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University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: CJP/ Course name: Academic English

PFAJAKA/07

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of ECTS credits: 2

Recommended semester/trimester of the course:

Course level: I.

Prerequisities:

Conditions for course completion:

Active classroom participation, assignments handed in on time, 2 absences tolerated

1 test (13th week), no retake.

Presentation on chosen topic

Final evaluation- average assessment of test (50%), and presentation (50%).

Grading scale: A 93-100%, B 86-92%, C 79-85%, D 72-78%, E 65-71%, FX 64% and less

Learning outcomes:

The development of students' language skills - reading, writing, listening, speaking, improvement of their linguistic competence - students acquire knowledge of selected phonological, lexical and syntactic aspects, development of pragmatic competence - students can efectively use the language for a given purpose, with focus on Academic English, level B2.

Brief outline of the course:

Formal and informal English

Academic English and its specific features

Key academic verbs and nouns

Linking words in academic writing, writing a paragraph, word-order, topic sentences

Word-formation - affixation

abstract

Selected aspects of English pronunciation, academic vocabulary

Selected functional grammar structures - defining, classifying, epressing opinion, cause-effect, paraphrasing

Recommended literature:

Seal B.: Academic Encounters, CUP, 2002

T. Armer: Cambridge English for Scientists, CUP 2011

M. McCarthy M., O'Dell F. - Academic Vocabulary in Use, CUP 2008

Zemach, D.E, Rumisek, L.A: Academic Writing, Macmillan 2005

Olsen, A.: Active Vocabulary, Pearson, 2013

www.bbclearningenglish.com

Cambridge Academic Content Dictionary, CUP, 2009

Course language:

English language, level B2 according to CEFR.

Notes:

Course assessment

Total number of assessed students: 416

A	В	С	D	Е	FX
36.54	21.63	15.14	9.38	6.01	11.3

Provides: Mgr. Viktória Mária Slovenská

Date of last modification: 20.09.2023

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚMV/ | Course name: Algebra I

ALG2a/22

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours):

Per week: 3 / 3 Per study period: 42 / 42

Course method: present

Number of ECTS credits: 6

Recommended semester/trimester of the course: 1.

Course level: I.

Prerequisities:

Conditions for course completion:

According to the results from the semester and in view of the results of the written and oral final exam..

Learning outcomes:

To acquire the methods of mathematical thinking and cognition. Gain basic knowledge of number theory related to divisibility, master the basic concepts of linear algebra and be able to apply them to specific problems and mathematical problems.

Brief outline of the course:

Divisibility in Z. Fields. Systems of linear equations, Gauss elimination. Maps, permutations. Computing with matrices. Determinants, Cramer rule.

Recommended literature:

T.S Blyth, E.F. Robertson: Basic linear algebra, Springer Verlag, 2001.

K. Jänich: Linear algebra, Springer Verlag, 1991.

Course language:

Slovak

Notes:

Course assessment

Total number of assessed students: 864

A	В	С	D	Е	FX
11.11	13.43	20.02	18.98	27.55	8.91

Provides: prof. RNDr. Danica Studenovská, CSc., RNDr. Lucia Janičková, PhD., Mgr. Martin Vodička

Date of last modification: 17.02.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚMV/ | Course name: Algebra II

ALG2b/22

Course type, scope and the method: Course type: Lecture / Practice

Recommended course-load (hours): Per week: 4 / 2 Per study period: 56 / 28

Course method: present

Number of ECTS credits: 6

Recommended semester/trimester of the course: 2.

Course level: I.

Prerequisities: ÚMV/ALG2a/22

Conditions for course completion:

According to tests and to the exam.

Learning outcomes:

To acquire the methods of mathematical thinking and cognition. To deepen and expand students' knowledge of systems of linear equations, to acquire basic knowledge about vector spaces, linear representations, polynomials and polynomial equations.

Brief outline of the course:

Linear spaces, bases. Rank of a matrix. Systems of homogeneous linear equations.

Linear transformations.

Ring, fields. Polynomials over a field. Factorization into irreducible factors, roots. Roots of complex numbers. Cubic equations.

Polynomials with several unknowns, symmetric polynomials.

Recommended literature:

T. Katriňák a kol.: Algebra a teoretická aritmetika 1, Alfa Bratislava, 1985.

A. Kurosh: Higher Algebra, Mir Publishers, 1975.

Course language:

Slovak

Notes:

Course assessment

Total number of assessed students: 245

A	В	С	D	Е	FX
23.27	15.92	15.51	15.92	25.71	3.67

Provides: prof. RNDr. Danica Studenovská, CSc., RNDr. Lucia Janičková, PhD.

Date of last modification: 16.04.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚMV/ | Course name: Algebra III

ALG2c/22

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours):

Per week: 2 / 2 Per study period: 28 / 28

Course method: present

Number of ECTS credits: 4

Recommended semester/trimester of the course: 6.

Course level: I.

Prerequisities:

Conditions for course completion:

According to tests and to the exam.

Learning outcomes:

To develop students' abstract thinking. Follow up on the acquired knowledge of algebra, expand it and generalize; be able to apply the acquired knowledge to specific examples. Demonstrate knowledge of mathematical content in context.

Brief outline of the course:

Relations, operations, algebraic structures.

Substructures.

Homomorphisms, isomorphisms.

Congruences, homomorphism theorems.

Terms, term operations, identities.

Recommended literature:

B. Jónsson: Topics in Universal Algebra, Springer-Verlag 1972

M. Kolibiar a kol.: Algebra a príbuzné disciplíny, Bratislava 1992

S.N. Burris and H.P. Sankappanavar: A Course in Universal Algebra 2000, http://www.math.uwaterloo.ca/~snburris/htdocs/ualg.html

Course language:

Slovak

Notes:

Course assessment

Total number of assessed students: 148

A	В	С	D	Е	FX
17.57	18.92	24.32	21.62	15.54	2.03

Provides: prof. RNDr. Danica Studenovská, CSc.

Date of last modification: 16.04.2022

University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚMV/ ATC/22	Course name: Algebra and number theory
Course type, scope a Course type: Lectur Recommended cour Per week: 2/1 Per Course method: pre	re / Practice rse-load (hours): study period: 28 / 14
Number of ECTS cr	edits: 3
Recommended seme	ster/trimester of the course: 4.
Course level: I.	
Prerequisities: ÚMV	/ALG2b/22
	se completion: alts of written checks carried out during the semester. Final evaluation is based ten checks carried out during the semester, of test, written and oral exam.
Learning outcomes: Obtain basic knowled	dge about groups and from the elementary number theory.
_	e ring of integers ex numbers scendent numbers, minimal polynomial of the field of rationals raic numbers oup s, Lagrange theorem , factorization
M. Harminc: Elemen T. Katriňák a kol.: Al A. Legéň: Grupy, oki I.R. Shafarevich: Bas	nture: ne: A Survey of Modern Algebra, New York 1965 tárna teória čísel (1.časť), PF UPJŠ Košice 2012 gebra a teoretická aritmetika 1, Alfa Bratislava 1985 ruhy a zväzy, Alfa Bratislava 1980 sic Notions of Algebra, Springer, 2005
Course language: Slovak	

Notes:

Course assessm	Course assessment					
Total number of assessed students: 353						
Α	В	С	D	Е	FX	
12.46	19.26	23.8	22.1	20.4	1.98	

Provides: doc. RNDr. Miroslav Ploščica, CSc.

Date of last modification: 23.08.2022

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: KPE/ **Course name:** Alternative Education ALP/06 Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: present **Number of ECTS credits: 2** Recommended semester/trimester of the course: 4. Course level: I. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 327 C Α В D Е FX 69.42 25.08 2.75 0.61 0.31 1.83

Provides: Mgr. Beáta Sakalová, doc. PaedDr. Renáta Orosová, PhD.

Date of last modification: 12.03.2024

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ | Course name: Analytical Chemistry

ANCHI/21

ANCHU/21

Course type, scope and the method:

Course type: Lecture / Practice Recommended course-load (hours):

Per week: 3 / 1 Per study period: 42 / 14

Course method: present

Number of ECTS credits: 5

Recommended semester/trimester of the course: 3.

Course level: I.

Prerequisities:

Conditions for course completion:

- 1. 3x test of analytical calculations (each 33%, minim. 50%).
- 2. Examination is composed of 3 questions (each for 33%, it is necessary to reach at least 50%).

Learning outcomes:

Survey of basic principles and tasks of analytical chemistry and applications of analytical methods in research and practice.

Brief outline of the course:

Subject and role of analytical chemistry. General principles and procedures - sampling, sample pretreatment. Preparation of solutions. Evaluation of the results.

Classification of analytical reactions. Qualitative analysis of cations and anions. Basic principles of organic analysis.

Methods of quantitative analysis. General principles of gravimetry. Volumetric analysis.

Instrumental methods of analytical chemistry (basic principles, instrumentaion and applications) - electroanalytical, optical and separation methods.

Recommended literature:

D.Harvey, Modern Analytical Chemistry. McGraw Hill, Boston, 2000

Skoog D.A., Principles of Instrumental Analysis. Saunders Col. Publishing, New York 1985

Course language:

Notes:

Course assessment

Total number of assessed students: 73

A	В	С	D	Е	FX
31.51	16.44	20.55	19.18	8.22	4.11

Provides: doc. RNDr. Taťána Gondová, CSc.

Date of last modification: 12.11.2021

Approved: doc. RNDr. Stanislav Lukáč, PhD., prof. RNDr. Vladimír Zeleňák, DrSc.

Page: 13

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ Course n

Course name: Bachelor Thesis and its Defence

BPO/21

Course type, scope and the method:

Course type:

Recommended course-load (hours):

Per week: Per study period: Course method: present

Number of ECTS credits: 4

Recommended semester/trimester of the course:

Course level: I.

Prerequisities:

Conditions for course completion:

Learning outcomes:

Brief outline of the course:

Oral presentation of the thesis results. Answering questions of the thesis oponent or members of the state examination board.

Recommended literature:

Course language:

slovak

Notes:

Course assessment

Total number of assessed students: 17

A	В	C	D	Е	FX
94.12	5.88	0.0	0.0	0.0	0.0

Provides:

Date of last modification: 07.12.2021

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚMV/ Course name: Bachelor project I BKPa/22 Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 1 Per study period: 14 Course method: present **Number of ECTS credits: 1 Recommended semester/trimester of the course:** 5. Course level: I. **Prerequisities: Conditions for course completion:** To prepare and present a contribution related to thesis and its topic. **Learning outcomes:** To get students familiar with basic knowledge on the form and content of thesis and thesis presentation as well as with the support for its realisation. **Brief outline of the course:** Necessary elements and formal aspects of a thesis. WYSIWYG editors, LaTeX, drawing programs. Presentation software, Microsoft PowerPoint and its clones, Beamer. Suggestions for presentation and contribution making. **Recommended literature:** electronic information sources Course language: Slovak and English **Notes:** Course assessment Total number of assessed students: 118 abs n 100.0 0.0 Provides: doc. RNDr. Dušan Šveda, CSc.

Approved: doc. RNDr. Stanislav Lukáč, PhD., prof. RNDr. Vladimír Zeleňák, DrSc.

Date of last modification: 24.08.2022

University: P. J. Šafá	rik University in Košio	ce			
Faculty: Faculty of S	cience				
Course ID: ÚMV/ BKPb/22	Course name: Bache	elor project II			
Course type, scope a Course type: Recommended cou Per week: Per stud Course method: pre	rse-load (hours): ly period:				
Number of ECTS cr	edits: 2				
Recommended seme	ster/trimester of the o	course: 6.			
Course level: I.					
Prerequisities:					
Conditions for cours	e completion:				
Learning outcomes:					
Brief outline of the c	ourse:				
Recommended litera	iture:				
Course language:					
Notes:					
Course assessment Total number of asse	ssed students: 101				
abs n					
	100.0 0.0				
Provides:		<u> </u>			
Date of last modifica	tion: 24.08.2022				
Approved: doc. RNI	Dr. Stanislav Lukáč. Ph	D., prof. RNDr. Vladimír Zeleňák, DrSc.			

University: P. J. Šafárik University in Košice					
Faculty: Faculty of S	cience				
Course ID: ÚMV/ BPO/14	Course name: Bachelor thesis and its defence				
Course type, scope a Course type: Recommended cour Per week: Per stud Course method: pre	rse-load (hours): y period:				
Number of ECTS cro	edits: 4				
Recommended seme	ster/trimester of the course:				
Course level: I.					
Prerequisities:					
fraud and must meet 21/2021, which lays Košice and its compo	e completion: It the result of the student's own work. It must not show elements of academic the criteria of good research practice defined in the Rector's Decision no. down the rules for assessing plagiarism at Pavol Jozef Šafárik University in ments. Fulfillment of the criteria is verified mainly in the supervision process thesis defense. Failure to do so is reason for disciplinary action.				
demonstrates mastery acquisition of knowled graduate of the study field problems. The batthe ability of indepen- on the bachelor thesi	's competences with respect to the profile of the graduate. The bachelor's thesis of the basics of theory and professional terminology of the field of study, edge, skills and competencies in accordance with the declared profile of the program, as well as the ability to apply them creatively in solving selected bachelor thesis may have elements of compilation. The student demonstrates dent professional work in terms of content, formal and ethical. Further details are determined by Directive no. 1/2011 on the basic requirements of final Regulations of UPJŠ in Košice.				
Brief outline of the course: 1. Elaboration of the bachelor thesis in accordance with the instructions of the supervisor. 2. Presentation of the results of the bachelor's thesis before the examination commission. 3. Answering questions related to the topic of the bachelor thesis within the discussion. Recommended literature:					
	erature is determined individually in accordance with the topic of the				
Course language:					

Notes:

Course assessment Total number of assessed students: 187					
A B C D E FX					
67.91	17.65	7.49	3.74	2.14	1.07
Provides:					
Date of last modification: 19.04.2022					

University: P. J. Šafárik University in Košice				
Faculty: Faculty of S	cience			
Course ID: ÚCHV/ BKPa/22				
Course type, scope a Course type: Recommended cou Per week: Per stud Course method: pre	rse-load (hours): ly period: esent			
Number of ECTS cr	edits: 3			
Recommended seme	ster/trimester of the cou	rse: 5.		
Course level: I.	,			
Prerequisities:				
Conditions for cours	se completion:			
Learning outcomes:				
Brief outline of the c	ourse:			
Recommended litera	nture:			
Course language:				
Notes:				
Course assessment Total number of asse	ssed students: 23			
abs				
100.0 0.0				
Provides:				
Date of last modifica	ation: 07.02.2022			
Approved: doc. RNI	Dr. Stanislav Lukáč, PhD	prof. RNDr. Vladimír Zeleňák, DrSc.		

University: P. J. Šafárik University in Košice				
Faculty: Faculty of S	cience			
Course ID: ÚCHV/ BKPb/22	Course ID: ÚCHV/ BKPb/22 Course name: Bakalársky projekt II			
Course type, scope a Course type: Recommended cour Per week: Per stud Course method: pre	rse-load (hours): ly period:			
Number of ECTS cr	edits: 3			
Recommended seme	ster/trimester of the o	course: 6.		
Course level: I.				
Prerequisities:				
Conditions for cours	e completion:			
Learning outcomes:				
Brief outline of the c	ourse:			
Recommended litera	iture:			
Course language:				
Notes:				
Course assessment Total number of asse	ssed students: 19			
abs n				
100.0 0.0				
Provides:		<u> </u>		
Date of last modifica	ition: 07.02.2022			
Approved: doc. RND	Dr. Stanislav Lukáč. Ph	D., prof. RNDr.	Vladimír Zeleňák, DrSc	 ;_

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ | Course name: Basis of Mineralogy

MIN1/14

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14

Course method: present

Number of ECTS credits: 5

Recommended semester/trimester of the course: 4.

Course level: I.

Prerequisities: ÚCHV/VCH/10 or ÚCHV/VCH/21 or ÚCHV/VCHU/10 or ÚCHV/ZAC2/10 or ÚCHV/VACH/10 or ÚCHV/CHG/09 or ÚCHV/ZCF/03 or ÚCHV/VCHU/15

Conditions for course completion:

Verification of theoretical knowledge and recognizing minerals.

A semester project about selected minerals (40 %), a practical test from recognizing of minerals (30 %), a written examination (30 %). The student must obtain totally at least 51%.

In a case of online education the practical test is canceled and the written examination contains more questions (60 %).

Learning outcomes:

To recognize the beauty of nature and to obtain basic knowledge from mineralogy. After completing the course, students will be familiar with the properties of commonly available minerals and will be able to recognize these minerals.

Brief outline of the course:

Basic terms and definitions, origin of minerals in nature. Basis of morphological and structural crystallography: characteristic properties of crystals, crystallographic laws, crystal structure, unit cells and their parameters, crystallographic systems with examples of minerals. Crystallochemistry: types of bonds and structures and their effect on the properties of minerals. Physical properties of minerals and their utilize in minerals classification. Basis of genetic and systematic mineralogy. Structure of silicates.

Recommended literature:

M. Košuth: Mineralógia. Elfa, s.r.o. Košice, 2001 V. Radzo: Mineralógia, Alfa Bratislava, 1987.

Course language:

Slovak

Notes:

Teaching is carried out in person or, if necessary, online using the MS Teams tool. The form of teaching is specified by the teacher at the beginning of the semester, updated continuously.

Course assessment					
Total number of assessed students: 140					
A	В	С	D	Е	FX
85.0	12.86	0.71	0.71	0.0	0.71

Provides: doc. RNDr. Ivan Potočňák, PhD.

Date of last modification: 21.07.2022

	COURSE INFORMATION LETTER			
University: P. J. Šafá	rik University in Košice			
Faculty: Faculty of S	cience			
Course ID: ÚCHV/ BCHU/21				
Course method: pre	re / Practice rse-load (hours): study period: 42 / 14 esent			
Number of ECTS cr				
Course level: I.	ster/trimester of the course: 5.			
-	V/VCHU/10 or ÚCHV/VCHU/15 or ÚCHV/VACH/10 or ÚCHV/VCHU/14			
student passes the ex	ne completion: In of the exam, which consists of two parts: (i) written and (ii) oral part. The am if he / she obtains at least 60% of the points in the written part and at the answers the asked questions in the oral part.			
fats and sugars) and	(i) the basic building blocks of biomacromolecules (proteins, DNA, RNA, their properties, (ii) the basic biochemical processes that take place in living ay energy is produced and used in cells.			
2. DNA and RNA an 3. Enzymes: Basic Co 4. Carbohydrates (Mo 5. Lipids and Cells Mo 6. Metabolis: Basic Co 7. Glycolysis and Glo 8. The Citric Acid Cy 9. Oxidative Phospho 10. The Calvine Cycl 11. Fatty Acids Meta 12. DNA Replication	nd Function, Exploring proteins. d the Flow of Genetic Information, Exploring genes. concepts and Kinetics, Catalytic Strategies and Regulatory Strategies. conosaccharides, Disaccharides, Polysaccharides – Functions and Properties). Itembranes, Membrane Channels and Pumps. Concepts and Design, Signal-Transduction Pathways. Inconeogenesis, Glycogen Metabolism. Inconeogenesis, Glycogen Metabolism. Incolored and Glyoxylate Cycle. Incolored and Glyoxylate Cycle. Incolored and Fernice Phosphate Pathway.			
Recommended litera	ture:			
Course language:				

Notes:

Course assessment					
Total number of assessed students: 84					
A	В	С	D	Е	FX
28.57	13.1	14.29	21.43	20.24	2.38

Provides: prof. RNDr. Erik Sedlák, DrSc., RNDr. Nataša Tomášková, PhD., prof. RNDr. Mária Kožurková, CSc.

Date of last modification: 14.11.2021

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ Cour

Course name: Biochemistry II

BCH1b/10

Course type, scope and the method: Course type: Lecture / Practice

Recommended course-load (hours): Per week: 3 / 1 Per study period: 42 / 14

Course method: present

Number of ECTS credits: 5

Recommended semester/trimester of the course: 6.

Course level: I.

Prerequisities: ÚCHV/BCH1a/03 or ÚCHV/BCH1a/21

Conditions for course completion:

Test and oral examination.

Learning outcomes:

The aim of biochemistry teaching is to acquire knowledge in the field of living organisms on the basis of their molecular structure information on cell metabolism.

Brief outline of the course:

Basic principle of metabolism, basic metabolic pathways and cycles, integration of cell metabolism.

Recommended literature:

Koolman J., Roehm K.H.: Color atlas of biochemistry. Thieme, Stuttgart, Germany, 2005. Kodíček M., Valentová O., Hynek R.: Biochemie, chemický pohled na biologický svět, Vysoká škola chemicko-technologická v Praze, Praha, 2022.

Course language:

Notes:

Course assessment

Total number of assessed students: 377

Α	В	С	D	Е	FX
9.02	19.63	31.03	18.3	21.22	0.8

Provides: prof. RNDr. Mária Kožurková, CSc., prof. RNDr. Erik Sedlák, DrSc., doc. RNDr. Rastislav Varhač, PhD., doc. RNDr. Viktor Víglaský, PhD., RNDr. Nataša Tomášková, PhD., RNDr. Danica Sabolová, PhD., univerzitná docentka

Date of last modification: 26.07.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ Course name

PBCHU/15

Course name: Biochemistry Practical

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 4 Per study period: 56

Course method: present

Number of ECTS credits: 4

Recommended semester/trimester of the course: 6.

Course level: I.

Prerequisities: ÚCHV/BCHU/03 or ÚCHV/BCHU/21

Conditions for course completion:

Active participation with a maximum of one excused absence without the need for compensation. In case of excused absence from two or more practical exercises (e.g. due to illness), the student agrees with the teacher on alternative dates for practice.

Correctly prepared protocols from all completed tasks.

At least 51% of points from each of the written tests.

Learning outcomes:

To allow students to get practical experience in experimental techniques and methods, currently used in a biochemical research: UV/VIS spectrophotometry, thin layer chromatography (TLC), gel electrophoresis, isolation of macromolecules and substances from biological materials and their quantitative and qualitative determination.

Brief outline of the course:

- 1. Biochemistry laboratory safety rules. Basic biochemical laboratory procedures.
- 2. Qualitative tests for amino acids and proteins.
- 3. Isolation of casein from milk. Determination of protein concentration by Lowry method.
- 4. Determination of the iodine number by Yasud method . Soap production. Reactions with soap. Oxidation of unsaturated fatty acids.
- 5. Saponification number of fats and oils. Qualitative test for cholesterol: Salkowsky reaction.
- 6. Qualitative tests for carbohydrates. Determination of reducing carbohydrates by the Schoorl's method
- 7. Determination of reducing and nonreducing carbohydrates in germinant plants.
- 8. Time-dependent course of enzyme-catalyzed reaction: digestion of gelatin by trypsine.
- 9. Determination of catalase activity and the first order rate constant. Effect of pH on alpha-amylase activity.
- 10. Effect of substrate concentration on initial rate of reaction, determination of Km and Vmax for urease-catalyzed hydrolysis of urea.
- 11. Isolation of DNA from spleen. Isolation of RNA from yeast. Qualitative tests for DNA and RNA components.
- 12. Determination of vitamin C concentration by 2,4-dinitrofenylhydrazine. Determination of vitamins A, B1, and C.

13. Final evaluation of students.

Recommended literature:

Sedlák, Varhač, Danko, Paulíková, Podhradský: Praktické cvičenia z biochémie, 2020, https://unibook.upjs.sk/sk/chemia/1411-prakticke-cvicenia-z-biochemie

Course language:

Slovak

Notes:

Teaching is carried out in person.

Course assessment

Total number of assessed students: 237

A	В	С	D	Е	FX
77.64	18.14	2.95	0.84	0.42	0.0

Provides: prof. RNDr. Mária Kožurková, CSc., RNDr. Nataša Tomášková, PhD., doc. RNDr. Rastislav Varhač, PhD., RNDr. Danica Sabolová, PhD., univerzitná docentka, RNDr. Eva Konkoľová, PhD.

Date of last modification: 19.11.2021

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ | Course name: Bioinorganic Chemistry I

BAC1/04

Course type, scope and the method: Course type: Lecture / Practice

Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14

Course method: present

Number of ECTS credits: 5

Recommended semester/trimester of the course: 5.

Course level: I., II.

Prerequisities:

Conditions for course completion:

Test or seminar works

examination

Learning outcomes:

The basic knowledges about biometal interactions with biomolecules, biomaterials, biominerals, biocatalysis, metals in biology and medicine, metal-based drugs, toxic metals for biosystems and metals in the environment.

Brief outline of the course:

Metalic and non-metalic elements and their roles in biological systems (biometals, bulk biological elements, essential trace elements). Biocoordination compounds, bioligands. Biocatalyzers. Oxygen carriers and oxygen transport proteins. Photochemical process. Catalysis and regulation processes. Calcium biominerals and biomineralization. Toxic metals. Application of knowledge of bioinorganic chemistry in pharmacy, chemotherapy (e.g. platinum complexes in cancer therapy) radiodiagnostics, mineral biotechnology, ecology and in other branches of life.

Recommended literature:

- 1. Shriver D. F., Atkins P. W., Overton T. L., Rourke J.P., Weller M.T., Amstrong F.A.: Shiver & Atkins. Inorganic Chemistry. Oxford University Press, Oxford 2006.
- 2. Kaim W., Schwederski B.: Bioinorganic Chemistry: Inorganic Elements in the Chemistry of Life. Wiley, Chichester 1998.
- 3. Wilkins P. C., Wilkins R. G.: Inorganic Chemistry in Biology. OCP, Oxford 1997.

Course language:

Notes:

Course assessment

Total number of assessed students: 373

A	В	С	D	Е	FX
42.36	27.61	19.03	5.9	4.83	0.27

Provides: prof. RNDr. Zuzana Vargová, Ph.D.

Date of last modification: 28.10.2021

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚBEV/ Course name: H

BDD/05

Course name: Biology of Children and Adolescents

Course type, scope and the method: Course type: Lecture / Practice

Recommended course-load (hours): Per week: 2 / 0 Per study period: 28 / 0

Course method: present

Number of ECTS credits: 2

Recommended semester/trimester of the course: 4., 6.

