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65. Linux and open source GIS.	
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68. Metageography and planetary geography	
69. Methods of human geographical research	
70. Methods of physical geographical research	
71. Methods of thematic cartography	
72. Microgeography	
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74. Multiculturalism and Multicultural Education	
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83. Physics for Chemists	
84. Political geography	
85. Population Geography	
86. Porous materials and their applications	
87. Positive Psychology	
88. Practical from Inorganic Chemistry	
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90. Psychology	
91. Psychology of Everyday Life	
92. Regional Geography of Europe	
93. Remote sensing applications	
94. Resolving Conflict Situations in Educational Practice	
95. School Administration and Legislation.	
96. Seaside Aerobic Exercise	
97. Selected Topics in Philosophy of Education (General Introduction)	

98. Self Marketing ECo-C2	
99. Seminar of human geography	
100. Seminar of physical geography	
101. Separation Methods	
102. Social and Political Context of Education	
103. Specialised German Language - Natural Sciences 1	
104. Sports Activities I	
105. Sports Activities II	
106. Sports Activities III	
107. Sports Activities IV	
108. Statistical Methods in Geography	
109. Structure determination - spectroscopic methods	
110. Student Scientific Conference in Geography	
111. Students Scientific Conference.	
112. Students' Digital Literacy	
113. Summer Course-Rafting of TISA River	
114. Teachers' Support Groups	
115. Team Work ECo-C1	
116. Theory of Education	

University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: CJP/ PFAJAKA/07	Course name: Academic English
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	ce rse-load (hours): dy period: 28
Number of ECTS cr	edits: 2
Recommended seme	ster/trimester of the course:
Course level: I.	
Prerequisities:	
1 test (13th week), no Presentation on chose Final evaluation- ave	ticipation, assignments handed in on time, 2 absences tolerated o retake.
of their linguistic cor syntactic aspects, dev	students' language skills - reading, writing, listening, speaking, improvement npetence - students acquire knowledge of selected phonological, lexical and elopment of pragmatic competence - students can effectively use the language with focus on Academic English, level B2.
Word-formation - aff abstract Selected aspects of E	English d its specific features and nouns demic writing, writing a paragraph, word-order, topic sentences
M. McCarthy M., O Zemach, D.E, Rumis Olsen, A. : Active Vo www.bbclearningeng	ncounters, CUP, 2002 E English for Scientists, CUP 2011 Dell F Academic Vocabulary in Use, CUP 2008 ek, L.A: Academic Writing, Macmillan 2005 Icabulary, Pearson, 2013

Course langua English langua	ge: ge, level B2 accor	rding to CEFR.				
Notes:						
Course assessm Total number of	nent of assessed studen	ts: 416				
А	В	С	D	Е	FX	
36.54	36.54 21.63 15.14 9.38 6.01 11.3					
Provides: Mgr.	Viktória Mária S	lovenská		<u>.</u>	1	
Date of last mo	odification: 11.09	.2024				
Approved: pro	f. Mgr. Jaroslav H	lofierka, PhD., p	rof. RNDr. Vlad	imír Zeleňák, Dr	Sc.	

University: P. J. Ša	fárik Universi	ty in Košice			
Faculty: Faculty of	Science				
Course ID: KPE/ ALP/06	Course na	me: Alternative	Education		
Course type, scope Course type: Prac Recommended co Per week: 2 Per s Course method: p	tice urse-load (ho tudy period:	ours):			
Number of ECTS					
Recommended sen	nester/trimes	ter of the cours	e: 4.		
Course level: I.					
Prerequisities:					
Conditions for cou	rse completio	on:			
Learning outcome	s:				
Brief outline of the	course:				
Recommended lite	rature:				
Course language:					
Notes:					
Course assessment Total number of ass		s: 356			
A	В	С	D	Е	FX
67.42	25.28	4.21	0.56	0.28	2.25
Provides: Mgr. Kat	arína Petríkov	vá, PhD., Mgr. Z	uzana Vagaská, I	PhD.	
Date of last modifi	cation: 12.03	.2024			
Approved: prof. M	gr. Jaroslav H	ofierka, PhD. n	rof. RNDr. Vladi	mír Zeleňák. Dr	Sc.

University: P. J	. Šafárik Univer	sity in Košice			
Faculty: Facult	y of Science				
Course ID: ÚC ANCHU/21	HV/ Course n	ame: Analytical (Chemistry		
Recommende	Lecture / Practic d course-load (l l Per study per	e 1ours):			
Number of EC	FS credits: 5				
Recommended	semester/trime	ster of the course	e: 3.		
Course level: I.					
Prerequisities:					
2. Examination Learning outco	lytical calculation is composed of mes:	ons (each 33%, m 3 questions (each	for 33%, it is no		
in research and	1 1	asks of analytical	chemistry and a	ipplications of ana	alytical methods
treatment. Prep Classification of of organic analy Methods of qua Instrumental m	e of analytical ch aration of solution of analytical read ysis. Intitative analysi ethods of analyt	emistry. General pons. Evaluation of etions. Qualitative s. General princip ical chemistry (ba paration methods.	the results. analysis of cat les of gravimeti sic principles, in	ions and anions.	Basic principles alysis.
Recommended D.Harvey, Mod	literature: ern Analytical (Chemistry. McGraumental Analysis.	w Hill, Boston,		York 1985
Course languag	ge:				
Notes:					
Course assessm Total number of	ent f assessed stude	nts: 75			
А	В	C	D	E	FX
30.67	17.33	20.0	18.67	9.33	4.0
Provides: doc.]	RNDr. Taťána G	ondová, CSc.		1	
Date of last mo	dification: 12.1	1.2021			
Approved: prof	. Mgr. Jaroslav	Hofierka, PhD., p	rof. RNDr. Vlad	limír Zeleňák, Dr	Sc.
<u> </u>	<u> </u>	, , , , ,		,	

University: P. J. Šaf	árik Univers	ity in Košice			
Faculty: Faculty of	Science				
Course ID: ÚGE/ SPB1/21	Course na	me: Bachelor Th	nesis Project Ser	ninar 1	
Course type, scope Course type: Pract Recommended co Per week: 2 Per st Course method: p	tice urse-load (h tudy period:	ours):			
Number of ECTS c	credits: 3				
Recommended sem	ester/trimes	ster of the cours	e: 5.		
Course level: I.					
Prerequisities:					
Conditions for cou	rse completi	on:			
Learning outcomes	3:				
Brief outline of the	course:				
Recommended lite	rature:				
Course language:					
Notes:					
Course assessment Total number of ass	essed studen	ts: 36			
A	В	С	D	E	FX
86.11	8.33	5.56	0.0	0.0	0.0
Provides: prof. Mgr	. Jaroslav Ho	ofierka, PhD., doo	c. Mgr. Ladislav	Novotný, PhD.	
Date of last modifie	cation: 27.06	0.2022			
Approved: prof. Mg	gr. Jaroslav H	Iofierka, PhD., p	rof. RNDr. Vlad	imír Zeleňák, Dr	Sc.

University: P. J. Š	Safárik Univers	ity in Košice				
Faculty: Faculty	of Science					
Course ID: ÚGE/ SPB2/21						
Course type, scop Course type: Pra Recommended o Per week: 2 Per Course method:	actice course-load (h study period:	ours):				
Number of ECTS	S credits: 3					
Recommended se	emester/trimes	ter of the cours	e: 6.			
Course level: I.						
Prerequisities:						
Conditions for co	ourse completi	on:				
Learning outcom	ies:					
Brief outline of tl	he course:					
Recommended li	terature:					
Course language	:					
Notes:						
Course assessme Total number of a		ts: 32				
А	В	С	D	Е	FX	
68.75	25.0	6.25	0.0	0.0	0.0	
Provides: prof. M Onačillová, PhD.	lgr. Jaroslav Ho	ofierka, PhD., do	c. Mgr. Ladislav	Novotný, PhD.,	Mgr. Katarína	
Date of last modi	fication: 27.06	5.2022				
Approved: prof. 1	Mgr. Jaroslav H	Iofierka, PhD., p	rof. RNDr. Vlad	imír Zeleňák, Dr	Sc.	

University: P. J. Š	afárik Universi	ty in Košice			
Faculty: Faculty of	of Science				
Course ID: ÚGE/ BPO/14	Course na	me: Bachelor Th	nesis and its Def	ence	
Course type, scop Course type: Recommended o Per week: Per s Course method:	course-load (ho tudy period: present				
Number of ECTS					
Recommended se	emester/trimes	ter of the cours	2:		
Course level: I.					
Prerequisities:					
Conditions for co	ourse completio	on:			
Learning outcom	es:				
Brief outline of th	ne course:				
Recommended lit	terature:				
Course language:	:				
Notes:					
Course assessmen Total number of a		s: 209			
A	В	С	D	Е	FX
38.76	26.79	16.75	8.61	7.66	1.44
Provides:	<u>_</u>				1
Date of last modi	fication: 07.12	.2021			
Approved: prof. N	Mgr. Jaroslav H	ofierka, PhD., p	of. RNDr. Vlad	imír Zeleňák, Dr	Sc.

University: P. J.	Šafárik Universi	ty in Košice			
Faculty: Faculty	of Science				
Course ID: ÚCI BPO/21	HV/ Course na	me: Bachelor T	hesis and its Def	<i>ience</i>	
Course type:	• •				
Number of ECT	FS credits: 4				
Recommended	semester/trimes	ter of the cours	e:		
Course level: I.					
Prerequisities:					
Conditions for a	course completio	on:			
Learning outco	mes:				
Brief outline of Oral presentatio the state examin	on of the thesis re	sults. Answerin	g questions of th	he thesis oponent	t or members of
Recommended	literature:				
Course languag slovak	e:				
Notes:					
Course assessm Total number of	ent assessed student	s: 27			
А	В	С	D	E	FX
88.89	11.11	0.0	0.0	0.0	0.0
Provides:			1	•	
Date of last mod	dification: 07.12	.2021			

University: P. J. Šafá	rik University in Košice						
Faculty: Faculty of S	cience						
Course ID: ÚCHV/ BKPa/22							
Course type, scope a Course type: Recommended cou Per week: Per stud Course method: pre	rse-load (hours): ly period: esent						
Number of ECTS cr							
	ster/trimester of the cours	e: 5.					
Course level: I.							
Prerequisities:							
Conditions for cours	e completion:						
Learning outcomes:							
Brief outline of the c	ourse:						
Recommended litera	iture:						
Course language:							
Notes:							
Course assessment Total number of asse	ssed students: 27						
	abs	n					
100.0 0.0							
Provides:							
Date of last modifica	ition: 07.02.2022						
Approved: prof. Mgr	. Jaroslav Hofierka, PhD., p	rof. RNDr. Vladimír Zeleňák, DrSc.					

University: P. J. Šafá	rik University in Košice				
Faculty: Faculty of S	cience				
Course ID: ÚCHV/ BKPb/22	Course name: Bakalársky	projekt II			
Course type, scope a Course type: Recommended cou Per week: Per stud Course method: pre	rse-load (hours): ly period: esent				
Number of ECTS cr					
	ster/trimester of the cours	e: 6.			
Course level: I.					
Prerequisities:					
Conditions for cours	e completion:				
Learning outcomes:					
Brief outline of the c	ourse:				
Recommended litera	ature:				
Course language:					
Notes:					
Course assessment Total number of asse	ssed students: 25				
	abs	n			
	100.0 0.0				
Provides:					
Date of last modifica	ition: 07.02.2022				
Approved: prof. Mgi	. Jaroslav Hofierka, PhD., p	rof. RNDr. Vladimír Zeleňák, DrSc.			

University: P. J. Ša	afárik Univers	ity in Košice			
Faculty: Faculty o	f Science				
Course ID: ÚGE/ ZKAR/21	Course na	me: Basics of K	arstology and Sp	eleology	
Course type, scop Course type: Lec Recommended co Per week: 1 / 1 P Course method:	ture / Practice ourse-load (h er study perio	ours):			
Number of ECTS	credits: 3				
Recommended ser	mester/trimes	ster of the cours	e: 4.		
Course level: I., II					
Prerequisities:					
Conditions for cou	urse completi	on:			
Learning outcome	es:				
Brief outline of th	e course:				
Recommended lite	erature:				
Course language:					
Notes:					
Course assessmen Total number of as	-	ts: 18			
А	В	С	D	Е	FX
66.67	11.11	11.11	11.11	0.0	0.0
Provides: RNDr. A	Alena Gessert,	PhD., univerzitn	á docentka, doc.	Ing. Katarína Bó	nová, PhD.
Date of last modif	ication: 20.02	2.2023			
Approved: prof. N	Igr. Jaroslav H	Iofierka, PhD., p	rof. RNDr. Vladi	mír Zeleňák, Dr	Sc.

University: P.	J Šafárik	University in	Košice
University. 1.	J. Darank	Oniversity in	RUSICC

Faculty: Faculty of Science

Course ID: ÚCHV/	Course name: Basis of Mineralogy
MIN1/14	

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours):

Per week: 2 / 1 Per study period: 28 / 14

Course method: present

Number of ECTS credits: 5

Recommended semester/trimester of the course: 4.

Course level: I.

Prerequisities: ÚCHV/VCH/10 or ÚCHV/VCH/21 or ÚCHV/VCHU/10 or ÚCHV/ZAC2/10 or ÚCHV/VACH/10 or ÚCHV/CHG/09 or ÚCHV/ZCF/03 or ÚCHV/VCHU/15

Conditions for course completion:

Verification of theoretical knowledge and recognizing minerals.

A semester project about selected minerals (40 %), a practical test from recognizing of minerals (30 %), a written examination (30 %). The student must obtain totally at least 51%.

In a case of online education the practical test is canceled and the written examination contains more questions (60 %).

Learning outcomes:

To recognize the beauty of nature and to obtain basic knowledge from mineralogy. After completing the course, students will be familiar with the properties of commonly available minerals and will be able to recognize these minerals.

Brief outline of the course:

Basic terms and definitions, origin of minerals in nature. Basis of morphological and structural crystallography: characteristic properties of crystals, crystallographic laws, crystal structure, unit cells and their parameters, crystallographic systems with examples of minerals. Crystallochemistry: types of bonds and structures and their effect on the properties of minerals. Physical properties of minerals and their utilize in minerals classification. Basis of genetic and systematic mineralogy. Structure of silicates.

Recommended literature:

M. Košuth: Mineralógia. Elfa, s.r.o. Košice, 2001 V. Radzo: Mineralógia, Alfa Bratislava, 1987.

Course language:

Slovak

Notes:

Teaching is carried out in person or, if necessary, online using the MS Teams tool. The form of teaching is specified by the teacher at the beginning of the semester, updated continuously.

Course assessment Total number of assessed students: 149						
А	В	С	D	Е	FX	
81.88	16.11	0.67	0.67	0.0	0.67	
Provides: doc.	Provides: doc. RNDr. Ivan Potočňák, PhD.					
Date of last mo	Date of last modification: 21.07.2022					
Approved: prof	f. Mgr. Jaroslav H	Iofierka, PhD., p	rof. RNDr. Vladi	mír Zeleňák, Dr	Sc.	

University: P. J. Š	Safárik University in Košice
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Faculty: Faculty of Science

Course ID: ÚCHV/	Course name: Biochemistry
BCHU/21	

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 3 / 1 Per study period: 42 / 14 Course method: present

Number of ECTS credits: 5

Recommended semester/trimester of the course: 5.

Course level: I.

Prerequisities: ÚCHV/VCHU/10 or ÚCHV/VCHU/15 or ÚCHV/VACH/10 or ÚCHV/VCHU/14

Conditions for course completion:

Successful completion of the exam, which consists of two parts: (i) written and (ii) oral part. The student passes the exam if he / she obtains at least 60% of the points in the written part and at the same time adequately answers the asked questions in the oral part.

Learning outcomes:

Gain knowledge of: (i) the basic building blocks of biomacromolecules (proteins, DNA, RNA, fats and sugars) and their properties, (ii) the basic biochemical processes that take place in living organisms, (iii) the way energy is produced and used in cells.

Brief outline of the course:

1. Protein Structure and Function, Exploring proteins.

- 2. DNA and RNA and the Flow of Genetic Information, Exploring genes.
- 3. Enzymes: Basic Concepts and Kinetics, Catalytic Strategies and Regulatory Strategies.
- 4. Carbohydrates (Monosaccharides, Disaccharides, Polysaccharides Functions and Properties).
- 5. Lipids and Cells Membranes, Membrane Channels and Pumps.
- 6. Metabolis: Basic Concepts and Design, Signal-Transduction Pathways.
- 7. Glycolysis and Gluconeogenesis, Glycogen Metabolism.
- 8. The Citric Acid Cycle and Glyoxylate Cycle.
- 9. Oxidative Phosphorylation, The Light Reactions of Photosyntesis.
- 10. The Calvine Cycle and the Pentose Phosphate Pathway.
- 11. Fatty Acids Metabolism, Urea Cycle.
- 12. DNA Replication, Transcription (RNA Synthesis).
- 13. Protein Synthesis & Degradation, the Integration of Metabolism.

Recommended literature:

Course language:

Notes:

Course assessm Total number of	lent f assessed studen	ts: 85				
A B C D E FX						
28.24	12.94	14.12	22.35	20.0	2.35	
Provides: prof. RNDr. Erik Sedlák, DrSc., RNDr. Nataša Tomášková, PhD., prof. RNDr. Mária Kožurková, CSc., Mgr. Mária Tomková, PhD.						
Date of last mo	dification: 14.11	.2021				
Approved: prof	f. Mgr. Jaroslav H	Iofierka, PhD., p	rof. RNDr. Vladi	mír Zeleňák, DrS	Sc.	

	Šafárik Univers	sity in Rosiec			
Faculty: Faculty	of Science				
Course ID: ÚC BCH1b/10	HV/ Course na	ame: Biochemist	ry II		
Recommended	Lecture / Practice l course-load (h Per study peri	e ours):			
Number of EC	FS credits: 5				
Recommended	semester/trimes	ster of the cours	e: 6.		
Course level: I.					
Prerequisities:	ÚCHV/BCH1a/()3 or ÚCHV/BC	HU/21 or ÚCHV	/BCH1a/21	
Conditions for Test and oral ex		on:			
basis of their me Brief outline of	the course:	e information on	cell metabolism.	field of living or	
Recommended	literature: ehm K.H.: Color	atlas of biochen	nistry. Thieme, S e, chemický pohl	tuttgart, Germany ed na biologický	y, 2005.
Kodíček M., Va škola chemicko	-technologická v	Praze, Praha, 20)22.		svet, vysoka
Kodíček M., Va škola chemicko Course languag	-technologická v	Praze, Praha, 20)22.		svet, vysoka
Kodíček M., Va škola chemicko Course languag Notes: Course assessm	-technologická v)22.		
Kodíček M., Va škola chemicko Course languag Notes: Course assessm	-technologická v ge: ent)22. D	E	FX
Kodíček M., Va škola chemicko Course languag Notes: Course assessm Total number of	-technologická v ge: ent assessed studen	ts: 397	I	E 20.91	
Kodíček M., Va škola chemicko Course languag Notes: Course assessm Total number of A 9.82 Provides: prof. Rastislav Varhač	ent assessed studen B 19.14 RNDr. Mária Ko 5, PhD., doc. RN	ts: 397 C 31.49 žurková, CSc., p	D 17.63 prof. RNDr. Erik ský, PhD., RNDr		FX 1.01 oc. RNDr.

	TĂCUL	TT · · · TZ ··
University: P.	J. Safarik	University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/	Course name: Biochemistry Practical
PBCHU/15	

Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 4 Per study period: 56

Course method: present

Number of ECTS credits: 4

Recommended semester/trimester of the course: 6.

Course level: I.

Prerequisities: ÚCHV/BCHU/03 or ÚCHV/BCHU/21

Conditions for course completion:

Active participation with a maximum of one excused absence without the need for compensation. In case of excused absence from two or more practical exercises (e.g. due to illness), the student agrees with the teacher on alternative dates for practice.

Correctly prepared protocols from all completed tasks.

At least 51% of points from each of the written tests.

Learning outcomes:

To allow students to get practical experience in experimental techniques and methods, currently used in a biochemical research: UV/VIS spectrophotometry, thin layer chromatography (TLC), gel electrophoresis, isolation of macromolecules and substances from biological materials and their quantitative and qualitative determination.

Brief outline of the course:

1. Biochemistry laboratory safety rules. Basic biochemical laboratory procedures.

- 2. Qualitative tests for amino acids and proteins.
- 3. Isolation of casein from milk. Determination of protein concentration by Lowry method.

4. Determination of the iodine number by Yasud method . Soap production. Reactions with soap. Oxidation of unsaturated fatty acids.

5. Saponification number of fats and oils. Qualitative test for cholesterol: Salkowsky reaction.

6. Qualitative tests for carbohydrates. Determination of reducing carbohydrates by the Schoorl's method.

7. Determination of reducing and nonreducing carbohydrates in germinant plants.

8. Time-dependent course of enzyme-catalyzed reaction: digestion of gelatin by trypsine.

9. Determination of catalase activity and the first order rate constant. Effect of pH on alpha-amylase activity.

10. Effect of substrate concentration on initial rate of reaction, determination of Km and Vmax for urease-catalyzed hydrolysis of urea.

11. Isolation of DNA from spleen. Isolation of RNA from yeast. Qualitative tests for DNA and RNA components.

12. Determination of vitamin C concentration by 2,4-dinitrofenylhydrazine. Determination of vitamins A, B1, and C.

13. Final evaluation of students.

Recommended literature:

Sedlák, Varhač, Danko, Paulíková, Podhradský: Praktické cvičenia z biochémie, 2020, https://unibook.upjs.sk/sk/chemia/1411-prakticke-cvicenia-z-biochemie

Course language:

Slovak

Notes:

Teaching is carried out in person.

Course assessment

Total number of assessed students: 244

А	В	С	D	Е	FX
77.46	18.44	2.87	0.82	0.41	0.0

Provides: prof. RNDr. Mária Kožurková, CSc., RNDr. Nataša Tomášková, PhD., doc. RNDr. Rastislav Varhač, PhD., RNDr. Danica Sabolová, PhD., univerzitná docentka, RNDr. Eva Konkoľová, PhD.

