						
Approved: doc	. Mgr. Daniel Jan	cura, PhD.				

University: P. J	. Šafá	rik Univers	ity in Košice					
Faculty: Facult	Faculty: Faculty of Science							
Course ID: ÚM ALG4a/22	Irse ID: ÚMV/ G4a/22Course name: Algebra I for physicists							
Course type, sc Course type: 1 Recommended Per week: 2 / 2 Course metho Number of EC	Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 2 Per study period: 28 / 28 Course method: present							
Recommended	seme		ster of the course	e: 3.				
Course level: I.								
Prerequisities:								
Conditions for course completion: According to the results from the semester and in view of the results of the written and oral final exam.								
Learning outco To obtain basic to apply it in co	Learning outcomes: To obtain basic knowledge from linear algebra concerning systems of linear equations. To be able to apply it in concrete excercises.							
Brief outline of the course: Systems of linear equations, Gauss elimination. Maps, permutations. Computing with matrices. Determinants, Cramer rule.								
Recommended literature: T. Katriňák a kol.: Algebra a teoretická aritmetika 1, Alfa Bratislava, 1985. T.S Blyth, E.F. Robertson: Basic linear algebra, Springer Verlag, 2001. K. Jänich: Linear algebra, Springer Verlag, 1991.								
Course language: Slovak								
Notes:								
Course assessm Total number o	lent f asses	ssed studen	ts: 824					
А		В	С	D	E	FX		
11.17		13.11	20.51	18.93	27.06	9.22		
Provides: prof. Vodička	Provides: prof. RNDr. Danica Studenovská, CSc., RNDr. Lucia Janičková, PhD., Mgr. Martin Vodička							
Date of last mo	Date of last modification: 16.04.2022							
Approved: doc.	Mgr.	Daniel Jan	cura, PhD.					

University: P. J.	. Šafárik Univer	sity in Košice						
Faculty: Faculty of Science								
Course ID: ÚM ALG4b/22	V/ Course n	ame: Algebra II f	for physicists					
Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 2 Per study period: 28 / 28 Course method: present								
Number of EC	FS credits: 5							
Recommended	semester/trime	ster of the cours	e: 4.					
Course level: I.								
Prerequisities:	ÚMV/ALG4a/2	2						
Conditions for Exam	course complet	ion:						
Learning outco To provide deep	Learning outcomes: To provide deeper knowledge on vector spaces, linear transformations and Euclidean spaces.							
Brief outline of the course: Vector spaces, subspaces. A basis, a dimension and a characterization of n-dimensional vector spaces. The rank of a matrix. Linear transformations and their matrices. Operations with linear transformations, matrices of sums and compositions of linear transformations. Regular linear transformations, regular matrices. Similar matrices. Characteristic vectors and characteristic values of linear transformations. Affine spaces, subspaces and their positions. Euclidean spaces, the distance of subspaces. Conics and quadrics.								
Recommended G. Birkhoff, S. T. T. Katriňák a ko M. Sekanina, L. M. Hejný, V. Za J. Eliaš, J. Horv A. F. Beardon: A	literature: Mac Lane: A Su bl.: Algebra a teo Boček, M. Koč aťko, P. Kršňák: áth, J. Kajan: Zl Algebra and Geo	urvey of Modern A pretická aritmetik fandrle, J.Šedivý: Geometria 1, SPI bierka úloh z vyšš pmetry, Cambridg	Algebra, New Yo a 1, Alfa Bratisla Geometrie 1, SP N Bratislava 1983 dej matematiky 1, ge University Pres	rk 1965 va, 1985 N Praha 1986 5 , Alfa Bratislava ss, 2005				
Course languag Slovak	ge:							
Notes:				_				
Course assessm Total number of	ent f assessed studer	nts: 322						
А	В	C	D	Е	FX			
16.15	10.25	13.04	18.63	31.99	9.94			

Provides: doc. RNDr. Roman Soták, PhD., Mgr. Martin Vodička

Date of last modification: 16.04.2022

University: P. J. Šaf	árik University in Košice					
Faculty: Faculty of	Faculty: Faculty of Science					
Course ID: ÚFV/ ABE/18	Course name: Analysis of Biophysical Experiments					
Course type, scope Course type: Lectu Recommended cou Per week: 2 / 1 Per Course method: pr	and the method: are / Practice arse-load (hours): ar study period: 28 / 14 aresent redits: 4					
Recommended sem	ester/trimester of the course: 6.					
Course level: I.						
Prerequisities:						
Conditions for cour Oral exam where the	se completion: e students present theoretical knowledge of topics listed in the course syllabus.					
Learning outcomes Students will get an experiments, verific learn the basic tools	: n overview of the basic knowledge related to data evaluation in biophysical ation of hypotheses and discrimination between different models. Students will for experimental data computer processing.					
Brief outline of the 1. Simple experime knowing the uncert discrepancy, compar 2. Checking relation uncertainties in dire differences, product variable, experiment	course: Ental measurements, physical units, errors and uncertainties, importance of tainties, estimating uncertainties in repeatable experiments, best estimates, crison of measured and accepted values, comparison of two measured numbers. ships with a graph, fractional uncertainties, multiplying two measured numbers, ect measurements, the square-root rule for counting experiments, sums and s and quotients, independent uncertainties in a sum, arbitrary functions of one tal examples.					

3. Analysis of random uncertainties, random and systematic errors, the mean and standard deviation, the normal distribution, hystograms and distributions, limiting distributions, the standard deviation as 68% confidence limit, rejection of data, Chauvenet's criterion, weighted averages, experimental examples.

4. Least-squares fitting, linear data: the slope and the constant parameter, uncertainty in the measured data, experimental examples, least-squares fits to other curves e.g. polynomial or exponential functions, multiple regression, calibration curves in biophysics and biochemistry.

5. Covariance and correlation, covariance in error propagation, coefficient of linear correlation, quantitative significance r, autocorrelation, cross-correlation, use of correlation functions in monitoring the dynamics of individual molecules.

6. The binomial distribution, probabilities in dice throwing, definition of the binomial distribution, the Gauss distribution of random errors, testing of hypothesis, the properties of the Poisson distribution, applications, Chi squared testing, degrees of freedom and reduced chi squared, probabilities for chi squared, experimental examples, solutions, using Excel calculations.

7. Noise sources in biophysical experiments, mechanical noise, electrical noise (thermal noise, shot noise, interference), noise sources in optical imaging experiments, noise characterisctics: color, power spectrum, signal-to-noise ratio, methods for noise reduction and spectral filtration.

8. Computer processing of experimental data (Origin, Igor), the usage of fitting algorithms, statistical analysis, data plotting in graphs, 3D graphs, statistical graphs, figure preparation for publications.

9. Matlab/Octave: a tool for numerical modeling, complex data fitting with shared parameters, examples and applications.

10. Python: simple still complex tool for data analysis, large set of libraries, application examples: polynomial fitting, Fourier transformation, machine learning.

11. Data analysis in the field of spectral data, data smoothing (moving average, Savitzky-Golay filter, Fourier filter), background subtraction (high-order polynomial fitting, rolling ball algorithm, iterative methods), searching for peak position and intensity, complex spectra as a linear combination of simple contributions.

12. Image processing: using Python for image processing, imaging the relevant regions (ROI – region of interest) and firther analysis, binary thresholding, region separation by color, intensity normalization, border detection.

Recommended literature:

1. J.R. Taylor. An Introduction to Error Analysis: The Study of Uncertainties in Physical Measurements, University Science Books, 1997.

J. Mandel. The Statistical Analysis of Experimental Data, Dover Publications. 1964
 E.J. Billo. Excel for Chemist, Wiley, 2011

Course language:

Slovak, English.

Notes:

Course assessment

Total number of assessed students: 4

А	В	С	D	Е	FX
75.0	25.0	0.0	0.0	0.0	0.0

Provides: doc. Mgr. Gregor Bánó, PhD., doc. RNDr. Gabriel Žoldák, DrSc., Mgr. Andrej Hovan, PhD.

Date of last modification: 22.09.2021

University: P. J	. Šafárik Univer	sity in Košice			
Faculty: Facult	y of Science				
Course ID: ÚF BPO/14	FV/ Course name: Bachelor Thesis and its Defence				
Course type, so Course type: Recommende Per week: Pe Course metho	cope and the me d course-load (l r study period: bd: present	thod: nours):			
Number of EC	TS credits: 4				
Recommended	semester/trime	ster of the cours	e:		
Course level: I.					
Prerequisities:					
Conditions for Required numb	course complet er of credits gain	ion: 1ed basedon subn	nitting the bache	lor thesis.	
Learning outco	omes:				
Brief outline of Presentation of professional co	f the course: the bachelor the mmission.	esis results, answ	ering questions	of the reviewer	and members of
Recommended	literature:				
Course language Slovak or Engli	ge: ish				
Notes:					
Course assessn Total number o	1ent f assessed studer	nts: 62			
А	В	С	D	Е	FX
85.48	8.06	3.23	3.23	0.0	0.0
Provides:	·				
Date of last mo	dification: 07.1	2.2021			
Approved: doc	. Mgr. Daniel Jai	ncura, PhD.			

University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚCHV/ BAM1/00	Course name: Biochemical Analytical Methods
Course type, scope a Course type: Lectur Recommended cour Per week: 2 / 1 Per Course method: pre	nd the method: re / Practice rse-load (hours): study period: 28 / 14 esent
Number of ECTS cro	edits: 4
Recommended seme	ster/trimester of the course: 5.
Course level: I., II.	
Prerequisities:	
Conditions for cours Absence of a maximu Exam carried out in v	e completion: um of three exercises. vriting with at least 51% score.
Learning outcomes: The student will gain in analyzes in the bio	comprehensive information about the methods and approaches that are used chemical laboratory.
 Brief outline of the c 1. Introduction to ana 2. Processing and inte 3. The effectiveness reliability 4. Spectral methods f 5. Spectroscopy 6. Biosensors 7. Enzymes in bioana 8. Separation method 9. Electroanalytical n 10. Immunochemical 	ourse: lytical methods in biochemistry erpretation of results of the chosen system of methods to ensure the required level of analytical for determination of biomacromolecules lytical chemistry s nethods methods
Recommended litera D. J. Holme, H. Peck S. R. Mikkelsen, E. C V. A. Gault, N. H. Mo applications, 2009	ture: : Analytical Biochemistry, 1998 Cortón: Bioanalytical Chemistry, 2004 cClenaghan: Understanding Bioanalytical Chemistry: Principles and
Course language: Slovak, English	

Notes:

Teaching is carried out in person or, if necessary, remotely using the tool MS Teams, BigBlueButton, etc. The form of teaching is specified by the teacher at the beginning of the semester, updated continuously.

Course assessment Total number of assessed students: 91						
A B C D E FX						
31.87	20.88	19.78	20.88	6.59	0.0	
Provides: doc. RNDr. Rastislav Varhač, PhD.						
Date of last modification: 16.11.2021						
Approved: doc	Approved: doc. Mgr. Daniel Jancura, PhD.					

University: P. J. Safári	k Universitv	in Košice
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Faculty: Faculty of Science

Course ID: ÚCHV/	Course name: Biochemistry Practical
PBC2/99	

Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 4 Per study period: 56

Course method: present

Number of ECTS credits: 4

Recommended semester/trimester of the course: 3.

Course level: I.

Prerequisities:

Conditions for course completion:

Active participation with a maximum of one excused absence without the need for compensation. In case of excused absence from two or more practical exercises (e.g. due to illness), the student agrees with the teacher on alternative dates for practice.

Correctly prepared protocols from all completed tasks.

At least 51% of points from each of the written tests.

Learning outcomes:

To allow students to get practical experience in experimental techniques and methods, currently used in a biochemical research: pipetting, titration, UV/VIS spectrophotometry, thin layer chromatography (TLC), gel electrophoresis, isolation of macromolecules and substances from biological materials and their quantitative and qualitative determination.

Brief outline of the course:

1. Biochemistry laboratory safety rules. Basic biochemical laboratory procedures.

- 2. Qualitative tests for amino acids and proteins.
- 3. Isolation of casein from milk. Determination of protein concentration by Lowry method.

4. Determination of the iodine number by Yasud method . Soap production. Reactions with soap. Oxidation of unsaturated fatty acids.

5. Saponification number of fats and oils. Qualitative test for cholesterol: Salkowsky reaction.

6. Qualitative tests for carbohydrates. Determination of reducing carbohydrates by the Schoorl's method.

7. Determination of reducing and nonreducing carbohydrates in germinant plants.

8. Time-dependent course of enzyme-catalyzed reaction: digestion of gelatin by trypsine.

9. Determination of catalase activity and the first order rate constant. Effect of pH on alpha-amylase activity.

10. Effect of substrate concentration on initial rate of reaction, determination of Km and Vmax for urease-catalyzed hydrolysis of urea.

11. Isolation of DNA from spleen. Isolation of RNA from yeast. Qualitative tests for DNA and RNA components.

12. Determination of vitamin C concentration by 2,4-dinitrofenylhydrazine. Determination of vitamins A, B1, and C.

13. Final evaluation of students.

Recommended literature:

Sedlák, Varhač, Danko, Paulíková, Podhradský: Praktické cvičenia z biochémie, 2020, https://unibook.upjs.sk/sk/chemia/1411-prakticke-cvicenia-z-biochemie

Course language:

Slovak

Notes:

Teaching is carried out in person.

Course assessment

Total number of assessed students: 945

А	В	С	D	Е	FX
57.78	25.82	10.16	4.44	1.59	0.21

Provides: prof. RNDr. Mária Kožurková, CSc., RNDr. Nataša Tomášková, PhD., doc. RNDr. Rastislav Varhač, PhD., RNDr. Danica Sabolová, PhD., univerzitná docentka

Date of last modification: 17.08.2022

University: P. J.	. Šafárik Univers	ity in Košice				
Faculty: Faculty	Faculty: Faculty of Science					
Course ID: ÚF BCHF1/18	Course ID: ÚFV/ Course name: Biochemistry for Physicists I CHF1/18					
Course type, sc Course type: I Recommended Per week: 3 / 2 Course metho	ope and the met Lecture / Practice d course-load (h 2 Per study peri d: present	thod: ours): od: 42 / 28				
Number of EC	FS credits: 6					
Recommended	semester/trimes	ster of the cours	e: 2.			
Course level: I.						
Prerequisities:						
Conditions for	course completi	on:				
Learning outco	mes:					
Brief outline of	the course:					
Recommended	literature:					
Course languag	ge:					
Notes:						
Course assessm Total number of	Course assessment Total number of assessed students: 12					
А	В	С	D	Е	FX	
33.33 16.67 16.67 8.33 25.0 0.0						
Provides: prof. RNDr. Erik Sedlák, DrSc., doc. RNDr. Gabriel Žoldák, DrSc.						
Date of last mo	dification: 27.09	9.2021				
Approved: doc.	Mgr. Daniel Jan	icura, PhD.		_		

University: P. J.	Šafárik Univers	ity in Košice				
Faculty: Faculty	y of Science					
Course ID: ÚF BCHF2/18	Course ID: ÚFV/ Course name: Biochemistry for Physicists II BCHF2/18					
Course type, sc Course type: I Recommended Per week: 3 / 2 Course metho	ope and the met Lecture / Practice I course-load (h 2 Per study peri d: present	thod: ours): od: 42 / 28				
Number of EC	FS credits: 6					
Recommended	semester/trimes	ster of the cours	e: 3.			
Course level: I.						
Prerequisities:	ÚFV/BCHF1/18					
Conditions for	course completi	on:				
Learning outco	mes:					
Brief outline of	the course:					
Recommended	literature:					
Course languag	ge:					
Notes:						
Course assessm Total number of	ent f assessed studen	ts: 11				
А	В	С	D	Е	FX	
27.27 18.18 27.27 18.18 9.09 0.0						
Provides: prof. RNDr. Erik Sedlák, DrSc., doc. RNDr. Gabriel Žoldák, DrSc.						
Date of last mo	dification: 16.12	2.2021				
Approved: doc.	Mgr. Daniel Jan	cura, PhD.				

University: P. J	. Šafárik Univers	sity in Košice					
Faculty: Facult	y of Science						
Course ID: ÚC BAC1/04	HV/ Course n	ame: Bioinorgan	ic Chemistry I				
Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14 Course method: present							
Number of EC	TS credits: 5						
Recommended	semester/trime	ster of the cours	e: 3.				
Course level: I.	, II.						
Prerequisities:	·						
Conditions for Test or seminar examination	course complet works	ion:					
Learning outco The basic know biocatalysis, mo metals in the en	mes: vledges about bi etals in biology a vironment.	ometal interactio and medicine, me	ns with biomolec etal-based drugs,	cules, biomateria toxic metals for	ls, biominerals, biosystems and		
Brief outline of Metalic and non elements, esse Oxygen carrier processes. Calc bioinorganic ch radiodiagnostic	Brief outline of the course: Metalic and non-metalic elements and their roles in biological systems (biometals, bulk biological elements, essential trace elements). Biocoordination compounds, bioligands. Biocatalyzers. Oxygen carriers and oxygen transport proteins. Photochemical process. Catalysis and regulation processes. Calcium biominerals and biomineralization.Toxic metals. Application of knowledge of bioinorganic chemistry in pharmacy, chemotherapy (e.g. platinum complexes in cancer therapy) radiadiagnestias minarel biotechnology acalegy and in other branches of life.						
 Recommended literature: 1. Shriver D. F., Atkins P. W., Overton T. L., Rourke J.P., Weller M.T., Amstrong F.A.: Shiver & Atkins. Inorganic Chemistry. Oxford University Press, Oxford 2006. 2. Kaim W., Schwederski B.: Bioinorganic Chemistry: Inorganic Elements in the Chemistry of Life. Wiley, Chichester 1998. 3. Wilkins P. C., Wilkins R. G.: Inorganic Chemistry in Biology. OCP, Oxford 1997. 							
Course language:							
Notes:							
Course assessment							
Total number of assessed students: 373							
A	D 27.(1		D 5.0	E	ГА 0.27		
42.36	27.61	19.03	5.9	4.83	0.27		
Provides: prof.	RNDr. Zuzana V	/argová, Ph.D.					

Date of last modification: 28.10.2021

University: P. J.	Šafárik Univer	sity in Košice				
Faculty: Faculty	of Science					
Course ID: ÚF BSIM1/14	Course ID: ÚFV/ Course name: Biomolecular Simulations BSIM1/14					
Course type, sc Course type: I Recommended Per week: 2 / 2 Course method	ope and the me Lecture / Practic l course-load (l 2 Per study per d: present	thod: e nours): iod: 28 / 28				
Number of EC	FS credits: 5					
Recommended	semester/trime	ster of the cours	e: 6.			
Course level: I.,	, II.					
Prerequisities:						
Conditions for Elaboration and programs on pro Q/A part.	course complet presentation of oject given at th	ion: the project on giv e exercises. Exan	ven actual subjec n. Might be subs	ct. Development o stituted by written	f own computer exam including	
Learning outco Introduction to	mes: actual problema	tics of biomolecu	lar simulations.			
Brief outline of Structural chara as flow of biolo mechanisms. Ex force fields an Carlo methods approaches. Co reactions, free approaches and	the course: cteristics of bio gical informatio xperimental me d methods of - algorithms and mputational ch energy evaluat heuristic approa	logical polymers. n. 3D-structure a thods of structur classical molecu d paralelization. allenges in biom ion, protein fold aches.	Foldamers. Cent nd function of for e determination lar dynamics. In <i>Ab initio</i> nolecular simular ding. Computat	ntral dogma of mo oldamers. Recent v and their limitat Molecular dynam > molecular dynar ations - simulatio ional complexity,	blecular biology view on enzyme ions. Empirical ics and Monte nics and hybrid ns of chemical , nontraditional	
Recommended	literature:					
Actual literature	e recommended	by lecturer.				
Course languag	ge:					
Notes:						
Course assessm Total number of	ent fassessed studer	nts: 56				
А	В	C	D	Е	FX	
76.79	76.79 7.14 12.5 1.79 1.79 0.0					
Provides: doc. I	RNDr. Jozef Uli	čný, CSc.				
Date of last mo	dification: 27.0	3.2020				
Approved: doc.	Mgr. Daniel Ja	ncura, PhD.				

University: P. J. Šafárik University in Košice					
Faculty: Faculty of Science					
Course ID: ÚFV/ BFSb1/18Course name: Biophysical Seminary I					
Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 1 Per study period: 14 Course method: present					
Number of ECTS credits: 1					
Recommended semester/trimester of the course: 3.					
Course level: I.					
Prerequisities:					
Conditions for course completion: Independent individual work on the thesis, active participation on seminars. Final diploma thesis.					
Learning outcomes: Completing this seminar, the students should be able to independently elaborate diploma thesis and in comprehensive way communicate the obtained results of their scientific work.					
Brief outline of the course: Seminar on selected topics from biophysical research and topics related to the final theses of the students.					
Recommended literature: The literature will be recommended by supervisors of the theses.					
Course language: English language					
Notes:					
Course assessment Total number of assessed students: 11					
A B C D E FX					
100.0 0.0 0.0 0.0 0.0					
Provides: doc. Mgr. Daniel Jancura, PhD.					
Date of last modification: 12.07.2022					
Approved: doc. Mgr. Daniel Jancura, PhD.					

University: P. J	University: P. J. Šafárik University in Košice								
Faculty: Facult	Faculty: Faculty of Science								
Course ID: ÚFV/ BFSb2/18Course name: Biophysical Seminary II									
Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 1 Per study period: 14 Course method: present									
Number of EC	TS credits: 1								
Recommended	semester/trime	ster of the cours	e: 4.						
Course level: I.									
Prerequisities:									
Conditions for	course complet	ion:							
Learning outco	omes:								
Brief outline of	the course:								
Recommended	literature:								
Course languag	ge:								
Notes:				=					
Course assessment Total number of assessed students: 6									
А	A B C D E FX								
100.0 0.0 0.0 0.0 0.0 0.0									
Provides: doc. Mgr. Daniel Jancura, PhD.									
Date of last modification: 12.07.2022									
Approved: doc.	. Mgr. Daniel Jar	ncura, PhD.			Approved: doc. Mgr. Daniel Jancura, PhD.				

University: P. J.	Šafárik Univers	sity in Košice				
Faculty: Faculty	of Science					
Course ID: ÚF BSSBF/18	V/ Course na	ame: Biophysics				
Course type, sc Course type: Recommended Per week: Per Course method	ope and the me I course-load (h • study period: d: present	thod: ours):				
Number of EC	FS credits: 4					
Recommended	semester/trime	ster of the cours	e:			
Course level: I.						
Prerequisities: ÚFV/EMBF2/18	ÚFV/MBF1/14 a 3 and ÚFV/EMB	and ÚFV/FCH1/ F3/18	02 and ÚFV/BF	B1/14 and ÚFV/E	EMBF1/18 and	
Conditions for	course completi	ion:				
Learning outco	mes:					
Brief outline of	the course:					
Recommended	literature:					
Course languag	ge:					
Notes:						
Course assessm Total number of	ent fassessed studen	its: 5				
A	В	С	D	Е	FX	
20.0	20.0 60.0 20.0 0.0 0.0 0.0					
Provides:			l			
Date of last mo	dification: 15.12	2.2021				
Approved: doc.	Mgr. Daniel Jar	ncura, PhD.				

University: P. J. Šaf	árik University in Košice
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Faculty: Faculty of Science

Course ID: ÚFV/	Course name: Biophysics in Biomedicine and Biotechnologies
BFBB/18	

Course type, scope and the method: Course type: Lecture Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of ECTS credits: 3

Recommended semester/trimester of the course: 2.

Course level: I.

Prerequisities:

Conditions for course completion:

Elaboration of a written essay on a tpoic related to the application of biophysics in biomedicine and biotechnologies.

Learning outcomes:

This course will provide an opportunity for students to obtain knowledge about the utility of biophysics in biomedical research and the application in biotechnology progress. Completing this course, the students should be able to evaluate importance of biophysics for development of biomedical and biotechnology industry.