Course level: I.

Prerequisities:

Conditions for course completion:

Written test

Learning outcomes:

Acquisition of basic morphological and physiological knowledge about individual organs and systems of the human body with a focus on the specifics of childhood and adolescence. Familiarity with developmental and growth characteristics and with the most common diseases in these stages of ontogenesis.

Brief outline of the course:

Human ontogenesis. Postnatal development. Age specific features of skeletal and muscalar, circulatory, respiratory, gastrointestinal and urinary systems. Reproductive system. Endocrine system. Nervous system. Age specifics of selected diseases and drug dependence arise. Human population and environment.

Recommended literature:

Drobný I., Drobná M.: Biológia dieťaťa pre špeciálnych pedagógov I. a II. Bratislava, PdF UK, 2000

Lipková V.: Somatický a fyziologický vývoj dieťaťa. Osveta Bratislava, 1980

Malá H., Klementa J.: Biológia detí a dorastu. Bratislava, SPN, 1989

Course language:

Notes:

Course assessment

Total number of assessed students: 1757

A	В	С	D	Е	FX
31.59	24.08	18.16	16.62	9.05	0.51

Provides: doc. RNDr. Monika Kassayová, CSc.

Date of last modification: 20.04.2022

COURSE INFORMATION LETTER					
University: P. J. Šafár	rik University in Košice				
Faculty: Faculty of S	cience				
Course ID: ÚMV/ ZBR/14					
Course type: Practic Recommended cour Per week: 2 Per stu	Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: present				
Number of ECTS cro	edits: 2				
Recommended seme	ster/trimester of the cours	e: 5.			
Course level: I.					
Prerequisities:					
Conditions for cours Active participation of	-				
	ainted with fundamentals of lates his/her habits of positive	of the contract bridge, develops his/her logical ve social behaviour.			
Brief outline of the course: Bridge rules. Principles of the bidding system Standard American. Basic techniques of declarer's play. Basic techniques of the defence. Lead conventions, signals. Common bidding conventions. Selected advanced techniques of the card play. Partnership cooperation in the contract bridge. Bridge ethics.					
Recommended literature: T. Menyhért: Kurz bridžu 2013, http://new.bridgekosice.sk/kurz-bridzu-2013/ R. Pavlicek: Learn To Play Bridge!, http://www.rpbridge.net/1a00.htm ACBL SAYC System Booklet, http://ebookbrowsee.net/acbl-sayc-pdf-d201415187					
Course language: Slovak or English					
Notes: Minimum number of participants is 4.					
Course assessment Total number of asses	ssed students: 35				
	abs	n			

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2.86

97.14

Provides: doc. RNDr. Miroslav Ploščica, CSc., Mgr. Martin Vodička

Date of last modification: 08.02.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ | **Course name:** Chemical calculations

CHV1/99

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of ECTS credits: 2

Recommended semester/trimester of the course: 1.

Course level: I.

Prerequisities:

Conditions for course completion:

Successful completion of two written tests in the middle and at the end of the semester. Accomplished test is with minimal 50% of point. The exact dates will be determined after mutual consultation between the teacher and the students.

The rating scale is determined as follows: A (100-91%), B (90-81%), C (80-71%), D (70-61%), E (60-51%), Fx (50-0%).

Learning outcomes:

To teach students how to calculate material balances in the systems with or without chemical processes and how to calculate examples concerning the chemical equilibrium.

Brief outline of the course:

Expression of the clear matter amount and the system composition. Stoichiometric formula. Material bilances for preparation, dissolving and mixing of solutions, and for separating of mixtures. Material bilances for combined processes. Chemical equations and material bilances in the systems with chemical processes. Acid-Base equilibrium and the pH calculations. The solubility product and solubility.

Recommended literature:

Potočňák I.: Chemické výpočty vo všeobecnej a anorganickej chémii (skriptum), PF UPJŠ, Košice, 2017.

https://unibook.upjs.sk/sk/chemia/843-chemicke-vypocty-vo-vseobecnej-a-anorganickej-chemii Any chemical laboratory tables.

Course language:

SK - slovak

Notes:

The subject is carried out in person or, if necessary, remotely using the online platform Big Blue Button (BBB). The form of teaching is specified by the teacher at the beginning of the semester and updated continuously.

Course assessment							
Total number of assessed students: 1722							
A	В	С	D	Е	FX		
26.02	18.99	22.18	19.57	11.96	1.28		

Provides: RNDr. Martin Vavra, PhD., doc. RNDr. Miroslav Almáši, PhD., Mgr. Nikolas Király, PhD.

Date of last modification: 15.11.2021

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/

Course name: Chemistry

SCHM/21

Course type, scope and the method:

Course type:

Recommended course-load (hours):

Per week: Per study period: Course method: present

Number of ECTS credits: 2

Recommended semester/trimester of the course:

Course level: I.

Prerequisities: (ÚCHV/OCHU/21 or ÚCHV/OCHU/03) and ÚCHV/ANCHU/21 and ÚCHV/BCHU/21 and (ÚCHV/ACHU/21 or ÚCHV/ACHU/03) and (ÚCHV/FCHU/22 or ÚCHV/FCHU/21 or ÚCHV/FCHU/10)

Conditions for course completion:

Learning outcomes:

Brief outline of the course:

Recommended literature:

Course language:

Notes:

Course assessment

Total number of assessed students: 59

A	В	С	D	Е	FX
10.17	27.12	22.03	18.64	16.95	5.08

Provides:

Date of last modification: 08.09.2021

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: Course name: Communication ECo-C4

KPPaPZ/ECo-C4/14

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: combined, present

Number of ECTS credits: 4

Recommended semester/trimester of the course: 4., 6.

Course level: I., N

Prerequisities:

Conditions for course completion:

1. Active participation in lessons (absence is allowed max. 90 min.), 2. Realization of assignments according to the teacher's instructions.

Detailed information in the electronic board of the course in AIS2. The teaching of the subject will be realized by a combined method.

Learning outcomes:

The student understands theoretical information about the basics of verbal and nonverbal communication, rhetoric and methods of visualization and interprets them adequately. Student is able to use the acquired communication skills in practice, can apply effective principles of communication with others, is able to anticipate and thus prevent possible misunderstandings, which will contribute to the development of his social and professional skills.

Brief outline of the course:

Basics of communication (Transmitter-receiver principle, "What is said is not equal to what is heard", "Internal dialogue", The concept of communication)

Active listening (The most important criteria for active listening)

Misunderstandings (How Misunderstandings Arise, How to Avoid Misunderstandings)

Body language (What is body language, Active / passive body language, Dress psychology)

Signs of Physical Expression, Disadvantages of Fake Physical Expression, Difference Between Active and Passive Body Expression

Personality development (Voices in us, "child in me" - identification of one's own personality)

Rhetoric (History of rhetoric, What is rhetoric, Vigor, alertness - assumptions, techniques, prompt reactions)

Visualization - optical display (Classic media - whiteboard, magnetic whiteboard, bulletin board, flipchart, Based on computer technology - PC + Beamer)

Recommended literature:

VÝROST, Jozef - SLAMĚNÍK, Ivan. Sociální psychologie. 2., přepr. a rozš. vyd. Praha : GRADA, 2008. 408 s.

VÝROST, Jozef - SLAMĚNÍK, Ivan. Aplikovaná sociální psychologie I : Člověk a sociální instituce. 1. vyd. Praha : Portál, 1998. 384 s. ISBN 80-7178-269-6.

KOMÁRKOVÁ, Růžena - SLAMĚNÍK, Ivan - VÝROST, Jozef. Aplikovaná sociální psychologie III : Sociálněpsychologický výcvik. 1. vyd. Praha : Grada Publishing, 2001. 224 s. VÝROST, Jozef - SLAMĚNÍK, Ivan. Aplikovaná sociální psychologie II. 1. vyd. Praha : Grada Publishing, 2001. 260 s.

Course language:

slovak

Notes:

After passing the certification exams from all 4 modules (Teamwork, Selfmarketing, Conflict Management, Communication) the student will receive an ECo-C card and an ECo-C certificate.

Course assessment

Total number of assessed students: 137

abs	n
86.13	13.87

Provides: Mgr. Lucia Barbierik, PhD.

Date of last modification: 24.06.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

PFAJKKA/07

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of ECTS credits: 2

Recommended semester/trimester of the course:

Course level: I.

Prerequisities:

Conditions for course completion:

Active participation in class and completed homework assignments. Students are allowed to miss two classes at the most.

2 credit tests (presumably in weeks 6/7 and 12/13) and an oral presentation in English.

Final evaluation consists of the scores obtained for the 2 tests (50%) and the presentation (50%). Final grade will be calculated as follows: A 93-100 %, B 86-92%, C 79-85%, D 72-78%, E 65-71%, FX 64 % and less.

Learning outcomes:

Brief outline of the course:

Recommended literature:

www.bbclearningenglish.com

Štěpánek, Libor a kol. Academic English-Akademická angličtina. Praha: Grada Publishing, a.s., 2011.

McCarthy M., O'Dell F.: English Vocabulary in Use, Upper-Intermediate. CUP, 1994.

Fictumova J., Ceccarelli J., Long T.: Angličtina, konverzace pro pokročilé. Barrister and Principal, 2008.

Peters S., Gráf T.: Time to practise. Polyglot, 2007.

Jones L.: Communicative Grammar Practice. CUP, 1985.

Additional study materials.

Course language:

English language, B2-C1 level according to CEFR

Notes:

Course assessment

Total number of assessed students: 299

A	В	С	D	Е	FX
45.48	20.74	17.39	7.69	6.02	2.68

Provides: Mgr. Ivana Kupková, PhD.

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Date of last modification: 11.02.2024

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: CJP/ Course name: Communicative Grammar in English

PFAJGA/07

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of ECTS credits: 2

Recommended semester/trimester of the course:

Course level: I.

Prerequisities:

Conditions for course completion:

Active classroom participation (maximum 2 absences tolerated), homework assignments completed by given deadlines.

Powerpoint presentation of a topic related to the study field.

Final Test - end of semester, no retake

Final assessment = average of test and presentation.

Grading scale: A 93-100%, B 86-92%, C 79-85%, D 72-78%, E 65-71%, FX 64% and less

Learning outcomes:

The development of students' language skills - reading, writing, listening, speaking, improvement of their communicative linguistic competence. Students acquire knowledge of selected phonological, lexical and syntactic aspects, development of pragmatic competence. Students can efectively use the language for a given purpose, with focus on Academic English and English on level B2.

Brief outline of the course:

Selected aspects of English grammar and pronunciation

Word formation

Contrast of tenses in English

The passive voice

Types of Conditionals

Phrasal verbs and English idioms

Words order and collocations, prepositional phrases

Recommended literature:

Vince M.: Macmillan Grammar in Context, Macmillan, 2008 McCarthy, O'Dell: English Vocabulary in Use, CUP, 1994

www.linguahouse.com

esllibrary.com

bbclearningenglish.com

ted.com/talks

Course language:

English language, level B2 according to CEFR.						
Notes:						
Course assessment Total number of assessed students: 446						
A B C D E FX						
41.48	19.51	15.7	7.85	5.61	9.87	

Provides: Mgr. Lenka Klimčáková

Date of last modification: 20.09.2023

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: KGER/ | Course name: Communicative Grammar in German Language

NJKG/07

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of ECTS credits: 2

Recommended semester/trimester of the course:

Course level: I.

Prerequisities:

Conditions for course completion:

Active participation in class and completed homework assignments. Students are allowed to miss 2 classes at the most (2x90 min.). 2 control tests during the semester. Final grade will be calculated as follows: A 93-100 %, B 86-92%, C 79-85%, D 72-78%, E 65-71%, FX 64 % and less.

Learning outcomes:

The aim of the course is to identify and eliminate the most frequent grammatical errors in oral and written communication, learning language skills of listening comprehension, speaking, reading and writing, increasing students 'language competence (acquisition of selected phonological, lexical and syntactic knowledge), development of students' pragmatic competence (acquisition of the ability to express selected language functions), development of presentation skills, etc.

Brief outline of the course:

The course is aimed at practicing and consolidating knowledge of morphology and syntax of German in order to show the context in grammar as a whole. The course is intended for students who often make grammatical errors in oral as well as written communication. Through the analysis of texts, audio recordings, tests, grammar exercises, monologic and dialogical expressions of students focused on specific grammatical structures, problematic cases are solved individually and in groups. Emphasis is placed on the balanced development of grammatical thinking in the communication process, which ultimately contributes to the development of all four language skills.

Recommended literature:

Dreyer, H. – Schmitt, R.: Lehr- und Übungsbuch der deutschen Grammatik. Hueber Verlag GmbH & Co. Ismaning, 2009.

Krüger, M.: Motive Kursbuch, Lektion 1 – 30. Huebert Verlag GmbH & Co. Ismaning, 2020. Brill, L.M. – Techmer, M.: Deutsch. Großes Übungsbuch. Wortschatz. Huebert Verlag GmbH & Co. Ismaning, 2011.

Földeak, Hans: Sag's besser!. Grammatik. Arbeitsbuch für Fortgeschrittene. Huebert Verlag GmbH & Co. Ismaning, 2001.

Geiger, S. – Dinsel, S.: Deutsch Übungsbuch Grammatik A2-B2. Huebert Verlag GmbH & Co. Ismaning, 2018.

Dittelová, E. – Zavatčanová, M.: Einführung in das Studium der deutschen Fachsprache. Košice: ES UPJŠ, 2000.

Course language:

German, Slovak language

Notes:

Course assessment

Total number of assessed students: 57

A	В	С	D	Е	FX
61.4	10.53	8.77	3.51	8.77	7.02

Provides: Mgr. Ulrika Strömplová, PhD.

Date of last modification: 12.07.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: Course name: Conflict Management ECo-C3

KPPaPZ/ECo-C3/14

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: combined, present

Number of ECTS credits: 4

Recommended semester/trimester of the course: 3., 5.

Course level: I., N

Prerequisities:

Conditions for course completion:

The conditions for completing the course are as follows:

- 1. Active participation in exercises
- 2. Submission of reflection within the set deadline on the selected topic.

Attendance at seminars is mandatory - the student may have two absences during the semester.

The evaluation of the course and its subsequent completion will be based on clearly and objectively set requirements, which will be set in advance and will not change. The aim of the assessment is to ensure an objective and fair mapping of the student's knowledge while adhering to all ethical and moral standards. There is no tolerance for students' fraudulent behavior, whether in the teaching process or in the assessment process.

Learning outcomes:

Successful mastery and demonstration of knowledge in the field of conflict management and control of basic rules.

The method of teaching the subject will be oriented to the student. Lecturers will be interested in students' needs, expectations and opinions so as to encourage them to think critically by expressing respect and feedback on their opinions and needs.

The content of the curriculum will be based on primary and high-quality sources that will reflect the topicality of the topics so as to ensure the connection of the curriculum with other subjects and also the connection of the curriculum with practice. Students will be expected to take an active approach in lectures and seminars with an emphasis on their independence and responsibility.

The student is able to demonstrate an understanding of an individual's behavior in various conflict situations. The student is able to describe, explain and evaluate their own internal resources, competencies as well as limitations and weaknesses that are directly related to conflict management. The student is able to apply theoretical knowledge and principles of conflict resolution to everyday situations.

Brief outline of the course:

Disputes and their causes (Types of disputes, External influences, Be able to reveal the causes of disputes), Dispute origin (Levels of disputes, Escalation warning signals, Escalation removal strategies, Know how to explain escalation stages; How do I approach a dispute?) Dispute Resolution, Dispute Resolution Strategies, Dispute Discussion, Dispute Settlement Initiatives,

Knowing how to handle a dispute and how to effectively resolve it), Dispute Resolution (Options, Public Struggle, Covert Struggle, Indefinite Postponement, Agreement, "Fair play", compromise, cooperation, capitulation, escape or separation), Prevention (Structures that produce disputes, The meaning and purpose of disputes, Stages and steps of dispute resolution, What does a positive corporate culture mean? Dispute is an incentive for change)

Recommended literature:

Course language:

Notes:

Course assessment

Total number of assessed students: 145

abs	n
94.48	5.52

Provides: Mgr. Ondrej Kalina, PhD.

Date of last modification: 24.06.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ Cour

Course name: Coordination Chemistry

KCHU/03

Course type, scope and the method: Course type: Lecture / Practice

Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14

Course method: present

Number of ECTS credits: 4

Recommended semester/trimester of the course: 5.

Course level: I.

Prerequisities: ÚCHV/ACHU/21

Conditions for course completion:

Final written exam

Learning outcomes:

The student acquires basic knowledge on the coordination compounds, preparation, isomerism and properties of coordination compounds as well as about the chemical bonding in coordination compounds.

Brief outline of the course:

- 1. Definition and nomenclature of coordination compounds.
- 2. Central atom and ligands
- 3. Coordination numbers, coordination polyhedra.
- 4. Isomerism of coordination compounds
- 5. Preparation of coordination compounds
- 6. Stability of coordination compounds
- 7. Chemical bonding in coordination compounds.

Recommended literature:

- J. Ribas: Coordination Chemistry, Wiley-VCH, Weinheim, 2008.
- J. C. Huheey, E. A. Keiter, R. L. Keiter: Inorganic Chemistry, Haper Collins, New York, 1993.
- G. A. Lawrance: Introduction to Coordination Chemistry, Wiley, 2010.

Course language:

Notes:

Course assessment

Total number of assessed students: 97

A	В	С	D	Е	FX
41.24	25.77	13.4	7.22	10.31	2.06

Provides: prof. RNDr. Juraj Černák, DrSc., doc. RNDr. Juraj Kuchár, PhD.

Date of last modification: 10.09.2021

	COURSE INFORMATION LETTER
University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚMV/ DSMa/10	Course name: Discrete mathematics I
Course method: pre	re / Practice rse-load (hours): study period: 28 / 28 esent
Number of ECTS cr	
Recommended seme	ster/trimester of the course: 3.
Course level: I.	
Prerequisities:	
Conditions for cours Examination.	e completion:
appreciate mathemat	ome factual knowledge of combinatorics and graph theory. To understand an ical notions, definitions, and proofs, to solve problems requiring more than and to express mathematical thoughts precisely and more rigorously.
Recurrence: Some m miscellaneous method The inclusion-exclusion Introduction to graphs Planarity. Polyhedra. Traveling round a graphs	al coefficients, Binomial theorem, polynomial theorem. iscellaneous problems, Fibonacci-type relations, Using generating functions,
2. J. Matoušek and J. New York 1999. 3. S. Jendrol', P. Miho	ature: It course in discrete mathematics, Springer-Verlag London, 2001. Nešetřil, Invitation to discrete mathematics, Oxford University Press Inc., Sk: Diskrétna matematika I, UPJŠ Košice 1992.
Course language: Slovak	

Notes:

Course assessment					
Total number of assessed students: 398					
A	В	С	D	Е	FX
17.84	20.35	21.86	22.11	14.82	3.02

Provides: doc. RNDr. Roman Soták, PhD., RNDr. Alfréd Onderko, PhD., RNDr. Zuzana Šárošiová, PhD.

Date of last modification: 16.04.2022

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚMV/ Course name: Discrete mathematics II DSM2b/22 Course type, scope and the method: Course type: Lecture / Practice **Recommended course-load (hours):** Per week: 2 / 2 Per study period: 28 / 28 Course method: present **Number of ECTS credits: 4** Recommended semester/trimester of the course: 4., 6. Course level: I. Prerequisities: ÚMV/DSMa/10 or ÚMV/DSM3a/10 **Conditions for course completion:** In the covered areas of graph theory, the ability to formulate definitions and statements, to present proofs of statements, to explain individual steps in proofs and to solve selected problems related to given topics is required. During the semester (continuous assessment) two tests take place, from which 50% of points can be obtained, and from the oral exam alike 50% can be obtained. Evaluation: A ... at least 90%, B ... at least 80%, C ... at least 70%, D ... at least 60%, E ... at least 50%, FX ... less than 50%. **Learning outcomes:** Acquired knowledge of basic areas of graph theory, overview of used objects and properties, understanding of important statements and methods, knowledge of possible applications and the ability to formulate and solve problems in this area. **Brief outline of the course:** - (week 1) Introduction to graphs (graph relations, graph operations, special graph classes) - (week 2-3) Connectivity and distance in graphs (connectedness of vertices, eccentricity, incidence matrix) - (week 4) (Spanning) Trees (trees isomorphism) - (week 5-6) Connectivity in graphs (vertex and edge k-connectedness) - (week (7-8) Independence and coverings (independent set, matching, vertex and edge covering) - (week 9-10) Extremal graph theory (Ramsey numbers, Turán graphs) - (week 11-13) Graph colorings (vertex coloring, chromatic polynomial, edge coloring) - (week 14) Directed graphs (strong/weak connectedness, tounaments, acyclic graphs) **Recommended literature:** 1. A. Bondy, U.S.R. Murty, Graph theory, Springer, 2008 2. G. Chartrand, L. Lesniak, P. Zhang, Graphs and digraphs, CRC Press, 2011 3. R. Diestel, Graph Theory, Springer, 2017 4. D. West, Introduction to Graph Theory, Pearson, 2001 Course language: Slovak

Notes:

Course assessm	Course assessment						
Total number of assessed students: 224							
A	В	С	D	Е	FX		
14.29	11.61	25.0	25.0	19.2	4.91		

Provides: RNDr. Igor Fabrici, Dr. rer. nat., univerzitný docent, RNDr. Daniela Matisová

Date of last modification: 16.04.2022

	COURSE INFORMATION LETTER
University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: KPPaPZ/PUDB/15	Course name: Drug Addiction Prevention in University Students
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	rse-load (hours): ady period: 28 esent
	ster/trimester of the course: 3., 5.
Course level: I.	
Prerequisities:	
participation in works 50 - 45: A; 44 - 40:	active participation in the training part (30p). 2nd part of the evaluation: active shops (20p). In total, students can get 50p and the final evaluation is as follows: B; 39-35: C; 34-30: D; 29 - 25: E 24 and less: FX. Detailed information in board of the course in AIS2. The teaching of the subject will be realized by
describe and explain substance use. Studer of substance and non The student is also a approaches in preven The student is able to	ands the principals of research data based prevention of risk behavior, can the determinants of risk behavior as well as protective and risk factors for at understands and adequately interprets the theory explaining the background substance addictions. The abelian conditions are the state and classify the types and forms of prevention, strategies and ation, can distinguish effective strategies from ineffective ones. In adequately interpret their experience with preventive activities in the group ative effect as well as limitations and threats.
Brief outline of the c	ourse:
internetu v školskej p Sloboda, Z., & Buko and Practice. New Yo National and internat	012). Základy prevencie užívania drog a problematického používania braxi. Košice: UPJŠ. ski, J. (Eds.). (2006). Handbook of Drug Abuse Prevention: Theory, Science,
Course language:	

Page: 52

slovak

Notes:

	Course assessment					
Total number of assessed students: 616						
	A	В	C	D	Е	FX
	78.41	15.91	3.73	1.46	0.16	0.32

Provides: prof. PhDr. Ol'ga Orosová, CSc., Mgr. Lucia Barbierik, PhD., Mgr. Viera Čurová, PhD., Mgr. Janka Liptáková

Date of last modification: 24.06.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚINF/ | **Course name:** Educational software

EDS/15

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of ECTS credits: 2

Recommended semester/trimester of the course: 5.

Course level: I.

Prerequisities:

Conditions for course completion:

Conditions for ongoing evaluation:

- 1. Creation of a worksheet for student.
- 2. Creation of a multimedia educational game.
- 3. Creation of an interactive educational guiz.
- 4. Creation of an instructional educational video.

Conditions for the final evaluation:

Creation and presentation of final project on the use of educational software in education.

Conditions for successful completion of the course:

Obtaining at least 50% of points for ongoing and final assignments.

Learning outcomes:

Students will receive, resp. deepen their basic skills in working with:

- a) presentation software, programs for creating and editing images, animations, diagrams, sounds, conceptual maps,
- b) programs for the creation of didactic tests, questionnaires, surveys,
- c) simulation and modeling software,
- d) selected subject-oriented educational programs,

Students present and discuss their idea of the use of educational software and educational Internet resources and tools in the selected school subject.

Brief outline of the course:

- 1. Overview of educational software and educational web resources and tools.
- 2. Creating and processing of materials for teaching aid.
- 3. Creation and use of electronic and interactive educational documents (worksheets, presentations, textbooks and workbooks).
- 4. Creation of instructional educational video.
- 5. Electronic voting and questionnaire creation.
- 6. Creation of didactic tests and educational games. Gamification elements, tools and environments.
- 7. Collaborative web applications.
- 8. Online communication tools.
- 9. Complex online learning environments.

- 10. Online educational platforms, repositories, projects and competitions.
- 11. Simulations and modelling. Subject-focused educational programmes.
- 12. Use digital tools to plan, monitor, differentiate and personalise learning. Accessibility of digital tools and learning resources.

Recommended literature:

SOLOMON, Gwen and Lynne SCHRUM, 2014. Web 2.0 How-to for Educators. Second. International Society for Technology in Education, 314 p. ISBN 978-1564843517.

STOBAUGH, Rebecca, 2019. Fifty Strategies to Boost Cognitive Engagement: Creating a Thinking Culture in the Classroom (50 Teaching Strategies to Support Cognitive Development). Solution Tree Press, 176 p. ISBN 978-1947604773.

LEMOV, Doug, 2015. Teach Like a Champion 2. 0: 62 Techniques That Put Students on the Path to College [online]. 2nd edition. John Wiley & Sons, Incorporated, 509 p. [cited 2021-7-10]. ISBN 9781118898628. Available from: https://ebookcentral.proquest.com/lib/upjs-ebooks/detail.action?docID=1895720

European Schoolnet: Transforming education in Europe [online]. [cited 2021-7-10]. Available from: http://www.eun.org/home

Science On Stage Europe [online]. Science on Stage Europe e.V. [cited 2021-7-10]. Available from: https://www.science-on-stage.eu/

Course language:

Slovak and partly English due to selected programs and information sources

Notes:

By default, teaching is carried out face to face. If this is not possible (eg due to a pandemic), teaching is provided at a distance through video conferencing programs and LMS.