Date of last modification: 19.11.2021

Course type, scope and the method: Course type; Lecture / Practice Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14 Course method: present Number of ECTS credits: 5 Recommended semester/trimester of the course: 5. Course level: 1, II. Prerequisities: Conditions for course completion: Test or seminar works examination Learning outcomes: The basic knowledges about biometal interactions with biomolecules, biomaterials, biominerals, biocatalysis, metals in biology and medicine, metal-based drugs, toxic metals for biosystems and metals in the environment. Brief outline of the course: Metalic and non-metalic elements and their roles in biological systems (biometals, bulk biological elements, essential trace clements). Biocordination compounds, bioligands. Biocatalyzers. Oxygen carriers and oxygen transport proteins. Photochemical process. Catalysis and regulation processes. Calcium biominerals and biomineralization. Toxic metals. Application of knowledge of bioinorganic chemistry in pharmacy, chemotherapy (e.g. platinum complexes in cancer therapy) radiodiagnostics, mineral biotechnology, ecology and in other branches of life. Recommended literature: 1. Shriver D. F., Atkins P. W., Overton T. L., Rourke J.P., Weller M.T., Amstrong F.A.: Shiver & Atkins. Inorganic Chemistry. Oxford University Press, Oxford 2006, 2. Kaim W., Schwederski B.: Bioinorganic Chemistry: Inorganic Elements in the Chemistry of Life. Wiley, Chichester 1998. 3. Wilkins P. C., Wilkins R. G.: Inorganic Chemistry in Biology. OCP, Oxford 1997. Course language:		CC	OURSE INFORM	MATION LET	TER			
Course ID: ÚCHV/ BAC1/04 Course name: Bioinorganic Chemistry I BAC1/04 Course type, scope and the method: Course type. Lecture / Practice Recommended course-load (hours): Per weck: 2 / 1 Per study period: 28 / 14 Course method: present Recommended course-load (hours): Per weck: 2 / 1 Per study period: 28 / 14 Course method: present Number of ECTS credits: 5 Recommended semester/trimester of the course: 5. Course level: 1, II. Prerequisities: Control of course completion: Test or seminar works examination Test or seminar works commended semester/trimester of the course; 5. Conditions for course completion: Test or seminar works examination Test or seminar works course inverses Test or seminar works course inverses The basic knowledges about biometal interactions with biomolecules, biomaterials, biominerals, biocatalysis, metals in biology and medicine, metal-based drugs, toxic metals for biosystems and metals in the environment. Brif outline of the course: Metalic and non-metalic elements and their roles in biological systems (biometals, bulk biological elements, essential trace elements). Biocoordination compounds, bioligands. Biocatalyzers. Oxygen carriers and oxygen transport proteins. Photochemical process. Catalysis and regulation processes. Catalymis and biomineralization. Toxic metals. Application of knowledge of bioinorganic chemistry in pharmacy, ehemotherapy (e.g. platinum complexes in cancer therapy) radiodiagnostics, mineral biotechnology, ecology and in other branches of life. Recommended literature: 1. Shriver D. F., Atkins P. W., Overton T. L., Rourke J.P., Weller M.T., Amstrong F.A.: S	University: P. J.	. Šafárik Univers	sity in Košice					
BAC1/04 Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14 Course method: present Number of ECTS credits: 5 Recommended semester/trimester of the course: 5. Course level: 1., II. Prerequisities: Continons for course completion: Test or seminar works examination Learning outcomes: The basic knowledges about biometal interactions with biomolecules, biomaterials, biominerals, biocatalysis, metals in biology and medicine, metal-based drugs, toxic metals for biosystems and metals in the environment. Brief outline of th course: Metalic and non-metalic elements and their roles in biological systems (biometals, bulk biological elements, essential trace clements). Biocoordination compounds, bioligands. Biocatalyzers. Oxygen carriers and oxygen transport proteins. Photochemical process. Catalysis and regulation processes. Calcium biominerals and biomineralization. Toxic metals. Application of knowledge of bioinorganic chemistry in pharmacy, chemotherapy (e.g. platinum complexes in cancer therapy) radiodiagnostics, mineral biotechnology, cology and in other branches of life. Recommended literature: 1. 1. Shriver D. F., Atkins P. W., Overton T. L., Rourke J.P., Weller M.T., Amstrong F.A.: Shiver & Atkins. Inorganic Chemistry. Oxford University Press, Oxford 2006. 2. Kaim W., Schwederski B.: Bioinorganic Chemistry in Biology. OCP, Oxford 1997. Course language: Notes:	Faculty: Faculty	y of Science						
Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14 Course method: present Number of ECTS credits: 5 Recommended semester/trimester of the course: 5. Course level: 1., II. Prerequisities: Conditions for course completion: Test or seminar works examination Learning outcomes: The basic knowledges about biometal interactions with biomolecules, biomaterials, biominerals, biocatalysis, metals in biology and medicine, metal-based drugs, toxic metals for biosystems and metals in the environment. Brief outline of the course: Metalic and non-metalic elements and their roles in biological systems (biometals, bulk biological elements, essential trace elements). Biocoordination compounds, bioligands. Biocatalyzers. Oxygen carriers and oxygen transport proteins. Photochemical process. Catalysis and regulation processes. Calcium biominerals and biomineralization. Toxic metals. Application of knowledge of bioinogranic chemistry in pharmacy, chemotherapy (e.g. platinum complexes in cancer therapy) radiodiagnostics, mineral biotechnology, ecology and in other branches of life. Recommendel literature: 1. Shriver D. F., Atkins P. W., Overton T. L., Rourke J.P., Weller M.T., Amstrong F.A.: Shiver & Atkins. Inorganic Chemistry. Oxford University Press, Oxford 2006. 2. Kaim W., Schwederski B.: Bioinorganic Chemistry: Inorganic Elements in the Chemistry of Life. Wiley, Chichester 1998.	Course ID: ÚC BAC1/04							
Recommended semester/trimester of the course: 5. Course level: 1., II. Prerequisities: Conditions for course completion: Test or seminar works examination Learning outcomes: The basic knowledges about biometal interactions with biomolecules, biomaterials, biominerals, biocatalysis, metals in biology and medicine, metal-based drugs, toxic metals for biosystems and metals in the environment. Brief outline of the course: Metalic and non-metalic elements and their roles in biological systems (biometals, bulk biological elements, essential trace elements). Biocoordination compounds, bioligands. Biocatalyzers. Oxygen carriers and oxygen transport proteins. Photochemical process. Catalysis and regulation processes. Calcium biominerals and biomineralization. Toxic metals. Application of knowledge of bioinorganic chemistry in pharmacy, chemotherapy (e.g. platinum complexes in cancer therapy) radiodiagnostics, mineral biotechnology, ecology and in other branches of life. Recommended literature: 1. Shriver D. F., Atkins P. W., Overton T. L., Rourke J.P., Weller M.T., Amstrong F.A.: Shiver & Atkins. Inorganic Chemistry. Oxford University Press, Oxford 2006. 2. Kaim W., Schwederski B.; Bioinorganic Chemistry: Inorganic Elements in the Chemistry of Life. Wiley, Chichester 1998. 3. Wilkins P. C., Wilkins R. G.: Inorganic Chemistry in Biology. OCP, Oxford 1997.	Course type: I Recommended Per week: 2 / 1	Lecture / Practice d course-load (h l Per study peri	e ours):					
Course level: 1., II. Prerequisities: Conditions for course completion: Test or seminar works examination Learning outcomes: The basic knowledges about biometal interactions with biomolecules, biomaterials, biominerals, biocatalysis, metals in biology and medicine, metal-based drugs, toxic metals for biosystems and metals in the environment. Brief outline of the course: Metalic and non-metalic elements and their roles in biological systems (biometals, bulk biological elements, essential trace elements). Biocoordination compounds, bioligands. Biocatalyzers. Oxygen carriers and oxygen transport proteins. Photochemical process. Catalysis and regulation processes. Calcium biominerals and biomineralization. Toxic metals. Application of knowledge of bioinorganic chemistry in pharmacy, chemotherapy (e.g. platinum complexes in cancer therapy) radiodiagnostics, mineral biotechnology, ecology and in other branches of life. Recommended literature: 1. Shriver D. F., Atkins P. W., Overton T. L., Rourke J.P., Weller M.T., Amstrong F.A.: Shiver & Atkins. Inorganic Chemistry. Oxford University Press, Oxford 2006. 2. Kaim W., Schwederski B.: Bioinorganic Chemistry: Inorganic Elements in the Chemistry of Life. Wiley, Chichester 1998. 3. Wilkins P. C., Wilkins R. G.: Inorganic Chemistry in Biology. OCP, Oxford 1997. Course assessment Total number of assessed students: 376 <	Number of EC	FS credits: 5						
Prerequisities: Conditions for course completion: Test or seminar works examination Learning outcomes: The basic knowledges about biometal interactions with biomolecules, biomaterials, biominerals, biocatalysis, metals in biology and medicine, metal-based drugs, toxic metals for biosystems and metals in the environment. Brief outline of the course: Metalic and non-metalic elements and their roles in biological systems (biometals, bulk biological elements, essential trace elements). Biocoordination compounds, bioligands. Biocatalyzers. Oxygen carriers and oxygen transport proteins. Photochemical process. Catalysis and regulation processes. Catcium biominerals and biomineralization. Toxic metals. Application of knowledge of bioinorganic chemistry in pharmacy, chemotherapy (e.g. platinum complexes in cancer therapy) radiodiagnostics, mineral biotechnology, ecology and in other branches of life. Recommended literature: 1. Shriver D. F., Atkins P. W., Overton T. L., Rourke J.P., Weller M.T., Amstrong F.A.: Shiver & Atkins. Inorganic Chemistry. Oxford University Press, Oxford 2006. 2. Kaim W., Schwederski B.: Bioinorganic Chemistry in Biology. OCP, Oxford 1997. Course language: Notes: Course language: A A A	Recommended	semester/trimes	ster of the cours	e: 5.				
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Test or seminar works examination Image: Construct on the construction of the co	Prerequisities:							
The basic knowledges about biometal interactions with biomolecules, biomaterials, biominerals, biocatalysis, metals in biology and medicine, metal-based drugs, toxic metals for biosystems and metals in the environment. Brief outline of the course: Metalic and non-metalic elements and their roles in biological systems (biometals, bulk biological elements, essential trace elements). Biocoordination compounds, bioligands. Biocatalyzers. Oxygen carriers and oxygen transport proteins. Photochemical process. Catalysis and regulation processes. Calcium biominerals and biomineralization. Toxic metals. Application of knowledge of bioinorganic chemistry in pharmacy, chemotherapy (e.g. platinum complexes in cancer therapy) radiodiagnostics, mineral biotechnology, ecology and in other branches of life. Recommended literature: 1. Shriver D. F., Atkins P. W., Overton T. L., Rourke J.P., Weller M.T., Amstrong F.A.: Shiver & Atkins. Inorganic Chemistry. Oxford University Press, Oxford 2006. 2. Kaim W., Schwederski B.: Bioinorganic Chemistry: Inorganic Elements in the Chemistry of Life. Wiley, Chichester 1998. 3. Wilkins P. C., Wilkins R. G.: Inorganic Chemistry in Biology. OCP, Oxford 1997. Course language: Notes: Course assessment Total number of assessed students: 376 A B C D E FX 42.02 27.39 19.15 5.85 5.32 0.27 <td>Test or seminar examination</td> <td>works</td> <td>ion:</td> <td></td> <td></td> <td></td>	Test or seminar examination	works	ion:					
Metalic and non-metalic elements and their roles in biological systems (biometals, bulk biological elements, essential trace elements). Biocoordination compounds, bioligands. Biocatalyzers. Oxygen carriers and oxygen transport proteins. Photochemical process. Catalysis and regulation processes. Calcium biominerals and biomineralization. Toxic metals. Application of knowledge of bioinorganic chemistry in pharmacy, chemotherapy (e.g. platinum complexes in cancer therapy) radiodiagnostics, mineral biotechnology, ecology and in other branches of life.Recommended literature: 1. Shriver D. F., Atkins P. W., Overton T. L., Rourke J.P., Weller M.T., Amstrong F.A.: Shiver & Atkins. Inorganic Chemistry. Oxford University Press, Oxford 2006. 2. Kaim W., Schwederski B.: Bioinorganic Chemistry: Inorganic Elements in the Chemistry of Life. Wiley, Chichester 1998. 3. Wilkins P. C., Wilkins R. G.: Inorganic Chemistry in Biology. OCP, Oxford 1997.Course language: Notes:Course assessment Total number of assessed students: 376ABCDEFX42.0227.3919.155.855.320.27	The basic know biocatalysis, me	vledges about bio etals in biology a			· ·	,		
1. Shriver D. F., Atkins P. W., Overton T. L., Rourke J.P., Weller M.T., Amstrong F.A.: Shiver & Atkins. Inorganic Chemistry. Oxford University Press, Oxford 2006. 2. Kaim W., Schwederski B.: Bioinorganic Chemistry: Inorganic Elements in the Chemistry of Life. Wiley, Chichester 1998. 3. Wilkins P. C., Wilkins R. G.: Inorganic Chemistry in Biology. OCP, Oxford 1997. Course language: Notes: Course assessment Total number of assessed students: 376 A B C D E FX 42.02 27.39 19.15 5.85 5.32 0.27	Metalic and nor elements, esser Oxygen carriers processes. Calc bioinorganic ch	n-metalic elemen ntial trace elem s and oxygen tra ium biominerals emistry in pharr	nents). Biocoord insport proteins. and biomineraliz nacy, chemother	ination compound Photochemical protocology vation.Toxic metapy (e.g. platinu	unds, bioligands. process. Catalysis als. Application o um complexes in o	Biocatalyzers. and regulation of knowledge of		
Notes: Course assessment Total number of assessed students: 376 B C D E FX 42.02 27.39 19.15 5.85 5.32 0.27	 Shriver D. F., Atkins. Inorgan Kaim W., Scl Life. Wiley, Chi 	, Atkins P. W., O ic Chemistry. Ox hwederski B.: Bi ichester 1998.	cford University oinorganic Chem	Press, Oxford 20 histry: Inorganic	006. Elements in the C	Chemistry of		
Course assessmentTotal number of assessed students: 376ABCDEFX42.0227.3919.155.855.320.27	Course languag	ge:						
Total number of assessed students: 376 A B C D E FX 42.02 27.39 19.15 5.85 5.32 0.27	Notes:							
42.02 27.39 19.15 5.85 5.32 0.27			its: 376					
	Α	В	C	D	E	FX		
Provides: prof. RNDr. Zuzana Vargová, Ph.D.	42.02	27.39	19.15	5.85	5.32	0.27		
	Provides: prof.	RNDr. Zuzana V	Vargová, Ph.D.					

Date of last modification: 28.10.2021

University: P. J.	Šafárik Univers	sity in Košice				
Faculty: Faculty	of Science					
Course ID: ÚBEV/Course name: Biology of Children and AdolescentsBDD/05						
Recommended	ecture / Practice course-load (h Per study peri	e iours):				
Number of ECT	S credits: 2					
Recommended	semester/trime	ster of the cour	se: 4., 6.			
Course level: I.						
Prerequisities:						
Conditions for o Written test	course complet	ion:				
systems of the h with developme of ontogenesis. Brief outline of Human ontogen	uman body with ntal and growth the course: nesis. Postnatal	a focus on the s characteristics a development.	ological knowled pecifics of childh nd with the most Age specific fea inary systems. F	ood and adolesce common disease tures of skeleta	ence. Familiarity es in these stages	
system. Nervou population and e		specifics of sele	cted diseases and	l drug dependend	ce arise. Human	
2000 Lipková V.: Son	ná M.: Biológia natický a fyziolo	ogický vývoj die	ciálnych pedagóg ťaťa. Osveta Brat ratislava, SPN, 1	tislava, 1980	ava, PdF UK,	
Course languag	e:					
Notes:						
Course assessm Total number of		nts: 1789				
А	В	C	D	E	FX	
31.25	24.04	18.28	16.71	9.11	0.61	
Provides: doc R	NDr Monika K	Lassayová, CSc.				
110/1405/ 400.1						
Date of last mod						

University: P. J. Šafárik University in Košice						
Faculty: Faculty of	Science					
Course ID: ÚGE/ KRT1/21	Course name: Cartography and Geoinformatics 1					
Course type: Lectu Recommended cou Per week: 2 / 2 Per	Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 2 Per study period: 28 / 28 Course method: present					
Number of ECTS c	Number of ECTS credits: 5					
Recommended semester/trimester of the course: 1.						
Course level: I.						
Prerequisities:						

Conditions for course completion:

During the semester, it is necessary to submit the results of the exercises. The acquired knowledge at the exercises will be verified by continuous examinations. The number of work outputs and written examinations will be announced at the beginning of the semester. It is possible to obtain 30% for meeting the evaluation criteria at the exercise (work outputs and written tests). The final evaluation of the exercises is determined by the instructor of the subject on the basis of completing the tasks in the exercises during the semester. The final evaluation of the course is based on a combination of meeting the evaluation conditions from the exercises and the final exam. A student who has met the conditions for passing the course at the seminars can apply for the final exam (70%). Credits will be awarded only to a student who achieves the final grade at least at the level of grade E. Credits will not be awarded to a student who does not meet the requirements of the exercises and the final exam is evaluated by FX. Rating scale: A (100-91%), B (81-90%,) C (71-80%), D (61-70%), E (51-60%).

Learning outcomes:

Knowledge: The student will gain theoretical knowledge in the field of cartography and geoinformatics. The student is able to understand cartographic and geoinformatics terminology, appropriately applies cartographic methods for displaying spatial information using a geographic information system, acquires a theoretical basis for the application of cartographic representations and coordinate systems and defines the composition of maps in GIS. The student acquires knowledge of the mathematical principles of mapping the Earth on a map and understands cartographic distortions, classification of cartographic representations, simple and false representations. The student acquires knowledge from the Slovak state map work (civil, military) and also acquires knowledge in cartographic expression methods (cartogram, cartodiagram) and the basics of cartometry.

Skills: The student will learn to acquire and work with the basics of the QGIS program, its control, purpose and structure. The student acquires basic orientations and work in the QGIS program, and work in the basic tools, setting layer properties and is capable of exporting data in different formats. The student understands cartographic representations in QGIS. The student acquires skills in working with paper maps, scale and measurements on maps, can orient in the field using a map, compass and can determine the azimuth. The student has skills in creating a point layer, has skills in

the principles of expressing point phenomena, the creation of a line layer as well as in the principles of expressing line phenomena, isolines. Student also has skills in creating a surface layer, in the principles of expressing surface phenomena. Controls the creation of map output, page settings, map export and output parameters settings. The student has skills in the composition of the map setting the compositional elements of the map and in creating the map output.

Competences: The student is able to work with a high degree of independence with geodata, to visualize them and create new layers, has all the prerequisites for independent creation of digital map output with available software support within GIS. The student is fully competent in the composition of the map - setting its compositional elements. When creating a map output, the student is able to independently or in cooperation in the relevant work team to communicate and collaborate with other experts, formulate opinions and recommendations in the creation and use of GIS in cartography.

Brief outline of the course:

Lectures: Cartography, basic concepts and position in the geosciences system. History and development of cartography. Geoinformatization cartography, digital cartography. Cartography and geoinformatics and their correlation. Geoinformatics, basic terms and definitions of GIS; online maps. Digital representation of objects and phenomena in GIS, vector and raster format. Principles of methodologies of cartographic modeling of geographical information in GIS. Design, use and evaluation of cartographic imaging properties in geoinformatics applications. Map - definition, map criteria, basic properties and elements of the map, categorization of maps, map scale. Principles of mapping the Earth, geoid, reference and display areas, global and local coordinate systems, the Earth and geographical lines and their importance for cartography and geoinformatics. Cartographic distortions, classification of cartographic representations, simple (azimuthal, conical, cylindrical) and false representations. Cartographic representations used in the Slovak state map work. Slovak state map work (civil, military), ZB-GIS, samples. Workflow for creating topographic maps, mapping, overview of 3D data collection in the field and used instrumentation. Map creation basics of map language, cartographic characters, map markers - point, line and area phenomena. Cartographic expression methods - cartogram, cartodiagram, classification and types of cartograms and cartodiagrams. Map composition, map content, map colors, map description, geographical nomenclature, map design. Basics of cartometry - positioning, measuring and determining distances, measuring and determining the size of surfaces, measuring oriented directions and angles, determining altitudes, determining the slope, profile construction, hypsometric curve. Classification of field formations. Thematic maps of various scales, applications, interpretation of maps. Maps on the Internet, map servers, Google Maps / Earth, Openstreetmaps. Office of Geodesy, Cartography and Cathars of the Slovak Republic - Geoportal.

Exercises: Basic introduction to ArcGIS, its purpose and control, program structure, data formats (* .mxd, * .shp), basic terminology - project, data layer - point, line, area, "features" and "graphics". Basic orientation in ArcMap, introduction of basic tools of the "Standard" and "Tools" packages, window "Table of contents", arrangement and properties of layers, tool "Select features" and "Data - Export Data". Defining a coordinate system, cartographic representations in ArcGIS. Introducing the options of the "Layer Properties" dialog box, working with the attribute table, working with files. Basic table editing, preparation and connection of databases (excel / shapefile) using the "Join" function. Working with paper maps, scale and measurement on maps. Orientation in the field using a map, compass, azimuth determination. Georeferencing. Point layer formation; principles of expressing linear phenomena in ArcGIS, isolines. Merge lines, Split lines. Formation; principles of expressing surface phenomena in ArcGIS, Polygon, Auto Complete Polygon, Cut Polygon Tools, Merge polygons. Cartogram, cartodiagram. Map output creation - Layout view, page settings, Map export and output

parameters settings. Map composition - setting the map composition elements and creating map output.

Recommended literature:

HOFIERKA, J., J. KAŇUK, M. GALLAY, 2014. Geoinformatika. Košice: Univerzita Pavla Jozefa Šafárika v Košiciach. ISBN 978-80-8152-178-2.

HOJOVEC, V. et al., 1987. Kartografie. Praha: Geodetický a kartografický podnik v Praze. ISBN 29-621-87.

LONGLEY, P.A., M. GOODCHILD, D. J. MAGUIRE, D. W. RHIND, 2010. Geographic Information Systems and Science. 3rd ed. Hoboken: Wiley & Sons, ISBN 978-0-470-72144-5. PRAVDA, J., D. KUSENDOVÁ, 2004. Počítačová tvorba tematických máp. Bratislava:

Univerzita Komenského v Bratislave. ISBN 80-223-2011-0.

ROBINSON, A. H. et al., 1995. Elements of Cartography. 6th ed. Hoboken: Wiley & Sons. ISBN 0-471-55579-7.

VOŽENÍLEK, V. et al., 2011. Metody tematické kartografie - Vizualizace prostorových jevů. Olomouc: Univerzita Palackého v Olomouci. ISBN 978-80-24427-90-4.

Course language:

Notes:

Course assessment

Total number of assessed students: 136

А	В	С	D	Е	FX
12.5	16.18	28.68	24.26	17.65	0.74

Provides: doc. RNDr. Ján Kaňuk, PhD., Mgr. Ondrej Tokarčík, PhD., Mgr. Michaela Nováková, PhD.

Date of last modification: 19.09.2023

University: P. J. Ša	fárik Univers	ity in Košice					
Faculty: Faculty of	Science						
Course ID: ÚGE/ KRT2/21	E/ Course name: Cartography and Geoinformatics 2						
Course type, scope Course type: Prac Recommended co Per week: 2 Per s Course method: p	ctice ourse-load (h tudy period:	ours):					
Number of ECTS	credits: 2						
Recommended sen	nester/trimes	ster of the cours	e: 2.				
Course level: I.							
Prerequisities:							
Conditions for cou	rse completi	on:					
Learning outcome	s:						
Brief outline of the	e course:						
Recommended lite	erature:						
Course language:							
Notes:							
Course assessment Total number of as		ts: 67					
A	В	С	D	Е	FX		
56.72	22.39	11.94	5.97	0.0	2.99		
Provides: Mgr. Ján	Šašak, PhD.,	doc. RNDr. Ján	Kaňuk, PhD., M	gr. Daniela Buch	alová		
Date of last modifi	cation: 27.06	5.2022					
Approved: prof. M	gr. Jaroslav H	Iofierka, PhD., p	of. RNDr. Vladi	mír Zeleňák, Dr	Sc.		

University: P.	J Šafárik	University in	Košice
University. 1.	J. Darank	Oniversity in	RUSICC

Faculty: Faculty of Science

Course ID: ÚCHV/	Course name: Chemical calculations
CHV1/99	

Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: present

Number of ECTS credits: 2

Recommended semester/trimester of the course: 1.

Course level: I.

Prerequisities:

Conditions for course completion:

Successful completion of two written tests in the middle and at the end of the semester. Accomplished test is with minimal 50% of point. The exact dates will be determined after mutual consultation between the teacher and the students.

The rating scale is determined as follows: A (100-91%), B (90-81%), C (80-71%), D (70-61%), E (60-51%), Fx (50- 0%).

Learning outcomes:

To teach students how to calculate material balances in the systems with or without chemical processes and how to calculate examples concerning the chemical equilibrium.

Brief outline of the course:

Expression of the clear matter amount and the system composition. Stoichiometric formula. Material bilances for preparation, dissolving and mixing of solutions, and for separating of mixtures. Material bilances for combined processes. Chemical equations and material bilances in the systems with chemical processes. Acid-Base equilibrium and the pH calculations. The solubility product and solubility.

Recommended literature:

Potočňák I.: Chemické výpočty vo všeobecnej a anorganickej chémii (skriptum), PF UPJŠ, Košice, 2017.

https://unibook.upjs.sk/sk/chemia/843-chemicke-vypocty-vo-vseobecnej-a-anorganickej-chemii Any chemical laboratory tables.

Course language:

SK - slovak

Notes:

The subject is carried out in person or, if necessary, remotely using the online platform Big Blue Button (BBB). The form of teaching is specified by the teacher at the beginning of the semester and updated continuously.

Course assessment Total number of assessed students: 1727						
A B C D E FX						
25.94	18.93	22.12	19.8	11.93	1.27	
Provides: RNDr. Martin Vavra, PhD., doc. RNDr. Miroslav Almáši, PhD., Mgr. Nikolas Király, PhD.						
Date of last modification: 15.11.2021						
Approved: prof. Mgr. Jaroslav Hofierka, PhD., prof. RNDr. Vladimír Zeleňák, DrSc.						

University: P. J.	Šafárik Univers	ity in Košice					
Faculty: Faculty	of Science						
Course ID: ÚCH SCHM/21	HV/ Course name: Chemistry						
Course type, sco Course type: Recommended Per week: Per Course method	course-load (he study period:						
Number of ECT	S credits: 2						
Recommended s	emester/trimes	ter of the cours	e:				
Course level: I.							
Prerequisities: ((BCHU/21 and (Ú FCHU/21 or ÚCI	CHV/ACHU/2						
Conditions for c	ourse completi	on:					
Learning outcom	nes:						
Brief outline of t	the course:						
Recommended l	iterature:						
Course language	e:						
Notes:							
Course assessme Total number of		ts: 81					
A	В	С	D	Е	FX		
12.35	25.93	23.46	16.05	17.28	4.94		
Provides:				<u>. </u>			
Date of last mod	ification: 08.09	.2021					
		lofierka, PhD., p					

Course ID:	Science
Course ID: KPPaPZ/ECo-C4/14	Course name: Communication ECo-C4
Course type, scope a Course type: Practi Recommended cou Per week: 2 Per stu Course method: pro	ce irse-load (hours): idy period: 28
Number of ECTS cr	redits: 4
Recommended seme	ester/trimester of the course: 3., 5.
Course level: I.	
Prerequisities:	
according to the teac	on in lessons (absence is allowed max. 90 min.), 2. Realization of assignment ther's instructions. In the electronic board of the course in AIS2. The teaching of the subject with
communication, rhet is able to use the ac communication with	tands theoretical information about the basics of verbal and nonverba- toric and methods of visualization and interprets them adequately. Studer cquired communication skills in practice, can apply effective principles of a others, is able to anticipate and thus prevent possible misunderstandings to the development of his social and professional skills.
heard", "Internal dial Active listening (The Misunderstandings (I Body language (Wha Signs of Physical Ex Active and Passive E Personality developm Rhetoric (History of reactions) Visualization - optica flipchart, Based on c	cation (Transmitter-receiver principle, "What is said is not equal to what is logue", The concept of communication) e most important criteria for active listening) How Misunderstandings Arise, How to Avoid Misunderstandings) at is body language, Active / passive body language, Dress psychology) xpression, Disadvantages of Fake Physical Expression, Difference Betwee Body Expression nent (Voices in us, "child in me" - identification of one's own personality) rhetoric, What is rhetoric, Vigor, alertness - assumptions, techniques, promp al display (Classic media - whiteboard, magnetic whiteboard, bulletin board computer technology - PC + Beamer)
Recommended liter	ature: 3. 2023. Nenásilná komunikácia. Aktuell. 234 s.

KOMÁRKOVÁ, Růžena - SLAMĚNÍK, Ivan - VÝROST, Jozef. Aplikovaná sociální psychologie III : Sociálněpsychologický výcvik. 1. vyd. Praha : Grada Publishing, 2001. 224 s. VÝROST, Jozef - SLAMĚNÍK, Ivan. Aplikovaná sociální psychologie II. 1. vyd. Praha : Grada Publishing, 2001. 260 s.

Course language:

slovak

Notes:

After passing the certification exams from all 4 modules (Teamwork, Selfmarketing, Conflict Management, Communication) the student will receive an ECo-C card and an ECo-C certificate.

Course assessment

Total number of assessed students: 169

abs	n
88.76	11.24

Provides: PhDr. Anna Janovská, PhD.

Date of last modification: 14.09.2024

Course type, scope and the method: Course type, scope and the method: Course type; Practice Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: present Number of ECTS credits: 2 Recommended semester/trimester of the course: Course level: I. Prerequisities: Conditions for course completion: Active participation in class and completed homework assignments. Students are allowed to miss two classes at the most. 2 credit tests (presumably in weeks 6/7 and 12/13) and an oral presentation in English. Final evaluation consists of the scores obtained for the 2 tests (50%) and the presentation (50%). Final grade will be calculated as follows: A 93-100 %, B 86-92%, C 79-85%, D 72-78%, E 65-71%, FX 64 % and less. Learning outcomes: Brief outline of the course: Recommended literature: www.bbclearningenglish.com Štěpánek, Libor a kol. Academic English-Akademická angličtina. Praha: Grada Publishing, a.s., 2011. McCarthy M., O'Dell F.: English Vocabulary in Use, Upper-Intermediate. CUP, 1994. Fictumova J., Ceccarelli J., Long T.: Angličtina, konverzace pro pokročilé. Barrister and Principal, 2008. Peters S., Gráf T.: Time to practise. Polyglot, 2007. Jones L.: Communicative Grammar Practice. CUP, 1985. Additional study materials. Course language: English language, B2-C1 level according to CEFR						
Course ID: CJP/ PFAJKKA/07 Course name: Communicative Competence in English PFAJKKA/07 Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: present Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: present Number of ECTS credits: 2 Recommended semester/trimester of the course: Course level: 1. Prerequisities: Conditions for course completion: Active participation in class and completed homework assignments. Students are allowed to miss two classes at the most. 2 credit tests (presumably in weeks 6/7 and 12/13) and an oral presentation in English. Final evaluation consists of the scores obtained for the 2 tests (50%) and the presentation (50%). Final grade will be calculated as follows: A 93-100 %, B 86-92%, C 79-85%, D 72-78%, E 65-71%, FX 64 % and less. Learning outcomes: Brief outline of the course: Recommended literature: www.bbclearningenglish.com Štēpánek, Libor a kol. Academic English-Akademická angličtina. Praha: Grada Publishing, a.s., 2011. McCarthy M., O'Dell F: English Vocabulary in Use, Upper-Intermediate, CUP, 1994. Fictumova J., Ceccarelli J., Long T: Angličtina, konverzace pro pokročilé. Barrister and Principal, 2008. Peters S., Graff T: Time to practise. Polyglot, 2007. Jones L.: Communicative Grammar Practice. CUP, 1985. Additional study materials. Course assessment Total number of assessed students: 301 E FX 45.18 20.93 17.61 7.64 5.98 2.66	University: P. J.	Šafárik Univers	ity in Košice			
PFAJKKA/07 Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: present Number of ECTS credits: 2 Recommended semester/trimester of the course: Course level: I. Prerequisities: Confittions for course completion: Active participation in class and completed homework assignments. Students are allowed to miss two classes at the most. 2 credit tests (presumably in weeks 6/7 and 12/13) and an oral presentation in English. Final grade will be calculated as follows: A 93-100 %, B 86-92%, C 79-85%, D 72-78%, E 65-71%, FX 64 % and less. Learning outcomes: Brief outline of the course: Recommended literature: www.bbclearningenglish.com Štěpánek, Libor a kol. Academic English-Akademická angličtina. Praha: Grada Publishing, a.s., 2011. McCarthy M., O'Dell F.: English Vocabulary in Use, Upper-Intermediate. CUP, 1994. Frietinova J., Cecearelli J., Long T.: Angličtina, konverzace pro pokročilé. Barrister and Principal, 2008. Peters S., Gráf T.: Time to practise. Polyglot, 2007. Jones L.: Communicative Grammar Practice. CUP, 1985. Additional study materials. Course assessment Total number of assessed students: 301 A B C	Faculty: Faculty	of Science				
Course type: Practice Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: present Number of ECTS credits: 2 Recommended semester/trimester of the course: Course level: I. Prerequisities: Conditions for course completion: Active participation in class and completed homework assignments. Students are allowed to miss two classes at the most. 2 credit tests (presumably in weeks 6/7 and 12/13) and an oral presentation in English. Final evaluation consists of the scores obtained for the 2 tests (50%) and the presentation (50%). Final evaluation consists of the scores obtained for the 2 tests (50%) and the presentation (50%). Final grade will be calculated as follows: A 93-100 %, B 86-92%, C 79-85%, D 72-78%, E 65-71%, FX 64 % and less. Learning outcomes: Recommended literature: www.bbclearningenglish.com Stepánek, Libor a kol. Academic English-Akademická angličtina. Praha: Grada Publishing, a.s., 2011. Meters S, Graft T: Time to practise. Polyglot, 2007. Jones L: Communicative Grammar Practice. CUP, 1985. Additional study materials. Course a	Course ID: CJP PFAJKKA/07	/ Course na	ame: Communic	ative Competend	e in English	
Recommended semester/trimester of the course: Course level: I. Prerequisities: Conditions for course completion: Active participation in class and completed homework assignments. Students are allowed to miss two classes at the most. 2 credit tests (presumably in weeks 6/7 and 12/13) and an oral presentation in English. Final evaluation consists of the scores obtained for the 2 tests (50%) and the presentation (50%). Final evaluation consists of the scores obtained for the 2 tests (50%) and the presentation (50%). Final evaluation consists of the scores obtained for the 2 tests (50%) and the presentation (50%). Final evaluation consists of the scores obtained for the 2 tests (50%) and the presentation (50%). Final evaluation consists of the scores obtained for the 2 tests (50%) and the presentation (50%). Final evaluation consists of the scores obtained for the 2 tests (50%) and the presentation (50%). Final evaluation consists of the scores obtained for the 2 tests (50%) and the presentation (50%). Final evaluation consists of the scores obtained for the 2 tests (50%) and the presentation (50%). Final evaluation consists of the scores obtained for the 2 tests (50%) and the presentation (50%). Final evaluation for the course: Recommended literature	Course type: F Recommended Per week: 2 Pe	Practice I course-load (h er study period:	ours):			
Course level: 1. Prerequisities: Conditions for course completion: Active participation in class and completed homework assignments. Students are allowed to miss two classes at the most. 2 credit tests (presumably in weeks 6/7 and 12/13) and an oral presentation in English. Final grade will be calculated as follows: A 93-100 %, B 86-92%, C 79-85%, D 72-78%, E 65-71%, FX 64 % and less. Learning outcomes: Brief outline of the course: Recommended literature: www.bbclearningenglish.com Štěpánek, Libor a kol. Academic English-Akademická angličtina. Praha: Grada Publishing, a.s., 2011. McCarthy M., O'Dell F.: English Vocabulary in Use, Upper-Intermediate. CUP, 1994. Fictumova J., Ceccarelli J., Long T.: Angličtina, konverzace pro pokročilé. Barrister and Principal, 2008. Peters S., Gráf T.: Time to practise. Polyglot, 2007. Jones L.: Communicative Grammar Practice. CUP, 1985. Additional study materials. Course language: English language, B2-C1 level according to CEFR Notes: C D E FX A B C D E FX A B C D E FX A B C D E FX						

Date of last modification: 11.02.2024

	Faculty: Faculty of Science	
Course ID: CJP/ PFAJGA/07	Course name: Communicative Grammar in English	
Course type, scope a Course type: Practic Recommended cou Per week: 2 Per stu Course method: pre	ce rse-load (hours): dy period: 28	
Number of ECTS credits: 2 Recommended semester/trimester of the course:		
		Course level: I.
Prerequisities:		
by given deadlines. Powerpoint presentat Final Test - end of se Final assessment = a	verage of test and presentation.	
Learning outcomes: The development of		
Learning outcomes: The development of a of their communic phonological, lexical	100%, B 86-92%, C 79-85%, D 72-78%, E 65-71%, FX 64% and less students' language skills - reading, writing, listening, speaking, improvement ative linguistic competence. Students acquire knowledge of selected and syntactic aspects, development of pragmatic competence. Students can aguage for a given purpose, with focus on Academic English and English on	
Learning outcomes: The development of a of their communic phonological, lexical efectively use the lar level B2. Brief outline of the c Selected aspects of E Word formation Contrast of tenses in The passive voice Types of Conditional Phrasal verbs and En	students' language skills - reading, writing, listening, speaking, improvement ative linguistic competence. Students acquire knowledge of selected and syntactic aspects, development of pragmatic competence. Students can aguage for a given purpose, with focus on Academic English and English on ourse: nglish grammar and pronunciation English	

English language, level B2 according to CEFR.