Brief outline of the course:

Week 1

Ion channels in personalized medicine. Basic knowledge about the structure and function of ion channels in cells. The relationship between the development of various diseases (cardiovascular, neurodegenerative, tumor) and ion channels. Ion channels as tools for drug development. Ion channels as targets for targeted therapy.

Week 2

Trans-disciplinary research - workflow for the identification of antivirals (high-throughput Xray crystallography of complexes, modeling, epitopes, in vitro essays, and the role of XBI). MicroCT with ROI for the study of the pathology of Covid-19-induced lung neovascularization. Experimental monitoring of non-equilibrium structural dynamics - photosynthetic reaction centers, opsins, photolysis of water for hydrogen energy.

Week 3

Application of NiR photobiostimulation in the therapy of neurodegenerative diseases - the principle of photobiostimulation, cell chromophores, application of NiR photobiostimulation in the treatment of Alzheimer's disease and Parkinson's disease

Week 4

In vitro evolution of proteins. Evolutionary techniques in the development of proteins / enzymes with new properties - interconnection of the methods of biophysics, bioinformatics, biochemistry and molecular biology. Design of mutations by methods of bioinformatics and molecular biology, selection and evolution by methods of molecular biology, analysis of properties - solubility, stability,

activity - of newly developed proteins by the methods of biochemistry and biophysics. Importance and application of proteins / enzymes with improved properties. Week 5

Carcinogenesis as an evolutionary process. According to current knowledge, cancer is the result of somatic mutations and epigenetic changes (epimutations), by which cells acquire new properties affecting the ratio between proliferation and apoptosis of individual cells, represented by the so-called the fitness of individual cells depending on the particular environment, which creates a selection pressure on the cells. Because cell properties are inherited during replication, all prerequisites for evolution are met and carcinogenesis can be studied and modeled as an evolutionary process.

Week 6

Lab-on-chip technology. Microfluidic systems. Light-controlled microrobots. Two-photon polymerization of microstructures. Construction of experimental facilities. Time-resolved laser spectroscopy of photosensitive drugs (phosphorescence, fluorescence, transient absorption). Singlet oxygen detection.

Week 7

Biophysical methods for the study of monoclonal antibodies and other therapeutic proteins. Biological function of antibodies, primary and secondary immune response, classes and subclasses of antibodies, role of individual types of antibodies, primary, secondary, tertiary and quaternary structure of antibodies, examples of therapeutic antibodies: trastuzumab and adalimumab. Preparation of antibodies, overview of monitored physicochemical properties of antibodies, critical quality attributes.

Week 8

Application of bioinformatics in biomedicine. The development of new technologies has provided us with an astonishing amount of information about the human gene, including individual differences. In addition to the natural need for database storage of information obtained in this way, there exists also a need to conceptualize procedures, analyzes and data processing in order to understand, model and address complex challenges such as pharmacogenetics and personalized medicine.

Week 9

Mitochondria in diseases and aging. The force of proton motion and the origin of life. Importance of cellular respiration for multicellular organisms. Mitochondria and the origin of the complexity of life. Mitochondrial aging theory. Mitochondria and diseases - origin, course and therapy. Week 10

Optical imaging, diagnostics and therapy in medicine. Modern trends in optical imaging methods (fluorescence and Raman imaging). Diagnosis and therapy (interaction of light with biological tissues, selected methods of medical optical diagnostics and therapy).

Weeks 11 and 12

Individual study of the selected texts about the applications of biophysics in biomedicine and biotechnologies.

Recommended literature:

1. E. Schrödinger. What is life? Cambridge University Press, 1992.

2. T. Hülswitt a R. Brinzanik. Budeme žít věčne? Kniha Zlín , 2012

3. J. Dowsett, P.A. Kenny a R.E. Johnston. The physics of diagnostic imaging. Hodder Arnold, 2006.

4. M.A. Hamblin a P. Mroz. Advances in photodynamic therapy. Artech House, 2008.

5. Súbor aktuálnych vedeckých publikácií

Course language:

English languag	ge						
Notes:							
Course assessment Total number of assessed students: 12							
А	B C D E FX						
91.67	8.33	0.0	0.0	0.0	0.0		
Provides: doc. RNDr. Katarína Štroffeková, PhD., doc. RNDr. Jozef Uličný, CSc., doc. RNDr. Gabriel Žoldák, DrSc., doc. Mgr. Daniel Jancura, PhD., RNDr. Branislav Brutovský, CSc., doc. Mgr. Gregor Bánó, PhD., RNDr. Gabriela Fabriciová, PhD., prof. RNDr. Erik Sedlák, DrSc.							
Date of last mo	dification: 12.07	7.2022					
Approved: doc.	. Mgr. Daniel Jan	cura, PhD.					

University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚBEV/ BS1/03	Course name: Biostatistics
Course type, scope a Course type: Lectu Recommended cou Per week: 2 / 2 Per Course method: pro	ind the method: re / Practice rse-load (hours): study period: 28 / 28 esent
Number of ECTS cr	redits: 6
Recommended seme	ester/trimester of the course: 3., 5.
Course level: I.	
Prerequisities:	
Conditions for cours Active participation of Passing the continua To absolve the final	se completion: on practicals, including successful solving of the assigned numerical examples. I testing. written test with at least 50% of the maximal score.
Learning outcomes: To provide the studer their scope of applic of the design of expe	nts with knowledge on basic principles of statistic methods used in biology and ation in statistical evaluation of experimental results, and with the principles priments, as well.
 Brief outline of the of 1. Sources and theored 2.Basic principles of and variability of dat 3. Theoretical and en 4. Reliability of estimute 5. Statistical sampline 6. One-way and multi 7. Regression analys 8. Correlations. 9. Non-parametrical 10. Design and planmetrical 11. Aanalysis of time 12. Analysis of quali 13. One- and multidi 	 course: etical background of biostatistics. the probability theory. Descriptive statistics: variables, measures of mean value a. npirical distributions. Experimental sampling from the normal distribution. nations. Testing of hypotheses. I and IItype errors. g. Comparison of two groups. tiple analysis of variance. Tests for multiple comparisons. is. methods. ning of biological experiments. e series. tative data. mensional methods, use of computer software.
Recommended litera Hassard, T. H.: Unde Snedecor,G.W., Cocl R.Forthofer, E.S.Lee Elsevier, Amsterdam	ature: rstanding biostatistics. Mosby Year Book, 1991 nran,W.G.: Statistical methods. The Iowa state university, Ames, 1972. , M.Hernandez: Biostatistics. A guide to design, analysis and dicovery. , 2007
Course language:	

Notes:							
Course assessment Total number of assessed students: 279							
А	В	С	D	Е	FX		
4.66	9.68	20.79	24.37	31.18	9.32		
Provides: prof. RNDr. Beňadik Šmajda, CSc.							
Date of last modification: 21.10.2021							
Approved: doc.	Approved: doc. Mgr. Daniel Jancura, PhD.						

University: P. J.	University: P. J. Šafárik University in Košice						
Faculty: Faculty of Science							
Course ID: ÚF BFB1/14	Course ID: ÚFV/ Course name: Cell Biophysics I BFB1/14						
Course type, scope and the method: Course type: Lecture Recommended course-load (hours): Per week: 3 Per study period: 42 Course method: present							
Number of EC	IS credits: 4						
Recommended	semester/trimes	ster of the cours	e: 5.				
Course level: I.	, II						
Prerequisities:							
Conditions for	course completi	on:					
Learning outco	mes:						
Brief outline of	the course:						
Recommended	literature:						
Course languag	ge:						
Notes:							
Course assessment Total number of assessed students: 31							
А	В	С	D	Е	FX		
45.16	45.16 22.58 12.9 19.35 0.0 0.0						
Provides: doc. RNDr. Katarína Štroffeková, PhD., RNDr. Gabriela Fabriciová, PhD.							
Date of last modification: 18.09.2023							
Approved: doc.	Mgr. Daniel Jan	icura, PhD.					

University: P. J. Šafárik University in Košice							
Faculty: Faculty of Science							
Course ID: CJF PFAJKKA/07	ourse ID: CJP/ Course name: Communicative Competence in English FAJKKA/07 FAJKKA/07						
Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: present							
Number of EC	FS credits: 2						
Recommended	semester/trimes	ster of the cours	e:				
Course level: I.							
Prerequisities:							
Conditions for Active participa two classes at th 2 credit tests (pr Final evaluation Final grade will FX 64 % and le Learning outco Brief outline of Recommended www.bbclearnin Štěpánek, Libon 2011. McCarthy M., C Fictumova J., C Principal, 2008. Peters S., Gráf	course completi ation in class and ne most. resumably in wea n consists of the s be calculated as ss. mes: the course: literature: ngenglish.com c a kol. Academic D'Dell F.: English deccarelli J., Long T.: Time to practi	on: completed home eks 6/7 and 12/13 scores obtained fo follows: A 93-10 c English-Akaden n Vocabulary in U g T.: Angličtina, 1 ise. Polyglot, 200	ework assignment 3) and an oral pro- or the 2 tests (50 0%, B 86-92%, of mická angličtina Jse, Upper-Intern konverzace pro p 07.	nts. Students are a esentation in Eng %) and the presen C 79-85%, D 72-7 Praha: Grada Pu mediate. CUP, 19 pokročilé. Barristo	allowed to miss lish. ntation (50%). 78%, E 65-71%, ublishing, a.s., 94. er and		
Additional stud	y materials.	mai Plactice. CO	P, 1985.				
Course language: English language, B2-C1 level according to CEFR							
Notes:							
Course assessment Total number of assessed students: 299							
А	В	С	D	E	FX		
45.48	20.74	17.39	7.69	6.02	2.68		
Provides: Mgr.	Ivana Kupková,	PhD.					

Date of last modification: 11.02.2024

University: P. J. Šafái	rik University in Košice
Faculty: Faculty of S	cience
Course ID: CJP/ PFAJGA/07	Course name: Communicative Grammar in English
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	nd the method: ce rse-load (hours): dy period: 28 esent
Number of ECTS cro	edits: 2
Recommended seme	ster/trimester of the course:
Course level: I.	
Prerequisities:	
Conditions for cours Active classroom part by given deadlines. Powerpoint presentat Final Test - end of ser Final assessment = av Grading scale: A 93-1	e completion: iccipation (maximum 2 absences tolerated), homework assignments completed ion of a topic related to the study field. mester, no retake verage of test and presentation. 100%, B 86-92%, C 79-85%, D 72-78%, E 65-71%, FX 64% and less
Learning outcomes: The development of s of their communica phonological, lexical efectively use the lan level B2.	students' language skills - reading, writing, listening, speaking, improvement ative linguistic competence. Students acquire knowledge of selected and syntactic aspects, development of pragmatic competence. Students can guage for a given purpose, with focus on Academic English and English on
Brief outline of the c Selected aspects of En Word formation Contrast of tenses in I The passive voice Types of Conditionals Phrasal verbs and Eng Words order and collo	ourse: nglish grammar and pronunciation English s glish idioms ocations, prepositional phrases
Recommended litera Vince M.: Macmillan McCarthy, O'Dell: Er www.linguahouse.con esllibrary.com bbclearningenglish.co ted.com/talks Course language:	ture: Grammar in Context, Macmillan, 2008 nglish Vocabulary in Use, CUP, 1994 m

English language, level B2 according to CEFR.							
Notes:							
Course assessment Total number of assessed students: 446							
А	В	С	D	Е	FX		
41.48	41.48 19.51 15.7 7.85 5.61 9.87						
Provides: Mgr. Lenka Klimčáková							
Date of last modification: 20.09.2023							
Approved: doc	. Mgr. Daniel Jar	cura, PhD.					

University: P. J. Šafárik University in Košice						
Faculty: Faculty of S	cience					
Course ID: KGER/ NJKG/07	Course name: Communicative Grammar in German Language					
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre Number of ECTS cre	nd the method: ce rse-load (hours): dy period: 28 esent edits: 2					

Recommended semester/trimester of the course:

Course level: I.

Prerequisities:

Conditions for course completion:

Active participation in class and completed homework assignments. Students are allowed to miss 2 classes at the most (2x90 min.). 2 control tests during the semester. Final grade will be calculated as follows: A 93-100 %, B 86-92%, C 79-85%, D 72-78%, E 65-71%, FX 64 % and less.

Learning outcomes:

The aim of the course is to identify and eliminate the most frequent grammatical errors in oral and written communication, learning language skills of listening comprehension, speaking, reading and writing, increasing students 'language competence (acquisition of selected phonological, lexical and syntactic knowledge), development of students' pragmatic competence (acquisition of the ability to express selected language functions), development of presentation skills, etc.

Brief outline of the course:

The course is aimed at practicing and consolidating knowledge of morphology and syntax of German in order to show the context in grammar as a whole. The course is intended for students who often make grammatical errors in oral as well as written communication. Through the analysis of texts, audio recordings, tests, grammar exercises, monologic and dialogical expressions of students focused on specific grammatical structures, problematic cases are solved individually and in groups. Emphasis is placed on the balanced development of grammatical thinking in the communication process, which ultimately contributes to the development of all four language skills.

Recommended literature:

Dreyer, H. – Schmitt, R.: Lehr- und Übungsbuch der deutschen Grammatik. Hueber Verlag GmbH & Co. Ismaning, 2009.

Krüger, M.: Motive Kursbuch, Lektion 1 – 30. Huebert Verlag GmbH & Co. Ismaning, 2020. Brill, L.M. – Techmer, M.: Deutsch. Großes Übungsbuch. Wortschatz. Huebert Verlag GmbH & Co. Ismaning, 2011.

Földeak, Hans: Sag's besser!. Grammatik. Arbeitsbuch für Fortgeschrittene. Huebert Verlag GmbH & Co. Ismaning, 2001.

Geiger, S. – Dinsel, S.: Deutsch Übungsbuch Grammatik A2-B2. Huebert Verlag GmbH & Co. Ismaning, 2018.

Dittelová, E. – Zavatčanová, M.: Einführung in das Studium der deutschen Fachsprache. Košice: ES UPJŠ, 2000.

Course language: German, Slovak language								
Notes:	Notes:							
Course assessn Total number o	nent f assessed studen	ts: 57						
А	В	С	D	Е	FX			
61.4	10.53 8.77 3.51 8.77 7.02							
Provides: Mgr.	Provides: Mgr. Ulrika Strömplová, PhD.							
Date of last modification: 12.07.2022								
Approved: doc	. Mgr. Daniel Jan	cura, PhD.						

University: P. J. Šafárik University in Košice					
Faculty: Faculty of S	Science				
Course ID: ÚFV/ POF1a/99	Course name: Computational Physics I				
Course type, scope a Course type: Lectu Recommended cou Per week: 2 / 1 Per Course method: pro	and the method: re / Practice rse-load (hours): study period: 28 / 14 esent				
Number of ECTS credits: 4					

Recommended semester/trimester of the course: 4., 6.

Course level: I.

Prerequisities: ÚFV/NUM/10

Conditions for course completion:

To successfully complete the course, the student must demonstrate a sufficient degree of understanding of the principles of computer solution of some typical physical problems. The basis of continuous assessment is participation and activity in exercises and work on assignments. The course ends with a final oral exam, the completion of which is conditional on the submission of all four assignments (projects) electronically and with the attached computer program. The credit evaluation of the course takes into account the following student workload: direct teaching (2 credits) and individual work on projects (2 credits). The minimum threshold for completing the course is to obtain at least 50% of the total score, using the following rating scale: A (90-100%), B (80-89%), C (70-79%), D (60- 69%), E (50-59%), F (0-49%).

Learning outcomes:

To teach the basic principles of computer solution of some typical physical problems. The course covers both the area of deterministic methods for solving problems by ordinary and partial differential equations as well as the area of stochastic Monte Carlo simulations and thus forms the basis for further study of more advanced computer methods contained in the follow-up course Computational Physics II.

Brief outline of the course:

- 1. Introduction to dynamical systems.
- 2. Numerical solution of systems of ordinary differential equations with initial condition.
- 3. Euler's method, convergence, error estimation and order of the method. One-step methods, Tylortype and Runge-Kuta (RK2, RK4) methods.
- 4. Multistep methods, general linear method (explicit, implicit). Methods based on numerical quadrature.
- 5. Boundary value problems for ordinary differential equations.
- 6. Numerical solution of partial differential equations (PDE). Difference methods, their consistence, convergence and stability. Elliptic PDE.
- 7. Parabolic PDE, diffusion equation. Explicit and implicit methods.

8. Introduction to the Monte Carlo method. Monte Carlo integration and application in statistical physics.

9. Basics of probability theory. Monte Carlo estimate of mean and standard deviation. Central theorem of Monte Carlo sampling.

10. Simple and importance sampling. Markov chain. Perron-Frobenius theorem. Metropolis algorithm, detailed balance condition.

11. Monte Carlo simulations of lattice spin systems - application to Ising model.

12. Statistical analysis of Monte Carlo data.

Recommended literature:

Basic literature:

POZRIKIDIS, C.: Num. Comp. in Science and Engineering, Oxford Univ. Press, 2008.

GARCIA A.L.: Numerical Methods for Physics, Prentice-Hall, 1994.

LANDAU D.P., BINDER K.: A Guide to Monte Carlo Simulations in Statistical Physics,

Cambridge Univ. Press, 5-th edition, 2021.

Other literature:

BERG, B.A.: Introduction to Markov Chain Monte Carlo Simulations and Their Statistical Analysis (http://www.worldscibooks.com/etextbook/5904/5904_intro.pdf)

JANKE, W.: Monte Carlo Simulations of Spin Systems (http://www.physik.uni-leipzig.de/~janke/ Paper/spinmc.pdf)

Course language:

Notes:

Course assessment

Total number of assessed students: 132

А	В	С	D	Е	FX	N	Р
29.55	18.18	12.88	15.15	17.42	2.27	0.0	4.55

Provides: prof. RNDr. Milan Žukovič, PhD.

Date of last modification: 14.09.2021
University: P. J. Šafărik University in Košice Faculty: Faculty of Science Course ID: ÚBEV/ Course name: Cytology CYT1/15 Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 3 / 2 Per study period: 42 / 28 Course method: present Number of ECTS credits: 6 Recommended semester/trimester of the course: 1. Course level: 1. Prerequisities: Conditions for course completion: Practicals graduation (without absence); Two written tests graduation (min. 70 % fruitfulness of each); Oral examination					
Faculty of Science Course ID: ÚBEV/ Course ID: ÚBEV/ Course name: Cytology CYT1/15 Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 3 / 2 Per study period: 42 / 28 Course method: present Number of ECTS credits: 6 Recommended semester/trimester of the course: 1. Course level: I. Prerequisities: Conditions for course completion: Practicals graduation (without absence); Two written tests graduation (min. 70 % fruitfulness of each); Oral examination	University: P. J. Šafárik University in Košice				
Course ID: ÚBEV/ CYT1/15 Course name: Cytology Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 3 / 2 Per study period: 42 / 28 Course method: present Number of ECTS credits: 6 Recommended semester/trimester of the course: 1. Course level: I. Prerequisities: Conditions for course completion: Practicals graduation (without absence); Two written tests graduation (min. 70 % fruitfulness of each); Oral examination	Faculty: Faculty of S	cience			
Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 3 / 2 Per study period: 42 / 28 Course method: present Number of ECTS credits: 6 Recommended semester/trimester of the course: 1. Course level: I. Prerequisities: Conditions for course completion: Practicals graduation (without absence); Two written tests graduation (min. 70 % fruitfulness of each); Oral examination	Course ID: ÚBEV/ CYT1/15	Course name: Cytology			
Number of ECTS credits: 6 Recommended semester/trimester of the course: 1. Course level: 1. Prerequisities: Conditions for course completion: Practicals graduation (without absence); Two written tests graduation (min. 70 % fruitfulness of each); Oral examination	Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 3 / 2 Per study period: 42 / 28 Course method: present				
Recommended semester/trimester of the course: 1. Course level: I. Prerequisities: Conditions for course completion: Practicals graduation (without absence); Two written tests graduation (min. 70 % fruitfulness of each); Oral examination	Number of ECTS credits: 6				
Course level: I. Prerequisities: Conditions for course completion: Practicals graduation (without absence); Two written tests graduation (min. 70 % fruitfulness of each); Oral examination	Recommended semester/trimester of the course: 1.				
Prerequisities: Conditions for course completion: Practicals graduation (without absence); Two written tests graduation (min. 70 % fruitfulness of each); Oral examination	Course level: I.				
Conditions for course completion: Practicals graduation (without absence); Two written tests graduation (min. 70 % fruitfulness of each); Oral examination	Prerequisities:				
	Conditions for cours Practicals graduation each); Oral examinat	se completion: (without absence); Two written tests graduation (min. 70 % fruitfulness of ion			

Learning outcomes:

To provide the students with knowledge of basic principles of cell microscopic and submicroscopic structure and function.

Brief outline of the course:

Lectures:

1.) Cell theory. Cell. 2.) Organization of living systems. 3.) Biological membranes. 4.) Transfer of substances across membranes. 5.) Cell wall of plant cells. 6.) Surface structures of cells. Extracellular matrix. Cell movement. 7.) Intercellular connections. 8.) Cytoskeleton. 9.) Cell nucleus. 10.) Mitochondria and cellular metabolism. 11.) Plastids and vacuoles. 12.) Ribosomes. Endoplasmic reticulum. Golgi apparatus. Lysosomes. 13.) Differentiation, aging and cell death, pathological changes in cells.

Exercises:

1.) Safety at work in a cytomorphological laboratory. Conditions for successful completion of exercises. 2.) Basics of optics. Origin and construction of the image with a magnifying glass and a microscope. 3.) Microscopic technique. 4.) Shape and size of cells. 5.) Principle of fluorescence and confocal microscopy. 6.) Control test. Vacuole. 7.) Cytoplasm movement. 8.) Nucleus and nucleolus. 9.) Cytoplasmic membrane. 10.) Osmotic processes. 11.) Cell inclusions. 12.) Cell walls of plant cells. 13.) Cell counting. Control test.

Recommended literature:

K.Kapeller, H.Strakele: Cytomorfológia. Osveta Martin, 1999

M.Babák, J.Šamaj: Cytológia. Univerzita Komenského Bratislava, 2002

Alberts B., Bray D., Johnson A., Lewis J.: Základy buněčné biologie. Espero Publishing, 2003 Campbell N. a Reece J.: Biologie. Computer Press, 2006

Kleban J., Mikeš J., Jendželovská Z., Jendželovský R., Fedoročko P.: Cytológia pracovný zošit na praktické cvičenia, 2018

Course language:

Notes:

10005.					
Course assessment Total number of assessed students: 1048					
А	В	С	D	Е	FX
12.98	19.75	28.82	20.8	16.6	1.05
Provides: doc. RNDr. Rastislav Jendželovský, PhD., RNDr. Zuzana Jendželovská, PhD., RNDr. Jana Vargová, PhD.					
Date of last modification: 19.02.2024					
Approved: doc. Mgr. Daniel Jancura, PhD.					

University: P. J. Šafá	rik University in Košice						
Faculty: Faculty of S	cience						
Course ID: CJP/ PFAJ4/07	Course name: English Language of Natural Science						
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: present						
Number of ECTS cr	edits: 2						
Recommended seme	ster/trimester of the course: 4.						
Course level: I.							
Prerequisities:							
Conditions for course Active participation is 2 classes at the most Continuous assessme 1 credit test taken pre 1 project (quiz on the 5 LMS quizzes (25% In order to be admitted assessment The exam test results represent the other 50 The final grade for the A 93-100, B 86-92, C Learning outcomes: Enhancement of stude in English for specifie Students obtain know English, improve their	n class and completed homework assignments. Students are allowed to miss nt: esumably in weeks 6/7 topic of the student's field of study) 25% of the continuous assessment of the continuous assessment) ed to the final exam, a student has to score at least 65 % from the continuous represent 50% of the final grade for the course, continuous assessment results 0% of the final grade. le course will be calculated as follows: 2 79-85, D 72-78, E 65-71, FX 64 and less.						
sciences.							
 Brief outline of the c Introduction to stude Selected aspects of Talking about acade Discussing science Defining scientific Expressing cause a Describing structure Explaining process Comparing objects 	ourse: dying language f scientific language lemic study terminology and concepts and effect res ses s, structures and concepts						

10. Talking about problem and solution

- 11. Referencing authors
- 12. Giving examples
- 13. Visual aids and numbers
- 14. Referencing time and place

Presentation topics related to students' study fields.