Course assessment

Total number of assessed students: 91

A	В	С	D	Е	FX
73.63	13.19	7.69	0.0	5.49	0.0

Provides: doc. RNDr. L'ubomír Šnajder, PhD., Mgr. Katarína Brinziková

Date of last modification: 16.03.2024

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: CJP/ Course n

PFAJ4/07

Course name: English Language of Natural Science

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of ECTS credits: 2

Recommended semester/trimester of the course: 4.

Course level: I.

Prerequisities:

Conditions for course completion:

Active participation in class and completed homework assignments. Students are allowed to miss 2 classes at the most

Continuous assessment:

1 credit test taken presumably in weeks 6/7

1 project (quiz on the topic of the student's field of study) 25% of the continuous assessment

5 LMS quizzes (25% of the continuous assessment)

In order to be admitted to the final exam, a student has to score at least 65 % from the continuous assessment

The exam test results represent 50% of the final grade for the course, continuous assessment results represent the other 50% of the final grade.

The final grade for the course will be calculated as follows:

A 93-100, B 86-92, C 79-85, D 72-78, E 65-71, FX 64 and less.

Learning outcomes:

Enhancement of students' language skills (speaking, writing, reading and listening comprehension) in English for specific and academic purposes and development of students' linguistic competence. Students obtain knowledge of selected phonological, lexical and syntactic aspects of professional English, improve their pragmatic competence - students can effectively use the language for a given purpose, and acquire presentation skills at B2 level (CEFR) with focus on terminology of natural sciences

Brief outline of the course:

- 1. Introduction to studying language
- 2. Selected aspects of scientific language
- 3. Talking about academic study
- 4. Discussing science
- 5. Defining scientific terminology and concepts
- 6. Expressing cause and effect
- 7. Describing structures
- 8. Explaining processes
- 9. Comparing objects, structures and concepts

- 10. Talking about problem and solution
- 11. Referencing authors
- 12. Giving examples
- 13. Visual aids and numbers
- 14. Referencing time and place

Presentation topics related to students' study fields.

Recommended literature:

lms.upjs.sk - e-kurz Odborný anglický jazyk pre prírodné vedy.

Redman, S.: English Vocabulary in Use, Pre-intermetdiate, Intermediate. Cambridge University Press, 2003.

Armer, T.: Cambridge English for Scientists. CUP, 2011.

Wharton J.: Academic Encounters. The Natural World. CUP, 2009.

P. Fitzgerald: English for ICT studies. Garnet Publishing, 2011.

https://worldservice/learningenglish, https://spectator.sme.sk

www.isllibrary.com

linguahouse.com

Course language:

English, level B2 (CEFR)

Notes:

Course assessment

Total number of assessed students: 3075

A	В	С	D	Е	FX
38.44	26.08	16.46	9.53	7.45	2.05

Provides: Mgr. Viktória Mária Slovenská, Mgr. Lenka Klimčáková

Date of last modification: 06.02.2024

Approved: doc. RNDr. Stanislav Lukáč, PhD., prof. RNDr. Vladimír Zeleňák, DrSc.

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University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ Course na

Course name: Food chemistry

PCH1/00

Course type, scope and the method:

Course type: Lecture / Practice Recommended course-load (hours):

Per week: 2 / 1 Per study period: 28 / 14

Course method: present

Number of ECTS credits: 4

Recommended semester/trimester of the course: 5.

Course level: I., II.

Prerequisities:

Conditions for course completion:

Active work during semester, presentation on certain theme. Two exams, one in the middle and second at the end of semester (min. 51%). A: 91-100b, B: 81-90b, C: 71-80b, D: 61-70b, E: 51-60b, FX: 0-50b.

Learning outcomes:

Students will recieve informations and knowledges about chemical substances in food, their importance and chemical changes in food during processing and storage.

Brief outline of the course:

The main categories of substances in the most important group of food. Aminoacids, proteins, lipids, carbohydrates. Water, minerals, low concentration anorganic compounds, vitamins. Hydrocarbons, colorants, toxic compounds, aditives. Chemical reactions in dairy products.

Recommended literature:

Course language:

english

Notes:

Teaching is carried out in person or, if necessary, online using the MS Teams or BBB (BigBlueButton) tool. The form of teaching is specified by the teacher at the beginning of the semester, updated continuously.

Course assessment

Total number of assessed students: 316

A	В	С	D	E	FX
68.04	27.53	4.11	0.0	0.0	0.32

Provides: RNDr. Ján Elečko, PhD.

Date of last modification: 28.01.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚMV/ | **Course name:** Function of real variable

FRPa/19

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 4 Per study period: 28 / 56

Course method: present

Number of ECTS credits: 7

Recommended semester/trimester of the course: 1.

Course level: I.

Prerequisities:

Conditions for course completion:

Continuous assessment of student's work during the semester (submission of compulsory homework, writing three tests). Final test and oral discussion on the topics of the subject.

Learning outcomes:

The course provides an introductory knowledge on basic tools of differential and integral calculus of real functions of one real variable, and a development of certain calculation skills in the field.

Brief outline of the course:

- 1. Basics of mathematical logic and notations (1 week)
- 2. Real functions basic notions, operation, graphs and their transformations (2 weeks)
- 3. Continuity of a real-valued function (1 week)
- 4. Derivative of a function using the geometric concepts, rules of differentiation (2 weeks)
- 5. Basic of differential calculus relations with monotonicity and convexity, extremas, using in optimisation, geometric and physics tasks (2 weeks)
- 6. Primitive function, methods of their finding (3 weeks)
- 7. Newton definite integral methods of its computation, using in geometric and physics tasks (2 weeks)

Recommended literature:

- 1. Kulcsár, Š. Kulcsárová, O.: Zbierka úloh z matematickej analýzy I., UPJŠ, 2002.
- 2. Kulcsár, Š. Kulcsárová, O.: Zbierka úloh z matematickej analýzy II., UPJŠ, 2003.
- 3. Hutník, O. Kulcsár, Š. Kulcsárová, O. Mojsej, I.: Zbierka úloh z matematickej analýzy III., UPJŠ, 2011.
- 4. Demidovič, B. P.: Sbírka úloh a cvičení z matematické analýzy, Fragment, Praha, 2003.
- 5. Brannan, D.: A First Course in Mathematical Analysis, Cambridge University Press, Cambridge 2006.
- 6. Bruckner, A. M., Bruckner J. B., Thomson, B. S.: Real Analysis, Second Edition, ClassicalRealAnalysis.com, 2008.
- 7. Zorich, V. A.: Mathematical Analysis I, Springer-Verlag 2002.

Course language:

Slovak

Notes:								
Course assessment Total number of assessed students: 839								
A B C D E FX								
8.82	8 22	16.92	21 33	31.7	12.99			

Provides: prof. RNDr. Ondrej Hutník, PhD., RNDr. Lenka Halčinová, PhD., RNDr. Jana Borzová, PhD., Mgr. Kristína Hurajová

Date of last modification: 16.04.2022

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ **Course name:** Fundamentals of Bioanalytical Chemistry BACHZ/06 Course type, scope and the method: Course type: Lecture / Practice **Recommended course-load (hours):** Per week: 2 / 1 Per study period: 28 / 14 Course method: present **Number of ECTS credits: 5** Recommended semester/trimester of the course: 3. Course level: I. **Prerequisities: Conditions for course completion:** Elaboration and presentation of a semester project with an assigned topic. Completion of block exercises. Oral examination. Detailed conditions for completing the subject are listed in the electronic bulletin board of the subject and in the repository of digital support materials LMS UPJS and are updated annually. **Learning outcomes:** After completing the course, the student has basic knowledge about biological samples, factors affecting biological samples and analytical methods used in clinical chemistry and bioanalysis. Brief outline of the course: Introduction to Bioanalytical Chemistry. Biological samples classification. Factors that affect analytes in biological samples. Collection, transport and storage of samples, the main principles of sampling, the suppressing of undesirable phenomena. Selected methods of pretreatment of biological samples. Analyzers, equipment and organization of work in a clinical laboratory. Control and management of quality in clinical laboratory. Quality manual, calibration, control, and reference materials. Validation and Good Laboratory Practice. Buffers in bioanalysis. Enzymes in bioanalysis, introduction, distribution, Mechanism of enzyme catalysis. The kinetics of enzymatic reactions with one substrate, the Michaelis constant, constant specificity, lag phase, kinetics of reactions with two substrates. Moderators of enzyme activity. Selected methods for the analysis of biomolecules. **Recommended literature:** 1. Chromý, V. a kol.: Bioanalytika, MU Brno, 2002 2. Kukačka, J. a kol.: Bioanalytická chemie v príkladech a cvičeních, Karolinum, 2010 3. Mikkelsen, S.R, Cortón E.: Bioanalytical Chemistry, Wiley, 2004 4. Wilson I.: Bioanalytical Separations 4, (Handbook of Analytical Separations), Elsevier, 2003 5.Lee, D.C., Webb, M.: Pharmaceutical Analysis, Blackwell, 2003

Course language:

Notes:

If necessary, the teaching also takes place in a distance form with the use of various tools of LMS UPJŠ, MS teams, etc. The form of teaching is specified by the teacher at the beginning of the semester, it is continuously updated.

Course assessment

Total number of assessed students: 108

A	В	С	D	Е	FX
33.33	30.56	30.56	4.63	0.0	0.93

Provides: doc. RNDr. Katarína Reiffová, PhD.

Date of last modification: 22.07.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ Course

Course name: General Chemistry

VCHU/15

Course type, scope and the method:

Course type: Lecture / Practice Recommended course-load (hours): Per week: 4 / 2 Per study period: 56 / 28

Course method: present

Number of ECTS credits: 7

Recommended semester/trimester of the course: 1.

Course level: I.

Prerequisities: ÚCHV/CHV1/99

Conditions for course completion:

Written test in the middle and the end of the semester followed by the oral examination. Active participation on seminars.

Learning outcomes:

To provide students with knowledge of atoms and molecules their electronic structure, theories of chemical bonds, physical and chemical properties of elements and compounds as well as their periodicity.

Brief outline of the course:

Main terms used in chemistry. Atoms – models of atoms, electron configuration, chemical periodicity and its effect on the properties of elements, radioactivity. Chemical bonds and intermolecular interactions. Chemical structure and physical properties of matter. State of matter. Solutions. Chemical equilibrium. Basis of chemical thermodynamics and chemical kinetics. Classification of chemical reactions. Electrochemistry.

Recommended literature:

- 1. Atkins P., Jones L.: Chemical Principles, 2nd ed., Freeman, New York 2002.
- 2. Russel J.B.: General Chemistry, 2nd ed., McGraw Hill, London 1992.

Course language:

Notes:

Course assessment

Total number of assessed students: 361

A	В	С	D	Е	FX
25.76	27.42	27.7	11.08	7.48	0.55

Provides: prof. RNDr. Vladimír Zeleňák, DrSc., doc. RNDr. Ivan Potočňák, PhD.

Date of last modification: 07.02.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ | Course name: General Course of Analytical Chemistry - Laboratory

PACU/03

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 4 Per study period: 56

Course method: present

Number of ECTS credits: 4

Recommended semester/trimester of the course: 4.

Course level: I.

Prerequisities: ÚCHV/ANCHU/03 or ÚCHV/ANCHU/21

Conditions for course completion:

Active participation in laboratory exercises and seminars; successful completion of the tests.

- 1. Participation in laboratory exercises is required. Assigned teacher who leads exercises might excuse without substitute the student's absence (incapacity for work, family reasons, etc.) for a maximum of two exercises during the semester with substitute supplying.
- 2. The assigned teacher, who leads the seminar, assesses the preparation of students and their activity in seminars. For the active participation in the exercises, the student can get a maximum of 10 points.
- 3. Two written tests are obligatory. The written test will consist of 15 questions with 15 points, together for 2 written testes of 30 points. To successful completion of the exam, it is necessary to achieve at least 8 points from each test.

Overall score: Max. number of points: 50 (elaboration of protocols / assignments - 10 points; active participation in practical exercises - 10 points; written tests - 2×15 points). Min. number of points to successful completion of course: 26.

Note: Detailed conditions are updated annually within the repository for digital support materials (LMS UPJŠ).

Learning outcomes:

Application of theoretical knowledge of qualitative and quantitative analytical chemistry into analytical laboratory practise.

Brief outline of the course:

Practical in qualitative and quantitative analysis. Qualitative analysis, separation by selective precipitation. Quantitative methods. Gravimetry, general principles of method. Volumetric methods. Preparation of accurate solutions. Indication of equvivalency point. Titration curves, calculations in volumetric analysis. Acidimetry, alkalimetry. Manganometry. Iodometry. Complexometry. Selected Instrumental analytical methods.

Recommended literature:

- 1. Y. Bazel a kol.: Praktikum z analytickej chémie, PF UPJŠ, Košice 2019.
- 2. T. Gondová a kol.: Praktikum z analytickej chémie, PF UPJŠ, Košice 1999.
- 3. V. Szmereková, P.Meľuch: Praktikum z analytickej chémie, PF UPJŠ, Košice 1988.
- 4. J. Labuda a kol. Analytická chémia, STU, Bratislava 2014.
- 5. Z. Holzbecher a kol: Analytická chemie, SNTL, ALFA Praha 1987.

6. L. Koller: Analytická chémia, TU Košice, 2002, skriptum a v digitálnej forme.

7.D. Harvey: Modern Analytical Chemistry. McGraw Hill, Boston, 2000.

Course language:

Slovak

Notes:

The course is implemented by full-time or, if necessary, distance method using the MS Teams or BBB or a combined method. The form of teaching is specified by the teacher at the beginning of the semester and updated continuously.

Course assessment

Total number of assessed students: 404

A	В	С	D	Е	FX
58.17	28.22	11.14	1.24	1.24	0.0

Provides: RNDr. Rastislav Serbin, PhD., RNDr. Jana Šandrejová, PhD., univerzitná docentka

Date of last modification: 15.11.2021

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚMV/ | **Course name:** Geometry I

GEO2a/22

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours):

Per week: 2 / 1 Per study period: 28 / 14

Course method: present

Number of ECTS credits: 3

Recommended semester/trimester of the course: 2.

Course level: I., II.

Prerequisities:

Conditions for course completion:

In the covered areas of geometry, the ability to formulate definitions and statements, to present proofs of statements, to explain individual steps in proofs and to solve selected problems related to given topics is required. Evaluation: A ... at least 90%, B ... at least 80%, C ... at least 70%, D ... at least 50%, FX ... less than 50%

Learning outcomes:

Acquired knowledge about the axiom system of Euclidean geometry, about the validity of the basic tools of planimetry, about sets of points of a given property, about congruence transformations and homothety in the plane, about important points, lines and circles in triangles and about quadrilaterals and their properties. The ability to use the above knowledges and tools to solve problems on this area. A new look at classical geometric results.

Brief outline of the course:

- (week 1-3) Hilbert's axiom system (axioms, triangle congruence theorems, pairs of congruent or "complementary" angles, basic proportionality theorem, triangle similarity theorems)
- (week 4-5) Basic tools of planimetry (Euclid's theorem, Pythagorean theorem, Thales' theorem, law of cosines, extended law of sines, central and inscribed angle theorem, area of a triangle)
- (week 6) Point sets of the given property (bisectors, equidistants, Apollonius circle)
- (week 7) Transformations (congruence transformations of the plane, homothety in the plane)
- (week 8-11) Points and lines connected with a triangle (Menelaus's theorem, Ceva's theorem, points of interest, the incircle and excircles, pedal triangles, Euler line, nine-point circle, Simson lines)
- (week 12-13) Quadrangles (Varignon's parallelogram, cyclic quadrangles, Ptolemy's theorem, Brahmagupta's formula)

Recommended literature:

- 1. D. Hilbert, Grundlagen der Geometrie, Teubner, 1968.
- 2. H.G. Forder, Foundations of Euclidean geometry, Dover Publ., 1958.
- 3. H.S.M. Coxeter, S.L. Greitzer, Geometry revisited, MAA, 1967.
- 4. R.A. Johnson, Advanced Euclidean geometry, Dover Publ., 2007.
- 5. D.A. Brannan, M.F. Esplen, J.J. Gray, Geometry, Cambridge Univ. Press, 2007.

Course language:

Slovak

Notes:

Course assessment

Total number of assessed students: 194

A	В	С	D	Е	FX
19.07	19.07	29.38	11.34	16.49	4.64

Provides: RNDr. Igor Fabrici, Dr. rer. nat., univerzitný docent

Date of last modification: 29.02.2024

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚMV/ | **Course name:** Geometry I

GEO2a/21

Course type, scope and the method: Course type: Lecture / Practice

Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14

Course method: present

Number of ECTS credits: 3

Recommended semester/trimester of the course: 3., 5.

Course level: I.

Prerequisities:

Conditions for course completion:

In the covered areas of geometry, the ability to formulate definitions and statements, to present proofs of statements, to explain individual steps in proofs and to solve selected problems related to given topics is required. Evaluation: A ... at least 90%, B ... at least 80%, C ... at least 70%, D ... at least 50%, FX ... less than 50%

Learning outcomes:

Acquired knowledge about the axiom system of Euclidean geometry, about the validity of the basic tools of planimetry, about sets of points of a given property, about congruence transformations and homothety in the plane, about important points, lines and circles in triangles and about quadrilaterals and their properties. The ability to use the above knowledges and tools to solve problems on this area. A new look at classical geometric results.

Brief outline of the course:

- (week 1-3) Hilbert's axiom system (axioms, triangle congruence theorems, pairs of congruent or "complementary" angles, basic proportionality theorem, triangle similarity theorems)
- (week 4-5) Basic tools of planimetry (Euclid's theorem, Pythagorean theorem, Thales' theorem, law of cosines, extended law of sines, central and inscribed angle theorem, area of a triangle)
- (week 6) Point sets of the given property (bisectors, equidistants, Apollonius circle)
- (week 7) Transformations (congruence transformations of the plane, homothety in the plane)
- (week 8-11) Points and lines connected with a triangle (Menelaus's theorem, Ceva's theorem, points of interest, the incircle and excircles, pedal triangles, Euler line, nine-point circle, Simson lines)
- (week 12-13) Quadrangles (Varignon's parallelogram, cyclic quadrangles, Ptolemy's theorem, Brahmagupta's formula)

Recommended literature:

- 1. D. Hilbert, Grundlagen der Geometrie, Teubner, 1968.
- 2. H.G. Forder, Foundations of Euclidean geometry, Dover Publ., 1958.
- 3. H.S.M. Coxeter, S.L. Greitzer, Geometry revisited, MAA, 1967.
- 4. R.A. Johnson, Advanced Euclidean geometry, Dover Publ., 2007.
- 5. D.A. Brannan, M.F. Esplen, J.J. Gray, Geometry, Cambridge Univ. Press, 2007.

Course language:

Slovak

Notes:

Course assessment

Total number of assessed students: 161

A	В	С	D	Е	FX
19.88	20.5	29.19	11.8	14.29	4.35

Provides: RNDr. Igor Fabrici, Dr. rer. nat., univerzitný docent

Date of last modification: 29.02.2024

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚMV/ | Course name: Geometry II

GEO2b/22

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 1 / 1 Per study period: 14 / 14

Course method: present

Number of ECTS credits: 2

Recommended semester/trimester of the course: 3.

Course level: I.

Prerequisities: ÚMV/GEO2a/22

Conditions for course completion:

Mastering the terminology of stereometry, basic properties of geometric solids, understanding concepts, basic stereometric definitions and theorems.

Understanding and using basic transformation methods for projection of solids,

effective use of suitable methods in the construction of planar cutting bodies, in the construction of the intersection of a line with a solid and in solving metric problems.

The conditions of the continuous assessment are active participation in the exercises, elaboration of home assignments and elaboration of two tests. Evaluation: A ... at least 90%, B ... at least 80%, C ... at least 70%, D ... at least 60%, E ... at least 50%, FX ... less than 50%

Learning outcomes:

An important result of education is the deepening and developing of knowledge of school stereometry and the development of the ability to apply a synthetic approach in deriving and proving relationships in stereometry and in their use in solving problems. The construction of solid images and problem solving will develop analytical thinking and spatial imagination of students.

Brief outline of the course:

- basic properties of geometric solids in space,
- images of solids in parallel projection,
- basic stereometric theorems (relative positions of straight lines, parallelism of a line and a plane, parallelism of two planes, relative position of three planes, perpendicularity of a line and a plane, perpendicularity of two planes),
- positional and metric properties of spatial solids (cuttings of polyhedrons, distances and angles of points, straight lines, planes, intersection of a straight line with a solid, intersection of planes),
- properties of polyhedrons, Euler's theorem, regular polyhedrons (Platonic solids, their number and properties)
- volume and surface area of solids and their parts, Cavalieri's principle
- projection methods (principle of parallel and central projection, axial affinity, use of axial affinity in the construction of cuts of prisms and cylinders, basics of Monge's Projection).

Recommended literature:

1. Pomykalová, E.: Matematika pro gymnázia - Stereometrie. Prometheus, 2009.

- 2. Šedivý, O., Pavlovičová, G., Rumanová, L., Vallo, D.: Stereometria. Umenie vidieť a predstavovať si priestor. Nitra, 2007.
- 3. Kuřina, F.: Deset pohledů na geometrii. Praha: MÚ AV ČR, 1996.

Course language:

Slovak

Notes:

Course assessment

Total number of assessed students: 18

A	4	В	С	D	Е	FX
11.	.11	5.56	16.67	16.67	44.44	5.56

Provides: doc. RNDr. Stanislav Lukáč, PhD.

Date of last modification: 20.04.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚMV/ | Course name: Geometry III

GEO2c/22

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours):

Per week: 2 / 2 Per study period: 28 / 28

Course method: present

Number of ECTS credits: 4

Recommended semester/trimester of the course: 4., 6.

Course level: I.

Prerequisities: ÚMV/ALG2b/22

Conditions for course completion:

Two written tests.

Written and oral examinations

For continuous evaluation - max. 40 points

for the written test - max. 20 points

for oral exams - max. 40 points)

Final score:

A: 100-91 points, B: 90-81, C: 80-71, D: 70-61, E: 60-51, F: less than 51 points Note: In each of the student needs to have at least 50% max. number of points

Learning outcomes:

Mastering the basics of the theory of linear and quadratic formations in the Affine and Euclidean space, mastering the methods of solving problems in analytical geometry in relation to the secondary school curriculum.

Brief outline of the course:

- 1. Affine n-dimensional space definition, linear coordinate system.
- 2. Subspace and its parametric expression, general equation of superplane, subspace as intersection of superstructures, general equations of subspace
- 3. Mutual position of subspaces, orientation of affine space, change of coordinate system
- 4. Arrangement of points on a line, half-spaces
- 5. Scalar product, external product, vector product of vectors and their basic properties
- 6. Euclidean space and its subspaces, Cartesian coordinate system
- 7. Perpendicularity of subspaces, distance of point from subspace, distance of point from superstructure, distance of subspaces,
- 8. Deviation of two lines, two superstructures, line and superplane, deviation of line and subspace
- 9. Axis of two extraterrestrial subspaces, Gram determinant, examples in E2 and E3

Recommended literature:

- 1. M.Sekanina, L.Boček, M.Kočandrle, J.Šedivý: Geometrie 1, SPN Praha 1986
- 2. M.Hejný, V.Zaťko, P.Kršňák: Geometria 1, SPN Bratislava 1985
- 3. J.Eliaš, J.Horváth, J.Kajan: Zbierka úloh z vyššej matematiky 1, Alfa Bratislava

Course language:

Slovak

Notes:

Course assessment

Total number of assessed students: 212

A	В	С	D	Е	FX
18.87	22.17	22.17	18.4	10.85	7.55

Provides: doc. RNDr. Dušan Šveda, CSc., Mgr. Daniela Šabaková, RNDr. Monika Krišáková

Date of last modification: 17.04.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚMV/ | Course name: Geometry IV

GEO2d/22

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 3 / 2 Per study period: 42 / 28

Course method: present

Number of ECTS credits: 5

Recommended semester/trimester of the course: 5.

Course level: I.

Prerequisities:

Conditions for course completion:

In the covered areas of geometry, the ability to formulate definitions and statements, to present proofs of statements, to explain individual steps in proofs and to solve selected problems related to given topics is required. During the semester (continuous assessment) two tests take place, from which 50% of points can be obtained, and from the oral exam alike 50% can be obtained. Evaluation: A ... at least 90%, B ... at least 80%, C ... at least 70%, D ... at least 60%, E ... at least 50%, FX ... less than 50%

Learning outcomes:

Acquired knowledge of the properties of affine, isometric and similarity transformations, understanding of important statements and methods, knowledge of the use of isometric and similarity transformations in construction and optimization problems and the ability to solve other problems in this area.

Brief outline of the course:

- (week 1-2) Quadric surfaces (circular and general quadric surfaces)
- (week 3-7) Affine transformations (associated transformation, matrix representation, affinities, fixed points and lines, pseudo-reflections)
- (week 8-10) Isometric transformations (matrix representation, isometries, classification in the plane, composition of reflections)
- (week 11-12) Similarity transformations (matrix representation, similarities, homothety, composition of homotheties)
- (week 13-14) Geometry of circles (the power of a point with respect to a circle, radical axis of two circles, pencils of circles)

Recommended literature:

- 1. M. Sekanina et al, Geometry 2, SPN, 1988 (in slovak).
- 2. O. Šedivý et al, Geometry 2, SPN, 1987 (in slovak).
- 3. H.S.M. Coxeter, Introduction to geometry, Wiley, 1989.
- 4. J.T. Smith, Methods of geometry, Wiley, 2000.