Notes

Notes:					
Course assessm Total number o	nent f assessed studen	ts: 446			
А	В	С	D	Е	FX
41.48	19.51	15.7	7.85	5.61	9.87
Provides: Mgr.	Viktória Mária S	lovenská, Mgr. I	ýdia Markovičov	vá, PhD.	·
Date of last mo	dification: 20.09	0.2023			
Approved: prof	f. Mgr. Jaroslav H	Iofierka, PhD., p	rof. RNDr. Vladi	mír Zeleňák, Dr	Sc.

University: P. J. Šafá	rik University in Košice			
Faculty: Faculty of S	cience			
Course ID: KGER/ NJKG/07				
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	ce rse-load (hours): dy period: 28			
Number of ECTS cr	edits: 2			

Recommended semester/trimester of the course:

Course level: I.

Prerequisities:

Conditions for course completion:

Active participation in class and completed homework assignments. Students are allowed to miss 2 classes at the most (2x90 min.). 2 control tests during the semester. Final grade will be calculated as follows: A 93-100 %, B 86-92%, C 79-85%, D 72-78%, E 65-71%, FX 64 % and less.

Learning outcomes:

The aim of the course is to identify and eliminate the most frequent grammatical errors in oral and written communication, learning language skills of listening comprehension, speaking, reading and writing, increasing students 'language competence (acquisition of selected phonological, lexical and syntactic knowledge), development of students' pragmatic competence (acquisition of the ability to express selected language functions), development of presentation skills, etc.

Brief outline of the course:

The course is aimed at practicing and consolidating knowledge of morphology and syntax of German in order to show the context in grammar as a whole. The course is intended for students who often make grammatical errors in oral as well as written communication. Through the analysis of texts, audio recordings, tests, grammar exercises, monologic and dialogical expressions of students focused on specific grammatical structures, problematic cases are solved individually and in groups. Emphasis is placed on the balanced development of grammatical thinking in the communication process, which ultimately contributes to the development of all four language skills.

Recommended literature:

Dreyer, H. – Schmitt, R.: Lehr- und Übungsbuch der deutschen Grammatik. Hueber Verlag GmbH & Co. Ismaning, 2009.

Krüger, M.: Motive Kursbuch, Lektion 1 – 30. Huebert Verlag GmbH & Co. Ismaning, 2020. Brill, L.M. – Techmer, M.: Deutsch. Großes Übungsbuch. Wortschatz. Huebert Verlag GmbH & Co. Ismaning, 2011.

Földeak, Hans: Sag's besser!. Grammatik. Arbeitsbuch für Fortgeschrittene. Huebert Verlag GmbH & Co. Ismaning, 2001.

Geiger, S. – Dinsel, S.: Deutsch Übungsbuch Grammatik A2-B2. Huebert Verlag GmbH & Co. Ismaning, 2018.

Dittelová, E. – Zavatčanová, M.: Einführung in das Studium der deutschen Fachsprache. Košice: ES UPJŠ, 2000.

Course languag German, Sloval	0				
Notes:					
Course assessm Total number o	nent f assessed studen	ts: 57			
А	В	С	D	Е	FX
61.4	10.53	8.77	3.51	8.77	7.02
Provides: Mgr.	Ulrika Strömplov	vá, PhD.			
Date of last mo	dification: 13.08	.2024			
Approved: prof	f. Mgr. Jaroslav H	lofierka, PhD., p	orof. RNDr. Vladi	mír Zeleňák, Dr	Sc.

University: P. J. Šafán	rik University in Košice
Faculty: Faculty of So	cience
Course ID: KPPaPZ/ECo-C3/14	Course name: Conflict Management ECo-C3
Course type, scope an Course type: Practic Recommended cour Per week: 2 Per stue Course method: pre	ce rse-load (hours): dy period: 28
Number of ECTS cro	edits: 4
Recommended seme	ster/trimester of the course: 3., 5.
Course level: I.	
Prerequisities:	
My strengths and we students will describe the form of deconstru Attendance at semina The evaluation of the set requirements, whi ensure an objective at	reflection on the selected topic within the specified time. Reflection topic: aknesses in conflict management. In a short presentation of their reflection, e their strengths and weaknesses in the management of conflict situations in action. rs is mandatory - the student may have two absences during the semester. course and its subsequent completion will be based on clearly and objectively ch will be set in advance and will not change. The aim of the assessment is to nd fair mapping of the student's knowledge while adhering to all ethical and re is no tolerance for students' fraudulent behavior, whether in the teaching
of basic rules. The method of teachi students' needs, expect respect and feedback The content of the cur topicality of the topics the connection of the c in lectures and semina The student is able to situations. The stude competencies as well The student is able to situations.	ad demonstration of knowledge in the field of conflict management and control ng the subject will be oriented to the student. Lecturers will be interested in etations and opinions so as to encourage them to think critically by expressing on their opinions and needs. riculum will be based on primary and high-quality sources that will reflect the s so as to ensure the connection of the curriculum with other subjects and also curriculum with practice. Students will be expected to take an active approach ars with an emphasis on their independence and responsibility. demonstrate an understanding of an individual's behavior in various conflict nt is able to describe, explain and evaluate their own internal resources, as limitations and weaknesses that are directly related to conflict management. apply theoretical knowledge and principles of conflict resolution to everyday
of disputes), Dispute	ourse: auses (Types of disputes, External influences, Be able to reveal the causes origin (Levels of disputes, Escalation warning signals, Escalation removal w to explain escalation stages; How do I approach a dispute?) Dispute

Resolution, Dispute Resolution Strategies, Dispute Discussion, Dispute Settlement Initiatives, Knowing how to handle a dispute and how to effectively resolve it), Dispute Resolution (Options, Public Struggle, Covert Struggle, Indefinite Postponement, Agreement, "Fair play", compromise, cooperation, capitulation, escape or separation), Prevention (Structures that produce disputes, The meaning and purpose of disputes, Stages and steps of dispute resolution, What does a positive corporate culture mean? Dispute is an incentive for change)

n

5.44

Recommended literature:

Course language:

Notes:

Course assessment

Total number of assessed students: 147

abs 94.56

Provides: Mgr. Ondrej Kalina, PhD.

Date of last modification: 12.09.2024

University: P. J. Šafărik University in Košice Faculty: Faculty of Science Course ID: ÚCHV/ KCHU/03 Course name: Coordination Chemistry Course type, scope and the method: Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14 Course method: present Number of ECTS credits: 4 Recommended semester/trimester of the course: 5. Course level: I. Prerequisities: ÚCHV/ACHU/21 Conditions for course completion: Final written exam Learning outcomes: The student acquires basic knowledge on the coordination compounds, preparation, isomerisi
Course ID: ÚCHV/ KCHU/03 Course name: Coordination Chemistry Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14 Course method: present Number of ECTS credits: 4 Recommended semester/trimester of the course: 5. Course level: I. Prerequisities: ÚCHV/ACHU/21 Conditions for course completion: Final written exam Learning outcomes: The student acquires basic knowledge on the coordination compounds, preparation, isomeristication
KCHU/03 Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14 Course method: present Number of ECTS credits: 4 Recommended semester/trimester of the course: 5. Course level: I. Prerequisities: ÚCHV/ACHU/21 Conditions for course completion: Final written exam Learning outcomes: The student acquires basic knowledge on the coordination compounds, preparation, isomerist
Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14 Course method: present Number of ECTS credits: 4 Recommended semester/trimester of the course: 5. Course level: I. Prerequisities: ÚCHV/ACHU/21 Conditions for course completion: Final written exam Learning outcomes: The student acquires basic knowledge on the coordination compounds, preparation, isomerist
Recommended semester/trimester of the course: 5. Course level: I. Prerequisities: ÚCHV/ACHU/21 Conditions for course completion: Final written exam Learning outcomes: The student acquires basic knowledge on the coordination compounds, preparation, isomerist
Course level: I. Prerequisities: ÚCHV/ACHU/21 Conditions for course completion: Final written exam Learning outcomes: The student acquires basic knowledge on the coordination compounds, preparation, isomerism
Prerequisities: ÚCHV/ACHU/21 Conditions for course completion: Final written exam Learning outcomes: The student acquires basic knowledge on the coordination compounds, preparation, isomerist
Conditions for course completion: Final written exam Learning outcomes: The student acquires basic knowledge on the coordination compounds, preparation, isomerism
Final written exam Learning outcomes: The student acquires basic knowledge on the coordination compounds, preparation, isomerism
The student acquires basic knowledge on the coordination compounds, preparation, isomerism
and properties of coordination compounds as well as about the chemical bonding in coordinatio compounds.
 Brief outline of the course: 1. Definition and nomenclature of coordination compounds. 2. Central atom and ligands 3. Coordination numbers, coordination polyhedra. 4. Isomerism of coordination compounds 5. Preparation of coordination compounds 6. Stability of coordination compounds 7. Chemical bonding in coordination compounds.
Recommended literature: J. Ribas: Coordination Chemistry, Wiley-VCH, Weinheim, 2008. J. C. Huheey, E. A. Keiter, R. L. Keiter: Inorganic Chemistry, Haper Collins, New York, 1993. G. A. Lawrance: Introduction to Coordination Chemistry, Wiley, 2010.
Course language:
Notes:
Course assessment Total number of assessed students: 98
A B C D E FX
40.82 25.51 13.27 7.14 11.22 2.04
Provides: prof. RNDr. Juraj Černák, DrSc., doc. RNDr. Juraj Kuchár, PhD.
Date of last modification: 10.09.2021

University: P. J. Ša	fárik Univers	ity in Košice			
Faculty: Faculty of	Science				
Course ID: ÚGE/ KULG/21	Course name: Cultural Geography				
Course type, scope Course type: Lect Recommended co Per week: 2 / 1 Pe Course method: p	ure / Practice urse-load (h er study perio	ours):			
Number of ECTS	credits: 4				
Recommended sen	nester/trimes	ster of the course	e: 5.	_	
Course level: I.					
Prerequisities:					
Conditions for cou	rse completi	on:			
Learning outcome	S:				
Brief outline of the	course:				
Recommended lite	rature:				
Course language:					
Notes:					
Course assessment Total number of ass		ts: 31			
A	В	С	D	Е	FX
64.52	12.9	19.35	3.23	0.0	0.0
Provides: Mgr. Mar	rián Kulla, Pł	nD., prof. Mgr. Ja	roslav Hofierka,	PhD.	1
Date of last modifi	cation: 27.06	5.2022			
Approved: prof. M	gr. Jaroslav H	Iofierka, PhD., pr	of. RNDr. Vladi	mír Zeleňák, Dr	Sc.

University: P. J. Šafá	rik University in Košice					
Faculty: Faculty of S	cience					
Course ID: ÚGE/ DTG/21						
Course type, scope a Course type: Practi Recommended cou Per week: 2 Per stu Course method: pre	ce rse-load (hours): Idy period: 28					
Number of ECTS cr	edits: 2					
Recommended seme	ester/trimester of the course: 1.					
Course level: I.						
Prerequisities:						
the semester. The ov evaluation. The evaluation	se completion: on a combination of midterm (30%) and final assessment (70%) at the end of erall evaluation is calculated as a weighted average of the final and midterm uation scheme applies to the overall evaluation: A (100-90 points), B (80-89 nts), D (60-69 points), E (50-59 points), FX (0 -49 points).					
technologies specific for and sort different professional literatur Skills: The student use databases of scie modifying different t acquainted with the l knowledge of using 0 Competences: The st of geography. The res	dent will gain knowledge in the field of information and communication to the study of geography and geoinformatics. The student will learn to search types of information. The acquired knowledge will be used in working with e published in scientific databases and selected geospatial databases. will learn to work with selected WebGIS portals publishing geodata and entific journals and citation manager. They will learn the basic methods of types of data in order to prepare them for integration into GIS. They will get icense conditions of the used software within the department. Gain advanced Office. udent will acquire basic competencies in the field of ICT needed for the study sult is the student's ability to manage the study fluently and smoothly in terms student is able to independently use ICT tools.					
university for stude operating systems, da SR, Soil portal, ŠGÚ the essence of vector databases (formulas,	course: I information regarding the study, standards and services provided by the nts (WiFi, information retrieval, websites, citation manager - CitacePro) ata types, file types, software used. Work with statistical data, DataCube, SO DŠ, Geoenviroportal, Geoportal and similar web applications. Explanation of and raster graphics, graphic formats and their use. Work with spreadsheet and contingency tables and graphs), advanced work and formatting in MS Word. nt to create presentations and posters.					
	ature: riestorové analýzy a modelovanie. Vysokoškolské učebné texty. Ita Univerzity Pavla Jozefa Šafárika v Košiciach. 114 s.					

ŽITNIAK, J., 2017. Microsoft Office 2016. Podrobná uživatelská příručka. Computer Press. 464

s.

KLATKOVSKÝ, K., 2016. Word 2016 nejen pro školy. Computer Media. 124 s.

KLATKOVSKÝ, K., 2016. Powerpoint 2016 nejen pro školy. Computer Media. 80 s.

LAURENČÍK, M., 2019. Excel 2016 a 2019 - pokročilé nástroje, Grada, 256 s.

Course languag	ge:				
Notes:					
Course assessm Total number of	ent f assessed studen	ts: 138			
А	В	С	D	Е	FX
50.72	27.54	13.77	4.35	1.45	2.17
Provides: doc. RNDr. Ján Kaňuk, PhD., Mgr. Daniela Buchalová					
Date of last modification: 27.06.2022					
Approved: prof	f. Mgr. Jaroslav H	Iofierka, PhD., p	rof. RNDr. Vladi	mír Zeleňák, Dr	Sc.

	rik University in Košice
Faculty: Faculty of S	cience
Course ID: KPPaPZ/PUDB/15	Course name: Drug Addiction Prevention in University Students
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	ce rse-load (hours): dy period: 28
Number of ECTS cr	edits: 2
Recommended seme	ster/trimester of the course: 3., 5.
Course level: I.	
Prerequisities:	
participation in works 50 - 45: A; 44 - 40:	te completion: active participation in the training part (30p). 2nd part of the evaluation: active shops (20p). In total, students can get 50p and the final evaluation is as follows B; 39-35: C; 34-30: D; 29 - 25: E 24 and less: FX. Detailed information in a board of the course in AIS2. The teaching of the subject will be realized by
describe and explain substance use. Studen of substance and non- The student is also a approaches in preven The student is able to	ands the principals of research data based prevention of risk behavior, can the determinants of risk behavior as well as protective and risk factors fo at understands and adequately interprets the theory explaining the background substance addictions. The to state and classify the types and forms of prevention, strategies and tion, can distinguish effective strategies from ineffective ones. To adequately interpret their experience with preventive activities in the group itive effect as well as limitations and threats.
Brief outline of the c	ourse:
internetu v školskej p Sloboda, Z., & Bukos and Practice. New Yo	012). Základy prevencie užívania drog a problematického používania oraxi. Košice: UPJŠ. ski, J. (Eds.). (2006). Handbook of Drug Abuse Prevention: Theory, Science
Course languages	
Course language: slovak	

Course assessment Total number of assessed students: 620						
А	В	С	D	Е	FX	
78.55	15.81	15.81 3.71 1.45 0.16 0.32				
Provides: prof. PhDr. Oľga Orosová, CSc., Mgr. Viera Čurová, PhD., Mgr. Janka Liptáková, PhDr. Anna Janovská, PhD., Mgr. Zuzana Michalove						
Date of last modification: 24.06.2022						
Approved: prof	Approved: prof. Mgr. Jaroslav Hofierka, PhD., prof. RNDr. Vladimír Zeleňák, DrSc.					

University: P. J. Šaf	árik Univers	ity in Košice			
Faculty: Faculty of	Science				
Course ID: ÚGE/ EKG/21	Course name: Economic geography				
Course type, scope Course type: Lect Recommended co Per week: 3 / 1 Pe Course method: p	ure / Practice urse-load (h r study perio	ours):			
Number of ECTS c	redits: 6				
Recommended sem	ester/trimes	ster of the cours	e: 3.		
Course level: I.					
Prerequisities:					
Conditions for cou	rse completi	on:			
Learning outcomes	:				
Brief outline of the	course:				
Recommended lite	rature:				
Course language:					
Notes:					
Course assessment Total number of ass	essed studen	ts: 77			
A	В	С	D	Е	FX
9.09	11.69	20.78	25.97	29.87	2.6
Provides: Mgr. Mar	ián Kulla, Pł	nD., doc. Mgr. La	dislav Novotný,	PhD., Mgr. Niko	la Svetozarov
Date of last modifie	cation: 27.06	5.2022			
Approved: prof. Mg	gr. Jaroslav H	Iofierka, PhD., p	of. RNDr. Vlad	imír Zeleňák, DrS	bc.

Faculty: Faculty of S	cience
Course ID: ÚINF/ EDS/15	Course name: Educational software
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	ce rse-load (hours): dy period: 28
Number of ECTS cr	edits: 2
Recommended seme	ster/trimester of the course: 5.
Course level: I.	
Prerequisities:	
 3. Creation of an inte 4. Creation of an inst Conditions for the fir Creation and presenta Conditions for succes Obtaining at least 50% Learning outcomes: Students will receive a) presentation software conceptual maps, b) programs for the c c) simulation and mod d) selected subject-or Students present and resources and tools in 	ng evaluation: sheet for student. imedia educational game. ractive educational quiz. ructional educational video. hal evaluation: ation of final project on the use of educational software in education. still completion of the course: % of points for ongoing and final assignments. , resp. deepen their basic skills in working with: are, programs for creating and editing images, animations, diagrams, sounds, reation of didactic tests, questionnaires, surveys, deling software, iented educational programs, discuss their idea of the use of educational software and educational Internet in the selected school subject.
 Creating and proce Creation and use of textbooks and workbe Creation of instruct Electronic voting and 	tional software and educational web resources and tools. essing of materials for teaching aid . f electronic and interactive educational documents (worksheets, presentations, ooks). tional educational video. and questionnaire creation. e tests and educational games. Gamification elements, tools and environments applications. ation tools.

10. Online educational platforms, repositories, projects and competitions.

11. Simulations and modelling. Subject-focused educational programmes.

12. Use digital tools to plan, monitor, differentiate and personalise learning. Accessibility of digital tools and learning resources.

Recommended literature:

SOLOMON, Gwen and Lynne SCHRUM, 2014. Web 2.0 How-to for Educators. Second. International Society for Technology in Education, 314 p. ISBN 978-1564843517.

STOBAUGH, Rebecca, 2019. Fifty Strategies to Boost Cognitive Engagement: Creating a Thinking Culture in the Classroom (50 Teaching Strategies to Support Cognitive Development). Solution Tree Press, 176 p. ISBN 978-1947604773.

LEMOV, Doug, 2015. Teach Like a Champion 2. 0: 62 Techniques That Put Students on the Path to College [online]. 2nd edition. John Wiley & Sons, Incorporated, 509 p. [cited 2021-7-10]. ISBN 9781118898628. Available from: https://ebookcentral.proquest.com/lib/upjs-ebooks/ detail.action?docID=1895720

European Schoolnet: Transforming education in Europe [online]. [cited 2021-7-10]. Available from: http://www.eun.org/home

Science On Stage Europe [online]. Science on Stage Europe e.V. [cited 2021-7-10]. Available from: https://www.science-on-stage.eu/

Course language:

Slovak and partly English due to selected programs and information sources

Notes:

By default, teaching is carried out face to face. If this is not possible (eg due to a pandemic), teaching is provided at a distance through video conferencing programs and LMS.

Course assessment

Total number of assessed students: 92

А	В	С	D	Е	FX
73.91	13.04	7.61	0.0	5.43	0.0

Provides: Ing. Zuzana Tkáčová, Ing.Paed.IGIP., doc. RNDr. Ľubomír Šnajder, PhD.

Date of last modification: 16.03.2024

University: P. J. Šafá	rik University in Košice			
Faculty: Faculty of S	cience			
Course ID: CJP/ PFAJ4/07	Course name: English Language of Natural Science			
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	ce rse-load (hours): dy period: 28			
Number of ECTS cr	edits: 2			
Recommended seme	ster/trimester of the course: 4.			
Course level: I.				
Prerequisities:				
2 classes at the most Continuous assessme 1 credit test taken pre 1 project (quiz on the 5 LMS quizzes (25% In order to be admitte assessment The exam test results represent the other 50 The final grade for th A 93-100, B 86-92, C	in class and completed homework assignments. Students are allowed to miss ent: esumably in weeks 6/7 topic of the student's field of study) 25% of the continuous assessment of the continuous assessment) ed to the final exam, a student has to score at least 65 % from the continuous represent 50% of the final grade for the course, continuous assessment results			
Learning outcomes: Enhancement of students' language skills (speaking, writing, reading and listening comprehension) in English for specific and academic purposes and development of students' linguistic competence. Students obtain knowledge of selected phonological, lexical and syntactic aspects of professional English, improve their pragmatic competence - students can effectively use the language for a given purpose, and acquire presentation skills at B2 level (CEFR) with focus on terminology of natural sciences.				
 6. Expressing cause a 7. Describing structure 8. Explaining process 	dying language f scientific language lemic study terminology and concepts and effect res			

10. Talking about problem and solution

- 11. Referencing authors
- 12. Giving examples
- 13. Visual aids and numbers
- 14. Referencing time and place

Presentation topics related to students' study fields.

Recommended literature:

lms.upjs.sk - e-kurz Odborný anglický jazyk pre prírodné vedy.

Redman, S.: English Vocabulary in Use, Pre-intermetdiate, Intermediate. Cambridge University Press, 2003.

Armer, T.: Cambridge English for Scientists. CUP, 2011.

Wharton J.: Academic Encounters. The Natural World. CUP, 2009.

P. Fitzgerald : English for ICT studies. Garnet Publishing, 2011.

https://worldservice/learningenglish, https://spectator.sme.sk

www.isllibrary.com

linguahouse.com

Course language:

English, level B2 (CEFR)

Notes:

Course assessment

Total number of assessed students: 3239

А	В	С	D	Е	FX
38.53	26.37	16.3	9.54	7.19	2.07

Provides: Mgr. Viktória Mária Slovenská, Mgr. Lenka Klimčáková, Mgr. Katarína Szabová, PhD.

Date of last modification: 06.02.2024

University: P. J. Šafárik University in Košice						
Faculty: Faculty of Science						
Course ID: ÚGE/ Course name: Environmental Geology ENG1/21						
Course type, scope Course type: Lectu Recommended cou Per week: 1 / 1 Pe Course method: p	ure / Practice urse-load (h r study perio	ours):				
Number of ECTS c	redits: 3					
Recommended sem	ester/trimes	ter of the course	e: 3.			
Course level: I., II.						
Prerequisities:						
Conditions for cour	rse completi	on:				
Learning outcomes	:					
Brief outline of the	course:					
Recommended liter	rature:					
Course language:						
Notes:						
Course assessment Total number of ass	essed studen	ts: 8				
A	В	С	D	Е	FX	
0.0	50.0	37.5	12.5	0.0	0.0	
Provides: doc. Ing.	Katarína Bór	nová, PhD., Mgr.	Imrich Sládek, F	PhD.		
Date of last modific	cation: 30.09	.2024				
Approved: prof. Mg	gr. Jaroslav H	lofierka, PhD., pi	of. RNDr. Vladi	mír Zeleňák, Dr	Sc.	

University: P. J. Ša	fárik Universi	ity in Košice					
Faculty: Faculty of Science							
Course ID: ÚGE/ HYP/15Course name: Fieldwork in Hydrology							
Course type, scope Course type: Prac Recommended co Per week: 2 Per st Course method: p	tice urse-load (ho tudy period:	ours):					
Number of ECTS of	credits: 3						
Recommended sem	nester/trimes	ter of the cours	e: 4.				
Course level: I.							
Prerequisities:							
Conditions for cou	rse completi	o n:					
Learning outcomes	5:						
Brief outline of the	course:						
Recommended lite	rature:						
Course language:							
Notes:							
Course assessment Total number of ass		ts: 80					
A	В	С	D	Е	FX		
93.75	5.0	0.0	1.25	0.0	0.0		
Provides: RNDr. D	ušan Barabas	, CSc.			1		
Date of last modified	cation: 27.06	.2022					
Approved: prof. M	gr. Jaroslav H	ofierka, PhD p	rof. RNDr. Vladi	mír Zeleňák, Dr	Sc.		

	Safárik Univers	ity in Košice					
Faculty: Faculty	of Science						
Course ID: ÚCH PCH1/00	V/ Course na	ame: Food chemis	stry				
Course type, sco Course type: La Recommended Per week: 2 / 1 Course method	ecture / Practice course-load (h Per study peri	e ours):					
Number of ECTS credits: 4							
Recommended s	emester/trimes	ster of the course	: 5.				
Course level: I.,	II.						
Prerequisities:							
	ing semester, p	resentation on ce		vo exams, one in C: 71-80b, D: 61-			
	cieve informat	ions and knowled es in food during	-	emical substances storage.	s in food, their		
carbohydrates. W	ries of substance /ater, minerals,		anorganic com	ood. Aminoacids, pounds, vitamins. products.			
Recommended l	iterature:						
Recommended la Course language english							
Course language english Notes: Teaching is carrie	et out in persor tool. The form		•	e MS Teams or BI acher at the begin			
Course language english Notes: Teaching is carrid (BigBlueButton) semester, update Course assessme	e: ed out in persor tool. The form d continuously. ent	of teaching is spe	•				
Course language english Notes: Teaching is carrie (BigBlueButton) semester, update	e: ed out in persor tool. The form d continuously. ent	of teaching is spe	•				
Course language english Notes: Teaching is carrie (BigBlueButton) semester, update Course assessme Total number of	ed out in persor tool. The form d continuously. ent assessed studen	of teaching is spe ts: 317	ecified by the te	acher at the begin	ning of the		
Course language english Notes: Teaching is carrid (BigBlueButton) semester, updated Course assessme Total number of A 68.14	ed out in persor tool. The form d continuously. ent assessed studen B 27.44	of teaching is spe ts: 317 C 4.1	D	acher at the begin	FX		
Course language english Notes: Teaching is carrie (BigBlueButton) semester, update Course assessme Total number of A	ed out in persor tool. The form d continuously. ent assessed studen B 27.44 Ján Elečko, Ph	of teaching is spe ts: 317 C 4.1 D.	D	acher at the begin	FX		

	University:	ΡJ	Šafárik	University	v in Košice
I	University.	1	Salarik	Oniversity	

Faculty: Faculty of Science

Course ID: ÚCHV/ **Course name:** Fundamentals of Bioanalytical Chemistry BACHZ/06

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours):

Per week: 2 / 1 **Per study period:** 28 / 14

Course method: present

Number of ECTS credits: 5

Recommended semester/trimester of the course: 3.