Recommended literature:

lms.upjs.sk - e-kurz Odborný anglický jazyk pre prírodné vedy.

Redman, S.: English Vocabulary in Use, Pre-intermetdiate, Intermediate. Cambridge University Press, 2003.

Armer, T.: Cambridge English for Scientists. CUP, 2011.

Wharton J.: Academic Encounters. The Natural World. CUP, 2009.

P. Fitzgerald : English for ICT studies. Garnet Publishing, 2011.

https://worldservice/learningenglish, https://spectator.sme.sk

www.isllibrary.com

linguahouse.com

Course language:

English, level B2 (CEFR)

Notes:

Course assessment

Total number of assessed students: 3075

А	В	С	D	Е	FX	
38.44	26.08	16.46	9.53	7.45	2.05	
Provides: Mgr. Viktória Mária Slovenská, Mgr. Lenka Klimčáková						
Date of last modification: 06.02.2024						

University: P. J. Šafá	rik University in Košice				
Faculty: Faculty of S	Faculty: Faculty of Science				
Course ID: ÚFV/ ZPU1/03	Course name: Essentials of UNIX Programming				
Course type, scope a Course type: Lectur Recommended cour Per week: 1 / 2 Per Course method: pre	nd the method: e / Practice rse-load (hours): study period: 14 / 28 esent				
Number of ECTS cr	edits: 4				
Recommended seme	ster/trimester of the course: 2.				
Course level: I.					
Prerequisities:					
Conditions for cours monitoring of studen unsupervised creation	e completion: t's programming skills to of the program to solve the given task at the end				
Learning outcomes: To provide students v applications of nume	with basic programming skills necessary for solving problems which require ric methods, simulation techniques and computer data processing.				
Brief outline of the c 1st week: Linux Basi (*,?). File ownership management. Manua 2nd week: C progran GCC Compiler. Forr Arithmetic operators. 3th week: Control f operators. Loops "wh logical operators. Con 4th week: Functions functions. User defin automatic variables. 5th week: Library fu functions (cos, sin, o (rint, round, floor, ce inclusion. Bit operator 6th week: Pointers an Functions for memo arguments. Formattee reading from file (fur 7th week: Summary.	 burse: cs: Characteristics. Linux distributions. UNIX/LINUX filesystem. Wildcards and permissions. Command line. Shell. Basic LINUX commands for file pages. ning language. Source code. C language syntax. Structure of C programs. natted output (printf). Declarations and types of variables. Operator sizeof. Assignement operators. Indexed variables (arrays). Text strings as arrays. low. Control structures. Statements and blocks. Increment and decrement ile", "for" and "do while". Break and continue statements. Relational and nditional expressions. Syntax of the "switch" statement. Declaration of function. Arguments of functions. Return of values by leed functions. Scope and lifetime of variables. Storage classes - static and unctions. Header files. Mathematical library (math.h). Basic mathematical exp, log). Generator of random numbers (function rand). Rounded values il). Symbolic constants. The C preprocessor: macro substitution, conditional rs. d addresses (&). Operator of dereferencing (*). Dynamic memory allocation. ry allocation and deallocation (malloc, calloc, free). Pointers and function d input (scanf). Structures and unions. Structure FILE. Formatted writing to/ netions fprintf, fcanf). 				

8th week: Basics of C++. OOP (Object oriented programming) paradigm. Data abstraction. Class, object. Data encapsulation. Member functions. Public and private parts of class. Difference between class and structures.

9th week: Constructor and destructor. Dynamic allocation and deallocation of memory by operators new and delete. Operator overloading. Polymorfism and inheritance.

10th week: Memory Layout of a Process in Linux. Monitoring of runing processes (commands ps, top). Filesystem /proc. Process priorities and scheduling. Signals. Running, stopping and ending processes in background. Ignoring hangup signal by command nohup. Commands bg, fg, jobs. Delayed start of processes - commands at, atq and atrm.

11th week: The Linux programming Interface. System calls. Process identifier (PID), function getpid. Signals -fundamental concepts. Interprocess communication via signals. Signal mask. Commands "kill" and "raise". Changing signal dispositions. Designing signal handlers, commands signal and sigaction. System data types.

12th week: Time Functions: time a gettimeofday. Time-conversion functions. Structures timeval, timespec and tm. Real vs. CPU time. Sleepers, functions sleep, usleep. Interval timers. Nanosecond timers and sleepers.

13th week: Multithreading (API Pthread). Thread ID. Thread creation and termination. Threads synchronization - mutexes. Communication between threads. Thread signal mask. Thread timers via signals.

Recommended literature:

William E. Shotts, Jr., The Linux Command Line: A Complete Introduction, No Starch Press, 2012

Kernighan, B. W., Ritchie, D. M., C programming language, 2nd edition, Prentice Hall PTR, 1988

Stroustrup, B., The C++ Programming Language, Pearson Education, 2013

Kerrisk, M, The Linux Programming Interface: A Linux and UNIX System Programming Handbook, No Starch Press, 2010

Course language:

Notes:

Course assessment

Total number of assessed students: 172

А	В	С	D	Е	FX
56.4	16.86	19.77	3.49	3.49	0.0
Provides: RNDr. Branislav Brutovský, CSc.					

Date of last modification: 20.09.2021

University: P. J	. Šafárik Univers	ity in Košice			
Faculty: Facult	y of Science				
Course ID: ÚF EMBF1/18	V/ Course na	Course name: Experimental Methods of Biophysics I			
Course type, sc Course type: I Recommended Per week: 2 Pe Course metho	Course type, scope and the method: Course type: Lecture Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: present				
Number of EC	FS credits: 3				
Recommended	semester/trimes	ster of the cours	e: 4.		
Course level: I.					
Prerequisities:					
Conditions for	course completi	on:			
Learning outco	omes:				
Brief outline of	the course:				
Recommended	literature:				
Course languag	Course language:				
Notes:					
Course assessment Total number of assessed students: 6					
А	В	С	D	Е	FX
16.67	50.0	50.0 33.33 0.0 0.0 0.0			
Provides: prof. RNDr. Pavol Miškovský, DrSc.					
Date of last modification: 30.03.2022					
Approved: doc.	Approved: doc. Mgr. Daniel Jancura, PhD.				

· · · · · · · · · · · · · · · · · · ·					
University: P. J	. Šafárik Univers	sity in Košice			
Faculty: Facult	y of Science				
Course ID: ÚF EMBF2/18	V/ Course na	Course name: Experimental Methods of Biophysics II			
Course type, sc Course type: I Recommended Per week: 2 Pe Course metho	ope and the met Lecture d course-load (h er study period: d: present	thod: ours): 28			
Number of EC	I'S credits: 3				
Recommended	semester/trimes	ster of the cours	e: 4.		
Course level: I.					
Prerequisities:					
Conditions for	course completi	on:			
Learning outco	Learning outcomes:				
Brief outline of	Brief outline of the course:				
Recommended	Recommended literature:				
Course languag	Course language:				
Notes:					
Course assessm Total number of	ent f assessed studen	its: 7			
А	В	С	D	Е	FX
42.86	14.29	42.86	0.0	0.0	0.0
Provides: prof. RNDr. Erik Sedlák, DrSc., RNDr. Gabriela Fabriciová, PhD., RNDr. Marián Fabián, CSc., doc. RNDr. Gabriel Žoldák, DrSc.					
Date of last mo	dification: 30.11	.2021			
Approved: doc.	. Mgr. Daniel Jan	icura, PhD.			

Faculty: Faculty of Science

Course ID: ÚFV/	Course name: Experimental Methods of Biophysics III
EMBF3/18	

Course type, scope and the method: Course type: Lecture

Recommended course-load (hours):

Per week: 2 Per study period: 28

Course method: present

Number of ECTS credits: 3

Recommended semester/trimester of the course: 5.

Course level: I.

Prerequisities:

Conditions for course completion:

Oral exam where the students present theoretical knowledge of topics listed in the course syllabus.

Learning outcomes:

Students will gain basic knowledge on confocal fluorescence microscopy, time-resolved fluorescence microscopy (FLIM / PLIM), high resolution microscopy, flow cytometry. Students will also learn the methods of optical imaging used in clinical practice, the basics of Lab on chip technology and the principles of building experimental equipment. Students will get a theoretical basis, which they will be able to utilize for the experiments using the listed methods.

Brief outline of the course:

1. Confocal fluorescence microscopy:

Principles of confocal fluorescence microscopy, preparation of samples for vital staining and for immunofluorescence, detection of cell organelles, localization and distribution of proteins in cells, colocalization of dyes, analysis of fluorescence image.

2. Time-resolved fluorescence microscopy (FLIM):

Principle of time-resolved fluorescence microscopy, time and frequency domain FLIM detection, Förster resonance energy transfer (FRET), the influence of the environment on the detection of fluorescent probes, measurement of the level of oxidative stress in cells.

3. Time-resolved phosphorescence microscopy (PLIM):

Principles of time-resolved phosphorescence microscopy, the influence of the environment on the detection of phosphorescent probes, quenching of phosphorescence, detection of temperature changes and oxygenation in cells and tissues. Possibilities of sensor application for detection of oxygen and oxygen deficiency in vitro and in vivo.

4. High resolution microscopy:

Principle of selected microscopic approaches for detection of proteins and molecules in cells at high resolution: structured illumination microscopy (SIM), stimulated emission depletion (STED), photo-activated localization microscopy (PALM), stochastic optical reconstruction microscopy (STORM).

5. Flow cytometry:

Principles of flow cytometry, use of fluorescence in cell counting and subsequent analysis of specific factors, cell cycle, changes caused by oxidative stress, separation of cells based on vital staining and immunolabeling, detection of apoptosis and necrosis in the cell population.

6. Optical imaging techniques used in clinical practice:

Examples of bioimaging through: autofluorescence, second harmonic generation (SHG), fluorescence of clinically approved contrast molecules, optical coherence tomography (OCT). 7. Advanced imaging methods:

Electron microscopy, transmission electron microscope, scanning electron microscope. Atomic force microscopy (AFM).

8. Lab on chip technology:

Advantages of LOC technologies. Overview of microfluidic systems and their applications in molecular and cell biology. Instrumentation and preparation of microfluidic devices.

Development of experimental equipment:

9. Electrical signals:

Excitation and detection of electrical signals. Arbitrary function generator, digital oscilloscope and multimeter, laboratory card. Voltage sources. Weak signal measurements: lock-in amplifier. Shielding and grounding of devices. Basics of PID control.

10. Sources of optical radiation:

Classic sources of optical radiation: gas discharge lamps, fluorescent lamps, light bulbs. LEDs. Pulsed LED emitters. Lasers. Selection of lasers based on laser radiation parameters: wavelength, power, beam quality, polarization. Pulsed laser radiation sources.

11. Detection of optical radiation:

Basic terms, flux, irradiance, light intensity. Optical detectors, photodiodes, avalanche photodiodes, pin diodes, photomultipliers, thermal detectors. CCD cameras, CMOS cameras.

12. Building experimental apparatuses:

Basic principles of planning experimental equipment. Detection sensitivity, noise sources. Mechanical and thermal stability. Overview of optical and optomechanical components.

Recommended literature:

1. Wolfgang Becker: The bh TCSPC Handbook Seventh Edition, Becker & Hickl GmbH 2017;

- 2. Guy Cox: Optical Imaging Techniques in Cell Biology, Taylor & Frances;
- 3. Howard M. Shapiro: Practical Flow Cytometry Fourth edition, 2003;

4. Nikolas Long and Wing-Tak Wong: The chemistry of molecular imaging, Wiley 2014

Course language:

Slovak language, English.

Notes:

notes:	Notes:					
Course assessment						
A	B	C	D	Е	FX	
100.0	0.0	0.0	0.0	0.0	0.0	
Provides: doc.	Provides: doc. Mgr. Gregor Bánó, PhD., RNDr. Veronika Huntošová, PhD.					
Date of last modification: 22.09.2021						
Approved: doc. Mgr. Daniel Jancura, PhD.						

University: P. J. Šafá	University: P. J. Šafárik University in Košice				
Faculty: Faculty of Science					
Course ID: ÚBEV/ ETB1/99	Course name: Experimental techniques in Biology				
Course type, scope a Course type: Practic Recommended cour Per week: 4 Per stu Course method: pre	nd the method: ce rse-load (hours): dy period: 56 esent				
Number of ECTS cr	edits: 4				
Recommended seme	ster/trimester of the course: 4., 6.				
Course level: I.					
Prerequisities: ÚBEV	V/CYT1/15				
Conditions for cours active participation o	e completion: n practicals/seminars, exam				
Learning outcomes: To provide the studer	ts with the knowledge of basic experimental techniques in biology.				
 Brief outline of the c Course management Molecular markers Molecular cytology Work in aseptic condition MTT test. Laboratory animalise Manipulation with Animal dissection; Surgery in experime Fluorescence and condition Fieldwork; Botann Use of scientific for the science of the s	ourse: nt. Laboratory safety. ; Polymerase chain reaction (PCR), RT-PCR, Real-Time PCR). y – Flow cytometry – principles and application in cell biology. nditions (in vitro); work with cell lines – subculturing, staining, cell counting, s, strains and inbreed lines; Breeding and manipulation with animals. laboratory animals; Behavioural tests. Anatomy of animals. nental research. confocal microscopy in experimental research. ical fieldwork and follow-up laboratory assessment. iterature. Presentation of own results. isms in biological research. Knockout.				
Recommended litera Zutphen, L. F. M., Ba Elsevier, Amsterdam	nture: numans, V., Beynen, A. C.: Principles of Laboratory Animal Science. , 1993				
Course language: English for Erasmus	students				

Notes:

Course assessment Total number of assessed students: 244								
А	В	B C D E FX						
56.56	12.7	12.7 12.3 4.1 13.11 1.23						
Provides: RNDr. Ján Košuth, PhD., prof. RNDr. Peter Fedoročko, CSc., RNDr. Anna Alexovič Matiašová, PhD., RNDr. Terézia Kisková, PhD., Mgr. Vladislav Kolarčik, PhD., univerzitný docent, doc. RNDr. Juraj Ševc, PhD., doc. RNDr. Rastislav Jendželovský, PhD., RNDr. Natália Pipová, PhD.								
Date of last modification: 15.10.2021								
Approved: doc. Mgr. Daniel Jancura, PhD.								

	CC	OURSE INFORM	MATION LETT	ER			
University: P. J. Š	afárik Univers	ity in Košice					
Faculty: Faculty of	of Science						
Course ID: ÚCH VCHU/15	V/ Course na	Course name: General Chemistry					
Course type, scop Course type: Le Recommended o Per week: 4 / 2 I Course method:	be and the me cture / Practice course-load (h Per study peri present	thod: e ours): od: 56 / 28					
Number of ECTS	S credits: 7						
Recommended se	mester/trimes	ster of the cours	e: 1.				
Course level: I.							
Prerequisities: Ú	CHV/CHV1/9	9					
Conditions for co Written test in the participation on se	urse completi e middle and t eminars.	on: he end of the ser	mester followed	by the oral exam	ination. Active		
Learning outcom To provide studer of chemical bond periodicity.	es: nts with know s, physical and	ledge of atoms a l chemical prope	nd molecules the rties of elements	eir electronic struard compounds	ucture, theories as well as their		
Brief outline of th Main terms used periodicity and i intermolecular int Solutions. Chemi Classification of c	ne course: I in chemistry ts effect on t ceractions. Che ical equilibriu chemical reacti	t. Atoms – mod the properties of emical structure a m. Basis of che ons. Electrochen	els of atoms, el f elements, radio and physical prop emical thermody histry.	ectron configura oactivity. Chemi erties of matter. namics and che	ation, chemical ical bonds and State of matter. mical kinetics.		
Recommended lin 1. Atkins P., Jones 2. Russel J.B.: Ge	terature: s L.: Chemical eneral Chemist	Principles, 2nd e ry, 2nd ed., McG	ed., Freeman, Nev raw Hill, London	w York 2002. 1992.			
Course language:	:						
Notes:	,						
Course assessmen Total number of a	nt ssessed studen	its: 361					
A	В	С	D	Е	FX		
25.76	27.42	27.7	11.08	7.48	0.55		
D d f D1	VIa direster	Zalažáli DrC-	daa DND- I	Dotočšál DLD			

Provides: prof. RNDr. Vladimír Zeleňák, DrSc., doc. RNDr. Ivan Potočňák, PhD.

Date of last modification: 07.02.2022

University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚFV/ VF1a/12	Course name: General Physics I
Course type, scope a Course type: Lectur Recommended cour Per week: 4 / 2 Per Course method: pre	nd the method: e / Practice rse-load (hours): study period: 56 / 28 sent
Number of ECTS cro	edits: 7
Recommended seme	ster/trimester of the course: 1.
Course level: I.	
Prerequisities:	
Conditions for cours Terms and conditions -participation in class -active participation a -submitting all the as -tests during the seme -project group work a Final assessment: -final oral examination Conditions for succes -participation in lesso -achieving the level h	e completion: of assessment during the semester es in accordance with study regulations and teacher's instructions it seminars and exercises signments in accordance with teacher's instruction ester and its successful presentation and defence an sful completion of the course: ons in accordance with the study regulations and teacher's instructions igher than 50 % in assessment during the semester and in final assessment
Learning outcomes: By the end of the comphysics and thermody course content and ap	urse student masters basic knowledge connected with mechanics, molecular ynamics. Student will be able to solve various problems connected with the ply gained knowledge in different situations.
Brief outline of the c 1. Basic knowledge of 2. Mechanics of parti 3. Gravitational field 4. Work, power and e 5. Mechanics of syste 6. Mechanics of rigid 7. Mechanics of fluid 9. Basics of molecular 10. Basics of thermood 11. Heat transfer. The 12. Structure and profile.	ourse: f the calculus, vector algebra. Standards and units. cle. nergy. m of particles. body. ic body. s. r physics. Structure and properties of gases. dynamics. perties of liquids

Recommended literature:

CUMMINGS, Karen, LAWS, Priscilla, REDISH, Edward, COONEY, Patrick: Understanding Physics, John Wiley & Sons, 2004

Course language:

English

Notes:

Course assessment

Total number of assessed students: 350

А	В	С	D	Е	FX
23.71	15.14	21.14	14.86	16.29	8.86
Provides: doc. RNDr. Zuzana Ješková, PhD.					

Date of last modification: 15.09.2021

University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚFV/ VF1b/03	Course name: General Physics II
Course type, scope a Course type: Lectur Recommended cour Per week: 4 / 2 Per Course method: pre	nd the method: re / Practice rse-load (hours): study period: 56 / 28 esent
Number of ECTS cr	edits: 7
Recommended seme	ster/trimester of the course: 2.
Course level: I.	
Prerequisities: ÚFV/	VF1a/12
Conditions for cours To successfully comp sufficient understand to continue the study Knowledge of individ Maxwell's equations Another requirement Credit evaluation take exercises, 4 credits), s that is part of the back is to obtain 50 points 50% of points from e Numerical exercises student must obtain a Oral exam with a ma level of at least 50%) Rating scale A 100-91 B 90-81 C 80-71 D 70-61 E 60-50 Fx 49-0	the completion: Note the course (presence, if necessary distance), the student must demonstrate ing of the basic concepts and laws of electromagnetism, so that it is possible of general physics III, IV and the discipline of electromagnetic field theory. dual laws of electricity and magnetism and their generalization in the form of is required. Knowledge of these laws in nature and in practical use is required. is adequate skills in solving the problems of electricity and magnetism. es into account the scope of teaching (4 hours of lectures, 2 hours of numerical self-study (1 credit), evaluation (2 credits) and the fact that it is a basic subject helor's state exam. The minimum limit for successful completion of the course from the subsequent point evaluation, while it is necessary to obtain at least ach part: maximum number of 20 points (usually 2 written tests of 10 points each, the t least 5 points from each test) uximum of 80 points (answer to three questions, each of which must reach a
After completing lect of electricity and ma He will also gain ade	tures and exercises, the student will have sufficient knowledge of the basics gnetism and will be able to solve numerical problems of electromagnetism. equate knowledge about electromagnetic phenomena in nature and the use of

electromagnetic phenomena in technical applications.

Brief outline of the course:

1. Week: Electrostatic field in vacuum. Culomb's law. Electric field. Electric dipole. Flux of electric field. Gauss' law.

2. Week: Work of forces in the electrostatic field. Potential. Relationship between electric fiel and electric potential. Potential and its measurement. Capacity of conductor and conductor system. Energy of electrostatic field.

3. Week: Stationary electric field and steady electric current. Ohm's law. Superconductivity. Equation of continuity of electric current. Electrical circuits with steady voltage. Kirchhoff's laws and their application. Work, power, energy and efficiency of the source of electromotive voltage.

4. Week: Electric current in electrolytes, semiconductors, gases and in vacuum. Thermoelectric phenomena and their use.

5. Week: Origin, properties and basic quantities of a stationary magnetic field in vacuum. Biot-Savart law and its application. Magnetic flux density.

6. Week: Interactions of a magnetic field with moving electrically charged particles and with electric currents. Ampere's law. Interaction between current conductors. Definition of ampere as current unit. Lorentz force.

7. Week: Quasi-stationary electric field. Capacitor charging and discharging process (R-C circuit). The phenomenon of electromagnetic induction. Faraday's law. Phenomenon of self-induction and mutual inductance, mutual inductance. Potential of magnetic field.

8. Week: Transient in the R-L circuit. Energy of magnetic field. Energy conservation law. Magnetic dipole. Alternating currents and basic circuits of alternating electric current. RLC circuit

9. Week: Serial and parallel resonance. Multiphase currents. Rotating magnetic field. Formation of multiphase currents. Electric motor. Power of alternating electric current.

10. Week: Electrical phenomena in the material environment. Dielectric polarization, mechanisms. Electric field in dielectric. Interaction of electric charges stored in a dielectric. Gauss' law. Polarization vector and electrical induction vector and their mutual relationship. Linear and nonlinear dielectrics.

11. Week: Magnetic properties of substances. Elementary magnetic field of an atom. Magnetic state of substances. Magnetic polarization. Diamagnetism and paramagnetism. Arranged magnetic structure. Ferromagnets.

12. Week: Unsteady electromagnetic field. Maxwell's equations.

Recommended literature:

T. Matsushita: Electricity and Magnetism, Springer, 2017

Course language:

english

Notes:

Presence form represents a standart form for the course, if a need arises, the course is performed using MS Teams.

Course assessment

Total number of assessed students: 368

А	В	С	D	Е	FX
34.78	14.95	15.76	12.23	9.78	12.5

Provides: prof. RNDr. Peter Kollár, DrSc., doc. RNDr. Adriana Zeleňáková, PhD., doc. RNDr. Erik Čižmár, PhD.