Course language:

Slovak

Notes:						
Course assessment Total number of assessed students: 195						
A B C D E FX						
15.38	15.9	24.1	19.49	18.46	6.67	

Provides: RNDr. Igor Fabrici, Dr. rer. nat., univerzitný docent, Mgr. Daniela Šabaková

Date of last modification: 14.04.2022

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: KPE/ Course name: Getting to know the Student in Education POŽ/21 Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: present **Number of ECTS credits: 2** Recommended semester/trimester of the course: 4. Course level: I. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 53 C Α В D Ε FX 75.47 13.21 3.77 0.0 0.0 7.55 Provides: PaedDr. Michal Novocký, PhD. Date of last modification: 12.03.2024 Approved: doc. RNDr. Stanislav Lukáč, PhD., prof. RNDr. Vladimír Zeleňák, DrSc.

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: KPE/ Course name: Inclusive Pedagogy **INP/17** Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: present **Number of ECTS credits: 2 Recommended semester/trimester of the course:** 5. Course level: I. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 107 C Α В D Е FX 69.16 22.43 3.74 1.87 2.8 0.0 Provides: PaedDr. Michal Novocký, PhD. Date of last modification: 12.03.2024

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University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚMV/ | Course name: Informatics course for teachers of mathematics

IPU/22

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 1 / 1 Per study period: 14 / 14

Course method: present

Number of ECTS credits: 2

Recommended semester/trimester of the course: 6.

Course level: I.

Prerequisities:

Conditions for course completion:

To master the use of basic algorithmic structures, to gain the ability to write algorithms for the construction of geometric shapes in the environment of turtle geometry. To be able to assess the possibilities of using interactive applications available on the Internet and to design procedures for the application of selected applications in the teaching of mathematics. To learn to use numerical and graphical means of a spreadsheet in data analysis, creating models to solve various mathematical problems.

Evaluation:

Algorithm creation paper - 6 b

Elaboration of dynamic constructions for solving geometric problems - 3 b

Seminar work on the use of interactive applications - 7 b + 3 b

Poll - 1 b

Tasks for creating numerical and graphical models in a spreadsheet - 4 b

Classification scale:

A: 91 % - 100 %,

B: 81 % - 90 %,

C: 71 % - 80 %,

D: 61 % - 70 %,

E: 51 % - 60 %,

FX: 0 % - 50 %.

Learning outcomes:

Knowledge and skills from the basics of working with standard information and communication technologies, which provide a variety of opportunities to support mathematics education. Skills to use basic commands of turtle geometry for generalization and writing algorithms for construction of geometric shapes. To master the basic principles of creating structures in the environment of dynamic geometry. Acquire creative and evaluative skills to plan and prepare a meaningful integration of modern technologies into mathematics education.

Brief outline of the course:

1-5: Use of basic algorithmic constructions in turtle geometry for the construction of geometric shapes,

- 6th 7th: Basics of work in the environment of dynamic geometry, creation of dynamic constructions,
- 8th 9th: Interactive teaching applications available on the Internet, selected possibilities of using digital technologies in mathematics education.
- 10. 12 .: Use of numerical and graphical representations of data and modeling in a spreadsheet environment in solving mathematical problems.

Recommended literature:

Brdička, B.: Role internetu ve vzdělávaní, 2003, http://it.pedf.cuni.cz/~bobr/role/econt.htm.

Lukáč, S. a kol.: IKT vo vyučovaní matematiky, Asociácia projektu Infovek 2002.

Vaníček, J.: Počítačové kognitivní technologie ve výuce geometrie. Pedagogická fakulta Univerzity Karlovy, 2009.

Šťastný, Z.: Matematické a statistické výpočty v Microsoft Excelu, Computer Press 2001.

Course language:

Slovak

Notes:

Course assessment

Total number of assessed students: 123

A	В	С	D	Е	FX
52.03	25.2	15.45	5.69	1.63	0.0

Provides: doc. RNDr. Stanislav Lukáč, PhD.

Date of last modification: 17.02.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ | Course name: Inorganic Chemistry

ACHU/21

Course type, scope and the method:

Course type: Lecture / Practice Recommended course-load (hours): Per week: 3 / 1 Per study period: 42 / 14

Course method: present

Number of ECTS credits: 5

Recommended semester/trimester of the course: 2.

Course level: I.

Prerequisities: ÚCHV/VCHU/15

Conditions for course completion:

Written test in the middle and the end of the semester followed by the oral examination. Active participation on seminars.

Learning outcomes:

Gaining knowledge about the properties and reactivity of elements and their compounds, the periodicity of their properties and the periodicity of the properties of their compounds. Knowledge of the basic physical and chemical properties of elements and their compounds, reactivity, their preparation, production and occurrence.

Brief outline of the course:

Electronic configuration, abundance, use, physical and chemical properties, preparation, reactivity of non-metallic elements hydrogen, halogens, oxygen, sulphur, nitrogen, phosphorus, carbon, silicon, boron and rare gases. Binary and other compounds formed by these elements, their properties and reactivity. Metals and transition elements. Abudance, properties, reactivity, important compounds.

Recommended literature:

Greenwood, N. N., Earnshaw, A: Chemistry of the Elements. Pergamon Press, Oxford, 1984 Atkins O., Overton T., Rourke J., Weller M., Armstrong F.: Inorganic Chemistry, University Press, Oxford, 2006.

Course language:

Notes:

Course assessment

Total number of assessed students: 55

A	В	С	D	Е	FX
36.36	32.73	14.55	9.09	7.27	0.0

Provides: prof. RNDr. Vladimír Zeleňák, DrSc., prof. RNDr. Juraj Černák, DrSc.

Date of last modification: 07.02.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ | Course name: Inorganic Chemistry II

ACH2/21

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 3 / 2 Per study period: 42 / 28

Course method: present

Number of ECTS credits: 6

Recommended semester/trimester of the course: 3., 5.

Course level: I.

Prerequisities: ÚCHV/ACH1/10 or ÚCHV/ACHU/21 or ÚCHV/ACHU/03

Conditions for course completion:

- 1. Students are required to attend seminars, this also applies to the online form of teaching. The relevant teacher who leads the seminar will justify the absence of the student (illnes, family reasons, etc.) in a maximum of two seminars during the semester without the need to replace the teaching hours. In the case of a longer justified absence (for example due to illness), the teacher will assign to the student alternative forms of duties;
- 2. Activity at seminars. The preparation of students and their activity in seminars is always assessed by the teacher who leads the seminar, within his / her competence.
- 3. Participation in 10 small written tests within the seminar, for each small test you can get 1 point. In the case of a justified absence of a student from a small test, the teacher may require to elaborate a written job. Successful completion is considered if the student obtains at least 5.5 points from these tests, which is a condition for participation in the exam. The points obtained from the seminar will be included in the total number of points obtained for the subject in the range of 10%.
- 4. The exam is usually carried out in written form (3 written tests, of which 2 tests during the semester) with the possibility of further oral examination, or, in case of restrictions of contact forms of the teaching, the exam will be performed in a suitable online electronic form.
- 5. To successfully complete the course, it is necessary to obtain at least 51% of the maximum number of points in each test and for seminars.

Learning outcomes:

To acquire knowledge about physical and chemical properties of metallic elements and their compounds.

Brief outline of the course:

General characterization of metals, chemistry of elements of the 1st and 2nd group, aluminum and other metals elements of groups 13 to 16. Chemistry of transition elements with emphasis on the 1st transition series.

Coordination compounds, chemistry of lanthanides and actinides. In all chapters are discussed the atomic properties of elements, properties of elements as substances, properties of their compounds. Emphasis is also put on environmental aspects of the properties of elements and their compounds. The lectures are discussed at the seminars in detail.

Recommended literature:

1. Greenwood, N.N., Earnshaw, A.: Chemistry of the elements, Pergamon Press N.Y., 1984.

2. D.F. Shriver, P.W. Atkins: Inorganic Chemistry, Oxford University Press, Oxford, 4th Ed., 2006.

Course language:

Notes

The subject can be realized in the form of personal attendance or, if necessary, also in online form.

Course assessment

Total number of assessed students: 29

A	В	С	D	Е	FX
13.79	24.14	44.83	10.34	3.45	3.45

Provides: prof. RNDr. Juraj Černák, DrSc., prof. RNDr. Vladimír Zeleňák, DrSc., RNDr. Miroslava Matiková Maľarová, PhD.

Date of last modification: 16.11.2021

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ Course name: Instrume

ANCH1b/21

Course name: Instrumental Analytical Chemistry

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 2 Per study period: 28 / 28

Course method: present

Number of ECTS credits: 5

Recommended semester/trimester of the course: 4., 6.

Course level: I.

Prerequisities:

Conditions for course completion:

Active participation in seminars; successful completion of the final test. Elaboration of 2 written assignments (or subject project), which will be one of the conditions for participation in the exam. The evaluation of the student's study results within the study of the subject is carried out by a combination of continuous control during the teaching part of the semester (50%) with an examination during the examination period (50%).

Note: Detailed conditions are updated annually within the repository for digital support materials (LMS UPJŠ).

Learning outcomes:

The student acquires knowledge of the theoretical foundations and instrumentation in analytical chemistry.

Brief outline of the course:

Classification of instrumental analytical methods. Basic parts of analytical instruments. Comparison of range, accuracy, detection limit, selectivity and economic characteristics of analytical methods. Analytical signal and calibration. Detection limit. Standard addition method. Accuracy and precision. Spectral methods. Electromagnetic radiation. Analytical signal of the optical methods. Classification of spectral and optical analytical methods. Instrumentation of spectral methods. Basic parts of instruments in spectral analysis: optical elements, radiation sources, monochromators, detectors (scheme, principle, basic characteristics, advantages and disadvantages). Molecular spectrometry. Nephelometry and turbidimetry. Luminescence analysis. Infrared spectroscopy. Raman spectroscopy. Refractometry. Chiroptical methods. Mass spectroscopy. Atomic spectral methods. Atomic absorption spectroscopy. Atomic emission spectral analysis. Atomic fluorescence spectrometry. Separation and preconcentration methods. Classification of separation methods. Chromatographic and non-chromatographic separation methods. Basic characteristics of separation methods. Non-chromatographic separation methods. Chromatographic methods of separation. Classification of chromatographic methods. Elution characteristics. Liquid chromatography. Gas chromatography. Supercritical fluid chromatography. Basic parts of instruments in chromatography. Electroanalytical methods. Basic principle of electroanalytical methods and their division. Potentiometry. Polarography. Voltammetry. Electrogravimetry. Coulometry. Conductometry.

Recommended literature:

- 1. Labuda a kol. Analytická chémia. ISBN: 9788022742429, Vydavateľstvo: STU Bratislava, Rok vydania: 2014, Počet strán: 671
- 2. Christian G.D. Analytical Chemistry. John Wiley & Sons, Inc. New York Chichester Brisbane Toronto Singapore 1994.
- 3. Holtzclaw H.F., Jr., Robinson W.R. College Chemistry with Qualitation Analysis. D.C. Heath and Company 1988.

Course language:

Slovak

Notes:

The course is implemented by full-time or, if necessary, distance method using the MS Teams or BBB or a combined method. The form of teaching is specified by the teacher at the beginning of the semester and updated continuously.

A calculator is required to master the calculation exercises. Not a cell phone!

Course assessment

Total number of assessed students: 1

A	В	С	D	Е	FX
0.0	0.0	100.0	0.0	0.0	0.0

Provides: prof. Mgr. Vasil' Andruch, DSc.

Date of last modification: 15.07.2022

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: KPE/ Course name: Integration and Inclusion in School Practice IIŠP/21 Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: present **Number of ECTS credits: 2 Recommended semester/trimester of the course:** 3. Course level: I. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 52 C Α В D Е FX 36.54 38.46 15.38 7.69 1.92 0.0 Provides: PaedDr. Michal Novocký, PhD.

Date of last modification: 12.03.2024

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University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ Course

Course name: Introduction to Environmental Chemistry

UECH/08

Course type, scope and the method:
Course type: Lecture / Practice

Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14

Course method: present

Number of ECTS credits: 5

Recommended semester/trimester of the course: 3., 4...

Course level: I.

Prerequisities:

Conditions for course completion:

Continuous test. Active participation in exercises - elaboration of semester work. Passing the final examination in the form of a written test.

Learning outcomes:

Introduction to topics in environmental chemistry and basic procedures applied for environmental protection. Discusses current and future environmental problems and possibilities how to solve them from chemichal point of view.

Brief outline of the course:

Introduction to Environmental Chemistry

Chemical aspects of pollution and environmental problems. Composition and behavior of the atmosphere. Energy balance of the Earth and climate changes. Principles of photochemistry, photoprocesses in the atmosphere. Petroleum, hydrocarbons and coal (characteristics, sources and environmental pollution). Soaps, polymers and synthetic surfactants. Haloorganics and pesticides. Environmental chemistry of some important elements (C, N, S, P, halogens, biologically important metals ...). Environmental chemistry in aqueous media. Aqueous systems, parameters, cycles and their protection. The Earth's crust (rocks, minerals, soils). Natural and artificial radioactivity, utilization. Energy and energy sources (fossil fuels, nuclear, geothermal, solar energy, wind and water energy). Solid waste disposal and recycling.

Recommended literature:

- 1. Gary W. van Loon, Stephen J. Duffy: Environmental Chemistry A Global Perspective, Oxford University Press, Oxford 2003
- 2. R.A. Bailey, H.M. Clark, J.P. Ferris, S. Krause, R.L. Strong: Chemistry of the Environment, Academic Press, San Diego 2002
- 3. G. Schwedt: The Essential Guide to Environmental Chemistry, Wiley and Sons, London 2001
- 4. R.N. Reeve, J.D. Barnes: General Environmental Chemistry, Wiley, London 1994
- 5. G. Burton, J. Holman, G. Pilling, D. Waddington: Chemical Storylines, Heinemann, Oxford, London 1994
- 6. www

Course language:

Notes:

Based on the current pandemic situation in Slovakia and in accordance with the conditions of the Faculty of Natural Sciences of UPJŠ in Košice, the education and examination can also be carried out in a distance form. The tutorial will be carried out in the form of online lectures and consultings in the BigBlueButton system. The written form of the exam takes place through the Google Forms app. Students prepare responses to the final written test. Test questions are randomly generated each time. The final oral exam is conducted through a webinar in BigBlueButton https://bbb.science.upjs.sk/b) system with online generation of random question numbers.

Course assessment

Total number of assessed students: 1

abs	n
100.0	0.0

Provides: doc. RNDr. Andrea Straková Fedorková, PhD.

Date of last modification: 18.07.2022

University: P. J. Šafárik University in Košice						
Faculty: Faculty of Science						
Course ID: Dek. PF UPJŠ/USPV/13	Course name: Introduction	n to Study of Sciences				
Course type, scope a Course type: Lectur Recommended cour Per week: Per stud Course method: pre	re / Practice rse-load (hours): ly period: 12s / 3d esent					
	euits: 2 ster/trimester of the cours					
	ster/trimester of the cours	e: 1.				
Course level: I.		-				
Prerequisities:						
Conditions for cours	e completion:					
Learning outcomes:						
Brief outline of the c	ourse:					
Recommended litera	nture:					
Course language:						
Notes:						
Course assessment Total number of asses	Course assessment Total number of assessed students: 2196					
	abs	n				
89.34 10.66						
Provides: doc. RNDr	Provides: doc. RNDr. Marián Kireš, PhD.					
Date of last modifica	ition: 30.08.2022					
Approved: doc. RND	Approved: doc. RNDr. Stanislav Lukáč, PhD., prof. RNDr. Vladimír Zeleňák, DrSc.					

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚMV/ Course name: Introduction to data analysis UAD/10Course type, scope and the method: Course type: Lecture / Practice **Recommended course-load (hours):** Per week: 1 / 1 Per study period: 14 / 14 Course method: present **Number of ECTS credits: 2 Recommended semester/trimester of the course:** 5. Course level: I. **Prerequisities: Conditions for course completion:** Test (40p) and individual project work (20p). Oral presentation of the individual project work (5p). At least 50% must be obtained from each part. Final evaluation: $\ge 90\%$ A; $\ge 80\%$ B; $\ge 70\%$ C; $\ge 60\%$ D; $\ge 50\%$ E; < 50% FX. **Learning outcomes:** To know the basic purpose of statistical data analysis, its methods and statistical thinking and understand its importance for science and practical life. To understand elementary statistical concepts. To gain experience in handling real data using spreadsheet Excel and statistical software R. **Brief outline of the course:** 1. Introduction (the basic philosophy and aim of statistical data analysis, descriptive and inductive statistics) 2. Collecting Data (types of data, random sample, randomized experiment) 3. Handling Data (visualization, summarizing – measures of center, measures of variability, skewness and kurtosis, empirical rule) - 5 weeks 4. Relationships in data (introduction to regression and correlation) - 4 weeks 5. Statistical inference (elementary view into estimation and testing hypothesis) - 2 weeks **Recommended literature:** 1. Anděl, J.: Statistické metody, Matfyzpress, Praha, 1998 (in Czech) 2. Rossman, A.J. et al.: Workshop Statistics: Discovery with Data and Fathom, 3rd ed. Wiley, 2009 3. Utts, J.M.: Seeing Through Statistics, 4th ed., Thomson Brooks/Cole, Belmont, 2014 4. Utts, J.M., Heckard R.F.: Mind on Statistics, 6th ed. Thomson Brooks/Cole, Belmont, 2021 5. Zvára, K., Štěpán, J.: Pravděpodobnost a matematická statistika, Matfyzpress, Praha, 2001 (in Czech) Course language: Slovak

Notes:

Course assessm	Course assessment							
Total number of assessed students: 434								
Α	В	С	D	Е	FX			
36.87	25.12	26.04	10.37	0.46	1.15			

Provides: doc. RNDr. Martina Hančová, PhD.

Date of last modification: 13.09.2021

University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚMV/ UDM/22	Course name: Introduction to mathematics
Course type, scope a Course type: Practic Recommended cour Per week: 4 Per stu Course method: pre	ce rse-load (hours): dy period: 56
Number of ECTS cr	edits: 3
Recommended seme	ster/trimester of the course: 1.
Course level: I.	
Prerequisities:	
Conditions for cours Two tests during the	•
•	natic sections of the secondary mathematics by interesting tasks. Explanation rties and proof methods used in various areas of mathematics.
and inequalities. Irra function; equations	ebraic expressions. Real number, absolute value of real numbers; equations tional equations and inequalities. Concept of function. Linear and quadratic and inequalities. Exponencial and logarithmic function; equations and etric functions; equations and inequalities. Complex numbers.
Bratislava, 1976 2. S. Richtárová - D. štúdium na vysokých 3. O. Hudec – Z. Kin štúdium na TU v Koš 4. F. Peller – V. Šáne uchádzačov o štúdium 5. F. Vesajda – F. Tala všeobecnovzdelávaci 6. J. Lukášová – O. C. 4. ročník gymnázia, S.	ík - T. Šalát: REPETITÓRIUM STREDOŠKOLSKEJ MATEMATIKY, Alfa Kyselová: MATEMATIKA (pomôcka pre maturantov a uchádzačov o školách), Enigma Nitra, 1998 náková – E. Švidroňová: PRÍKLADY Z MATEMATIKY (pre uchádzačov o šiciach), EF TU Košice, 1999 r – J. Eliáš – Ľ. Pinda: MATEMATIKA – Podklady na prijímacie testy pre m, Ekonóm Bratislava, 2000/2001 afous: ZBIERKA ÚLOH Z MATEMATIKY pre stredné e školy a gymnáziá, SPN Bratislava, 1973 Odvárko – B. Riečan – J. Šedivý – J. Vyšín: ÚLOHY Z MATEMATIKY pre
Course language: Slovak	

Notes:

Course assessm	Course assessment							
Total number of assessed students: 600								
A	В	С	D	Е	FX			
23.83	20.5	18.17	15.33	9.67	12.5			

Provides: RNDr. Veronika Hubeňáková, PhD., RNDr. Zuzana Gönciová

Date of last modification: 29.01.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚMV/ | Course name: Linear and integer programming

LCO/10

Course type, scope and the method: Course type: Lecture / Practice

Recommended course-load (hours): Per week: 2 / 2 Per study period: 28 / 28

Course method: present

Number of ECTS credits: 5

Recommended semester/trimester of the course: 5.

Course level: I.

Prerequisities: ÚMV/ALGa/10

Conditions for course completion:

Continuous evaluation: a small test during each tutorial, two large tests, a project with real data and commercial software. Bonus points awarded for homeworks (formulation of proofs). A necessary condition for final exam is at least 50% of points from th semester. Final exam: demonstrate the understanding of the theory and ability of argumentation.

Learning outcomes:

Ability to formulate practical tasks in a form of a linear program. Proficiency in solving linear programs by several methods, also using software. Understanding of the underlying theory and ability of exact argumentation.

Brief outline of the course:

Formulation of linear and integer programs. Geometric solution. Simplex method, its correctness an finiteness. Duality and its economic interpretation. Dual and revised simplex method. Sensitivity analysis and parametric programming. Algorithms for integer programming: branch and bound, Gomory cuts. Computational complexity of LP and ILP. Solution of practical problems.

Recommended literature:

lms.upjs.sk - podklady k prednáškam a zadania úloh na cvičenia.

Plesník, Dupačová, Vlach: Lineárne programovanie, Alfa, Bratislava 1990

Ch. Papadimitriou – K. Steiglitz: Combinatorial Optimization: Algorithms and Complexity, 1984 R.J. Vanderbei, Linear Programming:Foundations and Extentions, Springer 2020, electronic version: http://www.princeton.edu/~rvdb/LPbook/

Course language:

Slovak

Notes:

Course assessment

Total number of assessed students: 163

A	В	С	D	Е	FX
22.7	17.18	19.63	19.63	17.79	3.07

Provides: prof. RNDr. Katarína Cechlárová, DrSc., RNDr. Adam Marton

Date of last modification: 17.04.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚMV/ | Course name: Logic and set theory

LTM/10

Course type, scope and the method:

Course type: Lecture / Practice Recommended course-load (hours):

Per week: 3 / 2 Per study period: 42 / 28

Course method: present

Number of ECTS credits: 6

Recommended semester/trimester of the course: 5.

Course level: I., II.

Prerequisities: ÚMV/MANb/19 or ÚMV/FRPb/19 or ÚMV/MAN2b/22

Conditions for course completion:

Exam

Learning outcomes:

To obtain a basic knowledge on the mathematical notion of an infinity. Analysis of the notion of a proof.

Brief outline of the course:

Set as a mathematical formularization of an infinity. Properties of the set of reals. Relations and mappings.

Finite and countable sets. Cardinality of continuum. Elementary cardinal arithmetics.

Sentential calculus, an axiomatization. Completness Theorem. Methods of proofs. Language of predicate calculus, examples. Axiomatizations of predicate calculus and the notion of a proof. Methods of proofs in predicate calculus.

Recommended literature:

- L. Bukovský: Teória množín, ES UPJŠ, Košice, 1984.
- L. Bukovský: Množiny a všeličo okolo nich, ES UPJŠ, Košice, 2005.
- L. Bukovský, Úvod do matematickej logiky, elektronický učebný text.
- A. Sochor: Klasická matematická logika, Karolinum, Praha, 2001.
- E. Mendelson, Introduction to Mathematical Logic, van Nostrand 1964.

Course language:

Slovak

Notes:

Course assessment

Total number of assessed students: 276

A	В	С	D	Е	FX
13.04	18.84	19.2	16.3	30.8	1.81

Provides: RNDr. Jaroslav Šupina, PhD., RNDr. Adam Marton

Date of last modification: 19.04.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚMV/ | **Course name:** Macroeconomics

MAE/10

Course type, scope and the method:

Course type: Lecture / Practice

Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14

Course method: present

Number of ECTS credits: 4

Recommended semester/trimester of the course: 5.

Course level: I.

Prerequisities:

Conditions for course completion:

The final mark is given based on the results of the tests written during the semester ("small" exams every week, two written exams checking the ability of computations) and oral exam, that evaluates the verbal argument about the studied models.

Learning outcomes:

The student understands the basic economic models and is able to use them to explain the real economic phenomena.

Brief outline of the course:

Basic macroekonomic notions: Gross domestic product, inflation, unemployment.. Analysis of godds markets. Financial markets. IS-LM model in closed economy. Open economy. IS-LM model in open economy. Models of labour market. Inflation and economic growth. High depth.

Recommended literature:

- 1. Olivier Blanchard, Alessia Amighini, Francesco Giavazzi, Macroeconomics, a European perspective, Pearson Education, 2010
- 2. N. Gregory Mankiw, Macroeconomics, 7th Edition, Harvard University, Worth Publishers 2009

Course language:

Slovak

Notes:

Course assessment

Total number of assessed students: 86

A	В	С	D	Е	FX
25.58	13.95	20.93	19.77	13.95	5.81

Provides: prof. RNDr. Katarína Cechlárová, DrSc.

Date of last modification: 17.04.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚMV/ | **Course name:** Mathematical analysis III

MAN2c/22

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 2 Per study period: 28 / 28

Course method: present

Number of ECTS credits: 5

Recommended semester/trimester of the course: 3.

Course level: I.

Prerequisities: ÚMV/MAN2b/22

Conditions for course completion:

During the term, each student receives marks for two written exams each worth 25 points. Final marking is assigned based on the overall points for the work throughout the term followed by a written and oral examination where the student can obtain further 30+20 points.

Marking classification: A:91%-100%, B:81%-90%, C:71%-80%, D:61%-70%, E:51%-60%, FX:0%-50%

Learning outcomes:

Deepening the knowledge of real analysis of function with a single variable. The student will

- 1. familiarise themselves with mathematical culture, ways of thinking, self-expression and putting forward arguments,
- 2. gain a deeper understanding of the base terminology of real analysis, their properties and interconnections,
- 3. be able to define and interpret key terms, prove their basic properties and relationships,
- 4. know how to solve tasks focused on utilising the aforementioned concepts and interpret the obtained results.

Brief outline of the course:

Definite Riemann integral - definition, elementary properties, calculation methods, applications. Improper Riemann integral. Sequences and series of real functions – pointwise and uniform convergence, properties of the limit function and the sum. Power series, Taylor series and their applications.