Course level: I.

Prerequisities:

Conditions for course completion:

Elaboration and presentation of a semester project with an assigned topic. Completion of block exercises. Oral examination.

Detailed conditions for completing the subject are listed in the electronic bulletin board of the subject and in the repository of digital support materials LMS UPJŠ and are updated annually.

Learning outcomes:

After completing the course, the student has basic knowledge about biological samples, factors affecting biological samples and analytical methods used in clinical chemistry and bioanalysis.

Brief outline of the course:

Introduction to Bioanalytical Chemistry. Biological samples classification. Factors that affect analytes in biological samples. Collection, transport and storage of samples, the main principles of sampling, the suppressing of undesirable phenomena. Selected methods of pretreatment of biological samples. Analyzers, equipment and organization of work in a clinical laboratory. Control and management of quality in clinical laboratory. Quality manual, calibration, control, and reference materials. Validation and Good Laboratory Practice. Buffers in bioanalysis. Enzymes in bioanalysis, introduction, distribution, Mechanism of enzyme catalysis. The kinetics of enzymatic reactions with one substrate, the Michaelis constant, constant specificity, lag phase, kinetics of reactions with two substrates. Moderators of enzyme activity. Selected methods for the analysis of biomolecules.

Recommended literature:

1. Chromý, V. a kol.: Bioanalytika, MU Brno, 2002

2. Kukačka, J. a kol.: Bioanalytická chemie v príkladech a cvičeních, Karolinum, 2010

3. Mikkelsen, S.R, Cortón E.: Bioanalytical Chemistry, Wiley, 2004

4. Wilson I.: Bioanalytical Separations 4, (Handbook of Analytical

Separations), Elsevier, 2003

5.Lee, D.C., Webb, M.: Pharmaceutical Analysis, Blackwell, 2003

Course language:

Notes:

If necessary, the teaching also takes place in a distance form with the use of various tools of LMS UPJŠ, MS teams, etc. The form of teaching is specified by the teacher at the beginning of the semester, it is continuously updated.

Total number of assessed students: 108						
А	В	С	D	Е	FX	
33.33	30.56	30.56	4.63	0.0	0.93	
Provides: doc. RNDr. Katarína Reiffová, PhD.						

Date of last modification: 22.07.2022

E14 E 14	University: P. J. Šafárik University in Košice							
Faculty: Faculty of Science								
Course ID: ÚGE/ GEP2/18Course name: Fundamentals of Geology for Geographers								
Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 2 Per study period: 28 / 28 Course method: present								
Number of EC	TS credits: 6							
Recommended	semester/trime	ster of the cours	e: 1.					
Course level: I.								
Prerequisities:								
Conditions for	course complet	ion:						
Learning outco	omes:							
Brief outline of	the course:							
Courses have f occur in the Ear minerals, taxolo	following object th (global tector ogy of intrusive r	iics, species of ma ocks, taxology of	agmatism), secon sedimentary rock	ent theories of p dly, to describe th s and rocks which cs of the historic	ne rock-forming h had overcame			
Courses have f occur in the Ear minerals, taxolo metamorphosis	following object th (global tecton ogy of intrusive r , basics of the r	iics, species of ma ocks, taxology of	agmatism), secon sedimentary rock	dly, to describe th s and rocks which	ne rock-forming h had overcame			
Courses have f occur in the Ear minerals, taxolo metamorphosis paleontology.	following object th (global tecton ogy of intrusive r , basics of the r literature:	iics, species of ma ocks, taxology of	agmatism), secon sedimentary rock	dly, to describe th s and rocks which	ne rock-forming h had overcame			
Courses have f occur in the Ear minerals, taxolo metamorphosis paleontology. Recommended	following object th (global tecton ogy of intrusive r , basics of the r literature:	iics, species of ma ocks, taxology of	agmatism), secon sedimentary rock	dly, to describe th s and rocks which	ne rock-forming h had overcame			
Courses have f occur in the Ear minerals, taxolo metamorphosis paleontology. Recommended Course languag Notes: Course assessm	following object th (global tecton ogy of intrusive r , basics of the r literature: ge:	ics, species of ma ocks, taxology of egional geology	agmatism), secon sedimentary rock	dly, to describe th s and rocks which	ne rock-forming h had overcame			
Courses have f occur in the Ear minerals, taxolo metamorphosis paleontology. Recommended Course languag Notes: Course assessm	following object th (global tector ogy of intrusive r , basics of the r literature: ge:	ics, species of ma ocks, taxology of egional geology	agmatism), secon sedimentary rock	dly, to describe th s and rocks which	ne rock-forming h had overcame			
Courses have f occur in the Ear minerals, taxolo metamorphosis paleontology. Recommended Course languag Notes: Course assessm Total number o	following object th (global tector ogy of intrusive r , basics of the r literature: ge: nent f assessed studer	nts: 1212	agmatism), secon 'sedimentary roch of Slovakia, basi	dly, to describe th cs and rocks whic cs of the historic	he rock-forming h had overcame al geology and			
Courses have f occur in the Ear minerals, taxolo metamorphosis paleontology. Recommended Course languag Notes: Course assessm Total number o A	Tollowing object Th (global tector ogy of intrusive r , basics of the r literature: ge: Tent f assessed studer B 17.66	nts: 1212 C 32.59	agmatism), secon sedimentary roch of Slovakia, basi	dly, to describe the sand rocks which consider the historic constant of the historic E	he rock-forming h had overcame cal geology and FX			
Courses have f occur in the Ear minerals, taxolo metamorphosis paleontology. Recommended Course languag Notes: Course assessm Total number o A 7.84	following object th (global tector ogy of intrusive r , basics of the r literature: ge: nent f assessed studer B 17.66 Ing. Katarína Bó	nts: 1212 C 32.59 mová, PhD.	agmatism), secon sedimentary roch of Slovakia, basi	dly, to describe the sand rocks which consider the historic constant of the historic E	he rock-forming h had overcame cal geology and FX			

COURSE INFORMATION LETTER							
University: P. J. Šafárik University in Košice							
Faculty: Faculty of Science							
Course ID: ÚC VCHU/15	HV/ Course na	ame: General Ch	emistry				
Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 4 / 2 Per study period: 56 / 28 Course method: present Number of ECTS credits: 7							
		ster of the cours	a. 1				
Course level: I.		ster of the cours	e: 1.				
	ÚCHV/CHV1/9	9					
Written test in participation on	seminars.		mester followed	by the oral exan	nination. Active		
-	lents with know	-	and molecules th rties of elements				
 Brief outline of the course: Main terms used in chemistry. Atoms – models of atoms, electron configuration, chemical periodicity and its effect on the properties of elements, radioactivity. Chemical bonds and intermolecular interactions. Chemical structure and physical properties of matter. State of matter. Solutions. Chemical equilibrium. Basis of chemical thermodynamics and chemical kinetics. Classification of chemical reactions. Electrochemistry. 							
Recommended literature: 1. Atkins P., Jones L.: Chemical Principles, 2nd ed., Freeman, New York 2002. 2. Russel J.B.: General Chemistry, 2nd ed., McGraw Hill, London 1992.							
Course language:							
Notes:							
Course assessm Total number of	ent f assessed studen	nts: 362					
А	В	С	D	Е	FX		
25.69	27.35	27.62	11.05	7.73	0.55		
Provides: prof. RNDr. Vladimír Zeleňák, DrSc., doc. RNDr. Ivan Potočňák, PhD.							

Date of last modification: 07.02.2022

University:	ΡJ	Šafárik	University	in Košice
omversiey.	1.0.	Suluin	Oniversity	

Faculty: Faculty of Science

Course ID: ÚCHV/	Course name: General Course of Analytical Chemistry - Laboratory
PACU/03	

Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 4 Per study period: 56

Course method: present

Number of ECTS credits: 4

Recommended semester/trimester of the course: 4.

Course level: I.

Prerequisities: ÚCHV/ANCHU/03 or ÚCHV/ANCHU/21

Conditions for course completion:

Active participation in laboratory exercises and seminars; successful completion of the tests.

1. Participation in laboratory exercises is required. Assigned teacher who leads exercises might excuse without substitute the student's absence (incapacity for work, family reasons, etc.) for a maximum of two exercises during the semester with substitute supplying.

The assigned teacher, who leads the seminar, assesses the preparation of students and their activity in seminars. For the active participation in the exercises, the student can get a maximum of 10 points.
 Two written tests are obligatory. The written test will consist of 15 questions with 15 points, together for 2 written testes of 30 points. To successful completion of the exam, it is necessary to achieve at least 8 points from each test.

Overall score: Max. number of points: 50 (elaboration of protocols / assignments - 10 points; active participation in practical exercises - 10 points; written tests - 2×15 points). Min. number of points to successful completion of course: 26.

Note: Detailed conditions are updated annually within the repository for digital support materials (LMS UPJŠ).

Learning outcomes:

Application of theoretical knowledge of qualitative and quantitative analytical chemistry into analytical laboratory practise.

Brief outline of the course:

Practical in qualitative and quantitative analysis. Qualitative analysis, separation by selective precipitation. Quantitative methods. Gravimetry, general principles of method. Volumetric methods. Preparation of accurate solutions. Indication of equvivalency point. Titration curves, calculations in volumetric analysis. Acidimetry, alkalimetry. Manganometry. Iodometry. Complexometry. Selected Instrumental analytical methods.

Recommended literature:

- 1. Y. Bazel a kol.: Praktikum z analytickej chémie, PF UPJŠ, Košice 2019.
- 2. T. Gondová a kol.: Praktikum z analytickej chémie, PF UPJŠ, Košice 1999.
- 3. V. Szmereková, P.Meľuch: Praktikum z analytickej chémie, PF UPJŠ, Košice 1988.
- 4. J. Labuda a kol. Analytická chémia, STU, Bratislava 2014.
- 5. Z. Holzbecher a kol: Analytická chemie, SNTL, ALFA Praha 1987.

6. L. Koller: Analytická chémia, TU Košice, 2002, skriptum a v digitálnej forme. 7.D. Harvey: Modern Analytical Chemistry. McGraw Hill, Boston, 2000.

Course language:

Slovak

Notes:

The course is implemented by full-time or, if necessary, distance method using the MS Teams or BBB or a combined method. The form of teaching is specified by the teacher at the beginning of the semester and updated continuously.

Course assessment

Total number of assessed students: 425

А	В	С	D	Е	FX
60.0	26.82	10.59	1.41	1.18	0.0

Provides: RNDr. Rastislav Serbin, PhD., RNDr. Jana Šandrejová, PhD., univerzitná docentka

Date of last modification: 15.11.2021

University: P. J. Šaf	řárik University in Košice							
Faculty: Faculty of	Science							
Course ID: ÚGE/ GIS/15Course name: Geographic Information Systems								
Course type, scope Course type: Lectu Recommended cou Per week: 2 / 2 Per Course method: pr	ure / Practice urse-load (hours): r study period: 28 / 28							
Number of ECTS c	eredits: 6							
Recommended sem	nester/trimester of the course: 3.							
Course level: I.								
Prerequisities:								

Conditions for course completion:

The assessment is a combination of continual control during the practicals and the final exam in the examination period. The continual assessment is performed during the semester and it involves 2 written tests in the mid-term and end of the semester and a project report generated according to the assignment and practical skills acquired during the practicals. The student can proceed to the final exam in case he or she acquired at least 50 points of 100 in all elements of the the continual assessment. The final assessment mark is based on the average number points received in the midterm test, project report, practicals assessment, and final exam. The final exam is a written test comprising 3-4 questions. The credits are given in case the student had reached at least the E mark in continual assessment and final exam. The following marking scheme is applied in the assessment: A (100-90 points), B (80-89 points), C (70-79 points), D (60-69 points), E (50-59 points), FX (0-49 points).

Learning outcomes:

The students gain knowledge on the intermediate levele in the theory of geoinformation science, GIS, and Remote Sensing, GIS data models, methods of data processing and spatial analysis. They gain practical skills in processing of geographic data, management, analysis, and visualisation

of the geographic data in a GIS project.

Students acquire competence in defining a GIS project, suitabla data models, methods of data acquisition, data processing, analysis and visualisation, presentation skills and skills in team work.

Brief outline of the course:

The course is focused on the following topics: geoinformatics as a scientific discipline, components of geographic information system, digital landscape representation and data models, GIS standards for coordinate systems and transformations, collection of geographic data for GIS (GNSS, photogrammetry, multispectral satellite imagery, lidar, radar), data management in GIS, attribute and spatial demands, layer overlap, map algebra, spatial prediction, quality and uncertainty of geographic data, GIS web solutions, legislative aspects in GIS, GIS applications in practice.

Exercises are focused on working in ArcGIS Pro: basic and advanced vectorization, data organization in the geodatabase, import / export of various data formats to GIS, creation of color compositions from satellite images, mapping, 3D visualization and animation of geographic data, geoprocessing, map algebra, spatial and attribute demands, spatial prediction, analysis of digital

elevation models (DEM). Students learn the topics of the semester project in the middle of the semester and solve the assigned task in the team using the skills and knowledge acquired during the semester.

Recommended literature:

Course language:

Slovak or Czech or English

Notes:

Course assessment

Total number of assessed students: 393

А	В	С	D	Е	FX
27.99	26.72	27.23	12.47	5.6	0.0

Provides: doc. Mgr. Michal Gallay, PhD., Mgr. Michaela Nováková, PhD.

Date of last modification: 27.06.2022

University: P. J. Ša	fárik Univers	ity in Košice				
Faculty: Faculty of	Science					
Course ID: ÚGE/ GEOM1/21	Course name: Geography					
Course type, scope Course type: Recommended co Per week: Per stu Course method: p	urse-load (h 1dy period: present					
Number of ECTS						
Recommended sen	nester/trimes	ster of the cours	2.			
Course level: I.						
Prerequisities:						
Conditions for cou	rse completi	on:				
Learning outcome	s:					
Brief outline of the	course:					
Recommended lite	rature:					
Course language:						
Notes:						
Course assessment Total number of ass		ts: 36				
A	В	С	D	Е	FX	
19.44	11.11	11.11	25.0	30.56	2.78	
Provides:		<u> </u>		·		
Date of last modifi	cation: 27.06	5.2022				
Approved: prof. M	gr. Jaroslav H	Iofierka, PhD., p	of. RNDr. Vlad	imír Zeleňák, Dr	Sc.	

University: P. J. Šaf	árik Univers	ity in Košice				
Faculty: Faculty of	Science					
Course ID: ÚGE/ GNB/21	Course name: Geography of Religion					
Course type, scope Course type: Lect Recommended co Per week: 1 / 1 Pe Course method: p	ure / Practice urse-load (h r study perie	ours):				
Number of ECTS c	credits: 3					
Recommended sem	ester/trimes	ster of the course	e: 3.			
Course level: I.						
Prerequisities:						
Conditions for cou	rse completi	on:				
Learning outcomes	3:					
Brief outline of the	course:					
Recommended lite	rature:					
Course language:						
Notes:						
Course assessment Total number of ass	essed studen	ts: 28				
A	В	С	D	Е	FX	
17.86	14.29	32.14	25.0	10.71	0.0	
Provides: doc. Mgr.	Ladislav No	ovotný, PhD.		·1		
Date of last modifie	cation: 27.06	5.2022				
Approved: prof. Mg	gr. Jaroslav H	Iofierka, PhD., p	rof. RNDr. Vlad	imír Zeleňák, DrS	bc.	

University: P. J. Ša	fárik Univers	ity in Košice				
Faculty: Faculty of	Science					
Course ID: ÚGE/ GPOL/21	JGE/ Course name: Geography of agriculture and industry					
Course type, scope Course type: Lect Recommended co Per week: 1 / 1 Pe Course method: p	ure / Practice urse-load (h er study perio	ours):				
Number of ECTS of	credits: 3					
Recommended sem	nester/trimes	ster of the course	e: 4.			
Course level: I.						
Prerequisities:						
Conditions for cou	rse completi	on:				
Learning outcomes	5:					
Brief outline of the	course:					
Recommended lite	rature:					
Course language:						
Notes:						
Course assessment Total number of ass		ts: 19				
A	В	С	D	Е	FX	
31.58	15.79	26.32	10.53	15.79	0.0	
Provides: Mgr. Mar	rián Kulla, Ph	nD., doc. Mgr. La	dislav Novotný,	PhD.		
Date of last modified	cation: 14.02	2.2023				
Approved: prof. M	gr. Jaroslav H	Iofierka, PhD., p	rof. RNDr. Vladi	mír Zeleňák, DrS	bc.	

University: P. J. Šat	fárik Univers	ity in Košice				
Faculty: Faculty of	Science					
Course ID: ÚGE/ MOG/21	Course name: Geography of mining					
Course type, scope Course type: Lect Recommended co Per week: 2 Per st Course method: p	ure urse-load (h tudy period:	ours):				
Number of ECTS of	credits: 2					
Recommended sem	nester/trimes	ter of the course	e: 2.			
Course level: I.						
Prerequisities:						
Conditions for cou	rse completi	on:				
Learning outcomes	5:					
Brief outline of the	course:					
Recommended lite	rature:					
Course language:						
Notes:						
Course assessment Total number of ass		ts: 7				
A	В	С	D	Е	FX	
71.43	14.29	14.29	0.0	0.0	0.0	
Provides: doc. Ing.	Katarína Bór	nová, PhD., Mgr.	Imrich Sládek, I	PhD.	1	
Date of last modified	cation: 16.02	.2023				
Approved: prof. Ma	gr. Jaroslav H	lofierka, PhD., pr	of. RNDr. Vlad	imír Zeleňák, Dr	Sc.	

University: P. J.	Šafárik Univers	ity in Košice					
Faculty: Faculty	of Science						
Course ID: ÚGE GST/21	Course na	Course name: Geography of services and tourism					
Course type, sco Course type: Lo Recommended Per week: 1 / 1 Course method	ecture / Practice course-load (h Per study perio	ours):					
Number of ECT	S credits: 3						
Recommended s	emester/trimes	ster of the cours	e: 5.				
Course level: I.							
Prerequisities:							
Conditions for c	ourse completi	on:					
Learning outcon	nes:						
Brief outline of t	he course:						
Recommended l	iterature:						
Course language	2:						
Notes:							
Course assessme Total number of	-	ts: 11					
Α	В	С	D	Е	FX		
18.18	36.36	27.27	9.09	9.09	0.0		
Provides: Mgr. N PhD.	Iarián Kulla, Pł	nD., doc. Mgr. La	adislav Novotný,	PhD., doc. Mgr.	Michal Gallay		
Date of last mod	ification: 27.06	5.2022					
Approved: prof.	Mgr. Jaroslav H	Iofierka, PhD., p	rof. RNDr. Vladi	mír Zeleňák, Dr	Sc.		

University: P. J. Šar	fárik Univers	ity in Košice					
Faculty: Faculty of	Science						
Course ID: ÚGE/ GCR1/21							
Course type, scope Course type: Lect Recommended co Per week: 2 / 1 Pe Course method: p	ure / Practice urse-load (h r study perie	ours):					
Number of ECTS of							
Recommended sem	ester/trimes	ster of the cours	e: 5.				
Course level: I., II.							
Prerequisities:							
Conditions for cou	rse completi	on:					
Learning outcomes	5:						
Brief outline of the	course:						
Recommended lite	rature:						
Course language:							
Notes:							
Course assessment Total number of ass		ts: 11					
А	В	С	D	Е	FX		
18.18	18.18	45.45	18.18	0.0	0.0		
Provides: Mgr. Mar	ián Kulla, Pł	nD., doc. Mgr. La	dislav Novotný,	PhD., Mgr. Imri	ch Sládek, PhD.		
Date of last modified	cation: 27.06	5.2022					
Approved: prof. M	gr. Jaroslav H	Iofierka, PhD., p	rof. RNDr. Vladi	mír Zeleňák, Dr	Sc.		

University: P. J. Ša	afárik Univers	ity in Košice				
Faculty: Faculty o	f Science					
Course ID: ÚGE/ GAH/21	Course name: Geography of the atmosphere and hydrosphere					
Course type, scop Course type: Lec Recommended co Per week: 3 / 1 P Course method:	eture / Practice ourse-load (h er study perio present	ours):				
Number of ECTS	credits: 6					
Recommended ser	mester/trimes	ster of the cours	e: 3.			
Course level: I.						
Prerequisities:						
Conditions for cou	urse completi	on:				
Learning outcome	es:					
Brief outline of th	e course:					
Recommended lite	erature:					
Course language:						
Notes:						
Course assessmen Total number of as	-	ts: 74				
A	В	С	D	Е	FX	
2.7	20.27	35.14	35.14	6.76	0.0	
Provides: RNDr. E Mgr. Jaroslav Hofi				· ·	centka, prof.	
Date of last modif	ication: 27.06	5.2022				
Approved: prof. N	Igr. Jaroslav H	Iofierka, PhD., p	rof. RNDr. Vladi	imír Zeleňák, Dr	Sc.	

University: P. J. S	Šafárik Univers	ity in Košice					
Faculty: Faculty	of Science						
Course ID: ÚGE GPED/21	/ Course na	Course name: Geography of the pedosphere and biosphere					
Course type, sco Course type: Le Recommended Per week: 3 / 1 Course method	ecture / Practice course-load (h Per study peri : present	ours):					
Number of ECTS	S credits: 6						
Recommended se	emester/trimes	ster of the cours	e: 4.				
Course level: I.							
Prerequisities:							
Conditions for co	ourse completi	on:					
Learning outcom	nes:						
Brief outline of t	he course:						
Recommended li	terature:						
Course language	•						
Notes:				=			
Course assessme Total number of a	-	ts: 75					
A	В	С	D	Е	FX		
0.0	5.33	14.67	33.33	28.0	18.67		
Provides: RNDr. PhD., univerzitná		, CSc., doc. Mgr	. Michal Gallay,	PhD., RNDr. Ale	ena Gessert,		
Date of last mod	ification: 13.02	2.2023					
Approved: prof.	Mgr. Jaroslav H	Iofierka, PhD p	rof. RNDr. Vlad	mír Zeleňák, Dr	Sc.		

University: P. J. Šafa	arik University in Košice	
Faculty: Faculty of S	Science	
Course ID: ÚGE/ SGI2/21	Course name: Geoinforma	atics seminar
Course type, scope a Course type: Practi Recommended cou Per week: 2 Per stu Course method: pr	ce rse-load (hours): ıdy period: 28	
Number of ECTS cr	redits: 3	
Recommended sem	ester/trimester of the cours	e: 6.
Course level: I.		
Prerequisities:		
Conditions for cour	se completion:	
Learning outcomes:		
Brief outline of the	course:	
Recommended liter	ature:	
Course language:		
Notes:		
Course assessment Total number of asse	essed students: 13	
	abs	n
	100.0	0.0
Provides: doc. Mgr.	Michal Gallay, PhD., doc. R	NDr. Ján Kaňuk, PhD., Mgr. Ján Šašak, PhD.
Date of last modific	ation: 27.06.2022	
Approved: prof. Mg	r. Jaroslav Hofierka, PhD., p	rof. RNDr. Vladimír Zeleňák, DrSc.

University: P. J. Šaf	ărik University in Košice	
Faculty: Faculty of	Science	
Course ID: ÚGE/ GEX2/21	Course name: Geological	excursion
Course type, scope Course type: Pract Recommended course Per week: Per stu Course method: p	tice urse-load (hours): dy period: 3d	
Number of ECTS c		
Recommended sem	ester/trimester of the cours	e: 2.
Course level: I.		
Prerequisities:		
Conditions for cou	rse completion:	
Learning outcomes	:	
Brief outline of the	course:	
Recommended liter	rature:	
Course language:		
Notes:		
Course assessment Total number of ass	essed students: 93	
	abs	n
	100.0	0.0
Provides: doc. Ing.	Katarína Bónová, PhD.	
Date of last modific	cation: 27.06.2022	
Approved: prof. Ms	gr. Jaroslav Hofierka, PhD., r	rof. RNDr. Vladimír Zeleňák, DrSc.

University: P. J. Ša	fárik Univers	ity in Košice			
Faculty: Faculty of	Science				
Course ID: ÚGE/ GMP/21	Course na	me: Geomorpho	logical mapping		
Course type, scope Course type: Prac Recommended co Per week: 2 Per s Course method: p	tice urse-load (h tudy period: present	ours):			
Number of ECTS					
Recommended sen	nester/trimes	ter of the course	e: 4.		
Course level: I.					
Prerequisities:					
Conditions for cou	rse completi	on:			
Learning outcome	s:				
Brief outline of the	e course:				
Recommended lite	rature:				
Course language:					
Notes:					
Course assessment Total number of as		ts: 12			
A	В	С	D	Е	FX
0.0	0.0	91.67	0.0	8.33	0.0
Provides: RNDr. A	lena Gessert,	PhD., univerzitn	á docentka		
Date of last modifi	cation: 27.06	.2022			
Approved: prof. M	gr. Jaroslav H	lofierka, PhD pi	of. RNDr. Vlad	imír Zeleňák. Dr	Sc.

University: P. J. S	Šafárik Univers	ity in Košice			
Faculty: Faculty	of Science				
Course ID: ÚGE GEM2/18	Course na	ame: Geomorpho	ology		
Course type, sco Course type: Le Recommended Per week: 2 / 2 Course method	ecture / Practice course-load (h Per study peri	ours):			
Number of ECT	S credits: 6				
Recommended se	emester/trimes	ster of the cours	e: 2.		
Course level: I.					
Prerequisities:					
Conditions for co	ourse completi	on:			
Learning outcom	nes:				
Brief outline of t	he course:				
Recommended li	terature:				
Course language	:				
Notes:					
Course assessme Total number of a		ts: 1374			
А	В	С	D	E	FX
10.48	20.74	21.25	17.25	19.51	10.77
Provides: RNDr. Katarína Bónová,		PhD., univerzitn	á docentka, Mgi	. Imrich Sládek,	PhD., doc. Ing.
Date of last mod	ification: 13.02	2.2023			
Approved: prof.	Mgr. Jaroslav H	Iofierka, PhD., p	rof. RNDr. Vlad	imír Zeleňák. Dr	Sc.

University: P. J. Ša	fárik Universi	ty in Košice			
Faculty: Faculty of	Science				
Course ID: KPE/ POŽ/21	Course na	me: Getting to k	mow the Student	in Education	
Course type, scope Course type: Prac Recommended co Per week: 2 Per s Course method: p	ctice ourse-load (ho tudy period:	ours):			
Number of ECTS					
Recommended ser	nester/trimes	ter of the cours	e: 4.		
Course level: I.					
Prerequisities:					
Conditions for cou	irse completio	on:			
Learning outcome	s:				
Brief outline of the	e course:				
Recommended lite	erature:				
Course language:					
Notes:					
Course assessmen Total number of as	-	s: 105			
A	В	С	D	Е	FX
70.48	15.24	8.57	0.95	0.0	4.76
Provides: PaedDr.	Michal Novoc	ký, PhD., Mgr.	Beáta Sakalová, I	PhD.	
Date of last modifi	cation: 12.03	.2024			
Approved: prof. M	gr. Jaroslav H	ofierka, PhD., p	rof. RNDr. Vladi	mír Zeleňák. Dr	Sc.

University: P. J. Ša	fárik Universi	ity in Košice			
Faculty: Faculty of	Science				
Course ID: ÚGE/ EXH/21	Course na	me: Human Geo	graphy Excursio	on	
Course type, scope Course type: Prac Recommended co Per week: Per stu Course method: p	tice urse-load (ho 1dy period: 6	ours):			
Number of ECTS of					
Recommended sem	nester/trimes	ter of the course	e: 4.		
Course level: I.					
Prerequisities:					
Conditions for cou	rse completion	on:			
Learning outcomes	5:				
Brief outline of the	course:				
Recommended lite	rature:				
Course language:					
Notes:					
Course assessment Total number of ass		ts: 55			
A	В	С	D	Е	FX
54.55	29.09	16.36	0.0	0.0	0.0
Provides: Mgr. Mar	rián Kulla, Ph	D., doc. Mgr. La	dislav Novotný,	PhD.	1
Date of last modified	cation: 27.06	.2022			
Approved: prof. M	gr. Jaroslav H	ofierka, PhD., p	of. RNDr. Vlad	imír Zeleňák, Dr	Sc.