Date of last modification: 10.02.2023

University: P. J.	University: P. J. Šafárik University in Košice						
Faculty: Faculty of Science							
Course ID: ÚF VF1c/12	V/ Course na	Course name: General Physics III					
Course type, sc Course type: I Recommended Per week: 4/2 Course metho	ope and the me Lecture / Practice d course-load (h 2 Per study peri d: present	thod: e ours): od: 56 / 28					
Number of EC	FS credits: 7						
Recommended	semester/trime	ster of the cours	e: 3.				
Course level: I.							
Prerequisities:	ÚFV/VF1b/03 o	r ÚFV/VFM1b/1	5				
Conditions for Written test (2x Oral examination	course complet) from seminars on.	ion: during the semes	ter.				
Learning outco The objective is	mes: s to acquaint the	students with the	basis of oscilati	ons, waves and o	ptics.		
Brief outline of Undamped osc Fourier transfor Huyghens princ Geometrical op Light as electr Photon's theory	the course: ilations, Mather mation, Forced ciple. Reflection tics. Mirrors, len omagnetic wave of light. Law of	natical, Physical oscilations. Wave , difraction. Dop s. Fotometry. e. Dispersion, a emision and abs	and Torsional es, their generation pler effect. Wav bsorption, inter- orption, Planck's	pendulum, Damj ion, waves equati es speed in mater ference, difractio s law of radiation.	ped oscilations, on.Interference. rials. Acoustics. n, polarization. Lasers.		
Recommended 1. A. Hlavička (2. R.P. Feynman 3. D. Halliday (4. J. Fuka, B. H 5. A. Štrba, Vše	literature: et al., Fyzika pro n et al., Feynmar et al.,Fyzika-Vys avelka, Optika a cobecná Fyzika 3	pedagogické fak nove prednášky z okoškolská učeb atómová fyzika, – Optika, ALFA	culty, SPN, 1971 Fyziky I,II,III, z nice obecné fyzi SPN,1961 x, 1979	ALFA, 1985 ky, VUTIUM, 20	10		
Course languag 1. slovak 2. english	ge:						
Notes:							
Course assessm Total number of	ent f assessed studer	its: 150					
А	В	С	D	Е	FX		
30.67	26.0	26.67	12.0	4.67	0.0		
Provides: doc. 1	RNDr. Ján Füzer	, PhD.	1	I			

Date of last modification: 10.09.2021

Faculty: Faculty of Science

Course ID: ÚFV/	Course name: General Physics IV
VF1d/12	

Course type, scope and the method:

Course type: Lecture / Practice

Recommended course-load (hours): Per week: 4 / 2 **Per study period:** 56 / 28

Course method: present

Number of ECTS credits: 7

Recommended semester/trimester of the course: 4., 6.

Course level: I.

Prerequisities: ÚFV/VF1c/10 or ÚFV/VF1c/12 or ÚFV/VF1c/22

Conditions for course completion:

- active participation in lectures and excersises

- submission of solved tasks

- 2x test

- an exam

Credit evaluation of the subject: direct teaching and consultations (2credits), self-study

(1credit), practical activities- solved tasks (2redits), evaluation (2credits), a total of 7credits. Minimum limit for completion of the course is to obtain at least 51% of the total evaluation.

Learning outcomes:

Basic knowledge about the atomic structure and spectra and nuclei, and elementary particles. Basic experimental methods in nuclear physics and passage of nuclear radiation through media.

Brief outline of the course:

1.-6. week Atomic Physics - A.Kravčáková (P):

Corpuscular-wave dualism: De Broglie waves. Experimental confirmation of de Broglie's hypothesis. Uncertainty principle.

Atom structure: Atomic hypothesis. Rutherford's experiment. Bohr model of the atom.

Hydrogen radiation spectra. Combination principle. Quantum mechanical description of a hydrogen atom.

Electron shell: Spectra of hydrogen type atoms. Experimental verification of the existence of discrete levels of atoms (Franck-Hertz experiment). Angulat momentum of electron motion. Stern-Gerlach experiment. Quantum states of electrons. Atoms with more electrons. Alkali metal spectra. Total angular momentum of an atom. Magnetic momentum of an atom. An atom in an external magnetic and electric field. Zeeman's phenomenon. Selection rules. Pauli's principle. Periodic table of elements. X-ray spectra.

Molecules: Ion and covalent coupling, spectra of molecules.

7.-12. week Nuclear Physics - J.Vrláková (P):

Basic characteristics of atomic nuclei: Mass and electric charge. Radius of the atomic nucleus. Binding energy. Spin and magnetic momentum of the nucleus. Quadrupole momentum. Parity.

Nuclear forces and models of atomic nuclei: Properties of nuclear forces. Meson theory of nuclear forces. Models of atomic nuclei (droplet, layer and generalized model).

Radioactive radiation: Basic laws of radioactive decay. Law of decay. Alpha decay. Beta decay. Processes taking place in the nucleus during beta conversion. Neutrino existence hypothesis. Fermi's theory. Internal conversion. Gamma radiation.

Nuclear reactions: Basic terms and definitions. Classification of nuclear reactions. Conservation laws. Effective cross section. Mechanisms of nuclear reactions. Basic types of reactions. Breit-Wigner formula. Reactions with neutrons. Fission of atomic nuclei. Mechanism of fission. Nuclear reactor. Thermonuclear reactions.

Week 13 Subnuclear physics - A.Kravčáková (P):

Elementary particles: The concept of an elementary particle. Basic characteristics of particles. Conservation laws. Types of interactions. Antiparticles. Classification of elementary particles. Strange particles. Resonances. Quark model of hadrons.

Cosmic radiation: Primary and secondary components. Elementary particles and cosmology.

Week 14 Experimental methods - A.Kravčáková (P):

Passage of radiation through matter: The passage of heavy charged particles, electrons and gamma radiation through the matter.

Detectors: Basic characteristics of detectors. Volt-ampere characteristic. Gas detectors. Ionization chambers and Geiger-Müller computer. Scintillation, Cherenkov and semiconductor detectors. Track detectors.

Particle accelerators: Linear accelerator. Cyclic accelerators. Colliders.

Recommended literature:

1. Beiser A., Úvod do moderní fyziky, Praha, 1975.

2. Úlehla I., Suk M., Trka Z.: Atómy, jádra, částice, Praha, 1990.

3. Síleš E., Martinská G.: Všeobecná fyzika IV, skriptá PF UPJŠ, 2. vydanie, Košice, 1992.

4. Vrláková J., Kravčáková A., Vokál S.: Zbierka príkladov z atómovej a jadrovej fyziky, skriptá PF UPJŠ, Košice, 2016.

5. Hajko V. and team of authors, Physics in experiments, Bratislava, 1997.

6. Nosek D., Jádra a částice (Řešené příklady), Matfyzpress, MFF UK, Praha 2005,

7. Kravčáková A., Vokál S., Vrláková J., Všeobecná fyzika IV, 1.časť Atómová fyzika, skriptá PF UPJŠ, Košice, 2020.

8. Yang F., Hamilton J.H., Modern Atomic and Nuclear Physics, WSC Singapore, 2010.

Course language:

slovak and english

Notes:

Course assessment Total number of assessed students: 109

Total humber of assessed stadents. Toy						
А	В	С	D	Е	FX	
39.45	29.36	12.84	9.17	9.17	0.0	
Provides: doc. RNDr. Janka Vrláková, PhD., doc. RNDr. Adela Kravčáková, PhD.						

Date of last modification: 16.09.2021

University: P. J.	Šafárik Univers	sity in Košice			
Faculty: Faculty	y of Science				
Course ID: ÚF UPF1/12	V/ Course name: Introduction to Computational Physics				
Course type, sc Course type: I Recommended Per week: 2 / 1 Course metho	ope and the me Lecture / Practice I course-load (h Per study peri d: present	thod: e nours): iod: 28 / 14			
Number of ECT	FS credits: 4				
Recommended	semester/trime	ster of the cours	e: 1., 3.		
Course level: I.					
Prerequisities:					
Conditions for Elaboration of r Exam and discu	course complet nicroreferat on g ssion of the imp	ion: given topics. lementation of th	e given project.		
Learning outco The aim of the processes in co implement com	mes: lecture is to pro onventional com putational proce	ovide students wi nputers, as well sses using deeper	th the physical b as to provide le knowledge of p	background of th ess conventional hysical processe	le computational l possibilities to s.
Brief outline of Physical process point of view. P . Computer mod computing. Alte quantum compu	the course: ses utilised in con- hysical limits of deling and physic ernative method ating).	ntemporary comp current compute cal reality. Comp s of computation	outers. Computati r technologies (Notational completional completional completional completional completion)	onal processes / Moore, Amdahl l exity and paralel cal processors, I	thermodynamics aws lism. Distributed DNA computing,
Recommended Actual literature	literature: e provided by le	cturer.			
Course languag	ge:				
Notes:					
Course assessm Total number of	ent f assessed studer	nts: 48			
А	В	C	D	Е	FX
85.42	8.33	4.17	0.0	2.08	0.0
Provides: doc. I	RNDr. Jozef Ulio	čný, CSc.			
Date of last mo	dification: 22.0	9.2021			
Approved: doc.	Mgr. Daniel Jan	ncura, PhD.			

University: P. J. Šafá	árik University in Košice						
Faculty: Faculty of Science							
Course ID: ÚFV/ UVF/05	Course name: Introduction to General Physics						
Course type, scope a Course type: Practi Recommended cou Per week: 2 Per stu Course method: pr	and the method: ice irse-load (hours): idy period: 28 esent						
Number of ECTS ci	redits: 2						
Recommended seme	ester/trimester of the course: 1.						
Course level: I.							
Prerequisities:							
Terms and condition -participation in clas -active participation -submitting all the as -tests during the sem Final assessment: -based on assessmen Conditions for succe -participation in less -achieving the level	se completion: s of assessment during the semester ses in accordance with study regulations and teacher's instructions at seminars and exercises ssignments in accordance with teacher's instruction nester at during the semester essful completion of the course: ons in accordance with the study regulations and teacher's instructions higher than 50 % in assessment during the semester and in final assessment						
Learning outcomes: By the end of the co physics and thermoo collection, videomea	burse student is able to solve problems connected with mechanics, molecular dynamics. In solving problems student is able to apply digital tools for data assurement and computer modelling and data processing and their analysis.						
 Brief outline of the of the course is an aux and Thermodynamic connected with the full full the full thermodynamic connected with the full thermodynamic connected with the full thermodynamic of motion. Caravitational field full thermodynamic of the full thermodynamic of thermodynamic of thermodynamic of the ful	course: iliary subject to the course General physics 1 - Mechanics, Molecular Physics es aimed to development of conceptual understanding and problem solving following areas: dynamics of motion along a line and two-dimensional motion of particle. d. Projectile motion. energy. Law of energy conservation. . Equation of rotational motion. n conservation and angular momentum conservation. ok's law.						

11. Liquids. Surface tension.

12. Changes of state.

Recommended literature:

CUMMINGS, Karen, LAWS, Priscilla, REDISH, Edward, COONEY, Patrick: Understanding Physics, John Wiley & Sons, 2004

Course language:

English

Notes:

Course assessment

Total number of assessed students: 349

А	В	С	D	Е	FX	
36.96	20.34	24.64	13.18	4.58	0.29	
Provides: doc. RNDr. Zuzana Ješková, PhD.						

Date of last modification: 15.09.2021

University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	Science
Course ID: ÚFV/ UVF2/07	Course name: Introduction to General Physics II
Course type, scope a Course type: Practi Recommended cou Per week: 2 Per stu Course method: pro	and the method: ce rse-load (hours): ady period: 28 esent
Number of ECTS cr	redits: 2
Recommended seme	ester/trimester of the course: 2.
Course level: I.	
Prerequisities:	
Conditions for course Terms and conditions -participation in class -active participation -submitting all the as -tests during the sem -based on assessmen Conditions for succe -participation in less -achieving the level I	se completion: s of assessment during the semester ses in accordance with study regulations and teacher's instructions at seminars and exercises signments in accordance with teacher's instruction ester Final assessment: t during the semester ssful completion of the course: ons in accordance with the study regulations and teacher's instructions higher than 50 % in assessment during the semester and in final assessment
Learning outcomes: By the end of the cou connected with selec	rse student is able to solve problems and explain phemomena and experiments ted areas of Electricity and Magnetism.
 Brief outline of the of The course is an auxility to development of courses: 1. Electric field. Courses: 2. Work, electric poteds 3. Electric capacitantes 4. Electric current. O 5. Work and power. If 6. Magnetic field. 7. Interaction betwees 8. Transient phenomes 9. Electromagnetic in 10. Transient phenomes 11. Alternating curres 12. Resonance in serting the serting curres 	course: liary subject to the course General physics 2 - Electricity and Magnetism aimed onceptual understanding and problem solving connected with the following lomb's law. ential energy, electric potential. ce and capacitors. bhm's law, Kirchhoff's laws. Energy and efficiency of sources of electromotive force en magnetic field and electric charge. ena in RC circuit. nduction. nena in RL circuit. nt circuits. ies and paralel circuits.
Recommended liter	ature:

Matsushita, Ter CUMMINGS, I Physics, John V	Matsushita, Teruo. Electricity and Magnetism, Springer 2017 CUMMINGS, Karen, LAWS, Priscilla, REDISH, Edward, COONEY, Patrick: Understanding Physics, John Wiley & Sons, 2004							
Course languag English	ge:							
Notes:								
Course assessm Total number of	ent f assessed studen	ts: 272						
А	В	С	D	Е	FX			
38.97 22.43 20.96 8.46 9.19 0.0								
Provides: doc. RNDr. Zuzana Ješková, PhD.								

Date of last modification: 15.09.2021

University: P. J.	Šafárik Univers	ity in Košice					
Faculty: Faculty	of Science						
Course ID: ÚF ZMF/17	<i>V</i> / Course name: Introduction to Mathematics for Physicists						
Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 1 / 2 Per study period: 14 / 28 Course method: present							
Number of EC	IS credits: 3						
Recommended	semester/trimes	ster of the cours	e: 1.				
Course level: I.							
Prerequisities:							
Conditions for	course completi	on:					
Learning outco	mes:						
Brief outline of	the course:						
Recommended	literature:						
Course languag	ge:						
Notes:							
Course assessment Total number of assessed students: 300							
А	A B C D E FX						
39.67 21.0 19.0 10.67 8.67 1.0							
Provides: RNDr. Tomáš Lučivjanský, PhD., univerzitný docent, doc. RNDr. Jozef Hanč, PhD.							
Date of last modification: 16.11.2021							
Approved: doc.	Mgr. Daniel Jan	icura, PhD.		-			

University: P. J. Šafá	rik University in Košice					
Faculty: Faculty of S	cience					
Course ID: Dek. PF UPJŠ/USPV/13	Course name: Introduction	n to Study of Sciences				
Course type, scope a Course type: Lectur Recommended cour Per week: Per stud Course method: pre	Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: Per study period: 12s / 3d Course method: present					
Number of ECTS cr	edits: 2					
Recommended seme	ster/trimester of the cours	e: 1.				
Course level: I.						
Prerequisities:						
Conditions for cours	e completion:					
Learning outcomes:						
Brief outline of the c	ourse:					
Recommended litera	iture:					
Course language:						
Notes:						
Course assessment Total number of assessed students: 2196						
abs n						
89.34 10.66						
Provides: doc. RNDr. Marián Kireš, PhD.						
Date of last modification: 30.08.2022						
Approved: doc. Mgr. Daniel Jancura, PhD.						

University: P. J. Šafá	rik University in Košice							
Faculty: Faculty of S	cience							
Course ID: ÚMV/ UAD/10	Course name: Introduction to data analysis							
Course type, scope a Course type: Lectur Recommended cour Per week: 1 / 1 Per Course method: pre	Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 1 / 1 Per study period: 14 / 14 Course method: present							
Number of ECTS cr	edits: 2							
Recommended seme	ster/trimester of the course: 1.							
Course level: I.								
Prerequisities:								
Conditions for cours Test (40p) and individ Oral presentation of t At least 50% must be Final evaluation: ≥90	e completion: dual project work (20p). he individual project work (5p). obtained from each part. % A; ≥80% B; ≥70% C; ≥60% D; ≥50% E; <50% FX.							
To know the basic p understand its import To understand elemen To gain experience in	purpose of statistical data analysis, its methods and statistical thinking and ance for science and practical life. htary statistical concepts. handling real data using spreadsheet Excel and statistical software R.							
 Brief outline of the c 1. Introduction (the b statistics) 2. Collecting Data (ty 3. Handling Data (v skewness and kurtosi 4. Relationships in da 5. Statistical inference 	ourse: asic philosophy and aim of statistical data analysis, descriptive and inductive opes of data, random sample, randomized experiment) visualization, summarizing – measures of center, measures of variability, s, empirical rule) - 5 weeks ata (introduction to regression and correlation) - 4 weeks e (elementary view into estimation and testing hypothesis) - 2 weeks							
Recommended litera 1. Anděl, J.: Statistich 2. Rossman, A.J. et a 2009 3. Utts, J.M.: Seeing 4. Utts, J.M., Heckard 5. Zvára, K., Štěpán, Czech)	ture: ké metody, Matfyzpress, Praha, 1998 (in Czech) l.: Workshop Statistics: Discovery with Data and Fathom, 3rd ed. Wiley, Through Statistics, 4th ed., Thomson Brooks/Cole, Belmont, 2014 d R.F.: Mind on Statistics, 6th ed. Thomson Brooks/Cole, Belmont, 2021 J.: Pravděpodobnost a matematická statistika, Matfyzpress, Praha, 2001 (in							
Course language: Slovak								
Notes:								

Course assessment Total number of assessed students: 434							
А	A B C D E FX						
36.87	25.12	26.04	10.37	0.46	1.15		
Provides: doc. RNDr. Martina Hančová, PhD.							
Date of last modification: 13.09.2021							
Approved: doc. Mgr. Daniel Jancura, PhD.							

University: P. J. Safarik	University in Košice
Faculty: Faculty of Scie	ence
Course ID: ÚFV/ LTV/18	ourse name: Laboratory techniques
Course type, scope and Course type: Practice Recommended course Per week: 2 Per study Course method: prese	l the method: e-load (hours): period: 28 ent
Number of ECTS cred	its: 2
Recommended semeste	er/trimester of the course: 2.
Course level: I.	
Prerequisities:	
Conditions for course (1) Test-paper (written - at the end of the theor (2) Laboratory protocol	completion: exam during the semester - approximately in the 5th week of the semester etical-computational part of the course) (laboratory report)
Learning outcomes: Completing the course in biophysical (chemica	student will get knowledge and first experiences of safe and efficient work al, optical spectroscopy) laboratory.
Brief outline of the cou	irse:
Week 1 Course schedule and the fundamentals of la definition, presentation Interdisciplinary Biosci Week 2	requirements for successful completion of the course. Introduction to boratory work and safety, chemical and general safety. Introduction and a of the laboratories at the Department of Biophysics and Center for ences.
Composition of substant molecular weights, per- formula, mass and mass solution. Week 3	ces and solutions: basic characteristics of solutions. Chemical formula and centage composition from formulas, from empirical formula to molecular as fraction, molar weight, molar volume, molarity, the concentration of a
Mixtures and solutions concentration, the conc in chemical reactions, c Week 4	s: solubility of substances, solution and its concentration, mass/volume entration of a solution in %, molar concentration, mole-mass relationships oncentration units – ppm, ppb.
Mixtures and solutions: Week 5	diluting and mixing solutions.
Written exam. Laborato	bry safety rules and guidelines.
Proper and safe use of laboratory dryer, Milli- glassware/material use	Small laboratory equipment/instruments: automatic pipettes, centrifuge, Q ultrapure water system. Laboratory digester. Care and safe laboratory – handling, cleaning and storing.

Week 7

Preparation of solutions: Analytical balances and proper weighing practices; working safely with solvents, the storage and disposal of chemicals, solvents, stock solutions and chemical waste. What is Parafilm?

Week 8

Preparing buffer solutions. pH determination, acidity and alkalinity. Working principle of pH meter - calibration and working demonstration. Working with acids and bases.

Week 9

Introduction to spectroscopy. The light. Spectroscopic experiment. Spectroscopic techniques. Jablonski diagram. UV-Vis absorption spectrophotometry. Chromophore. Lambert-Beer's law. Absorption spectrum. Absorption spectrophotometer. Fluorescence spectroscopy. Fluorophore. Excitation and emission spectra. Characteristics of fluorescence spectra. Fluorescence quenching. Week 10

Introduction to spectrophotometry: working with spectroscopic equipment - preparation of solutions of selected molecules at different pH and measurement of their UV-Vis absorption spectra. Week 11

Introduction to spectrophotometry: working with spectroscopic equipment - measurement of fluorescence spectra of the selected molecules at different pH.

Week 12/13

Data collection, processing and analysis. Preparing a Practical/Scientific laboratory report. Evaluation of Laboratory reports.

Keeping the laboratory environment clean and safe.

Recommended literature:

Course language:

Course languag	course language.							
Notes:	Notes:							
Course assessment Total number of assessed students: 13								
А	В	С	D	E	FX			
76.92	23.08	0.0	0.0	0.0	0.0			
Provides: RND	Provides: RNDr. Zuzana Jurašeková, PhD.							
Date of last modification: 21.09.2021								
Approved: doc.	. Mgr. Daniel Jan	cura, PhD.						

University: P. J.	. Šafárik Univers	ity in Košice					
Faculty: Faculty	y of Science						
Course ID: ÚF MFY/12	Jourse ID: ÚFV/ Course name: Mathematical Physics IFY/12 IFY/12						
Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 3 / 1 Per study period: 42 / 14 Course method: present							
Number of EC	IS credits: 6		4				
Recommended	semester/trimes	ster of the cours	e: 4.				
Course level: 1.	,						
Prerequisities:	ÚMV/FRPb/19						
Conditions for	course completi	on:					
Learning outco	mes:						
Brief outline of	the course:						
Recommended	literature:						
Course languag	ge:						
Notes:							
Course assessment Total number of assessed students: 88							
А	A B C D E FX						
23.86	23.86 17.05 13.64 11.36 31.82 2.27						
Provides: RNDr. Tomáš Lučivjanský, PhD., univerzitný docent							
Date of last modification: 16.11.2021							
Approved: doc.	Mgr. Daniel Jan	cura, PhD.					

University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚMV/ MTFa/15	Course name: Mathematics I for physicists
Course type, scope a Course type: Lectur Recommended cour Per week: 2 / 2 Per Course method: pre	nd the method: re / Practice rse-load (hours): study period: 28 / 28 esent
Number of ECTS cr	edits: 5
Recommended seme	ster/trimester of the course: 1.
Course level: I.	
Prerequisities:	
Conditions for cours To complete the courterms and the ability is according to the re During the semester, (together 50 points). may write the exam. number of 30 points. 59-50-D, 49-40-E. If exam test (12 points)	e completion: rse, it is necessary to demonstrate the acquirement of basic mathematical to solve problems from selected thematic units. The evaluation of the subject esults from the semester and in view of the results of the written final test. students write tests at all seminars (together 20 points) and two extensive tests It is necessary to obtain at least 28 points during the semester. Then students To pass the exam, it is necessary to obtain at least 12 points from the maximum The scale for student evaluation is as follows: 100-80-A, 79-70-B, 69-60-C, a student does not achieve the required minimal number of points from the and during the semester (together 28 points), he/she is evaluated by FX.
Learning outcomes: After completing the equations and inequa differential and integr	e course, the student can use basic mathematical terms, can solve various ations, and is acquainted with basic mathematical knowledge from the ral calculus, and is able to apply the theory in concrete excercises.
Brief outline of the c Week 1-6: Definition functions. Compositi Week 7-14: Limit of Indefinite integrals, b	ourse: of function. Domain and range of functions. Elementary functions. Inverse ons of functions. functions. Continuity of functions. Derivation and its geometric aplications. asic methods of integration. Definite integral and its applications.
Recommended litera Huťka, Benko, Ďurik D. Studenovská, T. M odbory, UPJŠ 2006 D. Studenovská, T. M S. Lang: A First Cou	a ture: ovič: Matematika, Alfa, Bratislava 1991 Iadaras, S. Mockovčiak: Zbierka úloh z matematiky pre nematematické Iadaras: Matematika pre nematematické odbory, UPJŠ 2006 rse in Calculus, Springer Verlag, 1998
Course language: Slovak	
Notes:	

Course assessment Total number of assessed students: 101							
А	A B C D E FX						
21.78	21.78 12.87 19.8 15.84 18.81 10.89						
Provides: RNDr. Jana Borzová, PhD., RNDr. Miriam Kleinová, Mgr. Miriama Kmeciková, RNDr. Monika Krišáková							
Date of last modification: 18.04.2022							
Approved: doc. Mgr. Daniel Jancura, PhD.							