Recommended literature:

- 1. Mihalíková, B. Ohriska, J.: Matematická analýza II (skriptum), UPJŠ Košice, 2007.
- 2. Hutník, O.: Určitý integrál (elektronický učebný text), UPJŠ, Košice, 2012.
- 3. Kluvánek, I. Mišík, L. Švec, M.: Matematika I, ALFA, Bratislava, 1971.
- 4. Demidovič, B. P.: Sbírka úloh a cvičení z matematické analýzy, Fragment, Praha, 2003.
- 5. Eliaš, J. Horváth, J. Kajan, J.: Zbierka úloh z vyššej matematiky 2, 3, 4, Alfa, Bratislava, 1971.
- 6. Brannan, D.: A First Course in Mathematical Analysis, Cambridge University Press, Cambridge 2006.

- 7. Bruckner, A. M. Bruckner J. B. Thomson, B. S.: Real Analysis, Second Edition, ClassicalRealAnalysis.com, 2008.
- 8. Zorich, V. A.: Mathematical Analysis I, Springer-Verlag 2002.

Course language:

Slovak

Notes:

Course assessment

Total number of assessed students: 252

A	В	С	D	Е	FX
11.11	15.08	12.7	20.24	34.52	6.35

Provides: prof. RNDr. Jozef Doboš, CSc., prof. RNDr. Ondrej Hutník, PhD.

Date of last modification: 25.04.2022

	COURSE INFORMATION LETTER
University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚMV/ MAN2d/22	Course name: Mathematical analysis IV
Course method: pre	re / Practice rse-load (hours): study period: 28 / 28 esent
Number of ECTS cr	
Recommended seme	ster/trimester of the course: 4., 6.
Course level: I.	
Prerequisities: ÚMV	/MAN2b/22
l .	e completion: nt is taken the form of two main tests during the semester. Final evaluation is assessment (60%), written and oral part of the exam (40%).
the course. He has de	nds the basic concepts and their properties, which are defined in the content of veloped skills to use this theory in solving theoretical and practical problems. do connections in solving problem tasks.
2. Differential calculdirectional derivative3. Multivariable Rien	l real variables - basic notions, limits and continuity. (3 weeks) us of functions of several real variables - partial derivative, differentiability, , local and global extrema, constrained local extrema. (5 weeks) nann integral - definition, calculation methods, applications. (2 weeks) uclidean space, topological properties of points and sets in metric space,
2. L. Kluvánek, I. Mi 3. P. Vodstrčil, J. Bou 4. Z. Došlá, O. Došlý 5. J. Eliaš, J. Horváth 6. D. Hughes-Hallett 7. B. S. Thomson, J. (Pearson), Lexington Course language:	ferenciální počet funkcí více proměnných, Brno a Ostrava, 2012. šík, M. Švec: Matematika I, II, SVTL, Bratislava, 1959. ichala: Integrální počet funkcí více proměnných, Ostrava a Plzeň, 2012. "Metrické prostory, Teorie a príklady. 3.vydání, 2006. "J. Kajan: Zbierka úloh z vyššej matematiky 3, 4, SVTL, Bratislava, 1966. et al.: Calculus, Wiley, 1998. B. Bruckner, A. M. Bruckner: Elementary real analysis, Prentice Hall
Slovak	

Notes:

Course assessment						
Total number of assessed students: 65						
Α	В	С	D	Е	FX	
27.69	20.0	24.62	12.31	13.85	1.54	

Provides: RNDr. Lenka Halčinová, PhD.

Date of last modification: 17.04.2022

	COURSE INFORMATION LETTER					
University: P. J. Šafá	rik University in Košice					
Faculty: Faculty of Science						
Course ID: ÚMV/ MAN2b/22	Course name: Mathematical analysis of function of real variable					
Course type, scope a Course type: Lectur Recommended cour Per week: 4/3 Per Course method: pre	re / Practice rse-load (hours): study period: 56 / 42					
Number of ECTS cr	edits: 7					
Recommended seme	ster/trimester of the course: 2.					
Course level: I.						
Prerequisities: ÚMV	/FRPa/19					
	se completion: uring semeter and activity student to practice. Final evaluation is given by nt, written and oral part of the exam.					
1 1	ourse is to strengthen the knowledge in differential and integral calculus of real variable and to develop computational skills in the field.					
1	of real functions, elementary functions. Differential calculus - derivatives of orders, the basic theorems of differential calculus and their use to investigate					
2012. 2. Mihalíková, B O 3. Kluvánek, I Miši 4. Demidovič, B. P.: S 5. Brannan, D.: A Fir Cambridge 2006. 6. Bruckner, A. M., E ClassicalRealAnalysi	Ohriska, J.: Matematická analýza I (elektronický učebný text), UPJŠ Košice, Ohriska, J.: Matematická analýza II (skriptum), ES UPJŠ Košice, 2007. ík, L Švec, M.: Matematika I, ALFA, Bratislava, 1971. Sbírka úloh a cvičení z matematické analýzy, Fragment, Praha, 2003. est Course in Mathematical Analysis, Cambridge University Press, Bruckner J. B., Thomson, B. S.: Real Analysis, Second Edition,					
Course language: Slovak						

Notes:

Course assessment						
Total number of	Total number of assessed students: 109					
A	В	С	D	Е	FX	
13.76	16.51	20.18	19.27	22.02	8.26	

Provides: prof. RNDr. Ondrej Hutník, PhD., RNDr. Lenka Halčinová, PhD., RNDr. Jana Borzová, PhD.

Date of last modification: 17.04.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚMV/ | **Course name:** Mathematical modeling

MMD/22

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 3 Per study period: 42

Course method: present

Number of ECTS credits: 3

Recommended semester/trimester of the course: 5.

Course level: I.

Prerequisities:

Conditions for course completion:

Submitting a project from the specified list of projects and, possibly, a related short presentation.

Learning outcomes:

Using concrete examples of problems from real life, students will become familiar with several approaches and strategies for creating a mathematical model of specified problem as well as with defining the conditions related a real problem and transforming them into created mathematical model.

Brief outline of the course:

One specified real-life problem will be discussed, explored and modeled each week.

Recommended literature:

- 1. E. Lindner, A. Micheletti, C. Nunes (eds.), Mathematical Modelling in Real Life Problems, Springer, 2020.
- 2. K.K. Tung, Topics in Mathematical Modeling, Princeton University Press, 2007.
- 3. H. P. Williams, Model Building in Mathematical Programming, Wiley, 2013.

Course language:

Slovak

Notes:

Course assessment

Total number of assessed students: 28

A	В	С	D	Е	FX
89.29	10.71	0.0	0.0	0.0	0.0

Provides: RNDr. Jana Borzová, PhD., prof. RNDr. Katarína Cechlárová, DrSc., RNDr. Igor Fabrici, Dr. rer. nat., univerzitný docent, RNDr. Andrej Gajdoš, PhD., RNDr. Lenka Halčinová, PhD., RNDr. Jaroslav Šupina, PhD., doc. RNDr. Martina Hančová, PhD., Mgr. Martin Vodička, prof. RNDr. Ondrej Hutník, PhD., prof. RNDr. Ivan Žežula, CSc., RNDr. Lucia Janičková, PhD., doc. Mgr. Jozef Kiseľák, PhD., doc. RNDr. Daniel Klein, PhD., prof. RNDr. Tomáš Madaras, PhD.

Date of last modification: 25.08.2022

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚMV/ Course name: Mathematical problem solving strategies I MRUa/22 Course type, scope and the method: Course type: Practice **Recommended course-load (hours):** Per week: 2 Per study period: 28 Course method: present **Number of ECTS credits: 2** Recommended semester/trimester of the course: 4. Course level: I. **Prerequisities: Conditions for course completion:** Assessment is given on the basis of the results of written examinations carried out during the semester and active participation in exercises. Classification scale: A: 91 % - 100 %, B: 81 % - 90 %, C: 71 % - 80 %, D: 61 % - 70 %, E: 51 % - 60 %, FX: 0 % - 50 %. **Learning outcomes:** The student is able to explain the basic concepts and methods of solving mathematical problems selected from various areas of school mathematics. The student is able to apply the acquired knowledge in finding and using various strategies for solving problems. The student will get acquainted with typical and more demanding tasks in school mathematics and with specific problems and misconceptions that occur in their solution in the teaching of mathematics in primary and secondary school. **Brief outline of the course:** 1. - 7. Solving equations, inequalities and systems of equations (equations and inequalities with absolute values, equations with parameters, irrational equations and inequalities, exponential and logarithmic equations and inequalities, trigonometric equations and inequalities). 8. - 13. Concept of function, properties of elementary functions, graphs of functions. Recommended literature: Kubáček, Z., Černek, P., Žabka J. a kol.: Matematika a svet okolo nás, zbierka úloh. FMFI UK Bratislava, 2008 Kopka, J., Hrozny problémů ve školské matematice, Univerzita J. E. Purkyně, Ústí nad Labem, 1999. Učebnice a zbierky úloh z matematiky ZŠ a SŠ. Course language:

Page: 107

Slovak

Notes:

Course assessment						
Total number of assessed students: 236						
Α	В	С	D	Е	FX	
29.24	21.61	23.31	11.86	12.71	1.27	

Provides: prof. RNDr. Jozef Doboš, CSc.

Date of last modification: 25.04.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚMV/ | Course name: Mathematical problem solving strategies II

MRUb/15

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of ECTS credits: 2

Recommended semester/trimester of the course: 5.

Course level: I.

Prerequisities:

Conditions for course completion:

The resulting trial is granted on the basis of continuous assessment (on the results of written checks) and seminar work.

Learning outcomes:

Mastering the basic types of tasks and their methods of solving problems in primary and secondary school in the field of Planimetry, Stereometry and Goniometry.

Brief outline of the course:

Basic knowledge of school mathematics, various methods for the task, the role of mathematical competitions for thematic units Planimetry (4 w.), stereometry (3), goniometery (3).

Recommended literature:

- [1] Hejný, M. a kol., Teória vyučovania matematiky 2. SPN, Bratislava 1989 (in Slovak)
- [2] Kopka, J., Hrozny problémů ve školské matematice, Univerzita J. E. Purkyně, Ústí nad Labem 1999 (in Czech)
- [3] Jonson-Wilder.S., Mason.J.: Developing thinking in Geometry, Sage, 2009
- [4] Učebnice a zbierky úloh z matematiky ZŠ a SŠ

Course language:

Slovak

Notes:

Course assessment

Total number of assessed students: 188

A	В	С	D	Е	FX
31.91	30.32	25.0	8.51	4.26	0.0

Provides: doc. RNDr. Dušan Šveda, CSc.

Date of last modification: 19.09.2021

Approved: doc. RNDr. Stanislav Lukáč, PhD., prof. RNDr. Vladimír Zeleňák, DrSc.

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University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚMV/ | Course name: Mathematical problem solving strategies II

MRUb/22

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of ECTS credits: 2

Recommended semester/trimester of the course: 6.

Course level: I.

Prerequisities:

Conditions for course completion:

Conditions for continuous evaluation:

- 1. Participation in teaching in accordance with the study rules and instructions of the teacher.
- 2. Activity.
- 3. Homework and written test.
- 4. Conditions for successful completion of the course:
- 1. Participation in teaching in accordance with the study regulations and according to the instructions of the teacher;
- 2. Credits will be awarded to a student who scores at least 50% on homework assignments and at least 50% on written test. A grade of A requires at least 90%, a grade of B requires at least 80%, a grade of C requires at least 70%, a grade of D requires at least 60%, and a grade of E requires at least 50%.

Learning outcomes:

Students demonstrate a shift in different methods of problem-solving from combinatorics, probability and statistics. They will be aware of the connections between different methods of solution, and also the connections of these methods of solution with other topics of school mathematics.

While solving problems on written tests, the students will show that they have a conceptual understanding of the concepts of school combinatorics, probability and statistics. They are ready to use several methods of solving problems from these topics, they are able to consider whether a non-standard student's solution is correct or not, and they can explain this solution.

Brief outline of the course:

The content is focuses on different methods of problem-solving in combinatorics, probability and statistics. We are dealing with developing combinatorial, probabilistic and statistical thinking through different methods of problem-solving. The content of the course is based on current research results in this area. In solving combinatorial problems, students are introduced to the components of the model of combinatorial thinking - the listing of possibilities, the counting process, and combinatorial formulas and methods, and the connections between these components. When solving probability problems, we emphasize the different approaches to probability -

statistical, classical, geometric, and subjective and their connections. In part aimed at statistics, we focus on descriptive statistics and on the connection between probability and statistics.

Recommended literature:

Hecht, T., Sklenáriková, Z., Metódy riešenia matematických úloh, Bratislava, SPN, 1992. (in slovak)

Krantz, S.G., Techniques of Problem Solving, AMS, 1997.

Larson, L.C., Metódy riešenia matematických problémov, Bratislava, Alfa, 1990. (in slovak) Textbooks for secondary and middle schools.

Course language:

Slovak

Notes:

Course assessment

Total number of assessed students: 136

A	В	С	D	Е	FX
36.03	16.91	25.0	11.03	9.56	1.47

Provides: doc. RNDr. Ingrid Semanišinová, PhD.

Date of last modification: 17.04.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚMV/ | Course name: Mathematical problem solving strategies III

MRUc/15

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of ECTS credits: 2

Recommended semester/trimester of the course: 6.

Course level: I.

Prerequisities: ÚMV/MRUb/15

Conditions for course completion:

Conditions for continuous evaluation:

- 1. Participation in teaching in accordance with the study rules and instructions of the teacher.
- 2. Activity.
- 3. Homework and written test.
- 4. Conditions for successful completion of the course:
- 1. Participation in teaching in accordance with the study regulations and according to the instructions of the teacher;
- 2. Credits will be awarded to a student who scores at least 50% on homework assignments and at least 50% on written test. A grade of A requires at least 90%, a grade of B requires at least 80%, a grade of C requires at least 70%, a grade of D requires at least 60%, and a grade of E requires at least 50%.

Learning outcomes:

Students demonstrate a shift in different methods of problem-solving from combinatorics, probability and statistics. They will be aware of the connections between different methods of solution, and also the connections of these methods of solution with other topics of school mathematics.

While solving problems on written tests, the students will show that they have a conceptual understanding of the concepts of school combinatorics, probability and statistics. They are ready to use several methods of solving problems from these topics, they are able to consider whether a non-standard student's solution is correct or not, and they can explain this solution.

Brief outline of the course:

The content is focuses on different methods of problem-solving in combinatorics, probability and statistics. We are dealing with developing combinatorial, probabilistic and statistical thinking through different methods of problem-solving. The content of the course is based on current research results in this area.

In solving combinatorial problems, students are introduced to the components of the model of combinatorial thinking - the listing of possibilities, the counting process, and combinatorial formulas and methods, and the connections between these components.

When solving probability problems, we emphasize the different approaches to probability - statistical, classical, geometric, and subjective and their connections.

In part aimed at statistics, we focus on descriptive statistics and on the connection between probability and statistics.

Recommended literature:

Hecht, T., Sklenáriková, Z., Metódy riešenia matematických úloh, Bratislava, SPN, 1992. (in slovak)

Krantz, S.G., Techniques of Problem Solving, AMS, 1997.

Larson, L.C., Metódy riešenia matematických problémov, Bratislava, Alfa, 1990. (in slovak) Učebnice a zbierky úloh pre stredné a základné školy.

Course language:

Slovak

Notes:

Course assessment

Total number of assessed students: 197

A	В	С	D	Е	FX
30.46	26.9	23.86	11.17	6.6	1.02

Provides: doc. RNDr. Ingrid Semanišinová, PhD.

Date of last modification: 07.02.2022

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚMV/ **Course name:** Mathematical statistics MST/19 Course type, scope and the method: Course type: Lecture / Practice **Recommended course-load (hours):** Per week: 2 / 2 Per study period: 28 / 28 Course method: present Number of ECTS credits: 5 **Recommended semester/trimester of the course:** 5. Course level: I., II. **Prerequisities: Conditions for course completion:** Total evaluation based on two written tests during the semester (2x40p) and the result of the written (30p) and oral part of the exam (30p). At least 50% must be obtained from each part. Final evaluation: ≥90% A; ≥80% B; ≥70% C; ≥60% D; ≥50% E; <50% FX. **Learning outcomes:** Student should obtain the knowledge about basic statistical methods and the ability to apply theoretical knowledge in practical problems solving. **Brief outline of the course:** 1. Random vectors (definition, distributions, characteristics, joint and marginal distributions). 2. Covariance, correlation and regression. 3. Random sample, sampling distributions and characteristics. 4. Some important statistics and their distributions. 5. Point estimators and their properties. 6. Maximum likelihood method. 7. Interval estimates, confidence interval construction (2 weeks). 8. Testing of statistical hypothesis (critical region, level of significance and power of test, methods for searching optimal critical regions). 9. Some important parametric tests (2 weeks). 10. Some important nonparametric tests (2 weeks). **Recommended literature:** 1. Skřivánková V.: Pravdepodobnosť v príkladoch, UPJŠ, Košice, 2006 (in Slovak) 2. Skřivánková V.-Hančová M.: Štatistika v príkladoch, UPJŠ, Košice, 2005 (in Slovak) 3. Casella, G., Berger, R., Statistical Inference, 2nd ed., Duxbury Press, 2002 4. DeGroot, M. H., Schervish, M. J.: Probability and Statistics, 4th ed., Pearson, Boston, 2012 5. Anděl J.: Základy matematické statistiky, MatfyzPress, Praha, 2011 (in Czech) Course language: Slovak

Notes:

Course assessn	Course assessment						
Total number of assessed students: 174							
A	В	Е	FX				
25.29	12.07	7.47					

Provides: doc. RNDr. Martina Hančová, PhD.

Date of last modification: 14.04.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚMV/ | **Course name:** Mathematics

MTM/22

Course type, scope and the method:

Course type:

Recommended course-load (hours):

Per week: Per study period: Course method: present

Number of ECTS credits: 2

Recommended semester/trimester of the course:

Course level: I.

Prerequisities: ÚMV/MAN2c/22 and ÚMV/ATC/22

Conditions for course completion:

Acquiring the required number of credits in the structure defined by the study plan.

Learning outcomes:

Evaluation of student's competences with respect to the profile of the graduate.

Brief outline of the course:

Recommended literature:

Course language:

Slovak

Notes:

Course assessment

Total number of assessed students: 102

A	В	C	D	Е	FX
16.67	22.55	25.49	23.53	10.78	0.98

Provides:

Date of last modification: 26.01.2022

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚMV/ **Course name:** Mathematics I for chemists MTCa/22 Course type, scope and the method: Course type: Lecture / Practice **Recommended course-load (hours):** Per week: 2 / 2 Per study period: 28 / 28 Course method: present Number of ECTS credits: 5 Recommended semester/trimester of the course: 1., 3. Course level: I. **Prerequisities: Conditions for course completion:** To complete the course, it is necessary to demonstrate the acquirement of basic mathematical terms and the ability to solve problems from selected thematic units. The evaluation of the subject is according to the results from the semester and in view of the results of the written final test. During the semester, students write tests at all seminars (together 20 points) and two extensive tests (together 50 points). It is necessary to obtain at least 28 points during the semester. Then students may write the exam. To pass the exam, it is necessary to obtain at least 12 points from the maximum number of 30 points. The scale for student evaluation is as follows: 100-80-A, 79-70-B, 69-60-C, 59-50-D, 49-40-E. If a student does not achieve the required mninimal number of points from the exam test (12 points) and during the semester (together 28 points), he/ she is evaluated by FX. Learning outcomes: After completing the course, the student can use basic mathematical terms, can solve various equations and inequations, and is acquainted with basic mathematical knowledge from the differential and integral calculus, and is able to apply the theory in concrete excercises. **Brief outline of the course:** Week 1-6: Definition of function. Domain and range of functions. Elementary functions. Inverse functions. Compositions of functions. Week 7-14: Limit of functions. Continuity of functions. Derivation and its geometric aplications. Indefinite integrals, basic methods of integration. Definite integral and its applications. Recommended literature: Hut'ka, Benko, Ďurikovič: Matematika, Alfa, Bratislava 1991 D. Studenovská, T. Madaras, S. Mockovčiak: Zbierka úloh z matematiky pre nematematické odbory, UPJŠ 2006 D. Studenovská, T. Madaras: Matematika pre nematematické odbory, UPJŠ 2006 S. Lang: A First Course in Calculus, Springer Verlag, 1998 Course language: Slovak

Notes:

Course assessm	Course assessment						
Total number of assessed students: 640							
Α	В	C	D	Е	FX		
11.41	10.94	16.41	21.09	28.28	11.88		

Provides: RNDr. Jana Borzová, PhD., RNDr. Miriam Kleinová, Mgr. Miriama Kmeciková, RNDr. Monika Krišáková

Date of last modification: 18.04.2022

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: KPE/ Course name: Mentoring and Coaching in School Practice MKŠP/21 Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: present **Number of ECTS credits: 2 Recommended semester/trimester of the course:** 5. Course level: I. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 62 \mathbf{C} Α В D Е FX 83.87 12.9 3.23 0.0 0.0 0.0 Provides: Mgr. Katarína Petríková, PhD. Date of last modification: 12.03.2024

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚMV/ | **Course name:** Microeconomics

MIE/13

Course type, scope and the method:
Course type: Lecture / Practice

Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14

Course method: present

Number of ECTS credits: 4

Recommended semester/trimester of the course: 5.

Course level: I.

Prerequisities:

Conditions for course completion:

Continuous assessment: feedback in MOODLE, small tests during tutorial (notions), two written exams (solving problems). Final oral exam: ability of verbal argumentation and graphical explanation of studied models.

Learning outcomes:

Understanding of basic principles of microeconomics and ability to apply them in practical situations.

Brief outline of the course:

Economics and economy. Supply and demand. Consumer Theory. Theory of firm. Perfect competition. Monopoly. Labour market. Market failure. Externalities and Public goods.

Recommended literature:

- 1. lms.upjs.sk: lectures, tutorials and other material
- 2. H.L. Varian, Intermediate Mikroekonomics, WW Norton, 1993
- 3. J.M. Perloff, Microeconomics, 6th Edtion, Addison Wesley, 2012
- 4. J. Sloman, Economics, 6th Edition, Prentice Hall, 2006

Course language:

Slovak

Notes:

Course assessment

Total number of assessed students: 90

A	В	С	D	E	FX
24.44	22.22	18.89	18.89	13.33	2.22

Provides: prof. RNDr. Katarína Cechlárová, DrSc.

Date of last modification: 17.04.2022

Approved: doc. RNDr. Stanislav Lukáč, PhD., prof. RNDr. Vladimír Zeleňák, DrSc.

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University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: KPE/ **Course name:** Multiculturalism and Multicultural Education MMKV/17 Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: present **Number of ECTS credits: 2** Recommended semester/trimester of the course: 4. Course level: I. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 202 C Α В D Е FX 41.09 44.06 13.37 0.99 0.5 0.0 Provides: PaedDr. Michal Novocký, PhD. Date of last modification: 12.03.2024

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚMV/ | Course name: Numerical methods

NUM/19

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 3 Per study period: 28 / 42

Course method: present

Number of ECTS credits: 6

Recommended semester/trimester of the course: 6.

Course level: I.

Prerequisities: (ÚMV/MANb/19 or ÚMV/MAN2b/22 or ÚMV/FRPb/19) and (ÚMV/ALG1b/10 or ÚMV/ALG2b/22 or ÚMV/ALG3b/22 or ÚMV/ALG4b/22)

Conditions for course completion:

Form: Lectures and practices using computers. Solving problems and programming algorithms using the computational platform SageMath (including Python, NumPy, SciPy, SymPy, R, Maxima, matplotlib, GAP, FLINT, and many other packages).

Interim assessment (50% of the total assessment): Solving assigned tasks e.g. in the form of implementation of algorithms or their parts, modification of existing codes or use of available packages in solving real problems.

Final examination (50% of the total assessment): It consists of verifying the understanding of the theory taken over and demonstrating the practical skills acquired.

Learning outcomes:

After completing the course, the student will acquire theoretical knowledge and practical skills regarding the principles and implementation of basic numerical algorithms with emphasis on algorithms used in the field of data analysis.

The student should be able to understand and implement numerical algorithms in programming language independently, to be able to modify components of existing algorithms and also be able to solve (real) problems by selecting an appropriate numerical method with the

available effective computational packages.

Brief outline of the course:

- 1. Basic principles and techniques of numerical analysis computer implementation and representation of real numbers, numerical vs. symbolic (analytical) calculations, method vs. algorithm, error measurement of numerical solution, conditionality of numerical problems, stability and convergence of numerical algorithms.
- 2. Solution of nonlinear equations methods of bisection and simple iteration, the false position method and Newton method, Newton-Raphson method.
- 3. Numerical differentiation and integration trapezoidal method, Simpson method, Newton-Cotes formulas.
- 4. Approximation of functions and smoothing of data, using polynomials, interpolation, splines, kernel methods.

- 5. Linear systems Gaussian elimination with and without pivoting, forward and backward substitution, scaled partial pivoting, singularity and perturbation, matrix conditionality, Thomas method, iterative methods Jacobi, Gauss-Seidel, SOR method, gradient methods gradient descent, conjugate directions.
- 6. Eigenvalues and eigenvectors of matrices estimation of eigenvalues, partial eigenvalue problem (power method and Rayleigh method, Hessenberg shape), complete eigenvalue problem (calculation of dominant eigenvalue, LU, QU, QR decomposition, Jacobi method), SVD Singular Matrix Decomposition.
- 7. Optimization MLS, Cauchy method of the highest gradient, Newton method, conjugated gradient method of Fletcher-Reeves, Quasi-Newton methods, Regularization of ill-conditioned problems.

Recommended literature:

- 1. Ackleh, A. S., Allen, E. J., Kearfott, R. B., & Seshaiyer, P. (2009). Classical and Modern Numerical Analysis: Theory, Methods and Practice (1 edition). Boca Raton: Chapman and Hall/CRC.
- 2. Anastassiou, G. A., & Mezei, R. (2015). Numerical Analysis Using Sage. Springer International Publishing.
- 3. Cheney, E. W., & Kincaid, D. R. (2012). Numerical Mathematics and Computing (7 edition). Boston, MA: Cengage Learning.
- 4. O'Leary, D. P. (2008). Scientific Computing with Case Studies. Philadelphia: Society for Industrial and Applied Mathematics.
- 5. Sauer, T. (2017). Numerical Analysis. (3 edition). Hoboken, NJ? Pearson.
- 6. Segethová, J. (2002). Základy numerické matematiky. Karolinum.
- 7. M. Vicher (2003). Numerická matematika.