University: P. J. Ša	afárik Univers	ity in Košice			
Faculty: Faculty o	f Science				
Course ID: ÚGE/ HGS1/21	Course na	ame: Human Geo	ography of Slova	kia	
Course type, scop Course type: Lec Recommended co Per week: 2 / 1 P Course method:	ture / Practice ourse-load (h er study perio present	ours):			
Number of ECTS					
Recommended ser	nester/trimes	ster of the cours	e: 5.		
Course level: I.					
Prerequisities:					
Conditions for con	ırse completi	on:			
Learning outcome	es:				
Brief outline of th	e course:				
Recommended lite	erature:				
Course language:					
Notes:					
Course assessmen Total number of as		ts: 42			
A	В	С	D	Е	FX
2.38	7.14	26.19	26.19	38.1	0.0
Provides: RNDr. J doc. Mgr. Ladislav			univerzitná doce	entka, Mgr. Mari	án Kulla, PhD.,
Date of last modif	ication: 27.06	5.2022			
Approved: prof. M	lgr. Jaroslav H	Iofierka, PhD., p	rof. RNDr. Vladi	mír Zeleňák, Dr	Sc.

University: P. J. Ša	fárik Universi	ty in Košice			
Faculty: Faculty of	Science				
Course ID: KPE/ INP/17	Course na	me: Inclusive P	edagogy		
Course type, scope Course type: Prac Recommended co Per week: 2 Per se Course method: p	tice urse-load (ho tudy period:	ours):			
Number of ECTS of	credits: 2				
Recommended sen	nester/trimes	ter of the cours	e: 5.		
Course level: I.					
Prerequisities:					
Conditions for cou	rse completio	on:			
Learning outcomes	5:				
Brief outline of the	course:				
Recommended lite	rature:				
Course language:					
Notes:					
Course assessment Total number of ass		s: 111			
A	В	С	D	Е	FX
69.37	22.52	3.6	1.8	2.7	0.0
Provides: PaedDr. 1	Michal Novoc	ký, PhD.	1		
Date of last modified	cation: 14.09	.2024			
Approved: prof. M	gr. Jaroslav H	ofierka, PhD p	rof. RNDr. Vlad	imír Zeleňák, Dr	Sc.

	University:	ΡJ	Šafárik	University	v in Košice
I	University.	1	Salarik	Oniversity	

Faculty: Faculty of Science

Course ID: ÚCHV/	Course name: Inorganic Chemistry
ACHU/21	

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 3 / 1 Per study period: 42 / 14

Course method: present

Number of ECTS credits: 5

Recommended semester/trimester of the course: 2.

Course level: I.

Prerequisities: ÚCHV/VCHU/15

Conditions for course completion:

Written test in the middle and the end of the semester followed by the oral examination. Active participation on seminars.

Learning outcomes:

Gaining knowledge about the properties and reactivity of elements and their compounds, the periodicity of their properties and the periodicity of the properties of their compounds. Knowledge of the basic physical and chemical properties of elements and their compounds, reactivity, their preparation, production and occurrence.

Brief outline of the course:

Electronic configuration, abundance, use, physical and chemical properties, preparation, reactivity of non-metallic elements hydrogen, halogens, oxygen, sulphur, nitrogen, phosphorus, carbon, silicon, boron and rare gases. Binary and other compounds formed by these elements, their properties and reactivity. Metals and transition elements. Abudance, properties, reactivity, important compounds.

Recommended literature:

Greenwood, N. N., Earnshaw, A: Chemistry of the Elements. Pergamon Press, Oxford, 1984 Atkins O., Overton T., Rourke J., Weller M., Armstrong F.: Inorganic Chemistry, University Press, Oxford, 2006.

Course language:

Notes:

Course assessment

Total number of assessed students: 90

А	В	С	D	Е	FX
31.11	30.0	24.44	7.78	6.67	0.0
Provides: prof.	RNDr. Vladimír	Zeleňák, DrSc., j	prof. RNDr. Jura	j Černák, DrSc.	
Date of last mo	dification: 07.02	2.2022			

Approved: prof. Mgr. Jaroslav Hofierka, PhD., prof. RNDr. Vladimír Zeleňák, DrSc.

University: P. J. Šafárik University in Košice
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Faculty: Faculty of Science

Course ID: ÚCHV/	Course name: Inorganic Chemistry II
ACH2/21	

Course type, scope and the method:

Course type: Lecture / Practice

Recommended course-load (hours): Per week: 3 / 2 **Per study period:** 42 / 28

Course method: present

Number of ECTS credits: 6

Recommended semester/trimester of the course: 3., 5.

Course level: I.

Prerequisities: ÚCHV/ACH1/10 or ÚCHV/ACHU/21 or ÚCHV/ACHU/03

Conditions for course completion:

1. Students are required to attend seminars, this also applies to the online form of teaching. The relevant teacher who leads the seminar will justify the absence of the student (illnes, family reasons, etc.) in a maximum of two seminars during the semester without the need to replace the teaching hours. In the case of a longer justified absence (for example due to illness), the teacher will assign to the student alternative forms of duties;

2. Activity at seminars. The preparation of students and their activity in seminars is always assessed by the teacher who leads the seminar, within his / her competence.

3. Participation in 10 small written tests within the seminar, for each small test you can get 1 point. In the case of a justified absence of a student from a small test, the teacher may require to elaborate a written job. Successful completion is considered if the student obtains at least 5.5 points from these tests, which is a condition for participation in the exam. The points obtained from the seminar will be included in the total number of points obtained for the subject in the range of 10%.

4. The exam is usually carried out in written form (3 written tests, of which 2 tests during the semester) with the possibility of further oral examination, or, in case of restrictions of contact forms of the teaching, the exam will be performed in a suitable online - electronic form.

5. To successfully complete the course, it is necessary to obtain at least 51% of the maximum number of points in each test and for seminars.

Learning outcomes:

To acquire knowledge about physical and chemical properties of metallic elements and their compounds.

Brief outline of the course:

General characterization of metals, chemistry of elements of the 1st and 2nd group, aluminum and other metals elements of groups 13 to 16. Chemistry of transition elements with emphasis on the 1st transition series.

Coordination compounds, chemistry of lanthanides and actinides. In all chapters are discussed the atomic properties of elements, properties of elements as substances, properties of their compounds. Emphasis is also put on environmental aspects of the properties of elements and their compounds. The lectures are discussed at the seminars in detail.

Recommended literature:

Greenwood, N.N., Earnshaw, A.: Chemistry of the elements, Pergamon Press N.Y., 1984.
 D.F. Shriver, P.W. Atkins: Inorganic Chemistry, Oxford University Press, Oxford, 4th Ed., 2006.

Course language:

Notes:

The subject can be realized in the form of personal attendance or, if necessary, also in online form.

Course assessment

Total number of assessed students: 31

А	В	С	D	Е	FX
12.9	25.81	41.94	9.68	6.45	3.23

Provides: prof. RNDr. Juraj Černák, DrSc., prof. RNDr. Vladimír Zeleňák, DrSc., RNDr. Miroslava Matiková Maľarová, PhD.

Date of last modification: 16.11.2021

Approved: prof. Mgr. Jaroslav Hofierka, PhD., prof. RNDr. Vladimír Zeleňák, DrSc.

University: P.	J. Šafárik	University in Košice
University. 1.	J. Dululik	

Faculty: Faculty of Science

Course ID: ÚCHV/ **Course name:** Instrumental Analytical Chemistry ANCH1b/21

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours):

Per week: 2 / 2 Per study period: 28 / 28

Course method: present

Number of ECTS credits: 5

Recommended semester/trimester of the course: 4., 6.

Course level: I.

Prerequisities:

Conditions for course completion:

Active participation in seminars; successful completion of the final test. Elaboration of 2 written assignments (or subject project), which will be one of the conditions for participation in the exam. The evaluation of the student's study results within the study of the subject is carried out by a combination of continuous control during the teaching part of the semester (50%) with an examination during the examination period (50%).

Note: Detailed conditions are updated annually within the repository for digital support materials (LMS UPJŠ).

Learning outcomes:

The student acquires knowledge of the theoretical foundations and instrumentation in analytical chemistry.

Brief outline of the course:

Classification of instrumental analytical methods. Basic parts of analytical instruments. Comparison of range, accuracy, detection limit, selectivity and economic characteristics of analytical methods. Analytical signal and calibration. Detection limit. Standard addition method. Accuracy and precision. Spectral methods. Electromagnetic radiation. Analytical signal of the optical methods. Classification of spectral and optical analytical methods. Instrumentation of spectral methods. Basic parts of instruments in spectral analysis: optical elements, radiation sources, monochromators, detectors (scheme, principle, basic characteristics, advantages and disadvantages). Molecular spectrometry. Nephelometry and turbidimetry. Luminescence analysis. Infrared spectroscopy. Raman spectroscopy. Refractometry. Chiroptical methods. Mass spectroscopy. Atomic spectral methods. Atomic absorption spectroscopy. Atomic emission spectral analysis. Atomic fluorescence spectrometry. Separation and preconcentration methods. Classification of separation methods. Chromatographic and non-chromatographic separation methods. Basic characteristics of separation methods. Non-chromatographic separation methods. Chromatographic methods of separation. Classification of chromatographic methods. Elution characteristics. Liquid chromatography. Gas chromatography. Supercritical fluid chromatography. Basic parts of instruments in chromatography. Electroanalytical methods. Basic principle of electroanalytical methods and their division. Potentiometry. Polarography. Voltammetry. Electrogravimetry. Coulometry. Conductometry.

Recommended literature:

1. Labuda a kol. Analytická chémia. ISBN: 9788022742429, Vydavateľstvo: STU Bratislava, Rok vydania: 2014, Počet strán: 671

2. Christian G.D. Analytical Chemistry. John Wiley & Sons, Inc. New York – Chichester – Brisbane – Toronto – Singapore 1994.

3. Holtzclaw H.F., Jr., Robinson W.R. College Chemistry with Qualitation Analysis. D.C. Heath and Company 1988.

Course language:

Slovak

Notes:

The course is implemented by full-time or, if necessary, distance method using the MS Teams or BBB or a combined method. The form of teaching is specified by the teacher at the beginning of the semester and updated continuously.

A calculator is required to master the calculation exercises. Not a cell phone!

Course assessment

Total number of assessed students: 22

А	В	С	D	Е	FX
18.18	31.82	13.64	9.09	27.27	0.0

Provides: prof. Mgr. Vasil' Andruch, DSc.

Date of last modification: 15.07.2022

Approved: prof. Mgr. Jaroslav Hofierka, PhD., prof. RNDr. Vladimír Zeleňák, DrSc.

University: P. J. Ša	fárik Universi	ity in Košice						
Faculty: Faculty of	Science							
Course ID: KPE/ IIŠP/21								
Course type, scope Course type: Prac Recommended co Per week: 2 Per s Course method: p	tice ourse-load (ho tudy period:	ours):						
Number of ECTS	credits: 2							
Recommended sen	nester/trimes	ter of the course	e: 3.					
Course level: I.								
Prerequisities:								
Conditions for cou	rse completio	on:						
Learning outcome	s:							
Brief outline of the	course:							
Recommended lite	rature:							
Course language:								
Notes:								
Course assessment Total number of ass		ts: 54						
A	В	С	D	Е	FX			
37.04	38.89	14.81	7.41	1.85	0.0			
Provides: PaedDr. 1	Michal Novoc	ký, PhD., Mgr. Z	Zuzana Vagaská,	PhD.				
Date of last modifi	cation: 14.09	.2024						
Approved: prof. M	gr. Jaroslav H	ofierka PhD p	rof. RNDr. Vladi	mír Zeleňák, DrS	Sc.			

University: P. J. Šaf	ärik University in Košice					
Faculty: Faculty of Science						
Course ID: ÚGE/ ZEX1/21						
Course type, scope Course type: Pract Recommended cou Per week: Per stu Course method: pr	ice 1rse-load (hours): dy period: 10d					
Number of ECTS c	redits: 4					
Recommended sem	ester/trimester of the cours	e: 4.				
Course level: I.						
Prerequisities:						
Conditions for cour	se completion:					
Learning outcomes	:					
Brief outline of the	course:					
Recommended liter	ature:					
Course language:						
Notes:						
Course assessment Total number of ass	essed students: 33					
	abs	n				
	96.97 3.03					
Provides: doc. Mgr.	Ladislav Novotný, PhD., Mg	gr. Loránt Pregi, PhD., Mgr. Marián Kulla, PhD.				
Date of last modific	cation: 27.06.2022					
Approved: prof. Mg	gr. Jaroslav Hofierka, PhD., p	rof. RNDr. Vladimír Zeleňák, DrSc.				

University: P. J. Š	Safárik University in Košice
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Faculty: Faculty of Science

Course ID: ÚCHV/ **Course name:** Introduction to Environmental Chemistry UECH/08

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 2 / 1 Per study period: 28 / 14

Course method: present

Number of ECTS credits: 5

Recommended semester/trimester of the course: 3., 4..

Course level: I.

Prerequisities:

Conditions for course completion:

Continuous test. Active participation in exercises - elaboration of semester work. Passing the final examination in the form of a written test.

Learning outcomes:

Introduction to topics in environmental chemistry and basic procedures applied for environmental protection. Discusses current and future environmental problems and possibilities how to solve them from chemichal point of view.

Brief outline of the course:

Introduction to Environmental Chemistry

Chemical aspects of pollution and environmental problems. Composition and behavior of the atmosphere. Energy balance of the Earth and climate changes. Principles of photochemistry, photoprocesses in the atmosphere. Petroleum, hydrocarbons and coal (characteristics, sources and environmental pollution). Soaps, polymers and synthetic surfactants. Haloorganics and pesticides. Environmental chemistry of some important elements (C, N, S, P, halogens, biologically important metals ...). Environmental chemistry in aqueous media. Aqueous systems, parameters, cycles and their protection. The Earth's crust (rocks, minerals, soils). Natural and artificial radioactivity, utilization. Energy and energy sources (fossil fuels, nuclear, geothermal, solar energy, wind and water energy). Solid waste disposal and recycling.

Recommended literature:

1. Gary W. van Loon, Stephen J. Duffy : Environmental Chemistry - A Global Perspective, Oxford University Press, Oxford 2003

2. R.A. Bailey, H.M. Clark, J.P. Ferris, S. Krause, R.L. Strong : Chemistry of the Environment, Academic Press, San Diego 2002

3. G. Schwedt: The Essential Guide to Environmental Chemistry, Wiley and Sons, London 2001

4. R.N. Reeve, J.D. Barnes: General Environmental Chemistry, Wiley, London 1994

5. G. Burton, J. Holman, G. Pilling, D. Waddington: Chemical Storylines, Heinemann, Oxford, London 1994

6. www

Course language:

Notes:

Based on the current pandemic situation in Slovakia and in accordance with the conditions of the Faculty of Natural Sciences of UPJŠ in Košice, the education and examination can also be carried out in a distance form. The tutorial will be carried out in the form of online lectures and consultings in the BigBlueButton system. The written form of the exam takes place through the Google Forms app. Students prepare responses to the final written test. Test questions are randomly generated each time. The final oral exam is conducted through a webinar in BigBlueButton https://bbb.science.upjs.sk/b) system with online generation of random question numbers.

n

0.0

Course assessment

Total number of assessed students: 1

abs

100.0

Provides: doc. RNDr. Andrea Straková Fedorková, PhD.

Date of last modification: 18.07.2022

Approved: prof. Mgr. Jaroslav Hofierka, PhD., prof. RNDr. Vladimír Zeleňák, DrSc.

University: P. J. Šafá	rik University in Košice					
Faculty: Faculty of S	cience					
Course ID: Dek. PF UPJŠ/USPV/13	Course name: Introduction	to Study of Sciences				
Course type, scope a Course type: Lectur Recommended cour Per week: Per stud Course method: pre	re / Practice rse-load (hours): ly period: 12s / 3d					
Number of ECTS cr	edits: 2					
Recommended seme	ster/trimester of the cours	e: 1.				
Course level: I.						
Prerequisities:						
Conditions for cours	e completion:					
Learning outcomes:						
Brief outline of the c	ourse:					
Recommended litera	iture:					
Course language:						
Notes:						
Course assessment Total number of asse	ssed students: 2206					
	abs	n				
	89.39 10.61					
Provides: doc. RNDr	. Marián Kireš, PhD.					
Date of last modifica	tion: 30.08.2022					
Approved: prof. Mgr	. Jaroslav Hofierka, PhD., p	rof. RNDr. Vladimír Zeleňák, DrSc.				

University: P. J. Ša	fárik Universi	ity in Košice			
Faculty: Faculty of	Science				
Course ID: ÚGE/ UDID/21	Course na	me: Introduction	n to the didactics	of geography	
Course type, scope Course type: Lect Recommended co Per week: 1 / 1 Pe Course method: p	ture / Practice ourse-load (ho er study perio	ours):			
Number of ECTS	credits: 2				
Recommended sen	nester/trimes	ter of the cours	e: 6.		
Course level: I.					
Prerequisities:					
Conditions for cou	rse completio	o n:			
Learning outcome	s:				
Brief outline of the	e course:				
Recommended lite	erature:				
Course language:					
Notes:					
Course assessment Total number of as		ts: 9			
A	В	С	D	Е	FX
44.44	55.56	0.0	0.0	0.0	0.0
Provides: RNDr. St	tela Csachová	, PhD., doc. RNI	Dr. Ján Kaňuk, P	hD.	1
Date of last modifi	cation: 27.06	.2022			
Approved: prof. M	gr. Jaroslav H	ofierka, PhD., p	rof. RNDr. Vlad	imír Zeleňák, Dr	Sc.

University: P. J. Šat	árik Universi	ity in Košice			
Faculty: Faculty of	Science				
Course ID: ÚGE/ LOS/18	Course na	me: Linux and c	open source GIS		
Course type, scope Course type: Prac Recommended co Per week: 2 Per st Course method: p	tice urse-load (he udy period:	ours):			
Number of ECTS of	credits: 3				
Recommended sem	ester/trimes	ter of the cours	e: 3.		
Course level: I.					
Prerequisities:					
Conditions for cou	rse completio	on:			
Learning outcomes	:				
Brief outline of the	course:				
Recommended lite	rature:				
Course language:					
Notes:					
Course assessment Total number of ass	essed student	ts: 71			
A	В	С	D	Е	FX
61.97	33.8	4.23	0.0	0.0	0.0
Provides: Mgr. Mic	haela Nováko	ová, PhD., prof. 1	Mgr. Jaroslav Ho	ofierka, PhD.	
Date of last modified	cation: 30.09	.2021			
Approved: prof. Ma	gr. Jaroslav H	ofierka, PhD., p	rof. RNDr. Vlad	imír Zeleňák, Dr	Sc.

Faculty: Faculty of S	cience
Course ID: ÚMV/ MTCa/22	Course name: Mathematics I for chemists
Course type, scope a Course type: Lectu Recommended cou Per week: 2 / 2 Per Course method: pr	re / Practice rse-load (hours): study period: 28 / 28
Number of ECTS ci	edits: 5
Recommended seme	ester/trimester of the course: 1., 3.
Course level: I.	
Prerequisities:	
	ve problems from selected thematic units. subject is according to the results from the semester and in view of the results
of the written final te and two extensive te semester. Then stude 12 points from the m 100-80-A, 79-70-B, e	subject is according to the results from the semester and in view of the results st. During the semester, students write tests at all seminars (together 20 points) sts (together 50 points). It is necessary to obtain at least 28 points during the ents may write the exam. To pass the exam, it is necessary to obtain at least aximum number of 30 points. The scale for student evaluation is as follows 59-60-C, 59-50-D, 49-40-E. If a student does not achieve the required mninimal m the exam test (12 points) and during the semester (together 28 points), he
of the written final te and two extensive te semester. Then stude 12 points from the m 100-80-A, 79-70-B, on number of points from she is evaluated by F Learning outcomes: After completing th equations and inequ	subject is according to the results from the semester and in view of the results st. During the semester, students write tests at all seminars (together 20 points) sts (together 50 points). It is necessary to obtain at least 28 points during the ents may write the exam. To pass the exam, it is necessary to obtain at least aximum number of 30 points. The scale for student evaluation is as follows 59-60-C, 59-50-D, 49-40-E. If a student does not achieve the required mninimal m the exam test (12 points) and during the semester (together 28 points), he
of the written final te and two extensive te semester. Then stude 12 points from the m 100-80-A, 79-70-B, o number of points fro she is evaluated by F Learning outcomes: After completing th equations and inequal differential and integ Brief outline of the o Week 1-6: Definition functions. Compositi Week 7-14: Limit of	subject is according to the results from the semester and in view of the results st. During the semester, students write tests at all seminars (together 20 points) sts (together 50 points). It is necessary to obtain at least 28 points during the ents may write the exam. To pass the exam, it is necessary to obtain at least aximum number of 30 points. The scale for student evaluation is as follows 59-60-C, 59-50-D, 49-40-E. If a student does not achieve the required mninimal m the exam test (12 points) and during the semester (together 28 points), he X.

Notes:

Course assessment Total number of assessed students: 640								
A B C D E FX								
11.41	11.41 10.94 16.41 21.09 28.28 11.88							
	Provides: RNDr. Jana Borzová, PhD., RNDr. Miriam Kleinová, PhD., RNDr. Miriama Kmeciková, RNDr. Monika Krišáková							
Date of last modification: 18.04.2022								
Approved: prof	f. Mgr. Jaroslav H	Iofierka, PhD., p	rof. RNDr. Vladi	mír Zeleňák, DrS	Sc.			

University: P. J. Ša	fárik Univers	ity in Košice				
Faculty: Faculty of	Science					
Course ID: KPE/Course name: Mentoring and Coaching in School PracticeMKŠP/21						
Course type, scope Course type: Prac Recommended co Per week: 2 Per se Course method: p	tice urse-load (he tudy period: present	ours):				
Number of ECTS						
Recommended sen	nester/trimes	ter of the cours	e: 5.			
Course level: I.						
Prerequisities:						
Conditions for cou	rse completi	on:				
Learning outcomes	5:					
Brief outline of the	course:					
Recommended lite	rature:					
Course language:						
Notes:						
Course assessment Total number of ass		ts: 63				
А	В	С	D	Е	FX	
84.13	12.7	3.17	0.0	0.0	0.0	
Provides: Mgr. Zuz	ana Vagaská,	PhD.			1	
Date of last modified	cation: 18.09	.2024				
Approved: prof. M	gr. Jaroslav H	lofierka, PhD., p	rof. RNDr. Vlad	imír Zeleňák, Dr	Sc.	

University: P. J. Ša	fárik Univers	ity in Košice					
Faculty: Faculty of Science							
Course ID: ÚGE/Course name: Metageography and planetary geographyMPG/21							
Course type, scope Course type: Lect Recommended co Per week: 1 / 1 Pe Course method: p	ure / Practice urse-load (h er study perio	ours):					
Number of ECTS							
Recommended sen	nester/trimes	ster of the cours	e: 1.				
Course level: I.							
Prerequisities:							
Conditions for cou	rse completi	on:					
Learning outcome	5:						
Brief outline of the	course:						
Recommended lite	rature:						
Course language:							
Notes:							
Course assessment Total number of ass		ts: 138					
А	В	С	D	Е	FX		
42.03	46.38	8.7	0.72	0.0	2.17		
Provides: prof. Mg	r. Jaroslav Ho	ofierka, PhD., Mg	gr. Katarína Onač	illová, PhD.			
Date of last modifi	cation: 27.06	5.2022					
Approved: prof. M	gr. Jaroslav H	Iofierka, PhD., p	rof. RNDr. Vladi	mír Zeleňák, Dr	Sc.		

University: P. J. Š	afárik Univers	ity in Košice				
Faculty: Faculty o	f Science					
Course ID: ÚGE/ HGV/21Course name: Methods of human geographical research						
Course type, scop Course type: Pra Recommended c Per week: 3 Per Course method:	ctice ourse-load (h study period: present	ours):				
Number of ECTS						
Recommended se	mester/trimes	ster of the cours	e: 6.			
Course level: I.						
Prerequisities:						
Conditions for co	urse completi	on:				
Learning outcom	es:					
Brief outline of th	e course:					
Recommended lit	erature:					
Course language:						
Notes:						
Course assessmen Total number of as		ts: 15				
A	В	С	D	Е	FX	
100.0	0.0	0.0	0.0	0.0	0.0	
Provides: RNDr. S Dická, PhD., unive			,	,	Nestorová-	
Date of last modif	fication: 27.06	5.2022				
Approved: prof. N	Agr. Jaroslav H	Iofierka, PhD., p	rof. RNDr. Vlac	limír Zeleňák, Dr	Sc.	

University: P. J. Š	Śafárik Univers	ity in Košice				
Faculty: Faculty	of Science					
Course ID: ÚGE/ FGV/21Course name: Methods of physical geographical research						
Course type, scop Course type: Pra Recommended Per week: 3 Per Course method:	actice course-load (h study period: present	ours):				
Number of ECTS	S credits: 3					
Recommended so	emester/trimes	ster of the cours	e: 5.			
Course level: I.						
Prerequisities:						
Conditions for co	ourse completi	on:				
Learning outcom	nes:					
Brief outline of t	he course:					
Recommended li	terature:					
Course language	:					
Notes:						
Course assessme Total number of a		ts: 13				
А	В	С	D	Е	FX	
100.0	0.0	0.0	0.0	0.0	0.0	
Provides: RNDr. Ing. Katarína Bón				D., univerzitná do	ocentka, doc.	
Date of last modi	fication: 27.06	5.2022		_		
Approved: prof.]	Mgr. Jaroslav H	Iofierka, PhD., p	rof. RNDr. Vladi	mír Zeleňák, Dr	Sc.	

University: P. J. Šafárik University in Košice							
Faculty: Faculty of	Science						
Course ID: ÚGE/ Course name: Methods of thematic cartography MTK/21 Image: Methods of thematic cartography							
Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 2 Per study period: 28 Course method: present							
Number of ECTS c	redits: 3						
Recommended semester/trimester of the course: 2.							
Course level: I.							

Prerequisities:

Conditions for course completion:

The evaluation is based on the submitted assignments from the exercises.

Exercises are realized in the form of regular teaching, the introduction of the exercise is devoted to the theoretical basis, followed by the practical part of the exercise, which aims to work with spatial data in order to create a thematic map. During the semester, students will receive assignments aimed at creating a thematic map using selected methods of thematic cartography. Students submit assignments on an ongoing basis. Each assignment is evaluated separately. In order for the assignment to be accepted, it is necessary to obtain a minimum grade E from each assignment. The final evaluation is the average of the evaluation of individual assignments. Credits will be awarded only to a student who achieves a grade of at least E in the overall evaluation. Rating scale: A (100-91%), B (81-90%,) C (71-80%), D (61-70 %), E (51-60%).

Learning outcomes:

Knowledge: The student will gain knowledge and skills from thematic cartography. They will get acquainted with the theoretical aspects of the content and principles of creating thematic maps. He will gain theoretical foundations and an overview of various aspects of thematic cartography, such as color theory in cartography, types of scales and division of the statistical file into intervals. They will get acquainted with the means of expression cartographic and methods of thematic cartography and gain an overview of the use of dynamic elements of cartographic visualization. Skills: The student will learn to create thematic maps using GIS professionally and cartographically correctly. Can evaluate the suitability of the cartographic method for the representation of various geographical phenomena and determine the optimal procedure for creating thematic maps. Competences: The student is able to evaluate the thematic maps and the suitability of the methods of thematic cartography with a high degree of independence. He will get acquainted with professional terminology in the field of thematic cartography of geodesy, geoinformatics and cartography.

Brief outline of the course:

Exercises: Introduction to thematic cartography (content and types of thematic maps, phases and principles of creating thematic maps, compiling the content of the thematic map); Means of expression; Colors in maps; Scales (data evaluation, division of scales, creation of interval and

functional scales, methods for plotting extremes in a statistical file); Legend of thematic maps; Point character method; Line character method; Area character method; Comma method; Isolinia method; Cartographs and cartograms method; Cartographic anamorphosis and cartotypogram method; methods for expressing the dynamics of spatial phenomena; Description in maps; composition of thematic maps; Geospatial data topology control and map generalization. Evaluation of maps and atlases; Animations, interactive maps and virtual reality in cartography.

Recommended literature:

VOŽENÍLEK, V. (2005). Cartography for GIS: geovisualization and map communication. Olomouc, Vydavatelství UP.

KRAAK, M.J., ORMELING, F. (2003). Cartography. Visualization of Geospatial Data. Harlow. Prentice Hall, Pearson Education.

PETERSON, M. P. ET AL. (1995). Interactive and Animated Cartography. Upper Saddle River Prentice Hall.

VOŽENÍLEK, V., KAŇOK, J. A KOL. (2012). Metody tematické kartografie: vizualizace prostorových informací. Olomouc, Univerzita Palackého v Olomouci.

SLOCUM, T.A. ET AL. (2002). Thematic Cartography and Visualization. Upper Saddle River, Pearson/Prentice Hall.