University: P. J. Šafárik University in Košice		
Faculty: Faculty of Science		
Course ID: ÚMV/ MTFb/22	Course name: Mathematics II for physicists	
Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 2 Per study period: 28 / 28 Course method: present		
Number of ECTS credits: 5		
Recommended semester/trimester of the course: 2.		
Course level: I.		

Prerequisities: ÚMV/MTFa/15 or ÚMV/MTCb/13

Conditions for course completion:

Mastering standard procedures for solving systems of linear equations. Understanding the concept of function of several variables, mastering the definitions of limit of function, partial derivation of a function, differential of a function, local and global extrema of a function and acquiring skills associated with their use in calculations focused mainly on functions of two variables. Mastering standard procedures for solving basic types of ordinary differential equations of the 1st order. Understanding the concept of infinite series and acquiring skills to use the basic criteria of convergence of number series for deciding on the convergence or divergence of number series. Assessment is given on the basis of a continuous assessment and a written exam, which also includes an oral exam. Ongoing evaluation:

Two tests during the semester - 32 p. Small written tests during the semester - 10 p. Solving homework - 4 p. Active participation in exercises - 4. p. An exam: Final test and oral exam - 30 p. Classification scale: A: 91 % - 100 %, B: 81 % - 90 %, C: 71 % - 80 %, D: 61 % - 70 %, E: 51 % - 60 %, FX: 0 % - 50 %.

Learning outcomes:

The student should be able to explain the basic concepts and gain skills in using standard procedures for solving systems of linear equations using matrices and determinants. The student will expand his knowledge of the function of one variable and master the concept of a function of several variables, and will be able to explain the definitions of function limit, partial derivation of a function, differential of a function, local and global extrema of a function and acquire knowledge and skills oriented mainly on the functions of two variables. The student will learn standard procedures for solving basic types of ordinary differential equations of the 1st order. He will be able to use the acquired knowledge about solving differential equations in modeling and solving problems derived from real situations. The student will gain skills to use the basic criteria of convergence of number series when deciding on the convergence or divergence of number series.
The student will be able to use the acquired knowledge and skills in creating a mathematical model and will learn to effectively use the commands of the mathematical program Maple for routine calculations and visualization for solving created model.

Brief outline of the course:

1. - 3. Systems of linear equations, matrices, determinants.

4. - 7. Functions of several variables, continuity and limit, partial derivatives, differential, local and global extrema of a function of two variables.

8. - 11. Modeling of relations between quantities using differential equations. Methods for solving ordinary differential equations of the 1st order.

12. - 13. Sequences, infinite number series, convergence criteria of infinite number series, infinite functional series, Taylor series.

Recommended literature:

Huťka, V., Benko, E., Ďurikovič, V.: Matematika, Alfa, Bratislava 1991.

Kluvánek, I., Mišík, L., Švec, M.: Matematika II, Bratislava, 1961.

Osička, J.: Matematika pro chemiky, Brno, 2004.

Došlá, Z.: Matematika pro chemiky, Masarykova univerzita, Brno, 2011.

Hughes-Hallett, D., et al.: Applied Calculus. John Wiley & Sons, Inc., 2010.

Rogers, R., C.: The Calculus of Several Variables. 2011.

Course language:

Slovak

Notes:

Course assessment

Total number of assessed students: 25

А	В	С	D	Е	FX
52.0	24.0	16.0	4.0	4.0	0.0

Provides: doc. RNDr. Stanislav Lukáč, PhD., Mgr. Miriama Kmeciková, RNDr. Barbora Hennelová

Date of last modification: 18.04.2022

University: P. J. Šafá	University: P. J. Šafárik University in Košice					
Faculty: Faculty of S	Science					
Course ID: ÚFV/ MSA1/03	Course name: Methods of Structural Analysis					
Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 3 / 2 Per study period: 42 / 28 Course method: present						
Number of ECTS cr	redits: 7					
Recommended semester/trimester of the course: 6.						
Course level: I., II., III.						
Prerequisities						

Conditions for course completion:

For successful completing of the subject student has to show after taking exam adequate knowledge from the area using sophisticated research infrastructure for structural analysis of solids. Content of the subject is focused study of structure analytical methods as TEM, SEM, STEM and X-ray techniques. Credits evaluation takes into account taking part at the lectures - 3credits, study of recommended literature -1credit, working out of experimental protocol from OM and EM -2 credidts and study of recommended literature -2 credits, 2 credits – project, 1 credit – study for 2 written tests (EM and X-ray) - 1 credit. Minimal value to obtain evaluation for is reach 50% of each evaluation (tests and project) points. Point ratio protocol/test EM/TEST X-ray is 40/30/30.

Learning outcomes:

Student due to lecrures and experimental work after final exam demonstrates that he/she meets expectations according to the standards of the subject, which is predicted by short content andreferences. Student is able to use modern methods for structural analysis of metals. He has experiences with optic microscopy, electron microscopy (TEM, SEM, STEM), electron microprobe analysis and X-ray diffractometry.

Brief outline of the course:

Optic microscopy. Electron microscopy: Electron beam instruments, Electron optics, Electron lences and deflection systems, Transmission electron microscopy - principle and construction. Electron – specimen interactions. Electron diffraction. Kikuchy lines. Scanning electron microscopy – principle and cnstrucion. Scanning transmission electron microscopy. High Voltage electron microscopy. Electron microscopy. Electron microscopy. Convergent beam diffraction.

X-ray diffractometry: Scattering of x-rays, Neutrons and neutron scattering, CW - diffractometer, Ewald's sphere, Diffraction on powder samples, The main characteristics of powder diffraction pattern, Structure factor, Ocupation factor, Atomic displacement factor, Peak intensity, shape and symmetry, Sherrer equation. Peak profile, Rietweld method. Qualitative phase analysis, parameters of elementary cell, Profile analysis of diffraction peak and interpretation of profile analysis.

Recommended literature:

1. P.W. Hawkes, J.C.H. Spence, Science of Microscopy, Springer, 2007, ISBN: 10:0-387-25296-7.

2. Vitalij Pecharsky, Peter Zavalij, Fundamentals of Powder Diffraction and Structural characterization of Materials, Publisher: Springer (March 3, 2005)

ISBN-10: 0387241477, ISBN-13: 978-0387241470

3. Jens Als-Nielsen, Des McMorrow, Elements of Modern X-ray Physics, Publisher: Wiley; 2 edition (April 4, 2011),ISBN-10: 0470973943, ISBN-13: 978-0470973943.

4. Current Publications in the field of TEM, REM, X-ray

5. M.D. Graef, M.E. Henry, Structure of Materials, Cambridge Univ. Press, 2012, ISBN:978-1-107-00587-7.

6. S. Amelinckx, D. Dyck, et al, Electron Microscopy - Principle and Fundamentals, VCH, 1997, ISBN: 3-527-29479-1.

Course language:

1. English

Notes:

Lectures can be done at presence form or online using MS Teams. Education form is updated at the begining of the subject. All ppt presentations are accesible in LMS UPJŠ.

Course assessment

Total number of assessed students: 93

А	В	С	D	Е	FX	N	Р
38.71	21.51	7.53	1.08	0.0	0.0	0.0	31.18

Provides: prof. RNDr. Pavol Sovák, CSc., doc. Ing. Karel Saksl, DrSc., Ing. Vladimír Girman, PhD., Mgr. Maksym Lisnichuk, PhD.

Date of last modification: 21.09.2021

University: P. J. Šafárik University in Košice						
Faculty: Faculty of Science						
Course ID: ÚBEV/ MKV/15Course name: Microbiology and basics of virology						
Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 2 Per study period: 28 / 28 Course method: present						
Number of ECTS credits: 5						
Recommended semester/trimester of the course: 5.						
Course level: I.						
Prerequisities: ÚBEV/CYT1/15						
Conditions for course completion: Attendance of practicals (at least 90%), 2 written examinations during semester, final oral examination						
Learning outcomes: Students will obtain a basic informations on viruses, prokaryotic and eukaryotic microorganisms, their cytology, physiology, genetics, ecology, classification, and importance . Information on basic methods for studying microorganisms will be provided.						
Brief outline of the course: Viruses, prokaryotic and eukaryotic microorganisms, their cytology, physiology, genetics, ecology, classification. The importance of microorganisms for humans and environment.						
Recommended literature:						
Course language:						
Notes:						
Course assessment Total number of assessed students: 1500						
A B C D E FX						
24.07 13.47 18.33 18.93 20.93 4.27						
Provides: doc. RNDr. Peter Pristaš, CSc., RNDr. Mária Piknová, PhD., RNDr. Mariana Kolesárová, PhD., RNDr. Lenka Maliničová, PhD.						
Date of last modification: 10.12.2021						
Approved: doc. Mgr. Daniel Jancura, PhD.						

University: P. J. Šafárik University in Košice					
Faculty: Faculty of S	cience				
Course ID: ÚFV/ MTBF/18Course name: Modern Trends in Biophysics					
Course type, scope a Course type: Lectur Recommended cou Per week: 2 Per stu Course method: pre	nd the method: re rse-load (hours): Idy period: 28 esent				
Number of ECTS cr	edits: 2				

Recommended semester/trimester of the course: 1.

Course level: I.

Prerequisities:

Conditions for course completion:

Elaboration of a written essay on a selected topic from the course Modern trends in Biophysics.

Learning outcomes:

This course will provide an opportunity for students to be informed about the actual trends in biophysical research in the world as well about the research performed at the Department of Biophysics, Faculty of Science of UPJS and Center for Interdisciplinary Biosciences TIP-UPJS. Completing this course, the students should be able to understand texts from popular scientific literature about biophysics or bio-sciences in general.

Brief outline of the course:

Areas of interest in biophysics and its importance and position in science.

The structure of biophysics. Characterization of molecular, cellular, medical, environmental and radiation biophysics. Scientific disciplines related to biophysics. The future of biophysics. Lasers in life sciences.

History of laser development. Basic properties and principle of operation of lasers. Different types of lasers. Applications of lasers in biophysical experiments. Diagnostics and manipulation of samples using lasers, modern imaging methods, laser spectroscopy. Laser applications in clinical practice.

Raman spectroscopy and its application in biophysics.

Interaction of matter and light. Methods of optical spectroscopy (vibrational spectroscopy, Raman effect, mutual relation of Raman and infrared spectroscopy). Surface-enhanced Raman scattering. Raman macro- and micro-spectroscopy and imaging. Overview of the use of Raman spectroscopy with emphasis on biophysical applications (Raman spectra of proteins and other biomacromolecules, Raman cell imaging).

PickMolTM nanotechnology based on surface-enhanced Raman spectroscopy for water and food purity screening.

This technology detects whether water or any food matrix is contaminated with persistent organic pollutants (POPs)/pesticides/drugs. The patented PickMoITM technology has recently been verified by a certified laboratory and can be adapted for any organic molecule, which means its great potential for application in other areas, e.g. pharmaceutical and chemical industry, security and sports (doping control). PickMoITM technology is: highly sensitive (ppb concentration level),

selective (detection of specific molecules), efficient (up to 90% cost savings), fast (10 minutes per analysis), portable with immediate on-site analysis.

Methods of studying ligand-macromolecule interaction.

Use of optical spectroscopy methods in the study of ligand-macromolecule interactions: UV-vis absorption spectroscopy, fluorescence spectroscopy, Raman spectroscopy techniques (classical, surface-enhanced, DCDRS). Their advantages and limits. Use of thermodynamic methods: differential scanning calorimetry (study of the stability of the ligand-macromolecule complex) and isothermal titration calorimetry (direct measurement of thermodynamic parameters associated with the formation of complexes).

Scientific superhub European XFEL and biophysics.

X-ray optics and imaging techniques in nano- and mesoscopic scale. Serial protein crystallography and imaging of biological particles (viruses, supramolecular complexes and nature-inspired nanotechnologies). Structural dynamics of biomolecules - dynamic mixing and pump-probe experiments from Terraherz to the hard X-ray region. Mapping the electronic structure of molecules and clusters and their induced dynamics. Chemical imaging. Supercomputer clusters, protein engineering and production of protein nanocrystals, digital micro- and nano-fluidics, 3D printing and additive technologies. Complementary and supporting technologies - optical superresolution techniques, cryolelectron microscopy and electron diffraction. Diffraction limited X-ray sources in the future and atomistic cell mapping. "State-of-the-art" application - demos with commentary. The ecosystem of scientific super hubs as user equipment and opportunities to use EuXFEL. Protein evolution "in a test tube".

Principle of evolutionary methods for the development of proteins and enzymes. Examples of evolutionary methods - display technologies: ribosome display and yeast display. Application of display technology in the development of new or improved protein / enzyme properties for pharmaceutical and biotechnological applications.

Photobiostimulation

Principles of photobiostimulation, cell chromophores, application of therapeutic applications of photobiostimulation on absorption spectra of cell chromophores, perspective applications of photobiostimulation.

Metabolic flows in cells.

The importance of cell metabolism for the therapy of diseases, the principles of measuring metabolic flows, possible applications of methodology for the research in the field of cancer and neurodegenerative diseases.

Modern techniques for the study of individual molecules.

Force spectroscopy of proteins and nucleic acids (AFM, use of optical tweezers), determination of nanomechanical properties of biomolecules, overview of time trajectory analyzes of individual molecules, overview of applications of methods for studying properties of single (molecule) biomolecules: acoustic force spectroscopy, magnetic tweezers and mass photometry. Bioenergetics.

Central concept of bioenergetics - chemiosmotic theory. The main sources of energy in living organisms. Processes in biological systems in which energy is consumed. Mitochondria - structure and basic functions. Respiratory chain in mitochondria. Respiratory chain components. Mechanism of electron transport in the respiratory chain.

Recommended literature:

1. R. Glaser. Biophysics (2nd Edition), Springer-Verlach Berlin, 2012.

2. M.B. Jackson. Molecular and cellular biophysics, Cambridge University Press, 2006.

3. D.J. Dowsett, P.A. Kenny and R.E. Johnston. The physics of diagnostic imaging, Hodder Arnold, 2006.

4. M.R. Hamblin and P. Mroz. Advances in photodynamic therapy, Artech House,

2008.

4. D. Nicholls and S. Fergusson. Bioenergetics 4, Academic Press, 2013.

5. A.D.N.J. de Grey. The mitochondrial free radical theory of aging, R.G. Landis Company, 1999.

6. N. Lane. Síla, sexualita, sebevražda. Mitochondrie a smysl života, Academia, 2012.

Course language:

English language

Notes:

Course assessment

Total number	of assessed	students: 19

А	В	С	D	Е	FX
84.21	15.79	0.0	0.0	0.0	0.0

Provides: doc. Mgr. Daniel Jancura, PhD., doc. Mgr. Gregor Bánó, PhD., RNDr. Zuzana Jurašeková, PhD., prof. RNDr. Pavol Miškovský, DrSc., RNDr. Gabriela Fabriciová, PhD., doc. RNDr. Jozef Uličný, CSc., prof. RNDr. Erik Sedlák, DrSc., doc. RNDr. Katarína Štroffeková, PhD., doc. RNDr. Gabriel Žoldák, DrSc.

Date of last modification: 29.09.2022

University: P. J.	University: P. J. Šafárik University in Košice					
Faculty: Faculty	y of Science					
Course ID: ÚF MBF1/14	Course ID: ÚFV/ Course name: Molecular Biophysics I MBF1/14					
Course type, scope and the method: Course type: Lecture Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: present						
Number of EC	IS credits: 4					
Recommended	semester/trimes	ster of the cours	e: 4.			
Course level: I.	, II.					
Prerequisities:						
Conditions for	course completi	on:				
Learning outco	mes:					
Brief outline of	the course:					
Recommended	literature:					
Course languag	ge:					
Notes:						
Course assessment Total number of assessed students: 33						
А	В	С	D	Е	FX	
57.58	27.27	12.12	0.0	3.03	0.0	
Provides: doc. Mgr. Daniel Jancura, PhD., RNDr. Gabriela Fabriciová, PhD.						
Date of last modification: 24.11.2021						
Approved: doc.	Approved: doc. Mgr. Daniel Jancura, PhD.					

University: P. J.	. Šafárik Univers	ity in Košice				
Faculty: Faculty	y of Science					
Course ID: ÚF MBB1/18	Course ID: ÚFV/ Course name: Molecular and cell biology MBB1/18 Course name: Molecular and cell biology					
Course type, sc Course type: I Recommended Per week: 2 / 2 Course method	ope and the met Lecture / Practice I course-load (h 2 Per study peri d: present	thod: c ours): od: 28 / 28				
Number of EC	FS credits: 5					
Recommended	semester/trimes	ster of the cours	e: 4.			
Course level: I.						
Prerequisities:						
Conditions for	course completi	ion:				
Learning outco	mes:					
Brief outline of	the course:					
Recommended	literature:					
Course languag	ge:					
Notes:				_		
Course assessment Total number of assessed students: 7						
А	В	С	D	Е	FX	
71.43	14.29	0.0	0.0	14.29	0.0	
Provides: doc. RNDr. Katarína Štroffeková, PhD., RNDr. Zuzana Naďová, PhD.						
Date of last modification: 12.07.2022						
Approved: doc.	Mgr. Daniel Jan	icura, PhD.		_		

University: P. J. Šafárik U:	niversity in Košice						
Faculty: Faculty of Scienc	Faculty: Faculty of Science						
Course ID: ÚFV/ NUM/10Cou	rse name: Numerical Methods						
Course type, scope and th Course type: Lecture / Pr Recommended course-lo Per week: 2 / 1 Per study Course method: present	ne method: ractice pad (hours): y period: 28 / 14						
Number of ECTS credits:	: 4						
Recommended semester/t	trimester of the course: 3.						
Course level: I.							
Prerequisities:							
Conditions for course con To successfully complete understanding and ability algebra, which are necess evaluation is participation obtaining credits is passin electronically and with the into account the following projects (2 credits). The m the total score, using the f 69%), E (50-59%), F (0-49	npletion: e the course, the student must demonstrate a sufficient degree of to apply the basic numerical methods of mathematical analysis and sary for subsequent courses in computational physics. The basis of and activity in exercises and work on assignments. The condition for ng 2 written tests at seminars and submitting 4 assignments (projects) e attached computer program. The credit evaluation of the course takes g student workload: direct teaching (2 credits) and individual work on inimum threshold for completing the course is to obtain at least 50% of following rating scale: A (90-100%), B (80-89%), C (70-79%), D (60- 9%).						
Learning outcomes: To acquaint students with t for the next course of comp functions, solve systems of determine eigenvalues and	he basic numerical methods of mathematical analysis and algebra needed putational physics. The student will learn to approximate and interpolate of linear and nonlinear equations, numerically derive and integrate or l eigenvectors of matrices.						
 Brief outline of the course 1. Computational solution 2. Approximation of function 3. Interpolation of function 4. Approximation by trigot 5. Solution of nonlinear eq 6. Numerical methods for 7. Solution of systems of 1 8. Solution of systems of 1 9. Numerical integration (a 10. Numerical differentiation 11. Eigenvalues and eigen 12. The complete problem 	e: of problems and errors of numerical solution. ions. ns. nometric polynomials. Fast Fourier analysis. quations, convergence conditions and error estimation of the methods. solving nonlinear equations. inear equations - direct methods. inear equations - iterative methods. quadrature) of functions. ion of functions. vectors of a matrix - partial problem. of eigenvalues.						

Basic literature:

POZRIKIDIS, C.: Numerical Computation in Science and Engineering, Oxford University Press, 2008.

Other literature:

HAMMING, R.W.: Numerical Methods for Scientists and Engineers, Dover, 1973.

GARCIA, A.L.: Numerical Methods for Physics, Prentice-Hall, 1994.

Course	language:
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Notes:

Course assessment

Total number of assessed stud	dents: 181
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А	В	С	D	Е	FX
13.81	14.92	22.65	24.31	20.44	3.87
	~				

Provides: prof. RNDr. Milan Žukovič, PhD.

Date of last modification: 14.09.2021

Faculty: Faculty of Science

Course ID: ÚCHV/	Course name: Organic Chemistry
OCHB/10	

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours):

Per week: 3 / 1 **Per study period:** 42 / 14

Course method: present

Number of ECTS credits: 5

Recommended semester/trimester of the course: 2.

Course level: I.

Prerequisities: ÚCHV/VACH/10

Conditions for course completion:

1.Participation in seminars (also applies to the online form of teaching): justified non-participation of the student in two seminars will be excused by the teacher; longer-term justified non-participation of the student in seminars must be proven by the student's mastery of the subject matter in an alternative form determined by the teacher (e.g. preparation of assignments and others...).

2. Activity at seminars (also applies to the online form of teaching) - theoretical preparation of students is required for all seminars.

3.Short written examinations at seminars (max. 50b). Credit slips in the 7th and 14th week with a total sum of 100b. To pass the E rating, it is necessary to obtain 25.5b from each test.

4. The exam is a form of test. A minimum of 51 points is required to pass the exam. The final grade is calculated as the average of the evaluation of papers in seminars, credit papers and the exam itself. Final Grade: A: 91-100b, B: 81-90b, C: 71-80b, D: 61-70b, E: 51-60b, FX: 0-50b.

Learning outcomes:

After completing the course, the student, based on the study of common and different features of compounds, should be able to assess the properties of a given type of compound from the structure and name the corresponding type of compound based on nomenclature principles. From the acquired knowledge about the structure and properties of the relevant types of hydrocarbon compounds, the student should be able to independently derive the mechanisms of individual reactions.

Brief outline of the course:

Recommended literature:

- 1. Online ppt presentations in the system MOODLE na moodle science.upjs.sk
- 2. Organic chemistry, Clayden, Greeves Warren & Wothers, Oxford university Press, 2010.
- 3. Organická chémia, John McMurry, Vysoké učení technické v Brne, 2007, VUTIUM, ISBN: 978-80-214-3291-8 (VUT v Brne).
- 4. Organická chémia, Pavol Zahradník, Mária Mečiarová, Peter Magdolen, Univerzita
- Komenského v Bratislave, 2019, ISBN: 978-80-223-4589-7.

Course language:

anglický

Notes:

Teaching is carried out in person or, if necessary, online, using the MS Teams tool. The form of teaching is specified by the teacher at the beginning of the semester, updated continuously.

Course assessment Total number of assessed students: 299							
A B C D E FX							
20.07	20.74	32.78	19.06	7.02	0.33		
Provides: RNDr. Slávka Hamuľaková, PhD., univerzitná docentka, doc. RNDr. Miroslava Martinková, PhD., univerzitná profesorka, doc. RNDr. Mária Vilková, PhD.							
Date of last modification: 15.08.2022							
Approved: doc. Mgr. Daniel Jancura, PhD.							

University: P. J. Šafárik University in Košice					
Faculty: Faculty of Science					
Course ID: ÚFV/ FCH1/02	Course ID: ÚFV/ Course name: Physical Chemistry for Biological Sciences				
Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 3 / 2 Per study period: 42 / 28 Course method: present					
Number of ECTS credits: 6					
Recommended semester/trimester of the course: 3.					
Course level: I., II.					
Prerequisities:					
Conditions for course completion: Test					

Exam

During an exam, a student should demonstrate his/her ability to solve theoretical exercises from the selected parts of the Physical chemistry for biological sciences. Morever, the students should be able to manifest theoretical knowledge from the chapters which are present in the brief outline of the course.

Learning outcomes:

The introduction into the fundamental knowledge of selected parts of physical chemistry with emphasis on the utilization of these knowledge for the study of physico-chemical properties of biomacromolecules and biological systems. After completing the course, the students should understand physico-chemical mechanisms of many biological processes.