Course language:

Slovak

Notes:

Course assessment

Total number of assessed students: 138

A	В	С	D	Е	FX
13.77	16.67	7.25	14.49	35.51	12.32

Provides: RNDr. Andrej Gajdoš, PhD.

Date of last modification: 18.04.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ | Course name: Organic chemistry

OCHU/21

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 3 / 1 Per study period: 42 / 14

Course method: present

Number of ECTS credits: 5

Recommended semester/trimester of the course: 2.

Course level: I.

Prerequisities: ÚCHV/VCHU/15 or ÚCHV/VCHU/14 or ÚCHV/VCHU/10 or ÚCHV/VACH/10

Conditions for course completion:

Written test. Two tests, in 7th and 14th week. Test max 50 points. A student must obtain at least 51% of points. Writing of the tests is mandatory.

Written exam, 100 points. 69 Theoretical questions (69 points), 62 chemical formulas (31 points). A student must obtain at least 51% of points. Final evaluation: A 91-100 pts, B 81-90 pts, C 71-80 pts, D 61-70 pts, E 51-60 pts, FX 0-50 pts.

Learning outcomes:

Basic organic chemistry course.

Nomenclature of organic compounds, their chemical properties, structure, reactivity and characteristic reactions. Preparation of organic molecules, explanation of the basic mechanisms and principles of organic reactions.

After completing the subject, the student understands the studied theories, principles, methods and logical procedures of organic chemistry. He has knowledge of modern organic chemistry with an emphasis on the current development of knowledge in the aforementioned area.

Brief outline of the course:

Chemical bonding Hybridization and Bonding Covalent bonds Double bonds and Triple Bonds Structural Formulas of Organic Molecules Polar Covalent Bonds and Electronegativity Constitutional Isomers Alkenes Electrophilic Additions Strong Brønsted Acids Lewis Acids (non-Proton Electrophiles) Electrophilic Halogen Reagents Other Electrophilic Reagents Reduction Oxidation Radical Additions Allylic Substitution Alkynes Addition Reactions Hydrogenation Electrophiles Hydration & Tautomerism Hydroboration Nucleophilie Addition & Reduction Acidity of Terminal Alkynes (Substitution of H) Alkyl Halides General Reactivity Substitution(of X) SN2 Mechanism SN1 Mechanism Elimination (of HX) Summary of Substitution vs. Elimination Substitution by Metals Elimination Reactions of Dihalides Alcohols Reactions of Alcohols Substitution of the Hydroxyl H Substitution of the Hydroxyl Group Elimination of Water Oxidation of Alcohols Reactions of Phenols Acidity of Phenols Ring Substitution Mechanism Reactions of Substituted Benzenes Reaction Characteristics Reactions of Disubstituted Rings Reactions of Substituent Groups Nucleophilic Substitution, Elimination & Addition Reactions Amines Basicity of Nitrogen Compounds Acidity of Nitrogen Compounds Important Reagent Bases Reactions of

Amines Electrophilic Substitution at Nitrogen Preparation of 1°-Amines Preparation of 2° & 3°-Amines Reactions with Nitrous Acid Reactions of Aryl Diazonium Intermediates Elimination Reactions of Amines Oxidation States of Nitrogen Basic information: Aldehydes & Ketones Carboxylic Acids Derivatives of Carboxylic acids Natural products

Recommended literature:

- 1. Organic chemistry, J. Clayden, N. Greeves Warren, S. Wothers, Oxford University Press, 2012, ISBN 978-0-19-92-7029-3.
- 2. Organic chemistry, J. E. McMurry, Brooks/Cole, a Thomson Learning Company 2004, Sixth Eddition, ISBN 0534389996.
- 3. Organic chemistry, P. Zahradník, M. Mečiarová, P. Magdolen, Univerzita Komenského v Bratislave, 2019, ISBN: 978-80-223-4589-7.

Course language:

anglický

Notes:

Teaching is carried out in person or, if necessary, online using the MS Teams tool. The form of teaching is specified by the teacher at the beginning of the semester, updated continuously.

Course assessment

Total number of assessed students: 53

A	В	С	D	Е	FX
11.32	7.55	18.87	49.06	13.21	0.0

Provides: RNDr. Slávka Hamul'aková, PhD., univerzitná docentka, doc. RNDr. Miroslava Martinková, PhD., univerzitná profesorka, doc. RNDr. Mária Vilková, PhD.

Date of last modification: 04.08.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/

Course name: Organic chemistry - Lab.

POCHU/15

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 4 Per study period: 56

Course method: present

Number of ECTS credits: 4

Recommended semester/trimester of the course: 3.

Course level: I.

Prerequisities: ÚCHV/OCHU/03 or ÚCHV/OCHU/21

Conditions for course completion:

100% participations in practical exercises.

Two written tests 2 x 25 pts (a minimum of 13 points must be obtained in each test), twelve reports 12 x 2 pts, laboratory skills 12 pts, short quizzes and questions 14 pts.

A 100 pts. in total.

Assessment A: 91-100; B: 81-90; C: 71-80; D: 60-71; E: 51-60; FX: 0-50 pts.

Learning outcomes:

Students will become familiar with the basic isolation and purification methods used in a synthetic laboratory. Students should master basic laboratory technique and be able to apply the theoretical knowledge from the basic course of organic chemistry in simple synthetic projects.

Brief outline of the course:

Preparation, isolation, purification and identification of organic compounds. The emphasis is on gaining the experimental skills in synthesis of organic compounds, distillation, extraction, crystallization, sublimation and thin-layer chromatography.

- 1. Isolation and purification methods crystallization
- 2. Isolation and purification methods distillation
- 3. Preparation of ethyl acetate
- 4. Preparation of acetylsalicylic acid
- 5. Preparation of benzalaniline
- 6. Spectral methods in organic chemistry
- 7. Preparation of acetophenone oxime
- 8. Preparation of benzilic acid
- 9. Preparation of 4,5-diphenylimidazole
- 10. Isolation of caffeine from tea
- 11. Isolation of trimyristin from nutmeg

Recommended literature:

- 1. Handout with experimental procedures http://kekule.science.upjs.sk/pochu.
- 2. Organic chemistry lectures.

Course language:

Page: 126

Slovak **Notes: Course assessment** Total number of assessed students: 250 В \mathbf{C} D E FX Α 54.8 27.6 6.0 0.8 10.8 0.0

Provides: RNDr. Slávka Hamuľaková, PhD., univerzitná docentka, RNDr. Ján Elečko, PhD., RNDr. Jana Špaková Raschmanová, PhD., RNDr. Mariana Budovská, PhD., univerzitná docentka

Date of last modification: 28.01.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ Course name: Organická chémia II

OCH1b/21

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 2 Per study period: 28 / 28

Course method: present

Number of ECTS credits: 5

Recommended semester/trimester of the course: 3., 5.

Course level: I.

Prerequisities:

Conditions for course completion:

Written test. Two tests, in 7th and 14th week. Test max 50 points. A student must obtain at least 51% of points from each test. Writing of the tests is mandatory.

Written exam, 100 points. 69 Theoretical questions (69 points), 62 chemical formulas (31 points). A student must obtain at least 51% of points. Final evaluation: A 91-100 pts, B 81-90 pts, C 71-80 pts, D 61-70 pts, E 51-60 pts, FX 0-50 pts.

Learning outcomes:

The advanced organic chemistry. The structure, reactivity and synthesis of organic compounds with careful explanations of difficult concepts and reaction mechanisms.

After completing the subject, the student has deeper knowledge of organic chemistry, knows how to connect the properties of organic compounds with their structure and reactivity. He can explain the principles of the mechanisms of organic reactions and propose syntheses of various groups of organic compounds (also multi-stage).

The student understands the studied theories, principles, methods and logical procedures of advanced organic chemistry. He has knowledge of modern trends in the field of organic chemistry with an emphasis on the current development of knowledge in the aforementioned field.

Brief outline of the course:

Ethers - their nomenclature, preparation and reactions. Sigmatropic rearrangements, their selectivity. Preparation and reactions of epoxides.

Nitrogen compounds, Amines, their nomenclature, basicity and nucleophilicity, preparation amines, their reactions. Diazonium salts, their preparation and reactions. Nitro compounds, their preparation and reactions. Nitroso compounds, oximes, hydrazones, nitroaldol reaction.

Carbonyl compounds - aldehydes and ketones, their nomenclature and reactivity. Nucleophilic additions, addition of the primary and secondary amines and related nitrogen reagents, the aldol reaction, self-condensations, cross-condensations and related reactions. Claisen condensation and its variants. Alkylation of enolates and their applications. Benzilic acid rearrangement, Benzoin condensation, Cannizzaro reaction, Mannich reaction, Reformatsky reaction, Perkin synthesis, Knoevenagel condensation, Julia olefination, Julia-Kocienski and Petersen olefination, Wittig reaction, HWE olefination, Baylis-Hillman reaction, Darzens reaction, Baeyer-Villiger oxidation, conjugate addition, Michael addition (Michael's donors and acceptors), Robinson annulation.

Carboxylic acids, their nomenclature, properties and preparation. Reactions of carboxylic acids, Esterification. Carboxylic acid derivatives (acyl halides, anhydrides, esters, amides, – their nomenclature, properties, preparation and reactions). β -Oxoesters – their preparation and reactions. Acyloin condensation, Arndt-Eistert synthesis, Hofmann degradation, Lossen degradation, Curtius rearrangement, Wolff rearrangement.

Amino acids – their stereochemistry, properties, preparation and reactions, peptide bond - its structure, synthesis of peptides, the protective groups for amino acids.

Saccharides - classification, their nomenclature and stereochemistry. Fischer and Haworth projection, conformation of saccharides, reaction of saccharides (oxidation, reduction, production of the glycosidic bond). The protective groups. Oligosaccharides, polysaccharides.

Nucleotides and nucleic acids (structure of nucleoside, saccharides in NA, purine and pyrimidine bases in NA). Examples of nucleotides in RNA and DNA.

Heterocyclic compounds. Five and six membered heterocyclic compounds.

Terpenes, stereoids and alkaloids - their classification and properties.

Recommended literature:

Recommended literature:

- 1. J. Clayden, N. Greeves, S. Warren, P. Wothers: Organic Chemistry, Oxford University Press, 2012.
- 2. Solomons T.W. Graham: Solomon's Organic Chemistry, Willey&Sons Inc., 2017.
- 3. J. E. McMurry: Organic Chemistry, Vysoké učení technické v Brne, 2007, VUTIUM, ISBN: 978-80-214-3291-8 (VUT v Brne).
- 4. J. E. McMurry: Organic Chemistry, Cengage, 2015.

Course language:

english

Notes:

Teaching is carried out in person or, if necessary, online, using the MS Teams tool. The form of teaching is specified by the teacher at the beginning of the semester, updated continuously.

Course assessment

Total number of assessed students: 22

A	В	С	D	Е	FX
9.09	18.18	18.18	22.73	27.27	4.55

Provides: doc. RNDr. Miroslava Martinková, PhD., univerzitná profesorka

Date of last modification: 04.08.2022

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: KPE/ Course name: Pedagogy Pg/15 Course type, scope and the method: Course type: Lecture Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: present **Number of ECTS credits: 2 Recommended semester/trimester of the course:** 3. Course level: I. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 1139 C Α В D Е FX 23.97 28.8 22.91 13.78 8.6 1.93 Provides: PaedDr. Michal Novocký, PhD., doc. PaedDr. Renáta Orosová, PhD.

Date of last modification: 12.03.2024

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ | Course name: Physical Chemistry

FCHU/22

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 3 / 2 Per study period: 42 / 28

Course method: present

Number of ECTS credits: 6

Recommended semester/trimester of the course: 4.

Course level: I.

Prerequisities: ÚCHV/VCHU/14 or ÚCHV/VCHU/10 or ÚCHV/VACH/10 or ÚCHV/VCHU/15

Conditions for course completion:

Active participation in seminars. Two partial tests from computational seminars, each must be mastered at A-E. In the case of distance learning, it is necessary to prepare 2 assignments, each must be mastered at 80%.

Examination, unerstanding of three thematic areas of the subject (thermodynamics, electrochemistry, kinetics), must be mastered at A-E.

Learning outcomes:

Acquirement of the basics knowledgements of physical chemistry within the chapters: thermodynamics, phase equilibria, chemical equilibria, electrochemistry, chemical kinetics.

Brief outline of the course:

Fundamental concepts of thermodynamics, thermochemistry, chemical equilibrium, phase equilibria and diagrams, laws for ideal gas and reals gases, liquids, solutions, solutions of electrolytes. Electrochemistry: ionics and electrodics. Electrodes and electrochemical cells, corrosion. Chemical kinetics, catalysis. Adsorption.

Recommended literature:

T. Engel, P. Reid: Physical Chemistry, Pearson Educat. Inc., San Francisco 2006

P.W. Atkins: Physical Chemistry, Oxford University Presss, Oxford 1986, 1990, 1996

W.J. Moore: Physical Chemistry, Longman, London 1972 and newer editions

Course language:

Notes:

Teaching is carried out in person. If a distance form is required, the lectures will take place online, using the BigBlueButton (https://bbb.science.upjs.sk/). Other conditions will be specified by the teacher.

Course assessment

Total number of assessed students: 23

A	В	С	D	E	FX
0.0	17.39	30.43	21.74	17.39	13.04

Page: 131

Provides: RNDr. Andrea Morovská Turoňová, PhD., RNDr. Ján Macko, PhD., RNDr. Ivana

Šišoláková, PhD., univerzitná docentka, prof. RNDr. Renáta Oriňaková, DrSc.

Date of last modification: 22.07.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ | Course name: Physical Chemistry II

FCH1b/10

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 3 / 2 Per study period: 42 / 28

Course method: present

Number of ECTS credits: 6

Recommended semester/trimester of the course: 6.

Course level: I.

Prerequisities: ÚCHV/FCH1a/03 or ÚCHV/FCH1a/21 or ÚCHV/FCHU/10

Conditions for course completion:

- 1. Participation in seminars (also applies to the online form of teaching). Students are required to attend seminars. The relevant teacher who leads the seminar will justify the reasoned absence of the student (incapacity for work, family reasons, etc.) in a maximum of two seminars during the semester without the need for replacement. In the event of a longer-term reasoned absence (for example due to incapacity for work), the relevant teacher will provide the student with an alternative form of mastering the missed material.
- 2. Activity at seminars. The preparation of students and their regular monitoring is always assessed by the relevant teacher who conducts the seminar, within his/her competence.
- 3. Two tests from computational exercises, usually in the 6th and 12th week of the semester. To successfully pass each test, it is necessary to obtain at least 8 points (out of 15 points). Successful completion of continuous tests is a condition of admission to the oral exam.
- 4. The exam is observed in a regular oral form, resp. in case of restrictions of contact forms of the pedagogical process, the exam is performed by a suitable distance electronic form.
- 5. To successfully master the subject, it is necessary to prove mastery of the required curriculum at least 51%.

Learning outcomes:

Students will gain knowledge about the principles that govern the speed of chemical processes, the kinetics and mechanism of some selected reactions, the balance and kinetics of electrode processes. They will also learn the basics of electrochemistry and catalysis.

Brief outline of the course:

Electrochemistry. Equilibrium homogeneous processes electrolyte solutions. Charge transfer in electrolyte solutions. Nonequilibrium homogeneous processes. Transport processes in electrolyte solutions. Conductance and molar conductivity. Hindering effects. Transport numbers. Equilibrium in heterogeneous electrochemical systems. Pocesses on charged interfaces. Electrochemical cells and fuel cells. Classification of electrode types. Concentration cells. Electrolysis. Electrochemical power sources. Potentiometry. Electrical double layer. Surface tension.

Chemical kinetics. Homogeneous processes. Reaction rate. Reaction order. Classification of chemical reactions. Elementary chemical reactions. Mechanism and kinetics equations of complicated chemical processes. Methods of rate low determination. Theory of chemical kinetics.

Ttemperature dependence of reaction rates. Collision theory. Activated complex theory. Chain reactions. Structure and rate lows of chain reactions. Explosion. Polymerisation reactions. Photochemical reactions. Catalysis. Theory of homogeneous catalysis. Chemical oscillation reactions. Heterogeneous processes. Difusion. Physical and chemical adsorption. Adsorption and diffusion. Processes in heterogeneous electrochemical systems. Electrode kinetics, activation and diffusive mechanism of charge transfer.

Application of theoretical relationships on the solving of concrete problems and on the calculation of examples during seminars.

Recommended literature:

T. Engel, P. Reid: Physical Chemistry, Pearson Educat. Inc., San Francisco 2006

P.W. Atkins: Physical Chemistry, Oxford University Presss, Oxford 1986, 1990, 1994, 1998

W.J. Moore: Physical Chemistry, Longman, London 1972 and newer editions

Course language:

Slovak language

Notes:

Teaching is carried out in person or, if necessary, remotely using the bbb or MS Teams tool. The form of teaching is specified by the teacher at the beginning of the semester, updated continuously.

Course assessment

Total number of assessed students: 606

A	В	С	D	Е	FX
15.68	18.32	22.11	18.98	21.12	3.8

Provides: prof. RNDr. Renáta Oriňaková, DrSc., RNDr. Jana Shepa, PhD., RNDr. Ján Macko, PhD., RNDr. Ivana Šišoláková, PhD., univerzitná docentka

Date of last modification: 25.11.2021

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚFV/ | **Course name:** Physics for Chemists

FPCh/21

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours):

Per week: 2 / 2 Per study period: 28 / 28

Course method: present

Number of ECTS credits: 5

Recommended semester/trimester of the course: 1.

Course level: I.

Prerequisities:

Conditions for course completion:

Two written examinations during the semester, where students apply the new knowledge by solving problems.

Oral exam where students present theoretical knowledge of the thematic areas listed in the syllabus.

Learning outcomes:

Completing the course students will get knowledge of fundamental physical laws and will understand their relation to chemistry.

Brief outline of the course:

- 1. Kinematics of a point mass.
- Average and instantaneous velocity, 1D and 3D.
- Acceleration of a point mass (free fall, angled shot).
- Steady movement on a circle.
- 2. Dynamics of a mass point I.
- Newton's laws, applications. Different types of forces. Friction.
- 3. Dynamics of a mass point II.
- Mechanical work.
- Kinetic energy.
- Conservative force field, potential energy (gravitational, springs).
- The law of conservation of mechanical energy.
- The power.
- 4. System of mass points and rigid bodies I.
- Center of gravity. 1st impulse theorem.
- The law of momentum conservation.
- 5. System of mass points and rigid bodies II.
- Rotary motion. Angular momentum, moment of inertia. 2nd impulse theorem.
- The law of angular momentum conservation. Kinetic energy of rotational motion of rigid bodies.
- Balance.
- 6. Fluid mechanics I.
- Ideal fluid. Density and pressure.
- Hydrostatics, pressure measurement. Pascal's law. Archimedes' law.

- 7. Fluid mechanics II.
- Fluid dynamics.
- Continuity equation.
- Bernoulli equation, applications.
- 8. Molecular physics and thermodynamics I.
- Molecular structure of substances (osmosis, Brownian motion).
- Amount of substances, molar mass, Avogadro's law.
- Internal energy. Temperature and its measurement (Celsius, Kelvin).
- Heat, heat capacity. Latent heat.
- 9. Molecular physics and thermodynamics II.
- Ideal gas: state equation, internal energy, speed distribution.
- I. law of thermodynamics. Isothermal, adiabatic and cyclic processes.
- Heat transfer: conduction, convection, radiation.
- II. law of thermodynamics. Entropy.
- Heat engines, Carnot cycle.
- 10. Electricity and magnetism I.
- Electric charge. Coulomb's law. Electric field intensity and potential (voltage).
- Capacitor, capacity.
- Electric current. Ohm's law. Electrical power. Kirchhoff's laws.
- 11. Electricity and magnetism II.
- Magnetism. Magnetic induction, Lorentz force. Ampere's force. Biot-Savart law.
- Faraday's law of electromagnetic induction. Lenz's law.
- 12. Modern physics
- Relativity. Introduction to quantum physics.
- Atomic physics. Nuclear physics, applications. Elementary particles and cosmology.

Recommended literature:

- 1. V. Hajko, J. Daniel-Szabó: Základy fyziky. Veda, Bratislava, 1980.
- 2. Š. Veis, J. Maďar, V. Martišovič: Všeobecná fyzika 1, Mechanika a molekulová fyzika. Alfa, Bratislava, 1978.
- 3. P. Čičmanec: Všeobecná fyzika 2, Elektrina a magnetizmus. Alfa, Bratislava, 1980.
- 4. R.P. Feynman, R.B. Leighton, M. Sands: Feynmanove prednášky z fyziky 1-5. Alfa, Bratislava, 1985.
- 5. V. Hajko a kol.: Fyzika v príkladoch. Alfa, Bratislava, 1983.

Course language:

Slovak language.

Notes:

Course assessment

Total number of assessed students: 205

A	В	С	D	Е	FX
26.83	22.44	24.88	12.68	13.17	0.0

Provides: doc. Mgr. Gregor Bánó, PhD., RNDr. Zuzana Jurašeková, PhD., Mgr. Andrej Hovan, PhD.

Date of last modification: 22.09.2021

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ | **Course name:** Porous materials and their applications

ADP/03

Course type, scope and the method:

Course type: Lecture / Practice

Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14

Course method: present

Number of ECTS credits: 5

Recommended semester/trimester of the course: 6.

Course level: I., II., III.

Prerequisities:

Conditions for course completion:

Written test in the middle and the end of the semester.

Learning outcomes:

To make the acquaintance of various types of advanced porous solids and basic methods for their investigation. To gen up the students with the methods used in characterisation of specific surface area and pore size of different types of porous materials.

Brief outline of the course:

Terminology and principal terms associated with powders, porous solids and adsorption. Methodology of adsorption at the gas-solid interface, liquid-solid interface. Assessment of surface area and porosity. Inorganic materials (active carbon, metal oxides, zeolites, clay minerals, new advanced materials) and phenomenon of adsorption. Application in the industry and everyday life.

Recommended literature:

- 1. F. Rouquerol, J. Rouquerol, K. Sing: Adsorption by powders and porous solids, Academic press, London, UK, 1999
- 2. S. J. Gregg, K.S.W. Sing: Adsorption, surface area and porosity, Academic Press, London,, UK. 1982.
- 3. V. Zeleňák: Adsorption and porosity of solid substances, internal study text, PF UPJŠ, 2020.

Course language:

Notes:

The course is standardly realized in full-time form, in case of necessary circumstances by distance.

Course assessment

Total number of assessed students: 100

A	В	С	D	Е	FX	N	Р
77.0	10.0	4.0	0.0	0.0	0.0	0.0	9.0

Provides: prof. RNDr. Vladimír Zeleňák, DrSc.

Date of last modification: 21.11.2021

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course name: Positive Psychology

KPPaPZ/PP/15

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of ECTS credits: 2

Recommended semester/trimester of the course: 4., 6.

Course level: I.

Prerequisities:

Conditions for course completion:

Assessment is based on interim evaluation. The subject will be taught in both present and distance format. Up-to-date information concerning the subject for the given academic year can be found on the electronic board of the subject in the Academic information system of the UPJŠ.

Learning outcomes:

Students will acquire basic knowledge concerning the reasons for founding Positive psychology, its main theory, current research, as well as application of Positive psychology as a new and rapidly developing field within psychology. Students will also gain experience in applying critical thinking to the challenges and issues that Positive psychology brings and raises in the context of the individual in contemporary society. Emphasis is placed on the ability to critically evaluate current topics of positive psychology.

Brief outline of the course:

- 1. Different perspectives on well-being nad happiness in psychology
- 2. Main theoretical approaches to positive psychology
- 3. Positive emotions and positivity
- 4. Meaningfulness
- 5. Positive interpersonal relations
- 6. Post-traumatic growth
- 7. Hope and optimism
- 8. Gratitude
- 9. Spirituality as a personality dimension
- 10. Wisdom
- 11. Positive institutions
- 12. New themes and topics in PP

Recommended literature:

Brewer, M. B, Hwestone, M: Emotion and Motivation, Blackwell, 2004

Deci, E., Ryan R. M., Handbook of Self – Determination Reasearch, Rochester, 2002

Křivohlavý, J.: Pozitivní psychologie. Praha, Portál, 2003

Křivohlavý, J.: Psychologie vděčnosti a nevděčnosti. Praha, Grada, 2007

Křivohlavý, J.: Psychologie moudrosti a dobrého života, Praha, Grada, 2012

Křivohlavý, J.: Psychologie pocitu štěstí, Grada, 2013

McAdams, D. P., The Person, New York, 2002

Seligman, M. E. P., & Csikszentmihalyi, M. (Eds.). (2000). Positive psychology [Special issue] American Psychologist, 55(1).

Říčan, P.: Psychologie náboženství a spirituality, Praha, Portál, 2007

Slezáčková, A.: Pruvodce pozitivní psychologií, Praha, Grada, 2012

Course language:

Notes:

Course assessment

Total number of assessed students: 457

A	В	С	D	Е	FX
98.25	1.31	0.22	0.0	0.22	0.0

Provides: Mgr. Jozef Benka, PhD.

Date of last modification: 24.06.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ Cours

Course name: Practical from Inorganic Chemistry

PACHU/03

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 4 Per study period: 56

Course method: present

Number of ECTS credits: 4

Recommended semester/trimester of the course: 2.

Course level: I.

Prerequisities: ÚCHV/VCHU/14 or ÚCHV/VCHU/15 or ÚCHV/VCHU/10 or ÚCHV/VACH/10

Conditions for course completion:

Learning outcomes:

Acquisition of practical skills and knowledge necessary for work in a chemical laboratory in the preparation of inorganic and other compounds, in the preparation of solutions, methods of distillation and other basic techniques of work in the laboratory. Students will also be able to perform basic characterization of substances and proof reactions.