Course language:

Notes:

Course assessment

Total number of assessed students: 28

А	В	С	D	Е	FX			
42.86	42.86	10.71	0.0	0.0	3.57			
Provides: doc. RNDr. Ján Kaňuk, PhD., Mgr. Jozef Šupinský, PhD., Mgr. Loránt Pregi, PhD.								
Date of last modification: 27.06.2022								

Approved: prof. Mgr. Jaroslav Hofierka, PhD., prof. RNDr. Vladimír Zeleňák, DrSc.

University: P. J. Ša	fárik Universit	y in Košice			
Faculty: Faculty of	Science				
Course ID: ÚGE/ MKR/21	Course nam	ne: Microgeog	raphy		
Course type, scope Course type: Prac Recommended co Per week: 2 Per s Course method: p	tice burse-load (ho tudy period: 2 present	urs):			
Number of ECTS					
Recommended sen	nester/trimest	er of the cours	e: 6.		
Course level: I.					
Prerequisities:					
Conditions for cou	rse completio	n:			
Learning outcome	s:				
Brief outline of the	e course:				
Recommended lite	rature:				
Course language:					
Notes:					
Course assessment Total number of ass		: 25			
A	В	С	D	Е	FX
60.0	40.0	0.0	0.0	0.0	0.0
Provides: Mgr. Imr	ich Sládek, Ph	D., doc. Mgr. L	adislav Novotný	, PhD.	1
Date of last modifi	cation: 05.09.2	2024			
Approved: prof. M	gr. Jaroslav Ho	ofierka, PhD., n	rof. RNDr. Vladi	mír Zeleňák. Dr	Sc.

University: P. J. Šaf	árik Univers	ity in Košice			
Faculty: Faculty of	Science				
Course ID: ÚGE/ NSGE/15	Course na	me: Mineral Res	sources - geolog	ical and environn	nental relations
Course type, scope Course type: Lectu Recommended course Per week: 2 / 1 Pe Course method: p	ure / Practice urse-load (h r study perio	ours):			
Number of ECTS c	redits: 4				
Recommended sem	ester/trimes	ter of the cours	e: 6.		
Course level: I.					
Prerequisities:					
Conditions for cour	rse completi	on:			
Learning outcomes					
Brief outline of the	course:				
Recommended liter	rature:				
Course language:					
Notes:					
Course assessment Total number of ass	essed studen	ts: 152			
A	В	С	D	Е	FX
41.45	25.0	21.71	9.21	0.66	1.97
Provides: doc. Ing.	Katarína Bór	nová, PhD.			
Date of last modifie	cation: 30.09	.2021			
Approved: prof. Mg	gr. Jaroslav H	Iofierka, PhD., p	of. RNDr. Vlad	imír Zeleňák, Dr	Sc.

University: P. J. Ša	fárik Universi	ty in Košice				
Faculty: Faculty of	Science					
Course ID: KPE/ Course name: Multiculturalism and Multicultural Education MMKV/17 Course name: Multiculturalism and Multicultural Education						
Course type, scope Course type: Prac Recommended co Per week: 2 Per s Course method: p	tice ourse-load (ho tudy period:	ours):				
Number of ECTS	credits: 2					
Recommended sen	nester/trimes	ter of the cours	e: 4.			
Course level: I.						
Prerequisities:						
Conditions for cou	rse completio)n:				
Learning outcome	s:					
Brief outline of the	e course:					
Recommended lite	rature:					
Course language:						
Notes:						
Course assessment Total number of ass		s: 242				
A	В	С	D	Е	FX	
40.08	41.32	16.94	0.83	0.41	0.41	
Provides: PaedDr. 1	Michal Novoc	ký, PhD.				
Date of last modifi	cation: 12.03	.2024				
Approved: prof. M	gr. Jaroslav H	ofierka, PhD., p	rof. RNDr. Vlad	imír Zeleňák, Dr	Sc.	

University: P.	J Šafárik	University in	Košice
University. 1.	J. Darank	Oniversity in	RUSICC

Faculty: Faculty of Science

Course ID: ÚCHV/ Course name: Organic chemistry OCHU/21

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours): Per week: 3 / 1 Per study period: 42 / 14

Course method: present

Number of ECTS credits: 5

Recommended semester/trimester of the course: 2.

Course level: I.

Prerequisities: ÚCHV/VCHU/15 or ÚCHV/VCHU/14 or ÚCHV/VCHU/10 or ÚCHV/VACH/10

Conditions for course completion:

Written test. Two tests, in 7th and 14th week. Test max 50 points. A student must obtain at least 51% of points. Writing of the tests is mandatory.

Written exam, 100 points. 69 Theoretical questions (69 points), 62 chemical formulas (31 points). A student must obtain at least 51% of points. Final evaluation: A 91-100 pts, B 81-90 pts, C 71-80 pts, D 61-70 pts, E 51-60 pts, FX 0-50 pts.

Learning outcomes:

Basic organic chemistry course.

Nomenclature of organic compounds, their chemical properties, structure, reactivity and characteristic reactions. Preparation of organic molecules, explanation of the basic mechanisms and principles of organic reactions.

After completing the subject, the student understands the studied theories, principles, methods and logical procedures of organic chemistry. He has knowledge of modern organic chemistry with an emphasis on the current development of knowledge in the aforementioned area.

Brief outline of the course:

Chemical bonding Hybridization and Bonding Covalent bonds Double bonds and Triple Bonds Structural Formulas of Organic Molecules Polar Covalent Bonds and Electronegativity Constitutional Isomers Alkenes Electrophilic Additions Strong Brønsted Acids Lewis Acids (non-Proton Electrophiles) Electrophilic Halogen Reagents Other Electrophilic Reagents Reduction Oxidation Radical Additions Allylic Substitution Alkynes Addition Reactions Hydrogenation Electrophiles Hydration & Tautomerism Hydroboration Nucleophilie Addition & Reduction Acidity of Terminal Alkynes (Substitution of H) Alkyl Halides General Reactivity Substitution(of X) SN2 Mechanism SN1 Mechanism Elimination (of HX) Summary of Substitution vs. Elimination Substitution by Metals Elimination Reactions of Dihalides Alcohols Reactions of Alcohols Substitution of the Hydroxyl H Substitution of the Hydroxyl Group Elimination of Water Oxidation to Quinones Aromatic compounds Electrophilic Substitution A Substitution Mechanism Reactions of Substituted Benzenes Reaction Characteristics Reactions of Disubstituted Rings Reactions of Substituent Groups Nucleophilic Substitution, Elimination & Addition Reactions Amines Basicity of Nitrogen Compounds Acidity of Nitrogen Compounds Important Reagent Bases Reactions of Amines Electrophilic Substitution at Nitrogen Preparation of 1°-Amines Preparation of 2° & 3°-Amines Reactions with Nitrous Acid Reactions of Aryl Diazonium Intermediates Elimination Reactions of Amines Oxidation States of Nitrogen Basic information: Aldehydes & Ketones Carboxylic Acids Derivatives of Carboxylic acids Natural products

Recommended literature:

1. Organic chemistry, J. Clayden, N. Greeves Warren, S. Wothers, Oxford University Press, 2012, ISBN 978-0-19-92-7029-3.

2. Organic chemistry, J. E. McMurry, Brooks/Cole, a Thomson Learning Company 2004, Sixth Eddition, ISBN 0534389996.

3. Organic chemistry, P. Zahradník, M. Mečiarová, P. Magdolen, Univerzita Komenského v Bratislave, 2019, ISBN: 978-80-223-4589-7.

Course language:

anglický

Notes:

Teaching is carried out in person or, if necessary, online using the MS Teams tool. The form of teaching is specified by the teacher at the beginning of the semester, updated continuously.

Course assessment

Total number of assessed students: 83

А	В	С	D	Е	FX
12.05	9.64	22.89	42.17	12.05	1.2

Provides: RNDr. Slávka Hamuľaková, PhD., univerzitná docentka, doc. RNDr. Miroslava Martinková, PhD., univerzitná profesorka, doc. RNDr. Mária Vilková, PhD.

Date of last modification: 04.08.2022

Approved: prof. Mgr. Jaroslav Hofierka, PhD., prof. RNDr. Vladimír Zeleňák, DrSc.

	University:	ΡJ	Šafárik	University	/ in	Košice
I	Chiver sity.	1.5.	Suluin	Oniversity	111	1205100

Faculty: Faculty of Science

Course ID: ÚCHV/ **Course name:** Organic chemistry - Lab. POCHU/15

Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 4 Per study period: 56

Course method: present

Number of ECTS credits: 4

Recommended semester/trimester of the course: 3.

Course level: I.

Prerequisities: ÚCHV/OCHU/03 or ÚCHV/OCHU/21

Conditions for course completion:

100% participations in practical exercises.

Two written tests 2 x 25 pts (a minimum of 13 points must be obtained in each test), twelve reports 12 x 2 pts, laboratory skills 12 pts, short quizzes and questions 14 pts.

A 100 pts. in total.

Assessment A: 91-100; B: 81-90; C: 71-80; D: 60-71; E: 51-60; FX: 0-50 pts.

Learning outcomes:

Students will become familiar with the basic isolation and purification methods used in a synthetic laboratory. Students should master basic laboratory technique and be able to apply the theoretical knowledge from the basic course of organic chemistry in simple synthetic projects.

Brief outline of the course:

Preparation, isolation, purification and identification of organic compounds. The emphasis is on gaining the experimental skills in synthesis of organic compounds, distillation, extraction, crystallization, sublimation and thin-layer chromatography.

- 1. Isolation and purification methods crystallization
- 2. Isolation and purification methods distillation
- 3. Preparation of ethyl acetate
- 4. Preparation of acetylsalicylic acid
- 5. Preparation of benzalaniline
- 6. Spectral methods in organic chemistry
- 7. Preparation of acetophenone oxime
- 8. Preparation of benzilic acid
- 9. Preparation of 4,5-diphenylimidazole
- 10. Isolation of caffeine from tea
- 11. Isolation of trimyristin from nutmeg

Recommended literature:

- 1. Handout with experimental procedures http://kekule.science.upjs.sk/pochu.
- 2. Organic chemistry lectures.

Course language:

Slovak

Notes: **Course assessment** Total number of assessed students: 251 А В С D Е FX 54.58 5.98 27.89 10.76 0.8 0.0 Provides: RNDr. Slávka Hamul'aková, PhD., univerzitná docentka, RNDr. Ján Elečko, PhD., RNDr. Jana Špaková Raschmanová, PhD., doc. RNDr. Mariana Budovská, PhD., RNDr. Kvetoslava Stanková, PhD. Date of last modification: 28.01.2022 Approved: prof. Mgr. Jaroslav Hofierka, PhD., prof. RNDr. Vladimír Zeleňák, DrSc.

	~	
University D	I Cofómile	University in Vation
University: P	J Salalik	University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/	Course name: Organická chémia II
OCH1b/21	

Course type, scope and the method: Course type: Lecture / Practice

Recommended course-load (hours): Per week: 2 / 2 Per study period: 28 / 28

Course method: present

Number of ECTS credits: 5

Recommended semester/trimester of the course: 3., 5.

Course level: I.

Prerequisities:

Conditions for course completion:

Written test. Two tests, in 7th and 14th week. Test max 50 points. A student must obtain at least 51% of points from each test. Writing of the tests is mandatory.

Written exam, 100 points. 69 Theoretical questions (69 points), 62 chemical formulas (31 points). A student must obtain at least 51% of points. Final evaluation: A 91-100 pts, B 81-90 pts, C 71-80 pts, D 61-70 pts, E 51-60 pts, FX 0-50 pts.

Learning outcomes:

The advanced organic chemistry. The structure, reactivity and synthesis of organic compounds with careful explanations of difficult concepts and reaction mechanisms.

After completing the subject, the student has deeper knowledge of organic chemistry, knows how to connect the properties of organic compounds with their structure and reactivity. He can explain the principles of the mechanisms of organic reactions and propose syntheses of various groups of organic compounds (also multi-stage).

The student understands the studied theories, principles, methods and logical procedures of advanced organic chemistry. He has knowledge of modern trends in the field of organic chemistry with an emphasis on the current development of knowledge in the aforementioned field.

Brief outline of the course:

Ethers - their nomenclature, preparation and reactions. Sigmatropic rearrangements, their selectivity. Preparation and reactions of epoxides.

Nitrogen compounds, Amines, their nomenclature, basicity and nucleophilicity, preparation amines, their reactions. Diazonium salts, their preparation and reactions. Nitro compounds, their preparation and reactions. Nitroso compounds, oximes, hydrazones, nitroaldol reaction.

Carbonyl compounds - aldehydes and ketones, their nomenclature and reactivity. Nucleophilic additions, addition of the primary and secondary amines and related nitrogen reagents, the aldol reaction, self-condensations, cross-condensations and related reactions. Claisen condensation and its variants. Alkylation of enolates and their applications. Benzilic acid rearrangement, Benzoin condensation, Cannizzaro reaction, Mannich reaction, Reformatsky reaction, Perkin synthesis, Knoevenagel condensation, Julia olefination, Julia-Kocienski and Petersen olefination, Wittig reaction, HWE olefination, Baylis-Hillman reaction, Darzens reaction, Baeyer-Villiger oxidation, conjugate addition, Michael addition (Michael's donors and acceptors), Robinson annulation.

Carboxylic acids, their nomenclature, properties and preparation. Reactions of carboxylic acids, Esterification. Carboxylic acid derivatives (acyl halides, anhydrides, esters, amides, – their nomenclature, properties, preparation and reactions). β -Oxoesters – their preparation and reactions. Acyloin condensation, Arndt-Eistert synthesis, Hofmann degradation, Lossen degradation, Curtius rearrangement, Wolff rearrangement.

Amino acids – their stereochemistry, properties, preparation and reactions, peptide bond - its structure, synthesis of peptides, the protective groups for amino acids.

Saccharides - classification, their nomenclature and stereochemistry. Fischer and Haworth projection, conformation of saccharides, reaction of saccharides (oxidation, reduction, production of the glycosidic bond). The protective groups. Oligosaccharides, polysaccharides.

Nucleotides and nucleic acids (structure of nucleoside, saccharides in NA, purine and pyrimidine bases in NA). Examples of nucleotides in RNA and DNA.

Heterocyclic compounds. Five and six membered heterocyclic compounds.

Terpenes, stereoids and alkaloids - their classification and properties.

Recommended literature:

Recommended literature:

1. J. Clayden, N. Greeves, S. Warren, P. Wothers: Organic Chemistry, Oxford University Press, 2012.

2. Solomons T.W. Graham: Solomon's Organic Chemistry, Willey&Sons Inc., 2017.

3. J. E. McMurry: Organic Chemistry, Vysoké učení technické v Brne, 2007, VUTIUM, ISBN: 978-80-214-3291-8 (VUT v Brne).

4. J. E. McMurry: Organic Chemistry, Cengage, 2015.

Course language:

english

Notes:

Teaching is carried out in person or, if necessary, online, using the MS Teams tool. The form of teaching is specified by the teacher at the beginning of the semester, updated continuously.

Course assessment

Total number of assessed students: 22

А	В	С	D	Е	FX
9.09	18.18	18.18	22.73	27.27	4.55

Provides: doc. RNDr. Miroslava Martinková, PhD., univerzitná profesorka

Date of last modification: 04.08.2022

University: P. J. Ša	fárik Univers	ity in Košice			
Faculty: Faculty of	Science				
Course ID: KPE/ Pg/15	Course na	me: Pedagogy			
Course type, scope Course type: Lect Recommended co Per week: 2 Per se Course method: p	ure urse-load (h tudy period:	ours):			
Number of ECTS					
Recommended sen	nester/trimes	ter of the course	e: 3.		
Course level: I.					
Prerequisities:					
Conditions for cou	rse completi	on:			
Learning outcomes	S:				
Brief outline of the	course:				
Recommended lite	rature:				
Course language:					
Notes:					
Course assessment Total number of ass		ts: 1155			
A	В	С	D	Е	FX
23.81	28.57	22.68	13.85	9.18	1.9
Provides: PaedDr. 1	Michal Novo	cký, PhD., doc. P	aedDr. Renáta C	Prosová, PhD.	
Date of last modified	cation: 14.09	.2024			
Approved: prof. M	gr. Jaroslav H	Iofierka, PhD., pr	of. RNDr. Vladi	mír Zeleňák, Dr	Sc.

	University:	ΡJ	Šafárik	University	v in Košice
I	University.	1	Salarik	Oniversity	

Faculty: Faculty of Science

Course ID: ÚCHV/ **Course name:** Physical Chemistry FCHU/22

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours):

Per week: 3 / 2 Per study period: 42 / 28

Course method: present

Number of ECTS credits: 6

Recommended semester/trimester of the course: 4.

Course level: I.

Prerequisities: ÚCHV/VCHU/14 or ÚCHV/VCHU/10 or ÚCHV/VACH/10 or ÚCHV/VCHU/15

Conditions for course completion:

Active participation in seminars. Two partial tests from computational seminars, each must be mastered at A-E. In the case of distance learning, it is necessary to prepare 2 assignments, each must be mastered at 80%.

Examination, unerstanding of three thematic areas of the subject (thermodynamics, electrochemistry, kinetics), must be mastered at A-E.

Learning outcomes:

Acquirement of the basics knowledgements of physical chemistry within the chapters: thermodynamics, phase equilibria, chemical equilibria, electrochemistry, chemical kinetics.

Brief outline of the course:

Fundamental concepts of thermodynamics, thermochemistry, chemical equilibrium, phase equilibria and diagrams, laws for ideal gas and reals gases, liquids, solutions, solutions of electrolytes. Electrochemistry: ionics and electrodics. Electrodes and electrochemical cells, corrosion. Chemical kinetics, catalysis. Adsorption.

Recommended literature:

T. Engel, P. Reid: Physical Chemistry, Pearson Educat. Inc., San Francisco 2006 P.W. Atkins: Physical Chemistry, Oxford University Presss, Oxford 1986, 1990, 1996 W.J. Moore: Physical Chemistry, Longman, London 1972 and newer editions

Course language:

Notes:

Teaching is carried out in person. If a distance form is required, the lectures will take place online, using the BigBlueButton (https://bbb.science.upjs.sk/). Other conditions will be specified by the teacher.

Course assessment

Total number of assessed students: 43

А	В	С	D	Е	FX
30.23	11.63	20.93	18.6	11.63	6.98

Provides: RNDr. Ján Macko, PhD., RNDr. Ivana Šišoláková, PhD., univerzitná docentka, prof. RNDr. Renáta Oriňaková, DrSc., Mgr. Frederika Chovancová

Date of last modification: 22.07.2022

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II	I Cafémile	I Inizzanaity in Vation
University: P	J Salarik	University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/	Course name: Physical Chemistry II
FCH1b/10	

Course type, scope and the method: Course type: Lecture / Practice

Recommended course-load (hours): Per week: 3 / 2 **Per study period:** 42 / 28

Course method: present

Number of ECTS credits: 6

Recommended semester/trimester of the course: 6.

Course level: I.

Prerequisities: ÚCHV/FCH1a/03 or ÚCHV/FCH1a/21 or ÚCHV/FCHU/10

Conditions for course completion:

1. Participation in seminars (also applies to the online form of teaching). Students are required to attend seminars. The relevant teacher who leads the seminar will justify the reasoned absence of the student (incapacity for work, family reasons, etc.) in a maximum of two seminars during the semester without the need for replacement. In the event of a longer-term reasoned absence (for example due to incapacity for work), the relevant teacher will provide the student with an alternative form of mastering the missed material.

2. Activity at seminars. The preparation of students and their regular monitoring is always assessed by the relevant teacher who conducts the seminar, within his/her competence.

3. Two tests from computational exercises, usually in the 6th and 12th week of the semester. To successfully pass each test, it is necessary to obtain at least 8 points (out of 15 points). Successful completion of continuous tests is a condition of admission to the oral exam.

4. The exam is observed in a regular oral form, resp. in case of restrictions of contact forms of the pedagogical process, the exam is performed by a suitable distance - electronic form.

5. To successfully master the subject, it is necessary to prove mastery of the required curriculum at least 51%.

Learning outcomes:

Students will gain knowledge about the principles that govern the speed of chemical processes, the kinetics and mechanism of some selected reactions, the balance and kinetics of electrode processes. They will also learn the basics of electrochemistry and catalysis.

Brief outline of the course:

Electrochemistry. Equilibrium homogeneous processesn electrolyte solutions. Charge transfer in electrolyte solutions. Nonequilibrium homogeneous processes. Transport processes in electrolyte solutions. Conductance and molar conductivity. Hindering effects. Transport numbers. Equilibrium in heterogeneous electrochemical systems. Pocesses on charged interfaces. Electrochemical cells and fuel cells. Classification of electrode types. Concentration cells. Electrolysis. Electrochemical power sources. Potentiometry. Electrical double layer. Surface tension.

Chemical kinetics. Homogeneous processes. Reaction rate. Reaction order. Classification of chemical reactions. Elementary chemical reactions. Mechanism and kinetics equations of complicated chemical processes. Methods of rate low determination. Theory of chemical kinetics.

Ttemperature dependence of reaction rates. Collision theory. Activated complex theory. Chain reactions. Structure and rate lows of chain reactions. Explosion. Polymerisation reactions. Photochemical reactions. Catalysis. Theory of homogeneous catalysis. Chemical oscillation reactions. Heterogeneous processes. Difusion. Physical and chemical adsorption. Adsorption and diffusion. Processes in heterogeneous electrochemical systems. Electrode kinetics, activation and diffusive mechanism of charge transfer.

Application of theoretical relationships on the solving of concrete problems and on the calculation of examples during seminars.

Recommended literature:

T. Engel, P. Reid : Physical Chemistry, Pearson Educat. Inc., San Francisco 2006 P.W. Atkins : Physical Chemistry,Oxford University Presss, Oxford 1986, 1990, 1994, 1998 W.J. Moore : Physical Chemistry,Longman, London 1972 and newer editions

Course language:

Slovak language

Notes:

Teaching is carried out in person or, if necessary, remotely using the bbb or MS Teams tool. The form of teaching is specified by the teacher at the beginning of the semester, updated continuously.

Course assessment

Total number of assessed students: 623

А	В	С	D	Е	FX
15.41	18.62	22.47	18.46	21.35	3.69

Provides: prof. RNDr. Renáta Oriňaková, DrSc., RNDr. Jana Shepa, PhD., RNDr. Ján Macko, PhD., RNDr. Ivana Šišoláková, PhD., univerzitná docentka

Date of last modification: 25.11.2021

University: P. J. Šaf	ărik University in Košice	3
Faculty: Faculty of	Science	
Course ID: ÚGE/ EXF/21	Course name: Physica	al Geography Excursion
Course type, scope Course type: Pract Recommended co Per week: Per stu Course method: p	ice urse-load (hours): dy period: 6d	
Number of ECTS c	redits: 3	
Recommended sem	ester/trimester of the co	ourse: 4.
Course level: I.		
Prerequisities:		
Conditions for cou	rse completion:	
Learning outcomes	:	
Brief outline of the	course:	
Recommended liter	ature:	
Course language:		
Notes:		
Course assessment Total number of ass	essed students: 43	
	abs	n
	100.0	0.0
Provides: RNDr. Du Imrich Sládek, PhD.	ıšan Barabas, CSc., RND	Dr. Alena Gessert, PhD., univerzitná docentka, Mgr.
Date of last modific	eation: 27.06.2022	
Approved: prof. Mg	zr. Jaroslav Hofierka, Phl	D., prof. RNDr. Vladimír Zeleňák, DrSc.

University: P. J. Ša	fárik Univers	ity in Košice			
Faculty: Faculty of	Science				
Course ID: ÚGE/ FGS1/21	Course na	me: Physical Ge	ography of Slov	akia	
Course type, scope Course type: Lect Recommended co Per week: 2 / 1 Pe Course method: p	ure / Practice urse-load (h er study perio	ours):			
Number of ECTS					
Recommended sen	nester/trimes	ster of the course	e: 4.		
Course level: I.					
Prerequisities:					
Conditions for cou	rse completi	on:			
Learning outcomes	S:				
Brief outline of the	course:				
Recommended lite	rature:				
Course language:					
Notes:					
Course assessment Total number of ass		ts: 75			
A	В	С	D	Е	FX
13.33	26.67	29.33	12.0	6.67	12.0
Provides: RNDr. A	lena Gessert,	PhD., univerzitn	á docentka, doc.	Ing. Katarína Bó	onová, PhD.
Date of last modified	cation: 14.02	2.2023			
Approved: prof. M	gr. Jaroslav H	Iofierka, PhD., p	rof. RNDr. Vladi	mír Zeleňák, Dr	Sc.

	irik University in Košice
Faculty: Faculty of S	Science
Course ID: ÚFV/ FPCh/21	Course name: Physics for Chemists
Course type, scope a Course type: Lectur Recommended cou Per week: 2 / 2 Per Course method: pre	re / Practice rse-load (hours): study period: 28 / 28
Number of ECTS cr	redits: 5
Recommended seme	ester/trimester of the course: 1.
Course level: I.	
Prerequisities:	
problems.	tions during the semester, where students apply the new knowledge by solving dents present theoretical knowledge of the thematic areas listed in the syllabus
Learning outcomes: Completing the cou understand their relat	irse students will get knowledge of fundamental physical laws and wil
 Acceleration of a polysical sector of a polysical sector of a maximum sector	bint mass. taneous velocity, 1D and 3D. oint mass (free fall, angled shot). on a circle. ss point I. dications. Different types of forces. Friction. ss point II. field, potential energy (gravitational, springs). ation of mechanical energy.

7. Fluid mecha	nics II.				
- Fluid dynami					
- Continuity eq					
	ation, application				
-	hysics and thermo ucture of substand	•	winian motion)		
	ibstances, molar n		· · · · · · · · · · · · · · · · · · ·		
	gy. Temperature a	, U		vin).	
-	pacity. Latent heat			,	
-	hysics and thermo	•			
-	te equation, intern				
	modynamics. Isot			esses.	
	conduction, conv rmodynamics. Ent	,			
- Heat engines.	2	uopy.			
-	and magnetism I.				
- Electric charg	ge. Coulomb's law	. Electric field in	tensity and poter	ntial (voltage).	
- Capacitor, cap					
	ent. Ohm's law. Ele	-	irchhoff's laws.		
	and magnetism II. Magnetic induction		Ampara's force	Biot Savart law	
-	of electromagnet		-	Diot-Savart law.	
12. Modern ph	•				
-	troduction to quan	ntum physics.			
- Atomic physi	cs. Nuclear physi	cs, applications.	Elementary parti	cles and cosmolo	gy.
Recommended	l literature:				
	Daniel-Szabó: Zá				
	lad'ar, V. Martišov	vič: Všeobecná fy	zika 1, Mechani	ka a molekulová i	fyzika. Alfa,
Bratislava, 197		2 E1 1 / ¹	(* A 1		00
	: Všeobecná fyzik an, R.B. Leighton,	· ·	0		
4. K.F. Feynna Bratislava, 198		, IVI. Salius. Feyli	manove preunas	KY Z IYZIKY 1-3. A	.11a,
,	ol.: Fyzika v prík	ladoch. Alfa, Bra	tislava, 1983.		
Course langua					
Slovak languag					
Notes:					
Course assessr	nent				
Total number of	of assessed studen	ts: 206			
А	В	С	D	Е	FX
26.7	22.33	24.76	12.62	13.59	0.0
	Mgr. Gregor Bán	ó, PhD., RNDr. Z	Zuzana Jurašekov	vá, PhD., Mgr. An	drej Hovan,
PhD.					
Date of last mo	odification: 22.09	0.2021			

University: P. J. Ša	fárik Universi	ty in Košice			
Faculty: Faculty of	Science				
Course ID: ÚGE/ POL2/21	Course na	me: Political ge	ography		
Course type, scope Course type: Lect Recommended co Per week: 1 / 2 Pe Course method: p	ure / Practice urse-load (ho r study perio	ours):			
Number of ECTS of	credits: 5				
Recommended sem	nester/trimes	ter of the cours	e: 6.		
Course level: I.					
Prerequisities:					
Conditions for cou	rse completio	on:			
Learning outcomes	5:				
Brief outline of the	course:				
Recommended lite	rature:				
Course language:					
Notes:					
Course assessment Total number of ass		s: 16			
A	В	С	D	Е	FX
18.75	37.5	37.5	6.25	0.0	0.0
Provides: RNDr. St	ela Csachová	, PhD., doc. Mg	r. Ladislav Novot	tný, PhD.	
Date of last modified	cation: 27.06	.2022			
Approved: prof. M	gr. Jaroslav H	ofierka, PhD., p	rof. RNDr. Vladi	mír Zeleňák, Dr	Sc.