Brief outline of the course:

Week 1

Physical chemistry - areas of research, importance for science, definition. Thermodynamics - definition, areas of interest. Thermodynamic system. Properties of thermodynamic system. Basic thermodynamic quantities (pressure, volume, temperature, internal energy). Zero law of thermodynamics. Ideal gas. Equation of state of an ideal gas. Gas mixtures - Dalton's law. Real gas. Van der Waals equation of state.

Week 2

1st law of thermodynamics. Internal energy, work, heat. Mathematical formulation of the 1st law of thermodynamics. Enthalpy. Heat capacity. Relationship between heat capacities at constant pressure and volume. Isothermal expansion of an ideal gas. Work in reversible and irreversible isothermal expansion. Adiabatic expansion of an ideal gas. Exothermic and endothermic reactions and processes. Standard state of substances. Hess's law.

Week 3

Examples of spontaneous processes in nature. Definitions of the 2nd law of thermodynamics (Kelvin, Celsius). Entropy - introduction of the term. Thermodynamic definition of entropy. Entropy as a state function. Carnot cycle. Efficiency of a heat engine. Claussius inequality. Entropy

of isothermal expansion, gas mixing, melting and evaporation processes. Dependence of entropy on temperature. Nernst's heat theorem. 3rd law of thermodynamics. Week 4

Entropy as a property determining the spontaneity of processes. Criteria of process spontaneity at constant volume and constant pressure. Helmoltz and Gibbs free energy. Properties of Helmoltz energy. Properties of Gibbs energy. Standard Gibbs energy of a chemical reaction. Dependence of Gibbs energy on temperature - Gibbs-Helmoltz equation. Dependence of Gibbs energy on pressure for solids, liquids and gases. Simple mixtures. Partial molar volume. Partial molar Gibbs energy, chemical potential.

Week 5

Chemical potential in a liquid. Raoult's law, the ideal solution. Henry's law, ideally diluted solution. Mixing solutions, ideal solutions. Residual functions and regular solutions. Colligative properties. Increasing the boiling point and decreasing the melting point of the liquid in which the soluble chemical compound is located. Osmosis. Solvent activity, soluble substance activity. Week 6

Chemical equilibrium. Gibbs energy of a chemical reaction. Chemical equilibrium in an ideal gas. Equilibrium constant of chemical reaction. Temperature dependence of the equilibrium constant van't Hoff's equation. Stability of protein structure. Thermal denaturation of proteins. Van't Hoff enthalpy of protein denaturation. Chemical denaturation of proteins. Physiological consequences of incorrectly folded proteins.

Week 7

Examples of molecular associations and their significance for biological systems. Dissociation and association binding constants. Determination of dissociation binding constant - Langmuir isotherm. Cooperativity in ligand-macromolecule interactions. Cooperativity - simultaneous ligand binding, Hill's equation. Cooperativity - gradual binding of ligands. Allosteric interactions. Qualitative description of the Monod - Wyman - Changeaux model for cooperative binding of ligands to macromolecules. Experimental methods used to study the ligand - macromolecule interactions.

Week 8 Chemical and biochemical kinetics - basic definitions. Rates of chemical reactions. Rate constant. Order of chemical reaction. First order reactions. Second order reactions. Consecutive reactions. Determination of the rate law. Reverse chemical reactions. Relaxation processes. Temperature dependence of rate constants - Arrhenius equation. Experimental techniques used to determine the rates of chemical reactions. Transition state theory - Eyring's theory. Week 9

Enzymes - characterization and classification. Equilibrium model of enzyme kinetics. Steady state model of enzyme kinetics. Experimental determination of maximum rate and Michaelis-Menten constant in enzymatic reactions. Deviations from Michaelis-Menten kinetics. Enzyme inhibition. Reversible inhibition. Competitive, non-competitive and uncompetitive inhibition. Week 10

Kinetics of photophysical and photochemical processes. Jablonski diagram. Fluorescence, phosphorescence. Quantum yields of photophysical processes. Quenching of the excited states of molecules by external factors. Fluorescence quenching. Stern-Volmer equation. Förster resonance energy transfer (FRET). Biological application of FRET.

Week 11

Electrochemical reactions. Electrochemical cell. Standard redox potentials. Relationship between Gibbs energy change and electrochemical potential. Temperature dependence of electrochemical potential. Use of electrochemical cells. Determination of redox potential. Ionic electrochemical gradient. Proton motive force. Nernst potential. Introduction to the respiratory chain in mitochondria.

Week 12

Acids and bases. Acid-base properties of water. pH - measurement of environmental acidity. Dissociation of acids and bases - acid-base equilibrium. Henderson - Hasselbalch equation. Buffers.

Recommended literature:

Recommended	itterature:						
1. P. Atkins and	l J. de Paula. Atk	ins's Physical Ch	nemistry (9th Edi	tion), Oxford			
University Pres	s, 2010.						
2. P. Atkins. Fyzikálna chémia (slovenský preklad 6. vydania), STU Bratislava, 1999.							
3. P. Atkins, J. De Paula. Fyzikální chemie (český preklad 9. vydania), VŠCHT Praha,							
2013							
4. R.Chang. Phy	ysical Chemistry	for the Bioscien	ces, University S	cience Book, 200)6.		
5. D. Eisenberg	and D. Crothers	. Physical Chemi	stry with Applica	ations to the Life			
Sciences, Benja	min/Cummings,	1979.					
6. K. van Holde	e, W. Johnson and	d P. Ho. Principle	es of Physical Bio	ochemistry, Prent	ice		
Hall, 1988.							
7. D.T. Haynie.	Biological There	modynamics (2nd	d Edition), Camb	ridge University	Press,		
2008.							
8. A.P.H. Peters	. Concise Chemi	ical Thermodyna	mics (3rd Edition	n), CRC Press, Ta	ylor &		
Francis Group,	2010.						
9. I. Tinoco, jr.,	K. Sauer, J.C. W	/ang, J.C. Puglisi	, G. Harbison and	d D.Rovnyak.			
Physical Chemi	stry – Principles	and Application	s in Biological So	ciences (5th Editi	on),		
Pearson, 2014.							
10. A. Cooksy.	Physical Chemis	try- Thermodyna	mics, Statistical	Mechanics, and			
Kinetics, Pearso	on, 2014.						
Course languag	ze:						
English languag	ge						
Notes:							
Course assessm	ient						
Total number of	f assessed studen	ts: 118					
A B C D E FX							
18.64 27.97 33.05 11.02 9.32 0.0							
Provides: doc. 1	Mgr. Daniel Janc	ura, PhD.					
Date of last mo	dification: 17.09	9.2021					
Approved: doc.	Mgr. Daniel Jan	cura, PhD.					

University: P. J. Šafá	rik University in Košice					
Faculty: Faculty of Science						
Course ID: ÚFV/ ZFP1a/03	Course name: Physics Practical I					
Course type, scope a Course type: Practic Recommended cour Per week: 3 Per stu Course method: pre	nd the method: ce rse-load (hours): dy period: 42 esent					
Number of ECTS cr	edits: 3					
Recommended seme	ster/trimester of the course: 2.					
Course level: I.						
Prerequisities:						
Conditions for cours The active work durin Vindication of report	e completion: ng semester and hand in all reports. s.					
Learning outcomes: Developing proper la	boratory habits, skills and verify their theoretical knowledge.					
Brief outline of the c The goal of this labo with kinds and calcu results. The students introductory physics Laboratory assignme 1. Density measurem 2. Radius measureme surface using planime 3. Gravitational accel and physical pendulu	ourse: pratory exercises is to familiarize the students with measurement methods, lus of mistakes, with measured results processing, and with presentation of gain practical skills, and verify their theoretical knowledge of first semester course. They develop proper laboratory habits. nt: ents of liquids and solids. ents of spherical cap. Measurements of eter. leration measurements using mathematical m					
4. Moment of inertia pendulum.5. Measurements of Y	 and physical pendulum. 4. Moment of inertia measurement using physical and torsion pendulum. 5. Measurements of Young's modulus. 					
 6. Measurement of coefficient of viscosity. 7. Measurement of the speed of sound. 8. Measurements of general gas constant and Boltzmann constant. 9. Measurements of thermal expansivity of air. 10. Measurements of thermal capacity of matter. 11 Measurement of the surface tension 						
Recommended litera Degro, J., Ješková, Z measurements I), Ed. Standards STN ISO 3 standards in Bratislav	nture: ., Onderová, Ľ., Kireš, M.: Základné fyzikálne praktikum I. (Basic physical PF UPJŠ Košice 2007. 81. Slovenský inštitút normalizácie v Bratislave (Slovak institute of technical va),1997.					

Ješková, Z.: Computer based experiments in thermodynamics using IP COACH,ed. PF UPJŠ in Košice, 2004.

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Course language: english								
Notes:	Notes:							
Course assessment Total number of assessed students: 278								
А	В	B C D E FX						
56.83	56.83 25.54 12.95 3.96 0.72 0.0							
Provides: doc. RNDr. Marián Kireš, PhD., doc. RNDr. Jozef Hanč, PhD.								
Date of last modification: 29.03.2020								
Approved: doc. Mgr. Daniel Jancura, PhD.								

Faculty: Faculty of Science

Course ID: ÚFV/	Course name: Physics Practical II
ZFP1b/03	

Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 3 Per study period: 42 Course method: present

Number of ECTS credits: 3

Recommended semester/trimester of the course: 3.

Course level: I.

Prerequisities: (ÚFV/ZFP1a/03 or ÚFV/ZFP1a/22)

Conditions for course completion:

To successfully complete the course, the student must measure at least 11 experimental tasks, process and analyze the measured results and evaluate the experimental results in the form of a protocol.

The condition for the implementation of another experimental task is the submission of a protocol from the previous exercise.

The condition for the implementation of the practical task is sufficient theoretical training at home. If the student is not ready for the task in advance, the teacher can send him home and the student must replace the exercise at another time.

The credit evaluation of the course takes into account the following student workload:

1 credit: self-study of recommended literature and subsequent direct teaching

1 credits: realization of experimental exercise and subsequent defense of measuring procedure - it is obligatory to complete all practical tasks in the semester,

1 credit: elaboration and submission of protocols from measurements, which are evaluated.

Learning outcomes:

By completing the course, the student will get acquainted with selected physical experiments in the field of electricity and magnetism and supplement the theoretical knowledge acquired in the course General Physics in a practical way.

The result of education is:

a) Complementing and summarizing knowledge and experimental skills in the field of electricity and magnetism.

b) Gaining practical experience with recording, analysis and interpretation of experimental data from practical measurements.

c) Gaining experience with the presentation of experimental results in the form of a measurement protocol.

Brief outline of the course:

Students on practical exercises are working in pairs experimental tasks in the field of electrical, electromagnetic and magnetic properties of matters.

1. Electrical Resistivity

2. Self - and Mutual Inductance and Capacity

- 3. Serial and Parallel Resonance
- 4. Thermal Dependence of Selected Electrical Phenomena in Solids
- 5. The Characteristics of Semiconductor Diod
- 6. The Characteristics of Semiconductor Bipolar Transistor
- 7. Magnetic Hysteresis
- 8. Hall Constant Measurements
- 9. Measurements of Horizontal Component of Earth Magnetic Field
- 10. Measuring characteristics of switching components
- 11. Measuring the properties of optoelectronic components
- 12. Electric current in liquids and electrolysis

Recommended literature:

- 1. Tumanski S, Handbook of magnetic measurements, CRC press, 2011.
- 2. Fiorillo F, Characterization and Measurement of Magnetic Materials, Elsevier, 2004.

Course language:

english

Notes:

Teaching is carried out in person. If necessary, part of the teaching can be realized remotely using the MS Teams or BBB tool. At the beginning of the semester, the teacher sets the conditions for completing and mastering the course.

Course assessment

Total number of assessed students: 264

А	В	С	D	Е	FX
67.42	18.94	11.74	1.52	0.0	0.38

Provides: doc. RNDr. Adriana Zeleňáková, PhD., doc. RNDr. Ján Füzer, PhD.

Date of last modification: 30.09.2021

University: P. J	. Šafárik Univer	sity in Košice						
Faculty: Faculty of Science								
Course ID: ÚF ZFP1c/14	Course ID: ÚFV/ Course name: Physics Practical III CFP1c/14 Course name: Physics Practical III							
Course type, so Course type: 1 Recommended Per week: 3 P Course metho	ope and the m Practice d course-load (er study period d: present	ethod: hours): l: 42						
Number of EC	TS credits: 3							
Recommended	semester/trim	ester of the cours	se: 4.					
Course level: I.								
Prerequisities:								
Conditions for Measurements defended. As a of the task.	course comple of experimental part of evaluation	tion: tasks, their evalua on there is is also a	ation in the form good theoretica	of a written report ll preparation for th	t, which must be ne measurement			
Learning outco To gain some p practice in data report writing p	mes: hysical inside in collection, and presentation and	nto some of the co alysis and interpr results.	oncepts presente etation of resun	ed in the lectures. In nance. c. To gain	b. To gain some experience and			
Brief outline of Oscilations. Per sound. Refracti of waves. Polar	the course: ndulum. Compo ve index. Lense ization. The spe	sition and decom 's focal length. In eed of light. Quan	position of osci tterference. Diff tum optics.	llations. Resonanc raction. Diffraction	e. The speed of n and reflection			
Recommended Degro,J., Ješko 2006 P. Kollár a kol. J. Brož Základy	literature: vá, Z., Onderov Základné fyzika v fysikálních mě	á,Ľ., Kireš,M.: Za álne praktikum II, ření, SPN Praha,	ákladné fyzikáln PF UPJŠ Košic 1981.	e praktikum I, PF ce, 2006	UPJŠ Košice,			
Course languag slovak, english	ge:							
Notes:								
Course assessment Total number of assessed students: 99								
А	В	C	D	Е	FX			
66.67	66.67 21.21 7.07 2.02 3.03 0.0							
Provides: doc. RNDr. Marián Kireš, PhD., doc. RNDr. Ján Füzer, PhD., RNDr. Samuel Dobák, PhD.								
Date of last mo	dification: 01.0	02.2022	_					

University: P. J	. Šafárik Univers	ity in Košice			
Faculty: Faculty of Science					
Course ID: ÚF PEMBF1/18	V/ Course na	Course name: Practical exercises in experimental methods of biophysics I			
Course type, sc Course type: I Recommended Per week: 2 Pe Course metho	cope and the met Practice d course-load (h er study period: d: present	thod: ours): 28			
Number of EC	TS credits: 2		~		
Recommended	semester/trimes	ster of the cours	e: 5.		
Course level: 1.	,				
Prerequisities:	ÚFV/EMBF1/18	,			
Conditions for course completion:					
Learning outco	Learning outcomes:				
Brief outline of	Brief outline of the course:				
Recommended literature:					
Course language:					
Notes:					
Course assessm Total number of	nent f assessed studen	ts: 7			
А	B C D E FX				
71.43	28.57	28.57 0.0 0.0 0.0 0.0			
Provides: RNDr. Gabriela Fabriciová, PhD.					
Date of last mo	dification: 25.11	.2021			
Approved: doc.	. Mgr. Daniel Jan	icura, PhD.			

II	Č - C' -: 1- I I -:	:			
University: P. J.	. Safarik Univers	sity in Kosice			
Faculty: Faculty	y of Science				
Course ID: ÚF PEMBF2/18	V/ Course na	Course name: Practical exercises in experimental methods of biophysics II			
Course type, sc Course type: H Recommended Per week: 2 Po Course metho	ope and the met Practice d course-load (h er study period: d: present	thod: ours): 28			
Number of EC	1 S credits: 2				
Recommended	semester/trimes	ster of the cours	e: 5.		
Course level: I.					
Prerequisities:	ÚFV/EMBF2/18	5			
Conditions for	course completi	on:			
Learning outco	mes:				
Brief outline of	the course:				
Recommended	literature:				
Course languag	ge:				
Notes:					
Course assessm Total number of	ent f assessed studen	ts: 6			
А	В	С	D	E	FX
100.0	0.0	0.0	0.0	0.0	0.0
Provides: prof. Žoldák, DrSc.	RNDr. Erik Sedl	ák, DrSc., RND1	. Gabriela Fabric	ciová, PhD., doc.	RNDr. Gabriel
Date of last mo	dification: 30.11	.2021			
Approved: doc.	Mgr. Daniel Jan	cura, PhD.			

Faculty: Faculty of Science

Course ID: ÚFV/	Course name: Practical exercises in experimental methods of biophysics III
PEMBF3/18	

Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: present

Number of ECTS credits: 2

Recommended semester/trimester of the course: 6.

Course level: I.

Prerequisities: ÚFV/EMBF3/18

Conditions for course completion:

Elaboration of protocols.

Learning outcomes:

The students gain experimental skills in the methods of confocal fluorescence microscopy, time-resolved fluorescence microscopy (FLIM / PLIM), flow cytometry, microfluidics and the development of experimental apparatuses in the field of biophotonics.

Brief outline of the course:

1-2. Sample preparation for confocal fluorescence microscopy and vital fluorescence imaging in cells.

3-4. Time-resolved fluorescence microscopy (FLIM) - sample preparation and fluorescence lifetime imaging in cells.

5. Immuno-fluorescence imaging - sample preparation for confocal fluorescence microscopy and imaging of selected proteins in cells.

6. Flow cytometry - detection of fluorescent labels in living cells.

7. Setting up the function generator and the digital oscilloscope (the influence of input resistance, triggering and averaging of time courses). Noise monitoring, optimization of shielding and grounding of devices. Measurement using a laboratory card, evaluation of spectral noise density.

8. Safety training for working with optical radiation in practice. Safe work with lasers, setting the beam path through a cuvette. Measuring fluorescence spectra using a spectrograph.

9. Measurement of weak optical signals. Comparison of different types of photodiodes. Connection and setting of the lock-in amplifier when measuring with a chopped laser beam.

10. Temperature regulation and measurement. Connection and adjustment of a PID temperature regulator using a Peltier cell in combination with a thermocouple.

11. Preparation of a simple microfluidic system. Channel design. Mold preparation. Fabrication of a microfluidic system using PDMS.

12. Setting up the imaging system using a CMOS camera. Measurement of fluid flow rate in a microfluidic system based on video recording.

Recommended literature:

1. Wolfgang Becker: The bh TCSPC Handbook Seventh Edition, Becker & Hickl GmbH 2017;

2. Guy Cox: Optical Imaging Techniques in Cell Biology, Taylor & Frances;

 Howard M. Shapiro: Practical Flow Cytometry Fourth edition, 2003; Nikolas Long and Wing-Tak Wong: The chemistry of molecular imaging, Wiley 2014 					
Course language: Slovac language, English.					
Notes:					
Course assessn Total number o	nent f assessed studen	ıts: 5			
А	B C D E FX				
100.0	100.0 0.0 0.0 0.0 0.0 0.0				
Provides: doc. Mgr. Gregor Bánó, PhD., RNDr. Veronika Huntošová, PhD.					
Date of last modification: 22.09.2021					
Approved: doc. Mgr. Daniel Jancura, PhD.					

University: P. J. Šafa	árik University in Košice	
Faculty: Faculty of S	Science	
Course ID: ÚFV/ KVM I/11	Course name: Quantum Mechanics	
Course type, scope a Course type: Lectu Recommended cou Per week: 4 / 2 Per Course method: pr	and the method: re / Practice irse-load (hours): • study period: 56 / 28 resent	
Number of ECTS ci	redits: 8	
Recommended sem	ester/trimester of the course: 3.	
Course level: I.		
Prerequisities:		
Conditions for cour	se completion:	

To successfully complete the course, the student must demonstrate sufficient understanding of the basic terms, concepts and applications of quantum physics. Knowledge of basic concepts of quantum physics is required at the level of their mathematical definition, as well as their physical content and specific applications. During the semester, the student must continuously master the content of the curriculum so that he can actively and creatively use the acquired knowledge in solving specific computational tasks during the exercises and pass continuous written tests taken into account in the overall evaluation of the subject. The condition for obtaining credits is passing 2 continuous written tests in exercises and an oral exam, which consists of one more demanding calculation task and theoretical questions. The credit evaluation of the course takes into account the following student workload: direct teaching (3 credits), self-study (2 credits), individual consultations (1 credit) and assessment (2 credits). The minimum threshold for completing the course is to obtain at least 50% of the total score, using the following rating scale: A (90-100%), B (80-89%), C (70-79%), D (60- 69%), E (50-59%), F (0-49%).

Learning outcomes:

After completing lectures and exercises, the student will have sufficient physical skills,

knowledge and mathematical apparatus enabling independent solution of a wide range

traditional and current scientific problems in quantum physics. At the same time, he will gain an overview of the applications of quantum physics in various areas of physics - nuclear physics, condensed matter physics, statistical physics, quantum information theory, etc.

Brief outline of the course:

1. Subject matter, experimental and theoretical foundations of quantum mechanics (QM).

2. Wave formulation of QM. Postulate about wave function, superposition principle and postulate about operators.

3. Eigenvalues and eigenfunctions of operators. Measurement of quantities and reduction of wave function.

4. Time-independent and time-dependent Schrödinger equation. Ehrenfest equations and integrals of motion. Continuity equation.

5. Matrix formulation of QM, Dirac symbolics, calculation of mean values and density matrix.

6. Current immeasurability of physical quantities, Heisenberg uncertainty relations.

7. Solution of the Schrödinger equation for a particle in an infinitely deep potential well and a particle in a finite potential well. Bound and scattering states.

8. Passage of a particle through a potential barrier: tunneling and barrier reflection.

9. Solution of Schrödinger equation for linear harmonic oscillator.

10. Particle motion in the central potential field, angular part of the Schrödinger equation.

11. Particle motion in the central potential field, radial part of the Schrödinger equation. Hydrogen atom.

12. Electron spin, Pauli matrix. Principle of indistinguishability of identical particles, fermions and bosons. Pauli's exclusion principle.

13. Paradoxes and modern problems of QM. Quantum entanglement, nonlocality, computing, cryptography and teleportation.

Recommended literature:

1. Ľ. Tóth, M. Tóthová, Kvantová a štatistická fyzika I, Rektorát Univerzity P. J. Šafárika, 1982. (in Slovak language)

2. Ľ. Skála, Úvod do kvantovej mechaniky, Academia, Praha, 2005. (in Czech language)

3. J. Pišút, L. Gomolčák, Úvod do kvantovej mechaniky, Bratislava 1983. (in Slovak language)

4. W. Greiner, Quantum Mechanics, 4th edition, Springer, Berlin, 2000.

5. A. C. Philips, Introduction to Quantum Mechanics, Wiley, Weinheim, 2003.

6. D. J. Griffiths, Introduction to Quantum Mechanics, Prentice Hall, New Jersey, 1995.

7. G. Auletta, M. Fortunato, G. Parisi, Quantum Mechanics, Cambridge University Press, Cambridge, 2009.

Course language:

Notes:

Course assessment

Total number of assessed students: 111

А	В	С	D	Е	FX
23.42	21.62	18.92	11.71	18.02	6.31

Provides: doc. RNDr. Jozef Strečka, PhD.

Date of last modification: 19.09.2021

University: P. J. Šafá	rik University in Košice	
Faculty: Faculty of S	Science	
Course ID: ÚFV/ Course name: Quantum Mechanics II. KVM II/08		
Course type, scope a Course type: Lectu Recommended cou Per week: 3 / 1 Per Course method: pro	and the method: re / Practice rse-load (hours): study period: 42 / 14 esent	
Number of ECTS cr	redits: 6	
Recommended seme	ester/trimester of the course: 4.	
Course level: I		

Prerequisities: ÚFV/KVM/08 or ÚFV/KVM I/11

Conditions for course completion:

To successfully complete the course, the student must demonstrate sufficient understanding of the basics terms, concepts and applications of quantum physics. Knowledge of basic concepts is required from quantum physics at the level of their mathematical definition as well as their physical content and concrete applications. During the semester, the student must continuously master the content of the curriculum in order to gain the acquired knowledge in order to actively and creatively use them in solving specific tasks during the exercises and complete the continuous written test taken into account in the overall evaluation of the subject. The condition for obtaining credits is passing 1 continuous written test in exercises and an oral exam, which consists of one more complex computational task and theoretical questions. The credit evaluation of the course takes into account the following student workload: direct teaching (2 credits), self-study (2 credits), individual consultations (1 credit) and assessment (1 credit). Minimum threshold for passing the subject is to obtain at least 50% of the total score, while it is used the following rating scale: A (90-100%), B (80-89%), C (70-79%), D (60-69%), E (50-59%), F (0-49%).