Brief outline of the course:

The utilization of common laboratory techniques for preparation of elements (H2, O2, Cu, Ni), oxides(CO2, Al2O3·xH2O), nitrides(Mg3N2), acids (HNO3, H3BO3), salts((NH4)2SO4, KMnO4), binary salts(NH4)Fe(SO4)2·12H2O), halides (CuCl, CuCl2·2H2O, CuBr2) and coordination compounds [Cu(NH3)4]SO4·H2O, K3[Al(C2O4)3]·3H2O).

Recommended literature:

- J. Černák, J. Bubanec, M. Dzurillová, V. Zeleňák: Praktikum z anorganickej chémie. UPJŠ Košice, 1999.
- Z. Vargová, J. Kuchár: Základné praktikum z anorganickej chémie, UPJŠ, Košice, 2009. Z. Vargova, M.Almáši, J. Kuchár, J.Dinajová: Základné laboratórne cvičenia z anorganickej chémie, ŠafárikPress, 2020.

Course language:

Notes:

Course assessment

Total number of assessed students: 657

A	В	С	D	Е	FX
54.19	27.09	13.7	2.59	1.67	0.76

Provides: RNDr. Martin Vavra, PhD., RNDr. Miroslava Matiková Maľarová, PhD., Mgr. Nikolas Király, PhD., doc. RNDr. Juraj Kuchár, PhD., prof. RNDr. Zuzana Vargová, Ph.D.

Date of last modification: 22.07.2022

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ **Course name:** Practical in Physical Chemistry PFCU/22 Course type, scope and the method: Course type: Practice **Recommended course-load (hours):** Per week: 4 Per study period: 56 Course method: present **Number of ECTS credits: 4 Recommended semester/trimester of the course:** 5. Course level: I. Prerequisities: ÚCHV/FCHU/21 or ÚCHV/FCHU/22 or ÚCHV/FCHU/10 **Conditions for course completion:** 1. Adequate theoretical preparation for individual tasks of experimental practice according to the recommended literature. 2. Passing tasks with relevant results. 3. Processing of experimental work results in the form of a protocols and its acceptance. 4. Assessment of theoretical knowledges and practical skills. > In the case of distance learning: 1. Elaboration of a paper on a selected topic and its presentation. 2. Theoretical preparation in the form of protocols, where the basic principles of individual tasks are stated 3. Teaching is realized in blocks without limiting the scope in the alternative term. **Learning outcomes:** Theoretical principles, description of each technique and appropriate physical chemistry experiments. **Brief outline of the course:** Experimental verification of theoretical knowledge on thermodynamics, thermochemistry, chemical equilibria (determination of enthalpy, phase diagrams), colligative properties (cryoscopy, ebulioscopy), adsorption. Experimental verification of theoretical knowledge on electrochemistry (conductivity, dissociation constants, activity coefficients, electromotive force of galvanic cell, Daniell cell, potentials, polarography) and chemical kinetics (determination of rate constants). **Recommended literature:** B.P. Levitt: Findlay's Practical Physical Chemistry, Longman, London 1973 W.J. Moore: Physical Chemistry, Longman, London 1972 P.W. Atkins: Physical Chemistry, Oxford University Press, Oxford, New York 2002 Course language:

Notes:

Teaching is carried out in person. If a distance form is required, the conditions will be specified by the teacher.

Course assessment

Total number of assessed students: 46

A	В	С	D	Е	FX
97.83	2.17	0.0	0.0	0.0	0.0

Provides: RNDr. František Kaľavský, RNDr. Andrea Morovská Turoňová, PhD., RNDr. Jana Shepa, PhD., RNDr. Ján Macko, PhD.

Date of last modification: 22.07.2022

	COURSE INFORMATION LETTER
University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚMV/ TPP2/22	Course name: Probability theory
Course type, scope a Course type: Lectur Recommended cour Per week: 2 / 2 Per Course method: pre	re / Practice rse-load (hours): study period: 28 / 28
Number of ECTS cr	
Recommended seme	ster/trimester of the course: 6.
Course level: I.	
Prerequisities: ÚMV	/MAN2c/22
1	e completion: % in two written tests during the semester. d on written tests and oral exam.
_	ge of the axiomatic theory of probability, random variables and their applications.
Conditional probability Random variables, the Mean, variance and soluted Discrete and absoluted Quantile and character moments. Median and Transformation of rand Special types of descriptions.	finitions and properties of probability. Ity and independence. eir distribution function and characteristics. kewness. Ely continuous distributions. eristic functions, their properties. Relation between characteristic function and d mode. Indom variables. istributions with applications (binomial, Poisson, geometric, uniform, chi-square, Student, Fisher).
2. DeGroot, M. H., S. 3. Evans, M. J., Rose W. H. Freeman, 2009 4. Riečan et al.: Pravo	ravdepodobnosť v príkladoch, UPJŠ, Košice, 2006 (in Slovak) chervish, M. J.: Probability and Statistics, 4th ed., Pearson, Boston, 2012 nthal, J. S.: Probability and Statistics: The Science of Uncertainty, 2nd Ed.,
Course language: Slovak	

Notes:

Course assessment							
Total number of assessed students: 110							
Α	В	С	D	Е	FX		
30.91	13.64	10.0	9.09	36.36	0.0		

Provides: doc. RNDr. Daniel Klein, PhD., RNDr. Andrej Gajdoš, PhD.

Date of last modification: 17.02.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚINF/ | Course name: Programming, algorithms, and complexity

PAZ1a/15

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 3 / 4 Per study period: 42 / 56

Course method: present

Number of ECTS credits: 8

Recommended semester/trimester of the course: 3., 5.

Course level: I.

Prerequisities:

Conditions for course completion:

Graded activities during semester: assignments, small exams, midterm, final project.

Final examination: practical finalterm focused on a complex task.

Rules to pass the subject: Pass the minimal limit of points for category of homeworks (assignments, final project) and tests (small exams, midterm). Get at least 42% from the finalterm and pass the defined limit of total points for all graded activities.

Learning outcomes:

Get an ability to implement basic Java programs and obtain essential knowledge related to object-oriented programming.

Brief outline of the course:

- 1. Introduction to Java and JPAZ2 framework, first Eclipse project, interactive communication with objects using turtle graphics, repeating code in loops, notion of class, object, and method.
- 2. For-loops, local variables, variable types, arithmetic expressions, random numbers, random walk, conditions.
- 3. While-loop, returning a value from a method, reference and reference variables, debugging.
- 4. Primitive and reference types, chars, String objects (including basic algorithms), mouse events, instance variables.
- 5. Array of primitive values and array of references, simple array algorithms.
- 6. Advanced array algorithms, two-dimensional array.
- 7. Exceptions and exception handling, files and directories, writing to text files.
- 8. Reading from text files.
- 9. Creating classes, encapsulation, getters and setters, constructors and their hierarchy, method overloading.
- 10. Inheritance and polymorphism.
- 11. Java Collections Framework, ArrayList class, wrapper classes for primitive types and autoboxing, interfaces List, Set, Map and their implementations, methods equals and hashCode.
- 12. Access modifiers, abstract classes and methods, creating and implementing interfaces, sorting, static methods and variables.
- 13. Creating and throwing exceptions, checked and runtime exceptions, JavaDoc, Maven.

Recommended literature:

- 1. ECKEL, Bruce. Thinking in Java. Fourth edition. Upper Saddle River, NJ: Prentice Hall, c[2006]. ISBN 978-01-318-7248-6.
- 2. PECINOVSKÝ, Rudolf. OOP: naučte se myslet a programovat objektově. Brno: Computer Press, 2010. ISBN 978-80-251-2126-9.
- 3. SIERRA, Kathy a Bert BATES. Head first Java. Vyd. 2. Sebastopol: O'Reilly, 2005. ISBN 978-05-960-0920-5.

Course language:

Slovak language, english language is required only to read Java API documentation.

Notes:

Course assessment

Total number of assessed students: 891

A	В	С	D	Е	FX
16.16	8.53	11.78	18.29	13.8	31.43

Provides: RNDr. Juraj Šebej, PhD., RNDr. Miroslav Opiela, PhD., RNDr. Zoltán Szoplák, RNDr. Viktor Pristaš, doc. RNDr. Ondrej Krídlo, PhD., RNDr. Richard Staňa, Mgr. Viktor Olejár

Date of last modification: 04.01.2022

University: P. J. Šafárik University in Košice Faculty: Faculty of Science **Course ID:** Course name: Psychology KPPaPZ/Ps/15 Course type, scope and the method: Course type: Lecture Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: present **Number of ECTS credits: 2 Recommended semester/trimester of the course:** 3. Course level: I. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 858 C Α В D Е FX 37.41 20.98 16.2 12.59 11.07 1.75

Provides: PhDr. Anna Janovská, PhD., Mgr. Ondrej Kalina, PhD.

Date of last modification: 24.06.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: Course name: Psychology of Everyday Life

KPPaPZ/PKŽ/15

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of ECTS credits: 2

Recommended semester/trimester of the course: 3.

Course level: I.

Prerequisities:

Conditions for course completion:

The evaluation of the course and its subsequent completion will be based on clearly and objectively set requirements, which will be set in advance and will not change. The aim of the assessment is to ensure an objective and fair mapping of the student's knowledge while adhering to all ethical and moral standards. There is no tolerance for students' fraudulent behavior, whether in the teaching process or in the assessment process.

- 1. Active participation in seminars
- 2. Elaboration and presentation of PPT presentation on the assigned topic. Maximum number of points 20; minimum number of points 11.
- 3. Elaboration of an essay in the range of 4xA4 (standard pages). Maximum number of points 20; minimum number of points 11.

The final evaluation (grade) is the sum of points for the presentation and the essay.

A 40b - 37b

B 36b - 33b

C 32b - 29b

D 28b - 25b

E 24b - 21b

FX 20b - 0b

Learning outcomes:

The student is able to demonstrate an understanding of the individual's behavior in selected everyday situations such as conflict, group influence, empathy, helping, aggression, etc.

The student is able to describe, explain and evaluate the psychological mechanisms that occur in everyday situations.

The student is able to apply basic psychological knowledge to himself (self-regulation) but also in interaction with others (cooperation).

The method of teaching the subject will be oriented to the student. Speakers will be interested in the needs, expectations and opinions of students so as to encourage them to think critically by expressing respect and feedback on their opinions and needs.

The content of the curriculum will be based on primary and high-quality sources that will reflect the topicality of the topics so as to ensure the connection of the curriculum with other subjects and also

the connection of the curriculum with practice. Students will be expected to take an active approach in lectures and seminars with an emphasis on their independence and responsibility.

Brief outline of the course:

How to understand human behavior (overview of basic approaches in psychology); Basic overview of cognitive processes; Learning processes and their use in practice; Social influences, prosocial and antisocial behavior; How human emotions and motivations work; Deciding - why and when we take risks; Childhood experiences and their relationship to adulthood; Abnormal behavior, mental disorders and therapeutic approaches

Recommended literature:

Course language:

Notes:

Course assessment

Total number of assessed students: 228

A	В	С	D	Е	FX
42.11	25.0	26.32	4.82	1.32	0.44

Provides: Mgr. Ondrej Kalina, PhD.

Date of last modification: 24.06.2022

University: P. J. Šafárik University in Košice Faculty: Faculty of Science **Course ID:** Course name: Resolving Conflict Situations in Educational Practice KPPaPZ/RKS/14 Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 1/2 Per study period: 14/28 Course method: present **Number of ECTS credits: 4** Recommended semester/trimester of the course: 3., 5. Course level: I., N **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 178 abs n 94.38 5.62 Provides: PhDr. Anna Janovská, PhD., Mgr. Lucia Barbierik, PhD. Date of last modification: 24.06.2022 Approved: doc. RNDr. Stanislav Lukáč, PhD., prof. RNDr. Vladimír Zeleňák, DrSc.

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: KPE/ Course name: School Administration and Legislation OLŠ/15 Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: present **Number of ECTS credits: 2** Recommended semester/trimester of the course: 3., 5. Course level: I. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 322 C Α В D Е FX 45.65 29.81 14.29 6.52 3.11 0.62 Provides: PaedDr. Michal Novocký, PhD. Date of last modification: 12.03.2024

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚTVŠ/ | Course name: Seaside Aerobic Exercise

ÚTVŠ/CM/13

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of ECTS credits: 2

Recommended semester/trimester of the course:

Course level: I., II.

Prerequisities:

Conditions for course completion:

Completion: passed

Condition for successful course completion:

- active participation in line with the study rule of procedure and course guidelines
- effective performance of all tasks- aerobics, water exercise, yoga, Pilates and others

Learning outcomes:

Content standard:

The student demonstrates relevant knowledge and skills in the field, which content is defined in the course syllabus and recommended literature.

Performance standard:

Upon completion of the course students are able to meet the performance standard and:

- perform basic aerobics steps and basics of health exercises,
- conduct verbal and non-verbal communication with clients during exercise,
- organise and manage the process of physical recreation in leisure time

Brief outline of the course:

Brief outline of the course:

- 1. Basic aerobics low impact aerobics, high impact aerobics, basic steps and cuing
- 2. Basics of aqua fitness
- 3. Basics of Pilates
- 4. Health exercises
- 5. Bodyweight exercises
- 6. Swimming
- 7. Relaxing yoga exercises
- 8. Power yoga
- 9. Yoga relaxation
- 10 Final assessment

Students can engage in different sport activities offered by the sea resort – swimming, rafting, volleyball, football, table tennis, tennis and other water sports in particular.

Recommended literature:

1. BUZKOVÁ, K. 2006. Fitness jóga. Praha: Grada. 167 s.

- 2. ČECHOVSKÁ, I., MILEROVÁ, H., NOVOTNÁ, V. Aqua-fitness. Praha: Grada. 136 s.
- 3. EVANS, M., HUDSON, J., TUCKER, P. 2001. Umění harmonie: meditace, jóga, tai-či, strečink. 192 s.
- 4. JARKOVSKÁ, H., JARKOVSKÁ, M. 2005. Posilováni s vlastním tělem 417 krát jinak. Praha: Grada. 209 s.
- 5. KOVAŘÍKOVÁ, K. 2017. Aerobik a fitness. Karolium, 130 s.

Course language:

Slovak language

Notes:

Course assessment

Total number of assessed students: 54

abs	n
11.11	88.89

Provides: Mgr. Agata Dorota Horbacz, PhD.

Date of last modification: 29.03.2022

University: P. J. Šafárik University in Košice Faculty: Faculty of Science **Course ID:** KF/ Course name: Selected Topics in Philosophy of Education (General VKFV/07 Introduction) Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: present **Number of ECTS credits: 2** Recommended semester/trimester of the course: 3., 5. Course level: I. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 32 C Α В D Е FX 68.75 18.75 9.38 3.13 0.0 0.0

Provides: PhDr. Dušan Hruška, PhD.

Date of last modification: 13.04.2022

University: P. J. Šafárik University in Košice Faculty: Faculty of Science **Course ID:** KF/ Course name: Selected Topics in Philosophy of Education (General VKFV/07 Introduction) Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: present **Number of ECTS credits: 2** Recommended semester/trimester of the course: 6. Course level: I. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 32 C Α В D Е FX 68.75 18.75 9.38 3.13 0.0 0.0

Provides: PhDr. Dušan Hruška, PhD.

Date of last modification: 13.04.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚMV/ | **Course name:** Selected topics in elementary mathematics

VEM/22

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 1 / 1 Per study period: 14 / 14

Course method: present

Number of ECTS credits: 2

Recommended semester/trimester of the course: 5.

Course level: I.

Prerequisities: ÚMV/MAN2c/22

Conditions for course completion:

During the term, each student receives marks for two written exams. Final marking is assigned based on the overall points for the work throughout the term, for homework and their presentation. Marking classification: A:91%-100%, B:81%-90%, C:71%-80%, D:61%-70%, E:51%-60%, FX:0%-50%

Learning outcomes:

Obtain knowledge about the structure of elementary mathematics with respect to advanced mathematics; the development of mathematical skills of prospective teachers. The student will

- 1. familiarise themselves with mathematical culture, ways of thinking, self-expression and putting forward arguments,
- 2. gain a deeper understanding of the base terminology of real analysis, their properties and interconnections,
- 3. be able to define and interpret key terms, prove their basic properties and relationships,
- 4. know how to solve tasks focused on utilising the aforementioned concepts and interpret the obtained results.

Brief outline of the course:

Theory of Equations and Inequalities, Solving Higher Order Polynomials, The Role of CAS systems in Solving Equations and Inequalities,

Building the Real Number System, Rational and Irrational Numbers, Farey Sequences, Review of Geometric Series: Preparation for Decimal Representation, Decimal Expansion, Decimal Periodicity, Building the Complex Numbers, Operating on the Complex Numbers, Picturing Complex Numbers and Connections to Transformation Geometry, The Polar Form of Complex Numbers and De Moivre's Theorem, Some Connections to Roots of Polynomials, Euler's Identity and the Irrationality of e,

Functions and Modeling, Ways of Representing Functions, Solutions of Cubic Equations Using Trigonometry

Recommended literature:

J. Doboš: Rovnice a nerovnice, Bolchazy-Carducci Publ., 2003.

W.W. Esty: The language of mathematics, Montana State University, 2007.

F. Klein: Elementary Mathematics from an Advanced Standpoint, Dower Publications, 1945.

F. Kuřina, Z. Půlpán: Podivuhodný svět elementární matematiky, Academia, Praha, 2006.

P. Vrábel: Heuristika a metodológia matematiky, Nitra, 2005.

Course language:

Slovak

Notes:

Course assessment

Total number of assessed students: 58

A	В	С	D	Е	FX
6.9	27.59	13.79	24.14	27.59	0.0

Provides: prof. RNDr. Jozef Doboš, CSc.

Date of last modification: 25.04.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: Course name: Self Marketing ECo-C2

KPPaPZ/ECo-C2/14

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: combined, present

Number of ECTS credits: 4

Recommended semester/trimester of the course: 4., 6.

Course level: I., N

Prerequisities:

Conditions for course completion:

1. Active participation in lessons (absence is allowed max. 90 min.), 2. Realization of assignments according to the teacher's instructions.

Detailed information in the electronic bulletin board of the course in AIS2. The teaching of the subject will be realized by a combined method.

Learning outcomes:

The student is able to understand and explain the basic assumptions of good self-marketing, knows the possibilities for the correct presentation of his own person and understands the related knowledge and principles of personal and communication area. He / she can understand his / her competencies, his / her goals, how to make his / her strengths visible and he / she can apply this knowledge and social and professional skills in the personal and professional sphere of his / her life, which will also improve his / her employment opportunities.

Brief outline of the course:

What is marketing? (Marketing - Mix)

Basics of self-marketing (Personal opinion is crucial, Goal setting, Proper use of opportunity)

Me and my influence (What can I offer? What does he / she have unlike me? How do others see me? Ability to defend one's own opinion, Think positively!, I know how to explore myself - what options do I have?),

Competence (Have your own opinion, How to withstand criticism, Be a team player, Competence at work),

Draw attention to yourself (Voice and word selection, Active in meetings, Present yourself successfully).

Recommended literature:

VÝROST, Jozef - SLAMĚNÍK, Ivan. Sociální psychologie. 2., přepr. a rozš. vyd. Praha : GRADA, 2008. 408 s.

VÝROST, Jozef - SLAMĚNÍK, Ivan. Aplikovaná sociální psychologie I : Člověk a sociální instituce. 1. vyd. Praha : Portál, 1998. 384 s. ISBN 80-7178-269-6.

KOMÁRKOVÁ, Růžena - SLAMĚNÍK, Ivan - VÝROST, Jozef. Aplikovaná sociální

psychologie III : Sociálněpsychologický výcvik. 1. vyd. Praha : Grada Publishing, 2001. 224 s.

VÝROST, Jozef - SLAMĚNÍK, Ivan. Aplikovaná sociální psychologie II. 1. vyd. Praha : Grada Publishing, 2001. 260 s.

Course language:

slovak

Notes:

After passing the certification exams from all 4 modules (Teamwork, Selfmarketing, Conflict Management, Communication) the student will receive an ECo-C card and an ECo-C certificate.

Course assessment

Total number of assessed students: 163

abs	n
90.18	9.82

Provides: Mgr. Lucia Barbierik, PhD.

Date of last modification: 24.06.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚMV/ | Course name: Seminar to mathematical clubs

SMK/17

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of ECTS credits: 2

Recommended semester/trimester of the course: 6.

Course level: I.

Prerequisities:

Conditions for course completion:

Conditions for continuous evaluation:

- 1. Participation in teaching in accordance with the study rules and instructions of the teacher.
- 2. Activity.
- 3. Homework and written tests.
- 4. Seminar work and its presentation at the seminar plan the selected topic for one math circle Conditions for successful completion of the course:
- 1. Participation in teaching in accordance with the study regulations and according to the instructions of the teacher;
- 2. Credits will be awarded to a student who scores at least 50% on homework assignments, at least 50% on written tests, and at least 50% on a seminar work. A grade of A requires at least 90%, a grade of B requires at least 80%, a grade of C requires at least 70%, a grade of D requires at least 60%, and a grade of E requires at least 50%.

Learning outcomes:

While solving homework, the student will become familiar with different types of problems from mathematical competitions and demonstrate the ability to solve them with the mathematical apparatus of the student for whom the problem is intended.

While solving problems in written tests, the student will gain proficiency in solving problems from mathematical competitions such as Pythagorean and Mathematical Kangaroo.

The student will demonstrate in the seminar work that he/she can prepare the content of a mathematics circle that are motivating for his/her students.

Brief outline of the course:

The content is focuses on solving problems from mathematical competitions, and on familiarization with activities that will be motivating and fun for pupils and will develop their mathematical thinking

Students will also learn about the structure of mathematical competitions for middle and high school students and will be theoretically prepared for guiding mathematics circle.

The seminars focus on the following topics:

Number theory.

Equations, inequalities, inequalities.

Word problems.

Planimetry.

Stereometry.

Combinatorics. Dirichlet principle. Combinatorial geometry. Probability.

Mathematical games.

Recommended literature:

Acheson, D.: 1089 a další parádní čísla, Dokořán, 2006. (in czech)

Brožúry z edície Škola mladých matematikov. (in slovak)

Séria brožúr: XY. ročník matematickej olympiády. (in slovak)

Ziegler, G.M.: Matematika Vám to spočítá, Universum, Praha, 2011. (in czech)

Zhouf, J. a kol.: Matematické příběhy z korespondenčních seminářu, Prometheus, Praha, 2006.

(in czech)

Course language:

Slovak

Notes:

Course assessment

Total number of assessed students: 136

A	В	С	D	Е	FX
58.09	19.85	11.76	7.35	2.94	0.0

Provides: doc. RNDr. Ingrid Semanišinová, PhD.

Date of last modification: 18.04.2022

	COURSE INFORMATION LETTER
University: P. J. Šafár	ik University in Košice
Faculty: Faculty of Sc	ience
Course ID: ÚCHV/ ASM/03	Course name: Separation Methods
Course type, scope and Course type: Lecture Recommended course Per week: 2 / 1 Per s Course method: pres	e / Practice se-load (hours): tudy period: 28 / 14
Number of ECTS cre	dits: 5
Recommended semes	ter/trimester of the course: 6.
Course level: I.	
	V/ANCHU/03 or ÚCHV/ANCHU/21 or ÚCHV/ANCHE/09 or ÚCHV/ V/ANCH1b/21) and (ÚCHV/PAEC/03 or ÚCHV/PANCH/06 or ÚCHV/ V/PACU/03)
	e completion: sentation of a project focused on the application of separation methods. exam consists of 3 questions (each of 33%), 50% must be obtained for the
Learning outcomes: Survey of basic princ research and analytica	ciples, theoretical background and applications of separation methods in l practice.
SPE, SPME. Chromat phases. Instrumentation High-performance liquin LC, instrumentation Planar chromatograph	sification, theory and applications of separation methods. Extraction - LLE, tographic methods - theory, classification. Gas chromatography, stationary on, detectors in GC. Data evaluation - qualitative and quantitative analysis. and chromatography, principles, classification. Stationary and mobile phases
York 1997.	J.: Principles of instrumental analysis. Saunders College Publishing, NewL.: Handbook of sample preparation, Wiley 2010.
Course language: Slovak, english langua	ige

Notes:

Course assessment							
Total number of assessed students: 500							
A	В	С	D	Е	FX		
28.4	26.0	25.2	12.8	5.4	2.2		

Provides: doc. RNDr. Taťána Gondová, CSc.

Date of last modification: 01.08.2022

University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: KPO/ SPKVV/15	Course name: Social and Political Context of Education
Course type, scope a Course type: Lectur Recommended cour Per week: 2 Per stu Course method: pre	re rse-load (hours): dy period: 28
Number of ECTS cr	edits: 2
Recommended seme	ster/trimester of the course: 4., 6.
Course level: I.	
Prerequisities:	
Conditions for cours Evaluation of the dev A 100,00% - 91,00 B 90,99% - 81,00% C 80,99% - 71,00% D 70,99% - 61,00% E 60,99% - 51,00% FX 50,99% and les	reloped assignment. 0% 6 6 6 6
issues of education as Development of known related to the process The student will be a culturally. He/she wi	of teaching the subject is to impart knowledge and promote reflection on the nd training in the context of social and political change. wledge: the student will be able to know the current theoretical background of education and training in a modern democratic society. ble to navigate the social and political space - politically, legally, socially and ll be able to look for alternatives and solutions to dysfunctions, while at the opportunities and ways to implement them.
and economic object globalisation. Macro	ourse: I functions of education in human life and society. The political, social rives of education. Education, learning and social change in the context of social determinants of education. Current roles of education and training in and democratic society.
Course language: Slovak	
Notes:	

Course assessment							
Total number of assessed students: 161							
Α	В	С	D	Е	FX		
59.63	21.12	12.42	4.35	1.24	1.24		

Provides: Mgr. Ján Ruman, PhD.