University: P. J. Ša	afárik Univers	ity in Košice			
Faculty: Faculty o	f Science				
Course ID: ÚGE/ GOBY/21	Course na	me: Population	Geography		
Course type, scop Course type: Lec Recommended co Per week: 2 / 2 P Course method:	ture / Practice ourse-load (h er study perio	ours):			
Number of ECTS	credits: 5				
Recommended set	mester/trimes	ster of the cours	e: 2.		
Course level: I.	· · · ·				
Prerequisities:					
Conditions for co	urse completi	on:			
Learning outcome	25:				
Brief outline of th	e course:				
Recommended lite	erature:				
Course language:					
Notes:					
Course assessmen Total number of as		ts: 123			
A	В	С	D	Е	FX
7.32	4.88	25.2	34.96	21.95	5.69
Provides: doc. Mg docentka	r. Ladislav No	ovotný, PhD., RN	IDr. Janetta Nest	orová-Dická, PhI	D., univerzitná
Date of last modif	ication: 19.02	2.2024			
Approved: prof. M	lgr. Jaroslav H	Iofierka, PhD., p	rof. RNDr. Vladi	mír Zeleňák, Dr	Sc.

University: P. J. Šafárik University in Koši	sice
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Faculty: Faculty of Science

Course ID: ÚCHV/ **Course name:** Porous materials and their applications ADP/03

Course type, scope and the method: Course type: Lecture / Practice

Recommended course-load (hours):

Per week: 2 / 1 Per study period: 28 / 14

Course method: present

Number of ECTS credits: 5

Recommended semester/trimester of the course: 6.

Course level: I., II., III.

Prerequisities:

Conditions for course completion:

Written test in the middle and the end of the semester.

Learning outcomes:

To make the acquaintance of various types of advanced porous solids and basic methods for their investigation. To gen up the students with the methods used in characterisation of specific surface area and pore size of different types of porous materials.

Brief outline of the course:

Terminology and principal terms associated with powders, porous solids and adsorption. Methodology of adsorption at the gas-solid interface, liquid-solid interface. Assessment of surface area and porosity. Inorganic materials (active carbon, metal oxides, zeolites, clay minerals, new advanced materials) and phenomenon of adsorption. Application in the industry and everyday life.

Recommended literature:

1. F. Rouquerol, J. Rouquerol, K. Sing: Adsorption by powders and porous solids, Academic press, London, UK, 1999

2. S. J. Gregg, K.S.W. Sing: Adsorption, surface area and porosity, Academic Press, London,, UK, 1982.

3. V. Zeleňák: Adsorption and porosity of solid substances, internal study text, PF UPJŠ, 2020.

Course language:

Notes:

The course is standardly realized in full-time form, in case of necessary circumstances by distance.

Course assessment

Total number of assessed students: 104

А	В	С	D	Е	FX	Ν	Р
77.88	9.62	3.85	0.0	0.0	0.0	0.0	8.65
Provides: p	orof. RNDr. V	/ladimír Zele	eňák, DrSc.				
Date of last	t modificatio	on: 21.11.202	21				

University: P. J. Šafán	rik University in Košice
Faculty: Faculty of S	cience
Course ID: KPPaPZ/PP/15	Course name: Positive Psychology
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	ce rse-load (hours): dy period: 28
Number of ECTS cro	edits: 2
Recommended seme	ster/trimester of the course: 4., 6.
Course level: I.	
Prerequisities:	
format. Up-to-date in	e completion: on interim evaluation. The subject will be taught in both present and distance formation concerning the subject for the given academic year can be found of the subject in the Academic information system of the UPJŠ.
its main theory, curr rapidly developing fig thinking to the challer	basic knowledge concerning the reasons for founding Positive psychology, ent research, as well as application of Positive psychology as a new and eld within psychology. Students will also gain experience in applying critical nges and issues that Positive psychology brings and raises in the context of the porary society. Emphasis is placed on the ability to critically evaluate current chology.
	ves on well-being nad happiness in psychology oproaches to positive psychology and positivity nal relations wth n rsonality dimension
Deci, E., Ryan R. M., Křivohlavý, J.: Poziti Křivohlavý, J.: Psych	ture: one, M: Emotion and Motivation, Blackwell, 2004 Handbook of Self – Determination Reasearch, Rochester, 2002 vní psychologie. Praha, Portál, 2003 ologie vděčnosti a nevděčnosti. Praha, Grada, 2007 ologie moudrosti a dobrého života, Praha, Grada, 2012

Křivohlavý, J.: Psychologie pocitu štěstí, Grada, 2013 McAdams, D. P., The Person, New York, 2002 Seligman, M. E. P., & Csikszentmihalyi, M. (Eds.). (2000). Positive psychology [Special issue] American Psychologist, 55(1). Říčan, P.: Psychologie náboženství a spirituality, Praha, Portál, 2007 Slezáčková, A.:Pruvodce pozitivní psychologií, Praha, Grada, 2012

Course language:

Notes:

Course assessment

Total number of assessed students: 462

А	В	С	D	Е	FX
98.27	1.3	0.22	0.0	0.22	0.0

Provides: Mgr. Jozef Benka, PhD.

Date of last modification: 24.06.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/ **Course name:** Practical from Inorganic Chemistry PACHU/03

Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 4 Per study period: 56 Course method: present

Number of ECTS credits: 4

Recommended semester/trimester of the course: 2.

Course level: I.

Prerequisities: ÚCHV/VCHU/14 or ÚCHV/VCHU/15 or ÚCHV/VCHU/10 or ÚCHV/VACH/10

Conditions for course completion:

Learning outcomes:

Acquisition of practical skills and knowledge necessary for work in a chemical laboratory in the preparation of inorganic and other compounds, in the preparation of solutions, methods of distillation and other basic techniques of work in the laboratory. Students will also be able to perform basic characterization of substances and proof reactions.

Brief outline of the course:

The utilization of common laboratory techniques for preparation of elements (H2, O2, Cu, Ni), oxides(CO2, Al2O3·xH2O), nitrides(Mg3N2), acids (HNO3, H3BO3), salts((NH4)2SO4, KMnO4), binary salts(NH4)Fe(SO4)2·12H2O), halides (CuCl, CuCl2·2H2O, CuBr2) and coordination compounds [Cu(NH3)4]SO4·H2O, K3[Al(C2O4)3]·3H2O).

Recommended literature:

J. Černák, J. Bubanec, M. Dzurillová, V. Zeleňák: Praktikum z anorganickej chémie. UPJŠ Košice, 1999.

Z. Vargová, J. Kuchár: Základné praktikum z anorganickej chémie, UPJŠ, Košice, 2009. Z.Vargova, M.Almáši, J. Kuchár, J.Dinajová: Základné laboratórne cvičenia z anorganickej chémie, ŠafárikPress, 2020.

Course language:

Notes:

Course assessment

Total number of assessed students: 661

А	В	С	D	Е	FX
54.16	27.08	13.77	2.57	1.66	0.76

Provides: doc. RNDr. Juraj Kuchár, PhD., RNDr. Martin Vavra, PhD., RNDr. Miroslava Matiková Maľarová, PhD., prof. RNDr. Zuzana Vargová, Ph.D., Mgr. Michaela Rendošová, PhD.

Date of last modification: 22.07.2022

University: P. J. Šafárik University in Košice

Faculty: Faculty of Science

Course ID: ÚCHV/	Course name: Practical in Physical Chemistry
PFCU/22	

Course type, scope and the method: Course type: Practice Recommended course-load (hours):

Per week: 4 Per study period: 56

Course method: present

Number of ECTS credits: 4

Recommended semester/trimester of the course: 5.

Course level: I.

Prerequisities: ÚCHV/FCHU/21 or ÚCHV/FCHU/22 or ÚCHV/FCHU/10

Conditions for course completion:

1. Adequate theoretical preparation for individual tasks of experimental practice according to the recommended literature.

2. Passing tasks with relevant results.

- 3. Processing of experimental work results in the form of a protocols and its acceptance.
- 4. Assessment of theoretical knowledges and practical skills.

In the case of distance learning:

1. Elaboration of a paper on a selected topic and its presentation.

2. Theoretical preparation in the form of protocols, where the basic principles of individual tasks are stated.

3. Teaching is realized in blocks without limiting the scope in the alternative term.

Learning outcomes:

Theoretical principles, description of each technique and appropriate physical chemistry experiments.

Brief outline of the course:

Experimental verification of theoretical knowledge on thermodynamics, thermochemistry, chemical equilibria (determination of enthalpy, phase diagrams), colligative properties (cryoscopy, ebulioscopy), adsorption.

Experimental verification of theoretical knowledge on electrochemistry (conductivity, dissociation constants, activity coefficients, electromotive force of galvanic cell, Daniell cell, potentials, polarography) and chemical kinetics (determination of rate constants).

Recommended literature:

B.P. Levitt: Findlay's Practical Physical Chemistry, Longman, London 1973

W.J. Moore: Physical Chemistry, Longman, London 1972

P.W. Atkins: Physical Chemistry, Oxford University Press, Oxford, New York 2002

Course language:

Notes:

Teaching is carried out in person. If a distance form is required, the conditions will be specified by the teacher.

by the tedenet.					
Course assessm Total number of	lent f assessed studen	ts: 47			
А	В	С	D	Е	FX
97.87	2.13	0.0	0.0	0.0	0.0
Provides: RND Radka Gorejová		vský, RNDr. Jan	a Shepa, PhD., R	NDr. Ján Macko	, PhD., RNDr.
Date of last mo	dification: 22.07	7.2022			
Approved: prof	f. Mgr. Jaroslav H	Iofierka, PhD., p	rof. RNDr. Vladi	mír Zeleňák, Dr	Sc.

University: P. J. Š	Šafárik Univers	ity in Košice			
Faculty: Faculty	of Science				
Course ID: KPPaPZ/Ps/15	Course na	me: Psychology			
Course type, scop Course type: Le Recommended Per week: 2 Per Course method	cture course-load (h study period:	ours):			
Number of ECTS					
Recommended so	emester/trimes	ter of the cours	e: 3.		
Course level: I.					
Prerequisities:					
Conditions for co	ourse completi	on:			
Learning outcom	nes:				
Brief outline of t	he course:				
Recommended li	terature:				
Course language	•				
Notes:					
Course assessme Total number of a	-	ts: 870			
A	В	С	D	Е	FX
37.47	21.15	15.98	12.41	11.26	1.72
Provides: doc. M	gr. Gabriel Ban	ík, PhD.			
Date of last modi	ification: 24.06	.2022			
Approved: prof.	Mgr. Jaroslav H	lofierka, PhD., p	rof. RNDr. Vlad	imír Zeleňák, Dr	Sc.

KPPaPZ/PKŽ/15 Course type, scope and the Course type: Practice Recommended course-loa Per week: 2 Per study per Course method: present Number of ECTS credits: Recommended semester/tr Course level: I. Prerequisities: Conditions for course com The evaluation of the course set requirements, which wil ensure an objective and fair moral standards. There is n process or in the assessmen 1. Active participation in se 2. Elaboration and presenta points 20; minimum numbe 3. Elaboration of an essay i	rse name: Psychology of Everyday Life e method: ad (hours): riod: 28 2 rimester of the course: 3.
KPPaPZ/PKŽ/15 Course type, scope and the Course type: Practice Recommended course-loa Per week: 2 Per study per Course method: present Number of ECTS credits: Recommended semester/tr Course level: I. Prerequisities: Conditions for course com The evaluation of the course set requirements, which wil ensure an objective and fair moral standards. There is n process or in the assessmen 1. Active participation in se 2. Elaboration and presenta points 20; minimum numbe 3. Elaboration of an essay i	e method: ad (hours): riod: 28 2 rimester of the course: 3.
Course type: Practice Recommended course-loa Per week: 2 Per study per Course method: present Number of ECTS credits: Recommended semester/tr Course level: I. Prerequisities: Conditions for course com The evaluation of the course set requirements, which wil ensure an objective and fair moral standards. There is n process or in the assessmen 1. Active participation in se 2. Elaboration and presenta points 20; minimum numbe 3. Elaboration of an essay i	ad (hours): riod: 28 2 rimester of the course: 3.
Recommended semester/tr Course level: I. Prerequisities: Conditions for course com The evaluation of the course set requirements, which wil ensure an objective and fair moral standards. There is n process or in the assessmen 1. Active participation in se 2. Elaboration and presenta points 20; minimum numbe 3. Elaboration of an essay i	rimester of the course: 3.
Course level: I. Prerequisities: Conditions for course com The evaluation of the course set requirements, which wil ensure an objective and fair moral standards. There is n process or in the assessmen 1. Active participation in se 2. Elaboration and presenta points 20; minimum number 3. Elaboration of an essay i	pletion:
Prerequisities: Conditions for course com The evaluation of the course set requirements, which wil ensure an objective and fair moral standards. There is n process or in the assessmen 1. Active participation in se 2. Elaboration and presenta points 20; minimum numbe 3. Elaboration of an essay i	-
Conditions for course com The evaluation of the course set requirements, which wil ensure an objective and fair moral standards. There is n process or in the assessmen 1. Active participation in se 2. Elaboration and presenta points 20; minimum numbe 3. Elaboration of an essay i	-
The evaluation of the course set requirements, which will ensure an objective and fair moral standards. There is n process or in the assessmen 1. Active participation in se 2. Elaboration and presenta points 20; minimum number 3. Elaboration of an essay i	-
minimum number of points The final evaluation (grade) A 40b - 37b B 36b - 33b C 32b - 29b D 28b - 25b E 24b - 21b FX 20b - 0b Learning outcomes:	Il be set in advance and will not change. The aim of the assessment is to r mapping of the student's knowledge while adhering to all ethical and no tolerance for students' fraudulent behavior, whether in the teaching at process. eminars ation of PPT presentation on the assigned topic. Maximum number of er of points 11. in the range of 4xA4 (standard pages). Maximum number of points 20

The student is able to describe, explain and evaluate the psychological mechanisms that occur in everyday situations.

The student is able to apply basic psychological knowledge to himself (self-regulation) but also in interaction with others (cooperation).

The method of teaching the subject will be oriented to the student. Speakers will be interested in the needs, expectations and opinions of students so as to encourage them to think critically by expressing respect and feedback on their opinions and needs.

The content of the curriculum will be based on primary and high-quality sources that will reflect the topicality of the topics so as to ensure the connection of the curriculum with other subjects and also

the connection of the curriculum with practice. Students will be expected to take an active approach in lectures and seminars with an emphasis on their independence and responsibility.

Brief outline of the course:

How to understand human behavior (overview of basic approaches in psychology); Basic overview of cognitive processes; Learning processes and their use in practice; Social influences, prosocial and antisocial behavior; How human emotions and motivations work; Deciding - why and when we take risks; Childhood experiences and their relationship to adulthood; Abnormal behavior, mental disorders and therapeutic approaches

Recommended literature:

Course language:

Notes:

Course assessment

Total number of assessed students: 230

А	В	С	D	Е	FX
41.74	25.22	26.52	4.78	1.3	0.43

Provides: Mgr. Ondrej Kalina, PhD.

Date of last modification: 12.09.2024

University: P. J. Š	afárik Univers	ity in Košice			
Faculty: Faculty o	of Science				
Course ID: ÚGE/ RGE2/21	E/ Course name: Regional Geography of Europe				
Course type, scop Course type: Lec Recommended c Per week: 3 / 1 P Course method:	cture / Practice ourse-load (h 'er study perio	ours):			
Number of ECTS	credits: 5				
Recommended se	mester/trimes	ster of the cours	e: 6.		
Course level: I.					
Prerequisities:					
Conditions for co	urse completi	on:			
Learning outcom	es:				
Brief outline of th	e course:				
Recommended lit	erature:				
Course language:					
Notes:					
Course assessmen Total number of as		ts: 43			
A	В	С	D	E	FX
6.98	18.6	32.56	37.21	0.0	4.65
Provides: RNDr. S Mgr. Ladislav Nov			,	,	,
Date of last modif	fication: 27.06	5.2022			
Approved: prof. N	/Igr. Jaroslav H	Iofierka, PhD., p	rof. RNDr. Vlad	imír Zeleňák, Dr	Sc.

University: P. J.	Šafárik Univers	ity in Košice			
Faculty: Faculty	of Science				
Course ID: ÚGE ADPZ/22	Course name: Remote sensing applications				
Course type, sco Course type: Le Recommended Per week: 1 / 2 Course method	ecture / Practice course-load (h Per study peri	e ours):			
Number of ECT	S credits: 3				
Recommended s	emester/trimes	ster of the cours	e: 5.		
Course level: I.,	II.				
Prerequisities:					
Conditions for c	ourse completi	on:			
Learning outcon	nes:				
Brief outline of t	he course:				
Recommended li	iterature:				
Course language	2:				
Notes:					
Course assessme Total number of		ts: 11			
А	В	С	D	Е	FX
100.0	0.0	0.0	0.0	0.0	0.0
Provides: prof. M Onačillová, PhD.	•		c. RNDr. Ján Ka	ňuk, PhD., Mgr. 1	Katarína
Date of last mod	ification: 20.06	5.2022			
Approved: prof.	Mgr. Jaroslav H	Iofierka, PhD., p	rof. RNDr. Vlad	imír Zeleňák, Dr	Sc.

University: P. J. Šat	árik University in Koši	ce		
Faculty: Faculty of	Science			
Course ID: KPPaPZ/RKS/14	Course name: Resol	Course name: Resolving Conflict Situations in Educational Practice		
Course type, scope Course type: Lect Recommended co Per week: 1 / 2 Pe Course method: p	ure / Practice urse-load (hours): r study period: 14 / 28			
Number of ECTS of	redits: 4			
Recommended sem	ester/trimester of the	course: 3., 5.		
Course level: I.				
Prerequisities:				
Conditions for cou	rse completion:			
Learning outcomes	:			
Brief outline of the	course:			
Recommended lite	ature:			
Course language:				
Notes:				
Course assessment Total number of ass	essed students: 179			
	abs n			
94.41 5.59				
Provides: PhDr. An	na Janovská, PhD.			
Date of last modifie	cation: 27.05.2024			
Approved: prof. Mg	gr. Jaroslav Hofierka, P	hD., prof. RNDr. Vladimír Zeleňák, DrSc.		

University: P. J. Ša	fárik Universi	ity in Košice			
Faculty: Faculty of	Science				
Course ID: KPE/ OLŠ/15	Course name: School Administration and Legislation				
Course type, scope Course type: Prac Recommended co Per week: 2 Per s Course method: p	tice ourse-load (ho tudy period:	ours):			
Number of ECTS					
Recommended sen	nester/trimes	ter of the cours	e: 3., 5.		
Course level: I.					
Prerequisities:					
Conditions for cou	rse completio	on:			
Learning outcome	s:				
Brief outline of the	e course:				
Recommended lite	rature:				
Course language:					
Notes:					
Course assessment Total number of ass		ts: 325			
А	В	С	D	Е	FX
45.23	29.85	14.46	6.46	3.38	0.62
Provides: PaedDr. 1	Michal Novoc	ký, PhD.	1	1	
Date of last modifi	cation: 14.09	.2024			
Approved: prof. M	gr. Jaroslav H	ofierka, PhD., p	rof. RNDr. Vlad	imír Zeleňák, Dr	Sc.

University: P. J. Šafárik University in Košice				
Faculty: Faculty of Science				
Course ID: ÚTVŠ/ ÚTVŠ/CM/13	Course name: Seaside Aerobic Exercise			
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	ce rse-load (hours): dy period: 28			
Number of ECTS cro	edits: 2			
Recommended seme	ster/trimester of the course:			
Course level: I., II.				
Prerequisities:				
- active participation	e completion: sful course completion: in line with the study rule of procedure and course guidelines ce of all tasks- aerobics, water exercise, yoga, Pilates and others			
course syllabus and re Performance standard Upon completion of t - perform basic aerob - conduct verbal and p	rates relevant knowledge and skills in the field, which content is defined in the ecommended literature. d: the course students are able to meet the performance standard and: bics steps and basics of health exercises, non-verbal communication with clients during exercise, the process of physical recreation in leisure time			
Brief outline of the c Brief outline of the co 1. Basic aerobics – lo 2. Basics of aqua fithe 3. Basics of Pilates 4. Health exercises 5. Bodyweight exerci 6. Swimming	ourse: w impact aerobics, high impact aerobics, basic steps and cuing ess			

 ŽECHOVSKÁ, I., MILEROVÁ, H., NOVOTNÁ, V. Aqua-fitness. Praha: Grada. 136 s. EVANS, M., HUDSON, J., TUCKER, P. 2001. Umění harmonie: meditace, jóga, tai-či, strečink. 192 s. JARKOVSKÁ, H., JARKOVSKÁ, M. 2005. Posilováni s vlastním tělem 417 krát jinak. Praha: Grada. 209 s. KOVAŘÍKOVÁ, K. 2017. Aerobik a fitness. Karolium, 130 s. 			
Course language: Slovak language			
Notes:			
Course assessment Total number of assessed students: 62			
abs	n		
9.68 90.32			
Provides: Mgr. Agata Dorota Horbacz, PhD.			
Date of last modification: 29.03.2022			
Approved: prof. Mgr. Jaroslav Hofierka, PhD., pr	rof. RNDr. Vladimír Zeleňák, DrSc.		

University: P. J. Ša	afárik Univers	ity in Košice			
Faculty: Faculty o	f Science				
Course ID: KF/ VKFV/07	Course name: Selected Topics in Philosophy of Education (General Introduction)				
Course type, scop Course type: Prac Recommended co Per week: 2 Per s Course method:	ctice ourse-load (h study period:	ours):			
Number of ECTS	credits: 2				
Recommended ser	mester/trimes	ster of the cours	e: 3., 5.		
Course level: I.					
Prerequisities:					
Conditions for cou	urse completi	on:			
Learning outcome	es:				
Brief outline of th	e course:				
Recommended lite	erature:				
Course language:					
Notes:					
Course assessmen Total number of as		ts: 33			
А	В	С	D	Е	FX
66.67	18.18	12.12	3.03	0.0	0.0
Provides: PhDr. D	ušan Hruška, I	PhD.			
Date of last modif	ication: 13.04	.2022			
Approved: prof. N	Igr. Jaroslav H	Iofierka, PhD., p	rof. RNDr. Vlad	imír Zeleňák, Dr	Sc.

University: P. J. Šafá	University: P. J. Šafárik University in Košice				
Faculty: Faculty of S	cience				
Course ID: KPPaPZ/ECo-C2/14	Course name: Self Marketing ECo-C2				
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	ce rse-load (hours): dy period: 28				
Number of ECTS cr	edits: 4				
Recommended seme	ster/trimester of the course: 4., 6.				
Course level: I.					
Prerequisities:					
according to the teach Detailed information	n in lessons (absence is allowed max. 90 min.), 2. Realization of assignments				
knows the possibilitie knowledge and princ competencies, his / h knowledge and socia	to understand and explain the basic assumptions of good self-marketing, es for the correct presentation of his own person and understands the related iples of personal and communication area. He / she can understand his / her her goals, how to make his / her strengths visible and he / she can apply this and professional skills in the personal and professional sphere of his / her mprove his / her employment opportunities.				
Me and my influence me? Ability to defend options do I have?), Competence (Have y at work),					
GRADA, 2008. 408 s VÝROST, Jozef - SL instituce. 1. vyd. Prak KOMÁRKOVÁ, Růž	AMĚNÍK, Ivan. Sociální psychologie. 2., přepr. a rozš. vyd. Praha :				

VÝROST, Jozef - SLAMĚNÍK, Ivan. Aplikovaná sociální psychologie II. 1. vyd. Praha : Grada Publishing, 2001. 260 s.

Course language: slovak		
Notes: After passing the certification exams from all 4 me Management, Communication) the student will red	· · · · · · · · · · · · · · · · · · ·	
Course assessment Total number of assessed students: 171		
abs	n	
90.64 9.36		
Provides: Mgr. Ondrej Kalina, PhD.		
Date of last modification: 12.09.2024		
Approved: prof. Mgr. Jaroslav Hofierka, PhD., pro	of. RNDr. Vladimír Zeleňák, DrSc.	

University: P. J. Šaf	árik University in Košice			
Faculty: Faculty of	Science			
Course ID: ÚGE/ SHG/21	Course name: Seminar of	Course name: Seminar of human geography		
Course type, scope Course type: Pract Recommended cou Per week: 2 Per st Course method: pr	ice 1 rse-load (hours): udy period: 28			
Number of ECTS c	redits: 3			
Recommended sem	ester/trimester of the cours	se: 6.		
Course level: I.				
Prerequisities:				
Conditions for cour	se completion:			
Learning outcomes	:			
Brief outline of the	course:			
Recommended liter	ature:			
Course language:				
Notes:				
Course assessment Total number of ass	essed students: 10			
	abs	n		
	90.0 10.0			
6	· · · · ·	a Csachová, PhD., RNDr. Janetta Nestorová- dislav Novotný, PhD., Mgr. Loránt Pregi, PhD.		
Date of last modific	ation: 27.06.2022			
Approved: prof. Mg	r. Jaroslav Hofierka. PhD. 1	prof. RNDr. Vladimír Zeleňák, DrSc.		

University: P. J. Šafa	árik University in Košice		
Faculty: Faculty of S	Science		
Course ID: ÚGE/ SFG/21	Course name: Seminar of physical geography		
Course type, scope a Course type: Practi Recommended cou Per week: 2 Per stu Course method: pr	ice irse-load (hours): udy period: 28		
Number of ECTS c	redits: 3		
Recommended sem	ester/trimester of the cours	e: 6.	
Course level: I.			
Prerequisities:			
Conditions for cour	se completion:		
Learning outcomes			
Brief outline of the	course:		
Recommended liter	ature:		
Course language:			
Notes:			
Course assessment Total number of asse	essed students: 0		
	abs	n	
	0.0 0.0		
Provides: RNDr. Du PhD., univerzitná do	, , , ,	Katarína Bónová, PhD., RNDr. Alena Gessert,	
Date of last modific	ation: 27.06.2022		
Approved: prof. Mg	r. Jaroslav Hofierka, PhD., p	rof. RNDr. Vladimír Zeleňák, DrSc.	

University:	ΡJ	Šafárik	University	in Košice
Chiver Siey.	1.0.	Suluin	Oniversity .	

Faculty: Faculty of Science

Course ID: ÚCHV/	Course name: Separation Methods
ASM/03	

Course type, scope and the method: Course type: Lecture / Practice Recommended course-load (hours):

Per week: 2 / 1 Per study period: 28 / 14

Course method: present

Number of ECTS credits: 5

Recommended semester/trimester of the course: 6.

Course level: I.

Prerequisities: (ÚCHV/ANCHU/03 or ÚCHV/ANCHU/21 or ÚCHV/ANCHE/09 or ÚCHV/ ANCH1b/03 or ÚCHV/ANCH1b/21) and (ÚCHV/PAEC/03 or ÚCHV/PANCH/06 or ÚCHV/ PANCHE/09 or ÚCHV/PACU/03)

Conditions for course completion:

1. Preparation and presentation of a project focused on the application of separation methods.

2. Examination. The exam consists of 3 questions (each of 33%), 50% must be obtained for the pass exam.

Learning outcomes:

Survey of basic principles, theoretical background and applications of separation methods in research and analytical practice.

Brief outline of the course:

Basic principles, classification, theory and applications of separation methods. Extraction - LLE, SPE, SPME. Chromatographic methods - theory, classification. Gas chromatography, stationary phases. Instrumentation, detectors in GC. Data evaluation - qualitative and quantitative analysis. High-performance liquid chromatography, principles, classification. Stationary and mobile phases in LC, instrumentation. Applications.

Planar chromatographic methods - TLC, HPTLC, PC.

Electrophoretic techniques and their applications.

Recommended literature:

Skoog D. A., Leary J. J.: Principles of instrumental analysis. Saunders College Publishing, New York 1997.

Pawliszyn J., Lord H. L.: Handbook of sample preparation, Wiley 2010.

Current scientific literature

Course language:

Slovak, english language

Notes:

Course assessm	nent f assessed studen	ts: 506			
		C	D	Е	FX
А	В	C	D	E	ГА
28.66	26.09	25.1	12.65	5.34	2.17
Provides: doc. RNDr. Taťána Gondová, CSc.					
Date of last modification: 01.08.2022					
Approved: prof	Approved: prof. Mgr. Jaroslav Hofierka, PhD., prof. RNDr. Vladimír Zeleňák, DrSc.				

University: P. J. Šat	fárik University in Košice
Faculty: Faculty of	Science
Course ID: KPO/ SPKVV/15	Course name: Social and Political Context of Education
Course type, scope Course type: Lect Recommended co Per week: 2 Per st Course method: p	ure urse-load (hours): tudy period: 28
Number of ECTS of	redits: 2
Recommended sem	nester/trimester of the course: 4., 6.
Course level: I.	
Prerequisities:	
Conditions for cou Evaluation of the de A 100,00% - 91,0 B 90,99% - 81,00 C 80,99% - 71,00 D 70,99% - 61,00 E 60,99% - 51,00 FX 50,99% and I	eveloped assignment. 00% 0% 0% 0%
Learning outcomes The aim and purpos	se of teaching the subject is to impart knowledge and promote reflection on th

The aim and purpose of teaching the subject is to impart knowledge and promote reflection on the issues of education and training in the context of social and political change.