Learning outcomes:

After completing lectures and exercises, the student will have sufficient physical skills, knowledge and mathematical apparatus enabling independent solution of a wide range traditional and current scientific problems in quantum physics using approximate methods. At the same time, he will gain an overview of the applications of quantum physics in various fields of

same time, he will gain an overview of the applications of quantum physics in various fields of physics such as atomic and nuclear physics, condensed matter physics, statistical physics, quantum theory of magnetism, etc.

Brief outline of the course:

1. Stationary perturbation theory for non-degenerate quantum-mechanical systems with discrete energy spectrum.

2. Stationary perturbation theory for degenerate quantum-mechanical systems with discrete energy spectrum. Zeeman and Stark effects.

3. Stationary perturbation theory for two-level quantum-mechanical systems with two closelyspaced energy levels: crossing and selfavoided crossing of energy levels.

4. Ritz's variational method. Bound state of a quantum-mechanical system with attractive potential.

5. Applications of Ritz's variation method in finding the ground state of quantum spin models.

6. Nonstationary perturbation theory for non-degenerate quantum-mechanical systems with discrete energy spectrum. A special case of constant, adiabatic and short-rapid perturbation.

7. Nonstationary perturbation theory for quantum-mechanical systems with discrete-continuous energy spectrum. Harmonic perturbation and Fermi's golden rule.

8. Quantum-mechanical solution of the time-indepedent Schrődinger equation for the helium atom using stationary perturbation theory. Orthohelium and parahelium.

9. Quantum-mechanical solution of the time-indepedent Schrödinger equation for a hydrogen molecule using stationary perturbation theory. Heitler-London theory of valence bonds.

10. Quantum-mechanical solution of the time-indepedent Schrödinger equation for a hydrogen molecule using the Ritz variational method. LCAO method.

11. Hartree and Hartree-Fock method for multielectron atoms.

Recommended literature:

1. V. Ilkovič, Kvantová teória II, Scriptum UPJŠ, Košice, 1989. (in Slovak)

- 2. J. Pišút, L. Gomolčák, Úvod do kvantovej mechaniky, Bratislava 1983. (in Slovak)
- 3. W. Greiner, Quantum Mechanics, 4th edition, Springer, Berlin, 2000.
- 4. D. J. Griffiths, Introduction to Quantum Mechanics, Prentice Hall, New Jersey, 1995.

5. G. Auletta, M. Fortunato, G. Parisi, Quantum Mechanics, Cambridge University Press, Cambridge, 2009.

Course language:

EN - english

Notes:

Course assessment

Total number of assessed students: 120

А	В	С	D	Е	FX
30.0	15.0	16.67	15.0	19.17	4.17

Provides: doc. RNDr. Jozef Strečka, PhD.

Date of last modification: 19.09.2021

University: P. J. Šafán	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚTVŠ/ ÚTVŠ/CM/13	Course name: Seaside Aerobic Exercise
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	nd the method: ce cse-load (hours): dy period: 28 sent
Number of ECTS cro	edits: 2
Recommended seme	ster/trimester of the course:
Course level: I., II.	
Prerequisities:	
Conditions for cours Completion: passed Condition for success - active participation - effective performance	e completion: ful course completion: in line with the study rule of procedure and course guidelines ce of all tasks- aerobics, water exercise, yoga, Pilates and others
Learning outcomes: Content standard: The student demonstr course syllabus and re Performance standard Upon completion of t - perform basic aerob - conduct verbal and re - organise and manag	ates relevant knowledge and skills in the field, which content is defined in the ecommended literature. I: he course students are able to meet the performance standard and: ics steps and basics of health exercises, non-verbal communication with clients during exercise, e the process of physical recreation in leisure time
 Brief outline of the c Brief outline of the co 1. Basic aerobics – lo 2. Basics of aqua fithe 3. Basics of Pilates 4. Health exercises 5. Bodyweight exercises 5. Bodyweight exercises 6. Swimming 7. Relaxing yoga exercises 8. Power yoga 9. Yoga relaxation 10. Final assessment Students can engage volleyball, football, tag 	ourse: burse: w impact aerobics, high impact aerobics, basic steps and cuing ess ses rcises in different sport activities offered by the sea resort – swimming, rafting, able tennis, tennis and other water sports in particular.
Recommended litera 1. BUZKOVÁ, K. 20	ture: 06. Fitness jóga. Praha: Grada. 167 s.

 ŽECHOVSKÁ, I., MILEROVÁ, H., NOVOTNÁ, V. Aqua-fitness. Praha: Grada. 136 s. EVANS, M., HUDSON, J., TUCKER, P. 2001. Umění harmonie: meditace, jóga, tai-či, strečink. 192 s. JARKOVSKÁ, H., JARKOVSKÁ, M. 2005. Posilováni s vlastním tělem 417 krát jinak. Praha: Grada. 209 s. KOVAŘÍKOVÁ, K. 2017. Aerobik a fitness. Karolium, 130 s. 		
Course language: Slovak language		
Notes:		
Course assessment Total number of assessed students: 54		
abs	n	
11.11 88.89		
Provides: Mgr. Agata Dorota Horbacz, PhD.		
Date of last modification: 29.03.2022		
Approved: doc. Mgr. Daniel Jancura, PhD.		

University: P. J. Šafá	University: P. J. Šafárik University in Košice			
Faculty: Faculty of S	cience			
Course ID: ÚFV/ SPBFb1/18	Course name: Semestral thesis I			
Course type, scope a Course type: Recommended cou Per week: Per stud Course method: pre	and the method: rse-load (hours): ly period: esent			
Number of ECTS cr	redits: 4			
Recommended seme	ester/trimester of the course: 5.			
Course level: I.				
Prerequisities:				
Conditions for course Successful completing tasks set by the proj- required level. The a the project leader is u	se completion: ag the course, requires the student to demonstrate adequate level of the assigned ect leader at the beginning of the semester to the required extent and at the assignments are formulated by the teacher at the beginning of the semester, usually the supervisor of the final thesis. Tasks include e.g. study of literature			

preparation and implementation of the experiment, processing of the obtained data, or collaborating during the preparation of a scientific publication. Credit evaluation takes into account the time requirements of the student when working on a semester project in the range of 50 hours per semester. Individual activities of the student are evaluated by the project leader, the overall work of the student is evaluated by points on a point scale of 0 - 100 points. The minimum threshold for obtaining a rating is 50% of the rating scale, which is determined as follows: A 100-91% B 90-81% C 80-71% D 70-61% E 60-50% Fx 49-0%.

in the field, mastering the operation of experimental equipment, sample preparation technology,

Learning outcomes:

After completing the course, the student will acquire knowledge and skills associated with scientific work in the field of biophysics. By actively participating in individual research teams, students will extend their knowledge in the relevant part of biophysics, acquire experimental skills in operating contemporary scientific equipment, study of the literature will improve their language skills. Data processing resp. the creation of original software will improve their computer skills.

Brief outline of the course:

Program for semestral project is prepared individually for each student by supervisor of the project at the beginning of each semester and can be focused on search in literature for a selected area of research, preparation of experiment and its performing, creation of software for data acquisition and analysis, collaboration during preparation of manuscript, presentation of the obtained results for department audience. Supervisor of the project will specify the topic of the project.

Recommended literature:

Selected scientific journals and books.

Course language:

Notes:

Subject Semester work I is realized in attendance form. If necessary (e.g. Covid pandemic) it is		
taught online using software MS Teams, which allows to maintain contact with students even in		
adverse conditions and also allows to meet the requirements of the subject.		
Course assessment		
Total number of assessed students: 6		
abs	n	

100.0	0.0	
Provides:		
Date of last modification: 25.02.2022		
Approved: doc. Mgr. Daniel Jancura, PhD.		

University: P. J. Šafárik University in Košice		
Faculty: Faculty of Science		
Course ID: ÚFV/ SPBFb2/18	Course name: Semestral thesis II	
Course type, scope and the method: Course type: Recommended course-load (hours): Per week: Per study period: Course method: present		
Number of ECTS credits: 6		
Recommended semester/trimester of the course: 6.		
Course level: I.		
Prerequisities:		
Conditions for cours Successful completin tasks set by the proje	se completion: ag the course, requires the student to demonstrate adequate level of the assigned ect leader at the beginning of the semester to the required extent and at the	

tasks set by the project leader at the beginning of the semester to the required extent and at the required level. The assignments are formulated by the teacher at the beginning of the semester, the project leader is usually the supervisor of the final thesis. Tasks include e.g. study of literature in the field, mastering the operation of experimental equipment, sample preparation technology, preparation and implementation of the experiment, processing of the obtained data, or collaborating during the preparation of a scientific publication. Credit evaluation takes into account the time requirements of the student when working on a semester project in the range of 50 hours per semester. Individual activities of the student are evaluated by the project leader, the overall work of the student is evaluated by points on a point scale of 0 - 100 points. The minimum threshold for obtaining a rating is 50% of the rating scale, which is determined as follows: A 100-91% B 90-81% C 80-71% D 70-61% E 60-50% Fx 49-0%.

Learning outcomes:

After completing the course, the student will acquire knowledge and skills associated with scientific work in the field of biophysics. By actively participating in individual research teams, students will extend their knowledge in the relevant part of biophysics, acquire experimental skills in operating contemporary scientific equipment, study of the literature will improve their language skills. Data processing resp. the creation of original software will improve their computer skills.

Brief outline of the course:

Program for semestral project is prepared individually for each student by supervisor of the project at the beginning of each semester and can be focused on search in literature for a selected area of research, preparation of experiment and its performing, creation of software for data acquisition and analysis, collaboration during preparation of manuscript, presentation of the obtained results for department audience. Supervisor of the project will specify the topic of the project.

Recommended literature:

Selected scientific journals and books.

Course language:

Notes:

Subject Semester work I is realized in attendance form. If necessary (e.g. Covid pandemic) it is taught online using software MS Teams, which allows to maintain contact with students even in adverse conditions and also allows to meet the requirements of the subject.		
Course assessment Total number of assessed students: 5		
n		
0.0		
Date of last modification: 30.03.2022		
University: P. J. Šafárik	University in Košice	
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Faculty: Faculty of Scien	nce	
Course ID: ÚTVŠ/ Co TVa/11	ourse name: Sports Activities I.	
Course type, scope and Course type: Practice Recommended course- Per week: 2 Per study Course method: preser	the method: -load (hours): period: 28 nt	
Number of ECTS credi	ts: 2	
Recommended semester	r/trimester of the course: 1.	
Course level: I., II.		
Prerequisities:		
Conditions for course c Min. 80% of active parti	ompletion: cipation in classes.	
Learning outcomes: Sports activities in all the They have a great impace enables students to stree improve.	Fir forms prepare university students for their professional and personal life. et on physical fitness and performance. Specialization in sports activities ngthen their relationship towards the selected sport in which they also	
Brief outline of the cours Brief outline of the cours The Institute of physical activities aerobics; aikid yoga, power yoga, pilat tennis, chess, volleyball, Additionally, the Institu offers winter courses (sl the Tisza River) with an participation.	rse: se: education and sport at the Pavol Jozef Šafárik University offers 20 sports o, basketball, badminton, body-balance, body form, bouldering, floorball, es, swimming, fitness, indoor football, SM system, step aerobics, table tabata, cycling. te of physical education and sport at the Pavol Jozef Šafárik University ki course, survival) and summer courses (aerobics by the sea, rafting on attractive programme, sports competitions with national and international	
Recommended literatur BENCE, M. et al. 2005. [online] Dostupné na: ht BUZKOVÁ, K. 2006. F 8024715252. JARKOVSKÁ, H, JARH Grada. ISBN 978802475 KAČÁNI, L. 2002. Futb 8089197027. KRESTA, J. 2009. Futsa LAWRENCE, G. 2019. SNER, Wolfgang. 2004.	 Plávanie. Banská Bystrica: FHV UMB. 198s. ISBN 80-8083-140-8. tps://www.ff.umb.sk/app/cmsFile.php?disposition=a&ID=571 itness jóga, harmonické cvičení těla I duše. Praha: Grada. ISBN KOVSKÁ, M. 2005. Posilování s vlastním tělem 417 krát jinak. Praha: 67308. al:Tréning hrou. Bratislava: Peter Mačura – PEEM. 278s. ISBN I.Praha: Grada Publishing, a.s. 112s. ISBN 9788024725345. Power jóga nejen pro sportovce. Brno: CPress. ISBN 9788026427902. Posilování ve fitness. České Budějovice: Kopp. ISBN 8072322141. 	

STACKEOVÁ, D. 2014. Fitness programy z pohledu kinantropologie. Praha: Galén. ISBN 9788074921155.

VOMÁČKO, S. BOŠTÍKOVÁ, S. 2003. Lezení na umělých stěnách. Praha: Grada. 129s. ISBN 8024721743.

Course language:

Slovak language

Notes:

Course assessment

Total number of assessed students: 15193

abs	abs-A	abs-B	abs-C	abs-D	abs-E	n	neabs
86.05	0.07	0.0	0.0	0.0	0.05	8.69	5.15

Provides: Mgr. Patrik Berta, Mgr. Agata Dorota Horbacz, PhD., Mgr. Dávid Kaško, PhD., Mgr. Ladislav Kručanica, PhD., Mgr. Richard Melichar, Mgr. Petra Tomková, PhD., Mgr. Marcel Čurgali, Mgr. Alena Buková, PhD., doc. PaedDr. Ivan Uher, MPH, PhD., prof. RNDr. Stanislav Vokál, DrSc., Mgr. Zuzana Küchelová, PhD.

Date of last modification: 07.02.2024

University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚTVŠ/ TVb/11	Course name: Sports Activities II.
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	nd the method: ce cse-load (hours): dy period: 28 esent
Number of ECTS cr	edits: 2
Recommended seme	ster/trimester of the course: 2.
Course level: I., II.	
Prerequisities:	
Conditions for cours active participation ir	e completion: n classes - min. 80%.
Learning outcomes: Sports activities in all They have a great im enables students to s improve.	their forms prepare university students for their professional and personal life. apact on physical fitness and performance. Specialization in sports activities strengthen their relationship towards the selected sport in which they also
Brief outline of the c Brief outline of the co The Institute of physi activities aerobics; ai yoga, power yoga, p tennis, chess, volleyb Additionally, the Inst offers winter courses the Tisza River) with participation.	ourse: ourse: cal education and sport at the Pavol Jozef Šafárik University offers 20 sports kido, basketball, badminton, body-balance, body form, bouldering, floorball, ilates, swimming, fitness, indoor football, SM system, step aerobics, table all, tabata, cycling. titute of physical education and sport at the Pavol Jozef Šafárik University (ski course, survival) and summer courses (aerobics by the sea, rafting on an attractive programme, sports competitions with national and international
Recommended litera BENCE, M. et al. 200 [online] Dostupné na BUZKOVÁ, K. 2006 8024715252. JARKOVSKÁ, H, JA Grada. ISBN 978802 KAČÁNI, L. 2002. F 8089197027. KRESTA, J. 2009. Fu LAWRENCE, G. 201 SNER, Wolfgang. 20	 ture: D5. Plávanie. Banská Bystrica: FHV UMB. 198s. ISBN 80-8083-140-8. https://www.ff.umb.sk/app/cmsFile.php?disposition=a&ID=571 Fitness jóga, harmonické cvičení těla I duše. Praha: Grada. ISBN ARKOVSKÁ, M. 2005. Posilování s vlastním tělem 417 krát jinak. Praha: 4757308. utbal:Tréning hrou. Bratislava: Peter Mačura – PEEM. 278s. ISBN utsal.Praha: Grada Publishing, a.s. 112s. ISBN 9788024725345. Power jóga nejen pro sportovce. Brno: CPress. ISBN 9788026427902. Posilování ve fitness. České Budějovice: Kopp. ISBN 8072322141.

STACKEOVÁ, D. 2014. Fitness programy z pohledu kinantropologie. Praha: Galén. ISBN 9788074921155.

VOMÁČKO, S. BOŠTÍKOVÁ, S. 2003. Lezení na umělých stěnách. Praha: Grada. 129s. ISBN 8024721743.

Course language:

Slovak language

Notes:

Course assessment

Total number of assessed students: 13318

abs	abs-A	abs-B	abs-C	abs-D	abs-E	n	neabs
84.37	0.51	0.02	0.0	0.0	0.05	10.78	4.28

Provides: Mgr. Agata Dorota Horbacz, PhD., Mgr. Dávid Kaško, PhD., Mgr. Marcel Čurgali, Mgr. Patrik Berta, Mgr. Ladislav Kručanica, PhD., Mgr. Richard Melichar, Mgr. Petra Tomková, PhD., Mgr. Alena Buková, PhD., doc. PaedDr. Ivan Uher, MPH, PhD., prof. RNDr. Stanislav Vokál, DrSc., Mgr. Zuzana Küchelová, PhD.

Date of last modification: 07.02.2024

University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚTVŠ/ TVc/11	Course name: Sports Activities III.
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	nd the method: ce rse-load (hours): dy period: 28 esent
Number of ECTS cr	edits: 2
Recommended seme	ster/trimester of the course: 3.
Course level: I., II.	
Prerequisities:	
Conditions for cours min. 80% of active pa	e completion: articipation in classes
Learning outcomes: Sports activities in all They have a great im enables students to s improve.	their forms prepare university students for their professional and personal life. spact on physical fitness and performance. Specialization in sports activities strengthen their relationship towards the selected sport in which they also
Brief outline of the c Brief outline of the co The Institute of physi activities aerobics; ai yoga, power yoga, p tennis, chess, volleyb Additionally, the Inst offers winter courses the Tisza River) with participation.	ourse: ourse: cal education and sport at the Pavol Jozef Šafárik University offers 20 sports kido, basketball, badminton, body-balance, body form, bouldering, floorball, ilates, swimming, fitness, indoor football, SM system, step aerobics, table all, tabata, cycling. titute of physical education and sport at the Pavol Jozef Šafárik University (ski course, survival) and summer courses (aerobics by the sea, rafting on an attractive programme, sports competitions with national and international
Recommended litera BENCE, M. et al. 200 [online] Dostupné na BUZKOVÁ, K. 2006 8024715252. JARKOVSKÁ, H, JA Grada. ISBN 978802 KAČÁNI, L. 2002. F 8089197027. KRESTA, J. 2009. Fu LAWRENCE, G. 201 SNER, Wolfgang. 20	 ture: D5. Plávanie. Banská Bystrica: FHV UMB. 198s. ISBN 80-8083-140-8. https://www.ff.umb.sk/app/cmsFile.php?disposition=a&ID=571 Fitness jóga, harmonické cvičení těla I duše. Praha: Grada. ISBN ARKOVSKÁ, M. 2005. Posilování s vlastním tělem 417 krát jinak. Praha: 4757308. utbal:Tréning hrou. Bratislava: Peter Mačura – PEEM. 278s. ISBN utsal.Praha: Grada Publishing, a.s. 112s. ISBN 9788024725345. Power jóga nejen pro sportovce. Brno: CPress. ISBN 9788026427902. Posilování ve fitness. České Budějovice: Kopp. ISBN 8072322141.

STACKEOVÁ, D. 2014. Fitness programy z pohledu kinantropologie. Praha: Galén. ISBN 9788074921155.

VOMÁČKO, S. BOŠTÍKOVÁ, S. 2003. Lezení na umělých stěnách. Praha: Grada. 129s. ISBN 8024721743.

Course language:

Slovak language

Notes:

Course assessment

Total number of assessed students: 9100

abs	abs-A	abs-B	abs-C	abs-D	abs-E	n	neabs
88.37	0.07	0.01	0.0	0.0	0.02	4.46	7.07

Provides: Mgr. Marcel Čurgali, Mgr. Agata Dorota Horbacz, PhD., Mgr. Dávid Kaško, PhD., Mgr. Patrik Berta, Mgr. Ladislav Kručanica, PhD., Mgr. Richard Melichar, Mgr. Petra Tomková, PhD., Mgr. Alena Buková, PhD., doc. PaedDr. Ivan Uher, MPH, PhD., prof. RNDr. Stanislav Vokál, DrSc., Mgr. Zuzana Küchelová, PhD.

Date of last modification: 07.02.2024

University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚTVŠ/ TVd/11	Course name: Sports Activities IV.
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	nd the method: ce rse-load (hours): dy period: 28 esent
Number of ECTS cro	edits: 2
Recommended seme	ster/trimester of the course: 4.
Course level: I., II.	
Prerequisities:	
Conditions for cours min. 80% of active pa	e completion: articipation in classes
Learning outcomes: Sports activities in all They have a great im enables students to s improve.	their forms prepare university students for their professional and personal life. apact on physical fitness and performance. Specialization in sports activities strengthen their relationship towards the selected sport in which they also
Brief outline of the c Brief outline of the co The Institute of physic activities aerobics; ai yoga, power yoga, p tennis, chess, volleyb Additionally, the Inst offers winter courses the Tisza River) with participation.	ourse: burse: ccal education and sport at the Pavol Jozef Šafárik University offers 20 sports kido, basketball, badminton, body-balance, body form, bouldering, floorball, ilates, swimming, fitness, indoor football, SM system, step aerobics, table all, tabata, cycling. titute of physical education and sport at the Pavol Jozef Šafárik University (ski course, survival) and summer courses (aerobics by the sea, rafting on an attractive programme, sports competitions with national and international
Recommended litera BENCE, M. et al. 200 [online] Dostupné na BUZKOVÁ, K. 2006 8024715252. JARKOVSKÁ, H, JA Grada. ISBN 978802 KAČÁNI, L. 2002. F 8089197027. KRESTA, J. 2009. Fu LAWRENCE, G. 201 SNER, Wolfgang. 20	 Ature: O5. Plávanie. Banská Bystrica: FHV UMB. 198s. ISBN 80-8083-140-8. https://www.ff.umb.sk/app/cmsFile.php?disposition=a&ID=571 Fitness jóga, harmonické cvičení těla I duše. Praha: Grada. ISBN ARKOVSKÁ, M. 2005. Posilování s vlastním tělem 417 krát jinak. Praha: 4757308. 'utbal:Tréning hrou. Bratislava: Peter Mačura – PEEM. 278s. ISBN utsal.Praha: Grada Publishing, a.s. 112s. ISBN 9788024725345. Power jóga nejen pro sportovce. Brno: CPress. ISBN 9788026427902. O4. Posilování ve fitness. České Budějovice: Kopp. ISBN 8072322141.

STACKEOVÁ, D. 2014. Fitness programy z pohledu kinantropologie. Praha: Galén. ISBN 9788074921155.

VOMÁČKO, S. BOŠTÍKOVÁ, S. 2003. Lezení na umělých stěnách. Praha: Grada. 129s. ISBN 8024721743.

Course language:

Slovak language

Notes:

Course assessment

Total number of assessed students: 5671

abs	abs-A	abs-B	abs-C	abs-D	abs-E	n	neabs
82.81	0.28	0.04	0.0	0.0	0.0	7.97	8.9

Provides: Mgr. Marcel Čurgali, Mgr. Agata Dorota Horbacz, PhD., Mgr. Dávid Kaško, PhD., Mgr. Patrik Berta, Mgr. Ladislav Kručanica, PhD., Mgr. Richard Melichar, Mgr. Petra Tomková, PhD., Mgr. Alena Buková, PhD., doc. PaedDr. Ivan Uher, MPH, PhD., prof. RNDr. Stanislav Vokál, DrSc., Mgr. Zuzana Küchelová, PhD.