Date of last modification: 13.04.2022

COURSE INFORMATION LETTER	
University: P. J. Šafárik University in Košice	
Faculty: Faculty of Science	
Course ID: KGER/ Course name: Specialised German Language - Natural Sciences 1 OJPV1/07	
Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: present	
Number of ECTS credits: 2	
Recommended semester/trimester of the course: 4.	
Course level: I.	
Prerequisities:	
Conditions for course completion: Active participation in class and completed homework assignments. Students are allowed to classes at the most (2x90 min.). 1 control tests during the semester and written assignments grade will be calculated as follows: A 93-100 %, B 86-92%, C 79-85%, D 72-78%, E 65-71 64 % and less.	s. Final
Learning outcomes: The development of students' language skills - reading, writing, listening, speaking, improve of their linguistic competence - students acquire knowledge of selected phonological, lexic syntactic aspects, development of pragmatic competence - students can efectively use the lar for a given purpose, with focus on Academic English and English for specific/professional purpose, level B1.	cal and nguage
Brief outline of the course:	
Recommended literature: Duden Basiswissen Schule. Abitur: Enthält die Bände Mathematik, Physik, Chemie, Biolog Geographie, Geschichte. (2007). ISBN: 978-3411002511. Zettl, E. et al.: Aus moderner Technik und Naturwissenschaft. Ismaning: Hueber, 2003. Reiss, K.: Basiswissen Zahlentheorie: Eine Einführung in Zahlen und Zahlbereiche (Mathe für das Lehramt), Springer, 2007. ISBN: 978-3540453772. Meyer, L., Schmidt, G D.: Basiswissen Ausbildung: Physik. Bildungsverlag EINS, 2008. 978-3427799337. Duden. Schülerduden Biologie: Das Fachlexikon von A-Z. Bibliographisches Institut Berlit 2009. ISBN: 978-3411054275. Mortimer, Ch. E., Müller, U., Beck, J.: Chemie: Das Basiswissen der Chemie. Stuttgart: Th 2014. ISBN: 978-313484311 Deutsch perfekt, GEO, MaxPlanck Forschung a iné printové a elektronické médiá	matik ISBN: n,
Course language:	

German

Notes:

Course assessm	Course assessment				
Total number of	f assessed studen	ts: 148			
Α	В	С	D	Е	FX
24.32	22.97	24.32	20.27	7.43	0.68

Provides: Mgr. Ulrika Strömplová, PhD.

Date of last modification: 09.02.2023

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚTVŠ/ | **Course name:** Sports Activities I.

TVa/11

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of ECTS credits: 2

Recommended semester/trimester of the course: 1.

Course level: I., II.

Prerequisities:

Conditions for course completion:

Min. 80% of active participation in classes.

Learning outcomes:

Sports activities in all their forms prepare university students for their professional and personal life. They have a great impact on physical fitness and performance. Specialization in sports activities enables students to strengthen their relationship towards the selected sport in which they also improve.

Brief outline of the course:

Brief outline of the course:

The Institute of physical education and sport at the Pavol Jozef Šafárik University offers 20 sports activities aerobics; aikido, basketball, badminton, body-balance, body form, bouldering, floorball, yoga, power yoga, pilates, swimming, fitness, indoor football, SM system, step aerobics, table tennis, chess, volleyball, tabata, cycling.

Additionally, the Institute of physical education and sport at the Pavol Jozef Šafárik University offers winter courses (ski course, survival) and summer courses (aerobics by the sea, rafting on the Tisza River) with an attractive programme, sports competitions with national and international participation.

Recommended literature:

BENCE, M. et al. 2005. Plávanie. Banská Bystrica: FHV UMB. 198s. ISBN 80-8083-140-8. [online] Dostupné na: https://www.ff.umb.sk/app/cmsFile.php?disposition=a&ID=571 BUZKOVÁ, K. 2006. Fitness jóga, harmonické cvičení těla I duše. Praha: Grada. ISBN 8024715252.

JARKOVSKÁ, H, JARKOVSKÁ, M. 2005. Posilování s vlastním tělem 417 krát jinak. Praha: Grada. ISBN 9788024757308.

KAČÁNI, L. 2002. Futbal:Tréning hrou. Bratislava: Peter Mačura – PEEM. 278s. ISBN 8089197027.

KRESTA, J. 2009. Futsal. Praha: Grada Publishing, a.s. 112s. ISBN 9788024725345.

LAWRENCE, G. 2019. Power jóga nejen pro sportovce. Brno: CPress. ISBN 9788026427902.

SNER, Wolfgang. 2004. Posilování ve fitness. České Budějovice: Kopp. ISBN 8072322141.

STACKEOVÁ, D. 2014. Fitness programy z pohledu kinantropologie. Praha: Galén. ISBN 9788074921155.

VOMÁČKO, S. BOŠTÍKOVÁ, S. 2003. Lezení na umělých stěnách. Praha: Grada. 129s. ISBN 8024721743.

Course language:

Slovak language

Notes:

Course assessment

Total number of assessed students: 15193

abs	abs-A	abs-B	abs-C	abs-D	abs-E	n	neabs
86.05	0.07	0.0	0.0	0.0	0.05	8.69	5.15

Provides: Mgr. Patrik Berta, Mgr. Agata Dorota Horbacz, PhD., Mgr. Dávid Kaško, PhD., Mgr. Ladislav Kručanica, PhD., Mgr. Richard Melichar, Mgr. Petra Tomková, PhD., Mgr. Marcel Čurgali, Mgr. Alena Buková, PhD., doc. PaedDr. Ivan Uher, MPH, PhD., prof. RNDr. Stanislav Vokál, DrSc., Mgr. Zuzana Küchelová, PhD.

Date of last modification: 07.02.2024

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚTVŠ/ | Course name: Sports Activities II.

TVb/11

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of ECTS credits: 2

Recommended semester/trimester of the course: 2.

Course level: I., II.

Prerequisities:

Conditions for course completion:

active participation in classes - min. 80%.

Learning outcomes:

Sports activities in all their forms prepare university students for their professional and personal life. They have a great impact on physical fitness and performance. Specialization in sports activities enables students to strengthen their relationship towards the selected sport in which they also improve.

Brief outline of the course:

Brief outline of the course:

The Institute of physical education and sport at the Pavol Jozef Šafárik University offers 20 sports activities aerobics; aikido, basketball, badminton, body-balance, body form, bouldering, floorball, yoga, power yoga, pilates, swimming, fitness, indoor football, SM system, step aerobics, table tennis, chess, volleyball, tabata, cycling.

Additionally, the Institute of physical education and sport at the Pavol Jozef Šafárik University offers winter courses (ski course, survival) and summer courses (aerobics by the sea, rafting on the Tisza River) with an attractive programme, sports competitions with national and international participation.

Recommended literature:

BENCE, M. et al. 2005. Plávanie. Banská Bystrica: FHV UMB. 198s. ISBN 80-8083-140-8. [online] Dostupné na: https://www.ff.umb.sk/app/cmsFile.php?disposition=a&ID=571 BUZKOVÁ, K. 2006. Fitness jóga, harmonické cvičení těla I duše. Praha: Grada. ISBN 8024715252.

JARKOVSKÁ, H, JARKOVSKÁ, M. 2005. Posilování s vlastním tělem 417 krát jinak. Praha: Grada. ISBN 9788024757308.

KAČÁNI, L. 2002. Futbal:Tréning hrou. Bratislava: Peter Mačura – PEEM. 278s. ISBN 8089197027.

KRESTA, J. 2009. Futsal. Praha: Grada Publishing, a.s. 112s. ISBN 9788024725345.

LAWRENCE, G. 2019. Power jóga nejen pro sportovce. Brno: CPress. ISBN 9788026427902.

SNER, Wolfgang. 2004. Posilování ve fitness. České Budějovice: Kopp. ISBN 8072322141.

STACKEOVÁ, D. 2014. Fitness programy z pohledu kinantropologie. Praha: Galén. ISBN 9788074921155.

VOMÁČKO, S. BOŠTÍKOVÁ, S. 2003. Lezení na umělých stěnách. Praha: Grada. 129s. ISBN 8024721743.

Course language:

Slovak language

Notes:

Course assessment

Total number of assessed students: 13318

abs	abs-A	abs-B	abs-C	abs-D	abs-E	n	neabs
84.37	0.51	0.02	0.0	0.0	0.05	10.78	4.28

Provides: Mgr. Agata Dorota Horbacz, PhD., Mgr. Dávid Kaško, PhD., Mgr. Marcel Čurgali, Mgr. Patrik Berta, Mgr. Ladislav Kručanica, PhD., Mgr. Richard Melichar, Mgr. Petra Tomková, PhD., Mgr. Alena Buková, PhD., doc. PaedDr. Ivan Uher, MPH, PhD., prof. RNDr. Stanislav Vokál, DrSc., Mgr. Zuzana Küchelová, PhD.

Date of last modification: 07.02.2024

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚTVŠ/ | Course name: Sports Activities III.

TVc/11

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of ECTS credits: 2

Recommended semester/trimester of the course: 3.

Course level: I., II.

Prerequisities:

Conditions for course completion:

min. 80% of active participation in classes

Learning outcomes:

Sports activities in all their forms prepare university students for their professional and personal life. They have a great impact on physical fitness and performance. Specialization in sports activities enables students to strengthen their relationship towards the selected sport in which they also improve.

Brief outline of the course:

Brief outline of the course:

The Institute of physical education and sport at the Pavol Jozef Šafárik University offers 20 sports activities aerobics; aikido, basketball, badminton, body-balance, body form, bouldering, floorball, yoga, power yoga, pilates, swimming, fitness, indoor football, SM system, step aerobics, table tennis, chess, volleyball, tabata, cycling.

Additionally, the Institute of physical education and sport at the Pavol Jozef Šafárik University offers winter courses (ski course, survival) and summer courses (aerobics by the sea, rafting on the Tisza River) with an attractive programme, sports competitions with national and international participation.

Recommended literature:

BENCE, M. et al. 2005. Plávanie. Banská Bystrica: FHV UMB. 198s. ISBN 80-8083-140-8. [online] Dostupné na: https://www.ff.umb.sk/app/cmsFile.php?disposition=a&ID=571 BUZKOVÁ, K. 2006. Fitness jóga, harmonické cvičení těla I duše. Praha: Grada. ISBN 8024715252.

JARKOVSKÁ, H, JARKOVSKÁ, M. 2005. Posilování s vlastním tělem 417 krát jinak. Praha: Grada. ISBN 9788024757308.

KAČÁNI, L. 2002. Futbal:Tréning hrou. Bratislava: Peter Mačura – PEEM. 278s. ISBN 8089197027.

KRESTA, J. 2009. Futsal. Praha: Grada Publishing, a.s. 112s. ISBN 9788024725345.

LAWRENCE, G. 2019. Power jóga nejen pro sportovce. Brno: CPress. ISBN 9788026427902.

SNER, Wolfgang. 2004. Posilování ve fitness. České Budějovice: Kopp. ISBN 8072322141.

STACKEOVÁ, D. 2014. Fitness programy z pohledu kinantropologie. Praha: Galén. ISBN 9788074921155.

VOMÁČKO, S. BOŠTÍKOVÁ, S. 2003. Lezení na umělých stěnách. Praha: Grada. 129s. ISBN 8024721743.

Course language:

Slovak language

Notes:

Course assessment

Total number of assessed students: 9100

abs	abs-A	abs-B	abs-C	abs-D	abs-E	n	neabs
88.37	0.07	0.01	0.0	0.0	0.02	4.46	7.07

Provides: Mgr. Marcel Čurgali, Mgr. Agata Dorota Horbacz, PhD., Mgr. Dávid Kaško, PhD., Mgr. Patrik Berta, Mgr. Ladislav Kručanica, PhD., Mgr. Richard Melichar, Mgr. Petra Tomková, PhD., Mgr. Alena Buková, PhD., doc. PaedDr. Ivan Uher, MPH, PhD., prof. RNDr. Stanislav Vokál, DrSc., Mgr. Zuzana Küchelová, PhD.

Date of last modification: 07.02.2024

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚTVŠ/ | **Course name:** Sports Activities IV.

TVd/11

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of ECTS credits: 2

Recommended semester/trimester of the course: 4.

Course level: I., II.

Prerequisities:

Conditions for course completion:

min. 80% of active participation in classes

Learning outcomes:

Sports activities in all their forms prepare university students for their professional and personal life. They have a great impact on physical fitness and performance. Specialization in sports activities enables students to strengthen their relationship towards the selected sport in which they also improve.

Brief outline of the course:

Brief outline of the course:

The Institute of physical education and sport at the Pavol Jozef Šafárik University offers 20 sports activities aerobics; aikido, basketball, badminton, body-balance, body form, bouldering, floorball, yoga, power yoga, pilates, swimming, fitness, indoor football, SM system, step aerobics, table tennis, chess, volleyball, tabata, cycling.

Additionally, the Institute of physical education and sport at the Pavol Jozef Šafárik University offers winter courses (ski course, survival) and summer courses (aerobics by the sea, rafting on the Tisza River) with an attractive programme, sports competitions with national and international participation.

Recommended literature:

BENCE, M. et al. 2005. Plávanie. Banská Bystrica: FHV UMB. 198s. ISBN 80-8083-140-8. [online] Dostupné na: https://www.ff.umb.sk/app/cmsFile.php?disposition=a&ID=571 BUZKOVÁ, K. 2006. Fitness jóga, harmonické cvičení těla I duše. Praha: Grada. ISBN 8024715252.

JARKOVSKÁ, H, JARKOVSKÁ, M. 2005. Posilování s vlastním tělem 417 krát jinak. Praha: Grada. ISBN 9788024757308.

KAČÁNI, L. 2002. Futbal:Tréning hrou. Bratislava: Peter Mačura – PEEM. 278s. ISBN 8089197027.

KRESTA, J. 2009. Futsal. Praha: Grada Publishing, a.s. 112s. ISBN 9788024725345.

LAWRENCE, G. 2019. Power jóga nejen pro sportovce. Brno: CPress. ISBN 9788026427902.

SNER, Wolfgang. 2004. Posilování ve fitness. České Budějovice: Kopp. ISBN 8072322141.

STACKEOVÁ, D. 2014. Fitness programy z pohledu kinantropologie. Praha: Galén. ISBN 9788074921155.

VOMÁČKO, S. BOŠTÍKOVÁ, S. 2003. Lezení na umělých stěnách. Praha: Grada. 129s. ISBN 8024721743.

Course language:

Slovak language

Notes:

Course assessment

Total number of assessed students: 5671

abs	abs-A	abs-B	abs-C	abs-D	abs-E	n	neabs
82.81	0.28	0.04	0.0	0.0	0.0	7.97	8.9

Provides: Mgr. Marcel Čurgali, Mgr. Agata Dorota Horbacz, PhD., Mgr. Dávid Kaško, PhD., Mgr. Patrik Berta, Mgr. Ladislav Kručanica, PhD., Mgr. Richard Melichar, Mgr. Petra Tomková, PhD., Mgr. Alena Buková, PhD., doc. PaedDr. Ivan Uher, MPH, PhD., prof. RNDr. Stanislav Vokál, DrSc., Mgr. Zuzana Küchelová, PhD.

Date of last modification: 07.02.2024

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ Course na

Course name: Structure determination - spectroscopic methods

MUSU/22

Course type, scope and the method:

Course type: Lecture / Practice Recommended course-load (hours):

Per week: 2 / 3 Per study period: 28 / 42

Course method: present

Number of ECTS credits: 5

Recommended semester/trimester of the course: 6.

Course level: I.

Prerequisities: (ÚCHV/ACHU/21 or ÚCHV/ACHU/03) and (ÚCHV/ANCHU/21 or ÚCHV/ANCHU/03) and (ÚCHV/OCHU/21 or ÚCHV/OCHU/03)

Conditions for course completion:

- 1. Participation in exercises in accordance with the Study Rules of PF UPJŠ.
- 2. Successful execution of 3 control written works on exercises after 4., 8. and 12. weeks of teaching. Obtaining a minimum grade E from seminars.

The test consists of: 1. Solution of 2 structures of unknown compounds on the basis of combined application of spectral methods. 2. Theoretical and practical questions.

Percentage rating: 100-91% (A), 90-81% (B), 80-71% (C), 70-61% (D), 60-51% (E), 50% and less FX.

Learning outcomes:

Fundamentals of molecular spectroscopy and magnetic properties study, as powerful tools for structure determination in chemistry. Ultraviolet, visible, infrared and Raman spectroscopy, mass spectrometry and methods based on magnetic resonance (1H NMR, 13C NMR).

Brief outline of the course:

Fundamentals of molecular spectroscopy, mass spectrometry and magnetic methods as powerful tools for structure determination in chemistry. Ultraviolet and visible spectroscopy. Emission spectroscopy. Symmetry and group theory. Infrared and Raman spectroscopy. Mass spectrometry in organic and analytical chemistry and biochemistry. Nuclear magnetic resonance - NMR. Chemical shift and splitting of signals by spin-spin coupling. Coupling constants. 1H NMR, 13C NMR, NMR of other nuclei. Two- and more dimensional NMR. NMR applications. Methods and instruments used for spectra measurements. Combined application of spectral methods for solution of chemical problems.

Recommended literature:

- 1. Kováč Š., Ilavský D., Leško J.: Spektrálne metódy v organickej chémii a technológii, ALFA, Bratislava, 1987.
- 2. Milata V., Segl'a P.: Vybrané metódy molekulovej spektroskopie. STU BA, 2007.
- 3. Milata V., Segl'a P.: Spektrálne metódy v chémii. STU FCHPT Bratislava 2002.
- 4. Miertuš S. a kol.: Atómová a molekulová spektroskopia, ALFA, Bratislava 1991.
- 5. T. D. W. Claridge: High-Resolution NMR Techniques in Organic Chemistry, 5. Ed., Elsevier, 2016.

Course language:

slovak, english

Notes:

In-person course, alternatively online course using the BigBlueButton tool or MS Teams. The form of teaching is specified by the teacher at the beginning of the semester, updated continuously.

Course assessment

Total number of assessed students: 46

A	В	С	D	Е	FX
36.96	36.96	13.04	10.87	2.17	0.0

Provides: doc. RNDr. Ján Imrich, CSc., doc. RNDr. Juraj Kuchár, PhD., RNDr. Zuzana Kudličková, PhD., RNDr. Monika Tvrdoňová, PhD.

Date of last modification: 16.08.2022

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ Course name: Students Scientific Conference SVKB/04 Course type, scope and the method: **Course type:** Recommended course-load (hours): Per week: Per study period: Course method: present **Number of ECTS credits: 4** Recommended semester/trimester of the course: Course level: I. **Prerequisities: Conditions for course completion:** Present the results of student's work at the Student Scientific Conference and answer questions from committee members and others present. **Learning outcomes:** The student will acquire competences for independent scientific work in the laboratory, for analysis and written processing of obtained results and knowledge. By presenting the obtained results, the student prepares to present the obtained results in the defense of the bachelor's thesis and in front of the professional public at scientific conferences. **Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 12 abs n 100.0 0.0 **Provides:** Date of last modification: 22.07.2022

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: ÚMV/ Course name: Students scientific conference SVK/10 Course type, scope and the method: **Course type:** Recommended course-load (hours): Per week: Per study period: Course method: present **Number of ECTS credits: 4** Recommended semester/trimester of the course: Course level: I., II. **Prerequisities: Conditions for course completion: Learning outcomes:** Individual scientific work of students. Publishing of obtained results in a written form and as a public presentation. **Brief outline of the course: Recommended literature:** With respect to the research problematics (article in journals, books). Course language: Slovak or English **Notes:** Course assessment Total number of assessed students: 24 abs n 100.0 0.0 **Provides:** Date of last modification: 01.12.2021

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚFV/ | Course name: Students` Digital Literacy

DGS/21

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of ECTS credits: 2

Recommended semester/trimester of the course: 1.

Course level: I.

Prerequisities:

Conditions for course completion:

Summary evaluation based on ongoing assessment:

- 1. Practical ongoing assignments and their defense (at least 50% needed)
- 3. Active participation during face-to-face contact learning in classical or virtual classroom (3 absences allowed) and during online learning (no absence, uploading all individual ongoing assignments)

Learning outcomes:

The student should obtain and know to apply basic knowledge and skills in working with current digital technologies (mobile phone, tablet, laptop, web technologies):

- 1. according to the current European framework for the Digital competence DigComp and ECDL
- 2. for better and more effective learning, work and active life in higher education, later lifelong learning and further career prospects.

Brief outline of the course:

- 01.-02. Basic digital skills, DigComp framework, ECDL
- modern web browser and its personalization
- security, privacy, responsible use of DT
- 03.-05. Search, collection and evaluation of digital content
- scanning, audio recording and speech resolution, optical resolution (OCR)
- digital notebooks (Google keep, Evernote, Onenote)
- evaluation of digital resources (Google forms and sections)

06.-08. Editing and creating digital content

- cloud and interactive documents

(text and spreadsheet editors - Google, Microsoft, Jupyter)

- work with pdf documents, e-books and videos

(Kami, Google books, Screencasting)

09. - 10. Organization, protection and sharing of digital content

- modern LMS and cloud storage

(Google Classroom, Microsoft team, Google Drive, Dropbox)

- time management (Google Calendar)

11.-13. Digital communication and cooperation

- collaborative interactive whiteboards (Jamboard, Whiteboard)
- online presentations and online meetings (Google presentations, Powerpoint, Google meet, Microsoft teams)

Recommended literature:

- 1. Carretero Gomez, S., Vuorikari, R. and Punie, Y., DigComp 2.1: The Digital Competence Framework for Citizens with eight proficiency levels and examples of use, Luxembourg, 2017, ISBN 978-92-79-68006-9, https://www.ecdl.sk/
- 2. Bruff, D. (2019). Intentional Tech: Principles to Guide the Use of Educational Technology in College Teaching (1st edition). Morgantown: West Virginia University Press.
- 3. Baker, Y. (2020). Microsoft Teams for Education. Amazon Digital Services.
- 4. Miller, H. (2021). Google Classroom + Google Apps: 2021 Edition. Brentford: Orion Edition Limited.

Course language:

slovak

Notes:

Course assessment

Total number of assessed students: 160

A	В	С	D	Е	FX
69.38	4.38	4.38	0.0	21.88	0.0

Provides: doc. RNDr. Jozef Hanč, PhD.

Date of last modification: 26.01.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚTVŠ/ | Course name: Summer Course-Rafting of TISA River

LKSp/13

Course type, scope and the method:

Course type: Practice

Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of ECTS credits: 2

Recommended semester/trimester of the course:

Course level: I., II.

Prerequisities:

Conditions for course completion:

Completion: passed

Condition for successful course completion:

- active participation in line with the study rule of procedure and course guidelines
- effective performance of all tasks: carrying a canoe, entering and exiting a canoe, righting a canoe, paddling

Learning outcomes:

Content standard:

The student demonstrates relevant knowledge and skills in the field, which content is defined in the course syllabus and recommended literature.

Performance standard:

Upon completion of the course students are able to meet the performance standard and:

- implement the acquired knowledge in different situations and practice,
- implement basic skills to manipulate a canoe on a waterway,
- determine the right spot for camping,
- prepare a suitable material and equipment for camping.

Brief outline of the course:

Brief outline of the course:

- 1. Assessment of difficulty of waterways
- 2. Safety rules for rafting
- 3. Setting up a crew
- 4. Practical skills training using an empty canoe
- 5. Canoe lifting and carrying
- 6. Putting the canoe in the water without a shore contact
- 7. Getting in the canoe
- 8. Exiting the canoe
- 9. Taking the canoe out of the water
- 10. Steering
- a) The pry stroke (on fast waterways)
- b) The draw stroke

- 11. Capsizing
- 12. Commands

Recommended literature:

1. JUNGER, J. et al. Turistika a športy v prírode. Prešov: FHPV PU v Prešove. 2002. ISBN 8080680973.

Internetové zdroje:

1. STEJSKAL, T. Vodná turistika. Prešov: PU v Prešove. 1999.

Dostupné na: https://ulozto.sk/tamhle/UkyxQ2lYF8qh/name/Nahrane-7-5-2021-v-14-46-39#! ZGDjBGR2AQtkAzVkAzLkLJWuLwWxZ2ukBRLjnGqSomICMmOyZN==

Course language:

Slovak language

Notes:

Course assessment

Total number of assessed students: 209

abs	n
37.32	62.68

Provides: Mgr. Dávid Kaško, PhD.

Date of last modification: 29.03.2022

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: KPE/ Course name: Teachers' Support Groups **SSU/15** Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: present **Number of ECTS credits: 2** Recommended semester/trimester of the course: 6. Course level: I., II. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 44 \mathbf{C} Α В D Ε FX 86.36 13.64 0.0 0.0 0.0 0.0 Provides: doc. PaedDr. Renáta Orosová, PhD. Date of last modification: 12.03.2024

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University: P. J. Šafárik University in Košice Faculty: Faculty of Science **Course ID:** Course name: Team Work ECo-C1 KPPaPZ/ECo-C1/14 Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: combined, present **Number of ECTS credits: 4** Recommended semester/trimester of the course: 3., 5. Course level: I., N **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 142 abs n 97.89 2.11 Provides: PhDr. Anna Janovská, PhD. Date of last modification: 28.06.2021 Approved: doc. RNDr. Stanislav Lukáč, PhD., prof. RNDr. Vladimír Zeleňák, DrSc.

University: P. J. Šafárik University in Košice Faculty: Faculty of Science Course ID: KPE/ Course name: Theory of Education TVE/08 Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: present **Number of ECTS credits: 2** Recommended semester/trimester of the course: 4., 6. Course level: I. **Prerequisities: Conditions for course completion: Learning outcomes: Brief outline of the course: Recommended literature:** Course language: **Notes:** Course assessment Total number of assessed students: 645 C Α В D Ε FX 43.72 31.01 16.59 4.96 1.71 2.02

Provides: Mgr. Beáta Sakalová, doc. PaedDr. Renáta Orosová, PhD.

Date of last modification: 12.03.2024