Development of knowledge: the student will be able to know the current theoretical background related to the process of education and training in a modern democratic society.

The student will be able to navigate the social and political space - politically, legally, socially and culturally. He/she will be able to look for alternatives and solutions to dysfunctions, while at the same time exploiting opportunities and ways to implement them.

Brief outline of the course:

The status, role and functions of education in human life and society. The political, social and economic objectives of education. Education, learning and social change in the context of globalisation. Macrosocial determinants of education. Current roles of education and training in modern performance and democratic society.

Recommended literature:

Domestic and foreign journal literature

Kudláčová, B.(2007) Človek a výchova v dejinách európskeho myslenia. Trnava: PdF TU Zeus Leonardo (2010) Handbook of Cultural Politics and Education. Rotterdam, The Netherlands.

Course language:

Slovak

Notes:

Course assessm	ient					
Total number of	f assessed studen	ts: 201				
А	В	С	D	Е	FX	
60.7	0.7 20.9 10.95 4.48 1.49 1.49					
Provides: Mgr.	Ján Ruman, PhD			•		
Date of last mo	dification: 13.04	.2022				
Approved: prof	f. Mgr. Jaroslav H	Iofierka, PhD., p	rof. RNDr. Vlad	imír Zeleňák, DrS	Sc.	

University: P. J. Šafárik University in Košice
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Faculty: Faculty of Science

Course ID: KGER/	Course name: Specialised German Language - Natural Sciences 1
OJPV1/07	

Course type, scope and the method: Course type: Practice Recommended course-load (hours): Per week: 2 Per study period: 28

Course method: present

Number of ECTS credits: 2

Recommended semester/trimester of the course: 4.

Course level: I.

Prerequisities:

Conditions for course completion:

Active participation in class and completed homework assignments. Students are allowed to miss 2 classes at the most (2x90 min.). 1 control tests during the semester and written assignments. Final grade will be calculated as follows: A 93-100 %, B 86-92%, C 79-85%, D 72-78%, E 65-71%, FX 64 % and less.

Learning outcomes:

The development of students' language skills - reading, writing, listening, speaking, improvement of their linguistic competence - students acquire knowledge of selected phonological, lexical and syntactic aspects, development of pragmatic competence - students can effectively use the language for a given purpose, with focus on Academic English and English for specific/professional purposes - Natural Science, level B1.

Brief outline of the course:

Recommended literature:

Duden Basiswissen Schule. Abitur: Enthält die Bände Mathematik, Physik, Chemie, Biologie, Geographie, Geschichte. (2007). ISBN: 978-3411002511.

Zettl, E. et al.: Aus moderner Technik und Naturwissenschaft. Ismaning: Hueber, 2003.

Reiss, K.: Basiswissen Zahlentheorie: Eine Einführung in Zahlen und Zahlbereiche (Mathematik für das Lehramt), Springer, 2007. ISBN: 978-3540453772.

Meyer, L., Schmidt, G.- D.: Basiswissen Ausbildung: Physik. Bildungsverlag EINS, 2008. ISBN: 978-3427799337.

Duden. Schülerduden Biologie: Das Fachlexikon von A-Z. Bibliographisches Institut Berlin, 2009. ISBN: 978-3411054275.

Mortimer, Ch. E., Müller, U., Beck, J.: Chemie: Das Basiswissen der Chemie. Stuttgart: Thieme, 2014. ISBN: 978-313484311

Deutsch perfekt, GEO, MaxPlanck Forschung a iné printové a elektronické médiá

Course	language:
Germar	n

Notes:

Course assessm		ts: 149				
Total number of assessed students: 149ABCDEFX						
24.16	24.16 23.49 24.16 20.13 7.38 0.67					
Provides: Mgr. Ulrika Strömplová, PhD.						
Date of last modification: 09.02.2023						
Approved: prof	f. Mgr. Jaroslav H	Iofierka, PhD., p	rof. RNDr. Vladi	mír Zeleňák, Dr	Sc.	

Faculty: Faculty of S	cience
Course ID: ÚTVŠ/ TVa/11	Course name: Sports Activities I.
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	ce rse-load (hours): dy period: 28
Number of ECTS cr	edits: 2
Recommended seme	ster/trimester of the course: 1.
Course level: I., II.	
Prerequisities:	
Conditions for cours Min. 80% of active p	e completion: articipation in classes.
They have a great in	their forms prepare university students for their professional and personal life pact on physical fitness and performance. Specialization in sports activitie strengthen their relationship towards the selected sport in which they also
activities aerobics; ai yoga, power yoga, p tennis, chess, volleyb Additionally, the Inst offers winter courses	ourse: ical education and sport at the Pavol Jozef Šafárik University offers 20 sport kido, basketball, badminton, body-balance, body form, bouldering, floorball ilates, swimming, fitness, indoor football, SM system, step aerobics, table
[online] Dostupné na BUZKOVÁ, K. 2006 8024715252. JARKOVSKÁ, H, JA Grada. ISBN 978802 KAČÁNI, L. 2002. F 8089197027. KRESTA, J. 2009. Fu LAWRENCE, G. 201	05. Plávanie. Banská Bystrica: FHV UMB. 198s. ISBN 80-8083-140-8. : https://www.ff.umb.sk/app/cmsFile.php?disposition=a&ID=571 5. Fitness jóga, harmonické cvičení těla I duše. Praha: Grada. ISBN ARKOVSKÁ, M. 2005. Posilování s vlastním tělem 417 krát jinak. Praha:

STACKEOVÁ, D. 2014. Fitness programy z pohledu kinantropologie. Praha: Galén. ISBN 9788074921155.

VOMÁČKO, S. BOŠTÍKOVÁ, S. 2003. Lezení na umělých stěnách. Praha: Grada. 129s. ISBN 8024721743.

Course language:

Slovak language

Notes:

Course assessment

Total number of assessed students: 15203

abs	abs-A	abs-B	abs-C	abs-D	abs-E	n	neabs
86.07	0.07	0.0	0.0	0.0	0.05	8.67	5.15

Provides: Mgr. Patrik Berta, Mgr. Agata Dorota Horbacz, PhD., Mgr. Dávid Kaško, PhD., Mgr. Ladislav Kručanica, PhD., Mgr. Richard Melichar, Mgr. Petra Tomková, PhD., Mgr. Marcel Čurgali, Mgr. Alena Buková, PhD., univerzitná docentka, doc. PaedDr. Ivan Uher, MPH, PhD., prof. RNDr. Stanislav Vokál, DrSc., Mgr. Zuzana Küchelová, PhD.

Date of last modification: 07.02.2024

University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	science
Course ID: ÚTVŠ/ TVb/11	Course name: Sports Activities II.
Course type, scope a Course type: Practi Recommended cou Per week: 2 Per stu Course method: pro	ce rse-load (hours): ıdy period: 28
Number of ECTS cr	redits: 2
Recommended seme	ester/trimester of the course: 2.
Course level: I., II.	
Prerequisities:	
Conditions for cours active participation in	se completion: n classes - min. 80%.
They have a great in	I their forms prepare university students for their professional and personal life npact on physical fitness and performance. Specialization in sports activities strengthen their relationship towards the selected sport in which they also
activities aerobics; at yoga, power yoga, p tennis, chess, volley Additionally, the Ins offers winter courses	ourse: ical education and sport at the Pavol Jozef Šafárik University offers 20 sports ikido, basketball, badminton, body-balance, body form, bouldering, floorball bilates, swimming, fitness, indoor football, SM system, step aerobics, table
[online] Dostupné na BUZKOVÁ, K. 2000 8024715252. JARKOVSKÁ, H, JA Grada. ISBN 978802 KAČÁNI, L. 2002. H 8089197027. KRESTA, J. 2009. F LAWRENCE, G. 20	 005. Plávanie. Banská Bystrica: FHV UMB. 198s. ISBN 80-8083-140-8. a: https://www.ff.umb.sk/app/cmsFile.php?disposition=a&ID=571 6. Fitness jóga, harmonické cvičení těla I duše. Praha: Grada. ISBN ARKOVSKÁ, M. 2005. Posilování s vlastním tělem 417 krát jinak. Praha:

STACKEOVÁ, D. 2014. Fitness programy z pohledu kinantropologie. Praha: Galén. ISBN 9788074921155.

VOMÁČKO, S. BOŠTÍKOVÁ, S. 2003. Lezení na umělých stěnách. Praha: Grada. 129s. ISBN 8024721743.

Course language:

Slovak language

Notes:

Course assessment

Total number of assessed students: 13788

abs	abs-A	abs-B	abs-C	abs-D	abs-E	n	neabs
83.84	0.49	0.01	0.0	0.0	0.04	11.18	4.43

Provides: Mgr. Agata Dorota Horbacz, PhD., Mgr. Dávid Kaško, PhD., Mgr. Marcel Čurgali, Mgr. Patrik Berta, Mgr. Ladislav Kručanica, PhD., Mgr. Richard Melichar, Mgr. Petra Tomková, PhD., Mgr. Alena Buková, PhD., univerzitná docentka, doc. PaedDr. Ivan Uher, MPH, PhD., prof. RNDr. Stanislav Vokál, DrSc., Mgr. Zuzana Küchelová, PhD.

Date of last modification: 07.02.2024

University: P. J. Šafa	árik University in Košice
Faculty: Faculty of S	Science
Course ID: ÚTVŠ/ TVc/11	Course name: Sports Activities III.
Course type, scope a Course type: Practi Recommended cou Per week: 2 Per stu Course method: pr	ice irse-load (hours): udy period: 28
Number of ECTS cr	redits: 2
Recommended sem	ester/trimester of the course: 3.
Course level: I., II.	
Prerequisities:	
Conditions for cour min. 80% of active p	se completion: participation in classes
They have a great in	their forms prepare university students for their professional and personal life. npact on physical fitness and performance. Specialization in sports activities strengthen their relationship towards the selected sport in which they also
activities aerobics; a yoga, power yoga, j tennis, chess, volley Additionally, the Ins offers winter course	course: sical education and sport at the Pavol Jozef Šafárik University offers 20 sports ikido, basketball, badminton, body-balance, body form, bouldering, floorball, pilates, swimming, fitness, indoor football, SM system, step aerobics, table
[online] Dostupné na BUZKOVÁ, K. 200 8024715252. JARKOVSKÁ, H, J. Grada. ISBN 978802 KAČÁNI, L. 2002. I 8089197027. KRESTA, J. 2009. F LAWRENCE, G. 20	005. Plávanie. Banská Bystrica: FHV UMB. 198s. ISBN 80-8083-140-8. a: https://www.ff.umb.sk/app/cmsFile.php?disposition=a&ID=571 6. Fitness jóga, harmonické cvičení těla I duše. Praha: Grada. ISBN ARKOVSKÁ, M. 2005. Posilování s vlastním tělem 417 krát jinak. Praha:

STACKEOVÁ, D. 2014. Fitness programy z pohledu kinantropologie. Praha: Galén. ISBN 9788074921155.

VOMÁČKO, S. BOŠTÍKOVÁ, S. 2003. Lezení na umělých stěnách. Praha: Grada. 129s. ISBN 8024721743.

Course language:

Slovak language

Notes:

Course assessment

Total number of assessed students: 9104

abs	abs-A	abs-B	abs-C	abs-D	abs-E	n	neabs
88.38	0.07	0.01	0.0	0.0	0.02	4.46	7.06

Provides: Mgr. Marcel Čurgali, Mgr. Agata Dorota Horbacz, PhD., Mgr. Dávid Kaško, PhD., Mgr. Patrik Berta, Mgr. Ladislav Kručanica, PhD., Mgr. Richard Melichar, Mgr. Petra Tomková, PhD., Mgr. Alena Buková, PhD., univerzitná docentka, doc. PaedDr. Ivan Uher, MPH, PhD., prof. RNDr. Stanislav Vokál, DrSc., Mgr. Zuzana Küchelová, PhD.

Date of last modification: 07.02.2024

University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	cience
Course ID: ÚTVŠ/ TVd/11	Course name: Sports Activities IV.
Course type, scope a Course type: Practic Recommended cou Per week: 2 Per stu Course method: pre	ce rse-load (hours): Idy period: 28
Number of ECTS cr	edits: 2
Recommended seme	ster/trimester of the course: 4.
Course level: I., II.	
Prerequisities:	
Conditions for cours min. 80% of active p	se completion: articipation in classes
They have a great in	their forms prepare university students for their professional and personal life. npact on physical fitness and performance. Specialization in sports activities strengthen their relationship towards the selected sport in which they also
activities aerobics; ai yoga, power yoga, p tennis, chess, volleyb Additionally, the Ins offers winter courses	ourse: ical education and sport at the Pavol Jozef Šafárik University offers 20 sports kido, basketball, badminton, body-balance, body form, bouldering, floorball, bilates, swimming, fitness, indoor football, SM system, step aerobics, table
[online] Dostupné na BUZKOVÁ, K. 2006 8024715252. JARKOVSKÁ, H, JA Grada. ISBN 978802 KAČÁNI, L. 2002. F 8089197027. KRESTA, J. 2009. F LAWRENCE, G. 202	05. Plávanie. Banská Bystrica: FHV UMB. 198s. ISBN 80-8083-140-8. : https://www.ff.umb.sk/app/cmsFile.php?disposition=a&ID=571 5. Fitness jóga, harmonické cvičení těla I duše. Praha: Grada. ISBN ARKOVSKÁ, M. 2005. Posilování s vlastním tělem 417 krát jinak. Praha:

STACKEOVÁ, D. 2014. Fitness programy z pohledu kinantropologie. Praha: Galén. ISBN 9788074921155.

VOMÁČKO, S. BOŠTÍKOVÁ, S. 2003. Lezení na umělých stěnách. Praha: Grada. 129s. ISBN 8024721743.

Course language:

Slovak language

Notes:

Course assessment

Total number of assessed students: 5839

abs	abs-A	abs-B	abs-C	abs-D	abs-E	n	neabs
82.51	0.27	0.03	0.0	0.0	0.0	8.25	8.92

Provides: Mgr. Marcel Čurgali, Mgr. Agata Dorota Horbacz, PhD., Mgr. Dávid Kaško, PhD., Mgr. Patrik Berta, Mgr. Ladislav Kručanica, PhD., Mgr. Richard Melichar, Mgr. Petra Tomková, PhD., Mgr. Alena Buková, PhD., univerzitná docentka, doc. PaedDr. Ivan Uher, MPH, PhD., prof. RNDr. Stanislav Vokál, DrSc., Mgr. Zuzana Küchelová, PhD.

Date of last modification: 07.02.2024

University: P. J. Š	Šafárik Univers	ity in Košice				
Faculty: Faculty	of Science					
Course ID: ÚGE/ STMG/21	E/ Course name: Statistical Methods in Geography					
Course type, scop Course type: Le Recommended Per week: 1 / 2 1 Course method:	cture / Practice course-load (h Per study perio	ours):				
Number of ECTS	S credits: 3					
Recommended se	emester/trimes	ster of the cours	e: 2.			
Course level: I.						
Prerequisities:						
Conditions for co	ourse completi	on:				
Learning outcom	nes:					
Brief outline of t	he course:					
Recommended li	terature:					
Course language	:					
Notes:						
Course assessme Total number of a		ts: 118				
А	В	С	D	Е	FX	
27.97	20.34	16.95	15.25	19.49	0.0	
Provides: prof. M docentka	lgr. Jaroslav Ho	ofierka, PhD., RN	NDr. Janetta Nest	orová-Dická, PhI	D., univerzitná	
Date of last modi	ification: 12.02	2.2023				
Approved: prof.]	Mgr. Jaroslav H	Iofierka, PhD., p	rof. RNDr. Vladi	mír Zeleňák, DrS	Sc.	

University:	ΡJ	Šafárik	University	in Košice
Chiver Siey.	1.0	Juluin	Chiverbicy	

Faculty: Faculty of Science

Course ID: ÚCHV/
MUSU/22Course name: Structure determination - spectroscopic methods

Course type, scope and the method: Course type: Lecture / Practice

Recommended course-load (hours):

Per week: 2 / 3 **Per study period:** 28 / 42

Course method: present

Number of ECTS credits: 5

Recommended semester/trimester of the course: 6.

Course level: I.

Prerequisities: (ÚCHV/ACHU/21 or ÚCHV/ACHU/03) and (ÚCHV/ANCHU/21 or ÚCHV/ ANCHU/03) and (ÚCHV/OCHU/21 or ÚCHV/OCHU/03)

Conditions for course completion:

1. Participation in exercises in accordance with the Study Rules of PF UPJŠ.

2. Successful execution of 3 control written works on exercises after 4., 8. and 12. weeks of teaching. Obtaining a minimum grade E from seminars.

The test consists of: 1. Solution of 2 structures of unknown compounds on the basis of combined application of spectral methods. 2. Theoretical and practical questions.

Percentage rating: 100-91% (A), 90-81% (B), 80-71% (C), 70-61% (D), 60-51% (E), 50% and less FX.

Learning outcomes:

Fundamentals of molecular spectroscopy and magnetic properties study, as powerful tools for structure determination in chemistry. Ultraviolet, visible, infrared and Raman spectroscopy, mass spectrometry and methods based on magnetic resonance (1H NMR, 13C NMR).

Brief outline of the course:

Fundamentals of molecular spectroscopy, mass spectrometry and magnetic methods as powerful tools for structure determination in chemistry. Ultraviolet and visible spectroscopy. Emission spectroscopy. Symmetry and group theory. Infrared and Raman spectroscopy. Mass spectrometry in organic and analytical chemistry and biochemistry. Nuclear magnetic resonance - NMR. Chemical shift and splitting of signals by spin-spin coupling. Coupling constants. 1H NMR, 13C NMR, NMR of other nuclei. Two- and more dimensional NMR. NMR applications. Methods and instruments used for spectra measurements. Combined application of spectral methods for solution of chemical problems.

Recommended literature:

1. Kováč Š., Ilavský D., Leško J.: Spektrálne metódy v organickej chémii a technológii, ALFA, Bratislava, 1987.

2. Milata V., Segl'a P.: Vybrané metódy molekulovej spektroskopie. STU BA, 2007.

3. Milata V., Segl'a P.: Spektrálne metódy v chémii. STU FCHPT Bratislava 2002.

4. Miertuš S. a kol.: Atómová a molekulová spektroskopia, ALFA, Bratislava 1991.

5. T. D. W. Claridge: High-Resolution NMR Techniques in Organic Chemistry, 5. Ed., Elsevier, 2016.

Course language:

slovak, english

Notes:

In-person course, alternatively online course using the BigBlueButton tool or MS Teams. The form of teaching is specified by the teacher at the beginning of the semester, updated continuously.

5						
Course assessment Total number of assessed students: 47						
А	A B C D E FX					
36.17	36.17	14.89	10.64	2.13	0.0	
	Provides: doc. RNDr. Ján Imrich, CSc., doc. RNDr. Juraj Kuchár, PhD., RNDr. Zuzana Kudličková, PhD., RNDr. Monika Tvrdoňová, PhD.					
Date of last modification: 16.08.2022						
Approved: prof	Approved: prof. Mgr. Jaroslav Hofierka, PhD., prof. RNDr. Vladimír Zeleňák, DrSc.					

University: P. J. Šaf	ărik University in Košice					
Faculty: Faculty of	Science					
Course ID: ÚGE/ SVG/04	Course name: Student Sci	Course name: Student Scientific Conference in Geography				
Course type, scope Course type: Recommended co Per week: Per stu Course method: p	urse-load (hours): dy period:					
Number of ECTS c	redits: 4					
Recommended sem	ester/trimester of the cours	e:				
Course level: I., II.						
Prerequisities:						
Conditions for cou	rse completion:					
Learning outcomes	:					
work on the topic, w	vic suggested by supervisors in vrite a thesis and defense it be	mplying a geographical problem, the students will efore the committee.				
Recommended liter	rature:					
Course language:						
Notes:						
Course assessment Total number of ass	essed students: 12					
	abs n					
	100.0 0.0					
Janetta Nestorová-D		lena Gessert, PhD., univerzitná docentka, RNDr. ntka, Mgr. Marián Kulla, PhD., doc. Ing. Katarína				
Date of last modifie	cation: 01.12.2021					
Approved prof Ma	r Jaroslav Hofierka PhD n	rof RNDr Vladimír Zeleňák DrSc				

University: P. J. Šafárik University in Košice					
Faculty: Faculty of Science					
ourse ID: ÚCHV/ VKB/04Course name: Students Scientific Conference					
Course type, scope and the method: Course type: Recommended course-load (hours): Per week: Per study period: Course method: present					
Number of ECTS credits: 4					
Recommended semester/trimester of the course	se:				
Course level: I.					
Prerequisities:					
Conditions for course completion: Present the results of student's work at the Studen committee members and others present.	t Scientific Conference and answer questions from				
Learning outcomes: The student will acquire competences for independent scientific work in the laboratory, for analysis and written processing of obtained results and knowledge. By presenting the obtained results, the student prepares to present the obtained results in the defense of the bachelor's thesis and in front of the professional public at scientific conferences.					
Brief outline of the course:					
Recommended literature:					
Course language:					
Notes:					
Course assessment Total number of assessed students: 12					
abs	n				
100.0	100.0 0.0				
Provides:					
Date of last modification: 22.07.2022					
Approved: prof. Mgr. Jaroslav Hofierka, PhD., J	prof. RNDr. Vladimír Zeleňák, DrSc.				

University: P. J. Šafá	rik University in Košice						
Faculty: Faculty of S							
Course ID: ÚFV/ DGS/21							
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	ce rse-load (hours): dy period: 28						
Number of ECTS cr	edits: 2						
Recommended seme	ster/trimester of the course: 1.						
Course level: I.							
Prerequisities:							
 Practical ongoing a Active participation 	based on ongoing assessment: assignments and their defense (at least 50% needed) on during face-to-face contact learning in classical or virtual classroom (3 nd during online learning (no absence, uploading all individual ongoing						
digital technologies (1. according to the cu	btain and know to apply basic knowledge and skills in working with current mobile phone, tablet, laptop, web technologies): urrent European framework for the Digital competence DigComp and ECDL re effective learning, work and active life in higher education, later lifelong career prospects.						
 modern web browse security, privacy, re 0305. Search, colled scanning, audio rece digital notebooks (C evaluation of digital 0608. Editing and c cloud and interactive (text and spreadsheet work with pdf docu (Kami, Google books 09 10. Organization modern LMS and c (Google Classroom, I) time management (skills, DigComp framework, ECDL er and its personalization sponsible use of DT ction and evaluation of digital content ording and speech resolution, optical resolution (OCR) Google keep, Evernote, Onenote) I resources (Google forms and sections) reating digital content e documents editors - Google, Microsoft, Jupyter) ments, e-books and videos s, Screencasting) n, protection and sharing of digital content loud storage Microsoft team, Google Drive, Dropbox)						

- collaborative interactive whiteboards (Jamboard, Whiteboard)

- online presentations and online meetings

(Google presentations, Powerpoint, Google meet, Microsoft teams)

Recommended literature:

1. Carretero Gomez, S., Vuorikari, R. and Punie, Y., DigComp 2.1: The Digital Competence Framework for Citizens with eight proficiency levels and examples of use, Luxembourg, 2017, ISBN 978-92-79-68006-9, https://www.ecdl.sk/

2. Bruff, D. (2019). Intentional Tech: Principles to Guide the Use of Educational Technology in College Teaching (1st edition). Morgantown: West Virginia University Press.

3. Baker, Y. (2020). Microsoft Teams for Education. Amazon Digital Services.

4. Miller, H. (2021). Google Classroom + Google Apps: 2021 Edition. Brentford: Orion Edition Limited.

Course language:

slovak

Notes:

Notes:						
Course assessment Total number of assessed students: 163						
A	В	С	D	Е	FX	
69.33	4.29	4.29	0.0	22.09	0.0	
Provides: doc.	Provides: doc. RNDr. Jozef Hanč, PhD.					
Date of last modification: 26.01.2022						
Approved: prof	Approved: prof. Mgr. Jaroslav Hofierka, PhD., prof. RNDr. Vladimír Zeleňák, DrSc.					

University: P. J. Šafá	rik University in Košice
Faculty: Faculty of S	
Course ID: ÚTVŠ/ LKSp/13	Course name: Summer Course-Rafting of TISA River
Course type, scope a Course type: Practic Recommended cour Per week: 2 Per stu Course method: pre	ce rse-load (hours): dy period: 28
Number of ECTS cr	edits: 2
Recommended seme	ster/trimester of the course:
Course level: I., II.	
Prerequisities:	
- active participation	sful course completion: in line with the study rule of procedure and course guidelines ce of all tasks: carrying a canoe, entering and exiting a canoe, righting a canoe,
course syllabus and r Performance standard Upon completion of t - implement the acqu - implement basic ski - determine the right	the course students are able to meet the performance standard and: ired knowledge in different situations and practice, ills to manipulate a canoe on a waterway,
5. Canoe lifting and c	ourse: iculty of waterways iting ning using an empty canoe carrying n the water without a shore contact be out of the water

12. Commands

Recommended literature:

1. JUNGER, J. et al. Turistika a športy v prírode. Prešov: FHPV PU v Prešove. 2002. ISBN 8080680973.

Internetové zdroje:

1. STEJSKAL, T. Vodná turistika. Prešov: PU v Prešove. 1999.

Dostupné na: https://ulozto.sk/tamhle/UkyxQ2lYF8qh/name/Nahrane-7-5-2021-v-14-46-39#! ZGDjBGR2AQtkAzVkAzLkLJWuLwWxZ2ukBRLjnGqSomICMmOyZN==

Course language:

Slovak language

Notes:

Course assessment

Total number of assessed students: 232

abs	n
36.64	63.36

Provides: Mgr. Dávid Kaško, PhD.

Date of last modification: 29.03.2022

University: P. J. Ša	afárik Univers	ity in Košice					
Faculty: Faculty of	f Science						
Course ID: KPE/ SSU/15	/ Course name: Teachers' Support Groups						
Course type, scope Course type: Prace Recommended co Per week: 2 Per s Course method:]	ctice burse-load (h study period: present	ours):					
Number of ECTS							
Recommended ser		ter of the cours	e: 6.				
Course level: I., II.							
Prerequisities:							
Conditions for cou	ırse completi	on:					
Learning outcome	es:						
Brief outline of the	e course:						
Recommended lite	erature:						
Course language:							
Notes:							
Course assessmen Total number of as		ts: 59					
A	В	С	D	Е	FX		
88.14	10.17	0.0	0.0	0.0	1.69		
Provides: doc. Pae	dDr. Renáta C	Prosová, PhD., M	gr. Zuzana Vaga	ská, PhD.			
Date of last modif	ication: 12.03	.2024					
Approved: prof. M	lgr. Jaroslav H	lofierka, PhD., p	rof. RNDr. Vlad	imír Zeleňák, Dr	Sc.		

University: P. J. Šafa	árik University in Košic	3				
Faculty: Faculty of S	Science					
Course ID: KPPaPZ/ECo-C1/14	Course name: Team Work ECo-C1					
Course type, scope a Course type: Practa Recommended cou Per week: 2 Per sta Course method: pr	ice irse-load (hours): udy period: 28					
Number of ECTS cr	redits: 4					
Recommended sem	ester/trimester of the c	burse: 4., 6.				
Course level: I.						
Prerequisities:						
Conditions for cour	se completion:					
Learning outcomes:						
Brief outline of the	course:					
Recommended liter	ature:					
Course language:						
Notes:						
Course assessment Total number of asse	essed students: 142					
	abs	n				
	97.89 2.11					
Provides: PhDr. Anr	na Janovská, PhD.					
Date of last modific	ation: 14.09.2024					
Approved: prof. Mg	r. Jaroslav Hofierka, Ph	D., prof. RNDr. Vladimír Zeleňák, DrSc.				

University: P. J. Ša	fárik Univers	ity in Košice						
Faculty: Faculty of	Science							
Course ID: KPE/ TVE/08	Course na	Course name: Theory of Education						
Course type, scope Course type: Prac Recommended co Per week: 2 Per s Course method: p	tice ourse-load (he tudy period:	ours):						
Number of ECTS	credits: 2							
Recommended sen	nester/trimes	ter of the cours	e: 4., 6.					
Course level: I.								
Prerequisities:								
Conditions for cou	rse completi	o n:						
Learning outcome	s:							
Brief outline of the	e course:							
Recommended lite	rature:							
Course language:								
Notes:								
Course assessment Total number of ass		ts: 678						
A	В	С	D	Е	FX			
45.13	30.24	16.08	4.72	1.92	1.92			
Provides: Mgr. Kat	arína Petríkov	vá, PhD., Mgr. B	eáta Sakalová, P	hD.	1			
Date of last modifi	cation: 12.03	.2024						
Approved: prof. M	gr. Jaroslav H	lofierka, PhD., p	rof. RNDr. Vladi	mír Zeleňák, Dr	Sc.			