Date of last modification: 07.02.2024

University: P. J. Šafá	rik University in Košice					
Faculty: Faculty of S	Faculty: Faculty of Science					
Course ID: ÚFV/ SVK/13	Course ID: ÚFV/ Course name: Student Scientific Conference					
Course type, scope a Course type: Recommended cour Per week: Per stud Course method: pre	Course type, scope and the method: Course type: Recommended course-load (hours): Per week: Per study period: Course method: present					
Number of ECTS cr	edits: 4					
Recommended seme	ster/trimester of the cours	e:				
Course level: I., II.						
Prerequisities:						
Conditions for cours	e completion:					
Learning outcomes:						
Brief outline of the c	ourse:					
Recommended litera	iture:					
Course language:						
Notes:	Notes:					
Course assessment Total number of assessed students: 25						
abs n						
100.0 0.0						
Provides:						
Date of last modification: 30.11.2021						
Approved: doc. Mgr. Daniel Jancura, PhD.						

University: P. J. Šafán	rik University in Košice							
Faculty: Faculty of S	cience							
Course ID: ÚFV/ DGS/21	Course name: Students` Digital Literacy							
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: present							
Number of ECTS cro	edits: 2							
Recommended seme	ster/trimester of the course: 1.							
Course level: I.								
Prerequisities:								
Conditions for cours Summary evaluation 1. Practical ongoing a 3. Active participatic absences allowed) and assignments)	e completion: based on ongoing assessment: assignments and their defense (at least 50% needed) on during face-to-face contact learning in classical or virtual classroom (3 nd during online learning (no absence, uploading all individual ongoing							
Learning outcomes: The student should of digital technologies (i 1. according to the cu 2. for better and mor learning and further c	btain and know to apply basic knowledge and skills in working with current mobile phone, tablet, laptop, web technologies): rrent European framework for the Digital competence DigComp and ECDL e effective learning, work and active life in higher education, later lifelong areer prospects.							
Brief outline of the c 0102. Basic digital s - modern web browse - security, privacy, res 0305. Search, collec - scanning, audio reco - digital notebooks (C - evaluation of digital 0608. Editing and cr - cloud and interactiv (text and spreadsheet - work with pdf docur (Kami, Google books 09 10. Organization - modern LMS and cl (Google Classroom, I - time management (C 1113. Digital comm	ourse: skills, DigComp framework, ECDL er and its personalization sponsible use of DT ttion and evaluation of digital content ording and speech resolution, optical resolution (OCR) Google keep, Evernote, Onenote) resources (Google forms and sections) reating digital content e documents editors - Google, Microsoft, Jupyter) ments, e-books and videos b, Screencasting) n, protection and sharing of digital content oud storage Microsoft team, Google Drive, Dropbox) Google Calendar) unication and cooperation							

- collaborative interactive whiteboards (Jamboard, Whiteboard)

- online presentations and online meetings

(Google presentations, Powerpoint, Google meet, Microsoft teams)

Recommended literature:

1. Carretero Gomez, S., Vuorikari, R. and Punie, Y., DigComp 2.1: The Digital Competence Framework for Citizens with eight proficiency levels and examples of use, Luxembourg, 2017, ISBN 978-92-79-68006-9, https://www.ecdl.sk/

2. Bruff, D. (2019). Intentional Tech: Principles to Guide the Use of Educational Technology in College Teaching (1st edition). Morgantown: West Virginia University Press.

3. Baker, Y. (2020). Microsoft Teams for Education. Amazon Digital Services.

4. Miller, H. (2021). Google Classroom + Google Apps: 2021 Edition. Brentford: Orion Edition Limited.

Course language:

slovak

Not

Notes:							
Course assessment							
Total number o	f assessed studen	ts: 160					
А	В	С	D	E	FX		
69.38	3 4.38 4.38 0.0 21.88 0.0						
Provides: doc.]	Provides: doc. RNDr. Jozef Hanč, PhD.						
Date of last modification: 26.01.2022							
Approved: doc. Mgr. Daniel Jancura, PhD.							

X	
University: P. J. Safái	ik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚTVŠ/ LKSp/13	Course name: Summer Course-Rafting of TISA River
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	nd the method: se se-load (hours): dy period: 28 sent
Number of ECTS cro	edits: 2
Recommended seme	ster/trimester of the course:
Course level: I., II.	
Prerequisities:	
Conditions for cours Completion: passed Condition for success - active participation - effective performance paddling	e completion: ful course completion: in line with the study rule of procedure and course guidelines ce of all tasks: carrying a canoe, entering and exiting a canoe, righting a canoe,
Learning outcomes: Content standard: The student demonstr course syllabus and re Performance standard Upon completion of t - implement the acqui - implement basic ski - determine the right s - prepare a suitable m	ates relevant knowledge and skills in the field, which content is defined in the ecommended literature. I: the course students are able to meet the performance standard and: ired knowledge in different situations and practice, lls to manipulate a canoe on a waterway, spot for camping, aterial and equipment for camping.
Brief outline of the constraints of the constraints of the constraint of the constraints of the constraint of the constraints. Setting up a crew 4. Practical skills traints 5. Canoe lifting and constraints of the canoe lifting the canoe in the canoe in the canoe in the canoe of the pry stroke (on b) The draw stroke in the canoe of the canoe	burse: purse: iculty of waterways ting ning using an empty canoe arrying n the water without a shore contact e ut of the water fast waterways)

11. Capsizing				
12. Commands				
Recommended literature: 1. JUNGER, J. et al. Turistika a športy v prírode. Prešov: FHPV PU v Prešove. 2002. ISBN 8080680973. Internetové zdroje: 1. STEJSKAL, T. Vodná turistika. Prešov: PU v Prešove. 1999. Dostupné na: https://ulozto.sk/tamhle/UkyxQ2IYF8qh/name/Nahrane-7-5-2021-v-14-46-39#! ZGDjBGR2AQtkAzVkAzLkLJWuLwWxZ2ukBRLjnGqSomICMmOyZN==				
Course language: Slovak language				
Notes:				
Course assessment Total number of assessed students: 209				
abs	n			
37.32 62.68				
Provides: Mgr. Dávid Kaško, PhD.				
Date of last modification: 29.03.2022				
Approved: doc. Mgr. Daniel Jancura, PhD.				

University: P. J. Šafár	ik University in Košice				
Faculty: Faculty of Science					
Course ID: ÚTVŠ/ KP/12	Course name: Survival Course				
Course type, scope an Course type: Practic Recommended cour Per week: 2 Per stue Course method: pres	nd the method: e se-load (hours): hy period: 28 sent				
Number of ECTS cre	edits: 2				
Recommended semes	ster/trimester of the course:				
Course level: I., II.					
Prerequisities:					
Conditions for course Completion: passed Condition for success - active participation if - effective performance Learning outcomes: Content standard: The student demonstration course syllabus and rec Performance standard Upon completion of th - acquire knowledge a - obtain theoretical kn connected with survive - be able to resist ar	ful course completion: n line with the study rule of procedure and course guidelines, e of all the tasks defined in the course syllabus ates relevant knowledge and skills in the field, which content is defined in the ecommended literature. : he course students are able to meet the performance standard and should: bout safe stay and movement in natural environment, owledge and practical skills to solve extraordinary and demanding situations and minimization of damage to health, and face situations related to overcoming barriers and obstacles in natural				
environment, - be able implement children and youth wi	the acquired knowledge as an instructor during summer sport camps for thin recreational sport.				
 Brief outline of the constraints Brief outline of the constraints Brinciples of condunation Preparation and guidation Objective and subject Principles of hygicant Fire building Movement in the unit Shelters Food preparation and Rappelling, Tyroliant Transport of an interval 	Durse: urse: ct and safety in the movement in unfamiliar natural environment dance of a hike tour ective danger in the mountains ne and prevention of damage to health in extreme conditions nfamiliar terrain, orientation and navigation nd water filtering n traverse jured person, first aid				

Recommended literature:

1. JUNGER, J. et al. Turistika a športy v prírode. Prešov: Fakulta humanitných a prírodných vied PU v Prešove. 2002. 267s. ISBN 80-8068-097-3.

n

53.99

PAVLÍČEK, J. Člověk v drsné přírodě. 3. vyd. Praha: Práh. 2002. ISBN 8072520598.
 WISEMAN, J. SAS: příručka jak přežít. Praha: Svojtka & Co. 2004. 566s. ISBN 8072372807.

Course language:

Slovak language

Notes:

Course assessment

Total number of assessed students: 439

abs 46.01

Provides: Mgr. Ladislav Kručanica, PhD.

Date of last modification: 16.05.2023

University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚFV/ MSB/10	Course name: System Biology Modeling
Course type, scope a Course type: Lectur Recommended cour Per week: 2 / 0 Per Course method: pre	nd the method: e / Practice rse-load (hours): study period: 28 / 0 esent
Number of ECTS cro	edits: 3
Recommended seme	ster/trimester of the course: 5.
Course level: I.	
Prerequisities:	
Conditions for cours Solving intermediate Exam.	e completion: motivating challenges given at the lectures.
Learning outcomes: To provide an overvi field of systems biolo	ew of the computational techniques and achievable results in the emerging gy.
Brief outline of the c Basics of molecular and Anfinsen princip procedures and their Biological polymers a Biological databases as an example of non Molecular interactio approaches. Stochas perspectives of system	ourse: modeling. Physical structure of biopolymers. Foldamers, Levinthal paradox le. Essentials of molecular modeling and molecular simulations. Examples of results. as sequences. Sequence comparision. of sequences, acces and work. BLAS, FASTA, scoring matrices. Sugar code -linear code. Examples of use and results. n networks, modeling of reaction kinetics. Application of graph-based tic and deterministic modeling. Typical examples of use. Outlines and ns biology and systems medicine. Chalenges of synthetic biology.
Recommended litera Alon, Uri. *An Introc ed. Chapman and Hal Campbell, A. Malcol Bioinformatics*. 2nd Gabius, Hans-Joachin	ture: huction to Systems Biology: Design Principles of Biological Circuits*. 1st ll/CRC, 2006. m, and Laurie J. Heyer. *Discovering Genomics, Proteomics and ed. Benjamin Cummings, 2006. n. *The Sugar Code: Fundamentals of Glycosciences*. Wiley-VCH. 2009.
Course language:	<i>C </i>
Notes:	

Course assessment Total number of assessed students: 220							
A B C D E FX							
91.36	6.36	1.82	0.45	0.0	0.0		
Provides: doc. RNDr. Jozef Uličný, CSc.							
Date of last modification: 08.09.2021							
Approved: doc. Mgr. Daniel Jancura, PhD.							

University: P. J. Šafá	rik University in Košice	
Faculty: Faculty of S	Science	
Course ID: ÚFV/ TME1/03	Course name: Theoretical Mechanics	
Course type, scope a Course type: Lectur Recommended cou Per week: 3 / 2 Per Course method: pre	and the method: re / Practice rse-load (hours): study period: 42 / 28 esent	
Number of ECTS cr	redits: 6	
Recommended seme	ester/trimester of the course: 1., 3.	
Course level: I.		
Prerequisities: ÚFV/	/VF1a/12	

Conditions for course completion:

To successfully complete the course, the student must demonstrate sufficient understanding of all basic concepts and applications of theoretical mechanics. Knowledge of basic concepts at the level of their mathematical definition is required, as well as their physical content and principled applications. The student must be able to actively master the content of the curriculum continuously during the semester, so that he can actively and creatively use the acquired knowledge in solving specific problems in exercises and independent homework. In addition to direct participation in teaching, the student is obliged to independently study professional topics assigned by the teacher and also to develop and present two home assignments. The condition for obtaining credits is, in addition to participation in teaching, also the successful completion of the 3rd written tests from exercises and lectures and the elaboration of home assignments. The minimum limit for passing the exam is to obtain 51% of the total score, which takes into account all required activities with relevant weight.

Rating scale: A - 91% -100% points, B - 81% -90% points, C - 71% -80% points, D - 61% -70% points, E - 51% -60% points.

Learning outcomes:

The lecture on Theoretical Mechanics is the first lecture of an extensive university course in theoretical physics, where the student gets acquainted with fundamental theoretical concepts (e.g., generalized coordinates, velocities and momentum, phase space, Hamiltonian Lagrangian ...), which constitute the basis for understanding advanced theoretical methods of advanced courses such as quantum mechanics, statistical physics and quantum field theory. For this reason, attending this lecture is essential for all physics students. In addition to deep physical knowledge, students will also gain a practical experience in solving complex problems of mechanics of systems of mass points, continuum mechanics, hydrostatics and hydrodynamics.

Brief outline of the course:

1. Dynamics of a free system of mass points.

2. Motion of a constrained system of mass points. Constrains and their classification. The principle of virtual work and search for equilibrium positions.

3. D'Alembert's principle. Lagrange equations of the first kind. Generalized coordinates and generalized forces.

4. Lagrange equations of the second kind and generalized potential.

5. Basic properties of Lagrange equations. First integrals of equations of motion: Integral of energy and generalized momentum.

6. Integral principles. Variation of functions and integrals. Hamilton's principle.

7. Hamilton's function. Hamilton's canonical equations.

8. Mechanics of a perfectly rigid body. Position of a rigid body in space, independent coordinates. The speed of the points of a rigid body.

9. Center of gravity, linear and angular momentums of a rigid body. Tensor of inertia.

10. Kinetic energy of a rigid body. Euler angles and Euler kinematic equations. Euler's equations of motion of a perfectly rigid body.

11. Basic concepts of continuum mechanics. Vector and tensor of deformation. Stress vector and stress tensor. Equilibrium conditions and equations of motion of a continuum. Generalized Hooke's law. Waves in an elastic environment.

12. Mechanics of fluids. Conditions of a hydrostatic equilibrium. Continuity equation. Euler's equations of motion of an ideal fluid.

Recommended literature:

1. Meirovitch L.: Methods of Analytical dynamics, McGraw-Hill, New York, 1970.

- 2. Taylor T.T.: Mechanics: Classical and Quantum, Pergamon Press, Oxford, 1976.
- 3. Strelkov S.P.: Mechanics, Mir Publishers, Moscow, 1985.
- 4. Greiner W.: Classical Mechanics, Springer-Verlag, Berlin, 2010.
- 5. Goldstein H.: Classical Mechanics, Addison-Wesley, London, 1970.

6. Barger V., Olsson M.: Classical Mechanics: A Modern Perspective, McGraw-Hill, London, 1973.

Course language:

- 1. Slovak,
- 2. English

Notes:

Course assessment

Total number of a	sessed students: 196
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А	В	С	D	Е	FX
32.65	12.76	17.35	15.82	9.69	11.73

Provides: prof. RNDr. Michal Jaščur, CSc.

Date of last modification: 01.10.2021

University: P. J. Šafárik University in Košice				
Faculty: Faculty of S	cience			
Course ID: ÚFV/ TEP1/03Course name: Theory of the Electromagnetic Field				
Course type, scope a Course type: Lectur Recommended cou Per week: 3 / 1 Per Course method: pre	and the method: re / Practice rse-load (hours): study period: 42 / 14 esent			
Number of ECTS cr	edits: 5			
Recommended semester/trimester of the course: 4., 6.				

Course level: I.

Prerequisities: ÚFV/VFM1b/15 or ÚFV/VF1b/03

Conditions for course completion:

To successfully complete the course, the student must demonstrate sufficient understanding of the basics terms, concepts and applications of electromagnetic field theory. Knowledge of basic concepts is required at the level of their mathematical definition, as well as their physical content and specific applications. During the semester, the student must continuously master the content of the curriculum so that he can actively and creatively use the acquired knowledge in solving specific tasks during the exercises and pass continuous written tests taken into account in the overall evaluation of the subject. The condition for obtaining credits is passing 2 continuous written tests in exercises and an oral exam, which consists of theoretical questions covering the entire scope of the course. The credit evaluation of the course takes into account the following student workload: direct teaching (2 credits), self-study (1 credit), individual consultations (1 credit) and assessment (1 credit). The minimum threshold for completing the course is to obtain at least 50% of the total score, using the following rating scale: A (90-100%), B (80-89%), C (70-79%), D (60- 69%), E (50-59%), F (0-49%).

Learning outcomes:

After completing lectures and exercises, the student will have sufficient physical skills, knowledge and mathematical apparatus enabling independent solution of a wide range scientific problems in electromagnetic field theory. The student also gets an overview of applications of electromagnetic field theory in various fields of physics such as electricity, magnetism, optics, etc.

Brief outline of the course:

1. Charge density and current density. Continuity equation. Definition of electromagnetic field.

2. System of Maxwell's equations in vacuum: differential formulation of Gauss' law of electrostatics, law of total current. The absence of magnetic monopoles and the law of electromagnetic induction.

3. Scalar and vector potential, gauge transformation. Wave equations for potentials. Energy conservation law in electromagnetic field theory: Poynting vector.

4. Conservation law of momentum of electromagnetic field: Maxwell's stress tensor.

5. Electrostatic field in vacuum and its potential. Potential of charges distributed in space and on surfaces. Boundary conditions on a charged area.

6. Multipole development of charge system potential. Electrostatic field energy. Electrostatic potential energy of a charge system and its multipole development in an external electric field.

7. Dielectric polarization. Vector of electrical induction, dielectric susceptibility and permittivity. Electrostatic field induced by a system of free charges in a dielectric, boundary conditions at the interface of two dielectrics.

8. Magnetic fields of stationary currents in vacuum; Biot-Savart law.

9. Stationary magnetic field of closed elementary current system, magnetic moment. Magnetization of magnets, magnets in the magnetic field of stationary currents.

10. Magnetic field strength, magnetic susceptibility and permeability. Magnetic field of a system of conductive currents in magnetics, boundary conditions at the interface of two magnets.

11. System of Maxwell's equations in the material environment and the conservation law of electromagnetic field energy. Quasi-stationary electromagnetic field.

12. Electromagnetic waves in homogeneous non-conductive medium, plane electromagnetic wave. Monochromatic plane wave and its polarization.

13. Refraction and reflection of a plane monochromatic wave at the interface of two media.

Recommended literature:

Kvasnica J.: Teorie elektromagnetického pole. Academia Praha, 1985.

Bobák A.: Teória elektromagnetického polľa, UPJŠ Košice, 2002.

Bobák A., Vargová E.: Zbierka riešených úloh z elektromagnetického poľa, UPJŠ Košice, 2001. Greiner W.: Classical Electrodynamics, Springer-Verlag, New York, 1998.

Course language:

1. Slovak,

2. English

Notes:

Course assessment						
Total number of assessed students: 333						
A B C D E FX						
26.73	9.01	18.02	21.32	16.82	8.11	
Provides: doc. RNDr. Jozef Strečka, PhD.						
Date of last modification: 19.09.2021						

University: P. J. Šafárik University in Košice				
Faculty: Faculty of Science				
Course ID: ÚFV/ TSF/17	D: ÚFV/ Course name: Thermodynamics and Statistical physics			
Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 3 / 2 Per study period: 42 / 28 Course method: present				
Number of ECTS credits: 5				
Recommended semester/trimester of the course:				
Course level: I.				
Prerequisities:				

Conditions for course completion:

To successfully complete the course, the student must demonstrate sufficient understanding of all the basic concepts and applications of thermodynamics and classical statistical physics within the syllabus of the course. Knowledge of basic concepts of thermodynamics and classical statistical physics at the level of their mathematical definition, as well as their physical content and principled applications is required. The student must be able to actively master the content of the curriculum continuously during the semester, so that he can actively and creatively use the acquired knowledge in solving specific problems during exercises and for independent homework. In addition to direct participation in lectures, the student is obliged to study within the self-study professional topics assigned by the teacher and also to develop and present two homework assignments. The condition for obtaining credits is, in addition to participation in lectures, also the successful completion of three written tests from exercises and lectures and the elaboration of home assignments. The minimum limit for passing the exam is to obtain 51% of the total score, which takes into account all required activities with relevant weight.

Rating scale: A - 91% -100% points, B - 81% -90% points, C - 71% -80% points, D - 61% -70% points, E - 51% -60% points.

Learning outcomes:

After completing lectures and exercises, the student will acquire fundamental knowledge and skills in thermodynamics and classical statistical physics, which are prerequisites for completing advanced courses in quantum statistical physics, computer physics and condensed matter theory at the master's courses. The graduate of this course masters sufficient physical knowledge and mathematical apparatus to independently solve a wide range of current scientific problems in various fields of classical physics. These are mainly practical applications to systems consisting of a huge number of interacting particles described by the equations of classical physics. The graduate is able to apply the acquired knowledge in the field of life sciences (e.g. the spread of dangerous infectious diseases), but also in the field of big data processing and in the social and political sciences (e.g. prediction of election results).

Brief outline of the course:

1. Historical introduction and basic concepts of thermodynamics. Macroscopic system and macroscopic parameters. Internal, external, extensive and intensive macroscopic parameters. State

of system, state parameters and status functions. Basic division of thermodynamic systems - isolated, closed and open systems. Homogeneous and heterogeneous systems, thermaly homogeneous system. State of thermodynamic equilibrium. The first postulate of thermodynamics, transitivity and the principle of spontaneous inviolability of the equilibrium state.

2. The second postulate of thermodynamics and thermodynamic temperature. Natural, reversible, irreversible and quasi-static processes in thermodynamics. Internal energy, work and heat in thermodynamics. Thermal and caloric equation of state. The first law of thermodynamics. Heat capacity, specific and latent heat. Isothermal, isochoric, isobaric, adiabatic and polytropic processes in thermodynamics and their description.

3. Pfaff differential form, integrating factor, complete differential and their use in thermodynamics. Basic formulations of the second law of thermodynamics. Caratheodory's principle and mathematical formulation of the second law of thermodynamics for quasi-static processes. Introduction of absolute temperature and entropy in thermodynamics.

4. Relationship between thermodynamic and absolute temperature. Entropy and Claussius equation for reversible processes. Thermodynamic potentials for quasi-static processes. Maxwell's relations. The third law of thermodynamics. Unattainability of absolute zero temperature.

5. Dependence of thermodynamic quantities on the mass of the number of particles. Euler's theorem for homogeneous functions and its application. Thermodynamic potentials for systems with variable particle number. Non-static processes and nonequilibrium states. Slow and fast non-static processes. Mathematical formulation of the second law of thermodynamics for non-static processes. Clausius inequality.

6. Thermodynamic potentials of nonequilibrium systems and equilibrium conditions. Maximum work done by the body in the external environment. Heterogeneous systems. Gibbs phase rule.

7. Phase space, configuration space and impulse space. Statistical ensemble and distribution function. Stationary ensemble. Canonical invariance of phase volume. Calculation of mean values of physical quantities in classical statistical physics.

8. Microcanonical, canonical and grand canonical ensembles in classical statistical physics. Canonical and grand canonical partition function, internal energy, entropy, free energy and grand canonical potential.

9. Equipartition and virial theorems. Calculation of ideal gas entropy in a microcanonical ensemble, Gibbs paradox.

10. The ideal gas in the canonical ensemble and the classical theory of paramagnetism. Classical theory of heat capacity - Dulong's-Petit's law.

Recommended literature:

1) J. Kvasnica, Termodynamika, SNTL, Praha (1965).

2) J. Kvasnica, Statistická fyzika, ACADEMIA, Praha (1983).

3) M. Varady, Statisticka fyzika, UJEP Ústi nad Labem, 2007.

4) M. Jaščur, M. Hnatič, Úvod do termodynamiky, Univerzita P.J. Šafárika, Košice (2013).

Course language:

Notes:

Course assessment

Total number of assessed students: 27

А	В	С	D	Е	FX	
44.44	25.93	25.93	3.7	0.0	0.0	
Provides: prof. RNDr. Michal Jaščur, CSc.						

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